

## Supplementary Material for

# Products, Mechanism, and Kinetics of OH Radical-initiated Oxidation Degradation of 2,4,4'-trichlorobiphenyl in the Atmosphere

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### Keywords

PCB28, OH radicals, Reaction mechanism, Kinetic parameters

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Eighteen pages

Contains twelve Figures

**Fig. S1(a).** Optimized structures of pre-reactive complexes and product complexes at the MPWB1K/6-31+G(d,p) level of theory. Distances are in angstrom.

**Fig. S1(b).** Initial reaction schemes of OH radicals with PCB28.

**Fig. S1(c).** TST, CVT and CVT/SCT rate constants for  $\text{PCB28} + \text{OH} \rightarrow \text{IM16} + \text{H}_2\text{O}$ .

**Fig. S1(d).** The frontier molecular orbital analyses of IM5b-IM5g.

**Fig. S2(a).** Secondary reaction scheme of IM12 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

**Fig. S2(b).** Secondary reaction scheme of IM12b-IM12e with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

**Fig. S2(c).** Secondary reaction scheme of IM12b-1 and IM12c-1 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

**Fig. S3.** OH radical-initiated reaction scheme of PCB1 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

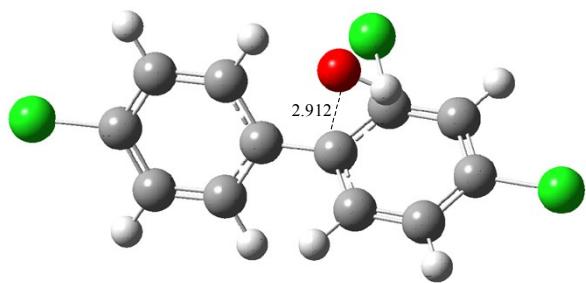
**Fig. S4.** OH radical-initiated reaction scheme of PCB4 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

**Fig. S5.** OH radical-initiated reaction scheme of PCB44 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

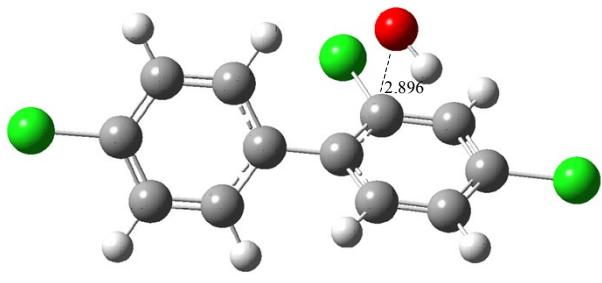
**Fig. S6.** OH radical-initiated reaction scheme of PCB110 with the potential barriers  $\Delta E$  (kcal/mol) and reaction heats  $\Delta H$  (kcal/mol).

**Fig. S7.** OH radical-initiated reaction scheme of PCB153 with the potential barriers

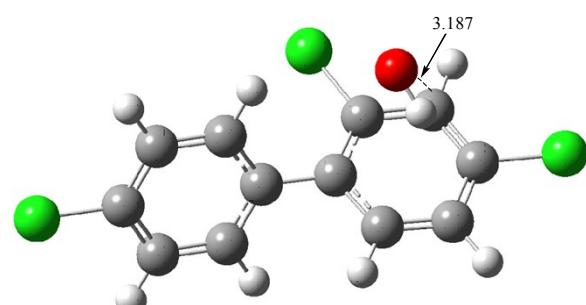
$\triangle E$  (kcal/mol) and reaction heats  $\triangle H$  (kcal/mol).



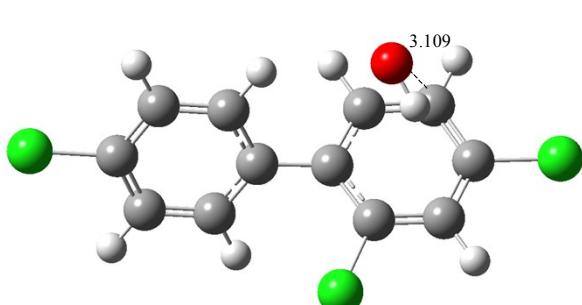
PTS1



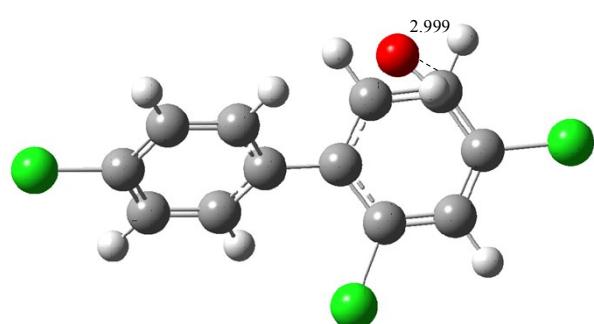
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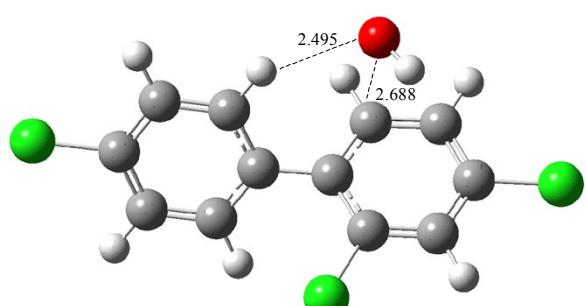
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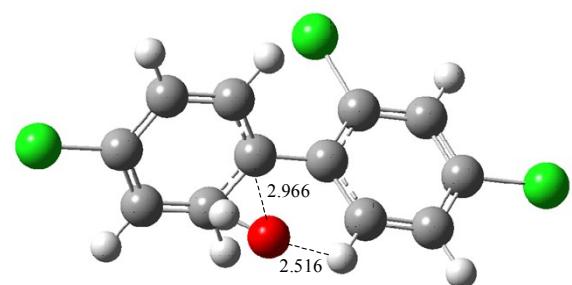
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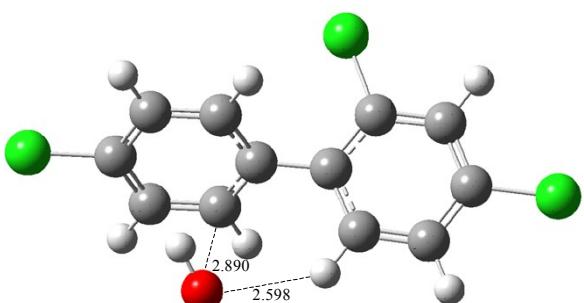
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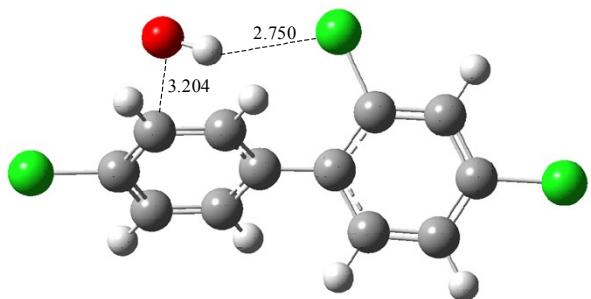
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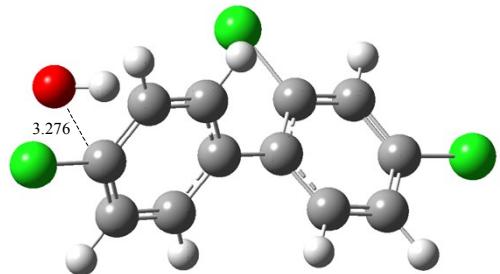
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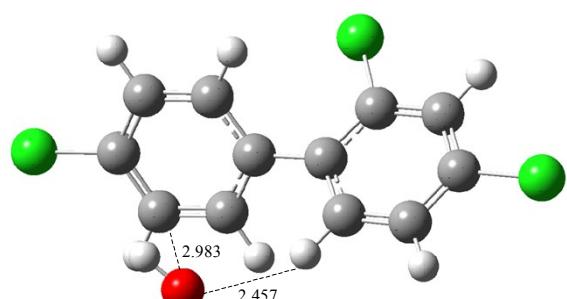
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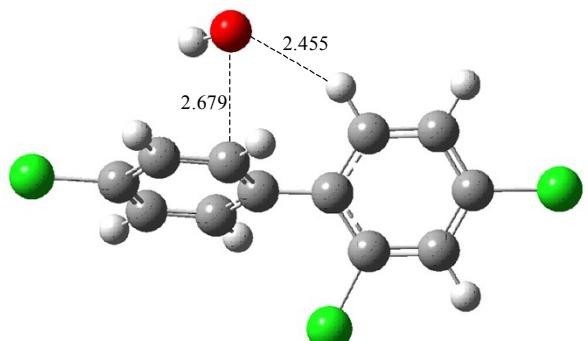
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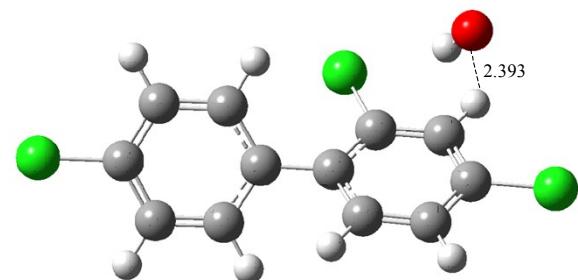
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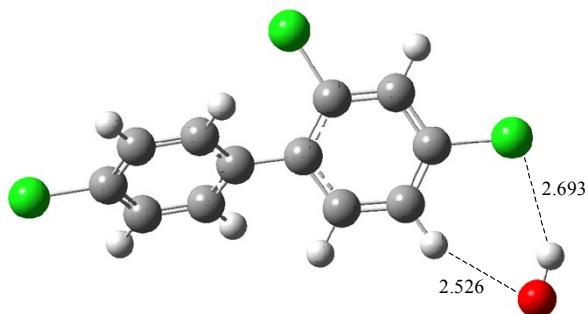
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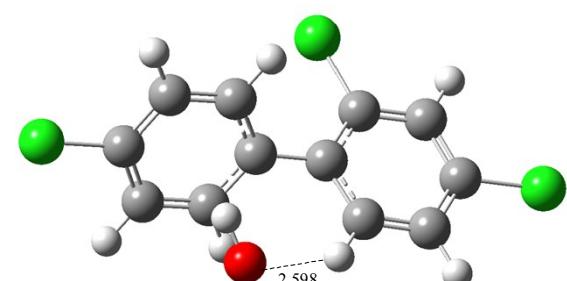
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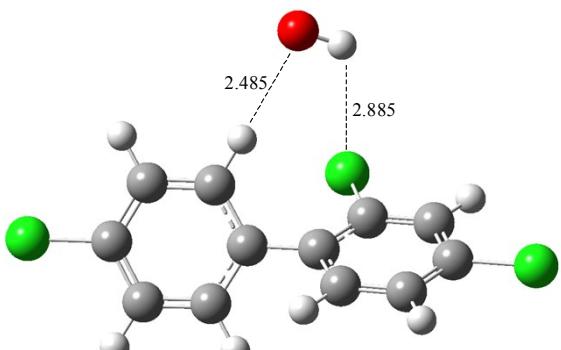
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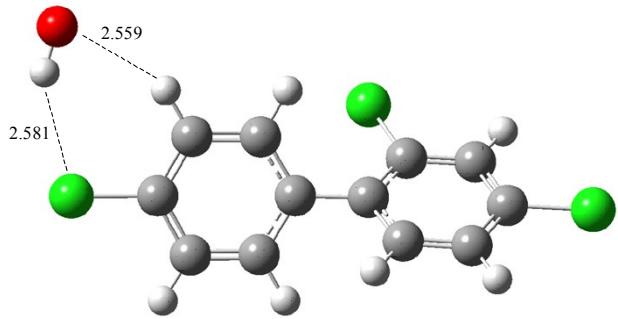
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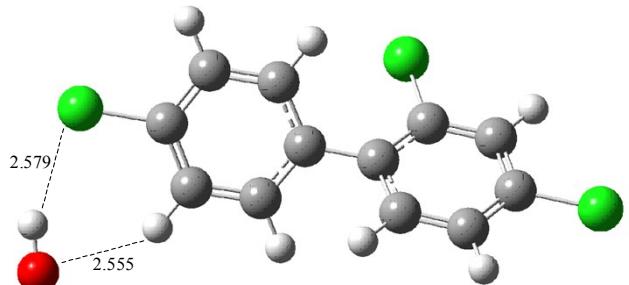
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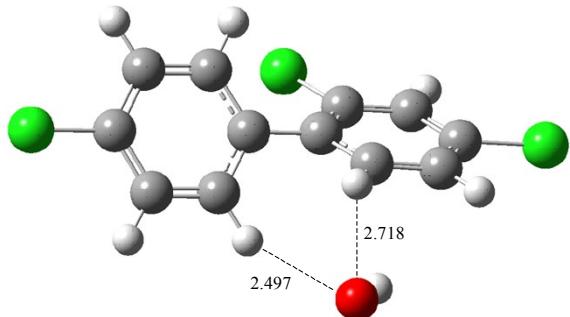
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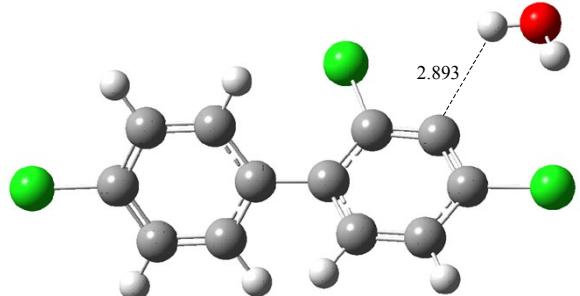
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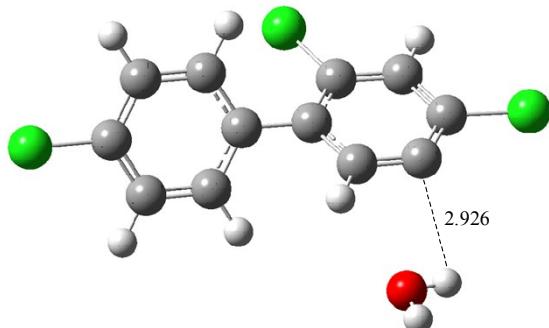
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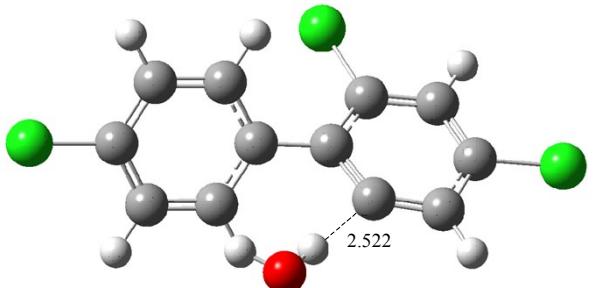
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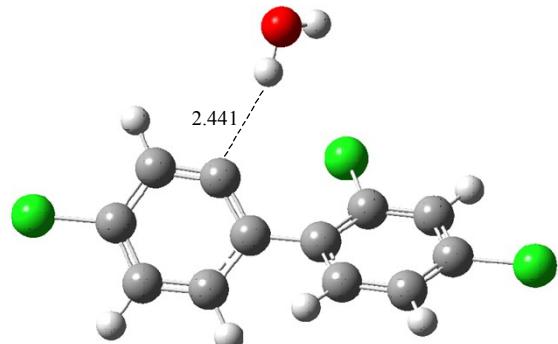
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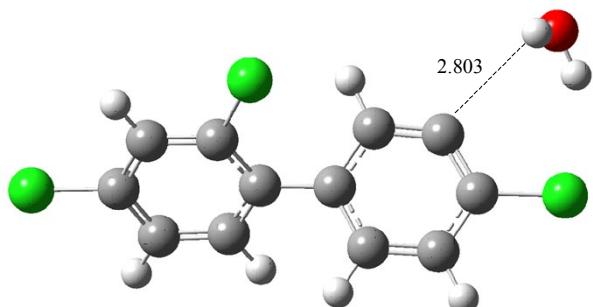
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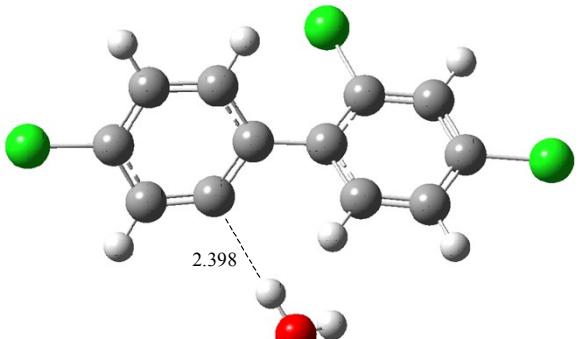
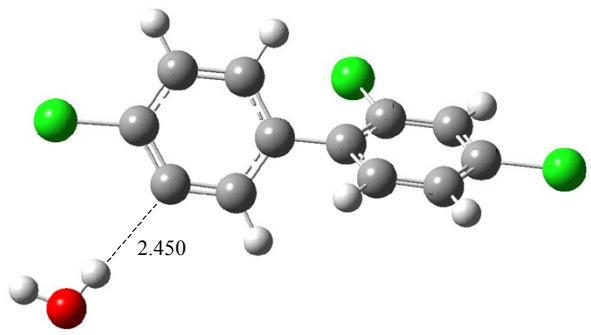
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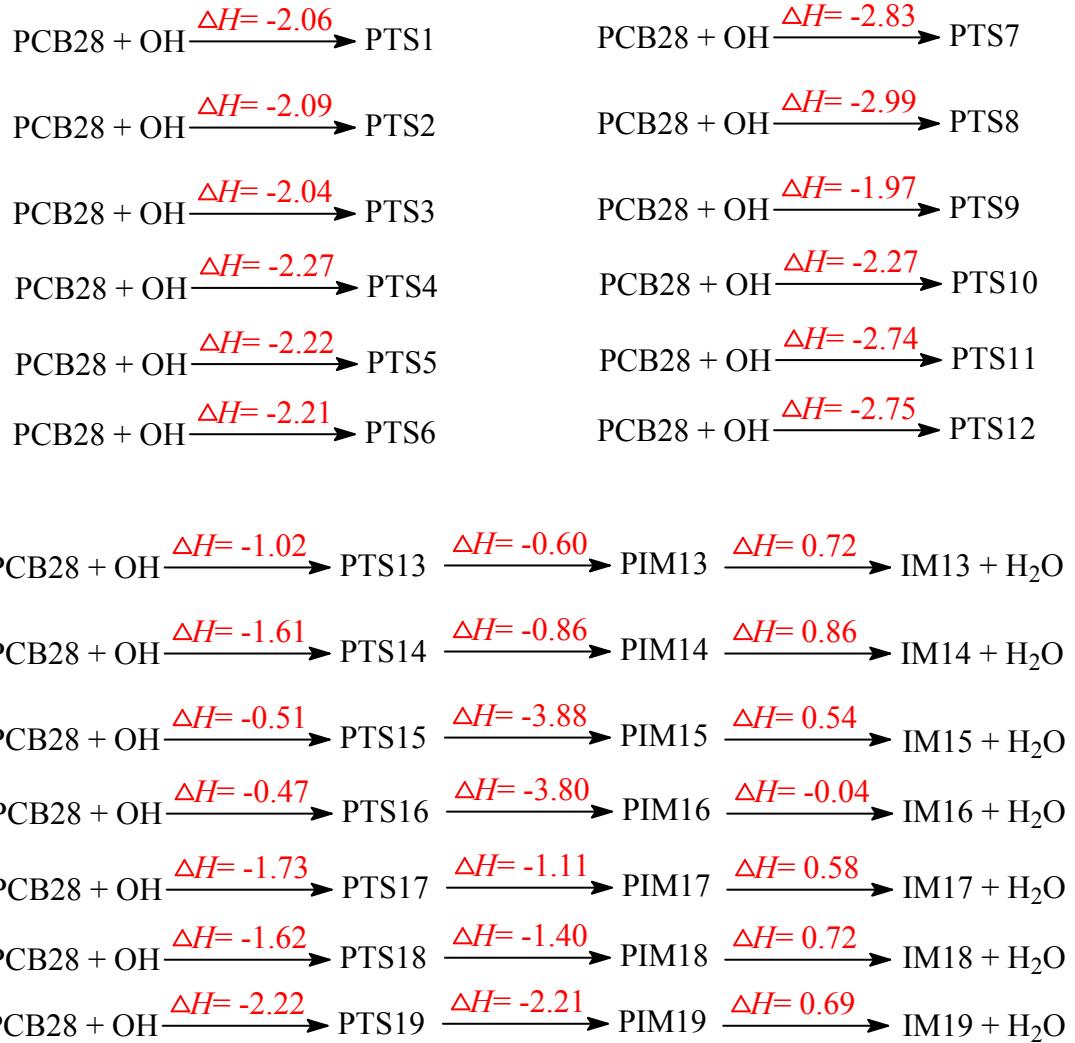
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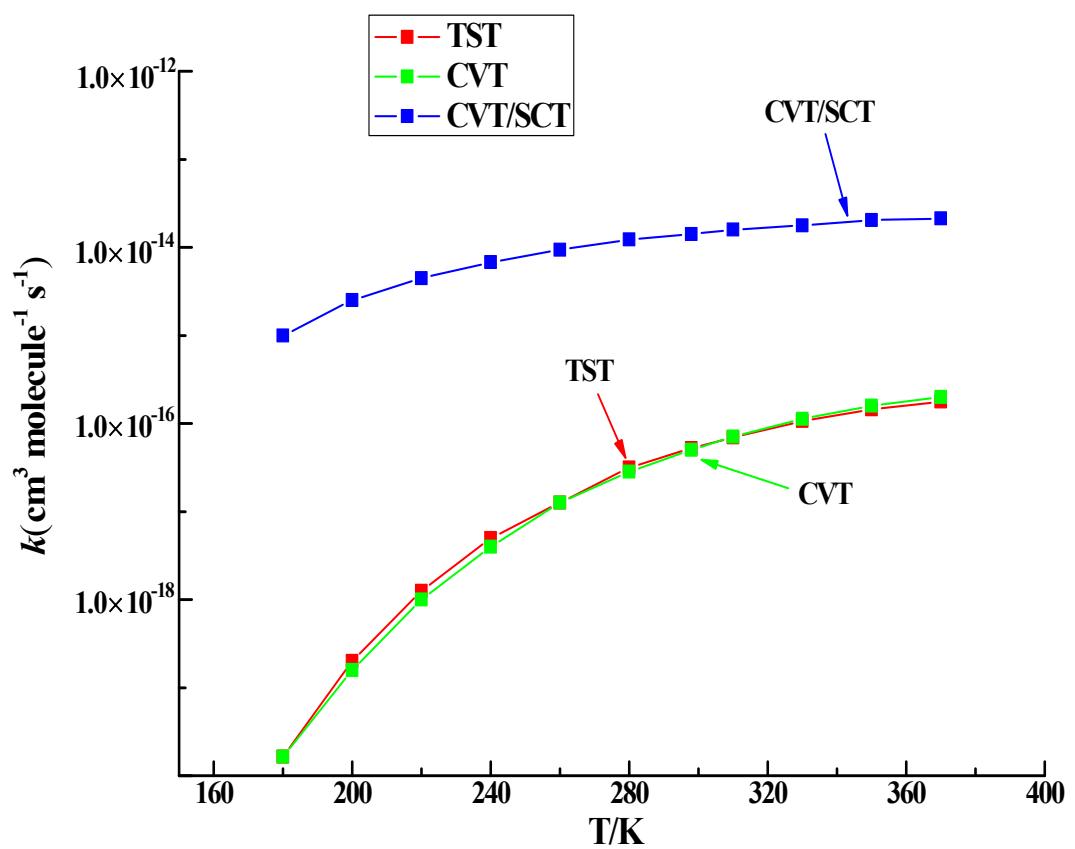
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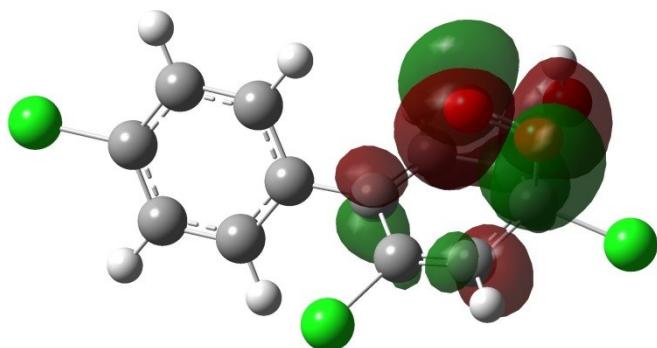
**Fig. S1(a)**



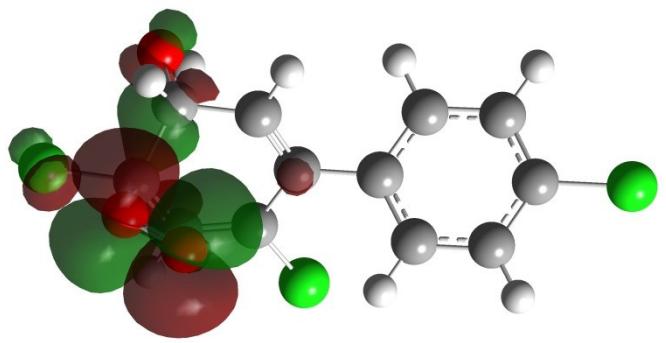
**Fig. S1(b)**



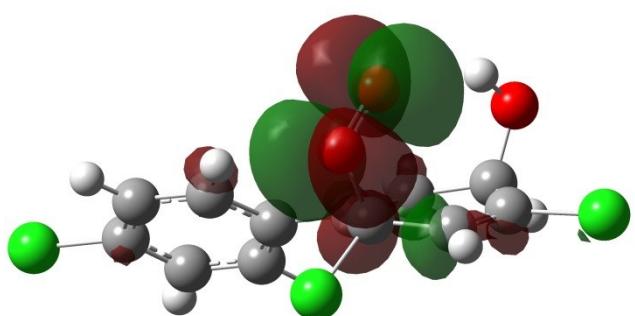
**Fig. S1(c)**



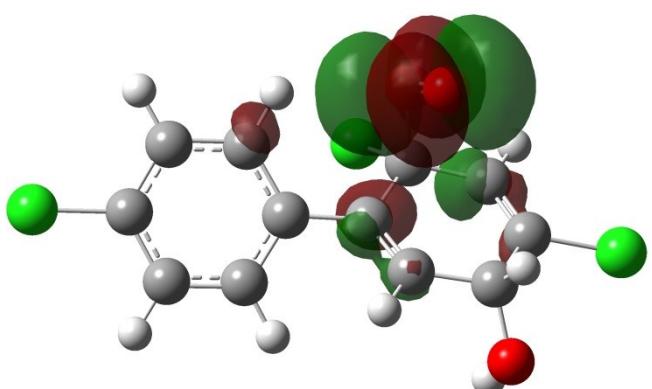
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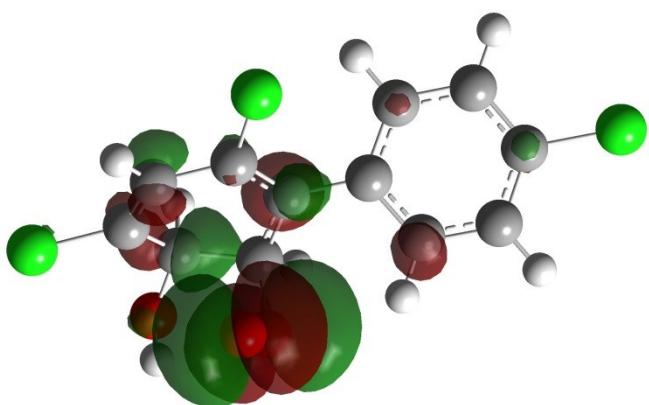
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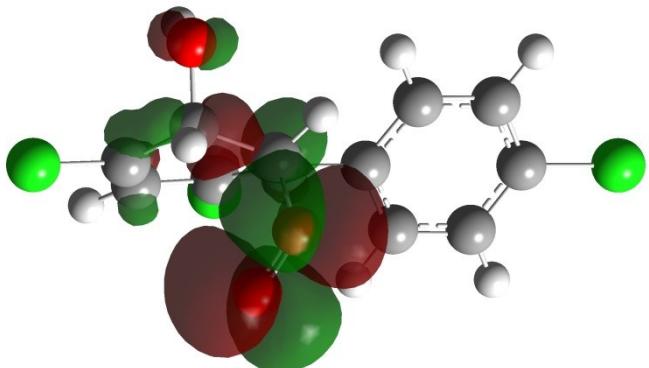
IM5d



IM5e

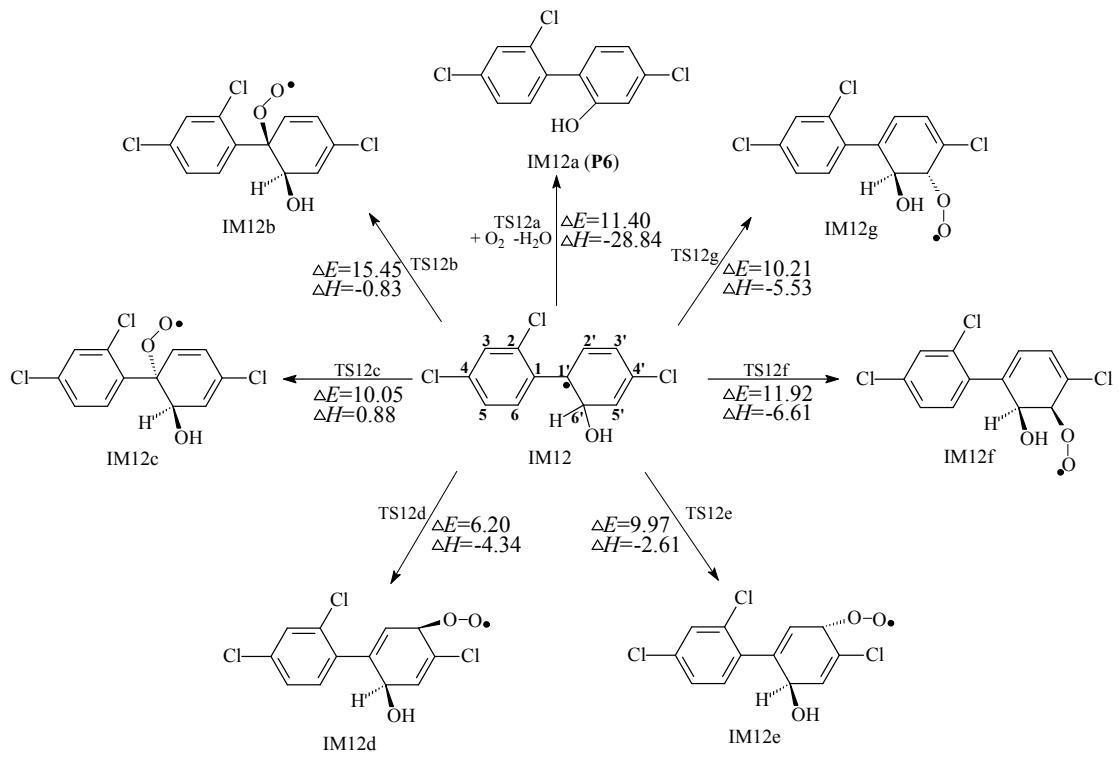


IM5f

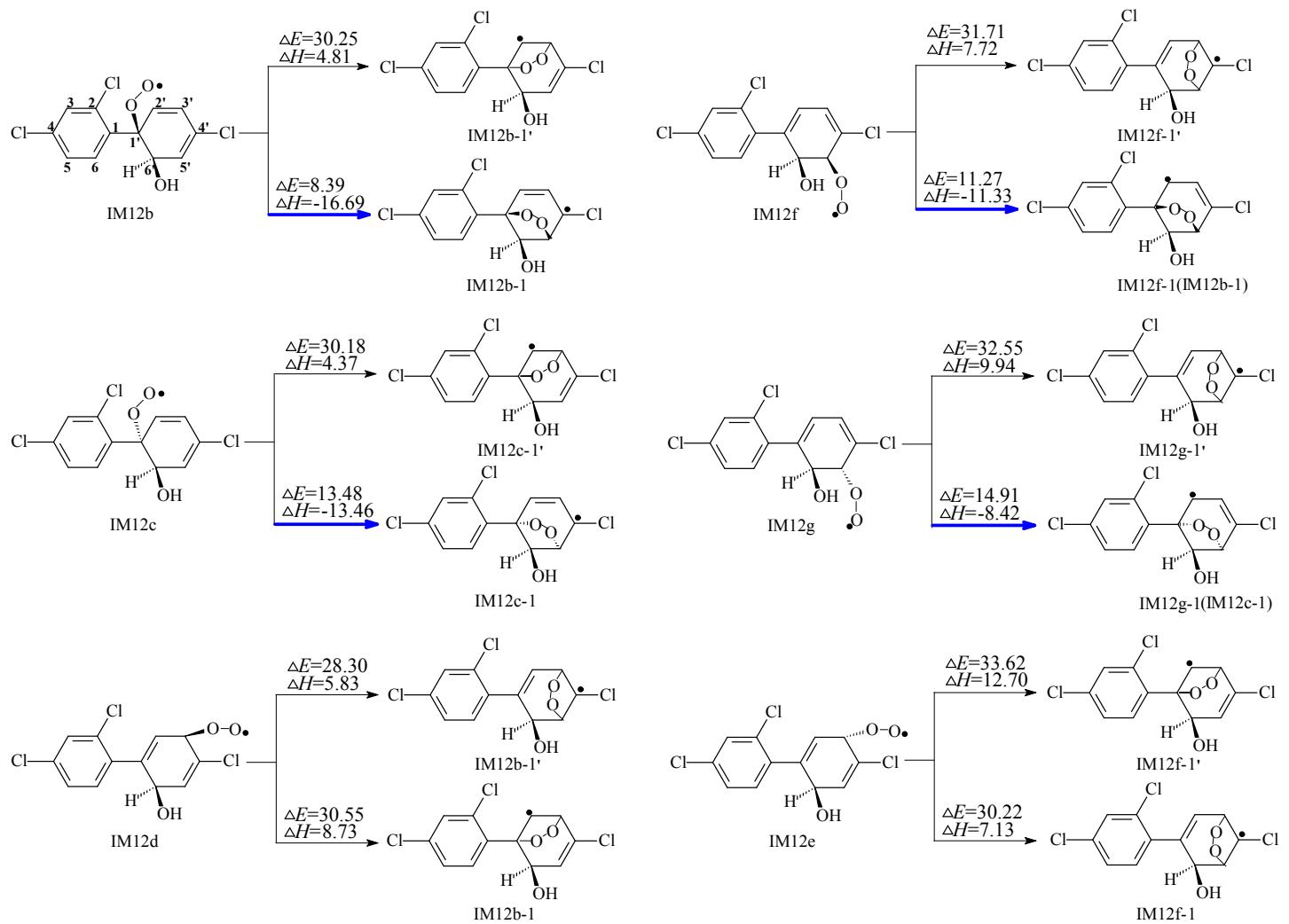


IM5g

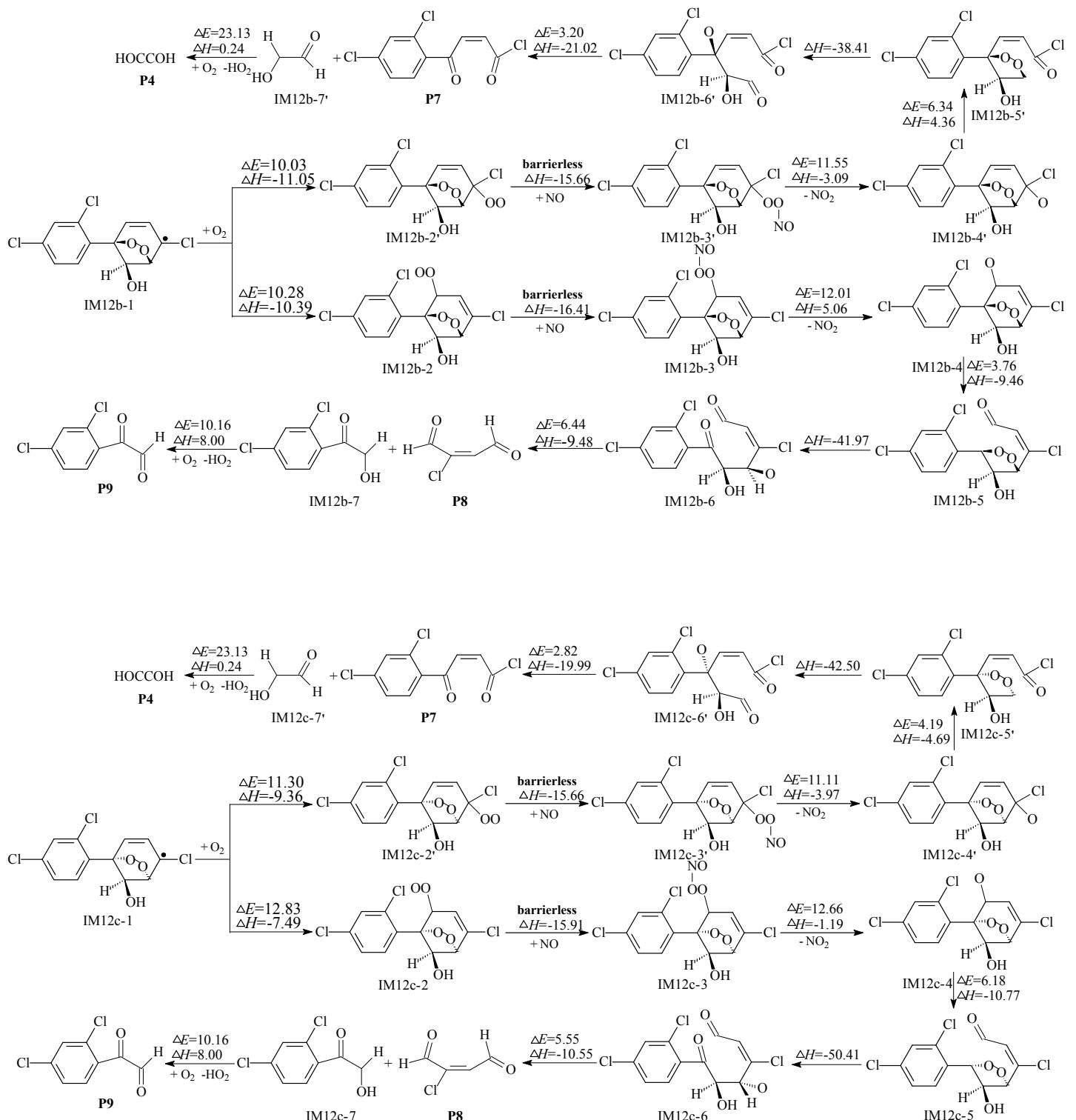
**Fig. S1(d)**



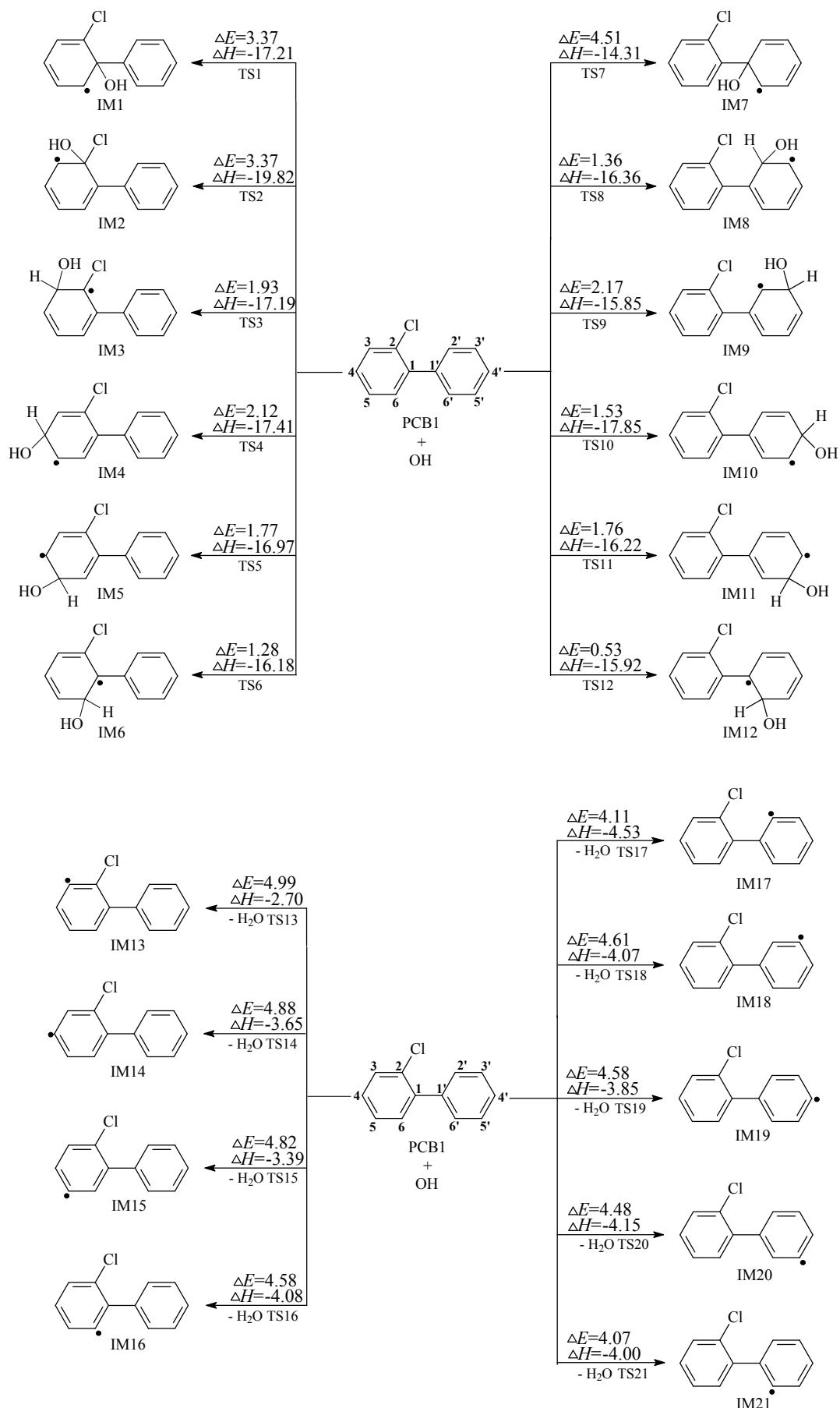
**Fig. S2(a)**



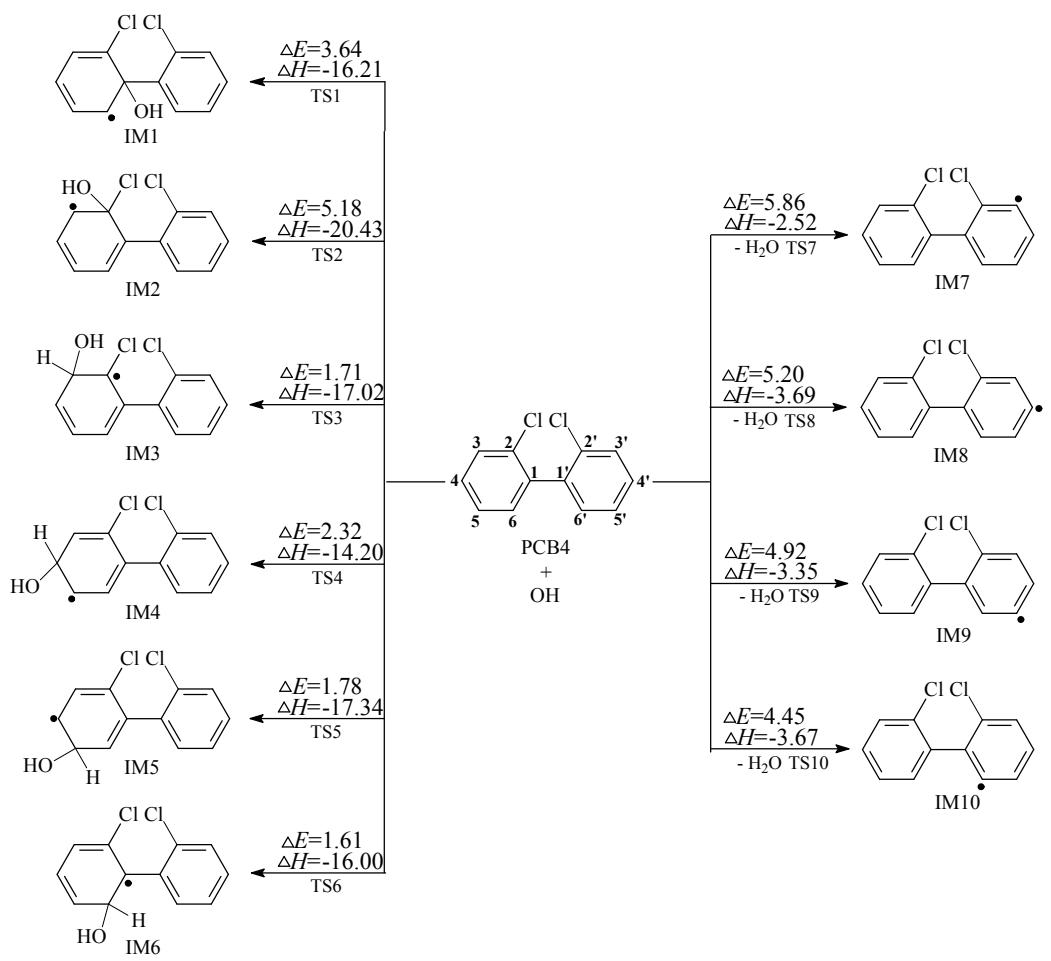
**Fig. S2(b)**



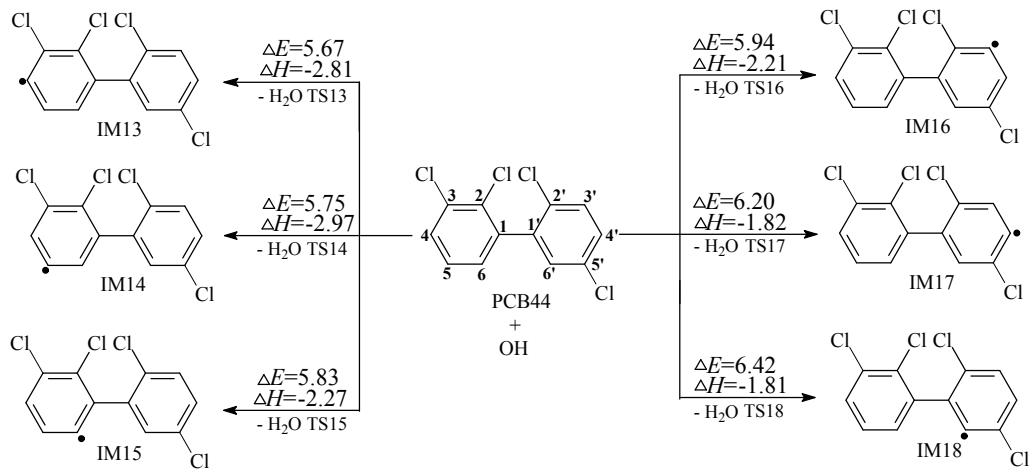
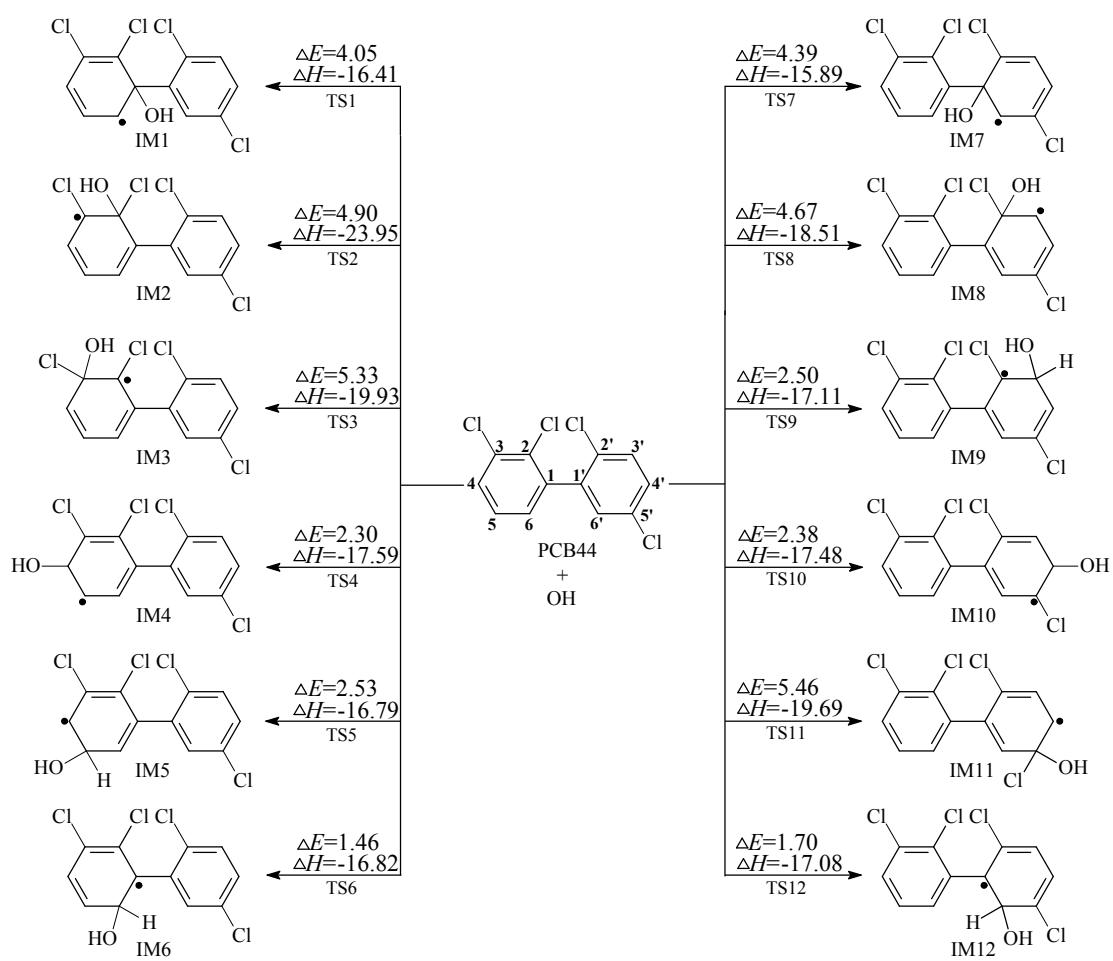
**Fig. S2(c)**



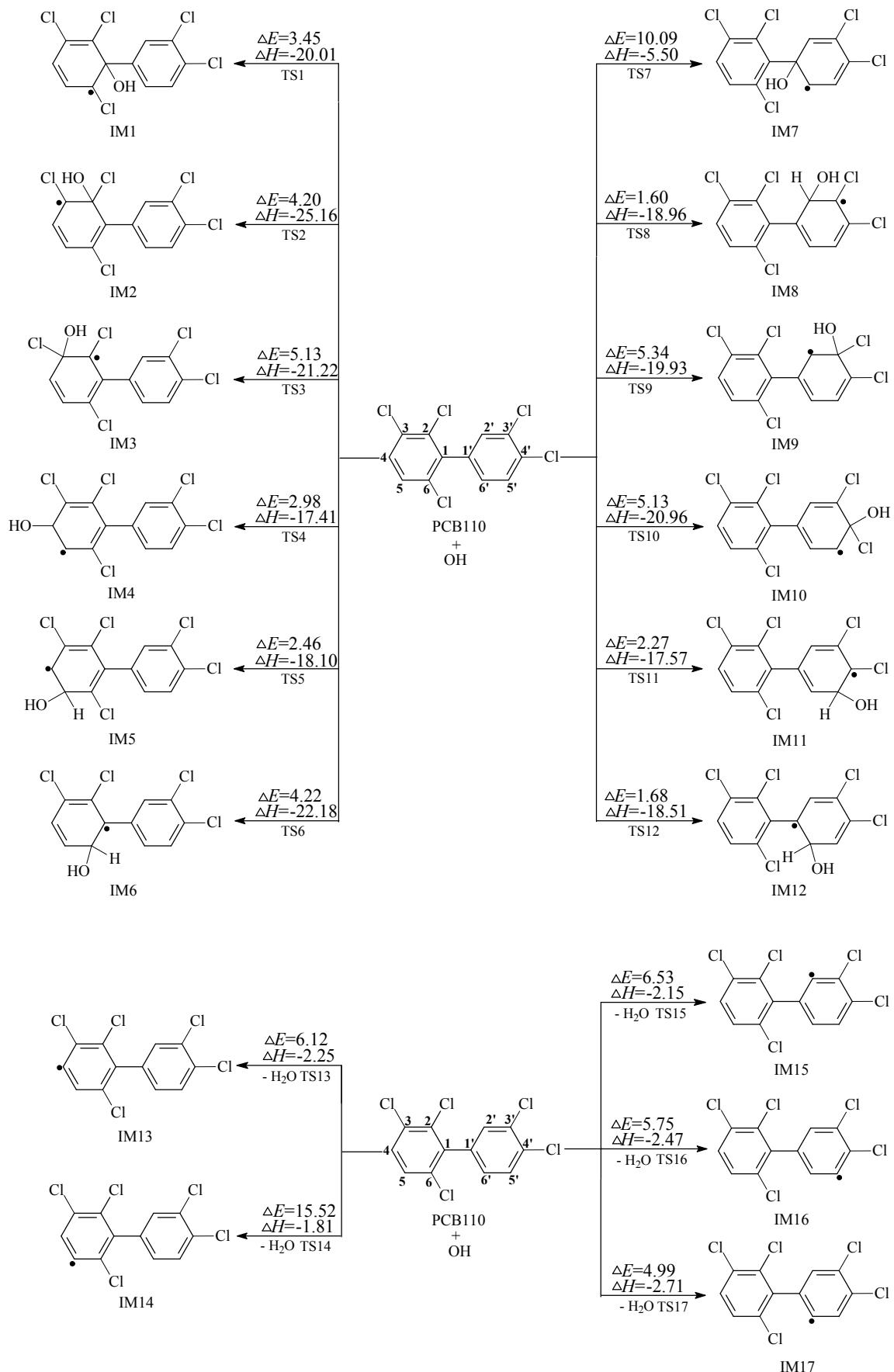
**Fig. S3**



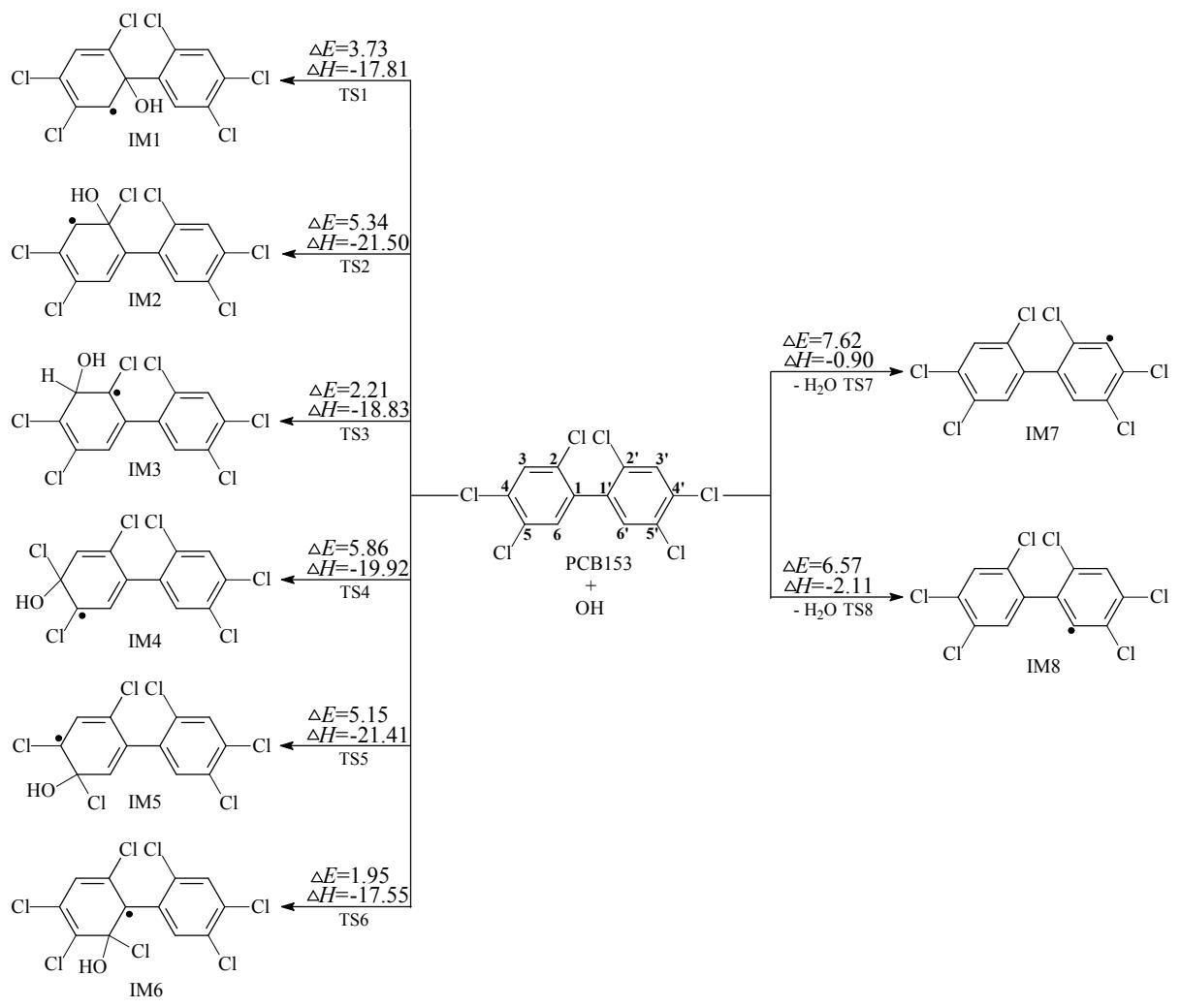
**Fig. S4**



**Fig. S5**



**Fig. S6**



**Fig. S7**