## Supplementary Information

An enhanced oxygen electrode catalyst by incorporating CoO/SnO<sub>2</sub> nanoparticles in crumpled nitrogen-doped graphene in alkaline media

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**Fig. S1** (a) XRD patterns of SnO<sub>2</sub>/NG-500 and CoO/SnO<sub>2</sub>/NG-500; (b-e) The nitrogen adsorptiondesorption isotherm and the pore distribution curves inset for CoO/NG-500 (b), SnO<sub>2</sub>/NG-500 (c), NG-500 (d) and r-GO-500 (e); (f) The average pore width values of CoO/SnO<sub>2</sub>/NG-500, SnO<sub>2</sub>/NG-500, CoO/NG-500, NG-500 and r-GO-500.

Sample	BET Surface Area (m <sup>2</sup> g <sup>-1</sup> )	Average Pore Size (nm)
CoO/SnO <sub>2</sub> /NG-500	282.2	45.1
CoO/NG-500	173.5	49.4
SnO <sub>2</sub> /NG-500	195.5	40.5
NG-500	359.7	64.0
r-GO-500	509.4	34.5

 Table S1 BET surface area and average pore size for the samples obtained from nitrogen

 adsorption-desorption isotherm.

Table S2 XPS datas of CoO/SnO<sub>2</sub>/NG-500, CoO/SnO<sub>2</sub>/NG, CoO/NG-500, SnO<sub>2</sub>/NG-500, NG-

500 and r-GO-500.
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Sample	C 1s	O 1s	N 1s	Co 2p	Sn 3d	$\Delta \mathrm{B} E_{\mathrm{O-Sn}}$
	(eV)	(eV)	(eV)	(eV)	(eV)	
CoO/SnO <sub>2</sub> /NG-500	284.8	530.2	400.1	781.1	487.2	43.0
CoO/SnO <sub>2</sub> /NG	284.50	531.8	399.2	781.4	486.6	45.2
CoO/NG-500	284.65	530.1	400.3	780.3	-	-
SnO <sub>2</sub> /NG-500	284.7	531.2	399.7	-	486.9	44.3
NG-500	284.8	531.9	398.5	-	-	-
r-GO-500	284.7	532.0	-	-	-	-



Fig. S2 (a,b,c) SEM images of r-GO-500 (a), CoO/NG-500 (b), and SnO<sub>2</sub>/NG-500 (c); (d-i) EDX

mapping of CoO/SnO<sub>2</sub>/NG-500.



Fig. S3 High resolution XPS spectra of (a) Sn 3d in SnO<sub>2</sub>/NG-500 and (b) Co 2p in CoO/NG-500.



**Fig. S4** (a, b) Cycle voltammograms (a) and linear sweep voltammogram measurements (b) of CoO/SnO<sub>2</sub>/NG at different temperature; (c, d) LSV polarization curves (c) and K-L plots (d) at different potentials for CoO/SnO<sub>2</sub>/NG-500 in an O<sub>2</sub>-saturated 0.1 M KOH at 1600 rpm; (e) The n values at various potentials of CoO/SnO<sub>2</sub>/NG-500.

Catalyst	Tafel Slope	Reference
	(mV dec <sup>-1</sup> )	
CoO/SnO <sub>2</sub> /NG-500	33	This work
N-CG-CoO	48	Energy Environ. Sci. 2014, 7, 609–616.
Co(OH) <sub>2</sub>	62	ACS Appl. Mater. Interfaces 2015, 7, 12930-12936.
Co-N-CNT	40	ACS Appl. Mater. Interfaces 2015, 7, 4048-4055.
Co <sub>0.5</sub> Fe <sub>0.5</sub> S@N-MC	67	ACS Appl. Mater. Interfaces 2015, 7, 1207–1218.
Co(OH) <sub>x</sub> -NCNT	54	ACS Appl. Mater. Interfaces 2016, 8, 1571–1577.
Co(OH) <sub>x</sub> -CNT	90	ACS Appl. Mater. Interfaces 2016, 8, 1571–1577.
CoS <sub>2</sub> (400)/N,S-GO	75	ACS Catal. 2015, 5, 3625–3637.

 Table S3 Recent literature on the ORR performance.

Catalyst	$\mathfrak{y}_{J=10} \ [mV$	Tafel Slope	Reference
	vs. SCE]	[mV dec <sup>-1</sup> ]	
CoO/SnO <sub>2</sub> /NG-500	285	41	This work
N-CG-CoO	334	71	Energy Environ. Sci., 2014, 7, 609–616.
CoFe <sub>2</sub> O <sub>4</sub> /PANI-	314	31	J. Mater. Chem. A, 2016.
MWCNT			
CoP/rGO	340	66	Chem. Sci., 2016, 7, 1690–1695.
CoSe	-	40	Electrochimica Acta, 2016, <b>194</b> , 59–66.
Co <sub>3</sub> O <sub>4</sub> /N-CNT–GNR	360	44	J. Mater. Chem. A, 2015, <b>3</b> , 13371–
			13376.
CNCNT	310	50	ACS Appl. Mater. Interfaces, 2015, 7,
			4048-4055.
NCNT/Co <sub>x</sub> Mn <sub>1-x</sub> O	340	40	Nano Energy, 2016, <b>20</b> , 315–325.
Co(OH) <sub>x</sub> -NCNT	380	36	ACS Appl. Mater. Interfaces, 2016, 8,
			1571–1577.

**Table S4** Recent literature on the OER performance.