

Steric and Chelating Ring Concerns on the L-Lactide Polymerization by Asymmetric β -Diketiminato Zinc Complexes

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Table S1. Crystal Data and Structure Refinement Parameters

	L¹H	1 [L¹ZnEt]₂	2 [L²ZnEt]	3 [L³ZnEt]₂	4 [L⁴ZnEt]
empirical formula	C ₂₃ H ₃₁ N ₃	C ₄₄ H ₅₈ N ₆ Zn ₂	C ₂₃ H ₃₁ N ₃ Zn	C ₅₀ H ₇₀ N ₆ Zn ₂	C ₂₆ H ₃₇ N ₃ Zn
formula weight	349.51	801.70	414.88	885.86	456.96
<i>T</i> (K)	200(2)	200(2)	200(2)	200(2)	200(2)
crystal size (mm ³)	0.68 x 0.55 x 0.08	0.23 x 0.19 x 0.06	0.18 x 0.17 x 0.08	0.35 x 0.31 x 0.24	0.20 x 0.13 x 0.05
crystal system	Triclinic	triclinic	monoclinic	triclinic	monoclinic
space group	P -1	P-1	P2(1)/n	P-1	P2(1)/n
<i>a</i> (Å)	9.3914(8)	8.137(9)	10.6411(6)	9.6519(7)	8.8377(8)
<i>b</i> (Å)	10.7363(10)	9.251(10)	14.0394(7)	9.7012(7)	34.635(3)
<i>c</i> (Å)	11.2888(9)	14.151(15)	14.2237(8)	14.3904(11)	9.0028(9)
<i>α</i> (deg)	99.922(4)	79.83(2)	90	76.206(3)	90
<i>β</i> (deg)	100.271(4)	78.73(2)	97.421(2)°	88.607(3)	118.1850(10)°
<i>γ</i> (deg)	101.376(5)	76.91(2)	90	63.321(3)	90
<i>V</i> (Å ³)	1072.30(16)	1007.6(18)	2107.1(2)	1163.94(15)	2428.9(4)
<i>Z</i>	2	1	4	1	4
<i>D</i> _{calcd} (g cm ⁻³)	1.082	1.321	1.308	1.264	1.250
<i>μ</i> (mm ⁻¹)	0.064	1.229	1.177	1.070	1.028
reflns mcsd/indep	7157 / 3667	6025/3483	11629/3726	9718/4053	15366/4249
data/rcstraints/params	3667 / 0 / 235	3483/0/235	3726/0/244	4053/0/262	4249/0/271
GOF	1.075	0.997	1.042	1.037	1.025
<i>R</i> _{int}	0.0388	0.0671	0.0337	0.0188	0.0398
<i>R</i> ₁ [<i>I</i> > 2σ](all data)	0.0886 (0.2536)	0.0866 (0.2035)	0.0374(0.0982)	0.0310 (0.0824)	0.0437(0.0979)
<i>R</i> _w [<i>I</i> > 2σ](all data)	0.1011 (0.2630)	0.1526(0.2400)	0.0493(0.1044)	0.0339(0.0842)	0.0625(0.1053)
max. peak/hole (e ⁻ / Å ³)	0.450/ -0.500	1.185/ -1.130	0.544/ -0.400	0.653/ -0.606	0.429/ -0.365

Table S2 Kinetic study of *L*-lactide polymerization with various Zn complexes

	1	2	3	4
Time (min)	Conv. of <i>L</i> -lactide			
1	0.32	0.38		
2	0.49	0.41		
4	0.67	0.51		
8		0.71		
12	0.97			
15	0.99	0.84		
20			0.39	
30			0.45	
37				0.45
60			0.54	
80			0.60	
87				0.55
107				0.58
127				0.61
160				0.65
190				0.68
220				0.71

$[LA]_0/[Zn + BnOH] = 100$; $[LA]_0 = 1.00$ M in CH_2Cl_2 for **1-3** or $[LA]_0/[Zn + BnOH] = 50$; $[LA]_0 = 1.32$ M in $CDCl_3$ for **4**.

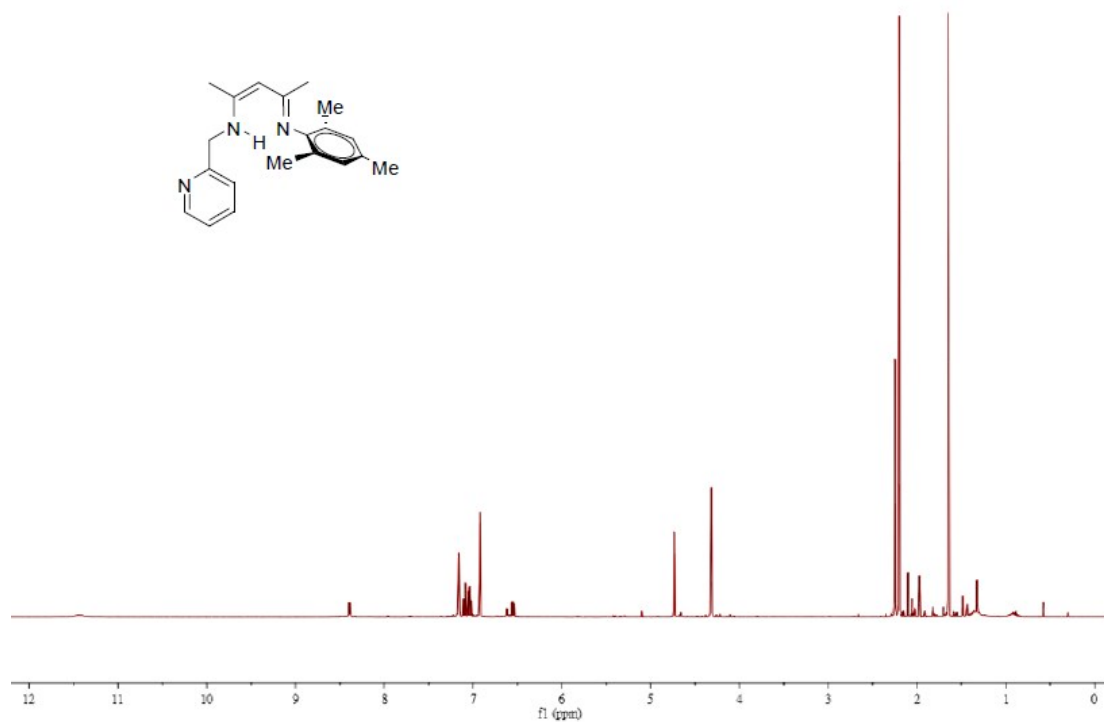


Figure S1. ^1H NMR spectrum of **HL¹** in C_6D_6 (400 MHz).

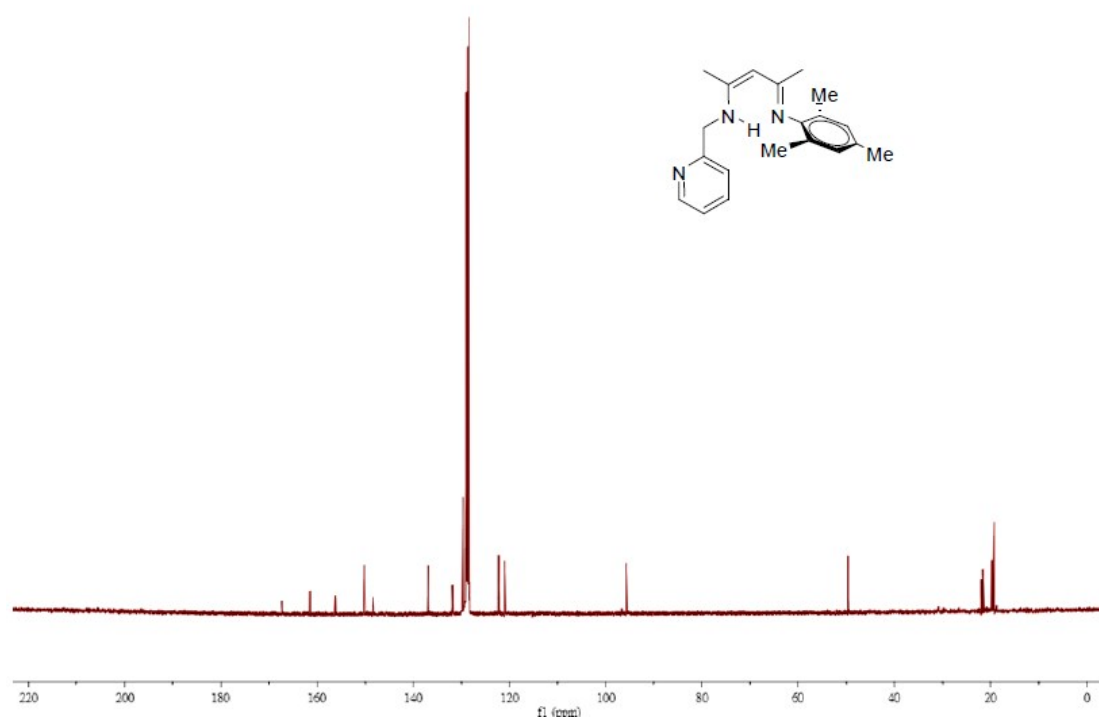


Figure S2. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **HL¹** in C_6D_6 (100 MHz).

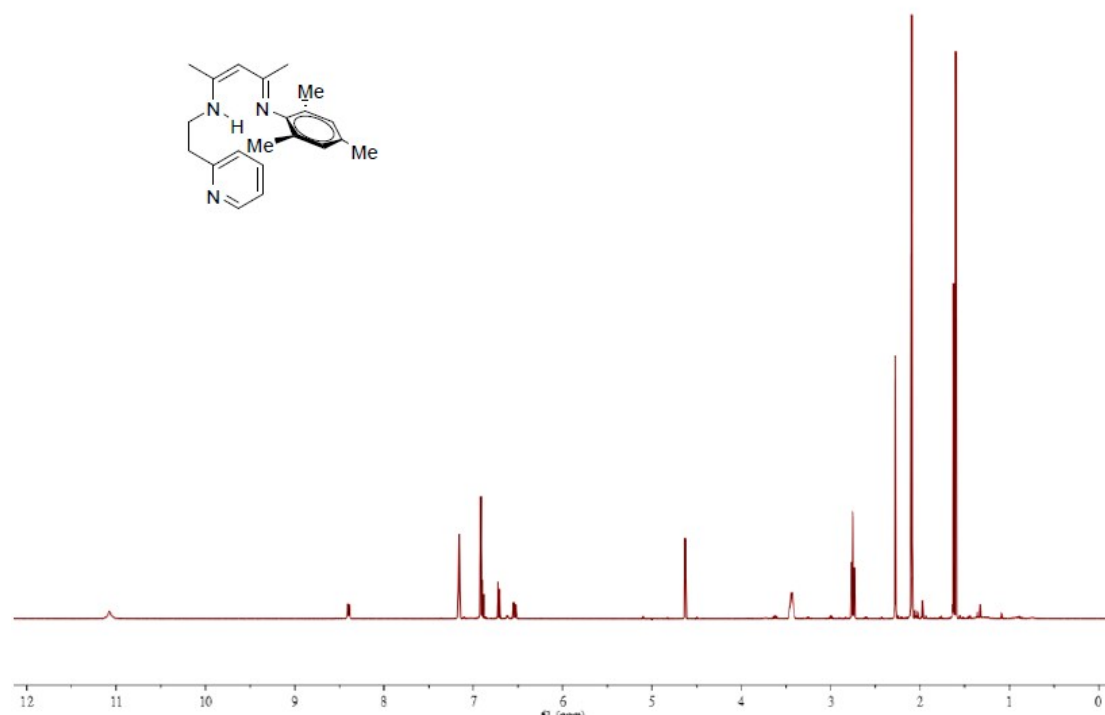


Figure S3. ¹H NMR spectrum of **HL²** in C₆D₆ (400 MHz).

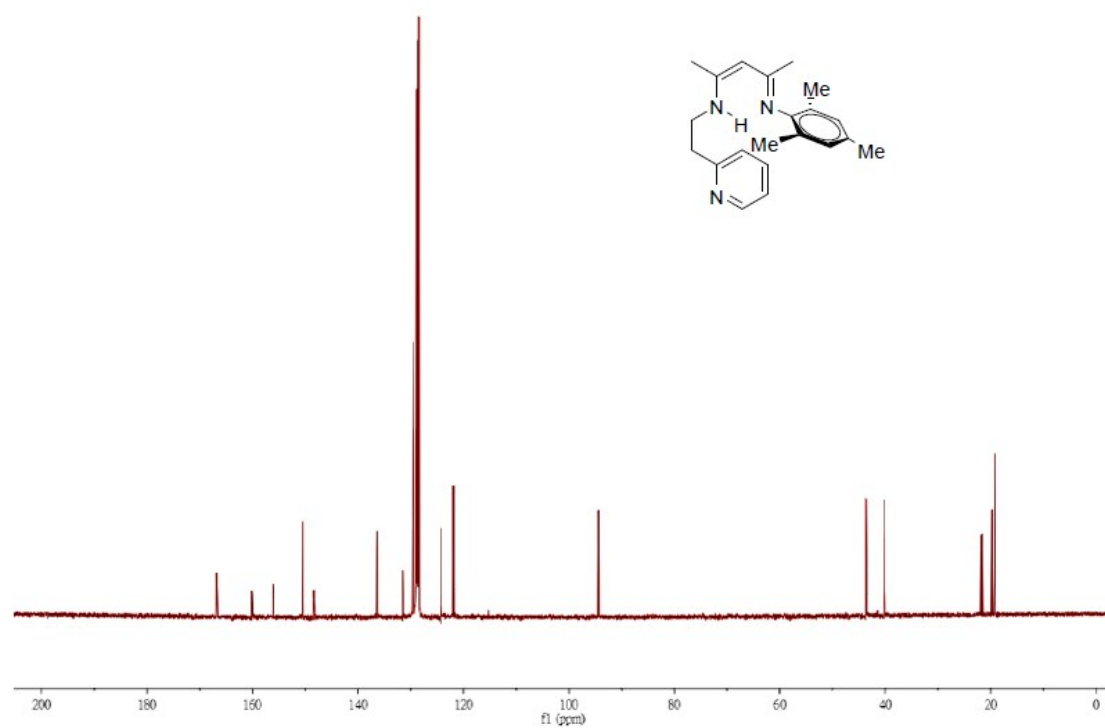


Figure S4. ¹³C {¹H} NMR spectrum of **HL²** in C₆D₆ (100 MHz).

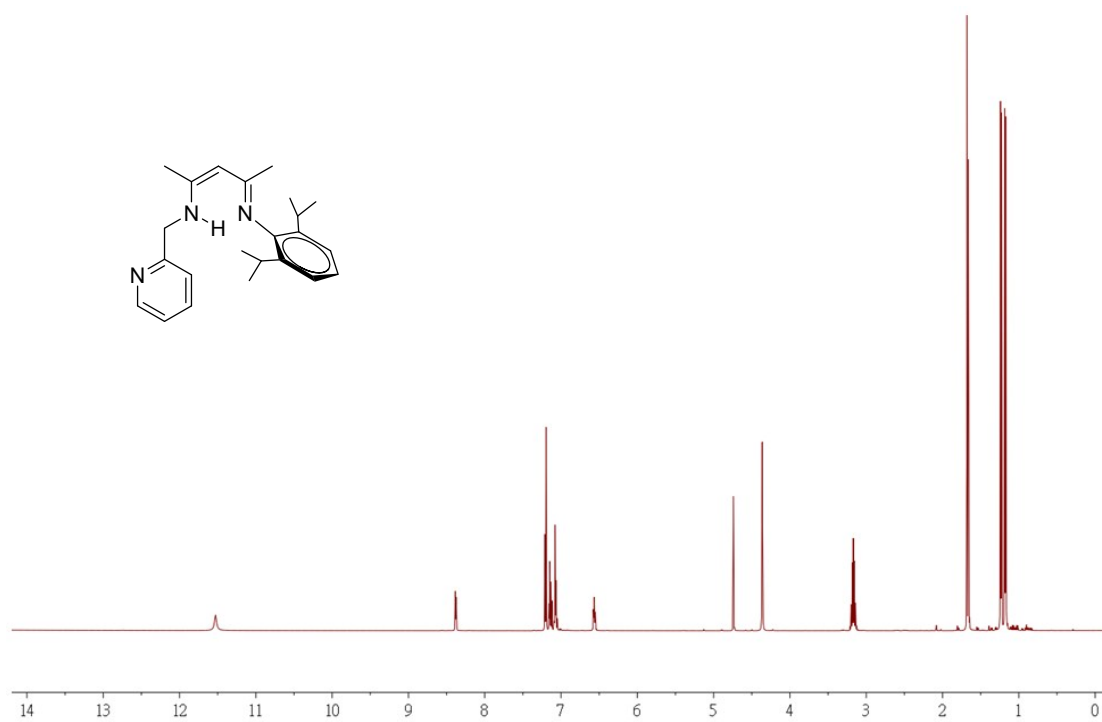


Figure S5. ¹H NMR spectrum of **HL³** in C₆D₆ (500 MHz).

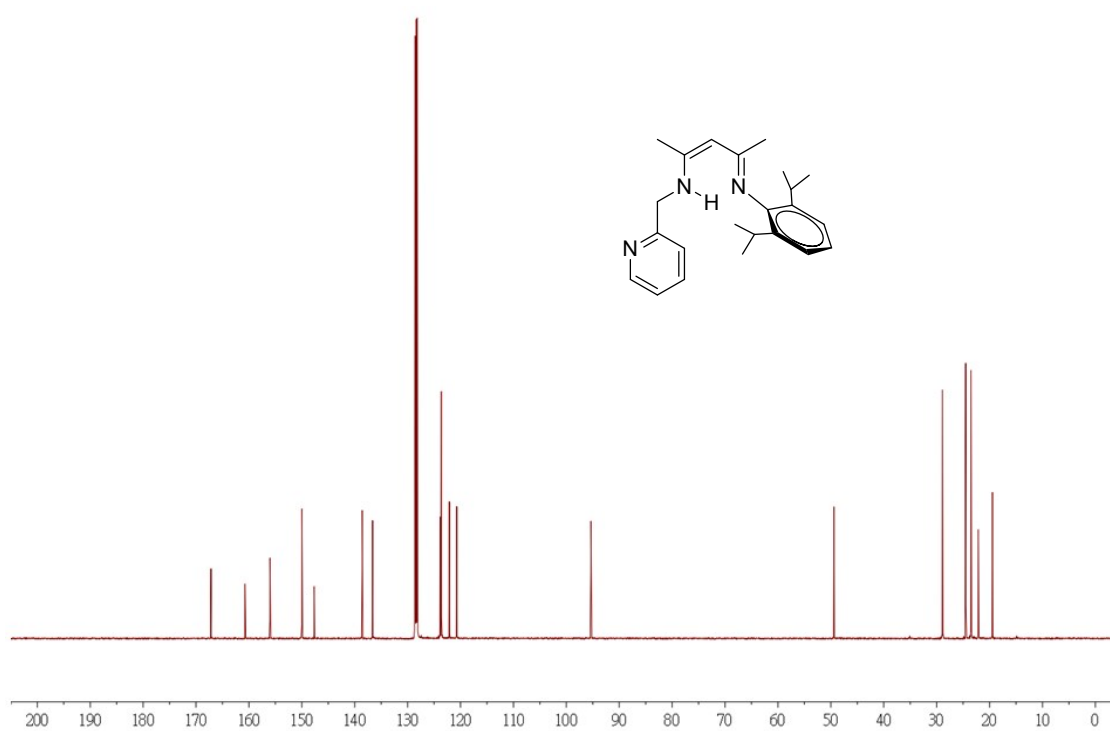


Figure S6. ¹³C {¹H} NMR spectrum of **HL³** in C₆D₆ (125 MHz).

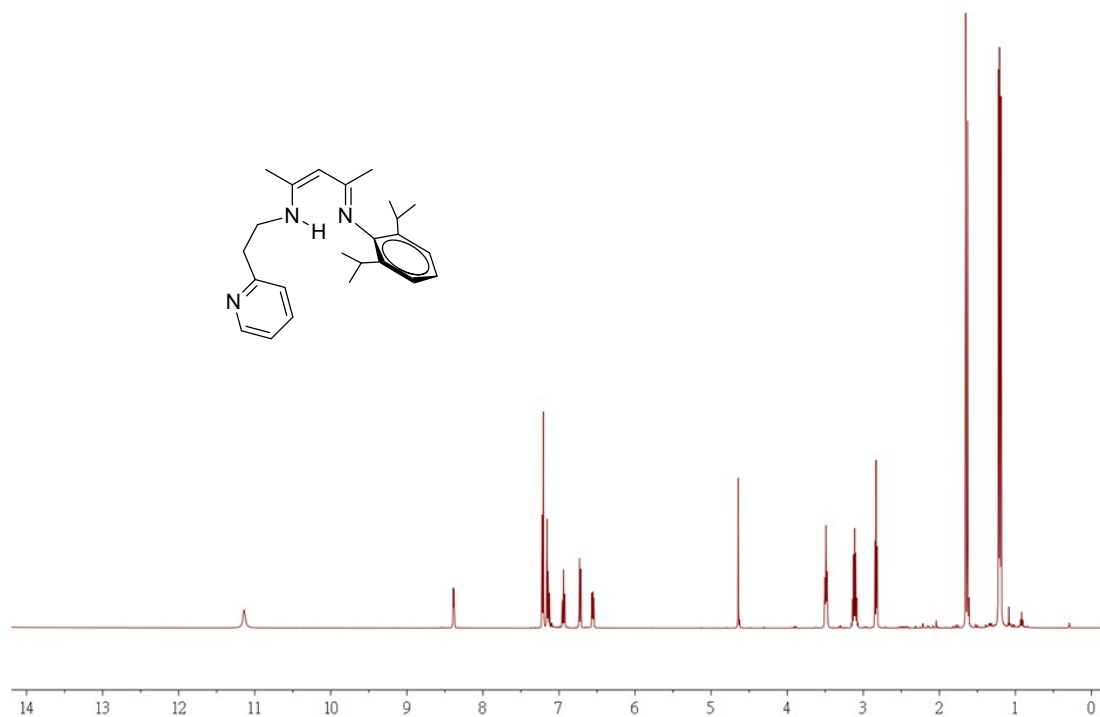


Figure S7. ^1H NMR spectrum of **HL⁴** in C_6D_6 (500 MHz).

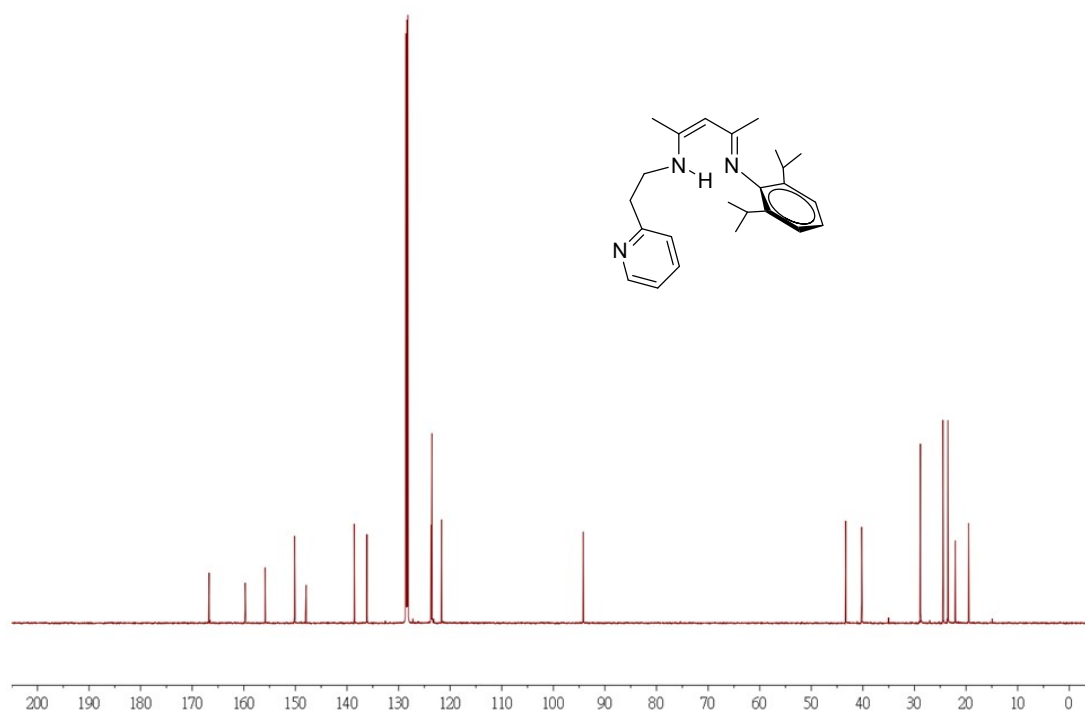


Figure S8. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **HL⁴** in C_6D_6 (125 MHz).

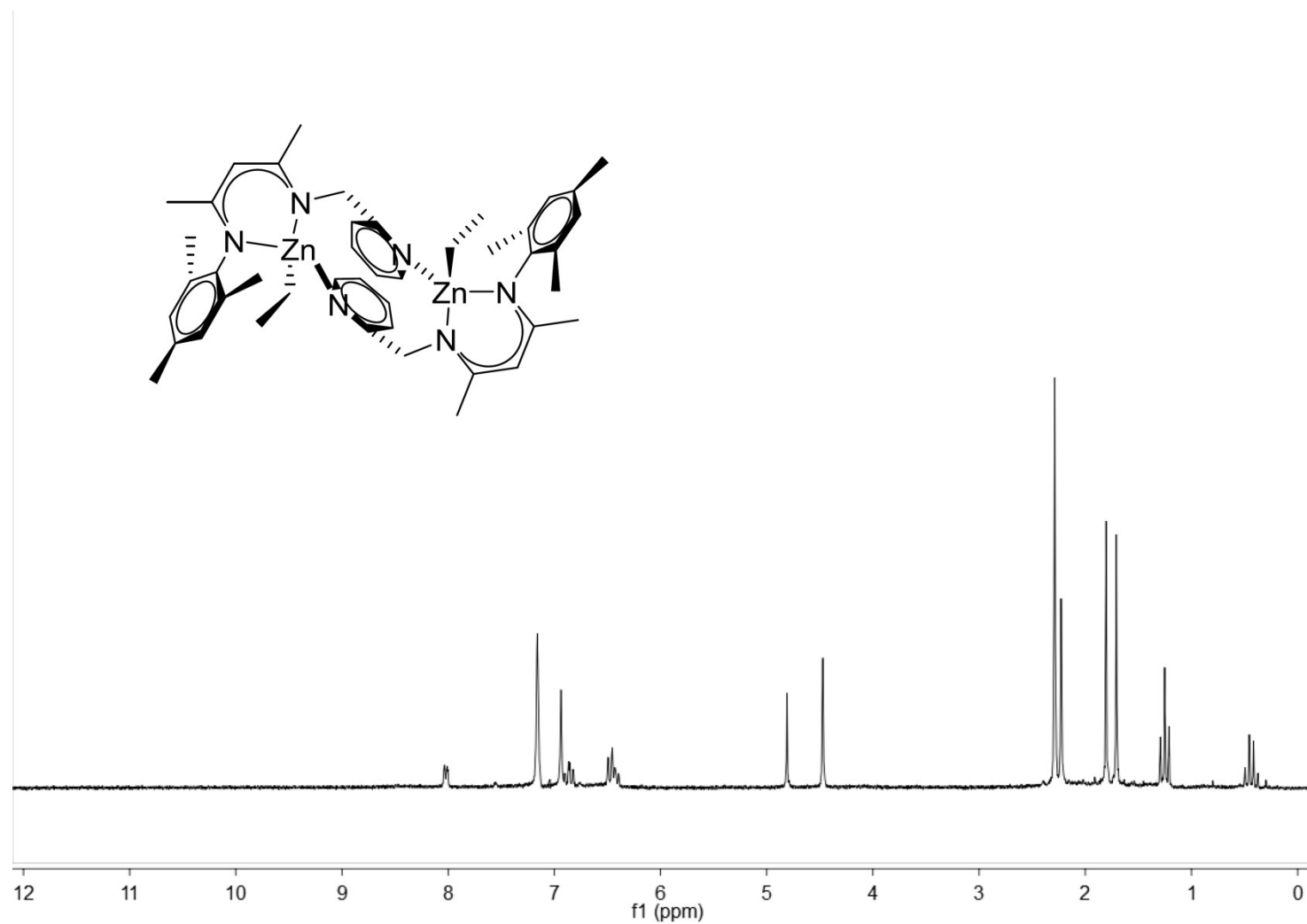


Figure S9. ^1H NMR spectrum of **1** in C_6D_6 (400 MHz).

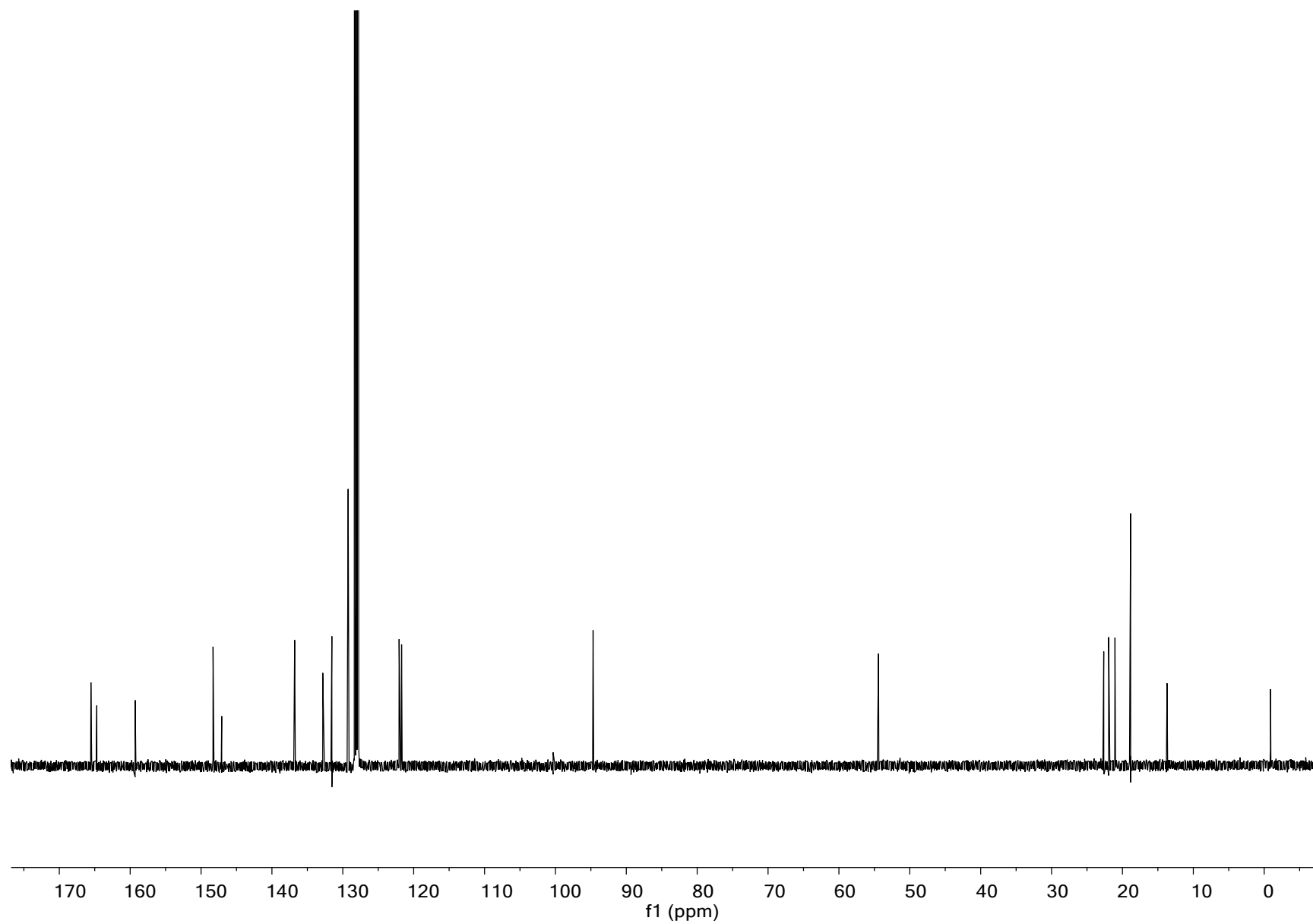


Figure S10. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **1** in C_6D_6 (100 MHz).

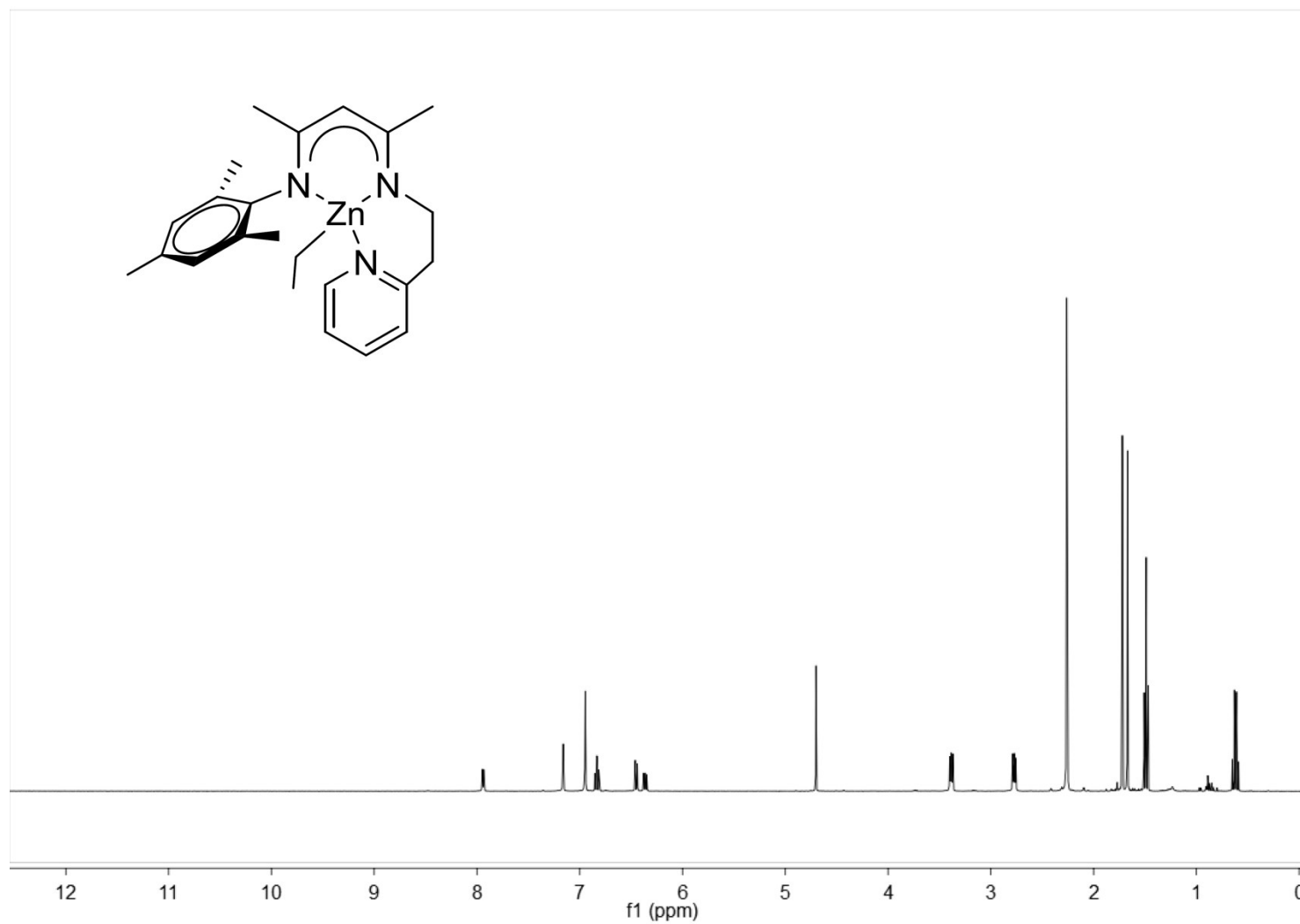


Figure S11. ^1H NMR spectrum of **2** in C_6D_6 (400 MHz).

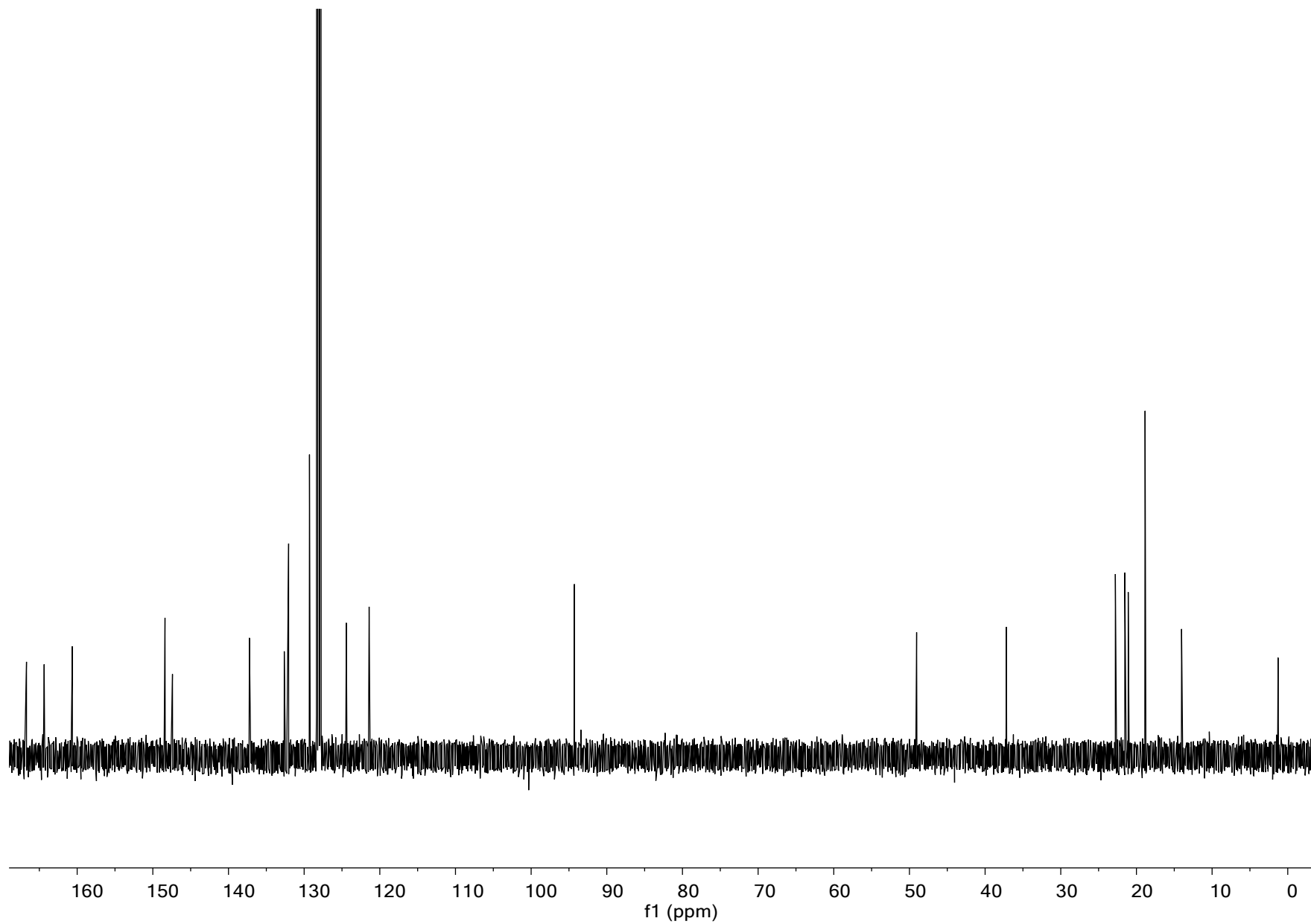


Figure S12. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **2** in C_6D_6 (100 MHz).

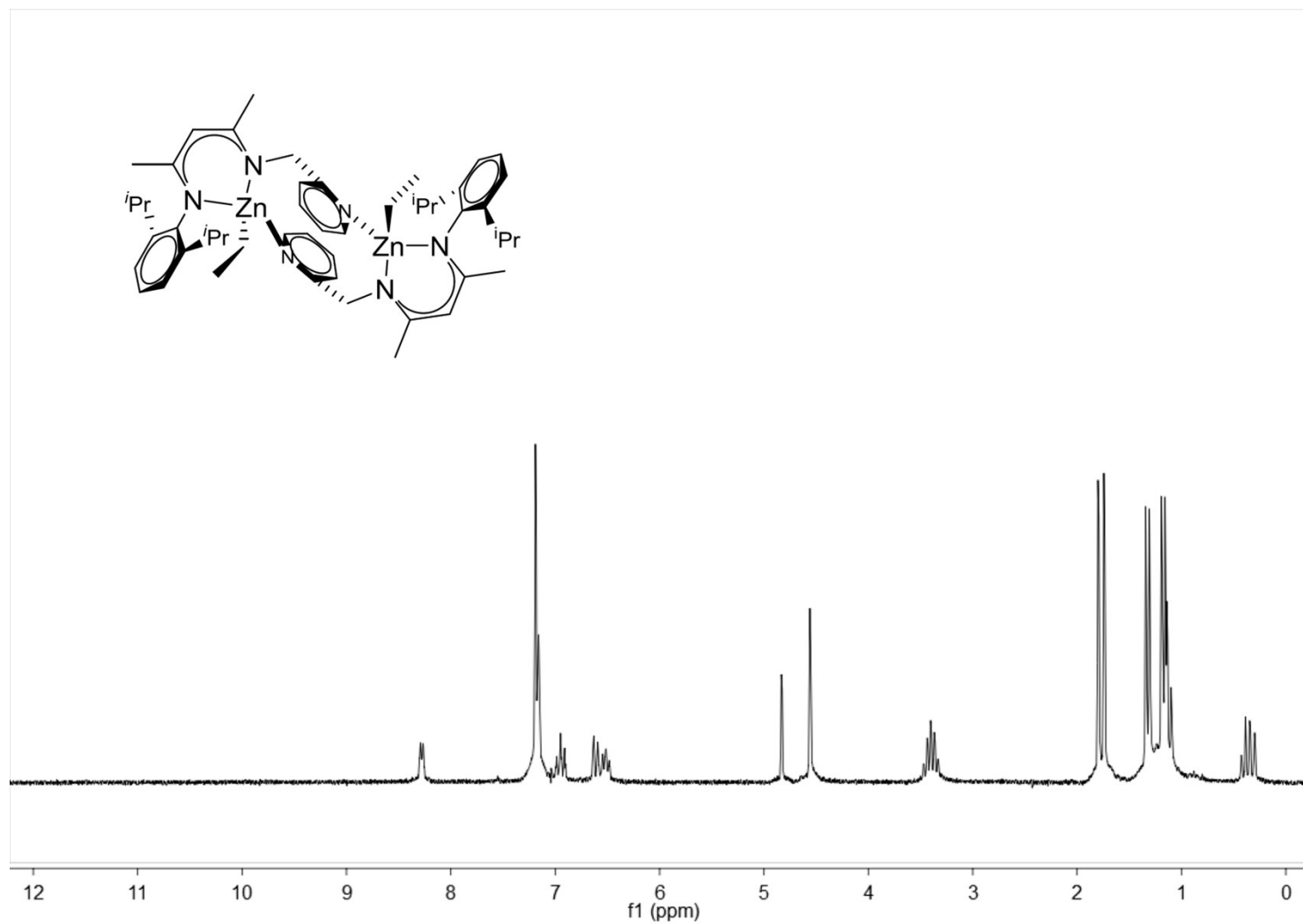


Figure S13. ^1H NMR spectrum of **3** in C_6D_6 (400 MHz).

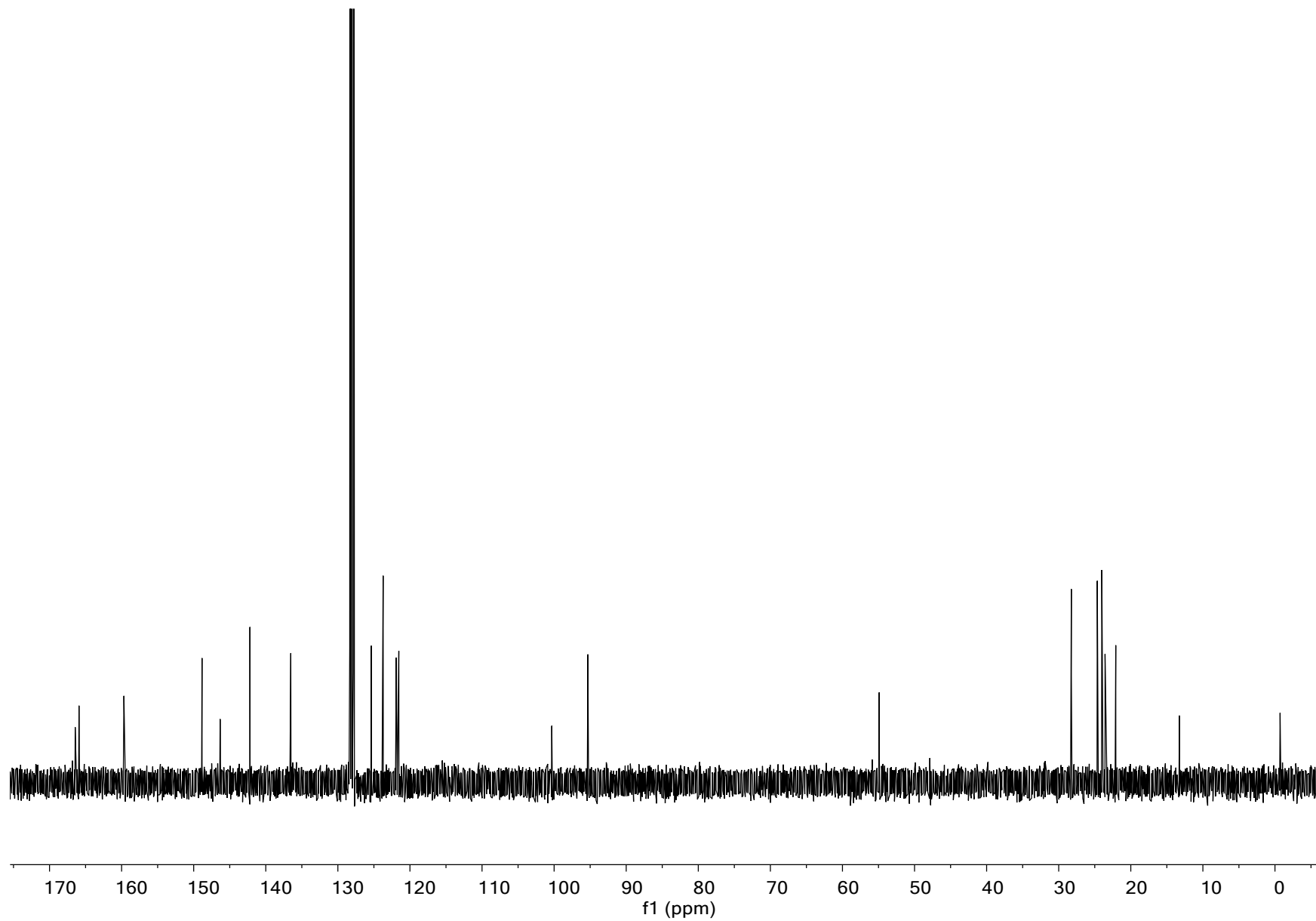


Figure S14. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **3** in C_6D_6 (100 MHz).

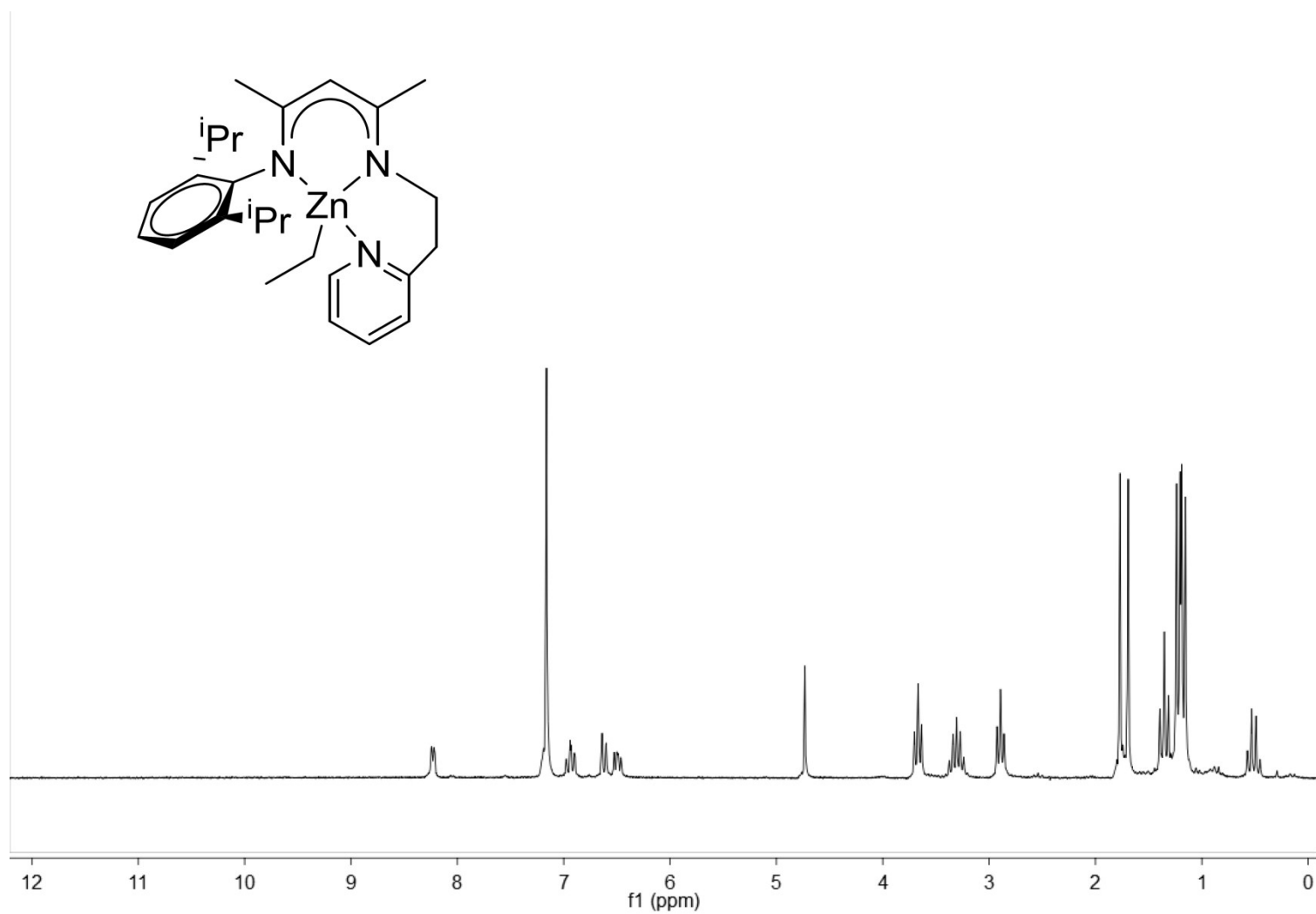


Figure S15. ^1H NMR spectrum of **4** in C_6D_6 (400 MHz).

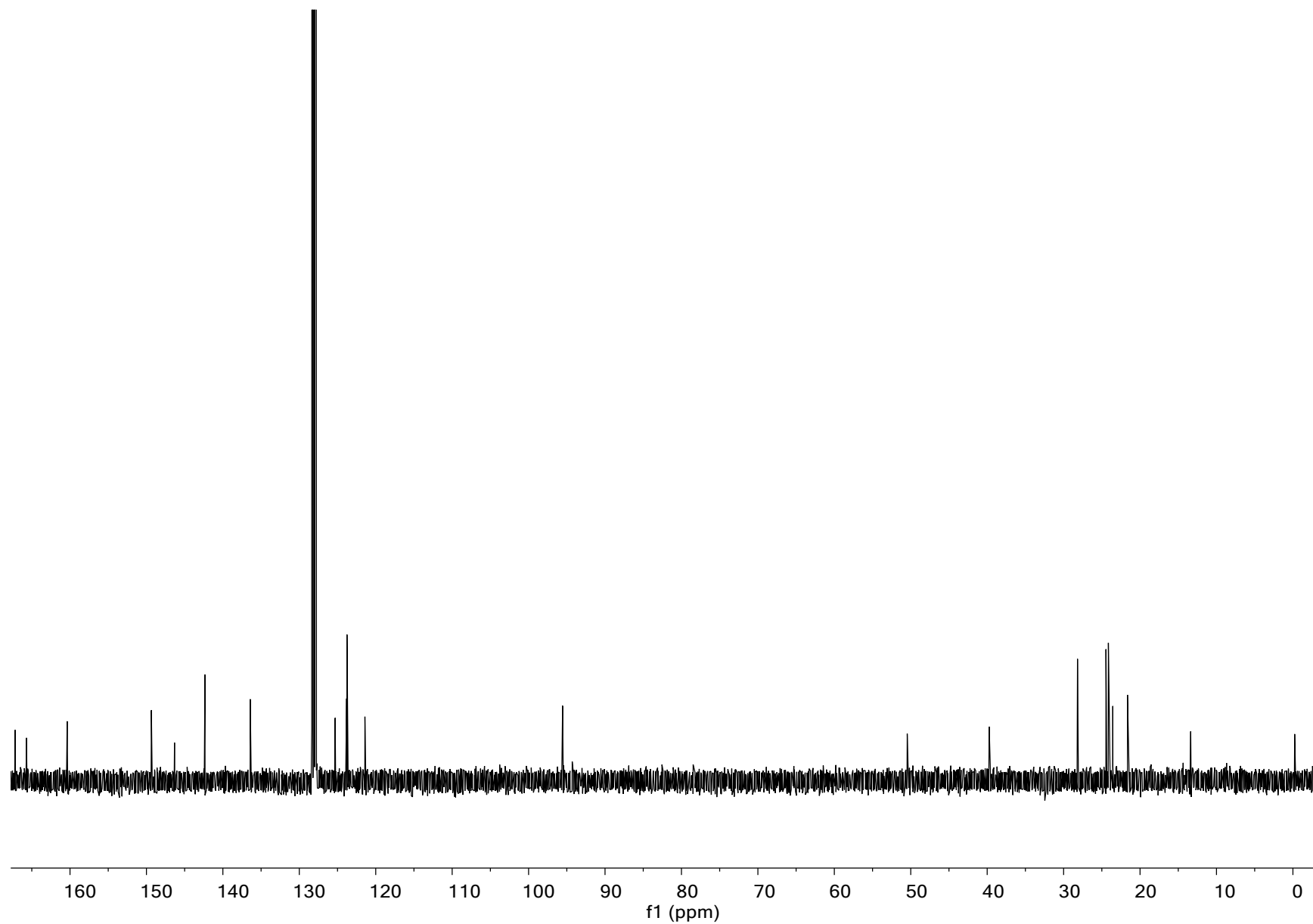


Figure S16. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **4** in C_6D_6 (100 MHz).

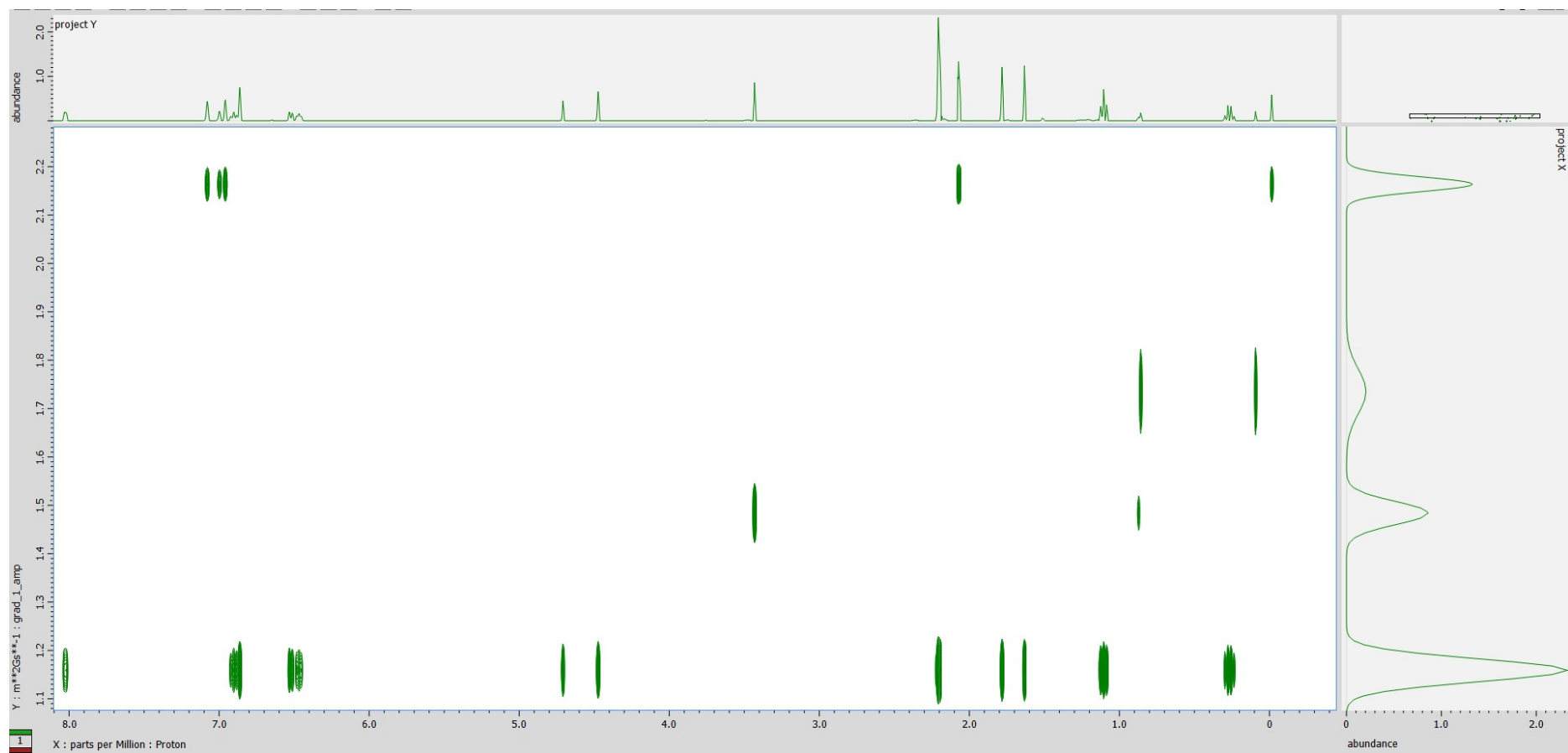
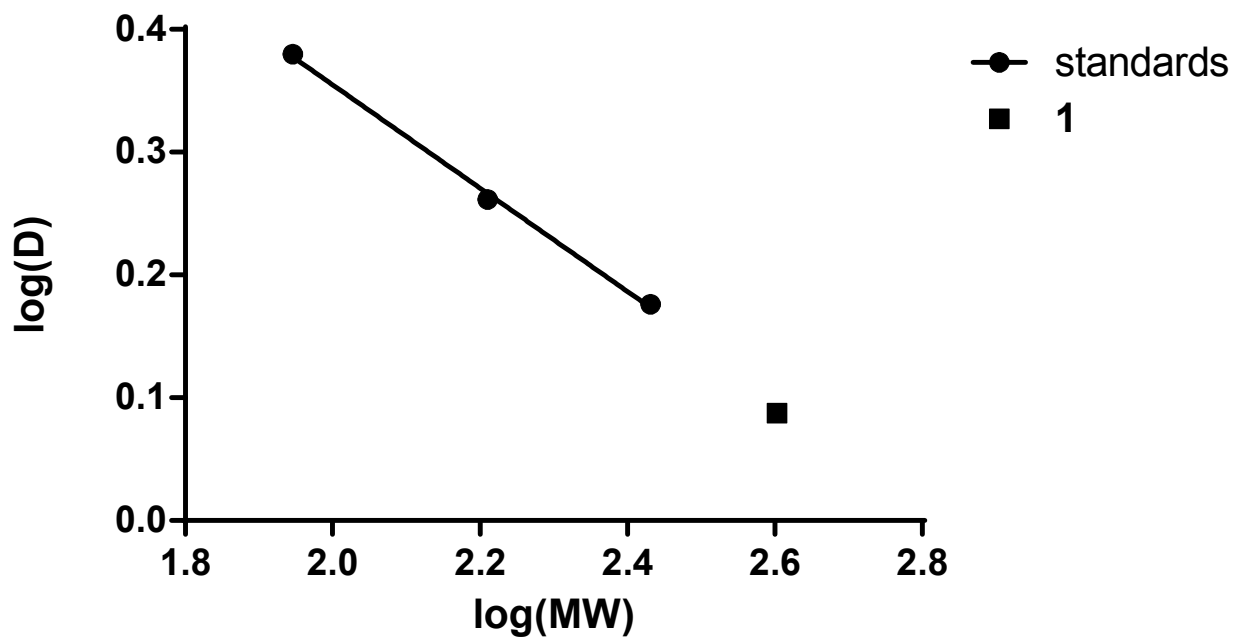
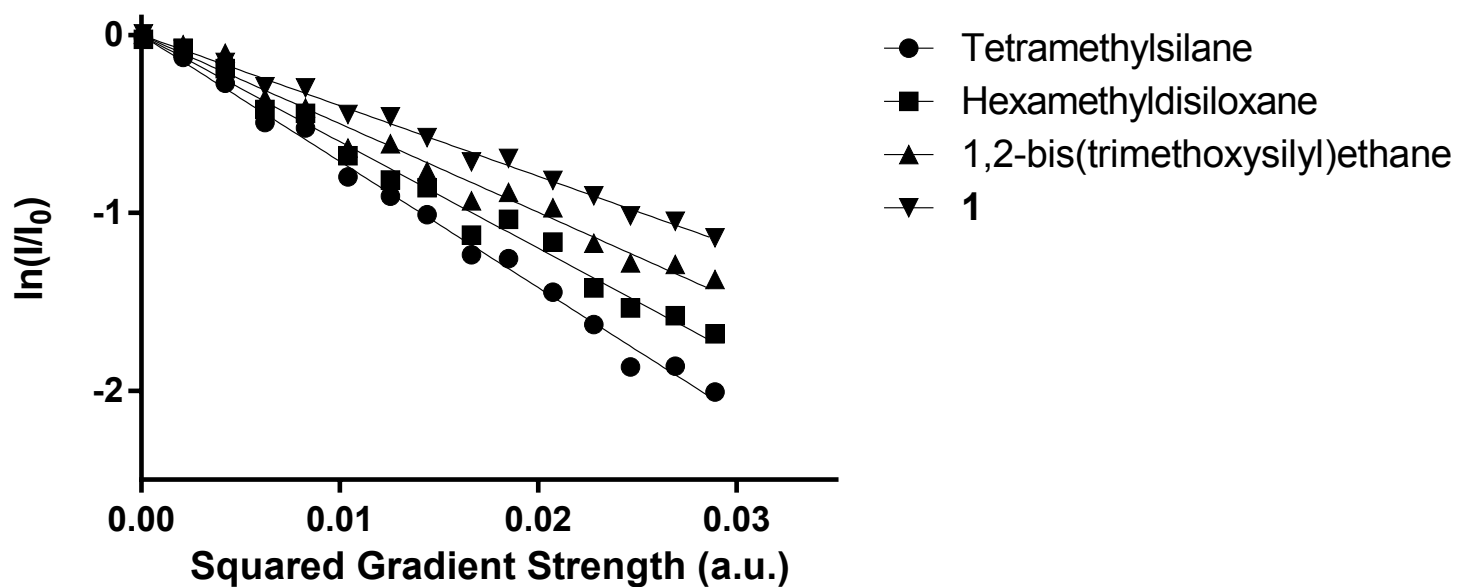


Figure S17. ^1H DOSY spectrum of **1** and three internal references in toluene- d_8 .



Equation	$Y = -0.4207 \cdot X + 1.196$
R square	0.9983
MW from eq.	431.84
ERROR	7.71%

Figure S18. D-FW analysis of ^1H DOSY data. Internal references are shown as solid circles and complex **1** is shown as solid square.



complex	Equation	R square
Tetramethylsilane	$Y = -70.97 \cdot X - 0.0008717$	0.9938
Hexamethyldisiloxane	$Y = -59.93 \cdot X - 0.0004703$	0.9867
1,2-bis(trimethoxysilyl)ethane	$Y = -49.83 \cdot X - 0.0006224$	0.9802
1	$Y = -39.69 \cdot X - 0.0004944$	0.9935

Figure S19. Stejskal-Tanner plot of experimental peaks for complex **1** and their best fit lines.

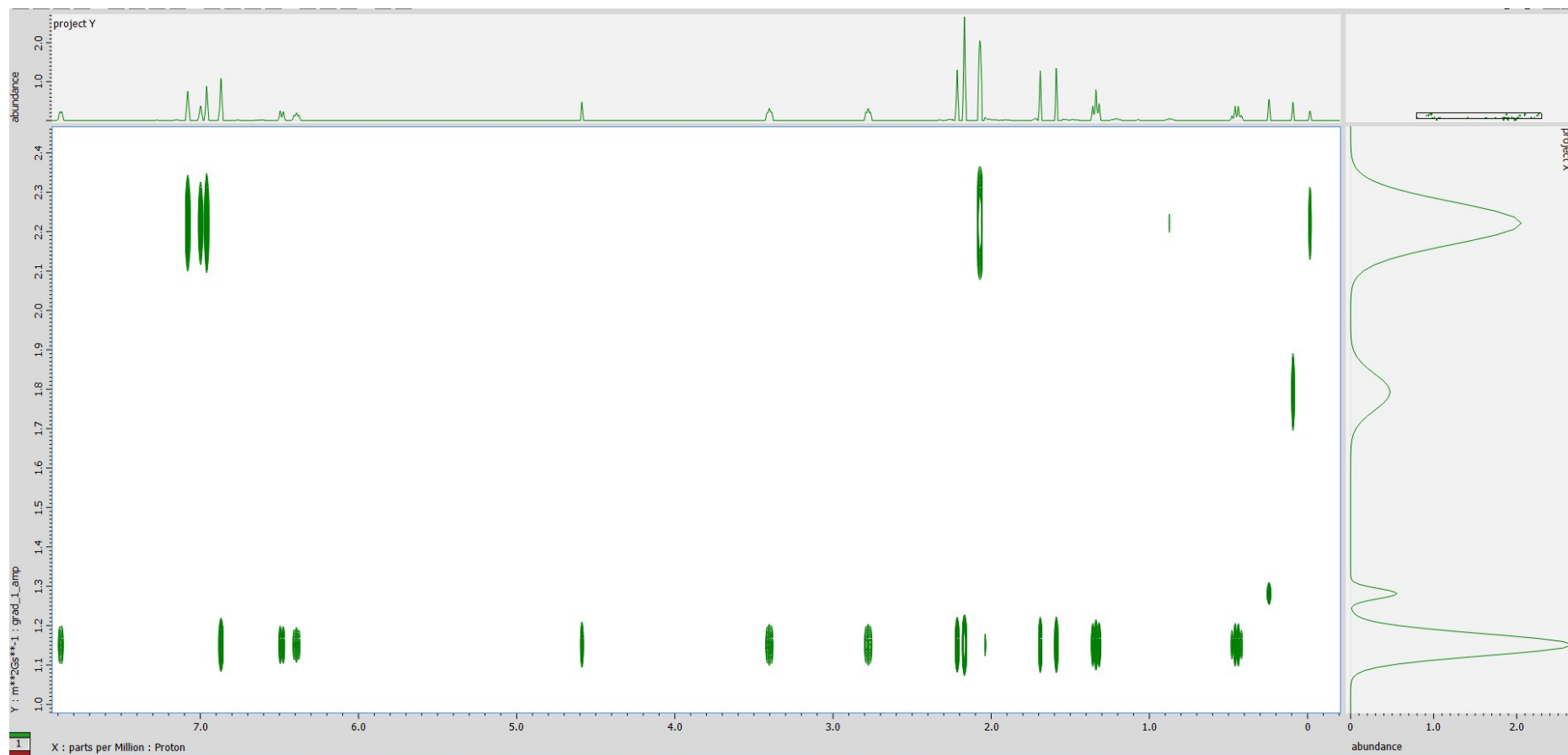
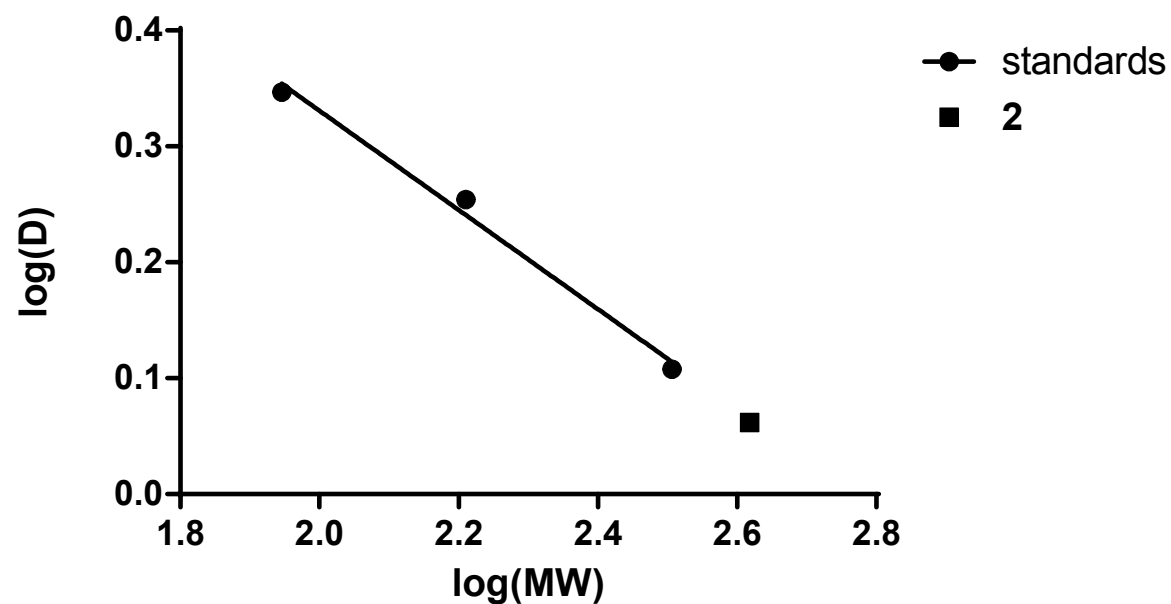
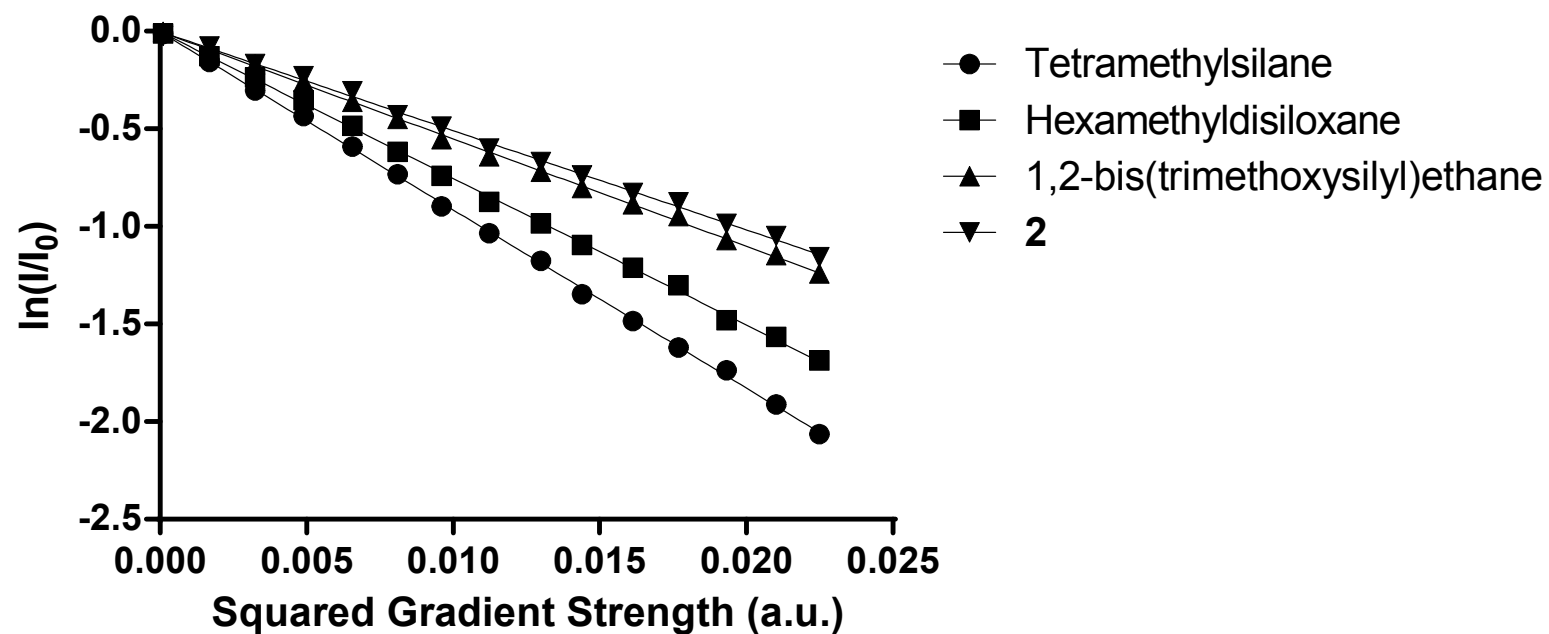


Figure S20. ^1H DOSY spectrum of **2** and three internal references in toluene- d_8 .



Equation	$Y = -0.4278 \cdot X + 1.186$
R square	0.9907
MW from eq.	425.09
ERROR	2.45%

Figure S21. D-FW analysis of ^1H DOSY data. Internal references are shown as solid circles and complex **2** is shown as solid square.



complex	Equation	R square
Tetramethylsilane	$Y = -91.24 \cdot X - 0.002636$	0.9995
Hexamethyldisiloxane	$Y = -75.13 \cdot X - 0.002174$	0.9991
1,2-bis(trimethoxysilyl)ethane	$Y = -54.98 \cdot X - 0.001595$	0.9989
1	$Y = -50.78 \cdot X - 0.001481$	0.9978

Figure S22. Stejskal-Tanner plot of experimental peaks for complex **2** and their best fit lines.

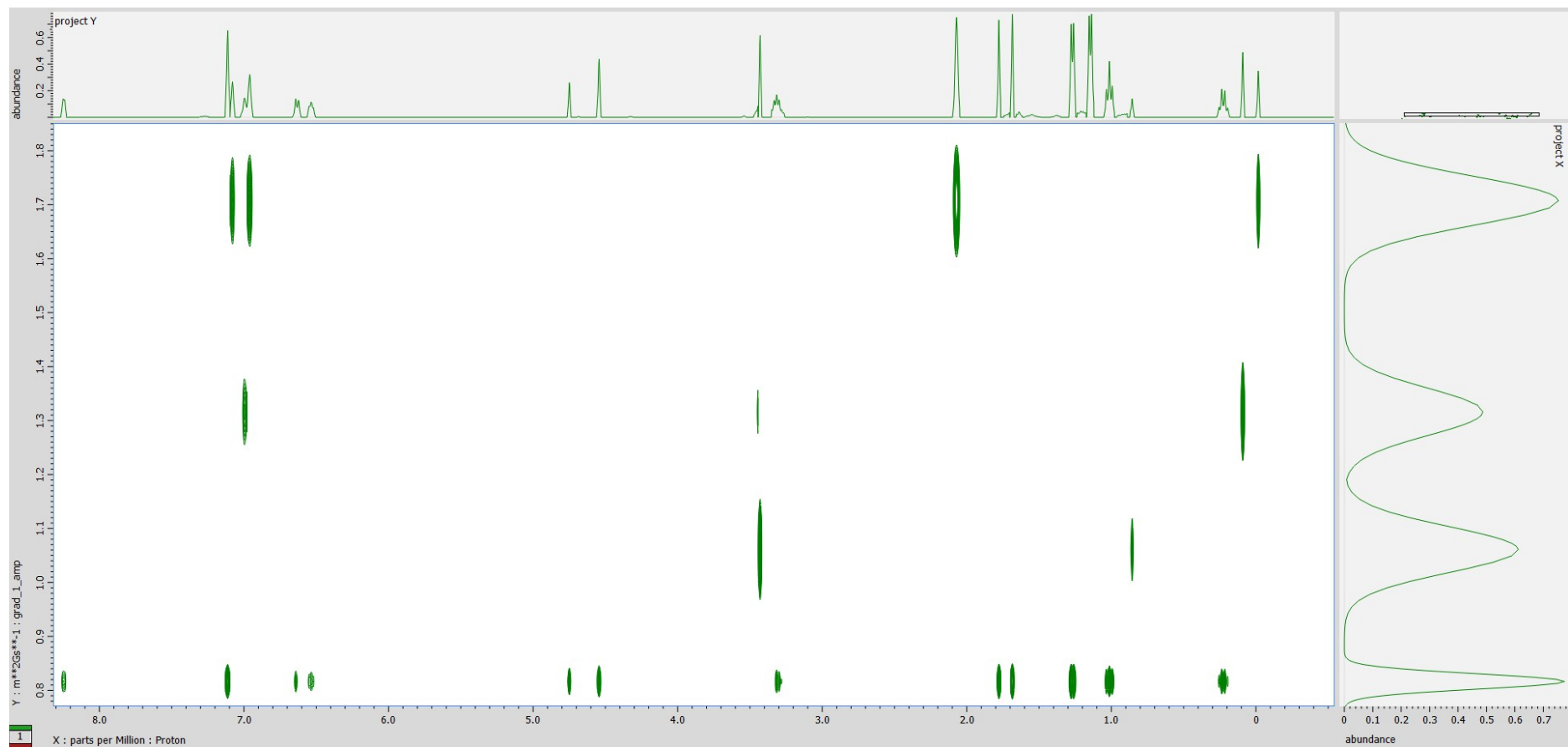
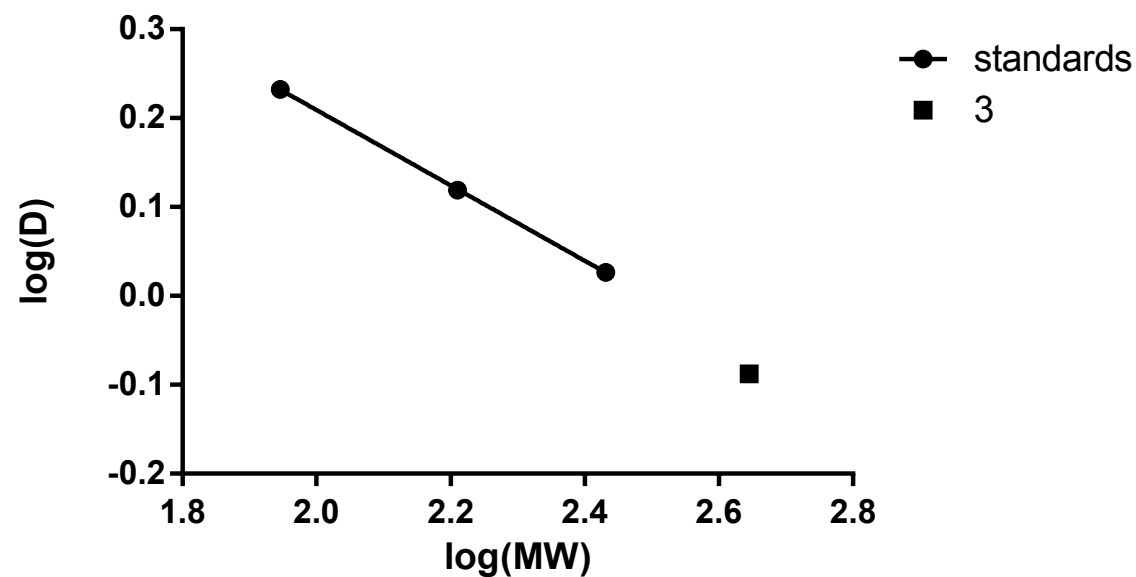


Figure S23. ^1H DOSY spectrum of complex **3** and three internal references in toluene- d_8 .



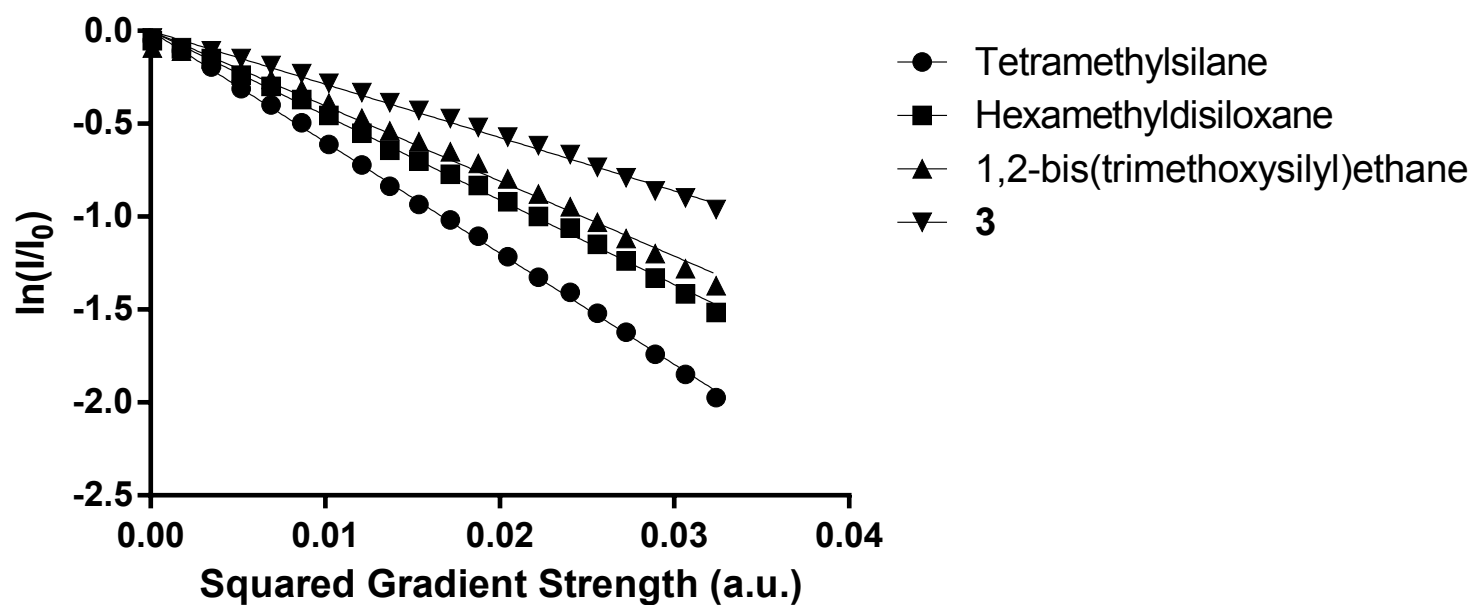
Equation	$Y = -0.4240 \cdot X + 1.057$
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R square	1.000
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MW from eq.	501.17
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ERROR	13.4%
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Figure S24. D-FW analysis of ^1H DOSY data. Internal references are shown as solid circles and complex **3** is shown as solid square.



complex	Equation	R square
Tetramethylsilane	$Y = -59.86 \cdot X + 0.0001005$	0.9990
Hexamethyldisiloxane	$Y = -45.52 \cdot X + 6.586e-005$	0.9979
1,2-bis(trimethoxysilyl)ethane	$Y = -40.42 \cdot X + 5.689e-006$	0.9921
3	$Y = -28.71 \cdot X - 3.200e-005$	0.9946

Figure S25. Stejskal-Tanner plot of experimental peaks for complex **3** and their best fit lines.

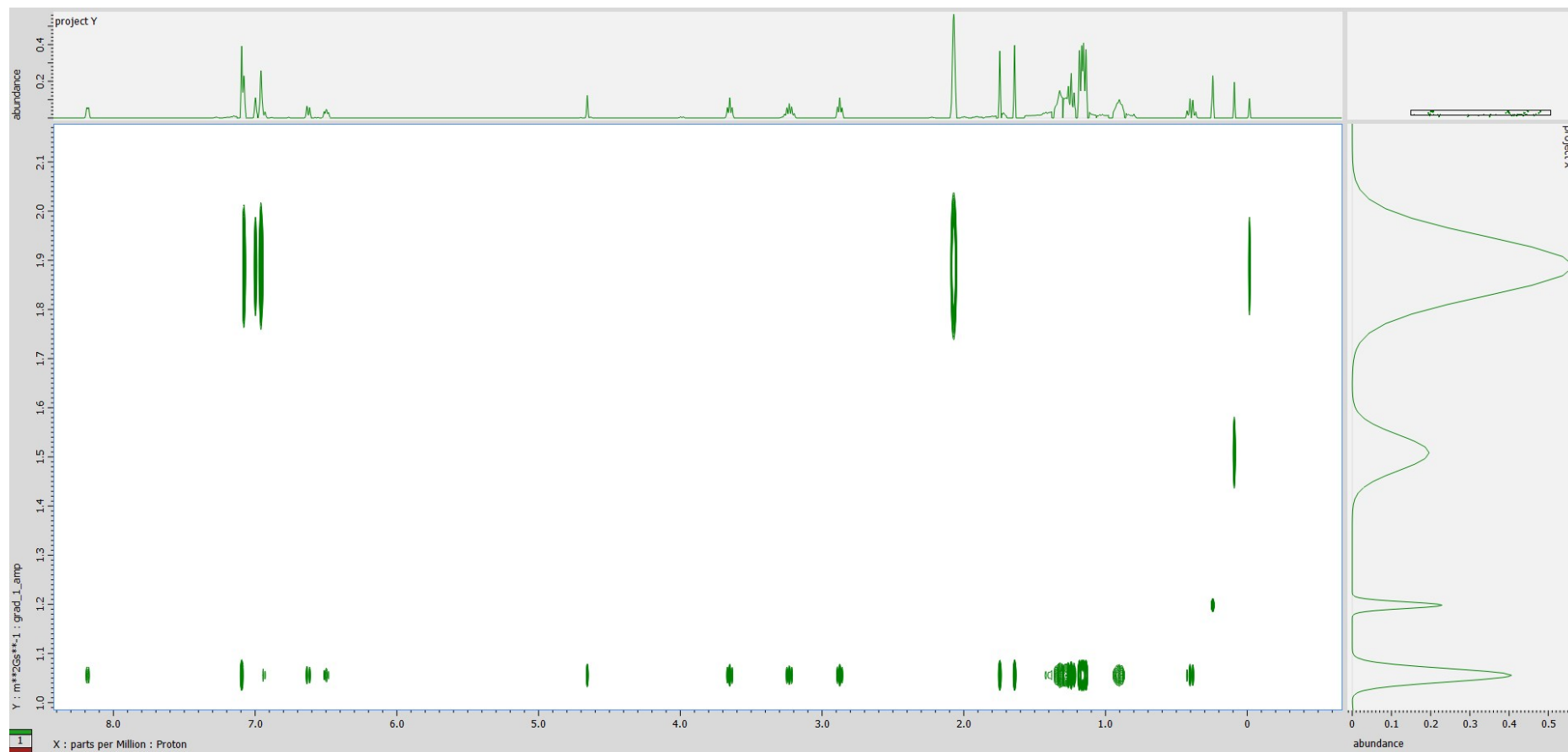
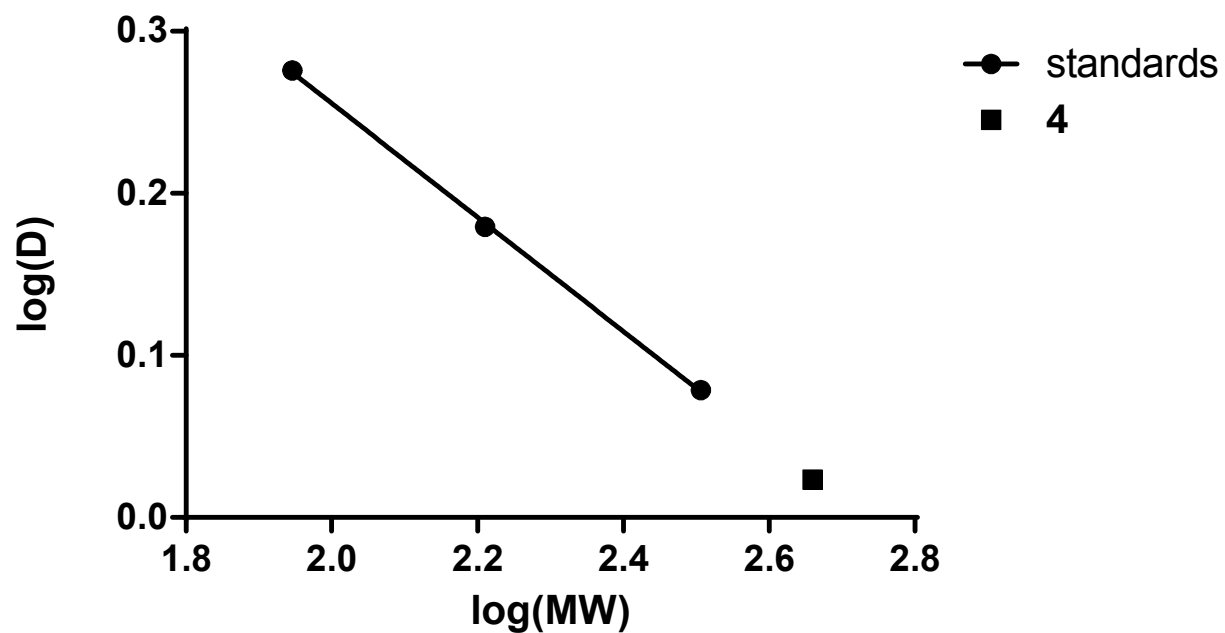


Figure S26. ^1H DOSY spectrum of complex **4** and three internal references in toluene- d_8 .



Equation

$$Y = -0.3516 \cdot X + 0.9588$$

R square

0.9996

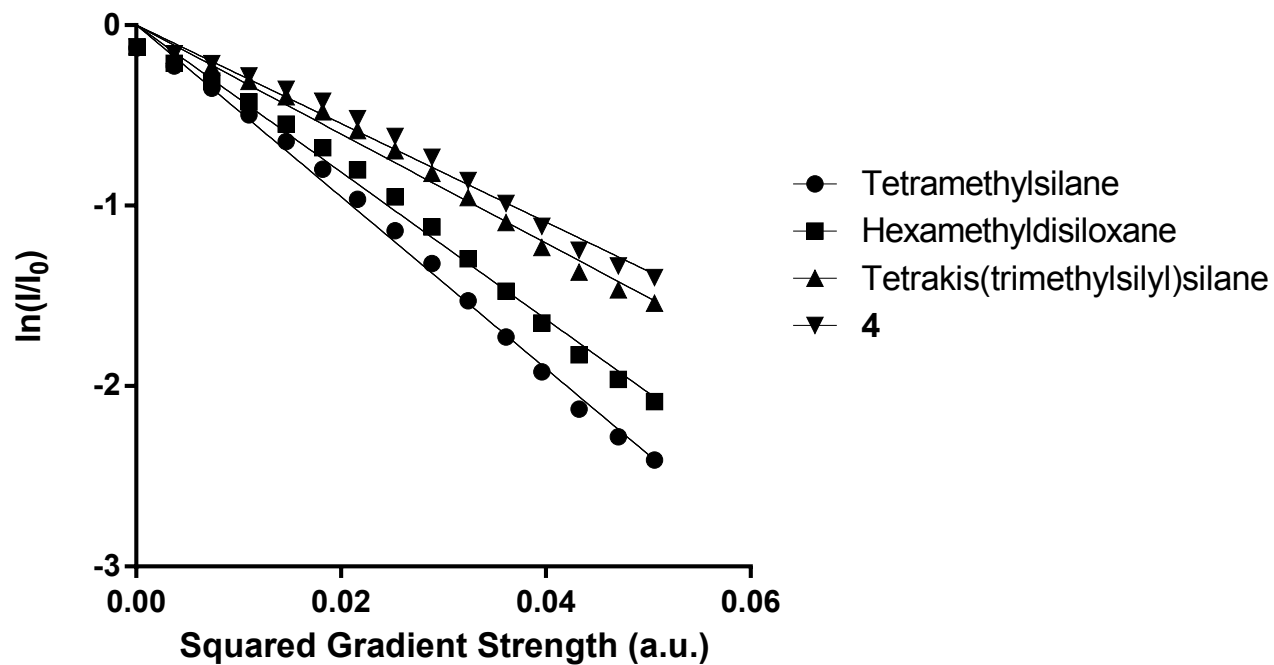
MW from eq.

457.78

ERROR

2.33%

Figure S27. D-FW analysis of ^1H DOSY data. Internal references are shown as solid circles and complex 4 is shown as solid square.



complex	Equation	R square
Tetramethylsilane	$Y = -47.62 \cdot X + 0.002039$	0.9946
Hexamethyldisiloxane	$Y = -40.77 \cdot X + 0.001773$	0.9920
1,2-bis(trimethoxysilyl)ethane	$Y = -30.20 \cdot X + 0.001304$	0.9873
4	$Y = -27.30 \cdot X + 0.001206$	0.9825

Figure S28. Stejskal-Tanner plot of experimental peaks for complex **4** and their best fit lines.

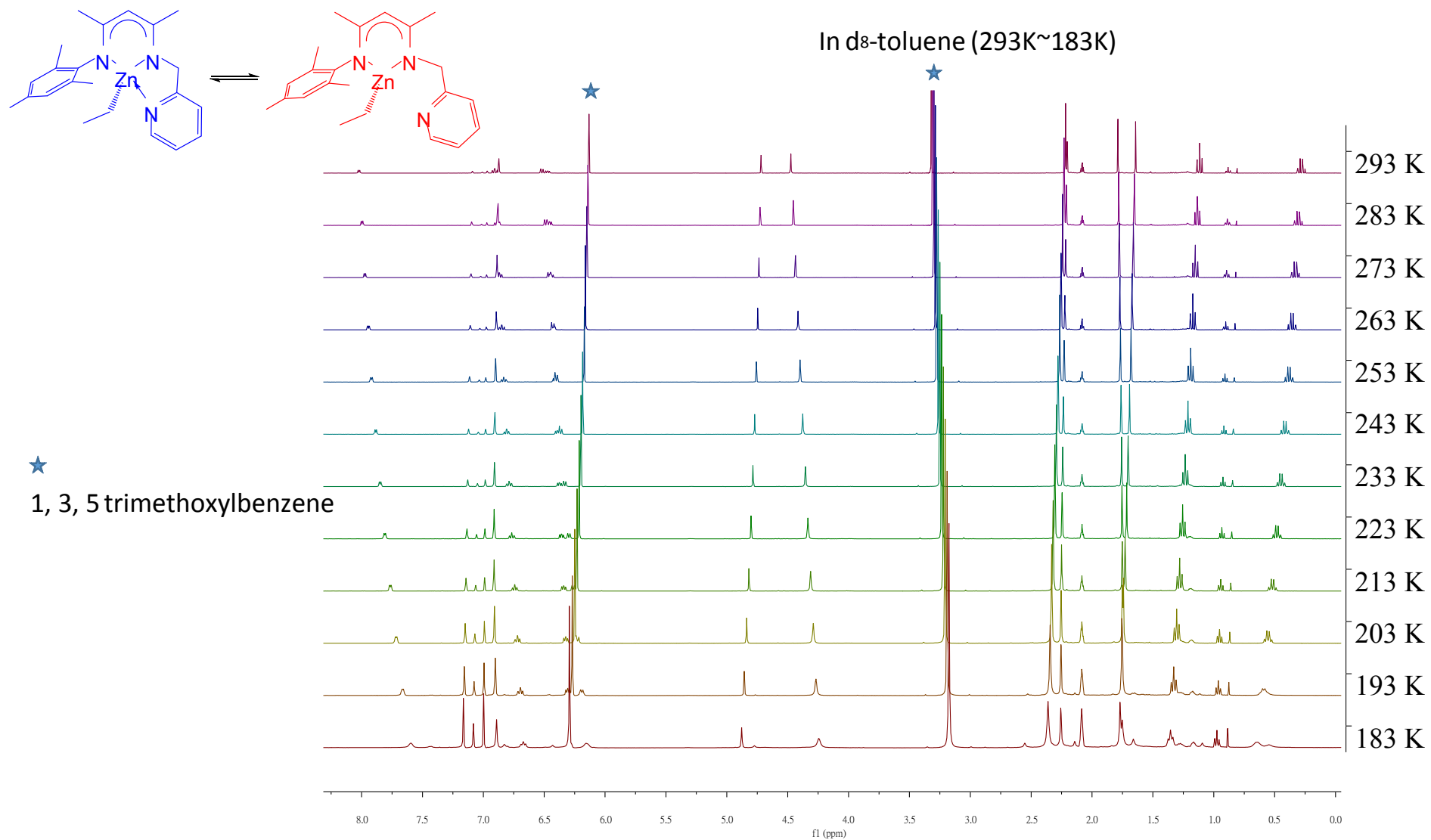


Figure S29. Variable Temperature ^1H NMR experiment of complex 1 (293K~183K)

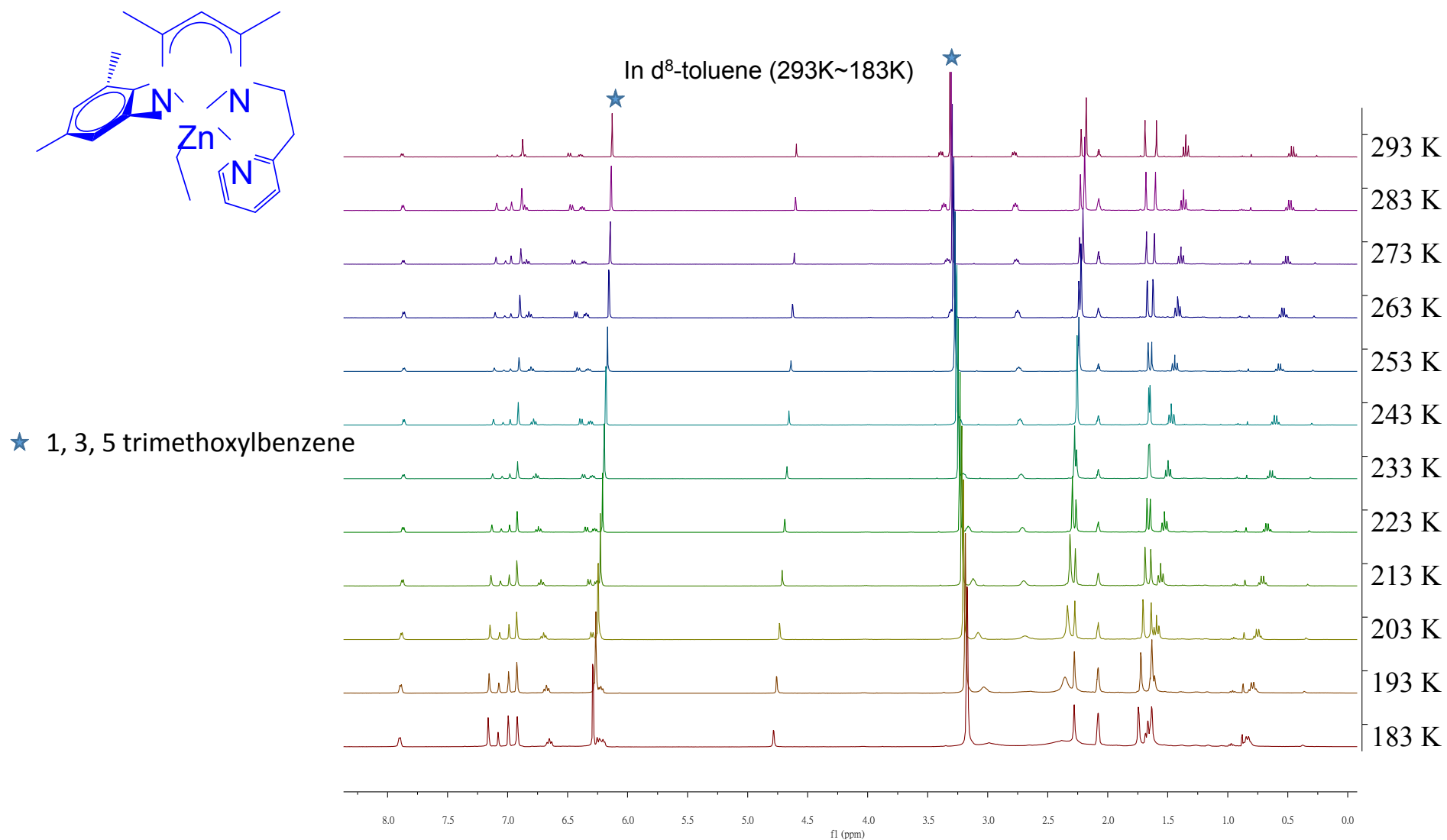


Figure S30. Variable Temperature ^1H NMR experiment of complex **2** (293K~183K)