

Electronic supplementary information for:

## Accessing Centnerszwer's Quasiracemate – Molecular Shape Controlled Molecular Recognition

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### Electronic Supplementary Information

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## **Supporting Information**

### **S1. Crystal growth attempts of Centnerszwer's Quasiracemate (+)-1/(-)-2**

Crystal growth attempts of Centnerszwer's Quasiracemate (+)-**1**/(-)-**2** were conducted using the slow evaporation method at room temperature and -5°C using a variety of single solvent (acetone, 2-butanone, dichloromethane, ethanol, methanol, and THF) and mixed solvent systems (1:1 dichloromethane:hexanes). Crystal growth experiments also explored the use of the vapor diffusion method (dichloromethane/pentane). In each case, materials retrieved from these studies consisted of either the starting homochiral Cl- or Br-succinic acids or amorphous conglomerate solids as determined by melting point or X-ray diffraction data.

This study also investigated the crystal growth behavior of components (+)-**1** and (-)-**2** from the melt. Samples of (+)-**1** and (-)-**2** were deposited under a cover slip on a glass slide to give an interface of contact between the two components. Heating the sample using an Instec HCS302 hot stage, a single eutectic point could be observed indicating the lack of a quasiracemic phase of (+)-**1**/(-)-**2**. These results were complicated by the possible decomposition of component (-)-**2**.

## S2. X-ray Crystallography

**Table S2. Crystallographic data for 2-chloro and 2-bromosuccinic acid structures.**

	(-)-1	(+)-2	(+)-1·Im	(-)-2·Im
Crystal data				
CCDC deposit no.	1470675	1470676	1470679	1470678
Empirical formula	C <sub>4</sub> H <sub>5</sub> ClO <sub>4</sub>	C <sub>4</sub> H <sub>5</sub> BrO <sub>4</sub>	C <sub>17</sub> H <sub>9</sub> ClN <sub>2</sub> O <sub>4</sub>	C <sub>17</sub> H <sub>9</sub> BrN <sub>2</sub> O <sub>4</sub>
Crystal System, space group	Orthorhombic <i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	Orthorhombic <i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	triclinic <i>P</i> 1	Monoclinic <i>P</i> 2 <sub>1</sub>
<i>M</i> <sub>r</sub>	152.5	196.99	220.61	265.07.41
<i>a</i> , Å	7.5593(3)	7.5647(5)	5.0265(3)	7.2825(2)
<i>b</i> , Å	8.5020(3)	8.6337(7)	6.1840(3)	8.8079(3)
<i>c</i> , Å	9.0604(4)	9.2380(7)	8.1859(4)	8.0544(2)
$\alpha$ , deg	90	90	93.095(3)	90
$\beta$ , deg	90	90	102.264(3)	108.936(2)
$\gamma$ , deg	90	90	106.883(3)	90
<i>V</i> , (Å <sup>3</sup> )	582.30(4)	603.35(8)	236.10(2)	488.68(3)
<i>Z</i> , <i>Z'</i>	4, 1	4, 1	1, 1	2, 1
<i>D</i> <sub>calc</sub> (g cm <sup>-3</sup> )	1.740	2.169	1.552	1.801
$\mu$ (mm <sup>-1</sup> ), rad. type	5.371, Cu <i>K</i> α	8.870, Cu <i>K</i> α	3.574, Cu <i>K</i> α	5.729, Cu <i>K</i> α
F <sub>000</sub>	312	384	114	264
temp (K)	100(2)	100(2)	100(2)	100(2)
Crystal form, color	plate, colorless	plate, colorless	plate, colorless	plate, colorless
Crystal size, mm	0.10 x 0.24 x 0.26	0.03 x 0.08 x 0.09	0.31 x 0.29 x 0.08	0.03 x 0.10 x 0.13
Data collection				
Diffractometer	Bruker Apex II	Bruker Apex II	Bruker Apex II	Bruker Apex II
<i>T</i> <sub>min</sub> / <i>T</i> <sub>max</sub>	0.487/0.753	0.475/0.753	0.609/0.753	0.609/0.753
No. of refls. (meas., uniq., and obs.)	8317/1068/1060	4637/1060/1010	5075/1486/1465	10932/1761/1740
<i>R</i> <sub>int</sub>	0.0380	0.0401	0.0276	0.0305
$\theta$ <sub>max</sub> (°)	68.22	68.24	68.04	68.17
Refinement				
<i>R</i> / <i>R</i> <sup>2</sup> <sub>w</sub> (obs data)	0.0400/0.1028	0.0264/0.0562	0.0234/0.0575	0.0184/0.0473
<i>R</i> / <i>R</i> <sup>2</sup> <sub>w</sub> (all data)	0.0401/0.1029	0.0284/0.0568	0.0238/0.0577	0.0187/0.0473
<i>S</i>	1.12	1.06	0.97	0.97
No. of refls.	5223	1060	1486	1740
No. of parameters	88	88	134	136
$\Delta\rho_{\text{max/min}}$ (e·Å <sup>-3</sup> )	0.613/-0.242	0.607/-0.350	0.233/-0.158	0.303/-0.210
Flack-x	0.08(4)	0.02(2)	0.052(19)	0.019(9)

**Table S2. Crystallographic data for 2-chloro and 2-bromosuccinic acid structures. (continued)**

	(±)-1·Im <sup>1</sup>	(±)-2·Im	(+)-1/(-)-2·Im	(+)-1/(-)-2·BPDO
Crystal data				
CCDC deposit no.	1211267	1470677	1470680	1470681
Empirical formula	C <sub>7</sub> H <sub>9</sub> ClN <sub>2</sub> O <sub>4</sub>	C <sub>7</sub> H <sub>9</sub> BrN <sub>2</sub> O <sub>4</sub>	C <sub>14</sub> H <sub>18</sub> BrClN <sub>4</sub> O <sub>8</sub>	C <sub>29</sub> H <sub>36</sub> N <sub>2</sub> O <sub>12</sub> S <sub>3</sub>
Crystal System, space group	triclinic <i>P</i> 	triclinic <i>P</i> 	triclinic <i>P</i> 1	triclinic <i>P</i> 1
<i>M</i> <sub>r</sub>	220.61	265.07	485.68	725.89
<i>a</i> , Å	8.084(3)	8.0994(4)	8.1185(8)	7.7692(3)
<i>b</i> , Å	8.253(5)	8.3605(4)	8.2996(8)	10.7311(4)
<i>c</i> , Å	8.990(5)	8.8967(7)	8.8998(9)	14.5005(13)
$\alpha$ , deg	101.10(5)	101.264(3)	100.911(5)	102.275(3)
$\beta$ , deg	107.72(4)	102.378(3)	106.931(5)	93.038(2)
$\gamma$ , deg	117.47(4)	106.468(3)	117.952(4)	107.620(3)
<i>V</i> , (Å <sup>3</sup> )	465.93	467.59(5)	467.58(8)	746.66(6)
<i>Z</i> , <i>Z'</i>	1, 1	2, 1	1, 1	1, 1
<i>D</i> <sub>calc</sub> (g cm <sup>-3</sup> )	1.572	1.883	1.725	1.614
$\mu$ (mm <sup>-1</sup> ), rad. type		5.987, Cu <i>K</i> 	4.798, Cu <i>K</i> 	3.337, Cu <i>K</i> 
F <sub>000</sub>		264	246	370
temp (K)	297(2)	100(2)	100(2)	100(2)
Crystal form, color		prism, colorless	prism, colorless	plate, colorless
Crystal size, mm		0.07 x 0.08 x 0.21	0.09 x 0.11 x 0.19	0.04 x 0.09 x 0.16
Data collection				
Diffractometer		Bruker Apex II	Bruker Apex II	Bruker Apex II
<i>T</i> <sub>min</sub> / <i>T</i> <sub>max</sub>		0.553/0.911	0.661/0.753	0.626/0.753
No. of refls. (meas., uniq., and obs.)		10122/1670/1605	10562/2949/2936	24878/2680/2263
<i>R</i> <sub>int</sub>		0.0420	0.0240	0.0456
$\theta$ <sub>max</sub> (°)		68.23	68.23	68.22
Refinement				
<i>R</i> / <i>R</i> <sup>2</sup> <sub>ω</sub> (obs data)		0.0425/0.1140	0.0331/0.0831	0.0345/0.0794
<i>R</i> / <i>R</i> <sup>2</sup> <sub>ω</sub> (all data)		0.0436/0.1158	0.0333/0.0832	0.0361/0.0803
<i>S</i>		1.10	0.98	1.06
No. of refls.		1670	2949	2680
No. of parameters		136	272	458
$\Delta\rho_{\text{max/min}}$ (e·Å <sup>-3</sup> )		1.786/-0.752	0.387/-0.298	0.310/-0.277
Flack-x		-	0.03(3)	0.018(18)

1. J. C. MacDonald, P. C. Dorrestein and M. M. Pilley, *Cryst. Growth Des.*, **2001**, 1, 29.

### S3. Hydrogen-Bond Tables

**Table S3. Hydrogen bond parameters for 2-chloro and 2-bromosuccinic acid structures.**

compound	$D\text{--H}\cdots A$	$D\text{--H} (\text{\AA})$	$H\cdots A (\text{\AA})$	$D\cdots A (\text{\AA})$	$D\text{--H}\cdots A (^{\circ})$
(-)- <b>1</b>	O1-H...O4 <sup>i</sup>	0.84(3)	1.81(3)	2.649(4)	173(6)
	O3-H...O2 <sup>ii</sup>	0.85(3)	1.82(3)	2.663(4)	174(6)
(+)- <b>2</b>	O1-H...O4 <sup>ii</sup>	0.84(3)	1.80(3)	2.647(4)	175(5)
	O3-H...O2 <sup>i</sup>	0.85(3)	1.81(3)	2.653(4)	172(5)
(+)- <b>1</b> -Im	N1-H...O1 <sup>iii</sup>	0.83(2)	1.92(2)	2.735(3)	166(4)
	O3-H...O2 <sup>iv</sup>	0.83(2)	1.77(2)	2.595(2)	176(4)
	N2-H...O2 <sup>v</sup>	0.88(2)	1.87(2)	2.752(3)	177(4)
(-)- <b>2</b> -Im	N1-H...O1 <sup>iii</sup>	0.83(3)	1.89(3)	2.670(4)	155(4)
	O3-H...O2 <sup>iv</sup>	0.82(3)	1.77(3)	2.591(4)	175(5)
	N2-H...O2 <sup>vi</sup>	0.84(3)	1.91(3)	2.743(4)	169(5)
( $\pm$ )- <b>2</b> -Im	N1-H...O1 <sup>iii</sup>	0.84(3)	1.91(2)	2.719(4)	162(5)
	O3-H...O2 <sup>ii</sup>	0.85(3)	1.69(2)	2.538(3)	174(5)
	N2-H...O2 <sup>vii</sup>	0.85(3)	1.90(2)	2.734(4)	171(5)
(+)- <b>1</b> /(-)- <b>2</b> -Im	N1-H...O1 <sup>iii</sup>	0.85(3)	1.88(3)	2.728(7)	175(9)
	O4-H...O1 <sup>iv</sup>	0.84(3)	1.73(2)	2.568(7)	171(8)
	N2-H...O2 <sup>viii</sup>	0.83(3)	1.97(4)	2.741(7)	154(6)
	N3-H...O5 <sup>ii</sup>	0.82(3)	1.94(3)	2.744(7)	165(8)
	O8-H...O5 <sup>i</sup>	0.84(3)	1.70(3)	2.535(7)	170(8)
	N4-H...O6 <sup>vii</sup>	0.83(3)	1.88(3)	2.715(7)	177(7)
(+)- <b>1</b> /(-)- <b>2</b> -BPDO	O1-H...O9 <sup>ii</sup>	0.86(3)	1.73(3)	2.561(7)	160(6)
	O3-H...O10 <sup>ix</sup>	0.85(3)	1.75(3)	2.574(7)	163(7)
	O5-H...O11 <sup>iii</sup>	0.86(3)	1.72(3)	2.561(7)	166(6)
	O7-H...O12 <sup>x</sup>	0.86(3)	1.71(3)	2.562(7)	175(7)

Symmetry codes: (i)  $x-1, y, z$ ; (ii)  $x+1, y, z$ ; (iii)  $x, y, z$ ; (iv)  $x, y, z+1$ ; (v)  $x+1, x-1, z$ ; (vi)  $x, y+1, z$ ; (vii)  $x-1, y-1, z-1$ ; (viii)  $x+1, y+1, z+1$ ; (ix)  $x+1, y+2, z+1$ ; (x)  $x-1, y-2, z-1$ .