Supplementary Information

Thermoresponsive dynamic covalent dendronized polymers

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Scheme S1 Synthetic procedures for the model compounds. Reagents and conditions: (a) Methyl 3-hydroxybenzoate, KI, K_2CO_3 , DMF, 80 °C (**2e**: 66%); (b)LiAlH₄, THF, -5 °C to 25 °C (**2f**: > 99%);(c) PCC, DCM, 0 °C to 25 °C (**5**: 65%); (d) hydrazine hydrate, MeOH (**6**: > 99%).



Scheme S2 Dimerization reaction of model system between 5 and 6.



Fig. S1 Plots of transmittance *vs* temperature for 0.25 wt % buffer solutions (pH = 1.6) of a) **PMeG1** after addition of **G1EtH**, b) **PEtG1** after addition of **G1MeH** and c) **PEtG2** after addition of **G1MeH**. Heating rate = $0.2 \text{ °C} \cdot \text{min}^{-1}$.



Fig. S2 UV/vis spectra of aqueous solutions of **PEtG1** at different temperatures, [**PEtG1**] = 39 μg·mL⁻¹.



Fig. S3 UV/vis spectra of aqueous solutions of **PEtG2** and CuSO₄ at different temperatures, [-C=N-]: [Cu²⁺] = 1:1, [**PEtG2**] = 25.5 μ g·mL⁻¹.

Compound 2e. According to general procedure A from Methyl 3-hydroxybenzoate (0.30 g, 1.97 mmol), **1b** (1.1 g, 1.79 mmol), KI (0.20 g, 1.20 mmol), K₂CO₃ (0.74 g, 5.35 mmol) and dry DMF, **2e** was yielded as a yellow oil (0.86 g, 66%). ¹H NMR (CDCl₃): δ = 1.21-1.24 (m, 9H, CH₃), 3.51-3.56 (m, 6H, CH₂), 3.59-3.62 (m, 6H, CH₂), 3.66-3.69 (m, 12H, CH₂), 3.80 (br, 6H, CH₂), 3.82 (br, 2H, CH₂), 3.88 (br, 4H, CH₂), 3.94 (s, 3H, OCH₃), 4.16-4.19 (m, 6H, CH₂), 5.01 (s, 2H, CH₂), 6.70 (s, 2H, CH), 7.16-7.18 (d, 1H, CH), 7.37 (t, 1H, CH), 7.65-7.68 (m, 2H, CH).

Compound 2f. According to general procedure B from LiAlH₄ (0.04 g, 1.05 mmol), **2e** (0.59 g, 0.81 mmol) and dry THF, **2f** was yielded as a bright yellow oil (0.57 g, > 99%).

Compound 5. According to general procedure C from PCC (0.23 g, 1.07 mmol), **2b** (0.57 g, 0.81 mmol) and dry DCM, **5** was yielded as a bright yellow oil (0.37 g, 65%). ¹H NMR (CDCl₃): δ = 1.20-1.24 (m, 9H, CH₃), 3.51-3.57 (m, 6H, CH₂), 3.59-3.62 (m, 6H, CH₂), 3.66-3.69 (m, 12H, CH₂), 3.74-3.76 (m, 6H, CH₂), 3.82 (t, 2H, CH₂), 3.87 (t, 4H, CH₂), 4.16-4.20 (m, 6H, CH₂), 5.03 (s, 2H, CH₂), 6.70 (s, 2H, CH), 7.25-7.27 (m, 1H, CH), 7.47-7.52 (m, 3H, CH), 10.00 (s,1H, HC=O).

Compound 6. According to general procedure D from **2e** (0.29 g, 0.40 mmol), hydrazine hydrate (0.90 mL, 22.04 mmol) and anhydrous MeOH, **6** was yielded as a bright yellow solid (0.29 g, > 99%). ¹H NMR (DMSO- d_6): δ = 1.01-1.08 (m, 9H, CH₃), 3.36-3.40 (m, 6H, CH₂), 3.42-3.45 (m, 6H, CH₂), 3.48-3.52 (m, 12H, CH₂), 3.55-3.58 (m, 6H, CH₂), 3.65 (t, 2H, CH₂), 3.72 (t, 4H, CH₂), 4.00 (t, 2H, CH₂), 4.08 (t, 4H, CH₂), 4.57 (br, 2H, NH₂), 5.03 (s, 2H, CH₂), 6.79 (s, 2H, CH), 7.13-7.16 (m, 1H, CH), 7.35-7.38 (m, 1H, CH), 7.41-7.43 (m, 1H, CH), 7.46-7.47 (m, 1H, CH), 9.77 (s, 1H, NH).



S6



Fig. S7 ¹H NMR spectrum of G1MeA in CDCl₃.



Fig. S9 ¹H NMR spectrum of **G1MeH** in DMSO-*d*₆.



Fig. S11 ¹H NMR spectrum of 2c in CDCl₃.







Fig. S13 ¹H NMR spectrum of 2d in CDCl₃.





Fig. S17 ¹H NMR spectrum of G1EtH in DMSO-d₆.





Fig. S21 ¹³C NMR spectrum of 4a in CDCl₃.



Fig. S23 ¹³C NMR spectrum of G2EtA in CDCl₃.



Fig. S25 ¹³C NMR spectrum of G2EtH in CDCl₃.



Fig. S27 ¹H NMR spectrum of 5 in CDCl₃.



Fig. S28 ¹H NMR spectrum of 6 in DMSO-d₆.