

Supporting Information of

A novel δ -MnO₂ with carbon nanotubes nanocomposites as enzyme-free sensor for
hydrogen peroxide electroensing

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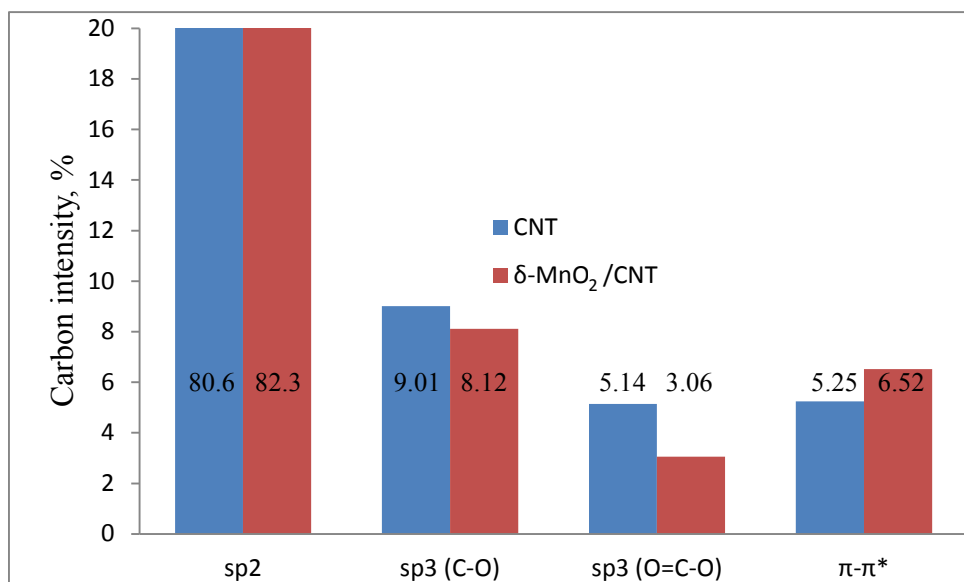


Figure S1: Distribution of carbon species obtained from the C1s peaks by XPS.

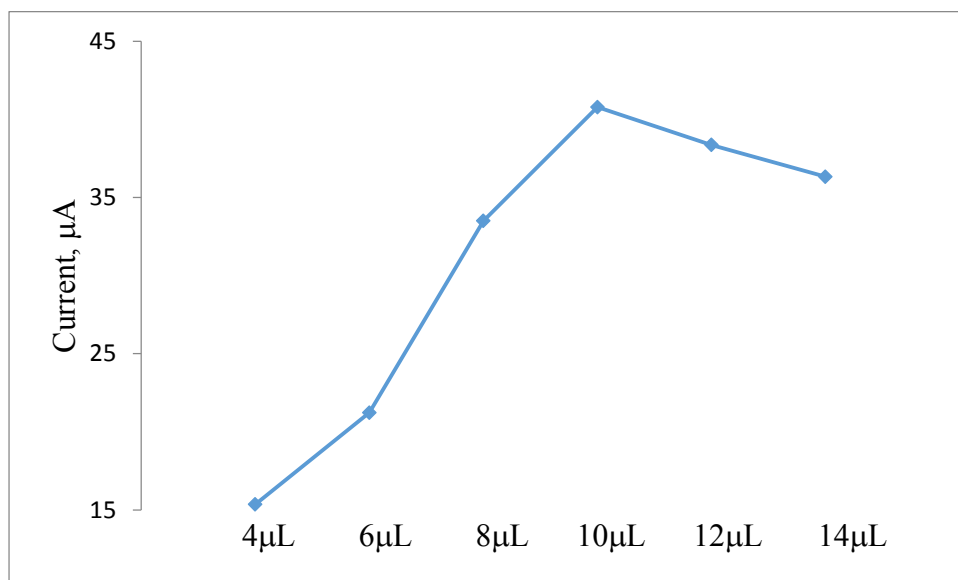


Figure S2: Plot of 1 mM H₂O₂ current response vs. δ-MnO₂/CNTs ink loading (μL) onto GCE by CVs in Ar-saturated 0.1 M PBS at 50 mV s⁻¹ scan rate.

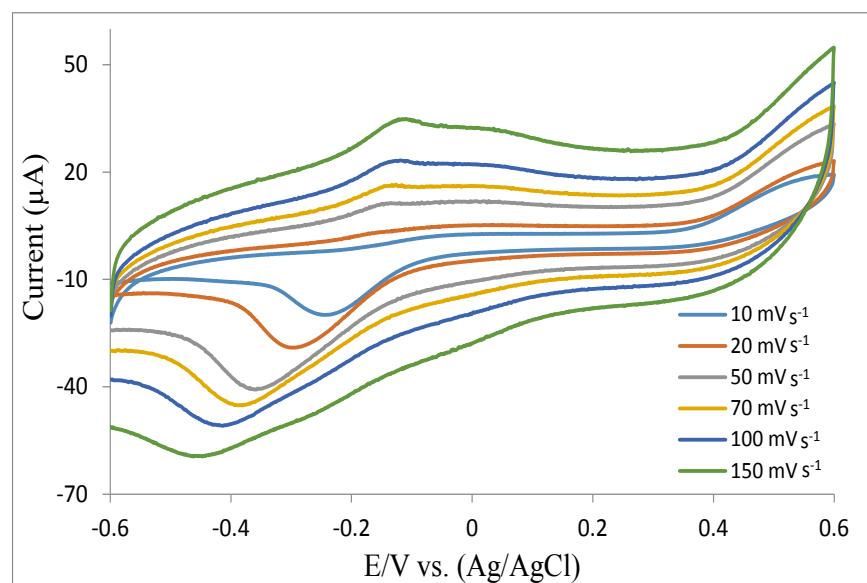


Figure S3: Scan rate dependent CVs recorded in Ar-saturated 0.1 M PBS at pH 7.4 on δ-MnO₂/CNTs/GCE at 10 to 150 mV s⁻¹ scan rates.

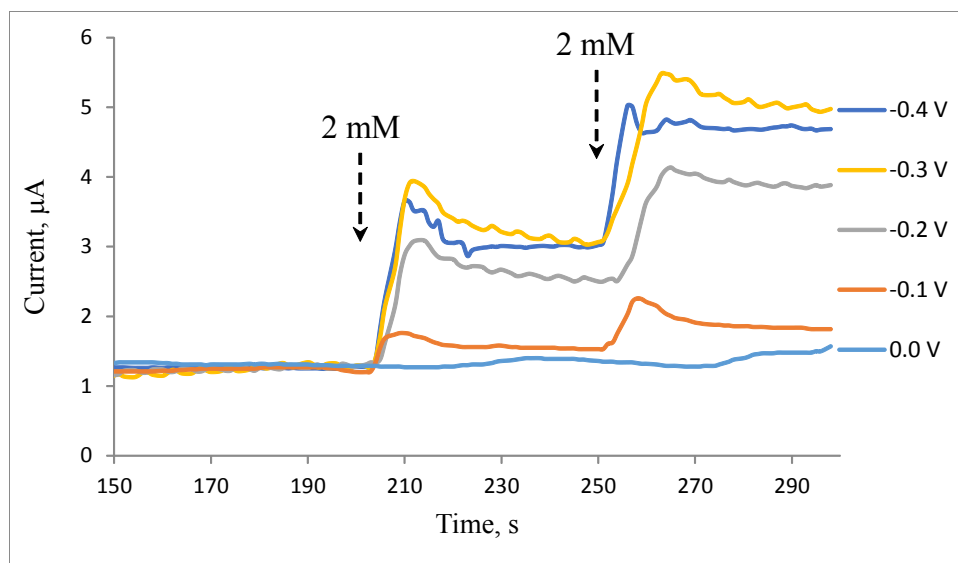


Figure S4: Amperometric response on δ -MnO₂/CNTs/GCE upon addition of 2 mM H₂O₂ in Ar-saturated 0.1 M PBS at different applied potentials for applied potential optimization.

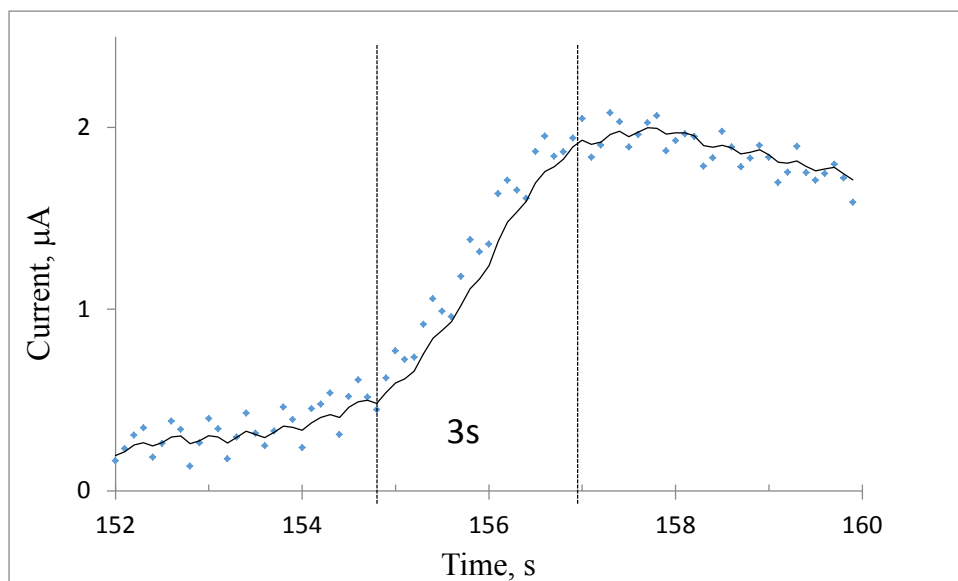


Figure S5: Enlarged amperometric response on δ -MnO₂/CNT/GCE upon addition H₂O₂ in 0.1 M PBS at an applied potential of -0.3 V at cited time range.

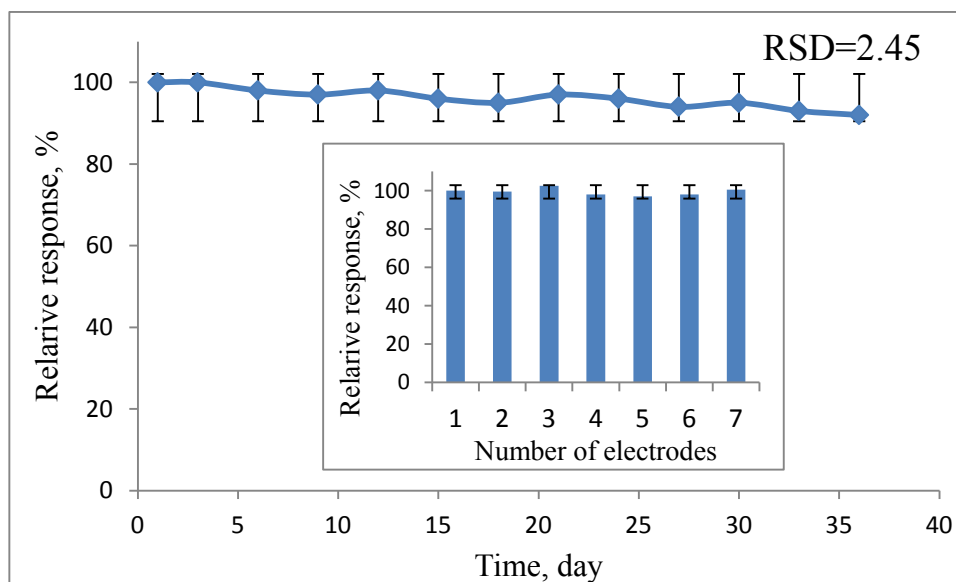


Figure S6: The stability and reproducibility of $\delta\text{-MnO}_2/\text{CNT}/\text{GCE}$ by CVs response in Ar-saturated 0.1 M PBS with 1 mM H_2O_2 at a scan rate of 50 mV s^{-1} ; inset: the reproducibility of seven different modified electrodes.

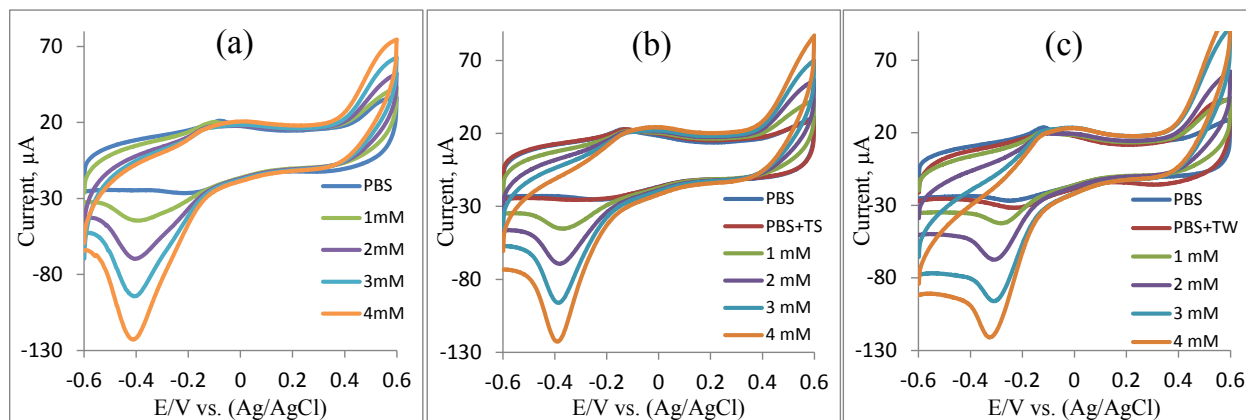


Figure S7: CVs response for real sample analysis on $\delta\text{-MnO}_2/\text{CNT}/\text{GCE}$ in Ar-saturated 0.1 M PBS (a), 0.1 M PBS + Tomato sauce (TS) (b), and 0.1 M PBS + Tap water (TW) (c) with the subsequent addition of 1, 2, 3, 4 mM H_2O_2 at a scan rate of 50 mV s^{-1} .