

Supporting information

Additive Effect of Fluoroethylene and Difluoroethylene Carbonates for the Solid Electrolyte Interphase Film Formation in Sodium-Ion Batteries: A Quantum Chemical Study

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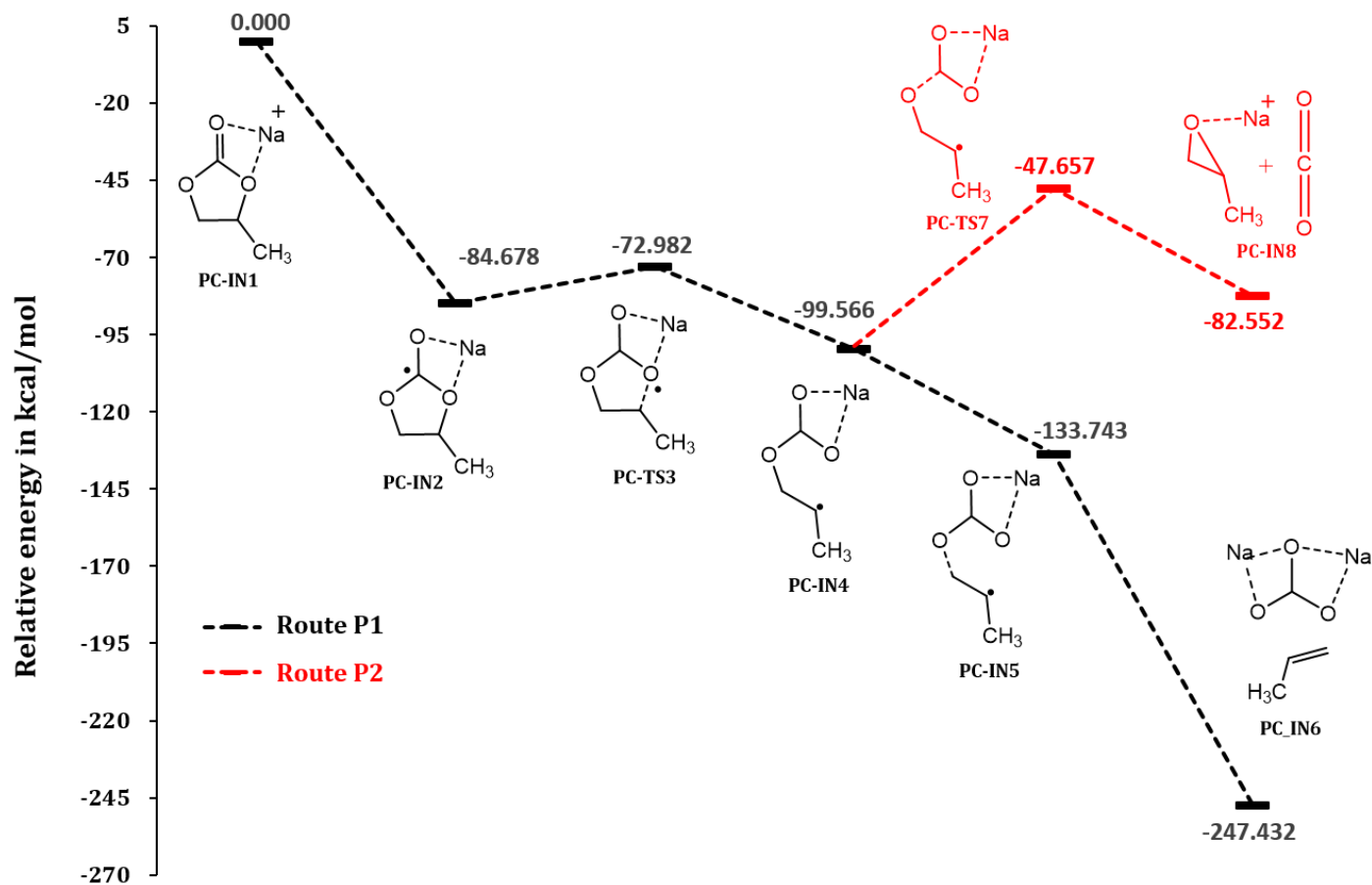


Figure S1: Potential energy profiles (ΔE in kcal/mol) for the possible reductive decomposition of Na^+ -propylene carbonate (Na^+ -PC) in gas phase.

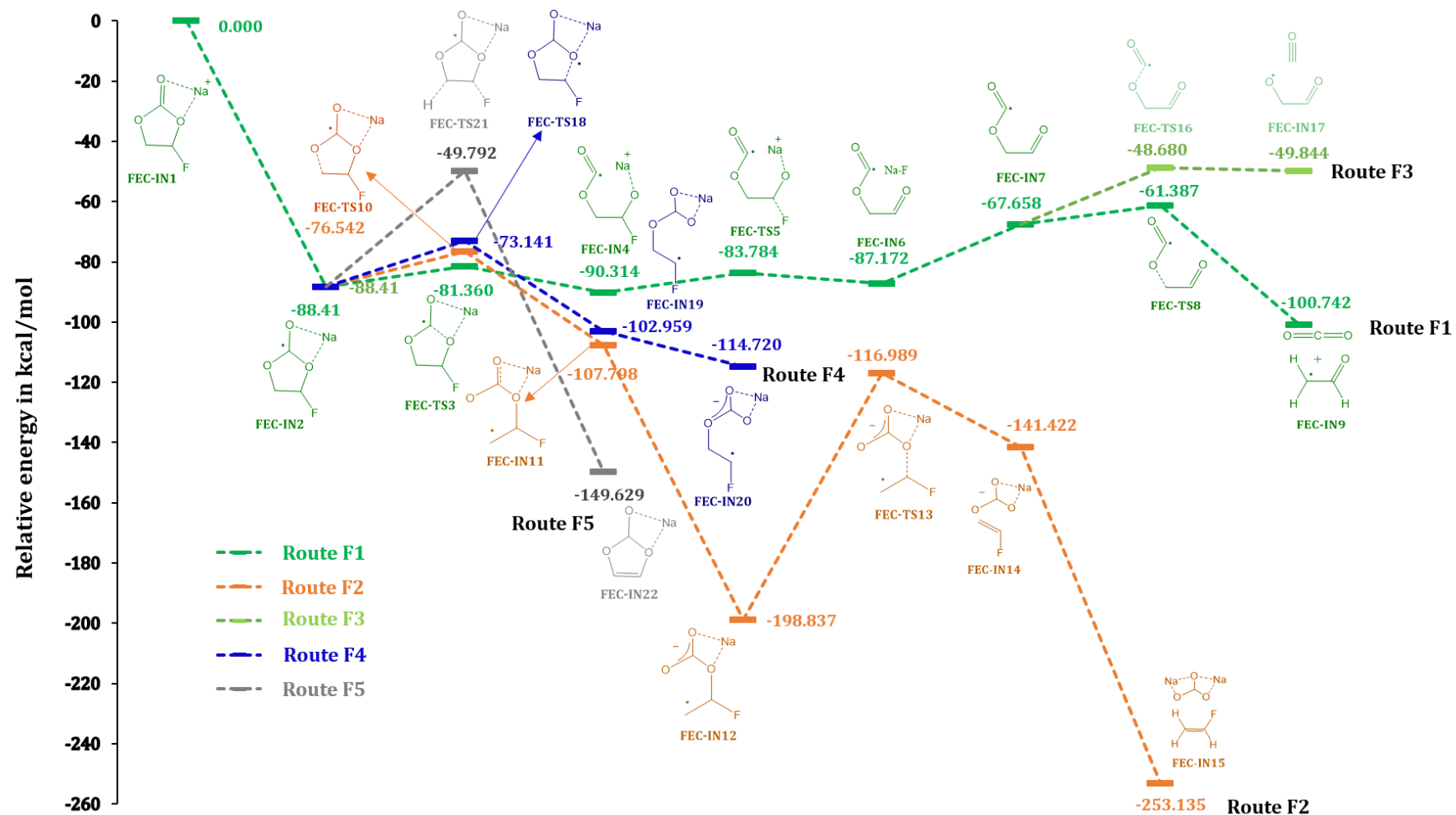


Figure S2: Potential energy profiles (ΔE in kcal/mol) for the possible reductive decomposition of Na^+ -FEC in gas phase.

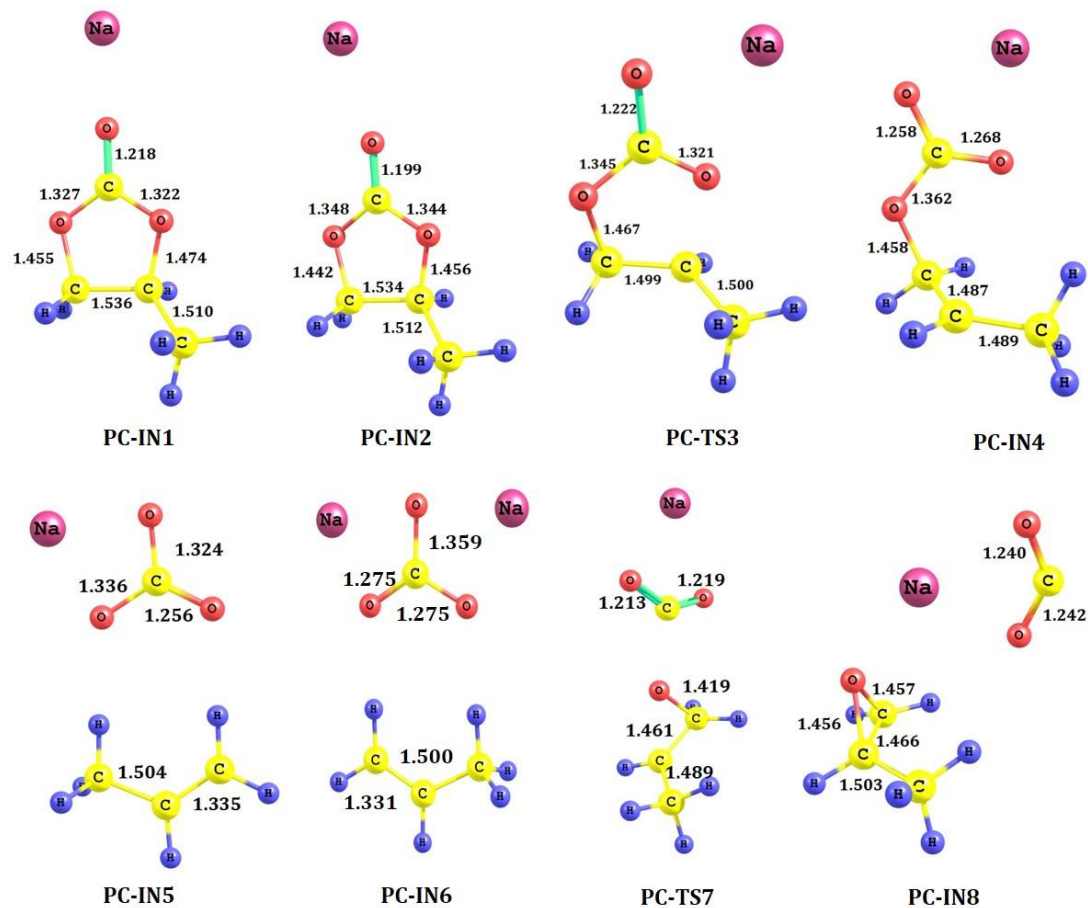
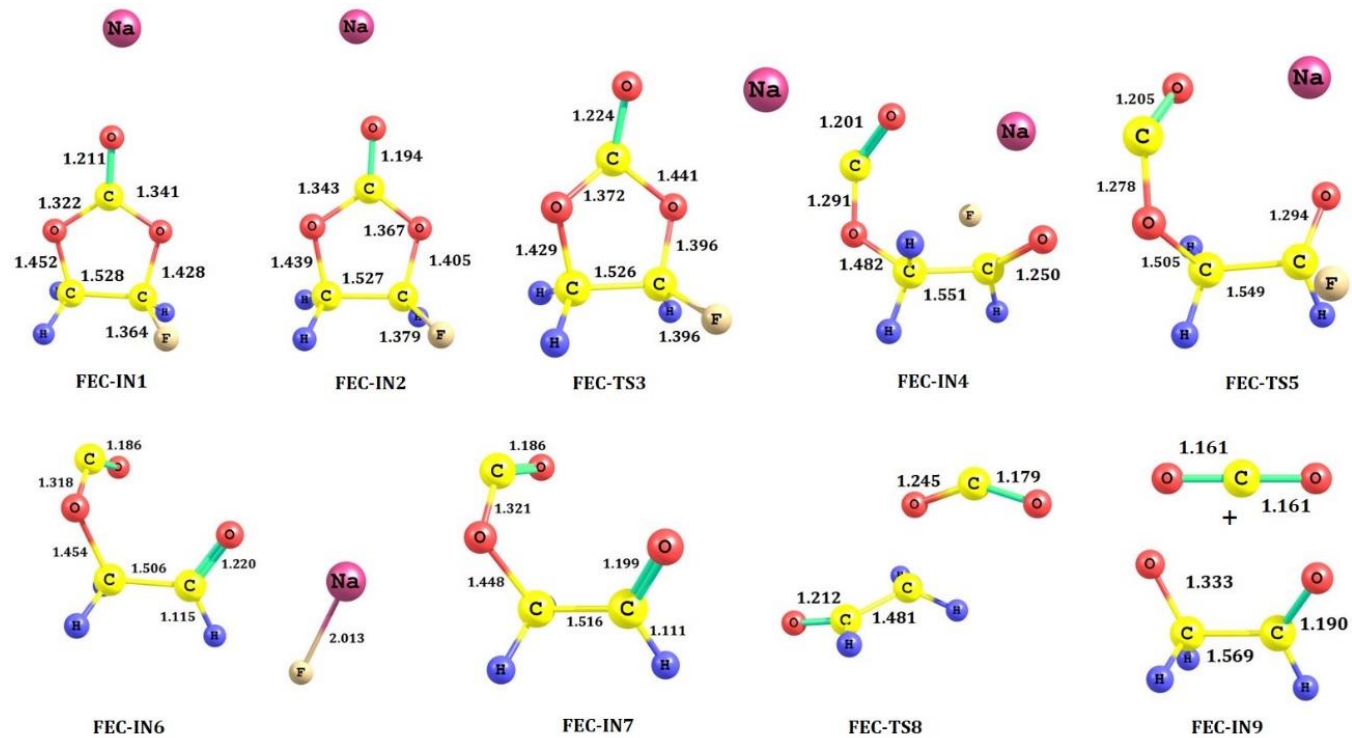


Figure S3: Optimized geometries with respective bond lengths of intermediates, products and transition states in the Na^+ -propylene carbonate (Na^+ -PC) reductive decomposition mechanism calculated at B3LYP/6-311++G(d,p) level of theory.



Route 1

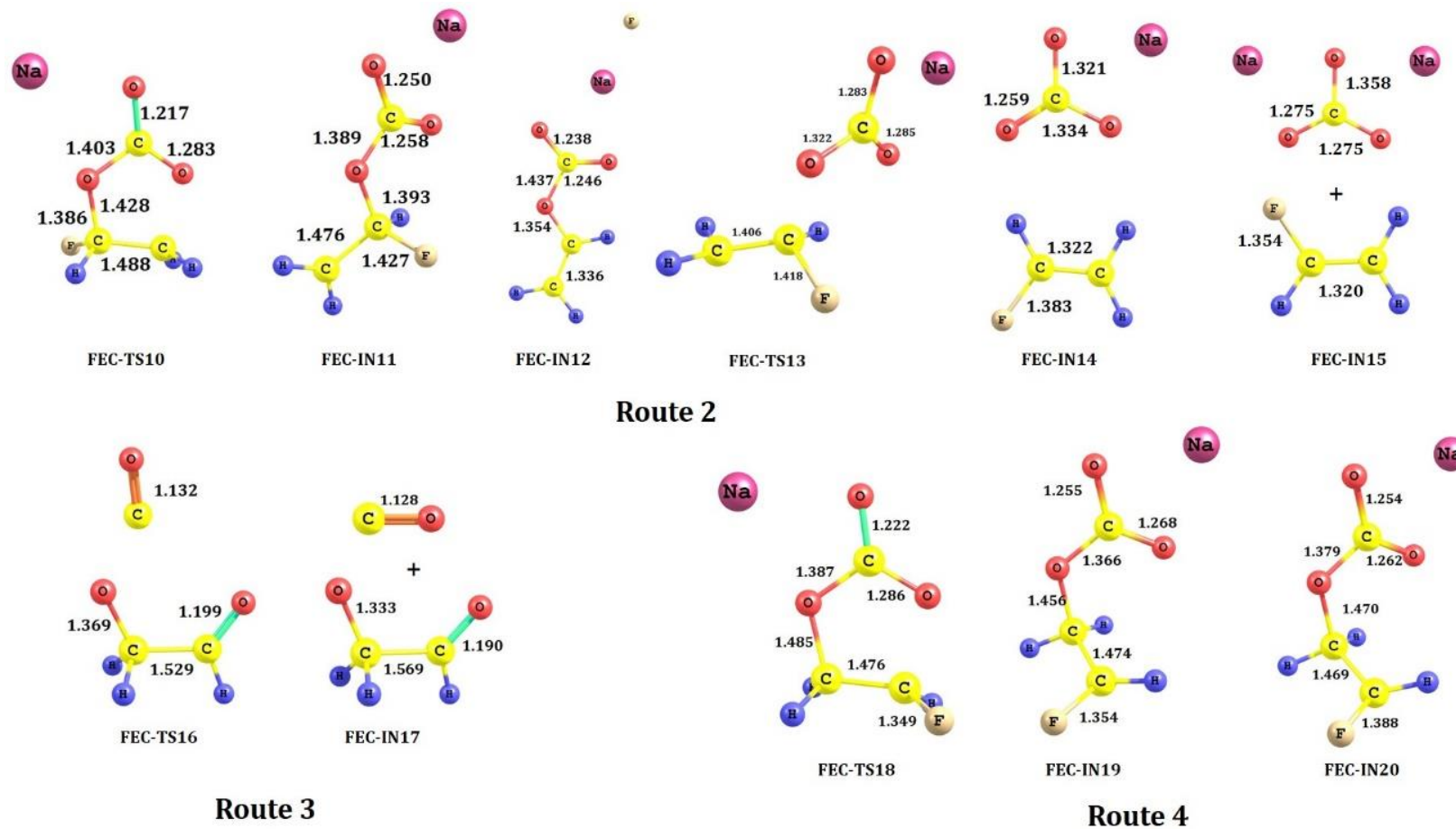
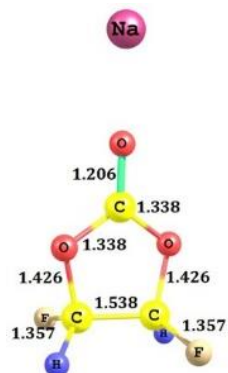
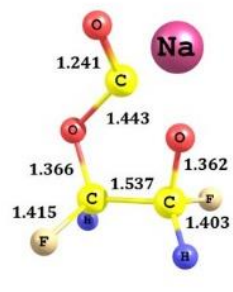


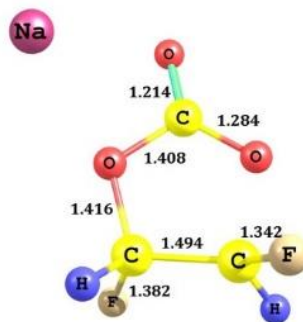
Figure S4: Optimized geometries with respective bond lengths of intermediates, products and transition states formed in the Na^+ -fluoroethylene carbonate (Na^+ -FEC) reductive decomposition mechanism calculated at B3LYP/6-311++G(d,p) level of theory.



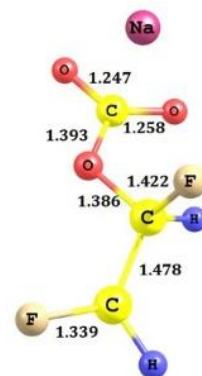
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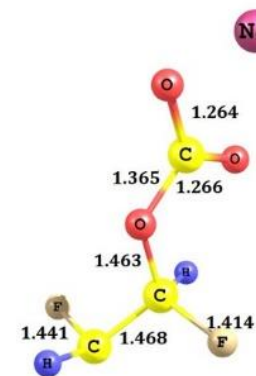
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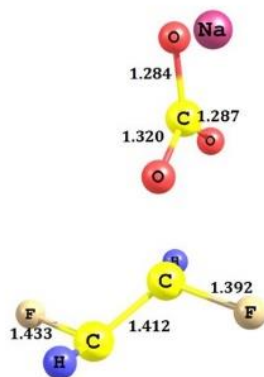
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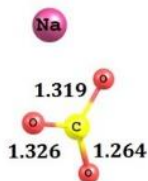
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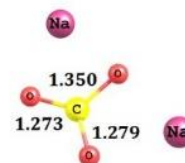
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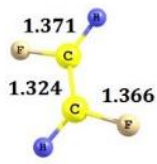
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DFEC-IN7



DFEC-IN8



Route 1

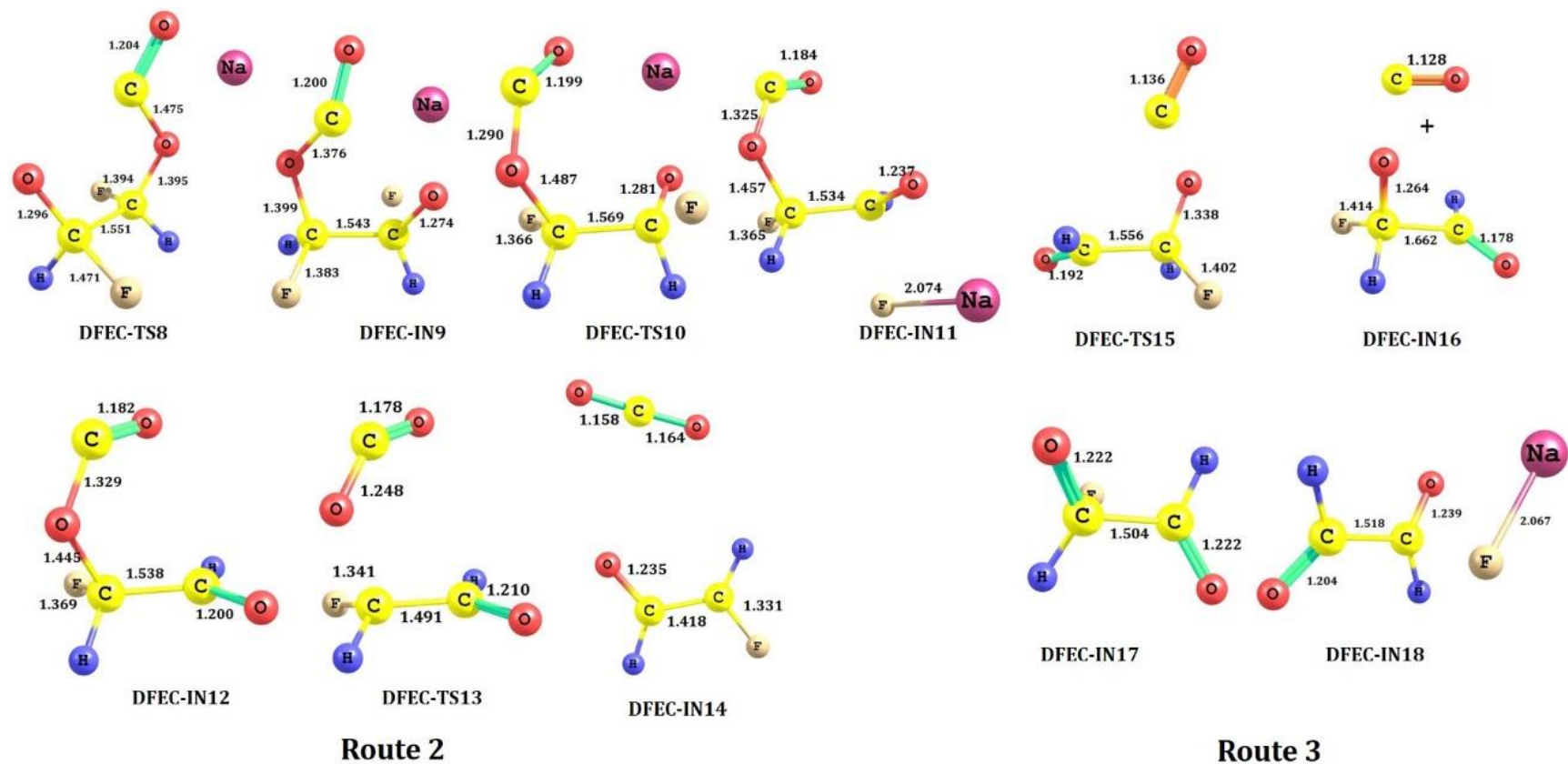


Figure S5: Optimized geometries with respective bond lengths of intermediates, products and transition states formed in the Na^+ -difluoroethylene carbonate (Na^+ -DFEC) reductive decomposition mechanism calculated at B3LYP/6-311++G(d,p) level of theory.