

Supplementary Material

The Electrochemical Sensor Based on Electrochemical Oxidation of Nitrite on Metalloporphyrin-Graphene Modified Glassy Carbon Electrode

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Table S1: Analytical parameters reported for some modified electrodes towards nitrite detection

Modified electrode	Fabrication method	Detection method	Linear range	Detection limit	Reference
PEDOT/Graphene modified electrode	Immobilize	Amperometry	0.5-240 μM	0.15 μM	[1]
Graphene nanoplatelet-protein composite film	Deposition	Amperometry	50 μM-2.5mM	10 mM	[2]
HAC modified GCE	Deposition	Amperometry and DPV	1-127 μM	0.07 μM	[3]
CR-GO modified GCE	Deposition	Amperometry	8.9-167 μM	1.0 μM	[4]
CoOx/MWCNTs/GCE	Deposition	Amperometry	0.5-250 μM	0.3 μM	[5]
GCE/f-ZnO@rFGO	Deposition	Amperometry	10 μM-8 mM	33 μM	[6]
GC/GO-Ag nanocomposite	Deposition	Amperometry LSV	10-180μM	37 μM 2.1 μM	[7]
GO-MnNH ₂ TPP	Deposition	Amperometry	10-160μM	2.5μM	This work

References

- 1 Q. Wang and Y. Yun, *Microchim Acta*, 2012, **177**, 411-418.
- 2 L. Cui, X. Meng, M. Xu, K. Shang, S. Ai and Y. Liu, *Electrochim. Acta*, 2011,

- 56**, 9769-9774.
- 3 R. Madhu, V. Veeramani and S. Chen, *Sci Rep-UK*, 2014, **4**.
- 4 V. Mani, A. P. Periasamy and S. Chen, *Electrochem. Commun.*, 2012, **17**, 75-78.
- 5 Z. Meng, B. Liu, J. Zheng, Q. Sheng and H. Zhang, *Microchim Acta*, 2011, **175**, 251-257.
- 6 A. Pandikumar, N. Yusoff, N. M. Huang and H. N. Lim, *Microchim Acta*, 2015, **182**, 1113-1122.
- 7 N. I. Ikhsan, P. Rameshkumar, A. Pandikumar, M. M. Shahid, N. M. Huang, S. V. Kumar and H. N. Lim, *Talanta*, 2015, **144**, 908-914.