## **Supplementary Information**

## Preparation of platinum nanoparticles immobilized on ordered

## mesoporous Co<sub>3</sub>O<sub>4</sub>-CeO<sub>2</sub> composites and enhanced catalytic activity

Qianli Wang <sup>a</sup>, Yiwei Zhang <sup>\*a</sup>, Yuming Zhou <sup>\*a</sup>, Zewu Zhang <sup>b</sup>,

Yuanmei Xu<sup>a</sup>, Chao Zhang<sup>a</sup>, Hongxing Zhang<sup>a</sup> and Xiaoli Sheng<sup>a</sup>

<sup>a</sup> School of Chemistry and Chemical Engineering, Southeast University, Nanjing 211189, China

<sup>b</sup> School of Materials Engineering, Nanjing Institute of Technology, Nanjing 211167, China



Fig. S1 TG curve of the mixture of  $Ce(NO_3)_3.6H_2O$  and  $Co(NO_3)_3.6H_2O$  loaded on KIT-6 (Co/(Ce+Co)=30 mol%).

E-mail: zhangchem@seu.edu.cn, ymzhou@seu.edu.cn Fax: +86 25 52090617 Tel: +86 25 52090617



Fig. S2 EDX analysis of (a) meso-CeO<sub>2</sub>, (b) meso-CeO<sub>2</sub>Co10 and (c) meso-CeO<sub>2</sub>Co30.



Fig. S3 TEM image of PAMAM dendrimers-templated Pt NPs (inset is the particle size distribution histogram of Pt NPs).

Catalyst	k (s <sup>-1</sup> )	$k_{nor} (k/m) (s^{-1} g^{-1})$	Ref.
Pt/meso-CeO <sub>2</sub> Co10	11.30×10 <sup>-3</sup>	75.33	This work
Pt/meso-CeO <sub>2</sub>	6.03×10 <sup>-3</sup>	40.20	This work
Pt nanoflowers	7.00×10 <sup>-4</sup>	7.80	1
RGO/Pt/CeO <sub>2</sub>	69.92×10 <sup>-3</sup>	11.28	2
Fe <sub>3</sub> O <sub>4</sub> @PDA-Pt	2.30×10 <sup>-3</sup>	57.50	3
PtNi/RGO	2.17×10 <sup>-3</sup>	43.40	4
mSiO <sub>2</sub> /Pt/CeO <sub>2</sub> /Fe	2.87×10 <sup>-3</sup>	15.06	5

 

 Table S1 Comparison of catalytic activity by Pt-based catalysts for the reduction of 4nitrophenol

## References

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