

Supporting Information

Diffusion Kinetics of Gold in TiO₂ Nanotube Arrays for formation of Au@TiO₂ Nanotube
Arrays

Wanggang Zhang^{1,2,3}, Yiming Liu^{1,2}, Diaoyu Zhou^{1,2}, Jing Wen^{1,2}, Liuwei Zheng^{1,2}, Wei
Liang^{1,2,*} and Fuqian Yang^{3,*}

¹ College of Materials Science and Engineering, Taiyuan University of Technology, Taiyuan
Shanxi 030024, China

² Key Laboratory of Interface Science and Engineering in Advanced Materials, Taiyuan
University of Technology, Ministry of Education, Taiyuan Shanxi 030024, China

³ Department of Chemical and Materials Engineering, University of Kentucky, Lexington,
KY 40506, USA

Figure S1 shows SEM images of pure TiO₂ nanotubes arrays prepared by the one-step anodization process and the two-step anodization process, and SEM image of the imprints left on the surface of the Ti sheet after ultrasonication.

Figure S2 shows the EDS analyses of the bottom and upper sides of the Au@TiO₂ nanotubes arrays after heat treatment at 450 °C for 3 hr.

Figure S3 shows SEM images of the bottom side of Au@TiO₂ nanotubes arrays with heat treatment at both 450 and 500 °C for 5 hr.

Figure S4 shows TEM images of Au@TiO₂ nanotubes arrays after heat treatment at 500 °C for three different heating times and the corresponding size distribution of Au nanocrystals.

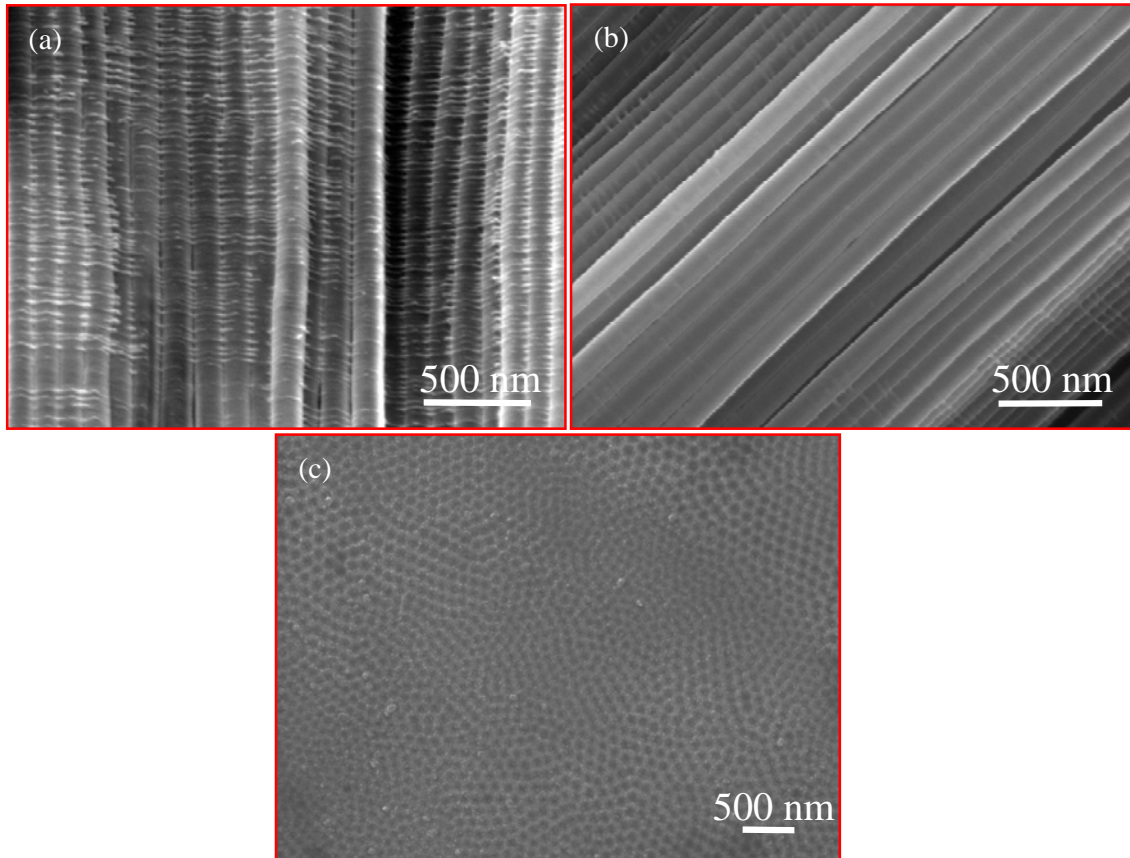
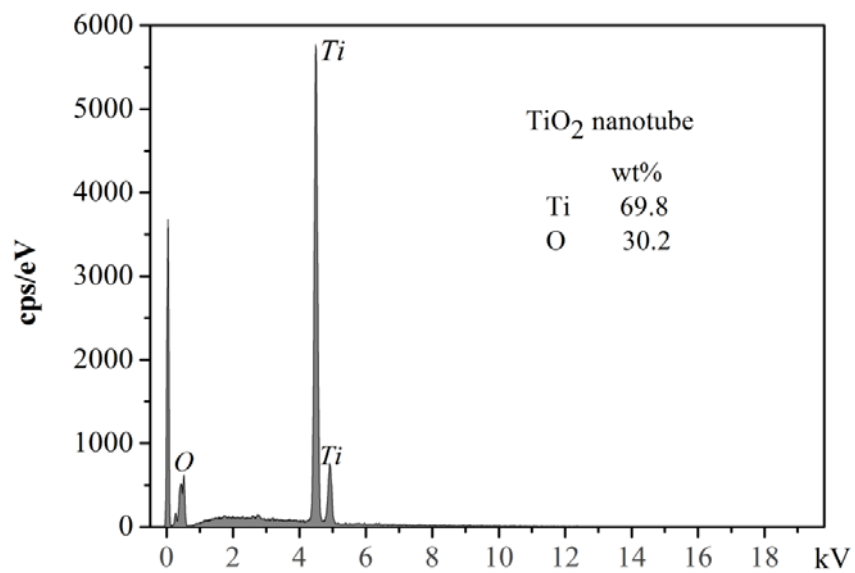
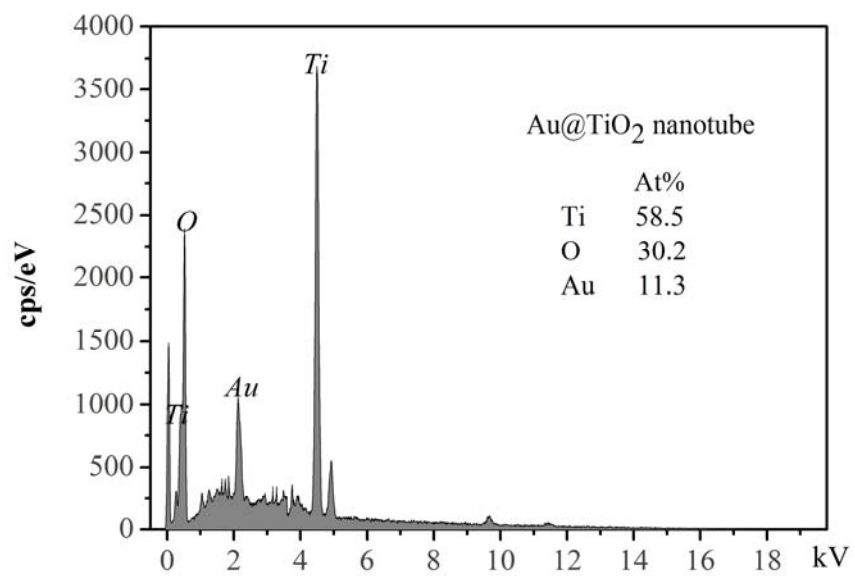


Figure S1. SEM images of pure TiO₂ nanotubes arrays; (A) TiO₂ nanotubes arrays prepared by the one-step anodization process, (b) TiO₂ nanotubes arrays prepared by the two-step anodization process, and (3) SEM image of the imprints left on the surface of the Ti sheet after ultrasonication



(a)



(b)

Figure S2. EDS analysis of Au@TiO_2 nanotubes arrays after heat treatment at 450°C for 3 hr; (a) bottom side and (b) upper side

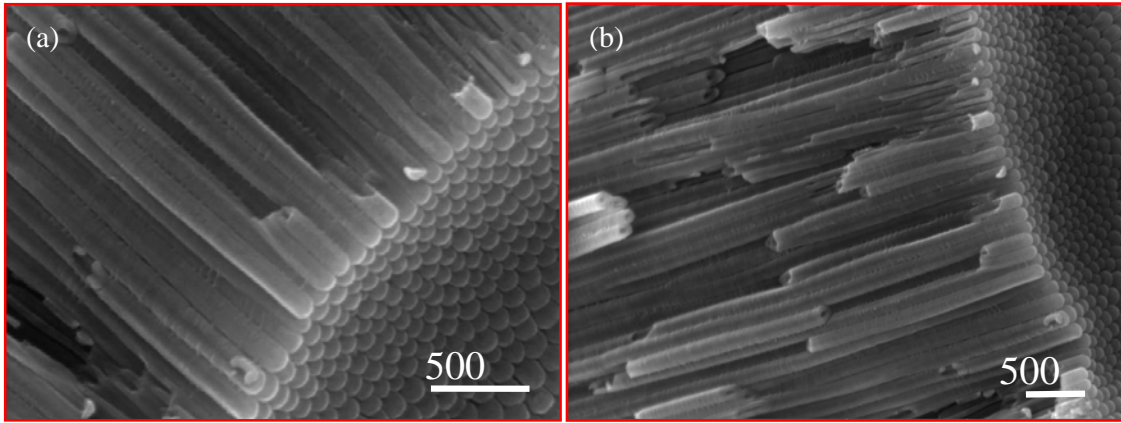


Figure S3. SEM images of the bottom side of Au@TiO₂ nanotubes arrays (a) with heat treatment at 450 °C for 5 hr and (b) with heat treatment at 500 °C for 5 hr

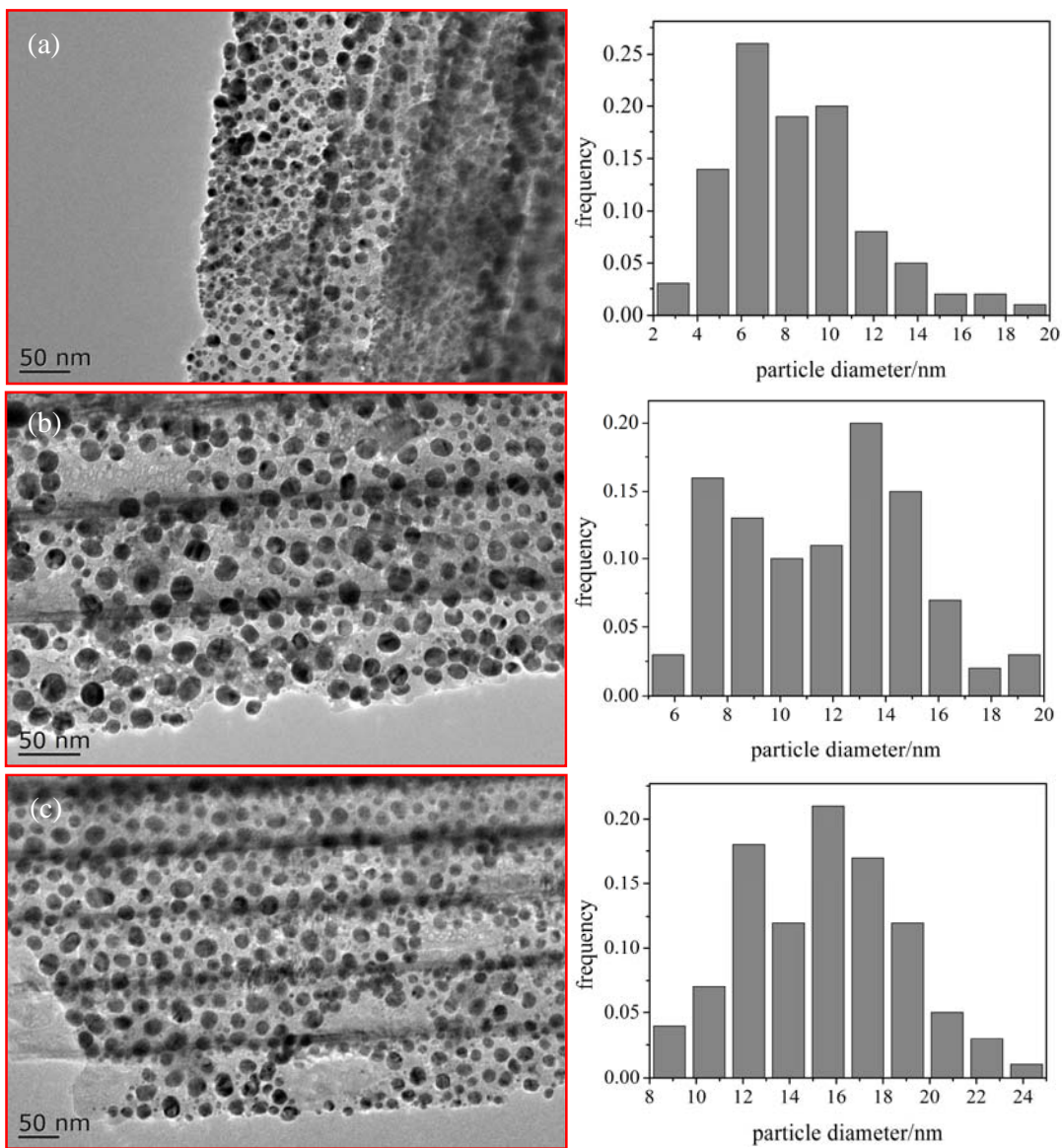


Figure S4. TEM images of Au@TiO₂ nanotubes arrays after heat treatment at 500 °C for three different times and the corresponding size distribution of Au nanocrystals; (a) 1 hr (average size: 8.55 nm), (b) 3 hr (average size: 11.68 nm); (c) 5 hr (average size: 15.53 nm)

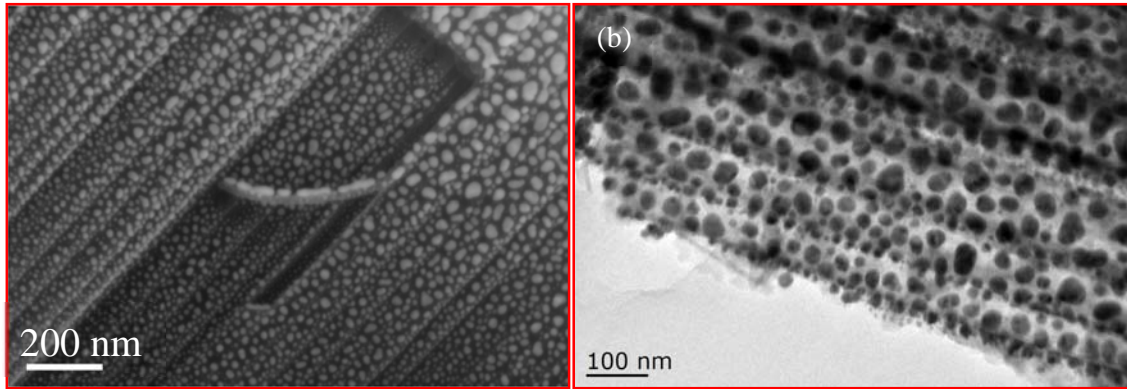


Figure S5 (a) SEM image of Au@TiO₂ nanotubes arrays after heat treatment at 450 °C for 3 hr in argon environment, and (b) TEM image of Au@TiO₂ nanotubes arrays after heat treatment at 450 °C for 3 hr in argon environment