

Electronic Supplementary Information

Self-assembly of fluorescent diimidazolium salts: tailor properties of the aggregates changing alkyl chain features

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Synthesis of *N,N'*-bis-(3-imidazol-yl-propyl)naphthalene-diimide.

1,4,5,8-naphthalenetetracarboxylic dianhydride (1.00 g; $3.73 \cdot 10^{-3}$ mol) was dissolved in 12.5 mL of anhydrous DMF, and the solution obtained was heated at 80 °C. Into a two-neck flask, 1-(3-aminopropyl)imidazole (1.026 g; $8.21 \cdot 10^{-3}$ mol) was dissolved in 12.5 mL of anhydrous DMF and obtained solution was heated at 140 °C.

The solution of 1,4,5,8-naphthalenetetracarboxylic dianhydride was added to the 1-(3-aminopropyl)imidazole solution. The mixture was stirred at 140 °C for 24 hours.

The resulting mixture was allowed to cool down, poured in water/acetone (3/1; 30 mL) mixture, then in cold diethyl ether. Stirring for 30 minutes afforded a dark precipitate. The solid was filtered off in vacuo and thoroughly washed with diethyl ether and small amounts of ethanol.

Yield: 70%; dark solid. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6); δ (ppm): 8.63 (s, 4H); 7.66 (s, 2H); 7.21 (s, 2H); 6.87 (s, 2H); 4.10 (m, 8H); 2.13 (m, 4H).

General synthesis of *N,N'*-bis-(1-alkyl-3-propylimidazolium)naphthalene-diimidediiodide salts.

N,N'-bis-(3-imidazol-yl-propyl)naphthalene-diimide (0.25 g; $5.19 \cdot 10^{-4}$ mol) was dissolved in 10 mL of anhydrous DMF, and solution obtained was heated at 80 °C. Into a two-neck flask, the suitable alkyl iodide ($1.14 \cdot 10^{-3}$ mol) was dissolved in 10 mL of anhydrous DMF; the solution was then heated at 90 °C.

The solution of *N,N'*-bis-(3-imidazol-yl-propyl)naphthalene-diimide was added to the alkyl iodide solution. The mixture was stirred at 90 °C for 72 hours.

The solvent was removed in vacuo and the dark solid obtained was washed with diethyl ether (20 mL), with ultrasounds irradiation until a colourless organic phase was obtained. Finally, the dark solid was washed with refluxing ethyl acetate (50 mL) overnight.

***N,N'*-bis-(1-hexyl-3-propylimidazolium)naphthalene-diimidediiodide [C_6NDI][I].**

Yield: 90%; dark solid; m. p.: 170.8 °C. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6); δ (ppm): 9.21 (s, 2H); 8.70 (s, 4H); 7.83 (d, $J = 8.0$ Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.28 (m, 4H), 1.79 (m, 4H), 1.27 (bs, 12H), 0.86 (m, 6H). $^{13}\text{C-NMR}$ (400 MHz, DMSO- d_6); δ (ppm): 163.3; 136.6; 130.9; 126.9; 122.9; 49.4; 47.5; 31.0; 29.7; 28.6; 25.5; 22.3; 14.2. HRMS calcd. for $\text{C}_{38}\text{H}_{48}\text{N}_6\text{O}_4$ 326.1863, found 326.1899.

***N,N'*-bis-(1-heptyl-3-propylimidazolium)naphthalene-diimidediiodide [C_7NDI][I].**

Yield: 87%; dark solid; m. p.: 170.1 °C. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6); δ (ppm): 9.25 (s, 2H); 8.70 (s, 4H); 7.83 (d, $J = 8.0$ Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.27 (m, 4H); 1.79 (m, 4H); 1.25 (m, 16H); 0.84 (t, $J = 4.0$ Hz, 6H). $^{13}\text{C-NMR}$ (400 MHz, DMSO- d_6); δ (ppm): 163.3; 136.6; 130.9; 126.9; 126.7; 122.9; 31.7; 29.4; 25.9; 22.5; 14.4. HRMS calcd. for $\text{C}_{40}\text{H}_{52}\text{N}_6\text{O}_4$ 340.2019, found 340.2057.

***N,N'*-bis-(1-octyl-3-propylimidazolium)naphthalene-diimidediiodide [C₈NDI][I].**

Yield: 83%; dark solid; m. p.: 146.9 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.21 (s, 2H); 8.70 (s, 4H); 7.83 (d, J = 8.0 Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.28 (m, 4H); 1.80 (m, 4H); 1.25 (m, 20H); 0.84 (m, 6H). ¹³C-NMR (400 MHz, DMSO-d₆); δ (ppm): 163.3; 130.9; 130.8; 126.9; 123.0; 122.9; 49.4; 47.4; 31.6; 31.4; 29.7; 28.9; 25.9; 22.4; 14.3. HRMS calcd. for C₄₂H₅₆N₆O₄ 354.2176, found 354.2213.

***N,N'*-bis-(1-nonyl-3-propylimidazolium)naphthalene-diimidediiodide [C₉NDI][I].**

Yield: 85%; dark solid; m. p.: 200.2 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.21 (s, 2H); 8.70 (s, 4H); 7.83 (d, J = 8.0 Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.28 (m, 4H); 1.79 (s, 4H); 1.24 (m, 24H); 0.83 (m, 6H). ¹³C-NMR (400 MHz, DMSO-d₆); δ (ppm): 163.3; 136.6; 130.9; 126.9; 122.9; 49.3; 37.6; 31.7; 30.6; 29.8; 29.0; 25.9; 22.5; 14.4. HRMS calcd. for C₄₄H₆₀N₆O₄ 368.2332, found 368.2337.

***N,N'*-bis-(1-decyl-3-propylimidazolium)naphthalene-diimidediiodide [C₁₀NDI][I].**

Yield: 85%; dark solid; m. p.: 104.0 – 106.8 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.22 (s, 2H); 8.70 (s, 4H); 7.84 (d, J = 8.0 Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.27 (m, 4H); 1.79 (m, 4H); 1.22 (m, 28H); 0.83 (m, 6H). ¹³C-NMR (300 MHz, DMSO-d₆); δ (ppm): 163.4; 136.6; 130.9; 126.9; 126.7; 122.9; 55.5; 49.4; 47.4; 37.7; 34.8; 31.7; 29.8; 29.4; 29.3; 29.1; 28.8; 25.9; 22.5; 14.4. HRMS calcd. for C₄₄H₆₀N₆O₄ 382.2489, found 382.2500

***N,N'*-bis-(1-propyl-3-undecylimidazolium)naphthalene-diimidediiodide [C₁₁NDI][I].**

Yield: 88%; dark solid; m. p.: 202.6 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.21 (s, 2H); 8.70 (s, 4H); 7.83 (d, J = 8.0 Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.27 (m, 4H); 1.79 (m, 4H); 1.22 (m, 32H); 0.83 (m, 6H). ¹³C-NMR (400 MHz, DMSO-d₆); δ (ppm): 163.3; 136.6; 130.9; 126.9; 126.7; 122.9; 49.4; 47.4; 46.6; 37.6; 34.9; 31.7; 29.7; 29.4; 29.2; 29.1; 28.8; 28.5; 25.9; 22.5; 14.4. HRMS calcd. for C₄₆H₆₄N₆O₄ 396.2645, found 396.2656.

***N,N'*-bis-(1-dodecyl-3-propylimidazolium)naphthalene-diimidediiodide [C₁₂NDI][I].**

Yield: 88%; dark solid; m. p.: 144.0 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.22 (s, 2H); 8.70 (s, 4H); 7.84 (d, J = 8.0 Hz, 4H); 4.32 (m, 5H); 4.17 (m, 7H); 2.27 (m, 4H); 1.79 (s, 4H); 1.20 (m, 36H); 0.83 (m, 6H). ¹³C-NMR (400 MHz, DMSO-d₆); δ (ppm): 163.3; 156.1; 136.6; 131.1; 130.9; 129.8; 126.9; 122.9; 108.2; 49.4; 47.4; 31.4; 29.8; 28.5; 25.9; 24.5; 22.4; 14.3; 9.1. HRMS calcd. for C₄₈H₆₈N₆O₄ 410.2802, found 410.2813.

Synthesis of *N,N'*-bis-[1-propyl-3-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluorooctyl)imidazolium]-naphthalene-diimidediiodide [C₈FNDI][I].

N,N'-bis-(3-imidazol-yl-propyl)naphthalene-diimide (0.100 g; 2.075·10⁻⁴mol) was dissolved in 5 mL of anhydrous DMF, and the solution obtained was heated at 80 °C. Into a two-neck flask, 1,1,2,2-tetrahydroperfluorooctylidide (0.216 g; 4.56·10⁻⁴mol) was dissolved in 5 mL of anhydrous DMF; the solution was heated at 80 °C. The solution of *N,N'*-bis-(3-imidazol-yl-propyl)naphthalene-diimide was added to the solution of alkyl iodide. The mixture was stirred, under argon atmosphere at 80 °C for 4 days.

The solvent was removed in vacuo and the dark solid obtained was washed with diethyl ether (20 mL), with ultrasounds irradiation until a colourless organic phase was obtained. Finally, the dark solid was washed with refluxing ethyl acetate (50 mL) overnight.

Yield: 81%; dark solid; m. p.: 179.2 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.30 (s, 2H); 8.86-8.71 (m, 6H); 7.91-7.54 (m, 8H); 4.59 (m, 2H); 4.37 (m, 5H); 4.15 (m, 5H); 2.27 (m, 8H).

HRMS calcd. for C₄₂H₃₀F₂₆N₆O₄ 588.0951, found 588.0945.

Synthesis of *N,N'*-bis-[1-(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,12-icosafuorododecyl)-3-propylimidazolium]-naphthalene-diimidediiodide [C₁₂FNDI][I].

N,N'-bis-(3-imidazol-yl-propyl)naphthalene-diimide (0.100 g; 2.075·10⁻⁴mol) was dissolved in 5 mL of anhydrous DMF and the solution obtained was heated at 80 °C. Into a two-neck flask, 1,1,2,2-tetrahydroperfluorododecylidide (0.31 g; 4.56·10⁻⁴mol) was dissolved in 5 mL of anhydrous DMF; the solution was heated at 120 °C.

The solution of *N,N'*-bis-(3-imidazol-yl-propyl)naphthalene-diimide was added to the solution of alkyl iodide. The mixture obtained was stirred, under argon atmosphere, at 120 °C for 4 days.

The solvent was removed in vacuo and the dark solid obtained was washed with diethyl ether (20 mL) with ultrasounds irradiation until a colourless organic phase was obtained. Finally, the dark solid was washed with refluxing ethyl acetate (50 mL) overnight.

Yield: 50%; dark solid; m. p.: 182.2 °C. ¹H-NMR (400 MHz, DMSO-d₆); δ (ppm): 9.30 (s, 2H); 8.87-8.70 (m, 6H); 7.92-7.56 (m, 8H); 4.58 (m, 2H); 4.30 (m, 5H); 4.11 (m, 5H); 2.25 (m, 8H).

HRMS calcd. for C₅₀H₃₀F₄₂N₆O₄ 788.0902, found 788.0823.

Determination of thermodynamic parameters from temperature dependent measurements

To determine the thermodynamic parameters we used the procedure reported by Meijer et al.^[1]

To this aim, we first calculated the number averaged degree of polymerization at each temperature

$DP_n(T)$ by means of equation (1).

$$DP_n(T) = \frac{1}{\sqrt{1 - \alpha_{agg}(T)}} \quad (1)$$

At a given temperature DP_n is related to the association constant K_{ass} by equation (2), rearrangement of which yields K_{ass} as expressed by equation (3), where C_T is the total concentration of salt.

$$DP_n(T) = \frac{1}{2} + \frac{1}{2} \sqrt{4 \cdot K_{ass}(T) \cdot C_T + 1} \quad (2)$$

$$K_{ass} = \frac{[(2 \cdot DP_n(T) - 1)^2 - 1]}{4 \cdot C_T} \quad (3)$$

- [1] M. M. J. Smulders, M. M. L. Nieuwenhuizen, T. F. A. de Greef, P. van der Schoot, A. P. H. J. Schenning, E. W. Meijer, *Chem. Eur. J.* **2010**, *16*, 362-367.

Table S1. Melting temperatures, enthalpy and entropy values determined by DSC measurements.

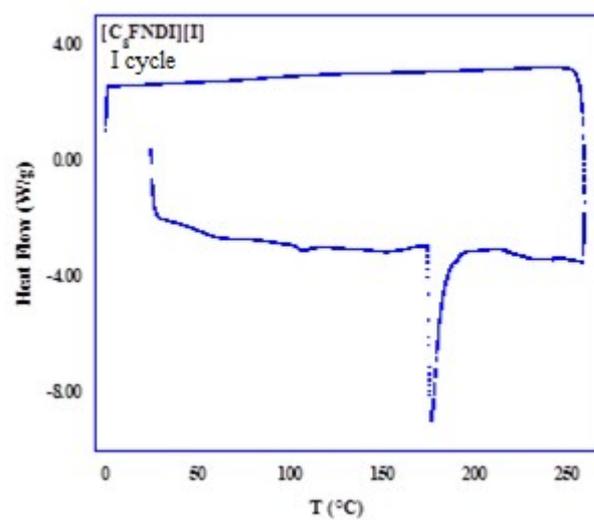
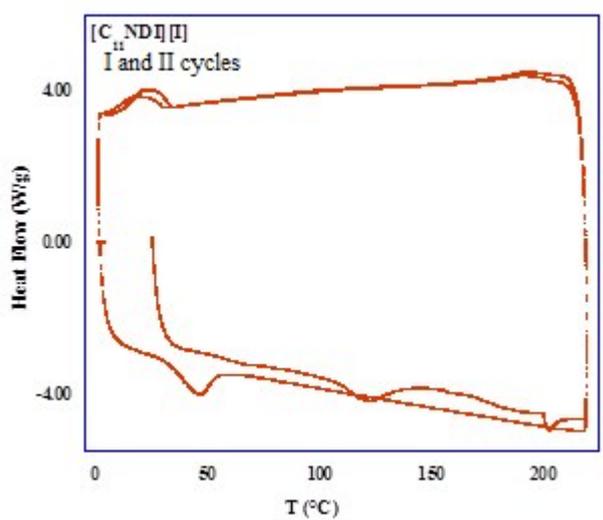
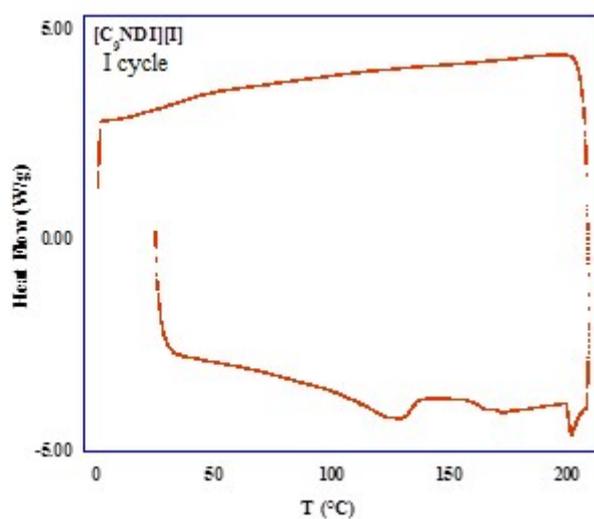
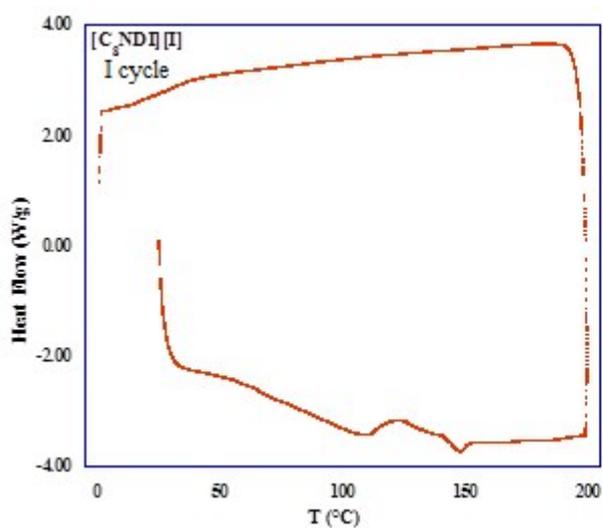
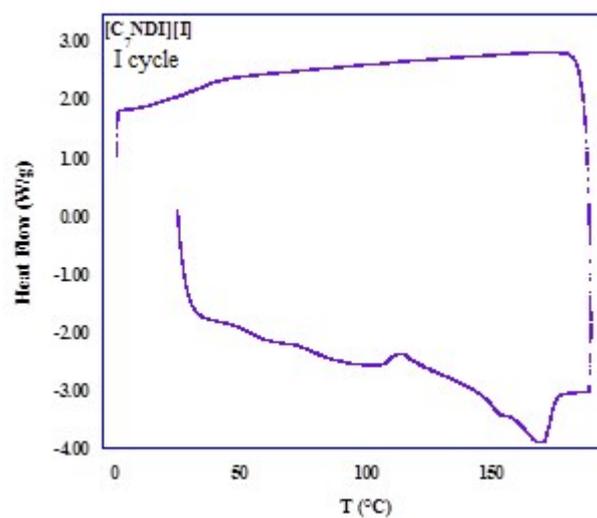
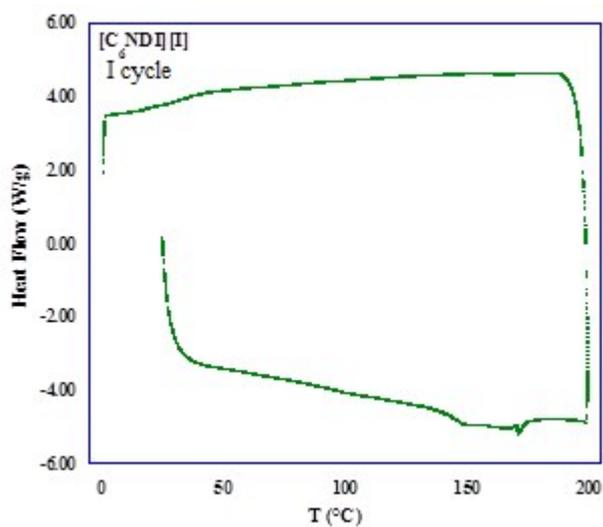
Salt	T_m (°C)	ΔH_m (J/mol)	ΔS_m (J/K mol)
[C ₆ NDI][I]	170.8	316	1.0
[C ₇ NDI][I]	170.1	4000	9.0
[C ₈ NDI][I]	146.9	792	2.0
[C ₉ NDI][I]	200.2	1344	3.0
[C ₁₀ NDI][I]	-	-	-
[C ₁₁ NDI][I]	202.6	1420	3.0
[C ₁₂ NDI][I]	144.0	4600	11.0
[C ₈ FNDI][I]	179.2	3440	8.0
[C ₁₂ FNDI][I]	182.2	4920	11.0

Table S2. Salts concentration corresponding to the onset of aggregation (C_{on}) for [C_nNDI][I] salts as a function of solvent.

Salt	$10^6 \cdot C_{on,THF}$ [M]	$10^6 \cdot C_{on,DMF}$ [M]	$10^6 \cdot C_{on,CHCl_3}$ [M]	$10^6 \cdot C_{on,1,4-Diox}$ [M]
[C ₆ NDI][I]	2.6	2.3		
[C ₇ NDI][I]	5.5	1.8		
[C ₈ NDI][I]	3.1	1.1		
[C ₉ NDI][I]	2.4	2.0		
[C ₁₀ NDI][I]	3.3	1.9		
[C ₁₁ NDI][I]	1.9	1.3		
[C ₁₂ NDI][I]	1.0	2.4	3.0	2.6
[C ₈ FNDI][I]		2.7		
[C ₁₂ FNDI][I]		3.3		

Table S3. Position of main emission band (λ_{max}) in solution and in solid state as a function of alkyl chain and solvent.

Salt	λ_{max} (nm)							
	THF		DMF		CHCl ₃		1,4-dioxane	
	Solid state	Solution	Solid state	Solution	Solid state	Solution	Solid state	Solution
[C ₆ NDI][I]	376	410						
[C ₇ NDI][I]	375	408						
[C ₈ NDI][I]	375	408						
[C ₉ NDI][I]	375	393						
[C ₁₀ NDI][I]	375	408						
[C ₁₁ NDI][I]	375	408						
[C ₁₂ NDI][I]	375	410	375	419	375	410	376	420



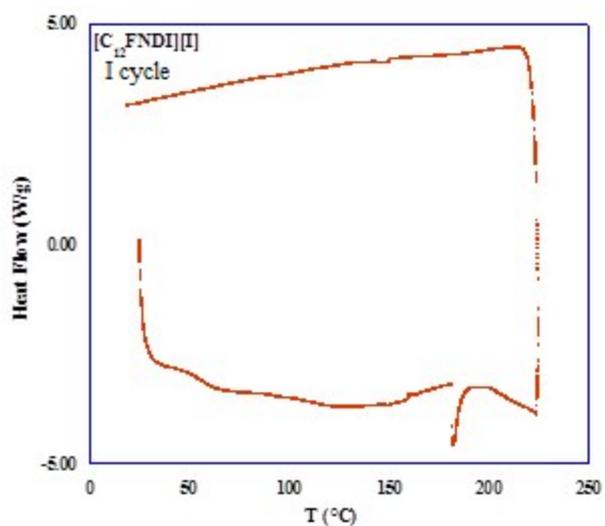
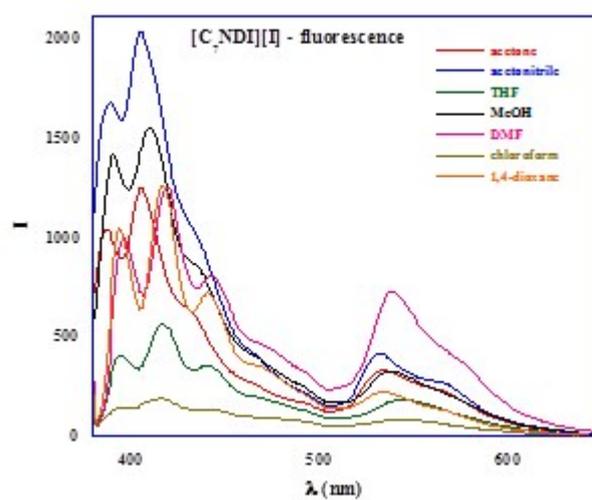
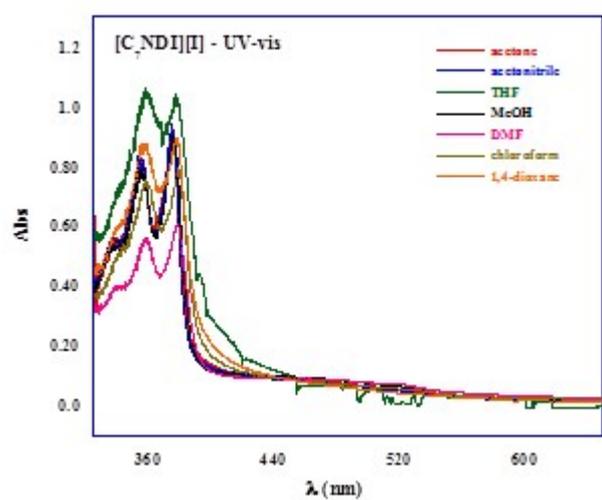
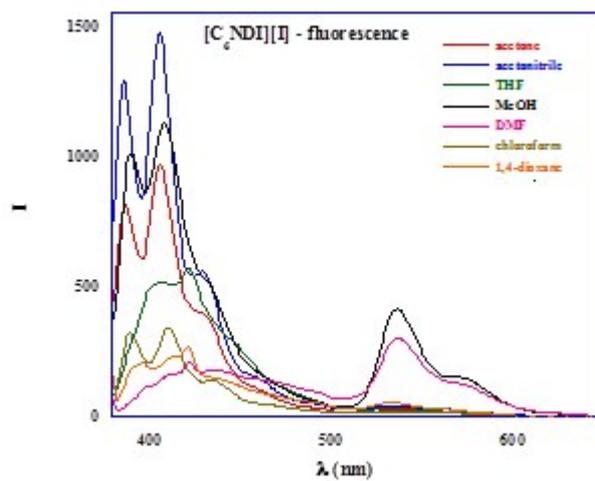
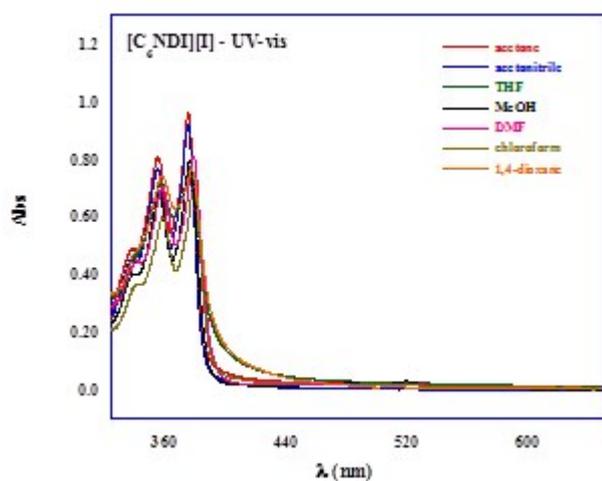
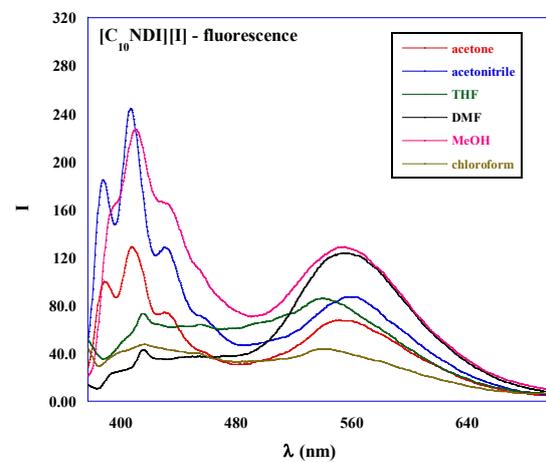
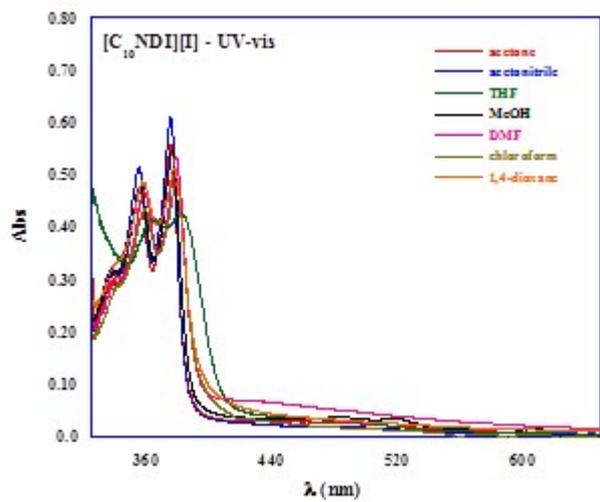
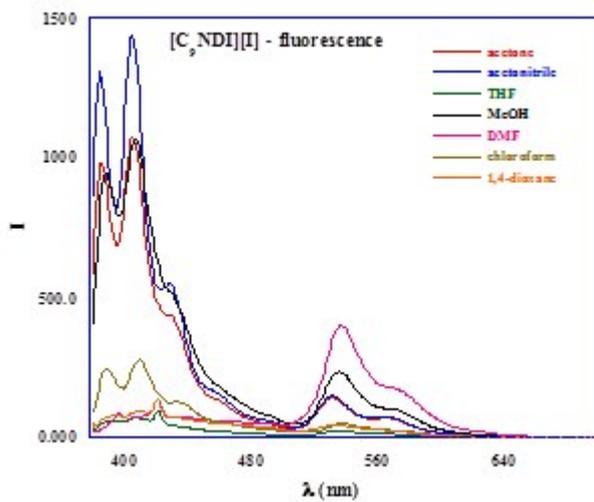
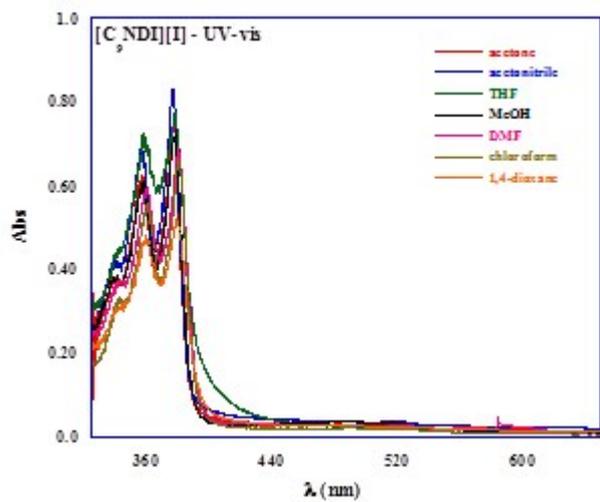
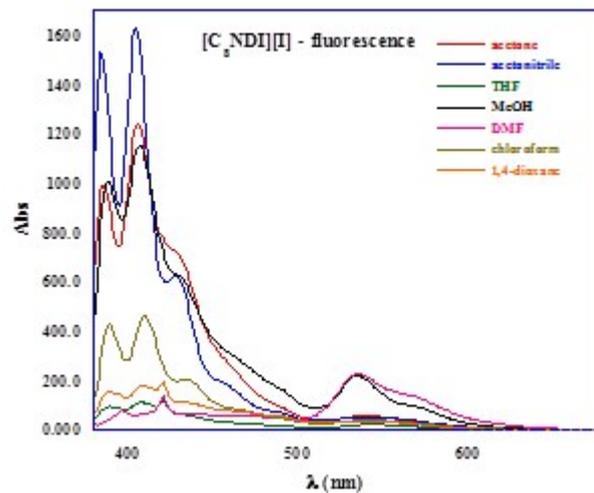
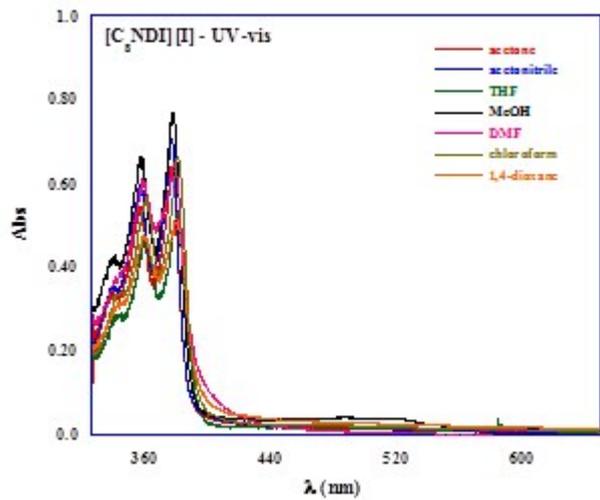


Figure S1. DSC thermograms of the synthesized salts.





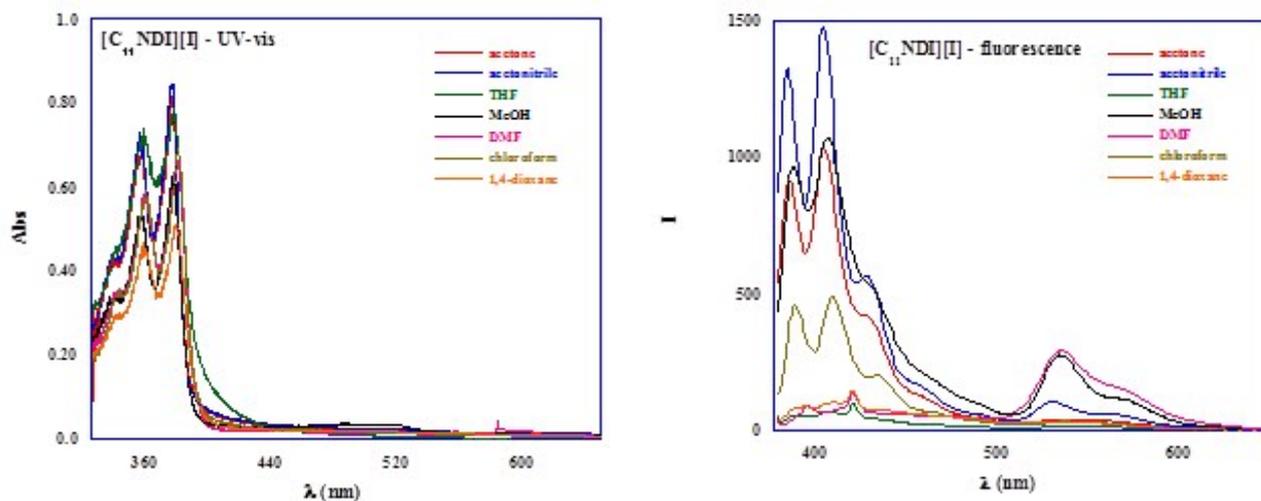
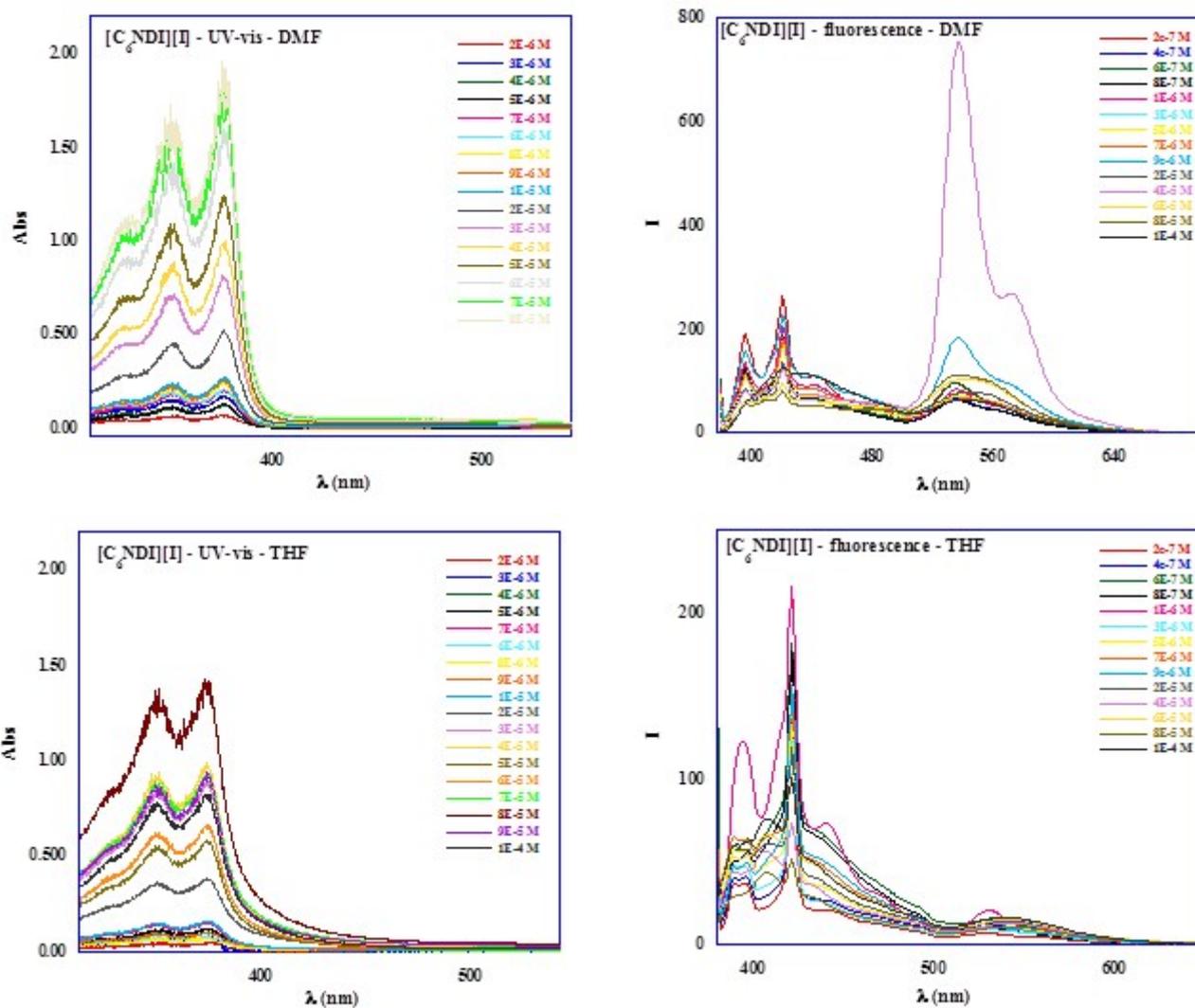
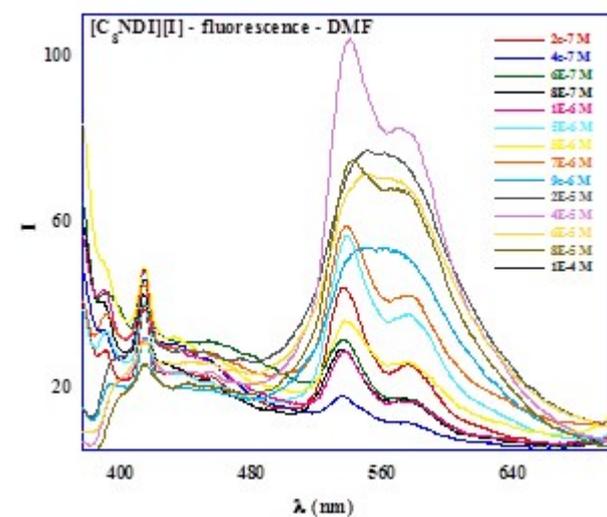
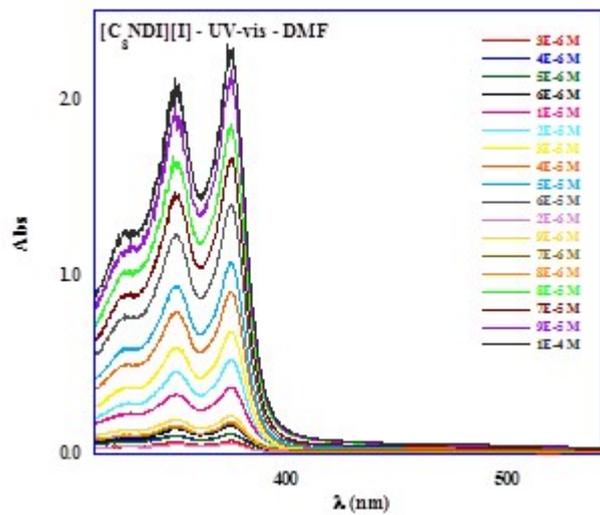
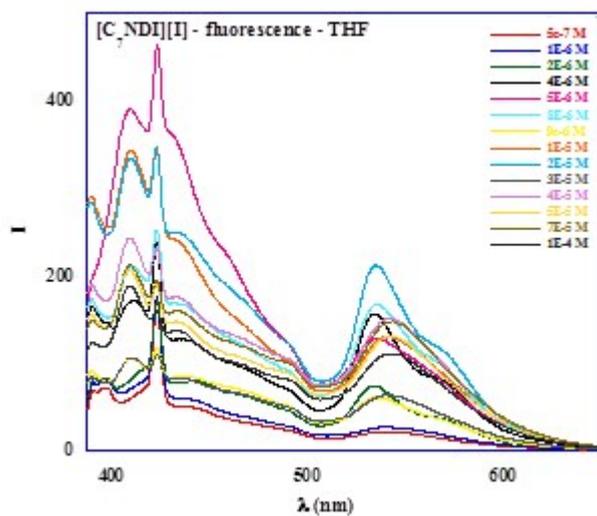
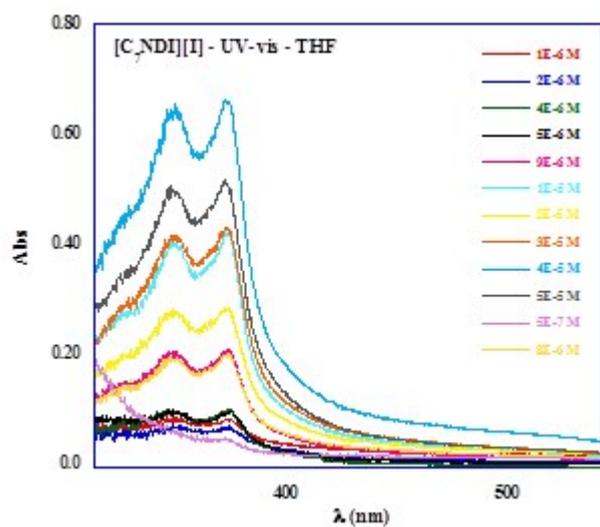
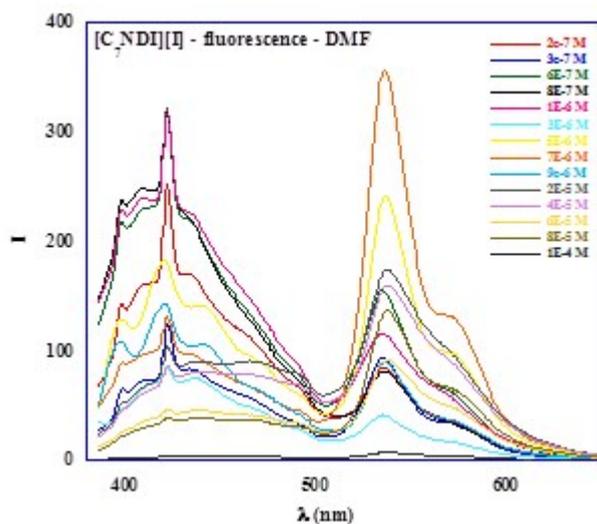
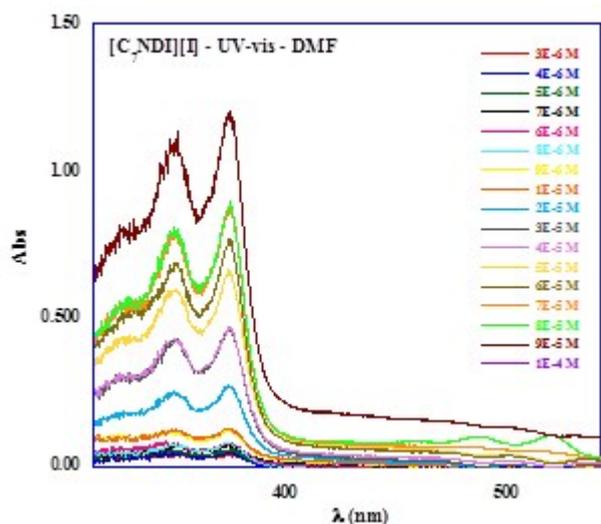
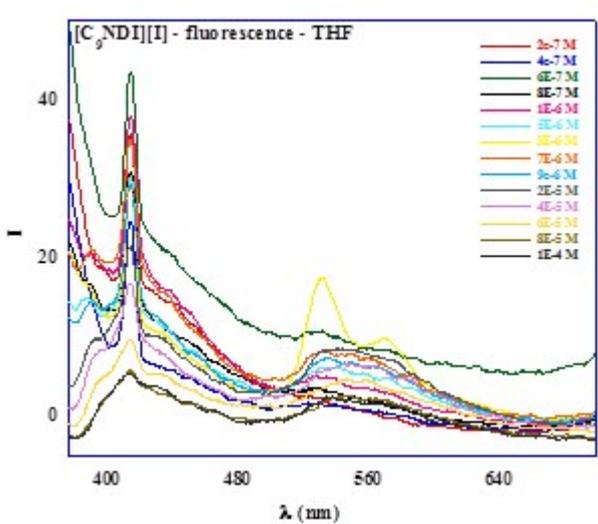
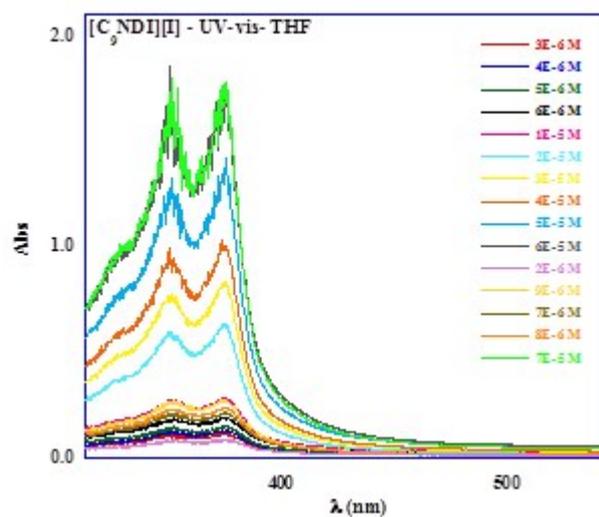
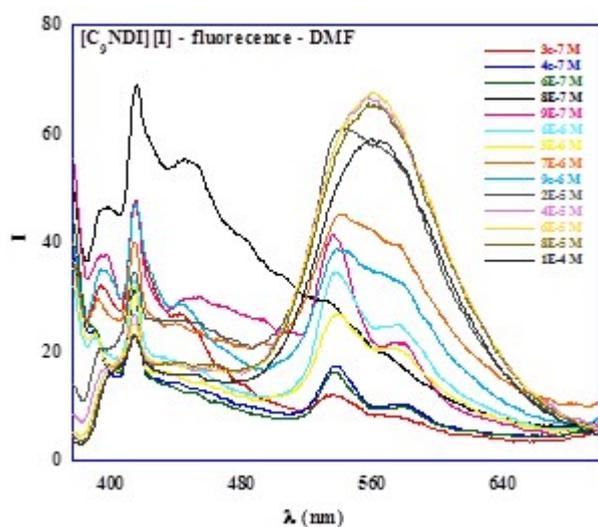
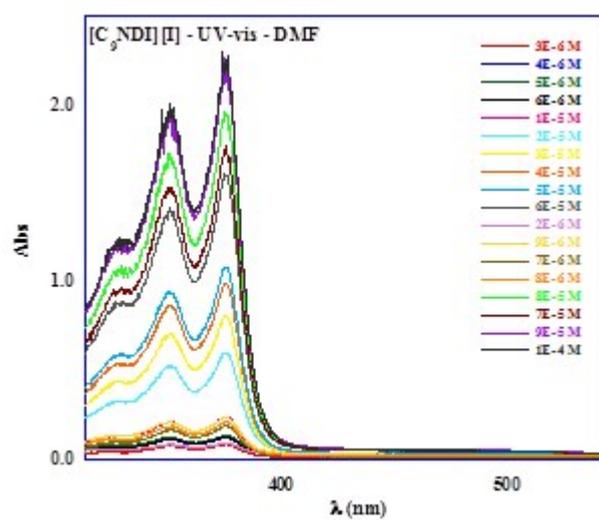
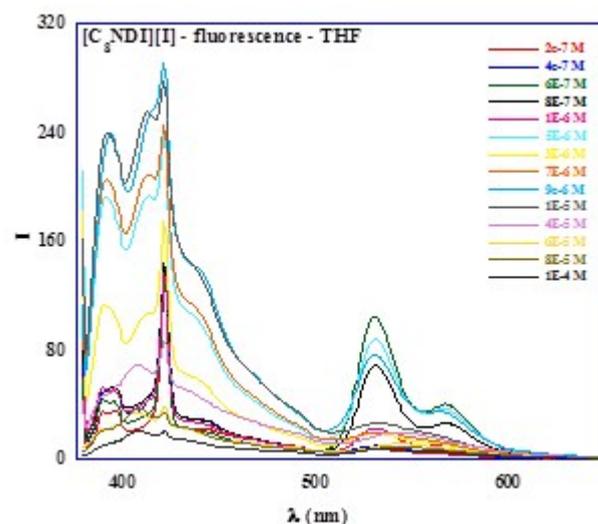
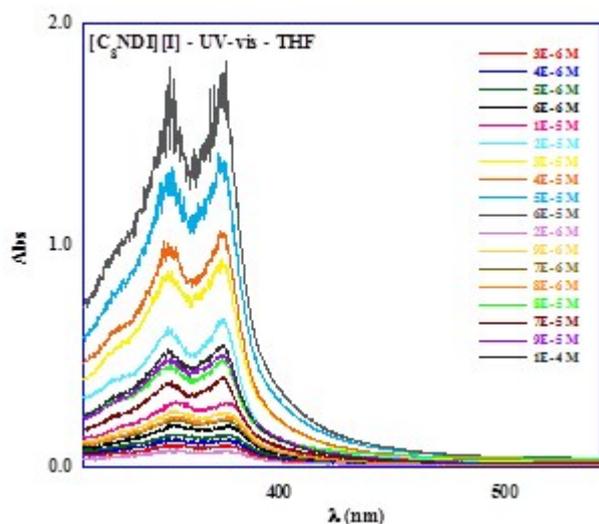
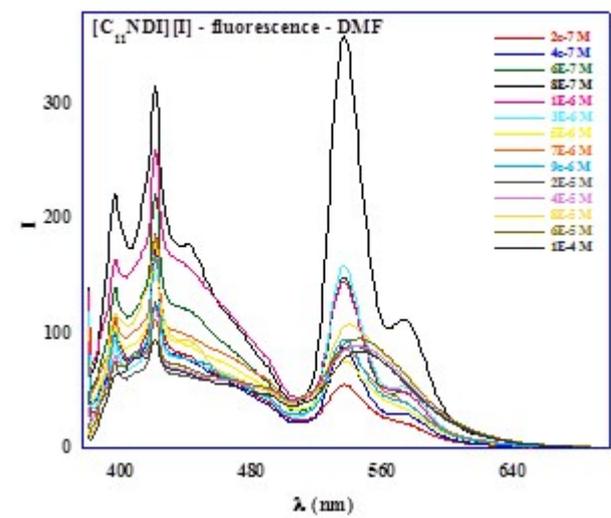
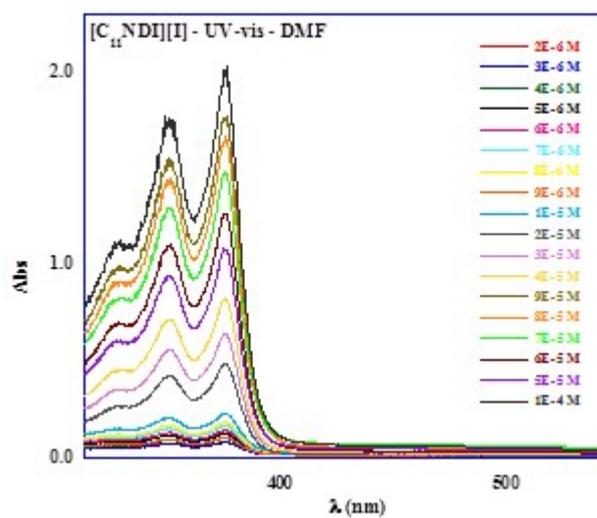
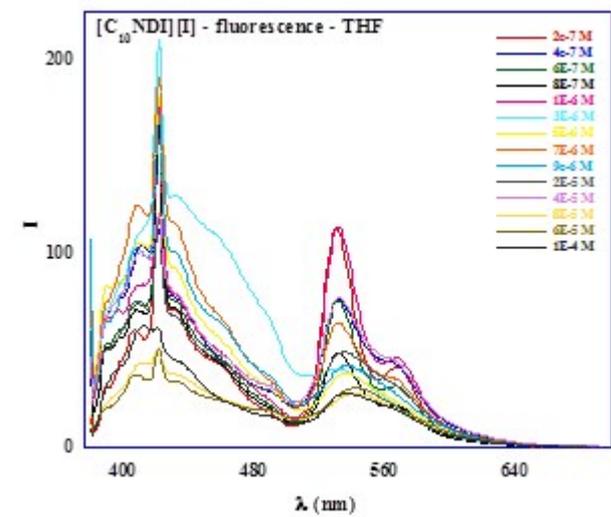
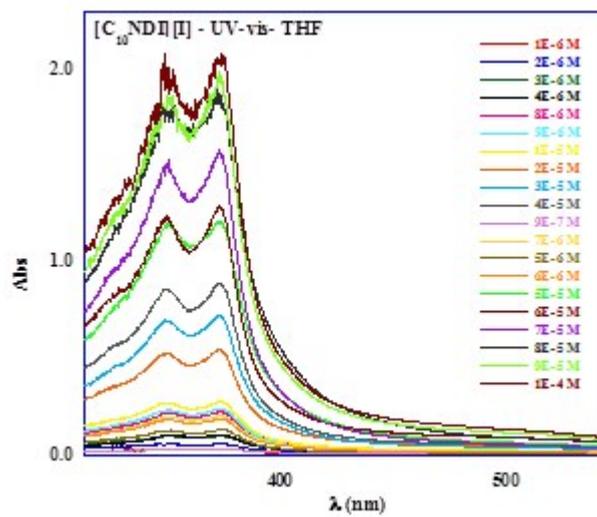
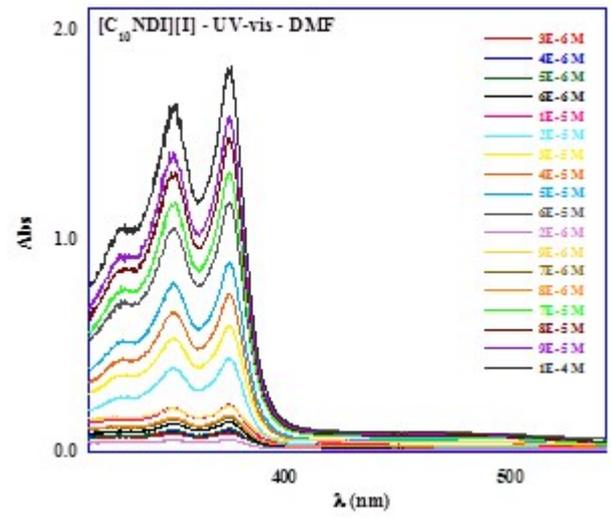
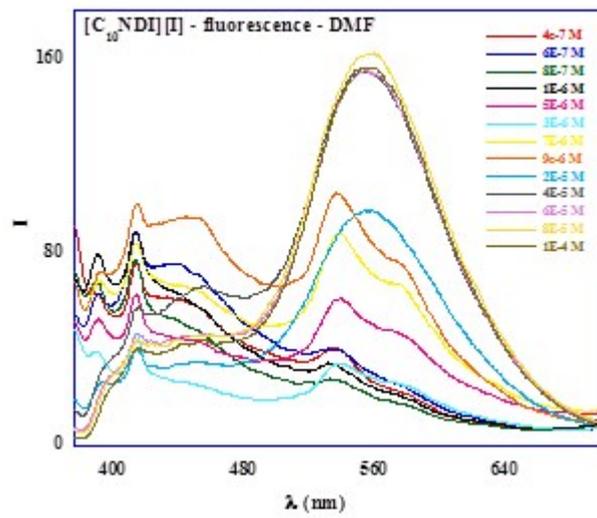


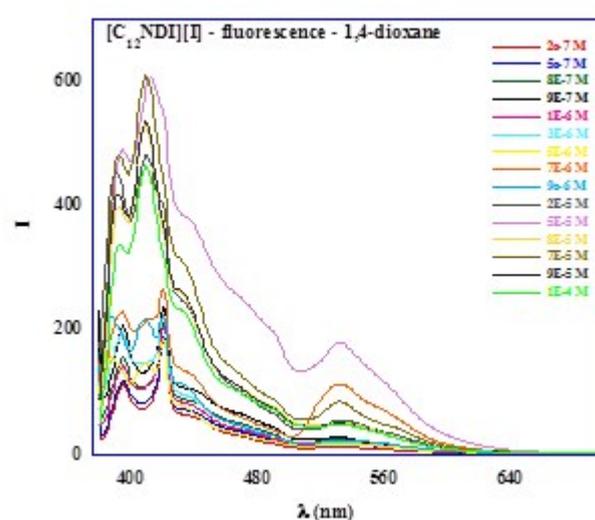
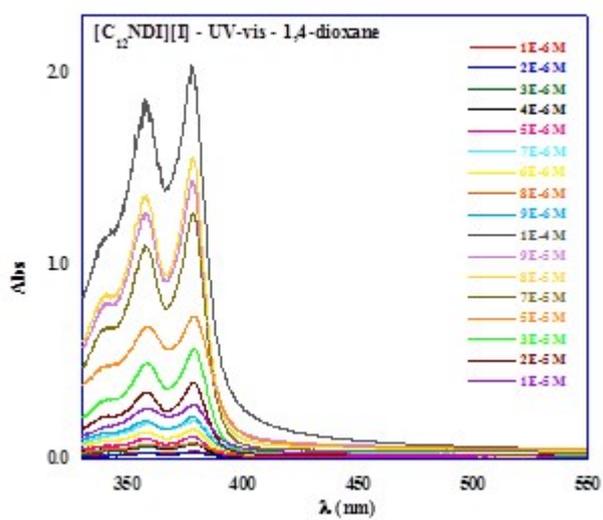
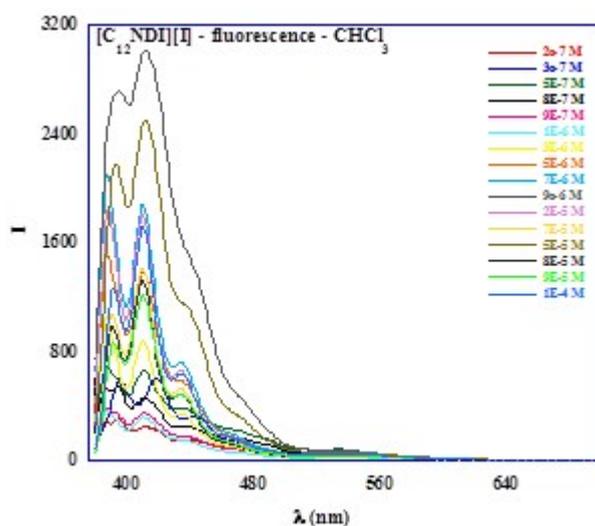
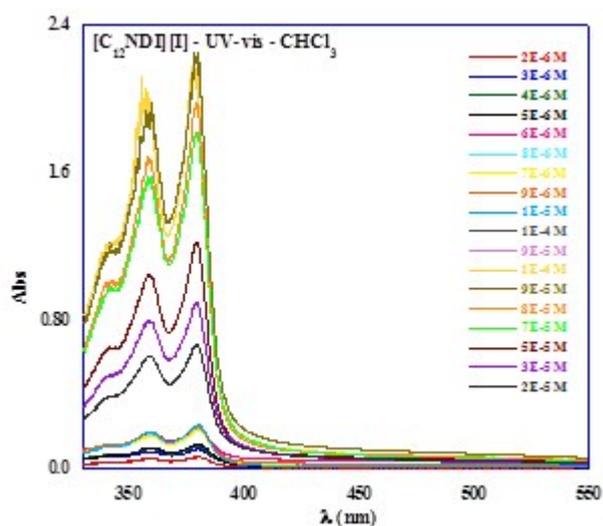
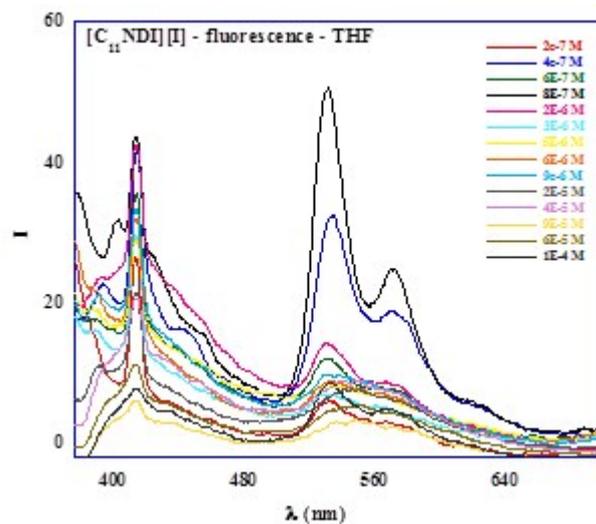
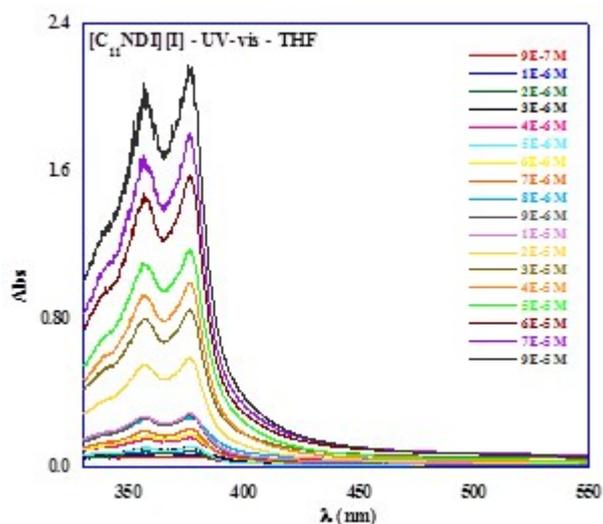
Figure S2. UV-vis and fluorescence spectra of salts as function of solvent. Fluorescence intensities are in arbitrary units.











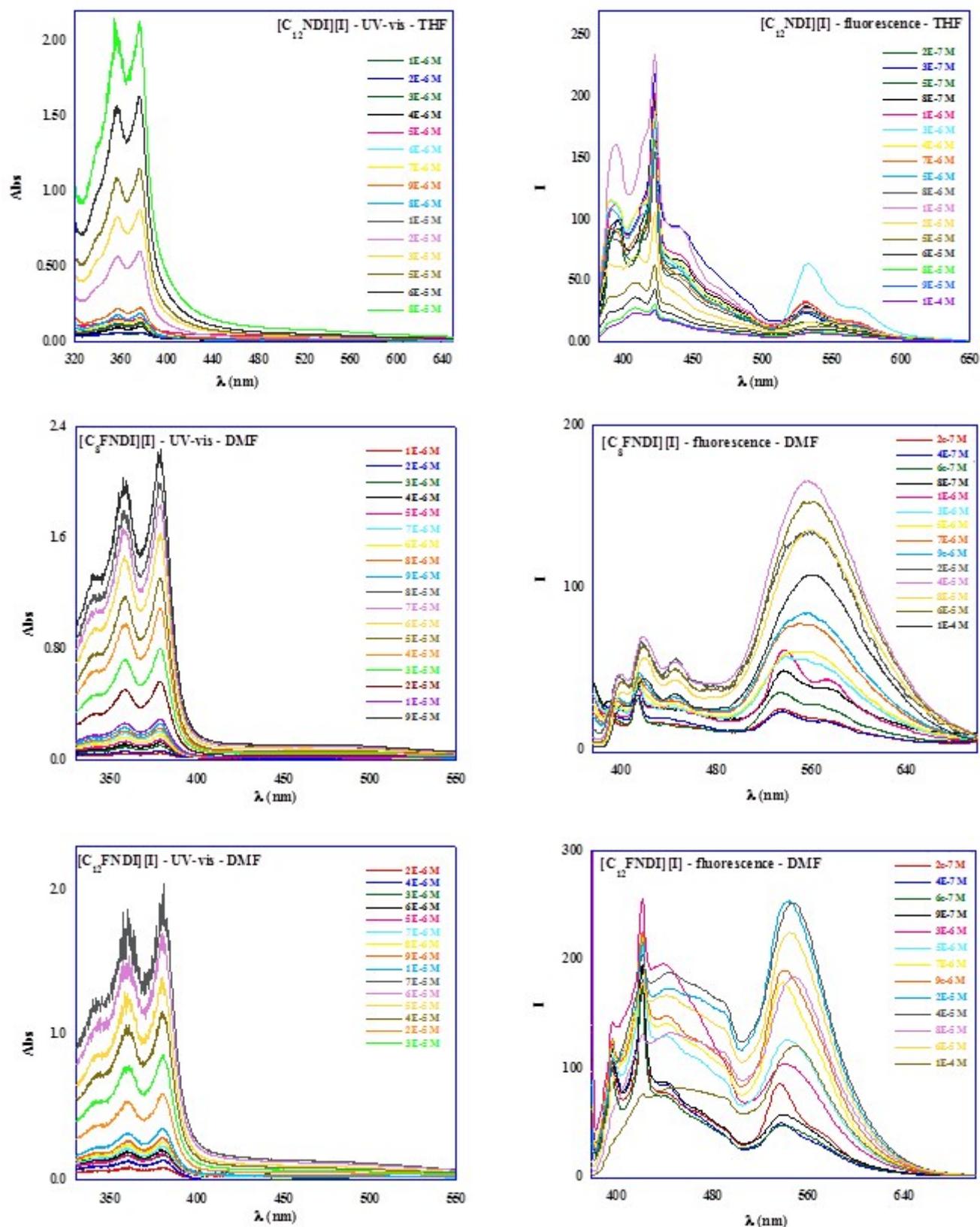
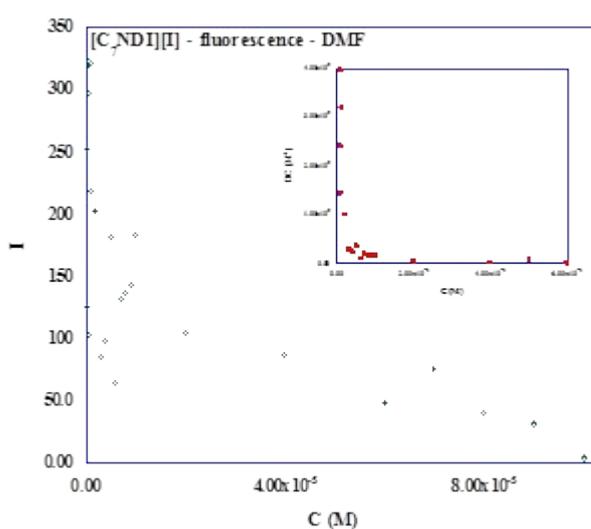
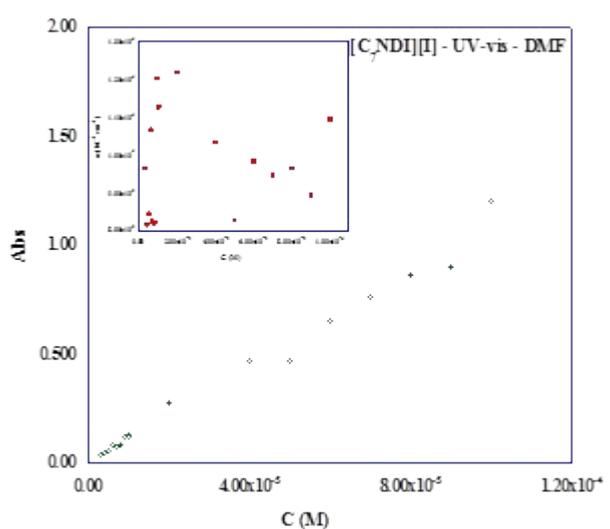
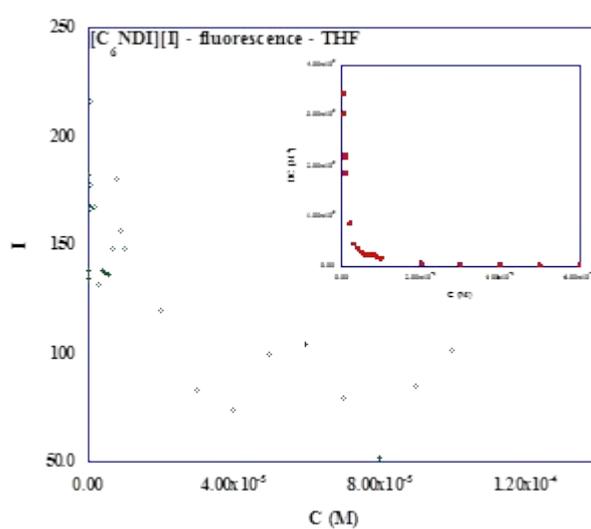
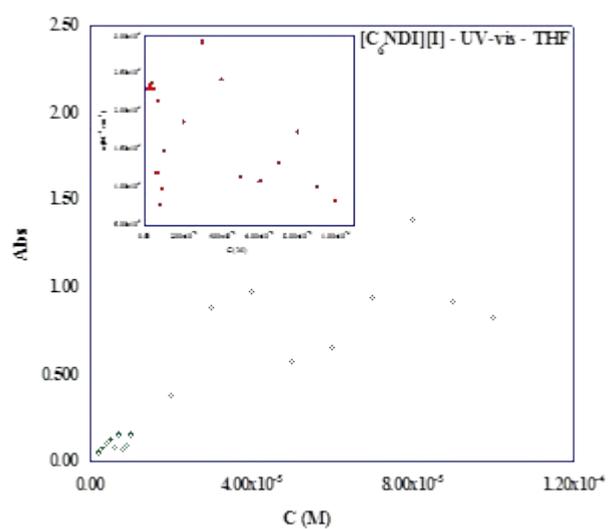
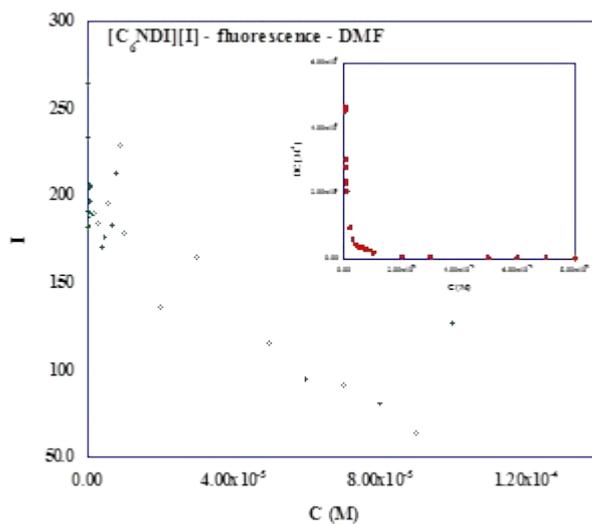
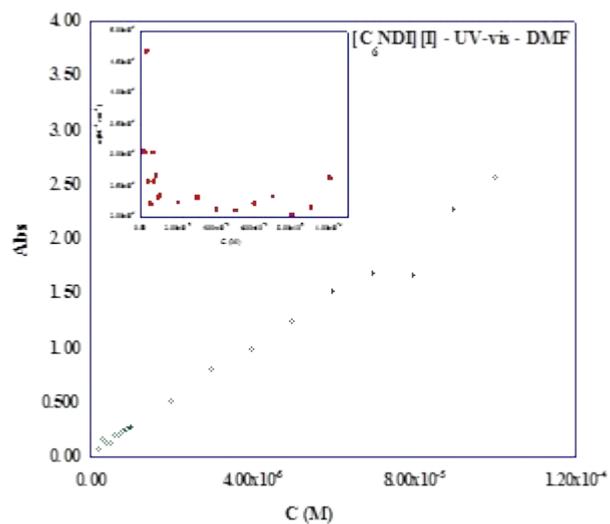
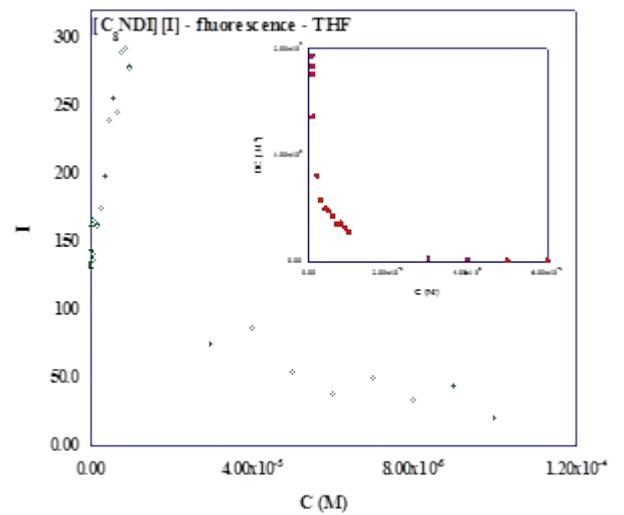
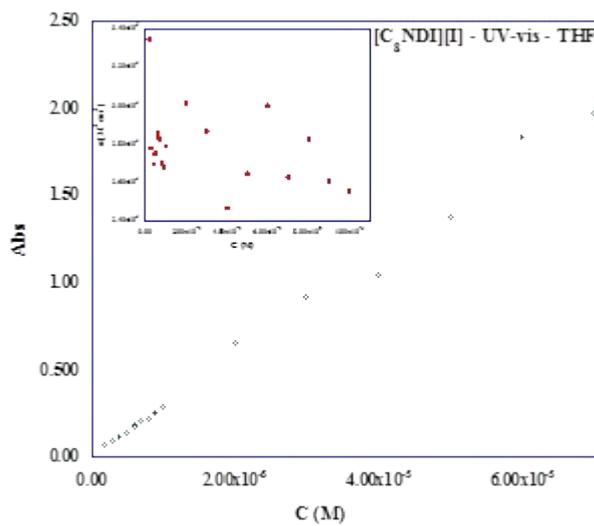
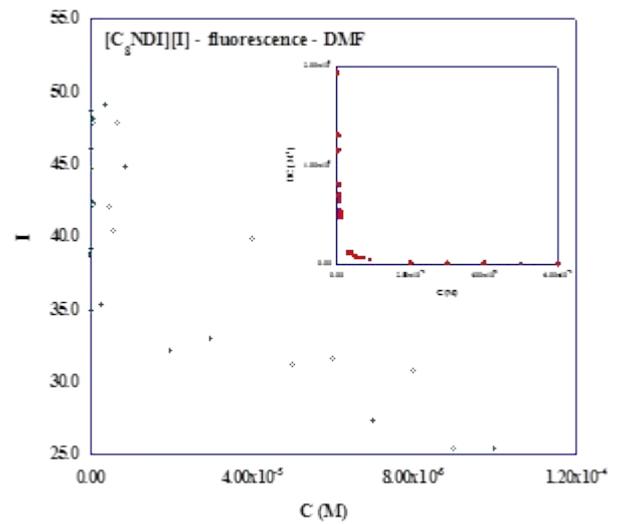
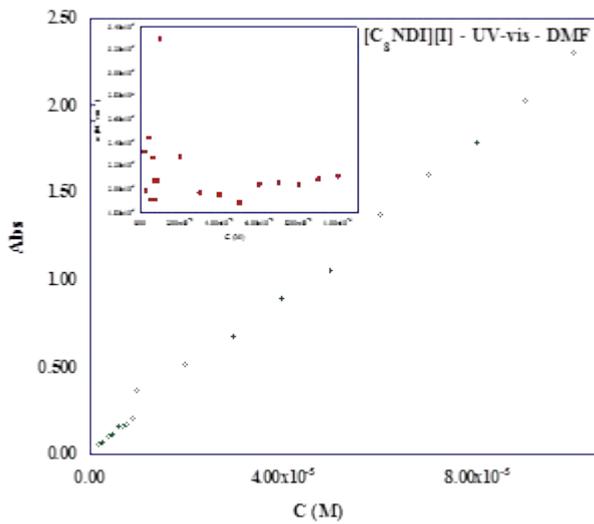
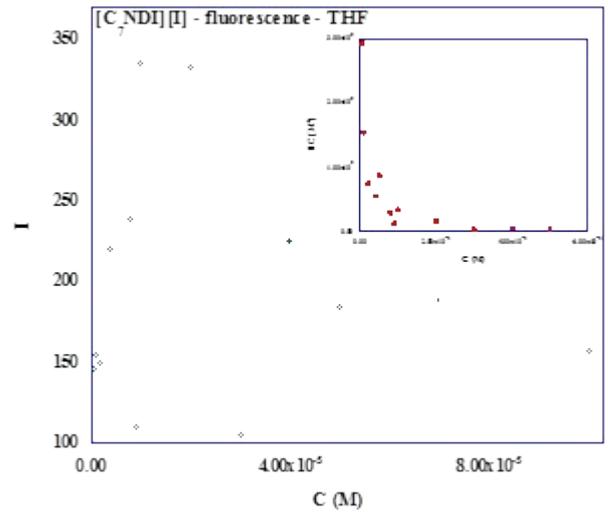
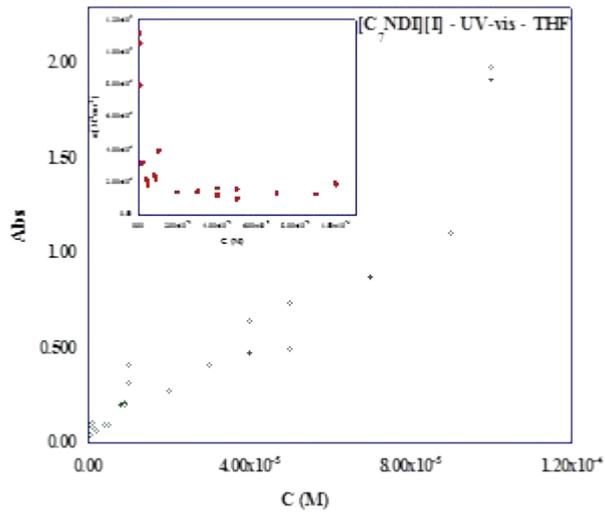
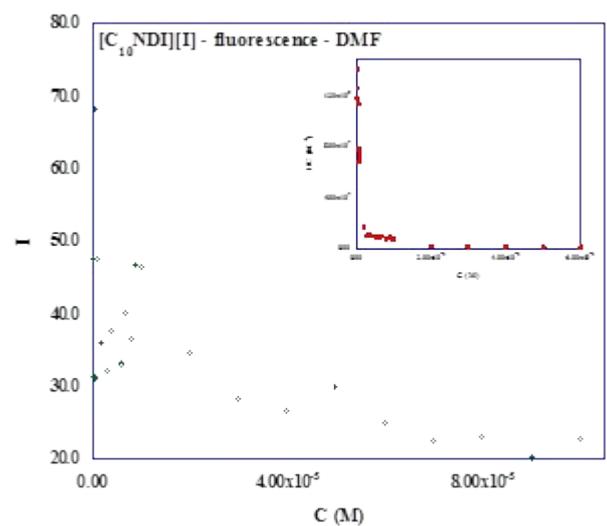
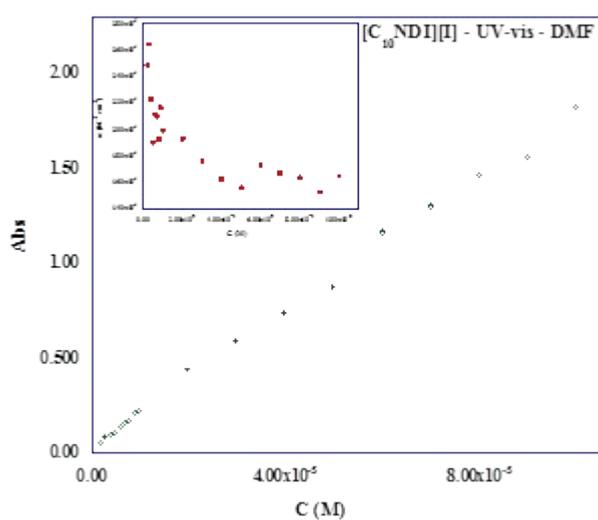
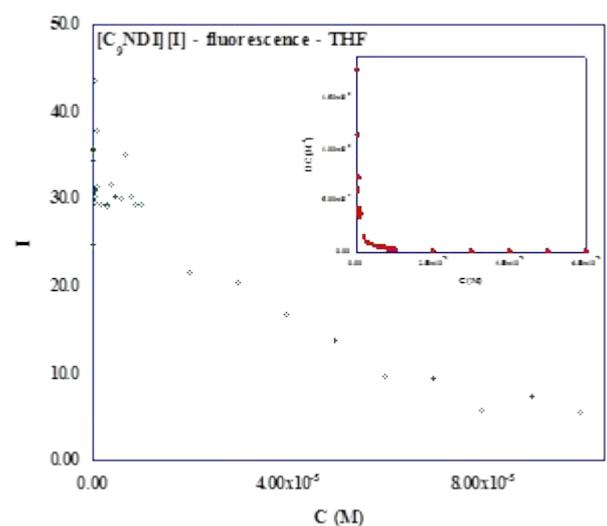
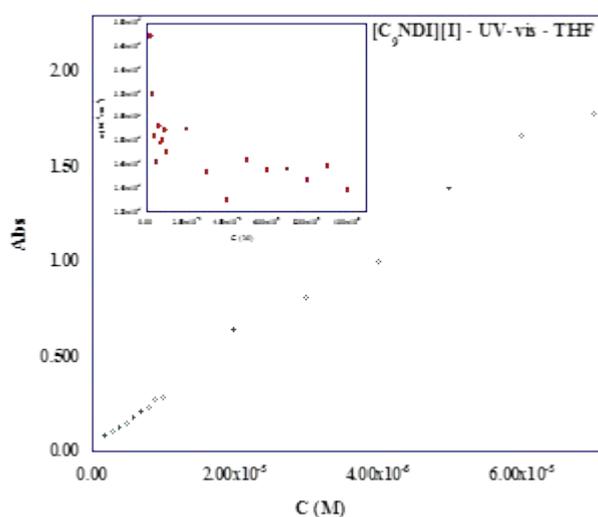
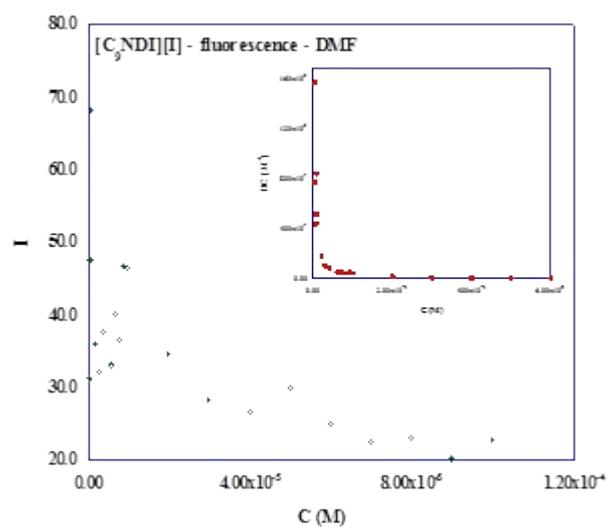
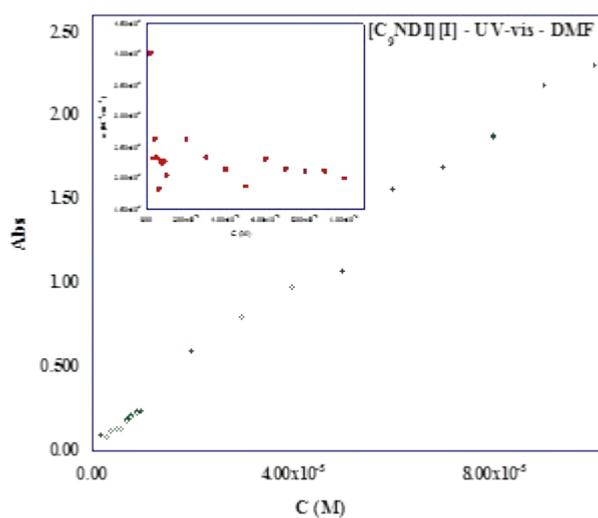
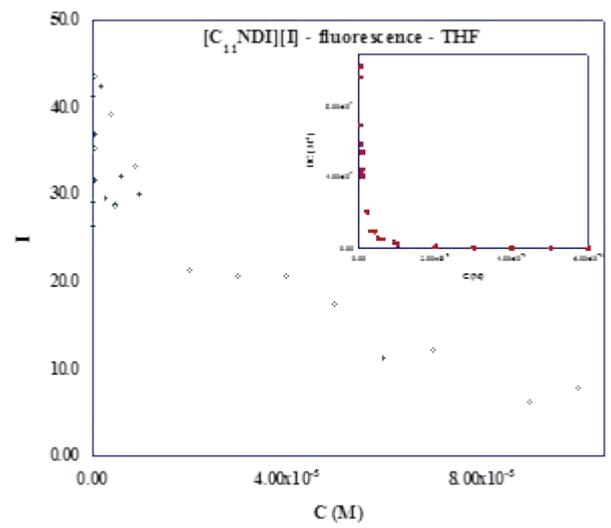
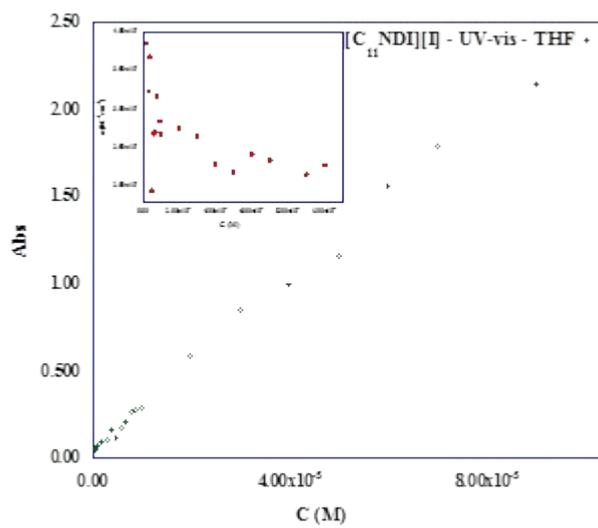
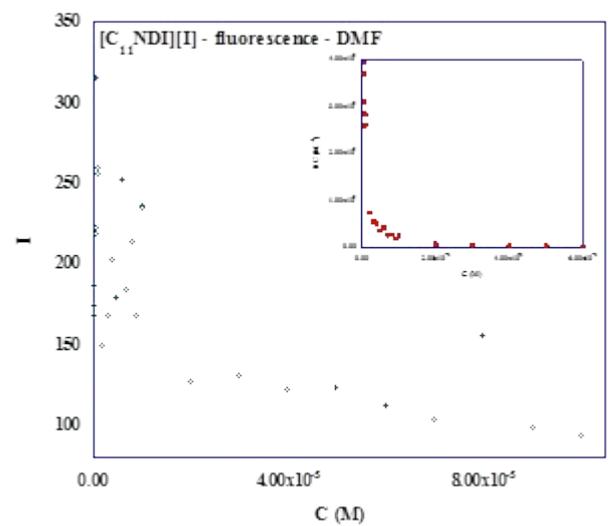
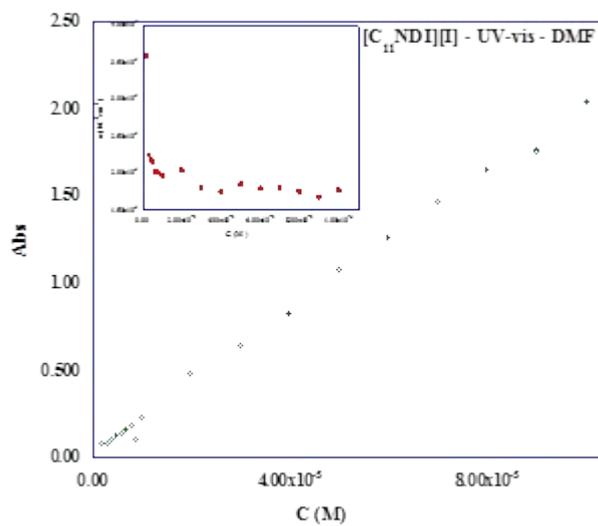
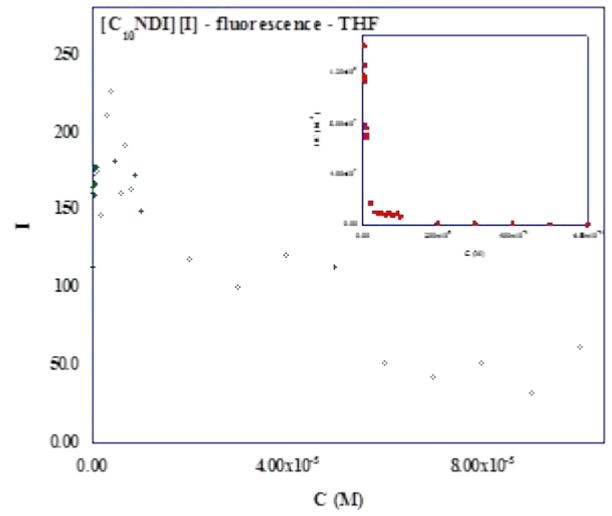
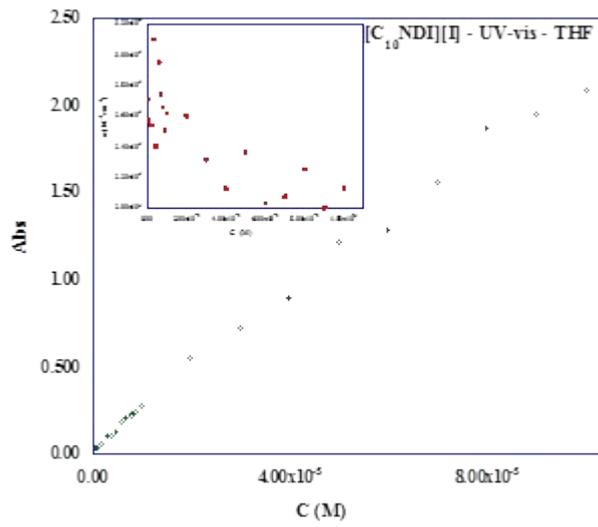


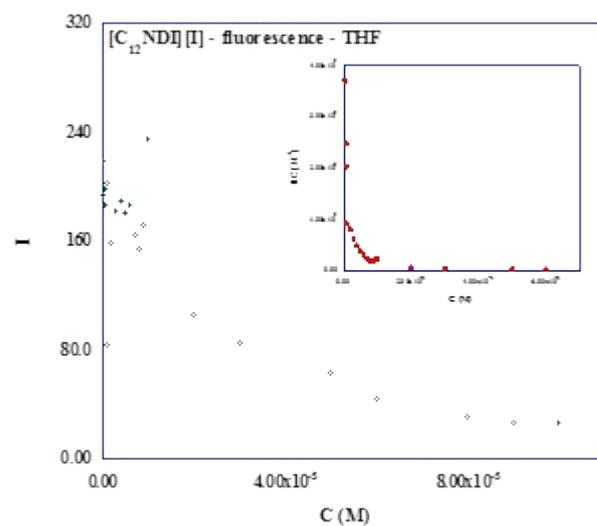
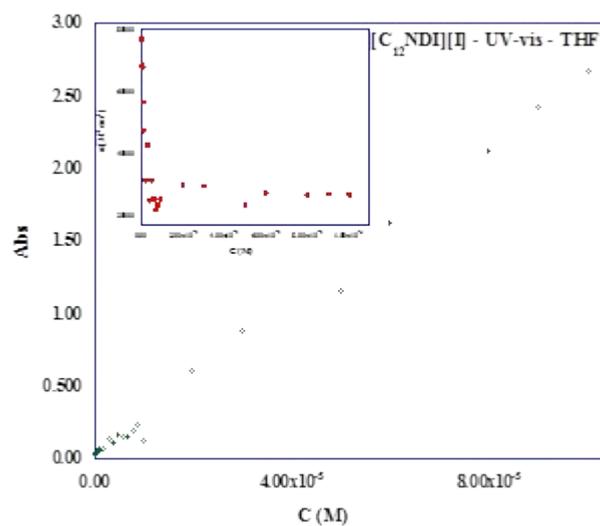
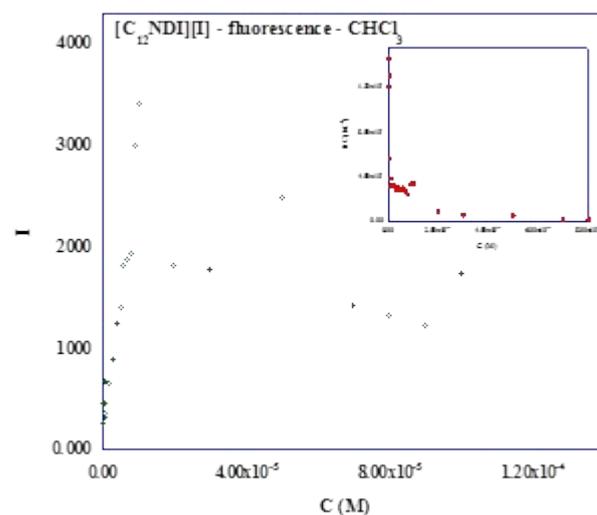
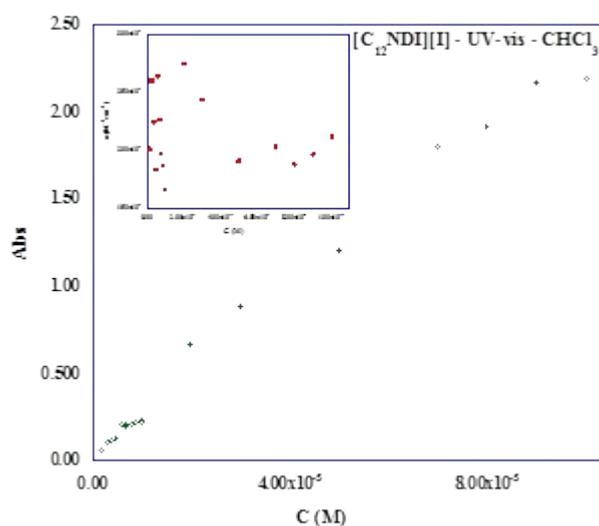
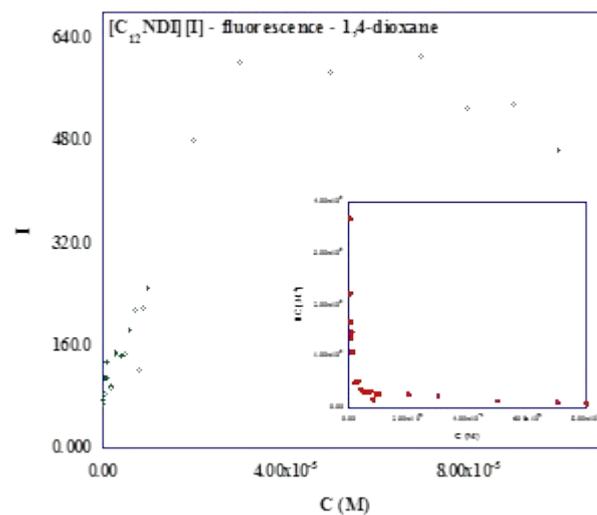
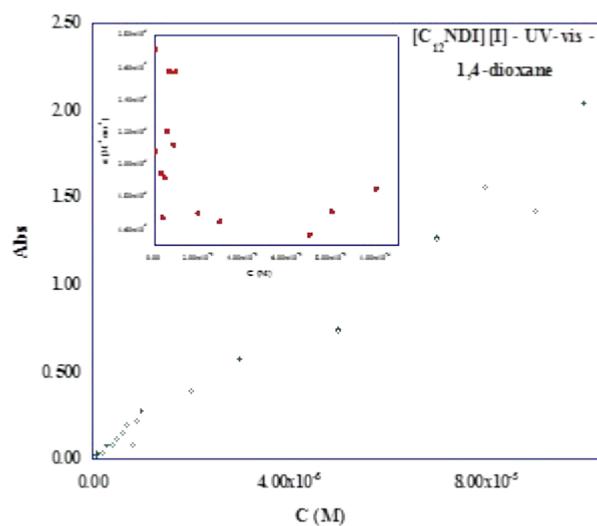
Figure S3. UV-vis and fluorescence ($\lambda_{\text{ex}} = 362$ nm) of salts as a function of concentration and solvent. Fluorescence intensities are in arbitrary units.











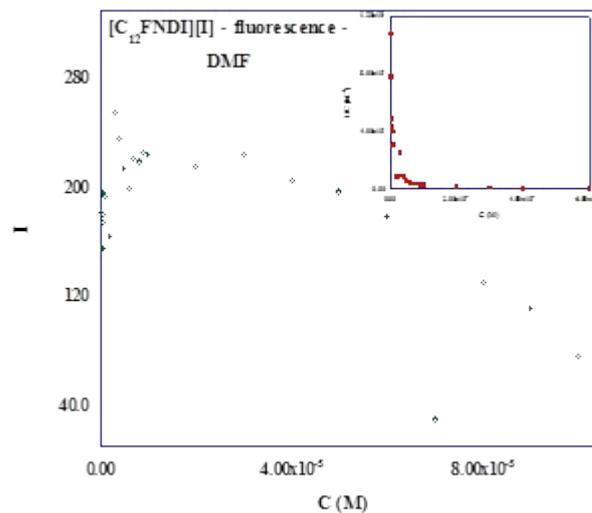
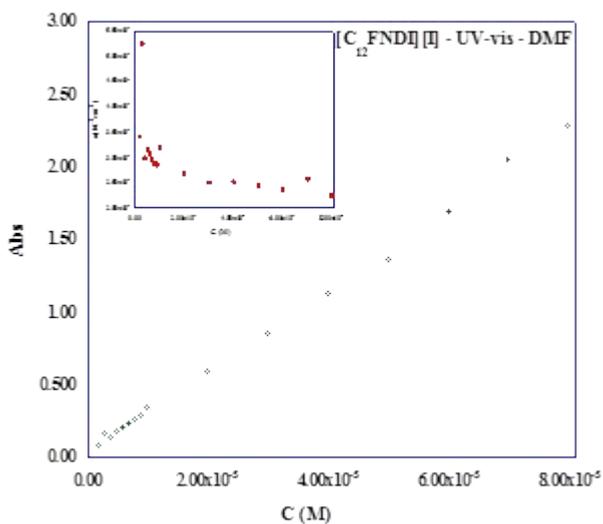
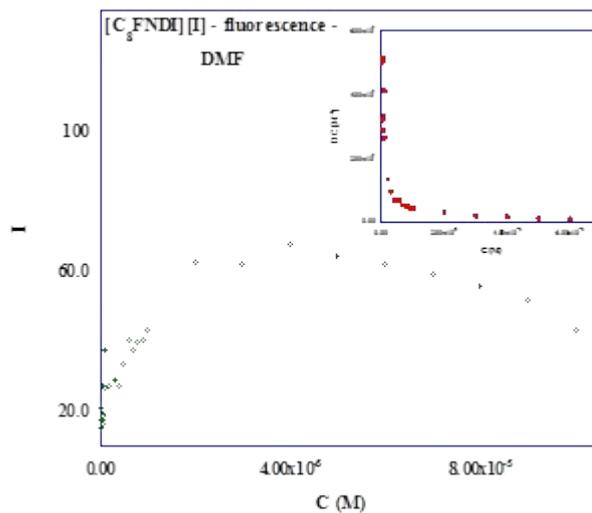
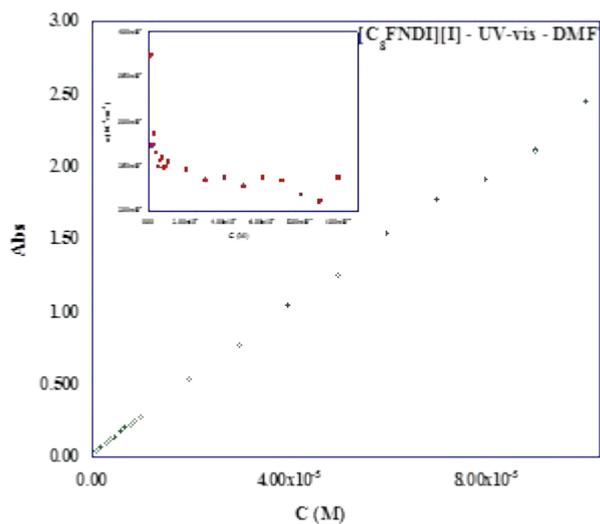
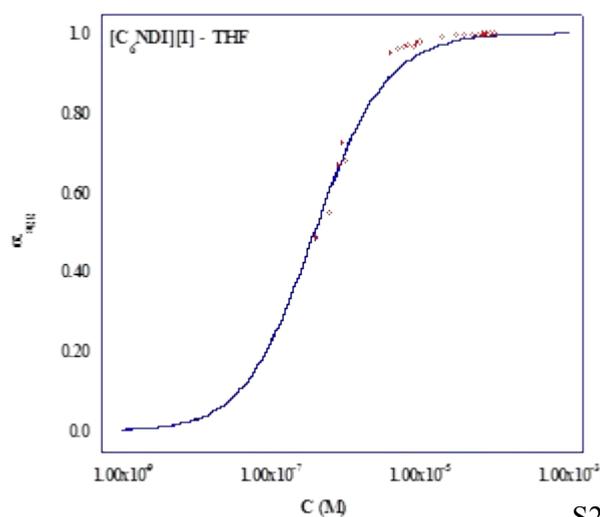
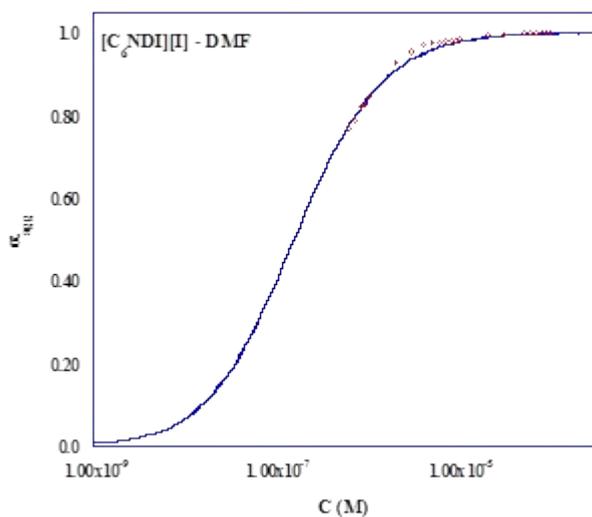
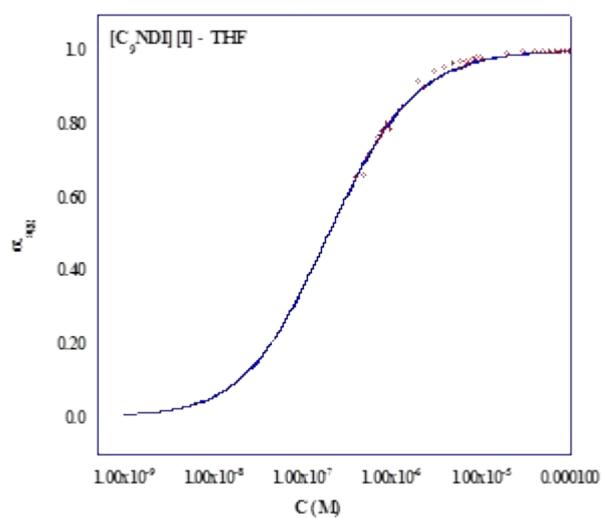
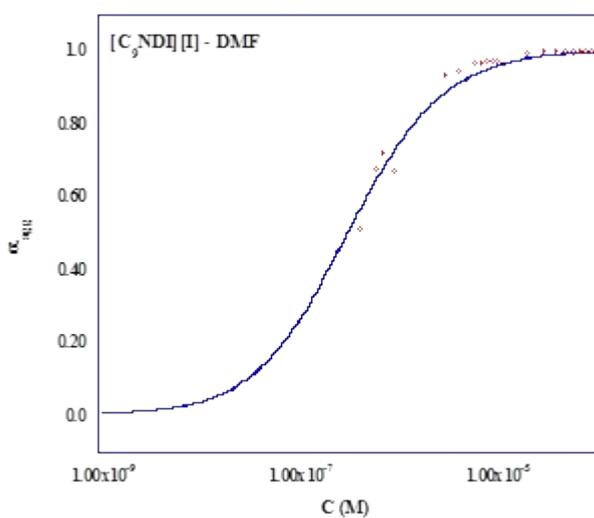
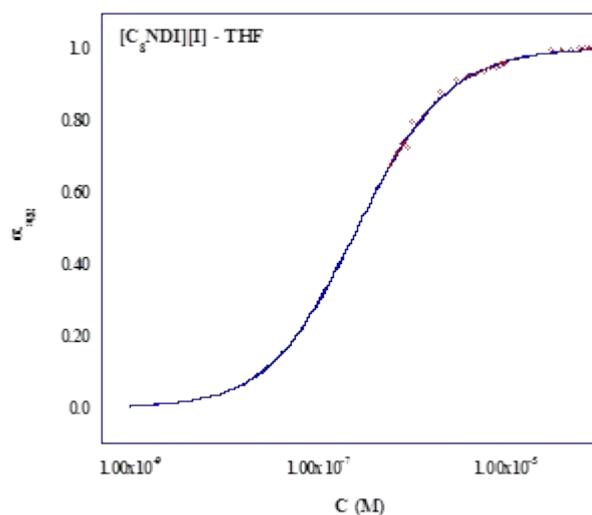
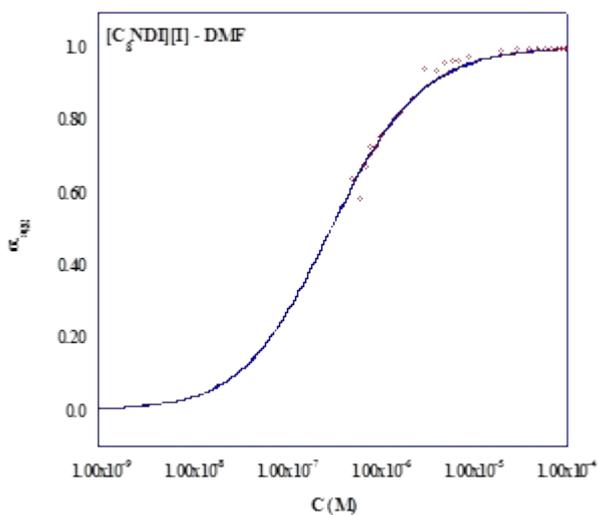
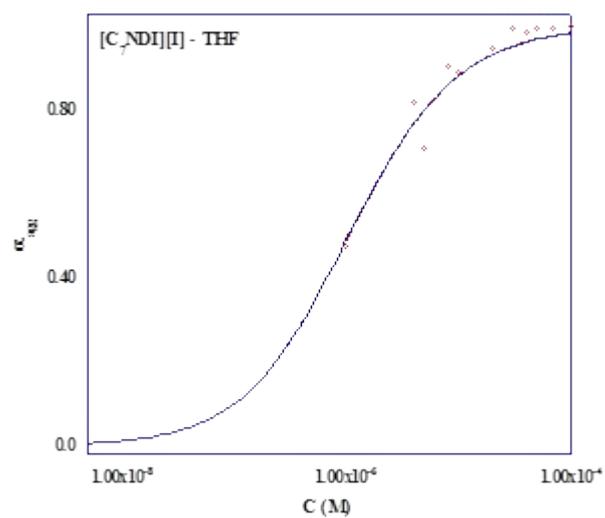
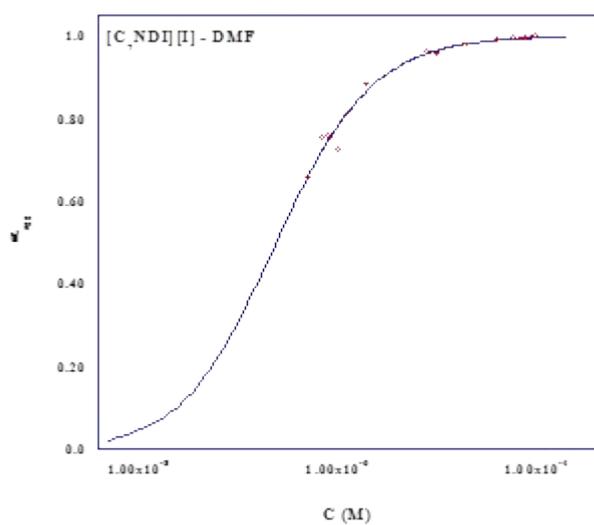
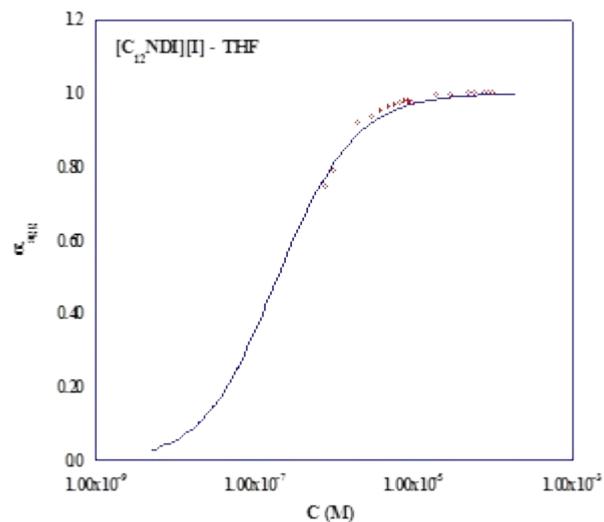
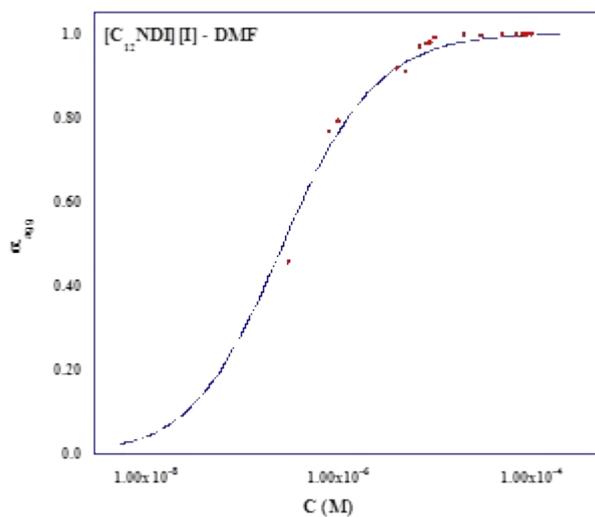
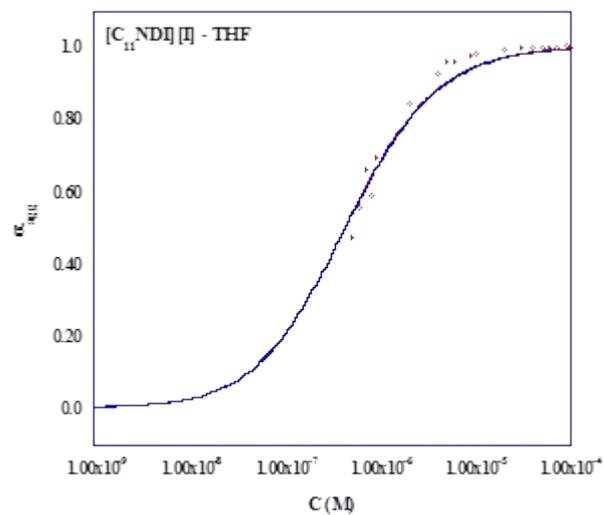
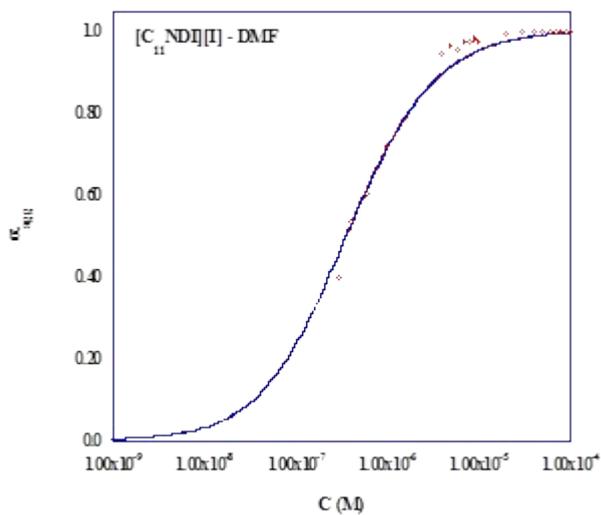
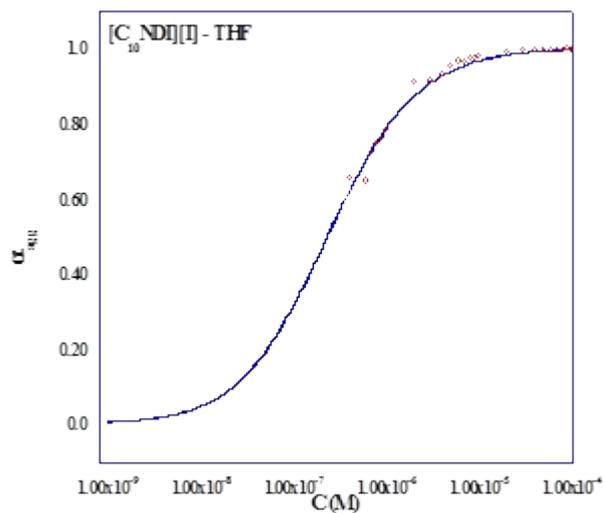
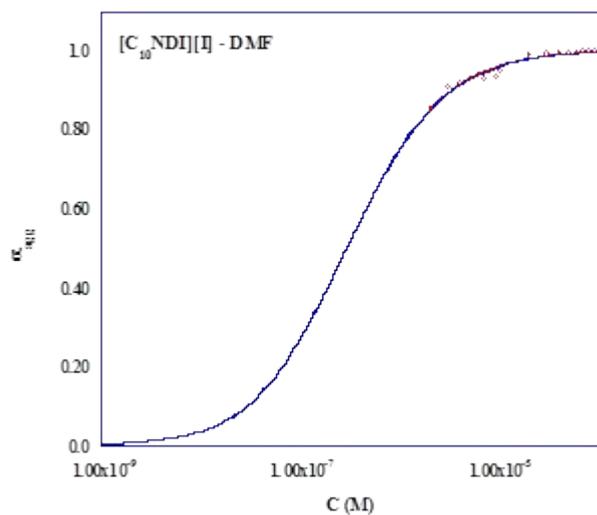


Figure S4. Absorbance and fluorescence intensity (arbitrary units) as a function of concentration and solvent. (Inset: trend of ϵ as a function of concentration; trend of I/C as function of concentration). I/C values were determined at 415 nm for $[C_8\text{NDI}][I]$ in DMF, $[C_{10}\text{NDI}][I]$ in THF and DMF. Moreover, I/C was determined at 533 nm for $[C_{12}\text{NDI}][I]$ in dioxane, 435 nm $[C_{12}\text{NDI}][I]$ in CHCl_3 and at 422 nm in all other cases.







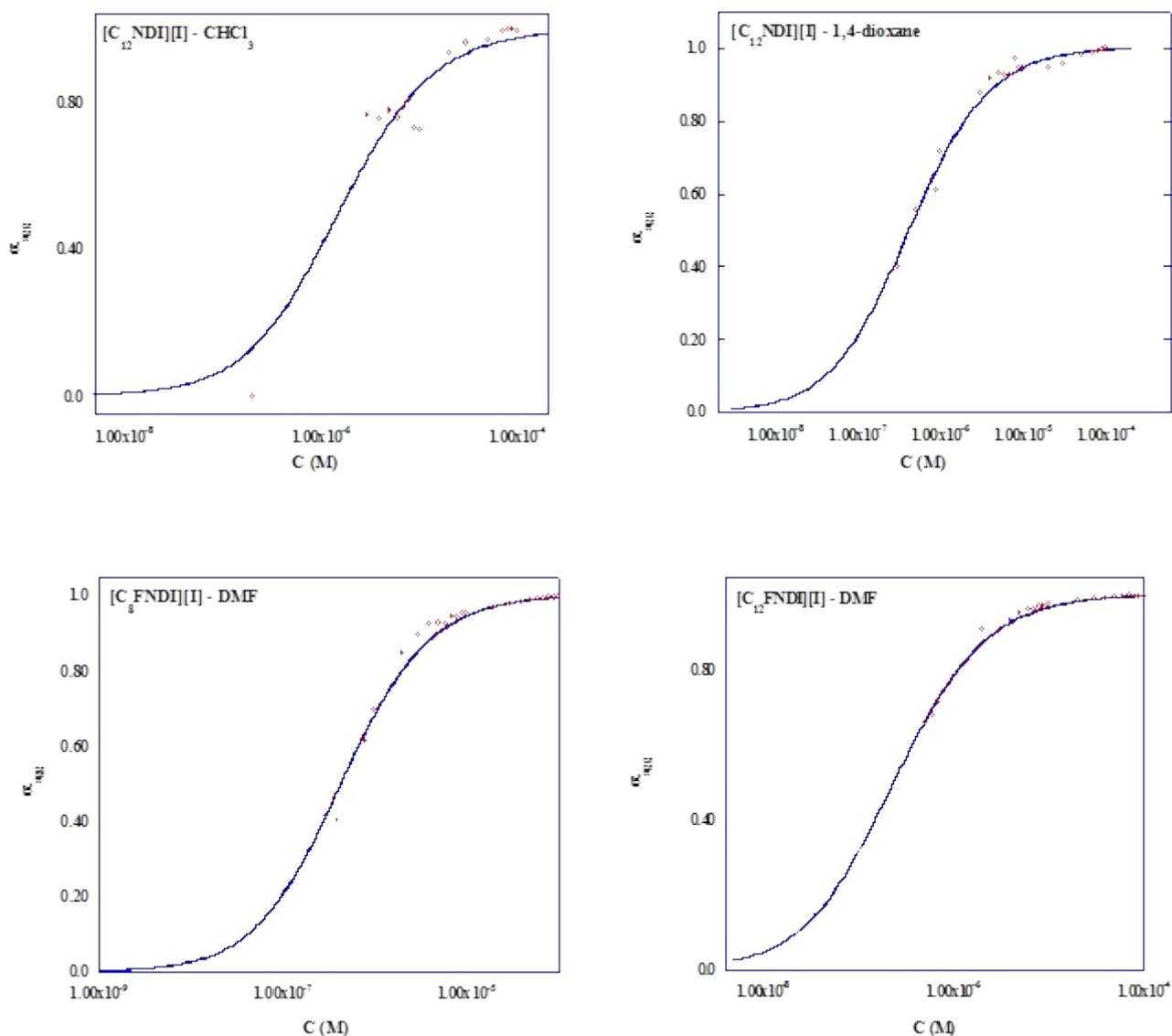
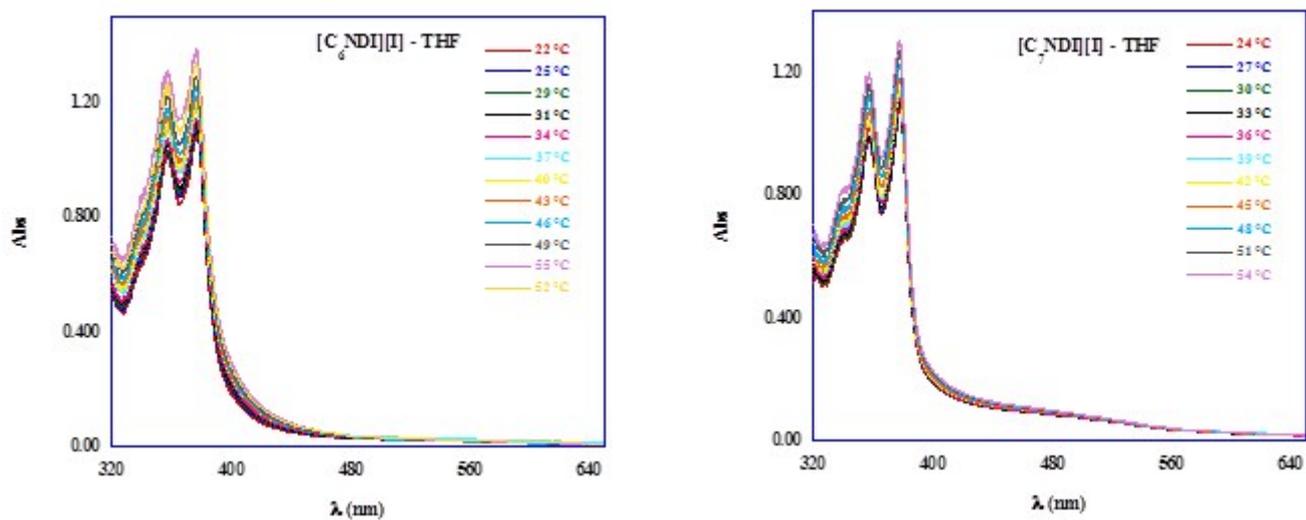


Figure S5. Plot of α_{agg} as a function of concentration and solvent. α_{agg} was determined at 415 nm for $[C_8NDI][I]$ in DMF, $[C_{10}NDI][I]$ in THF and DMF. α_{agg} was determined at 533 nm for $[C_{12}NDI][I]$ in dioxane, at 435 nm at $[C_{12}NDI][I]$ in $CHCl_3$ and at 422 nm in all the other cases.



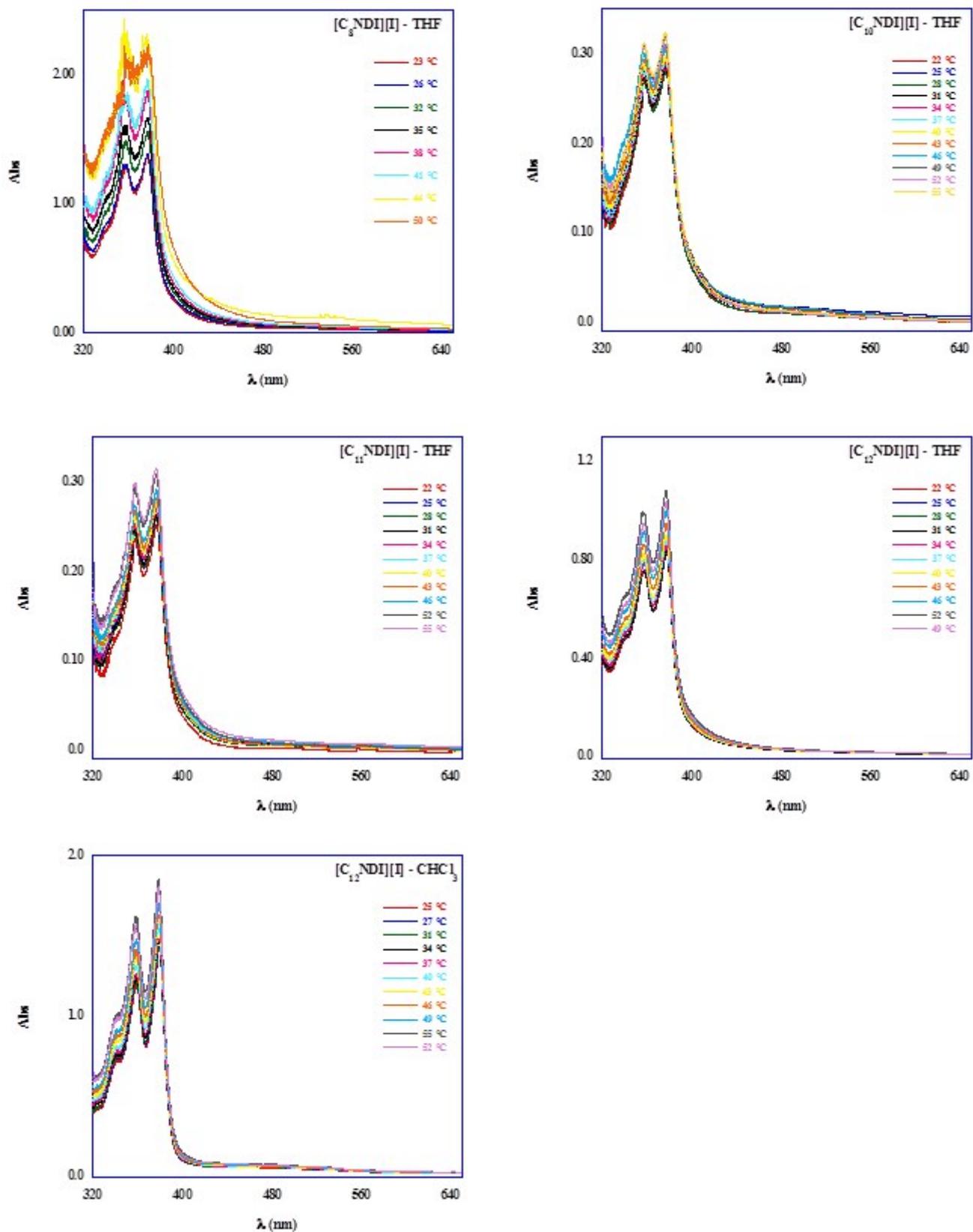
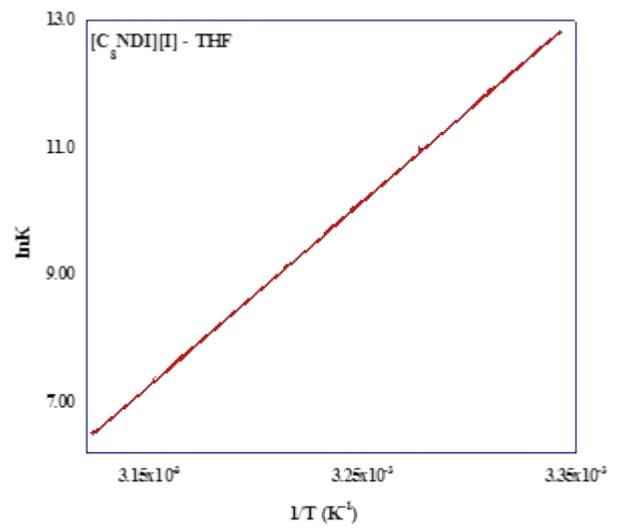
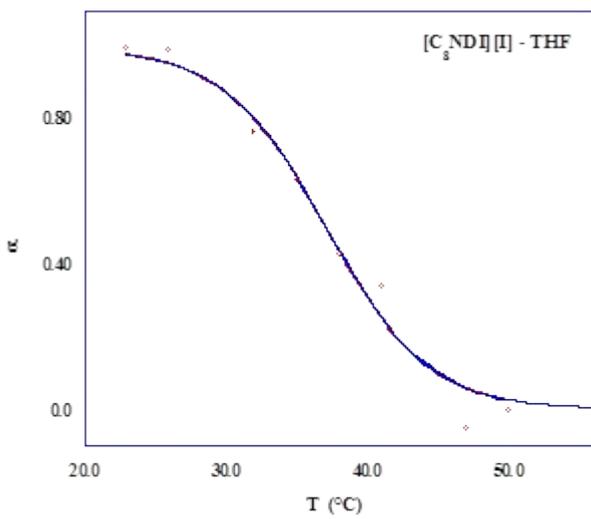
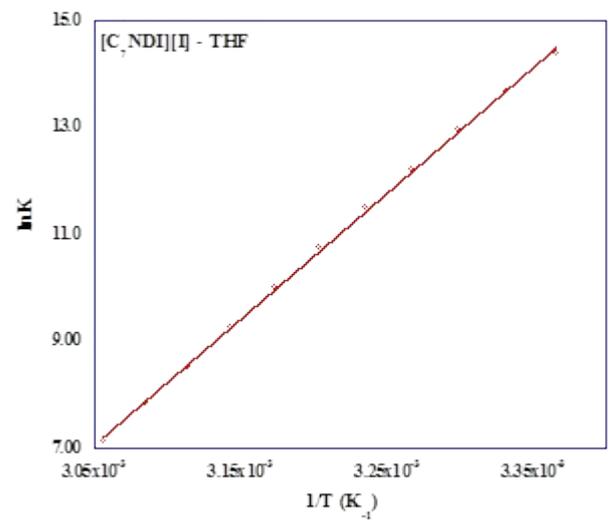
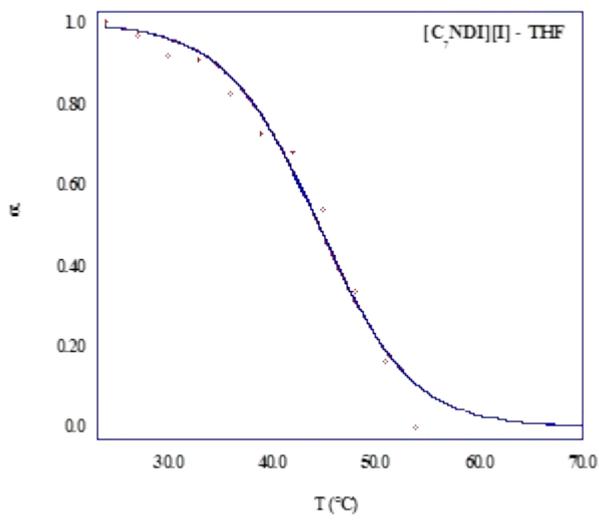
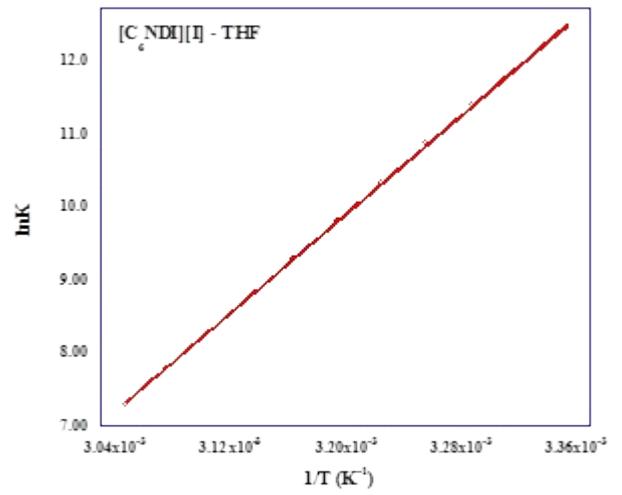
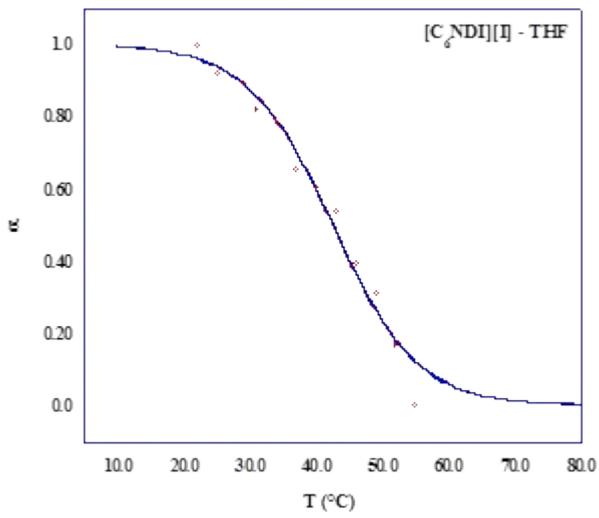
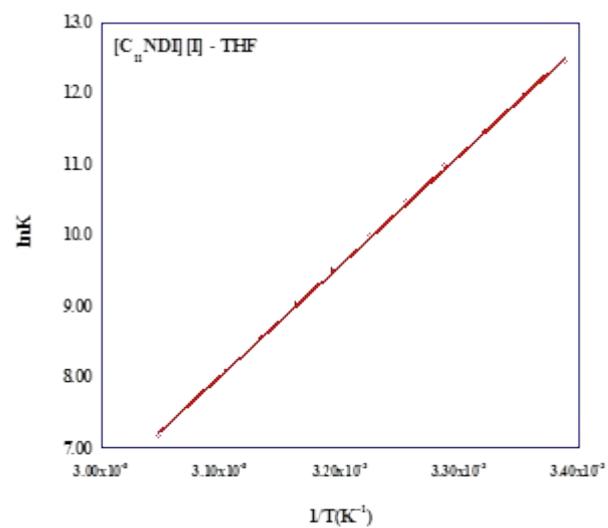
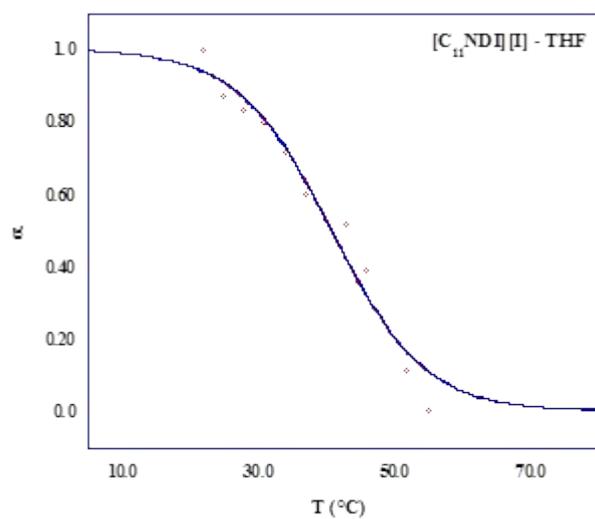
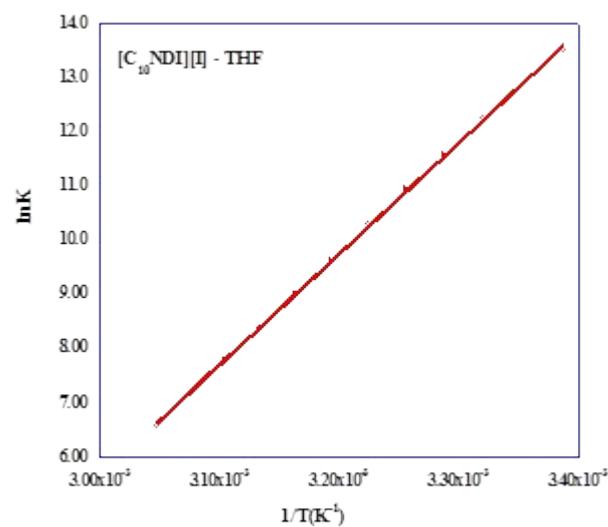
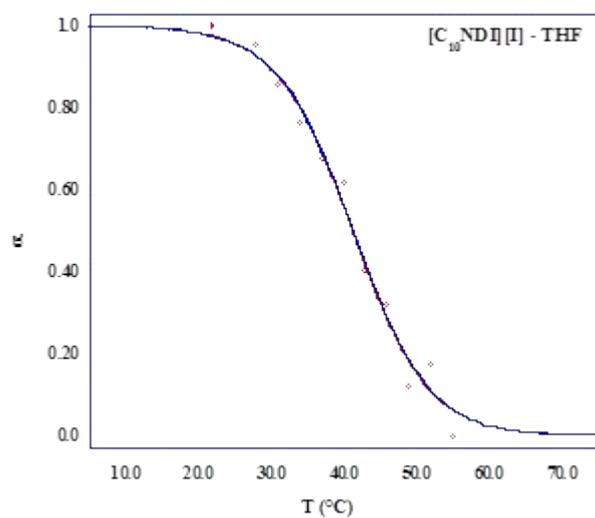
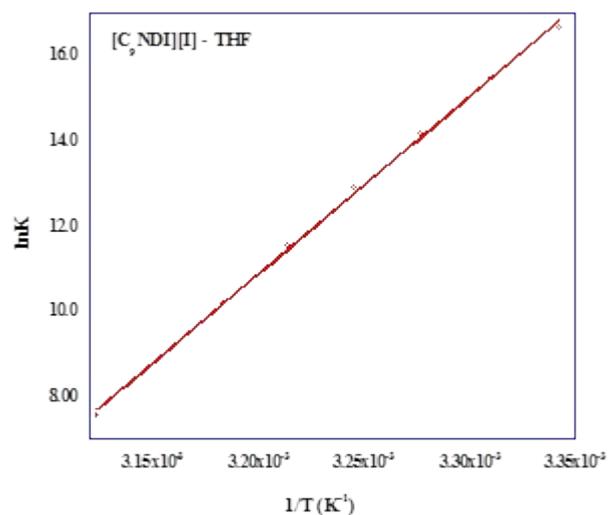
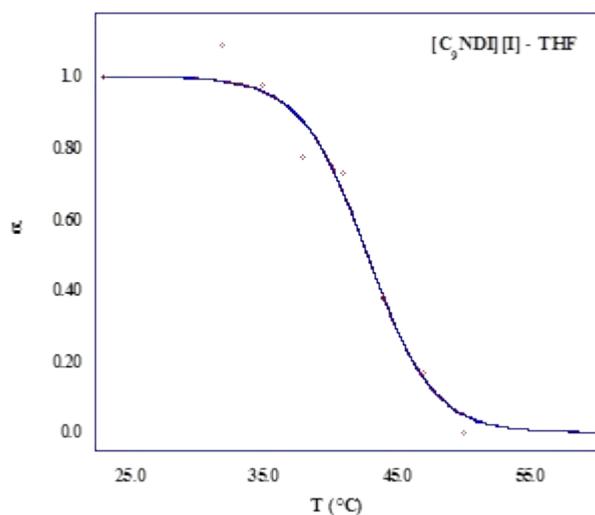


Figure S6. UV-vis spectra at fixed concentration ($5 \cdot 10^{-5}$ M) as a function of temperature.





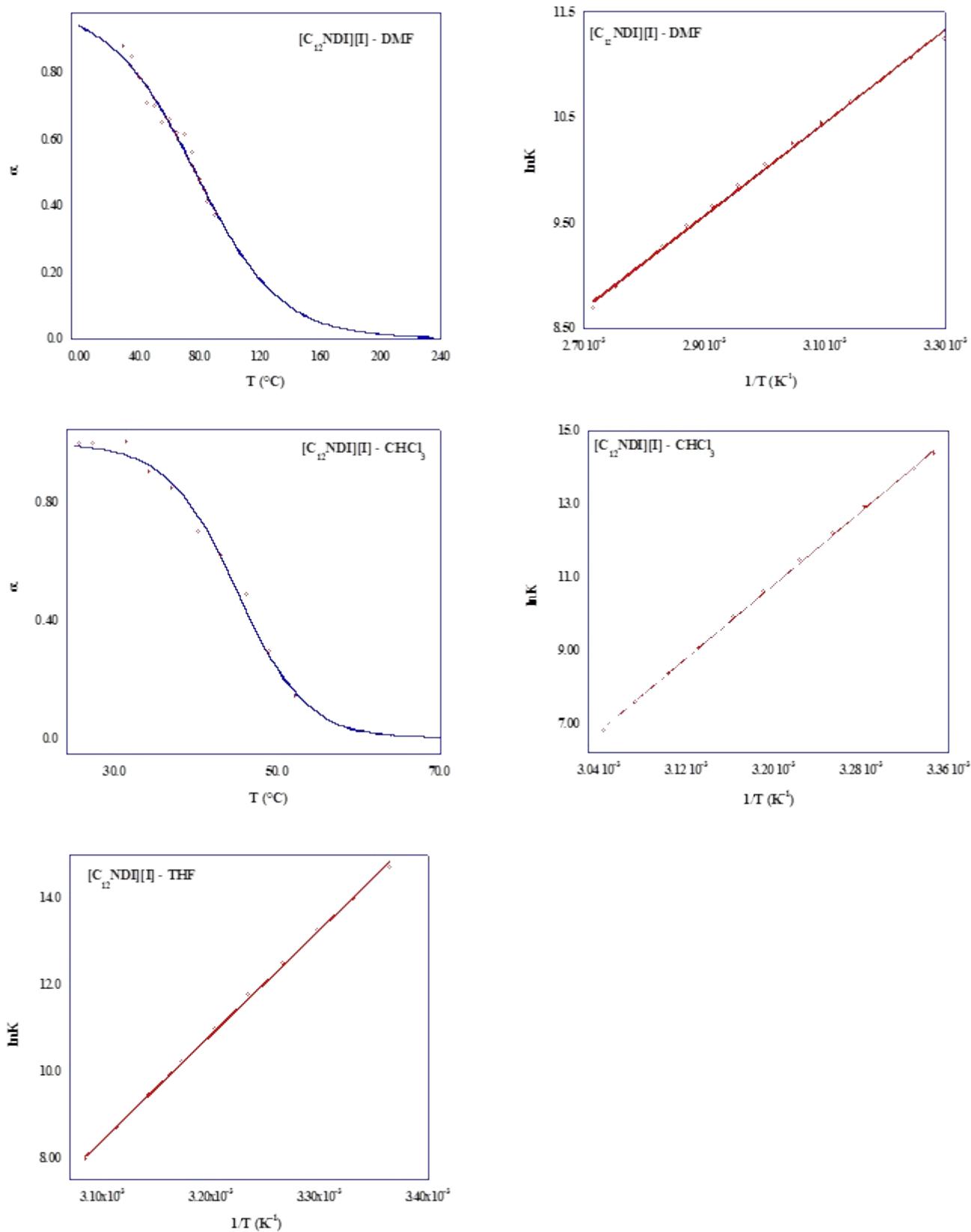


Figure S7. α and van't Hoff plots of UV-vis analysis at variable temperature and fixed concentration ($5 \cdot 10^{-5}$ M). α_{agg} was determined at 382 nm for $[C_6NDI][I]$, at 380 nm for $[C_7NDI][I]$, at 375 nm for $[C_8NDI][I]$, at 377 for $[C_9NDI][I]$, at 375 nm for $[C_{10}NDI][I]$ and at 374 nm $[C_{11}NDI][I]$.

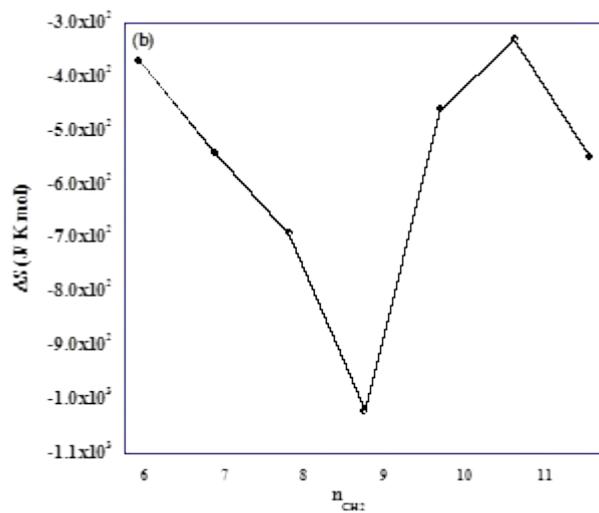
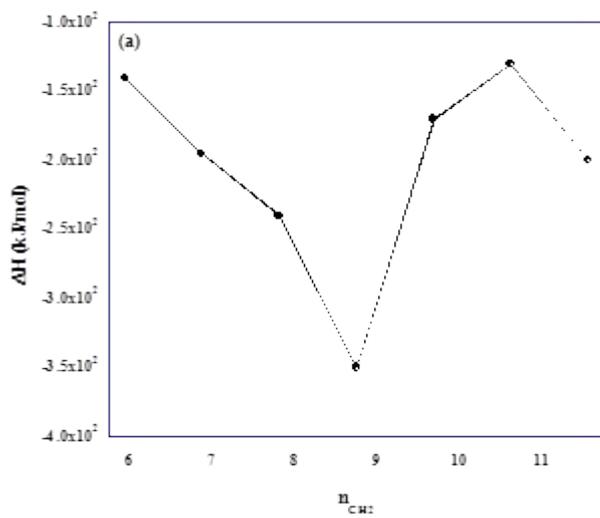
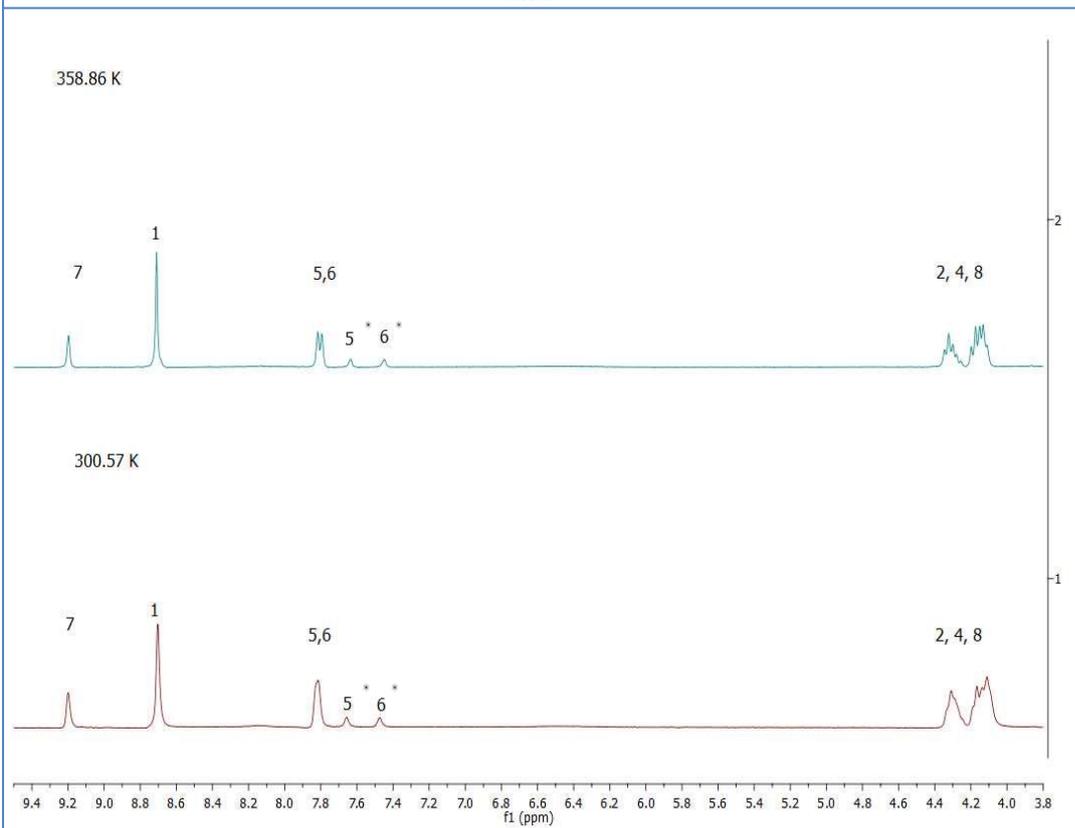
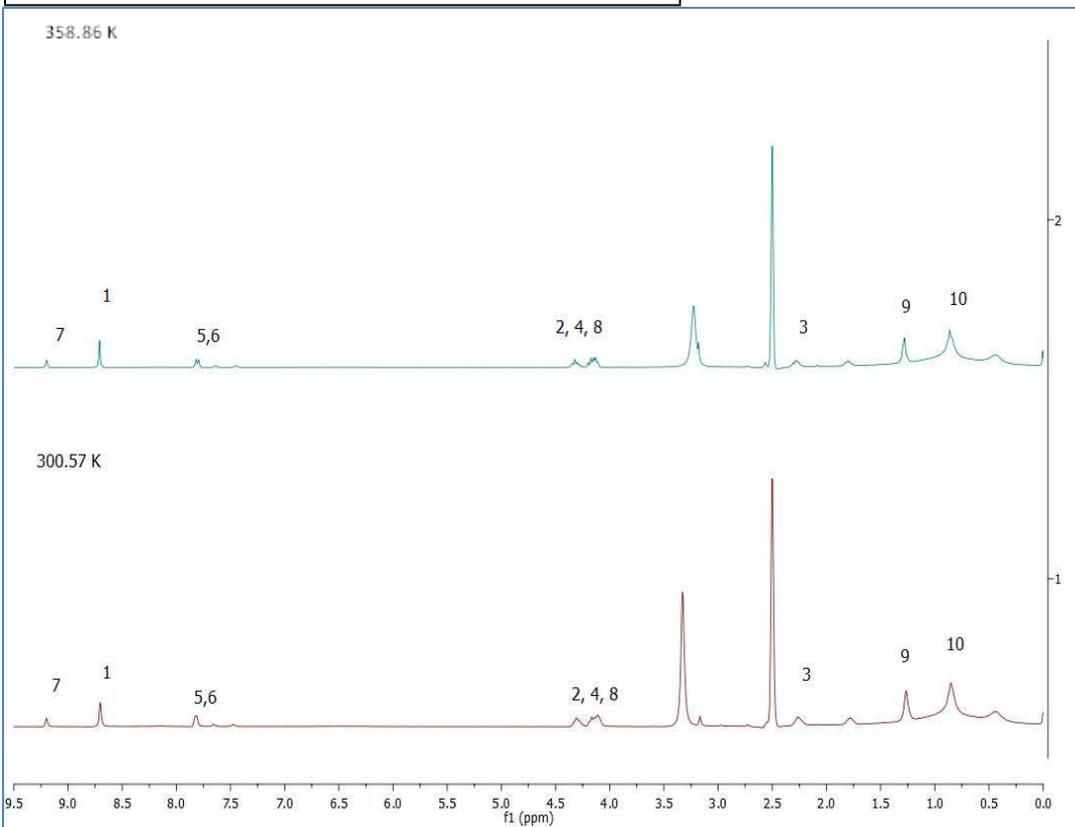
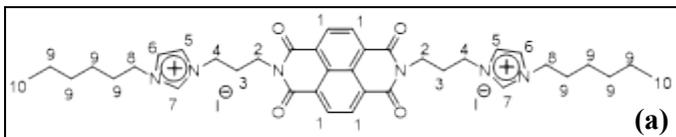
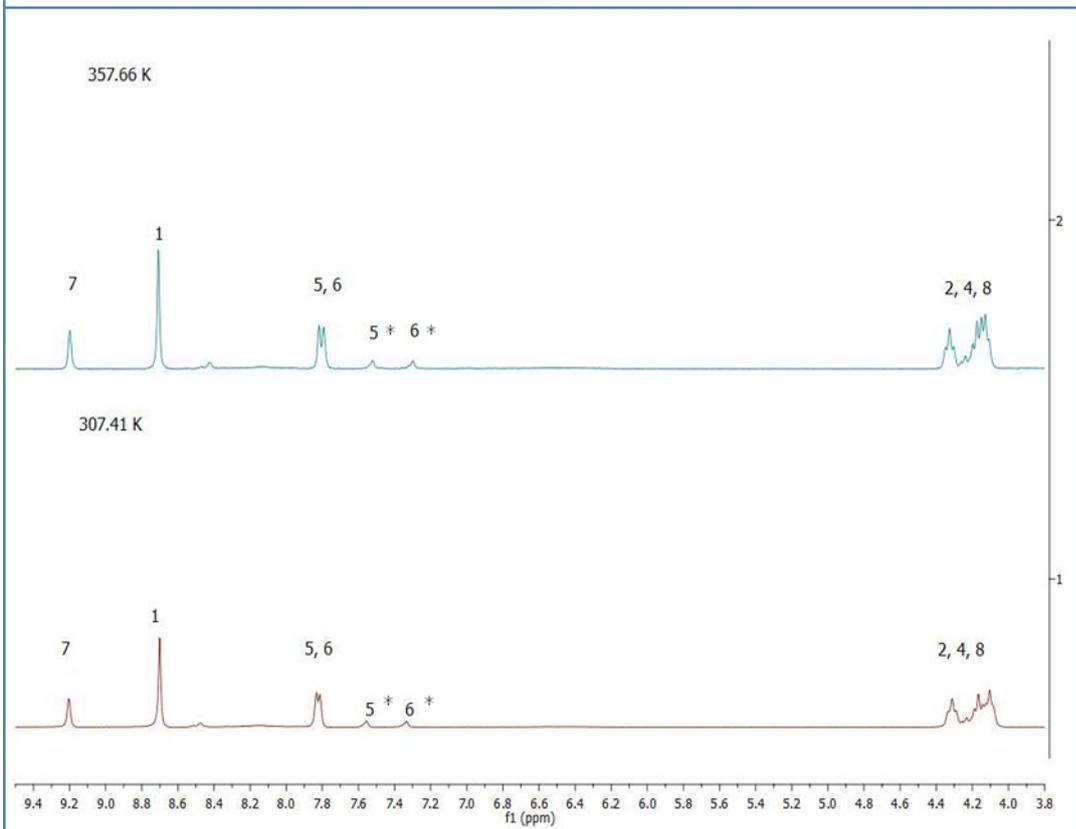
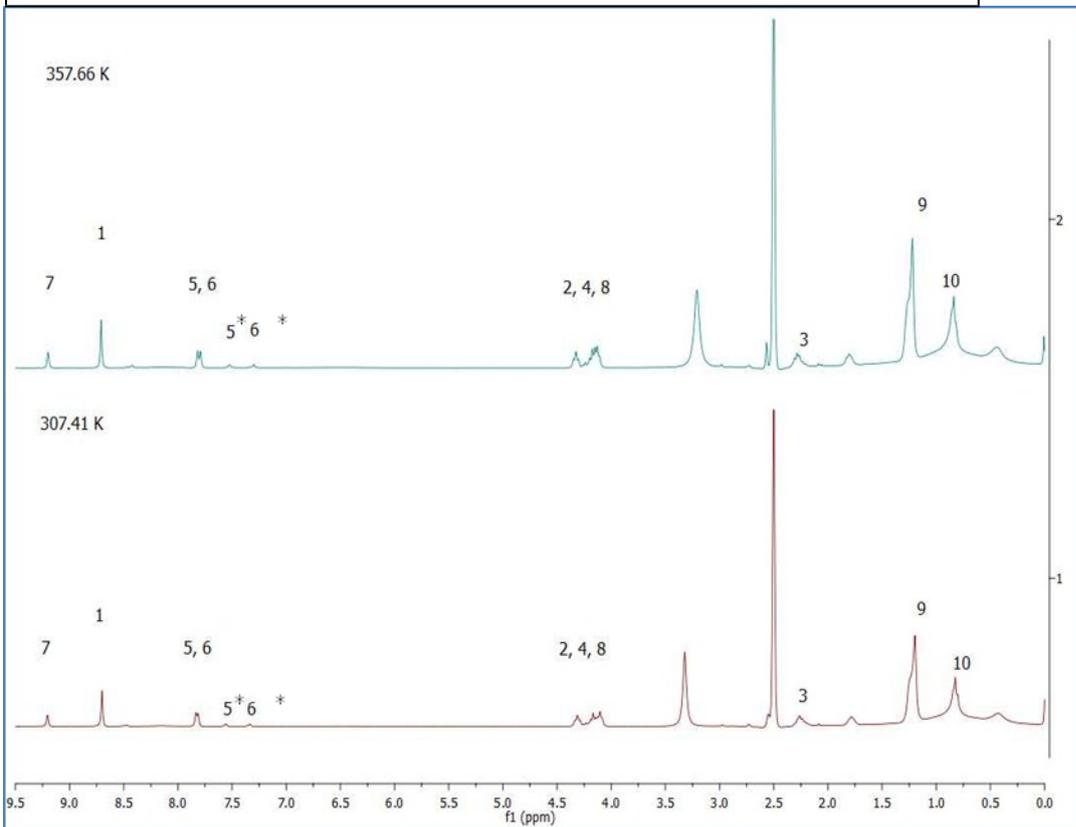
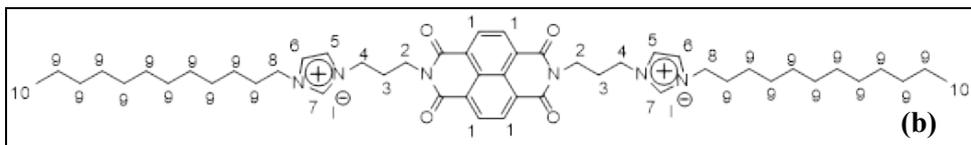


Figure S8. Plots of (a) ΔH and (b) ΔS as a function of the alkyl chain length.





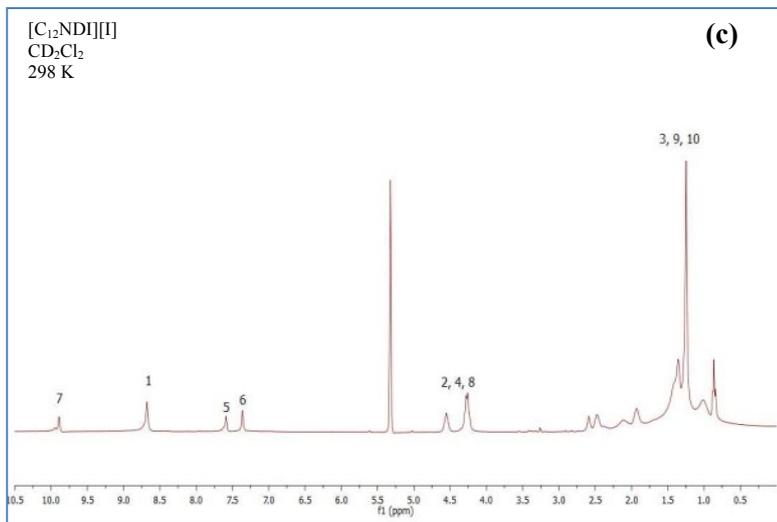
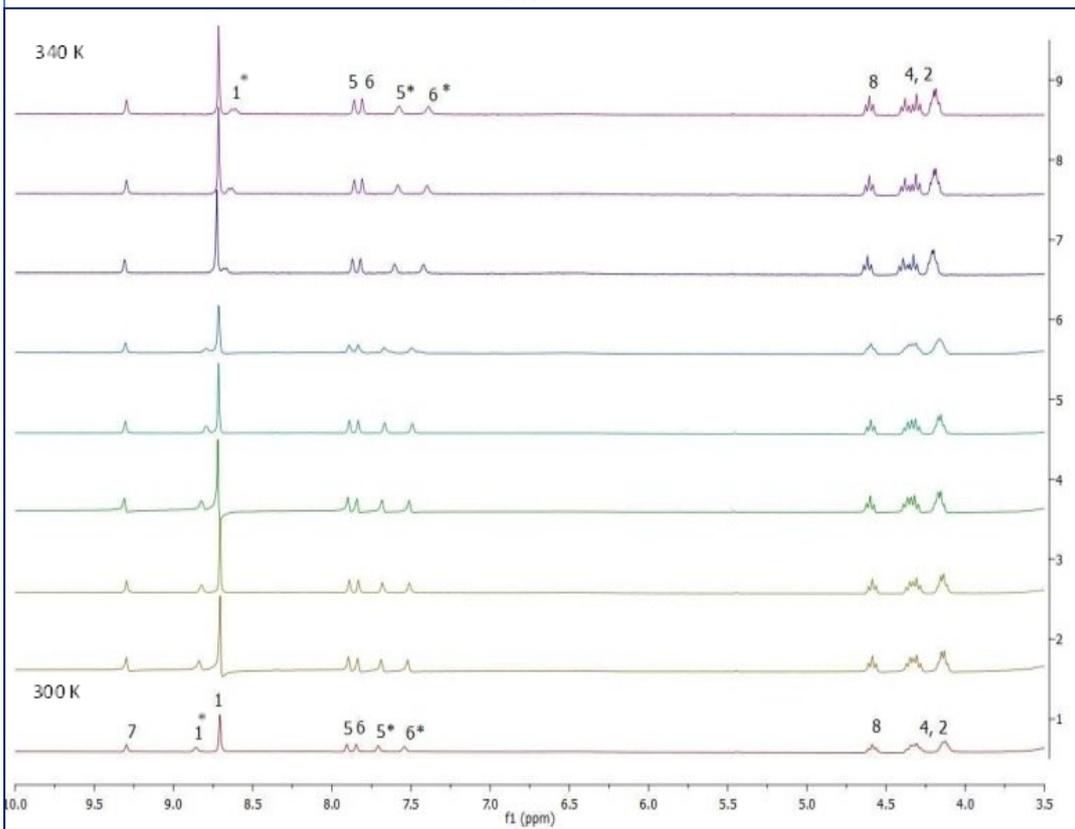
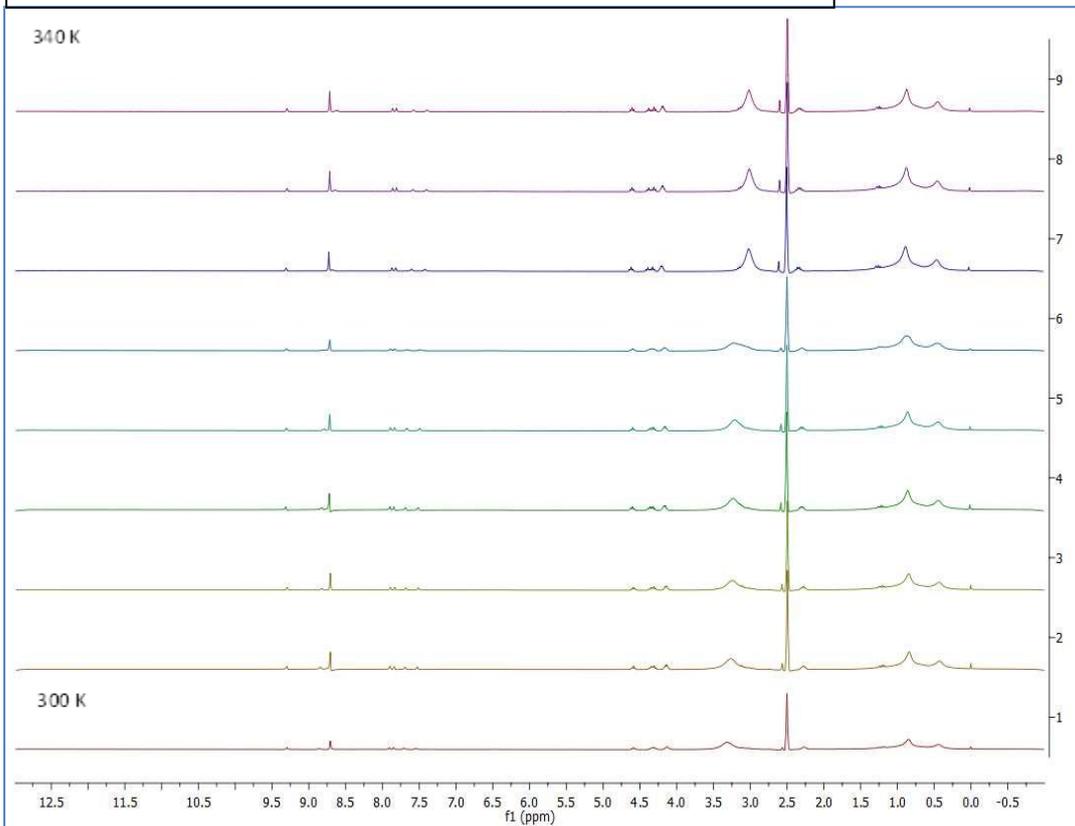
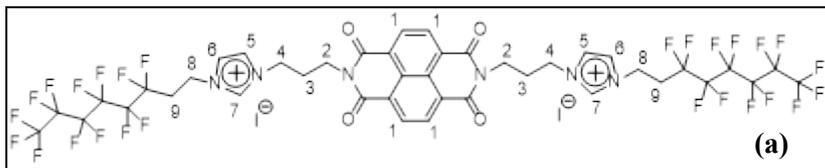


Figure S9. ¹H NMR spectra at variable temperature of (a)[C₆NDI][I] in DMSO-[d₆], (b)[C₁₂NDI][I] in DMSO-[d₆] and (c)[C₁₂NDI][I] in CD₂Cl₂-[d₂] respectively.



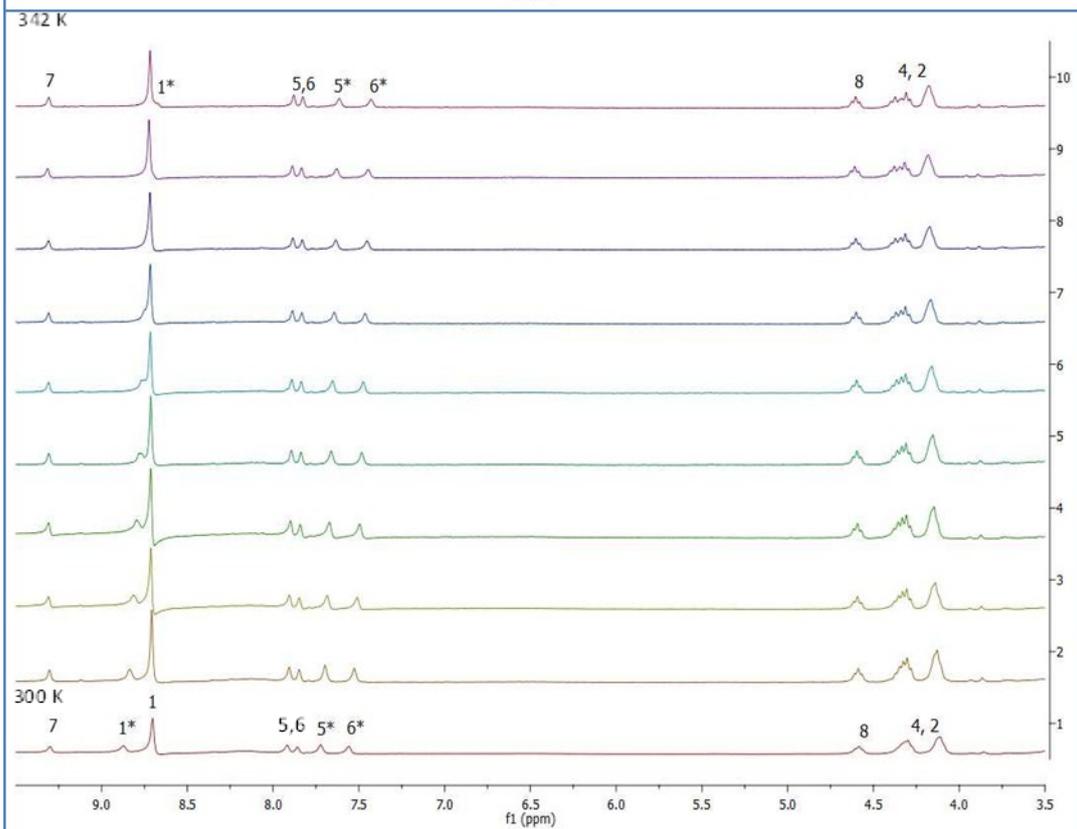
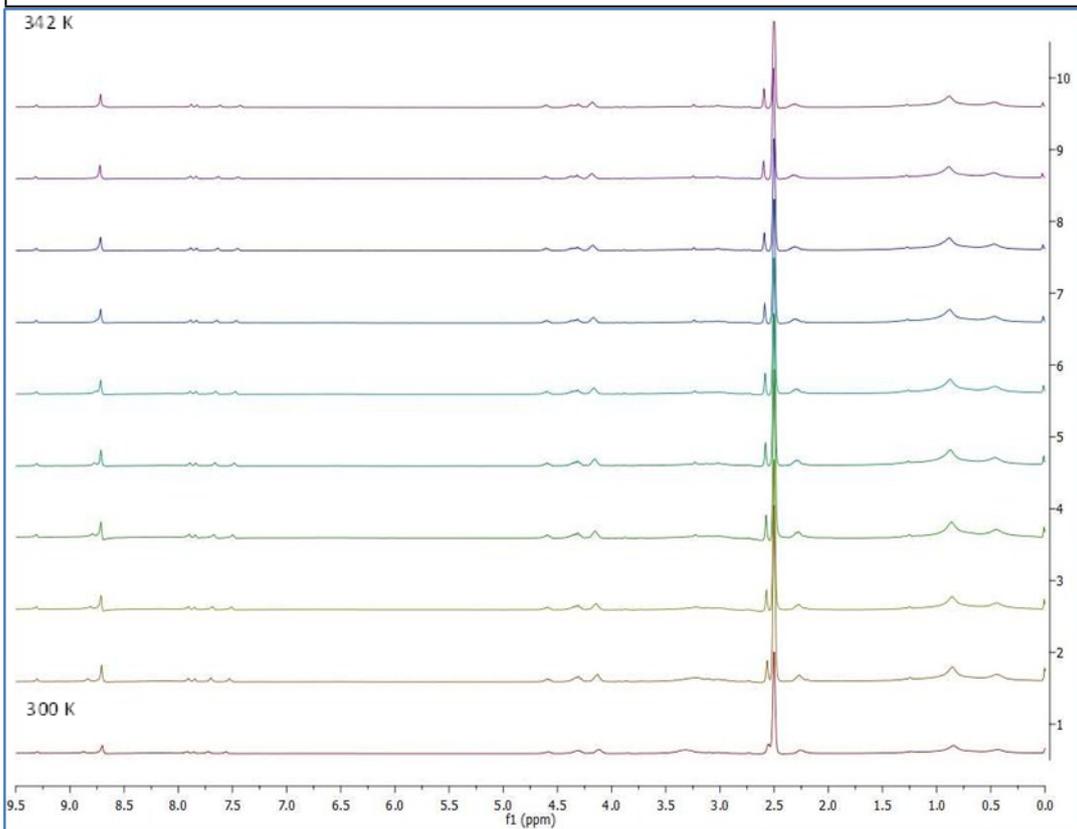
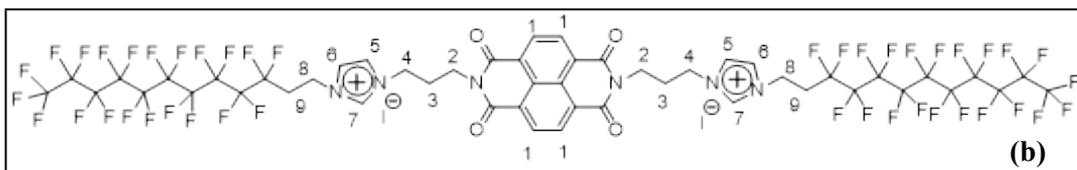
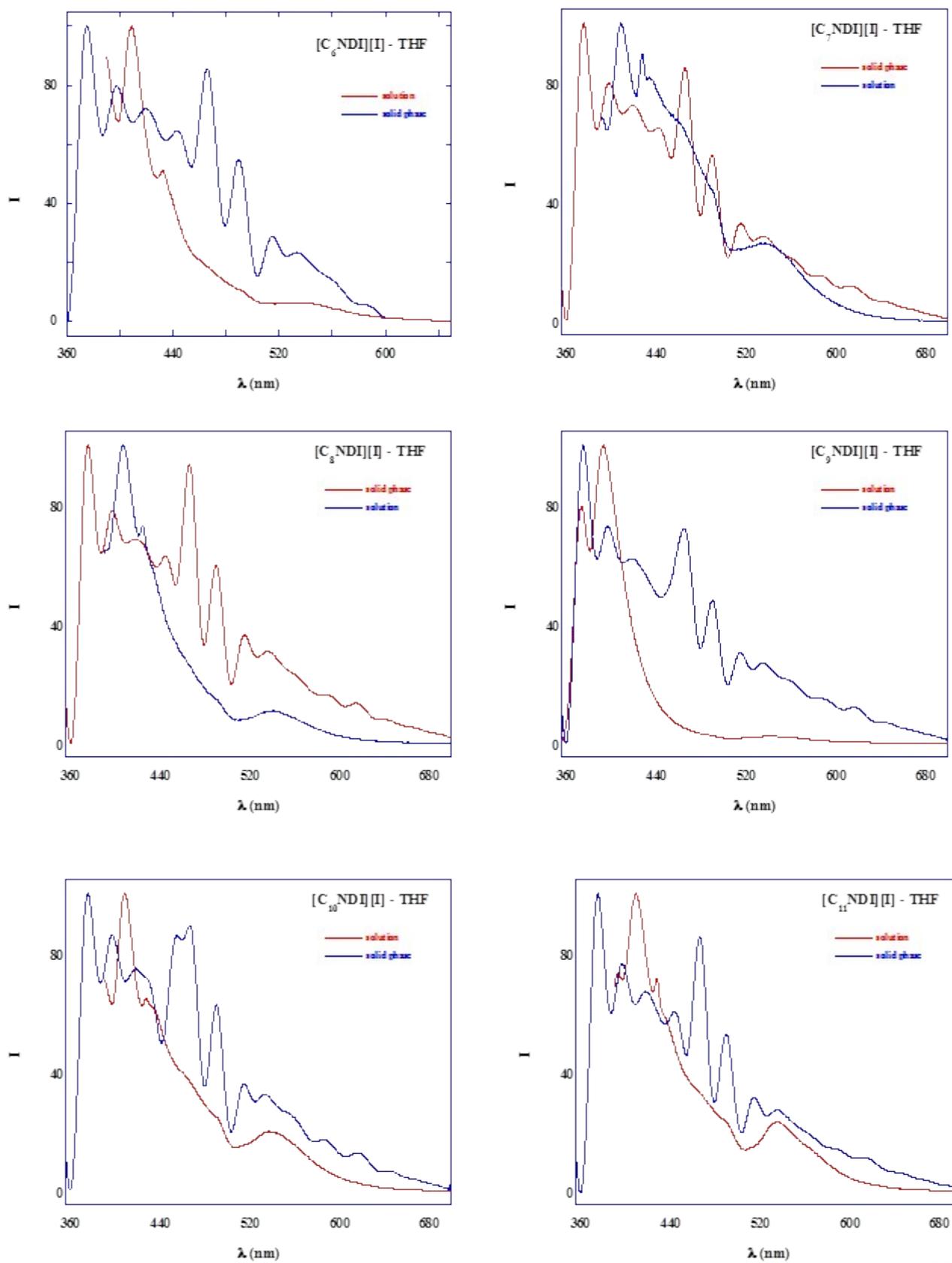


Figure S10. ^1H NMR spectra at variable temperature of (a) $[\text{C}_8\text{FNDI}][\text{I}]$ in $\text{DMSO}-[d_6]$ and (b) $[\text{C}_{12}\text{FNDI}][\text{I}]$ in $\text{DMSO}-[d_6]$ respectively.



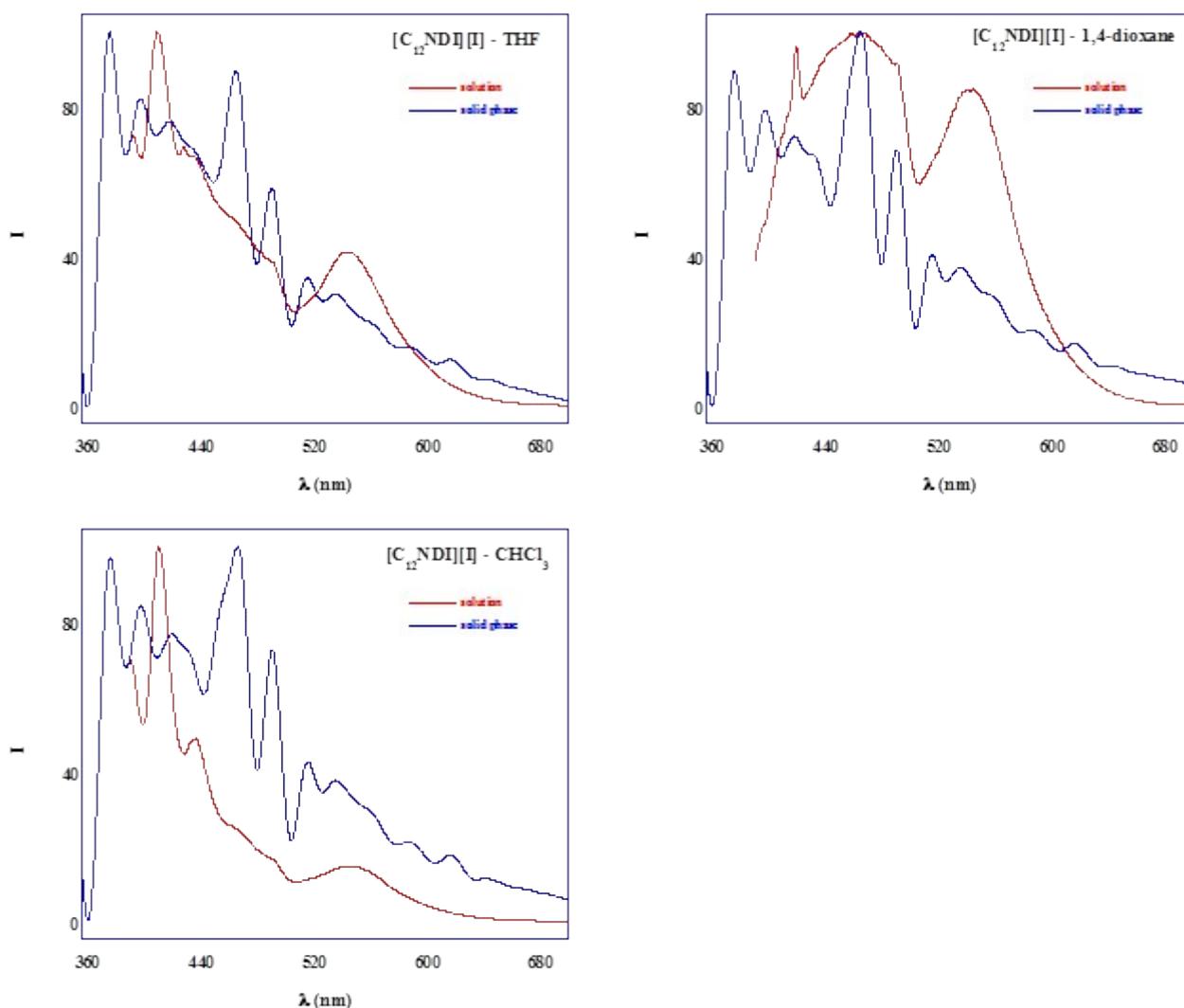


Figure S11. Fluorescent spectra in solution and in solid phase of salts. Fluorescence intensities are in arbitrary units. Excitation wavelength for solid-phase spectra are: 275 nm for [C₆NDI][I], 277 nm for [C₇NDI][I], 272 nm for [C₈NDI][I], [C₉NDI][I], and [C₁₀NDI][I], 274 nm for [C₁₂NDI][I] deriving from drop-casting of solution in dioxane and chloroform. Excitation wavelength for solution-phase spectra are: 375 nm for [C₆NDI][I], 380 nm for [C₇NDI][I], 378 nm for [C₈NDI][I], 382 nm for [C₉NDI][I] and [C₁₀NDI][I], 381 nm for [C₁₁NDI][I], 380 nm for [C₁₂NDI][I] in dioxane and 383 nm for [C₁₂NDI][I] in chloroform.

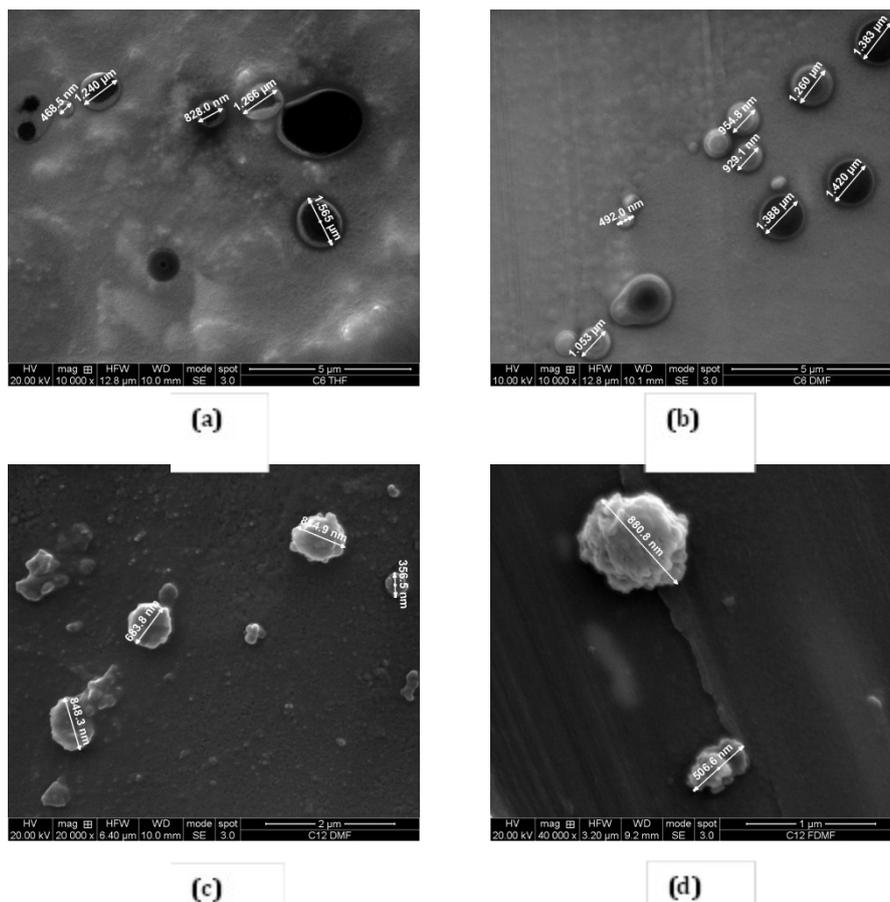
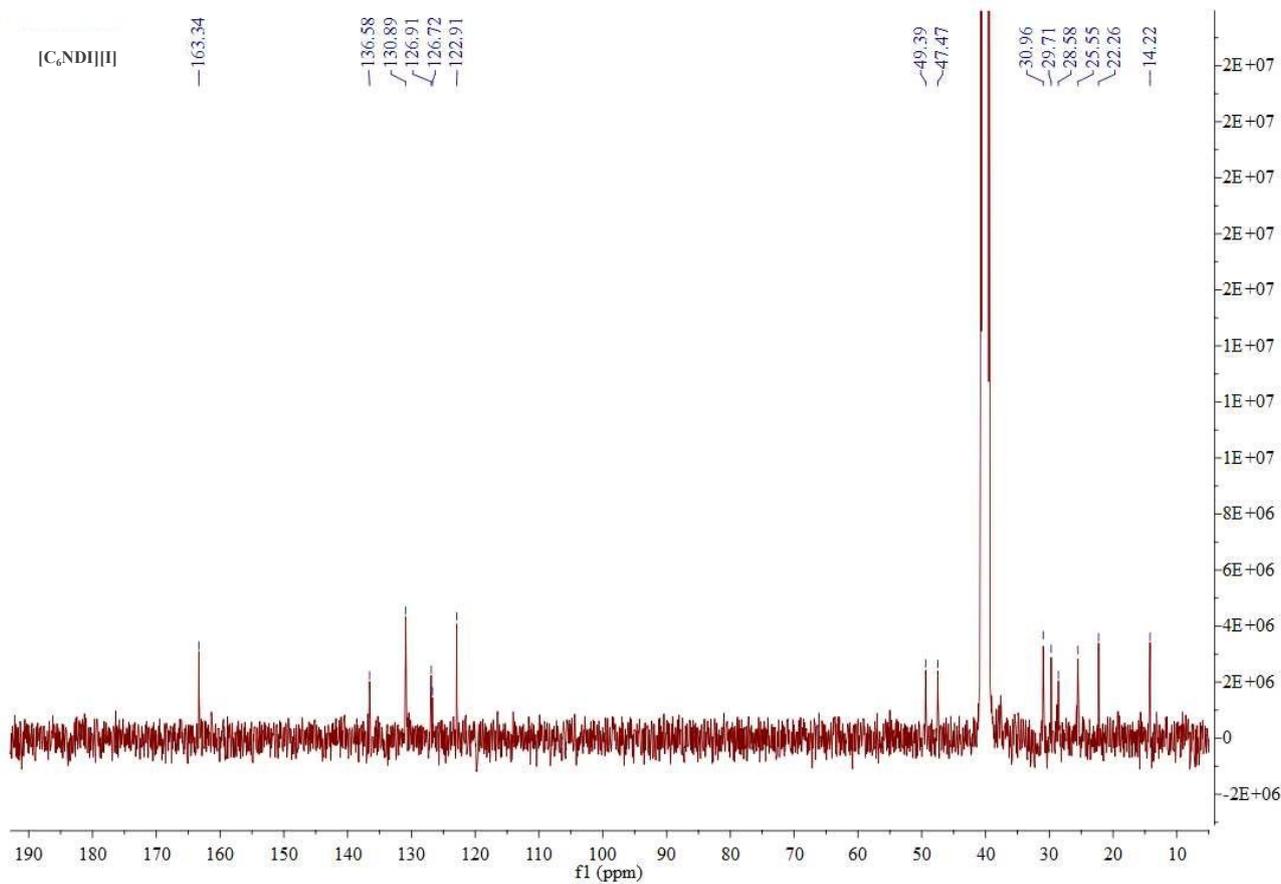
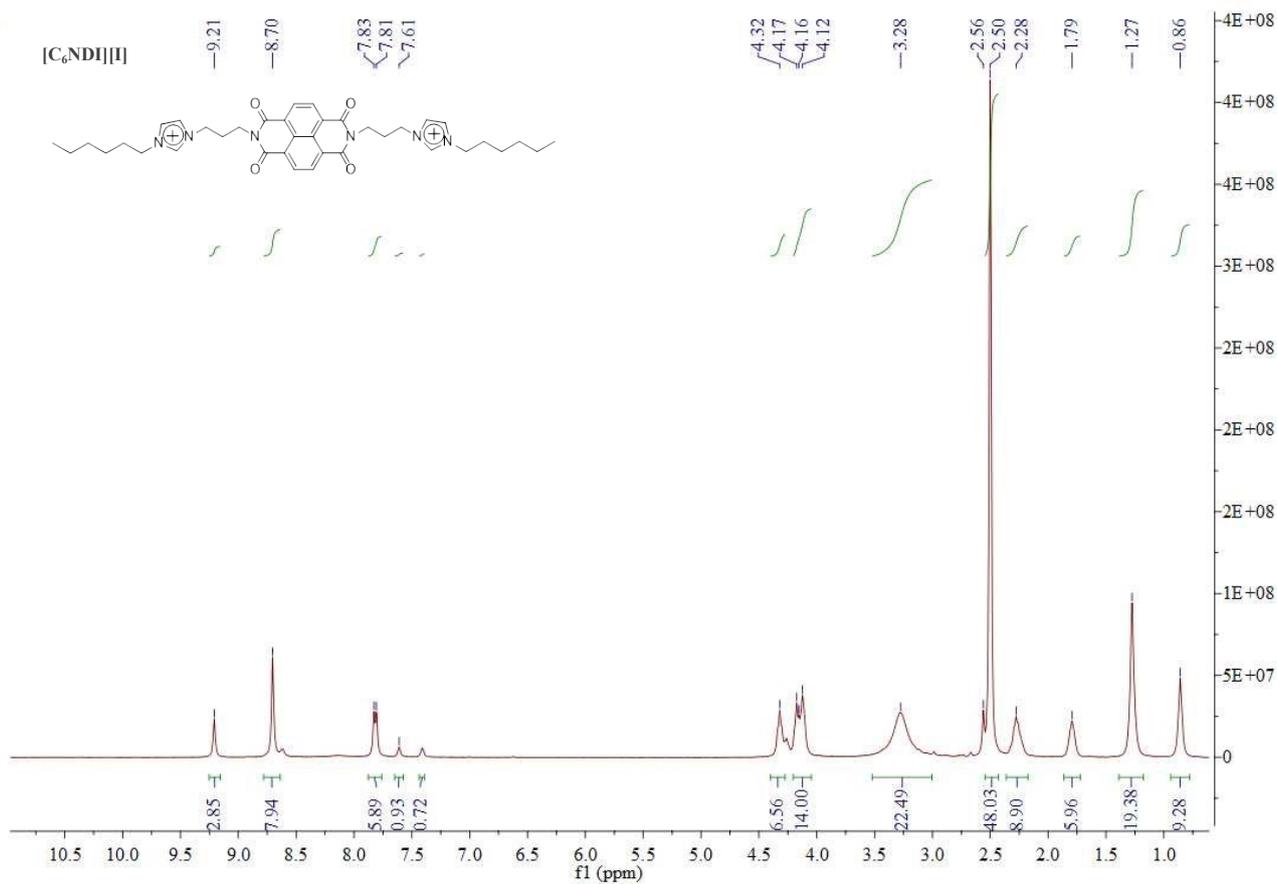
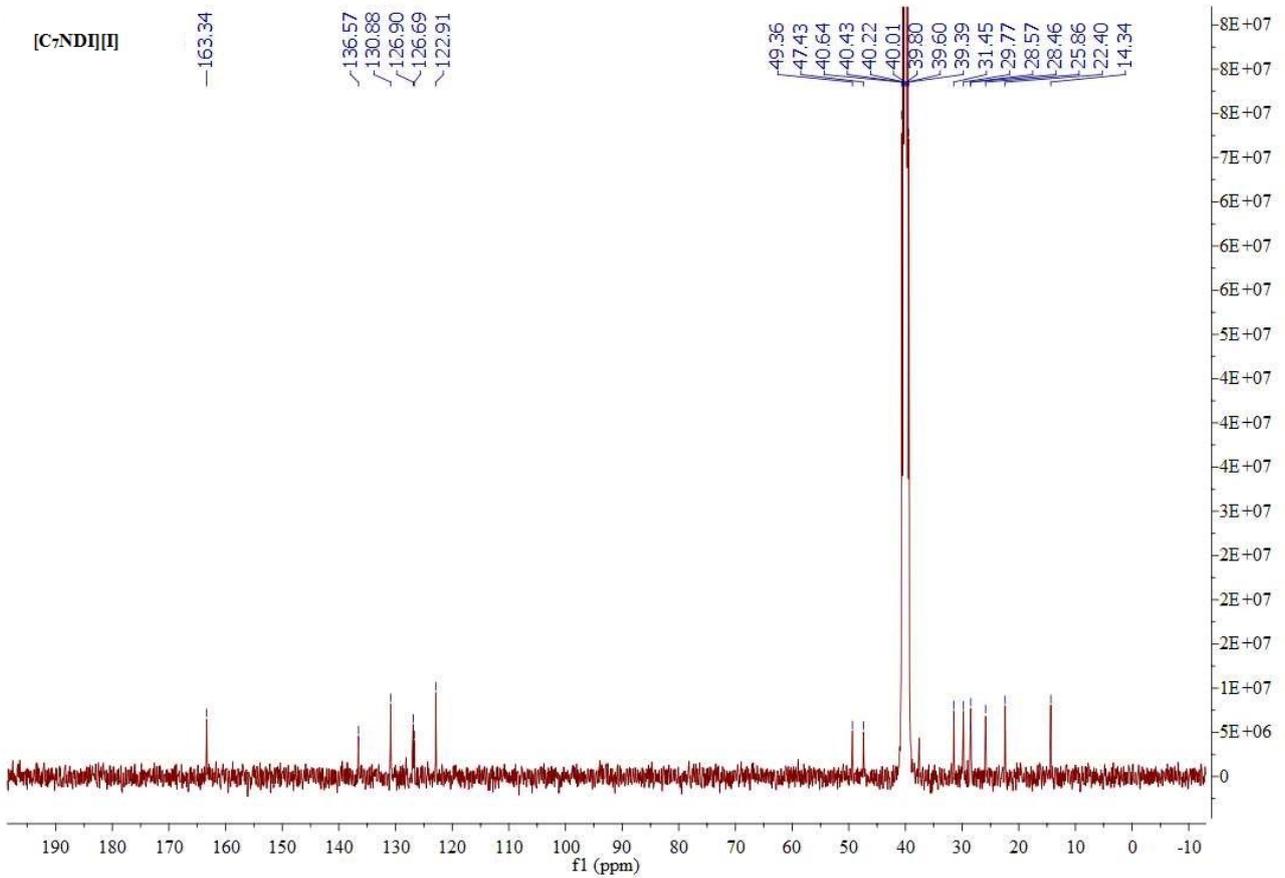
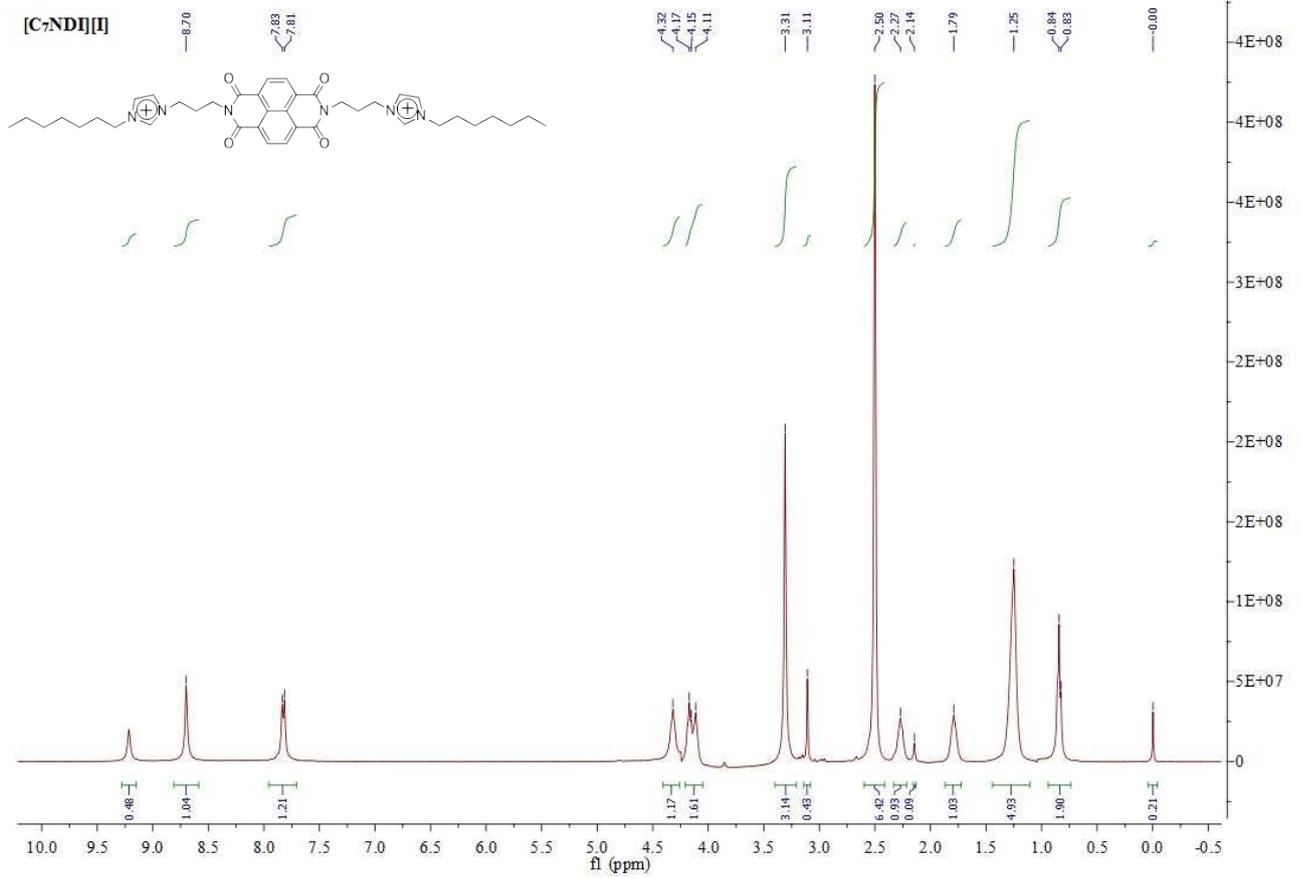
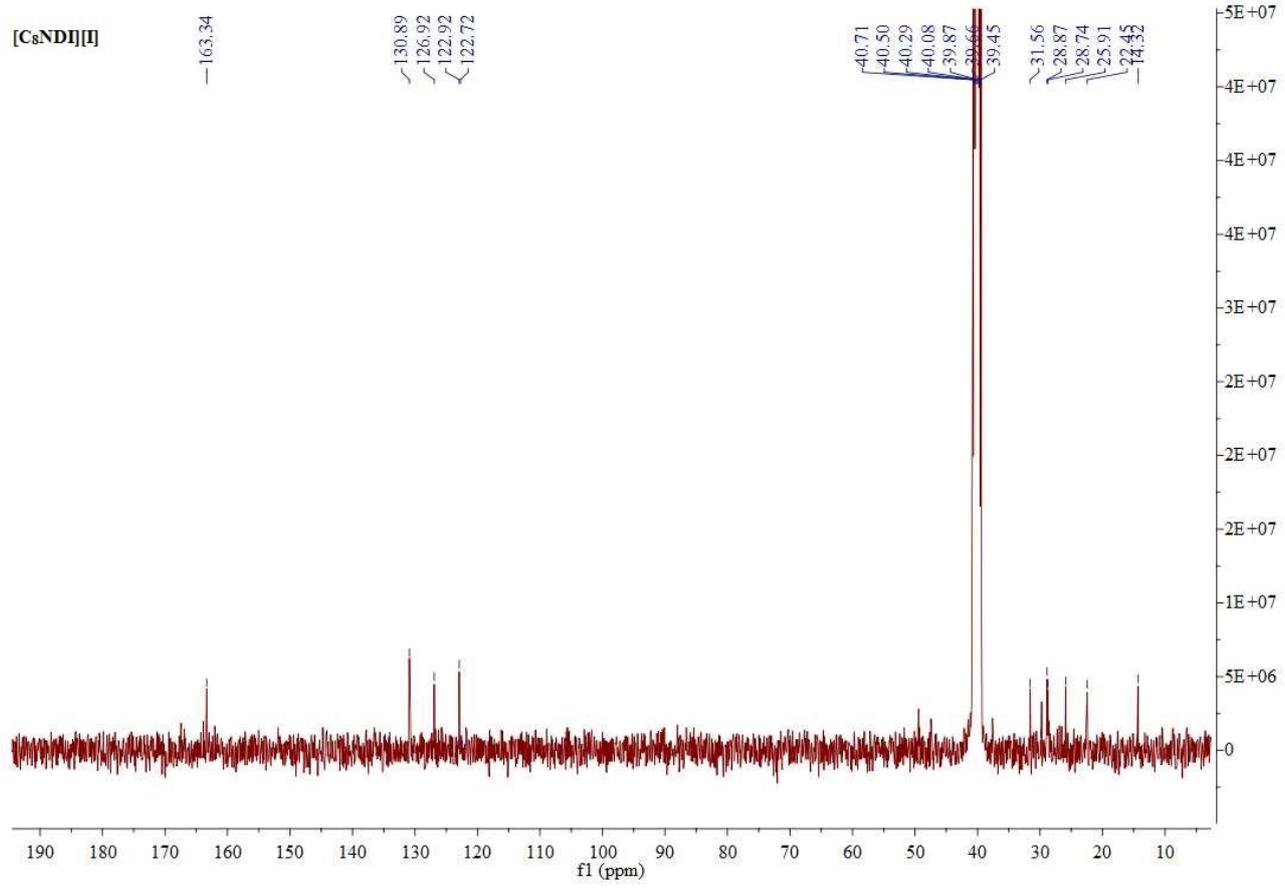
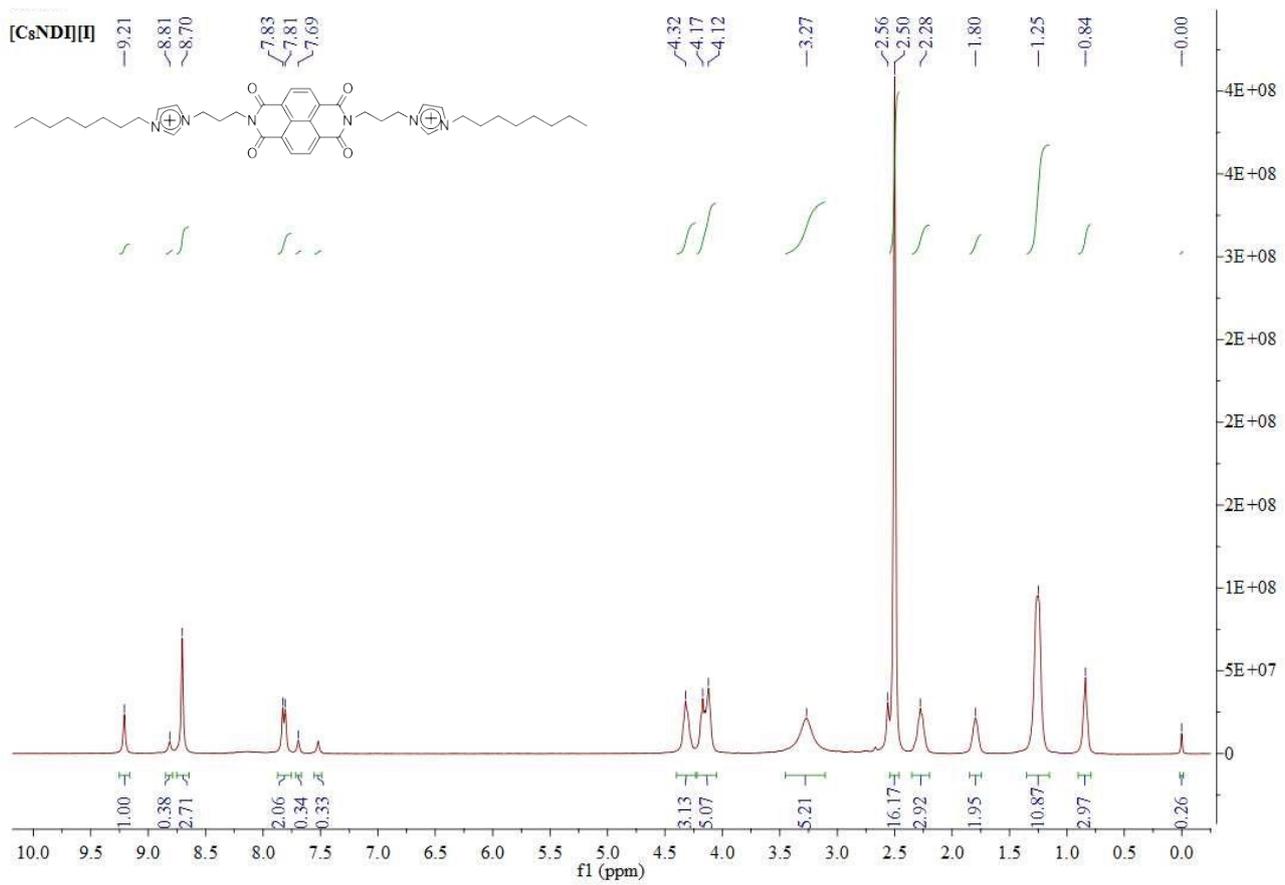
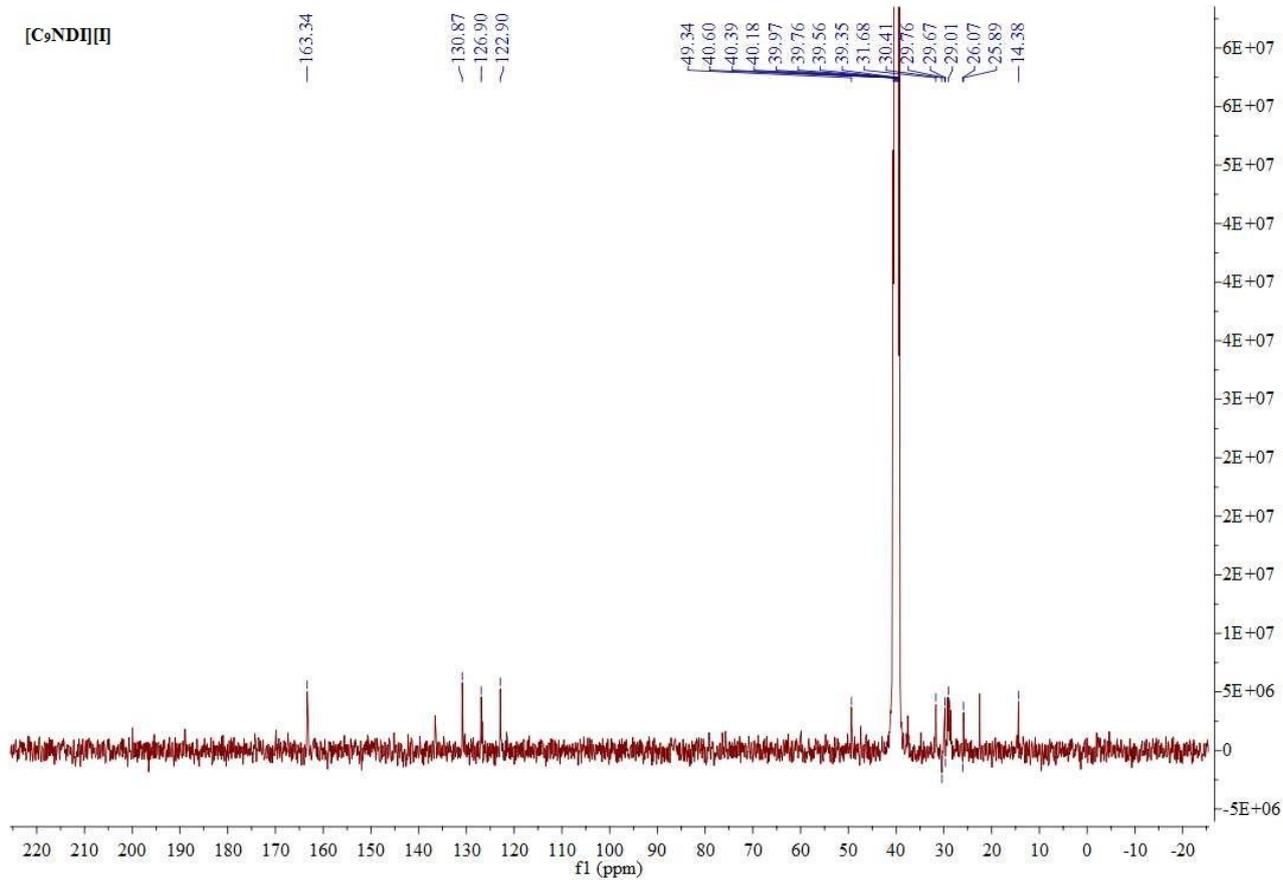
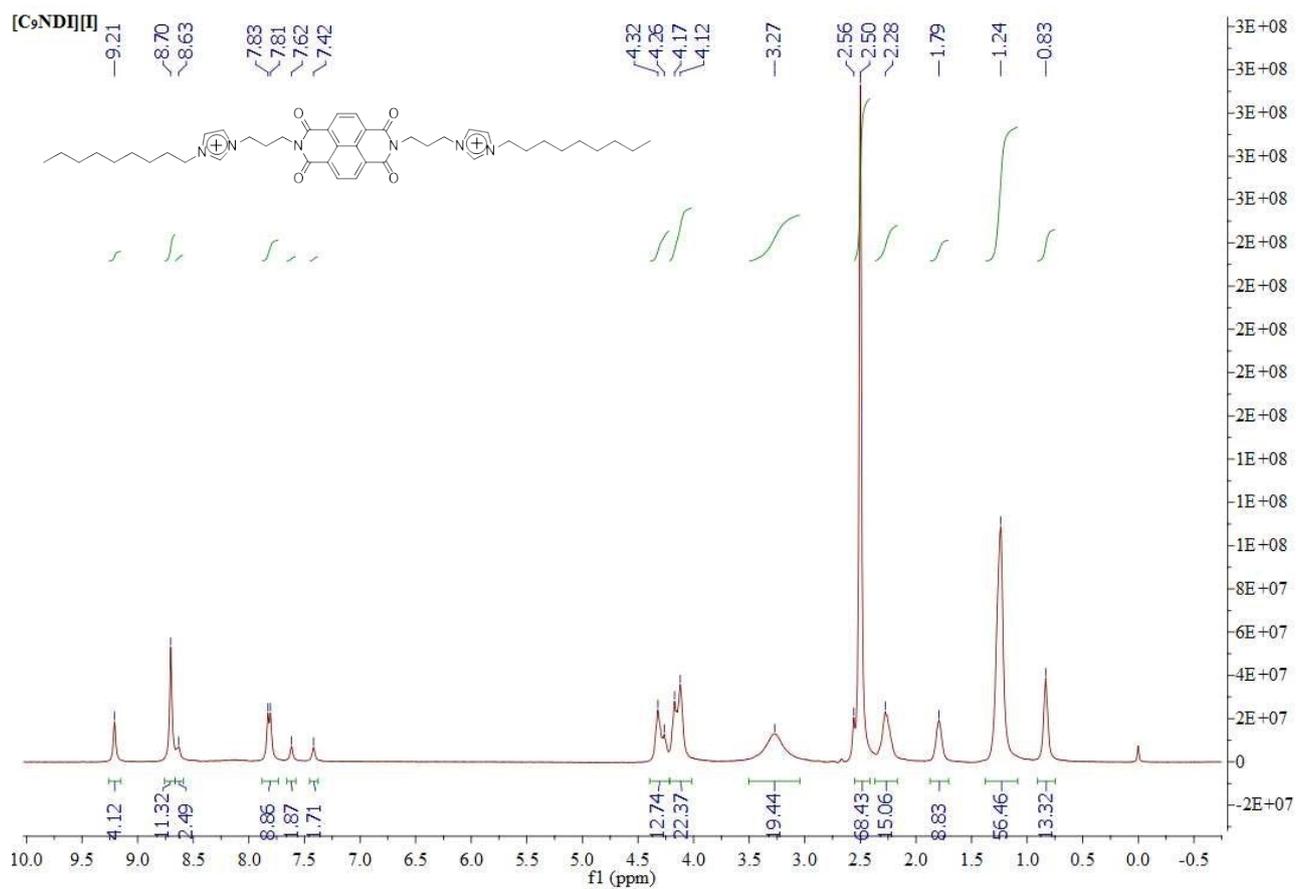


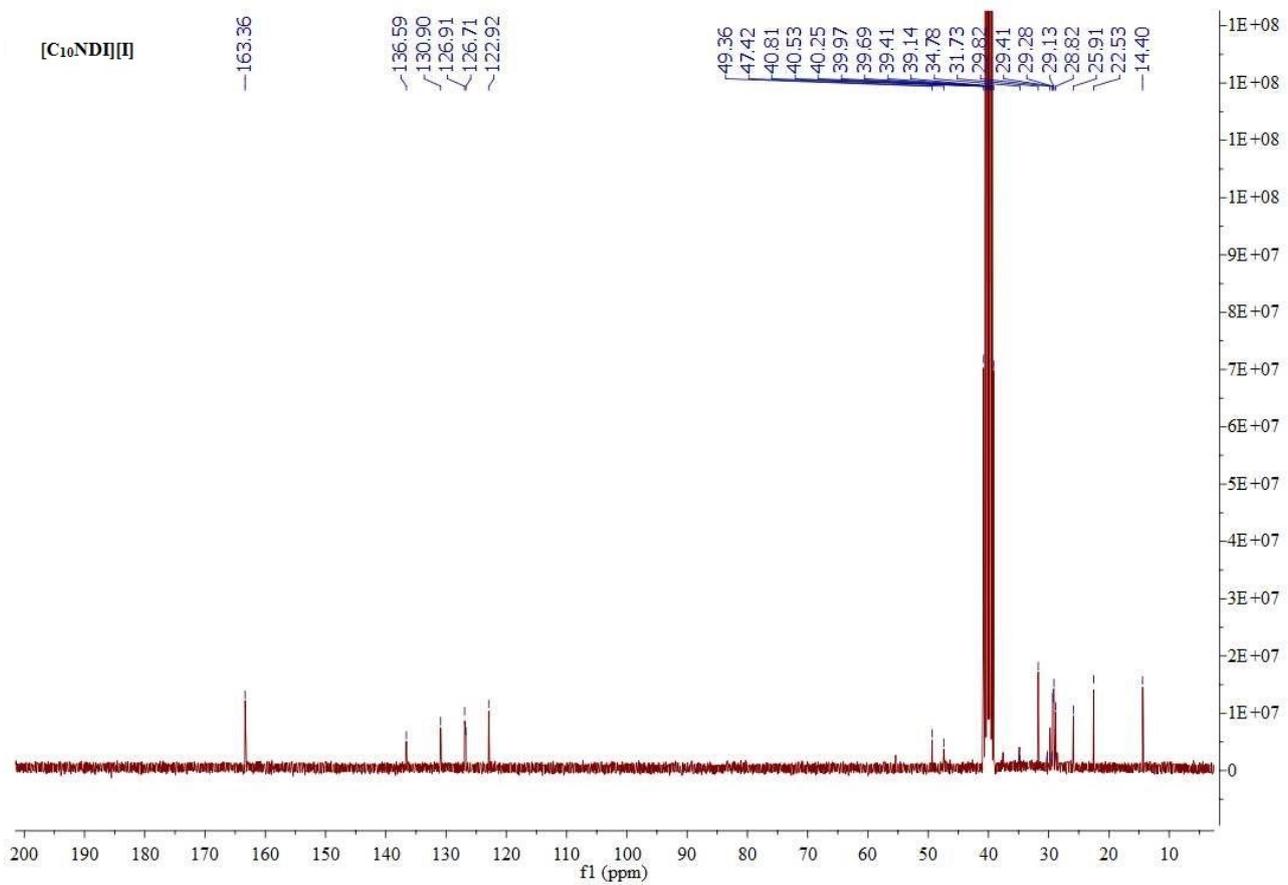
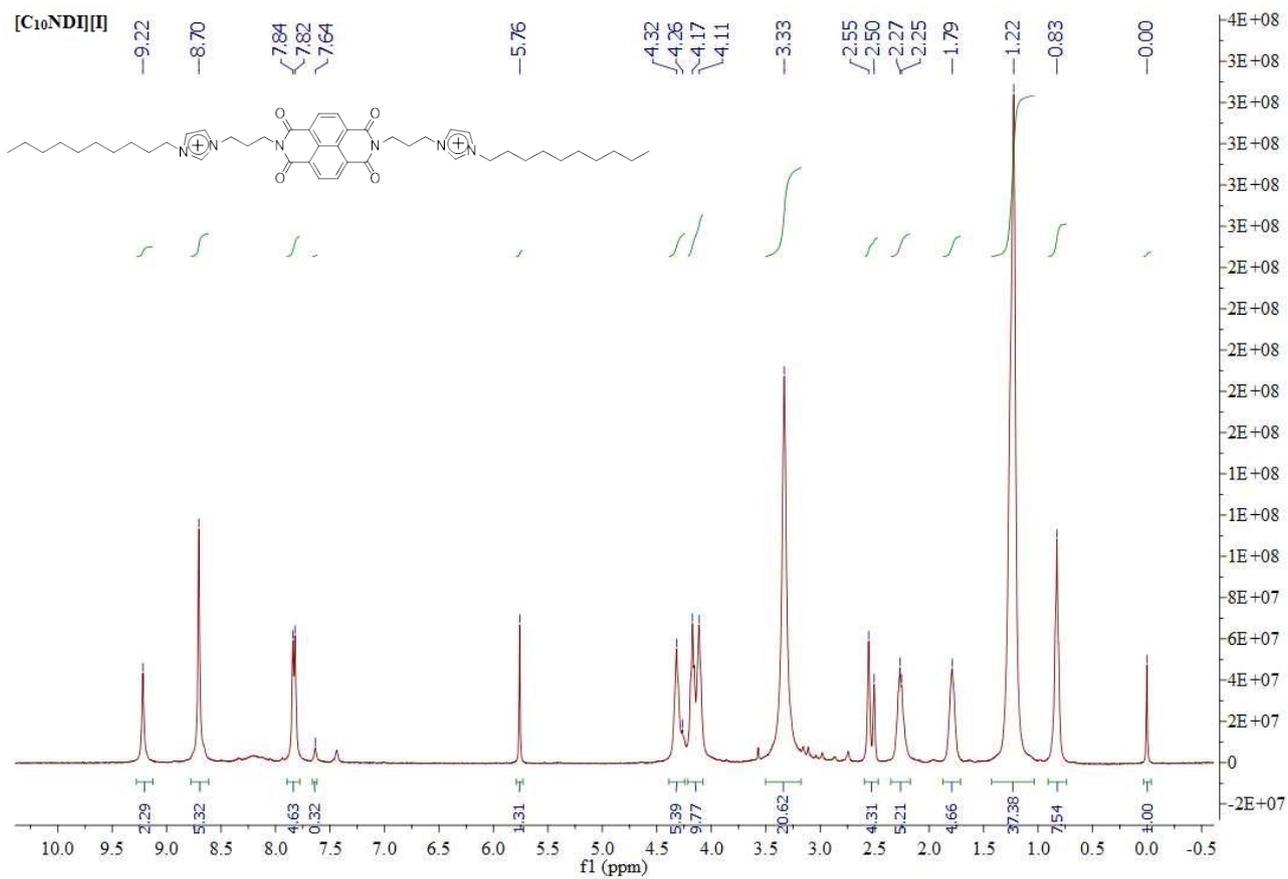
Figure S12. SEM images collected at $5 \cdot 10^{-5}$ M from casting of (a) $[C_6NDI][I]$ in THF; (b) $[C_6NDI][I]$ in DMF; (c) $[C_{12}NDI][I]$ in DMF; (d) and (e) $[C_{12}FNDI][I]$ in DMF.

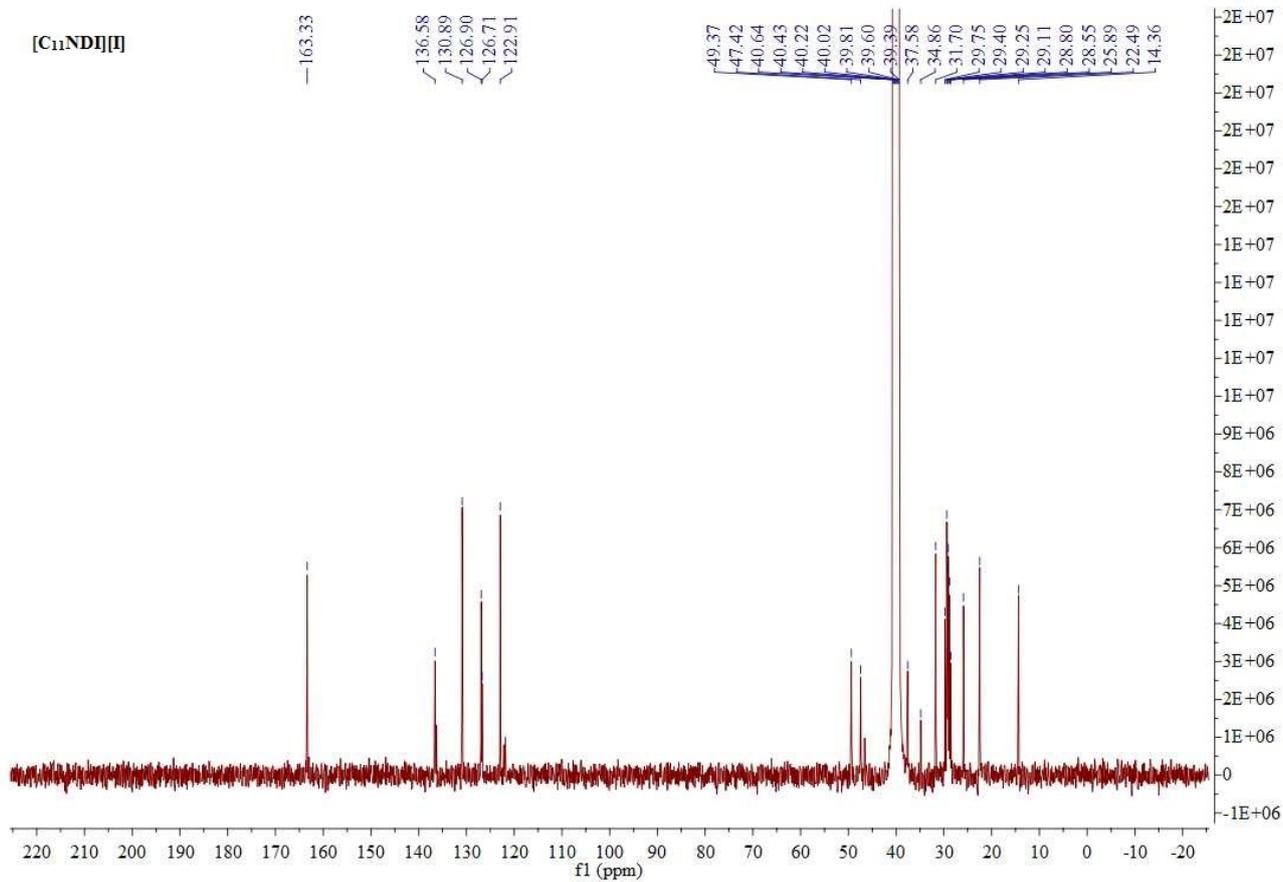
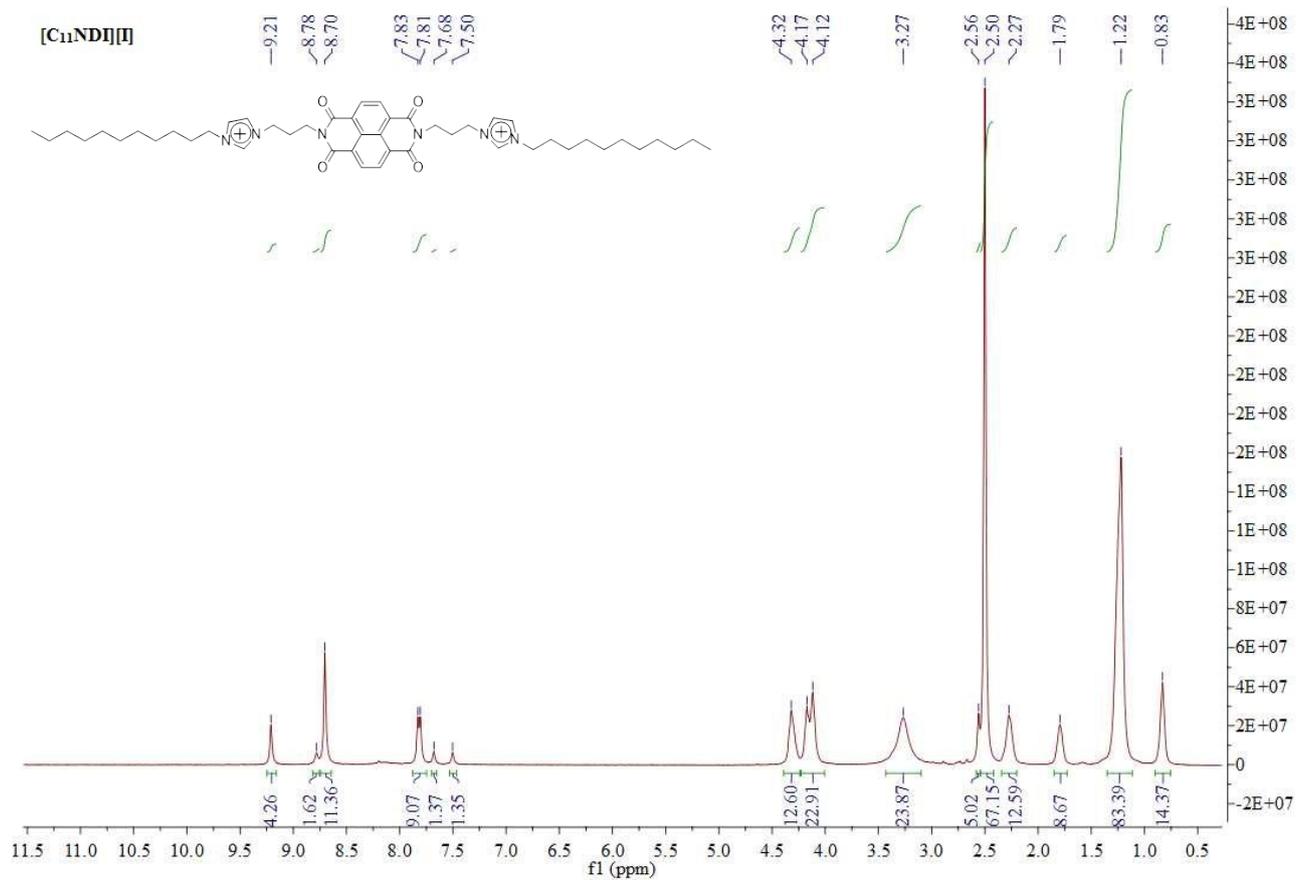


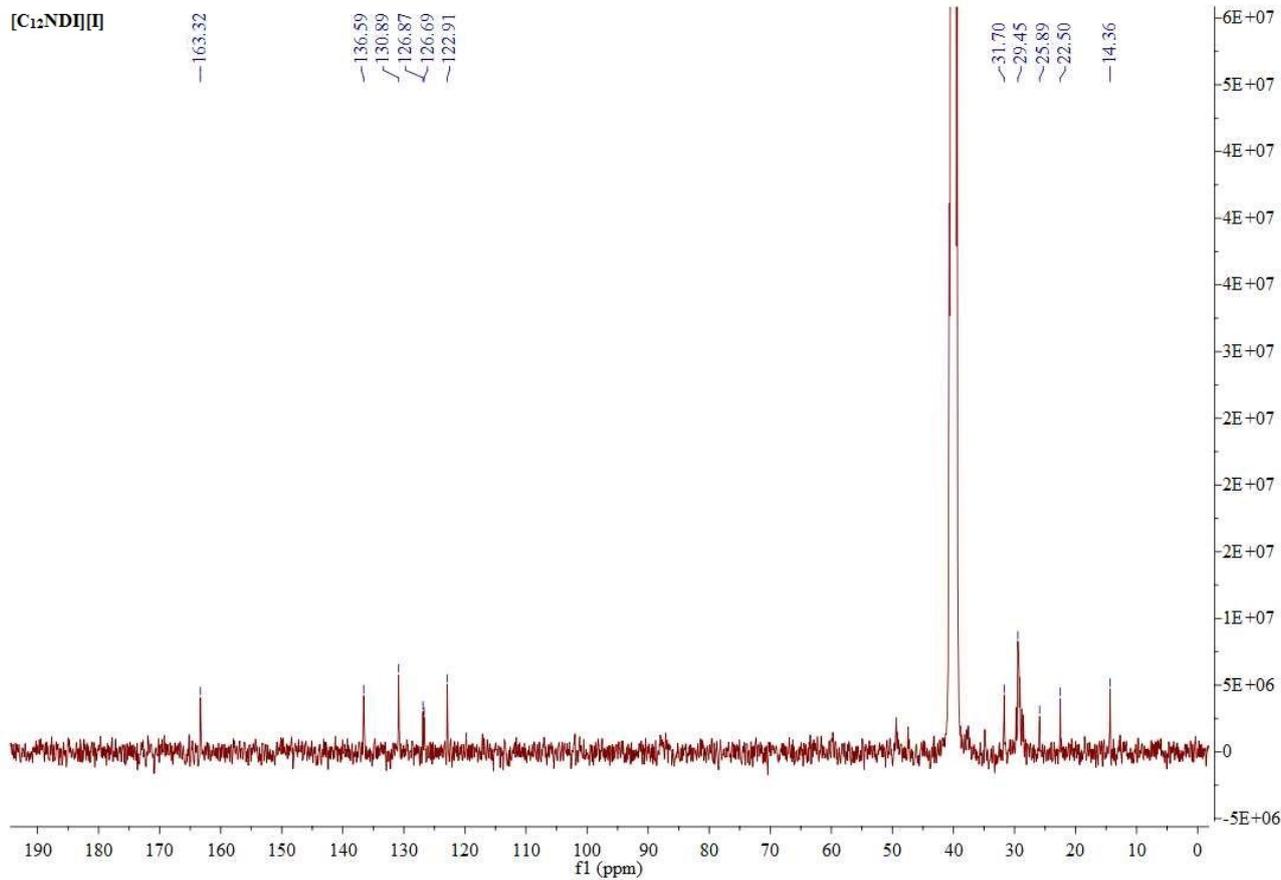
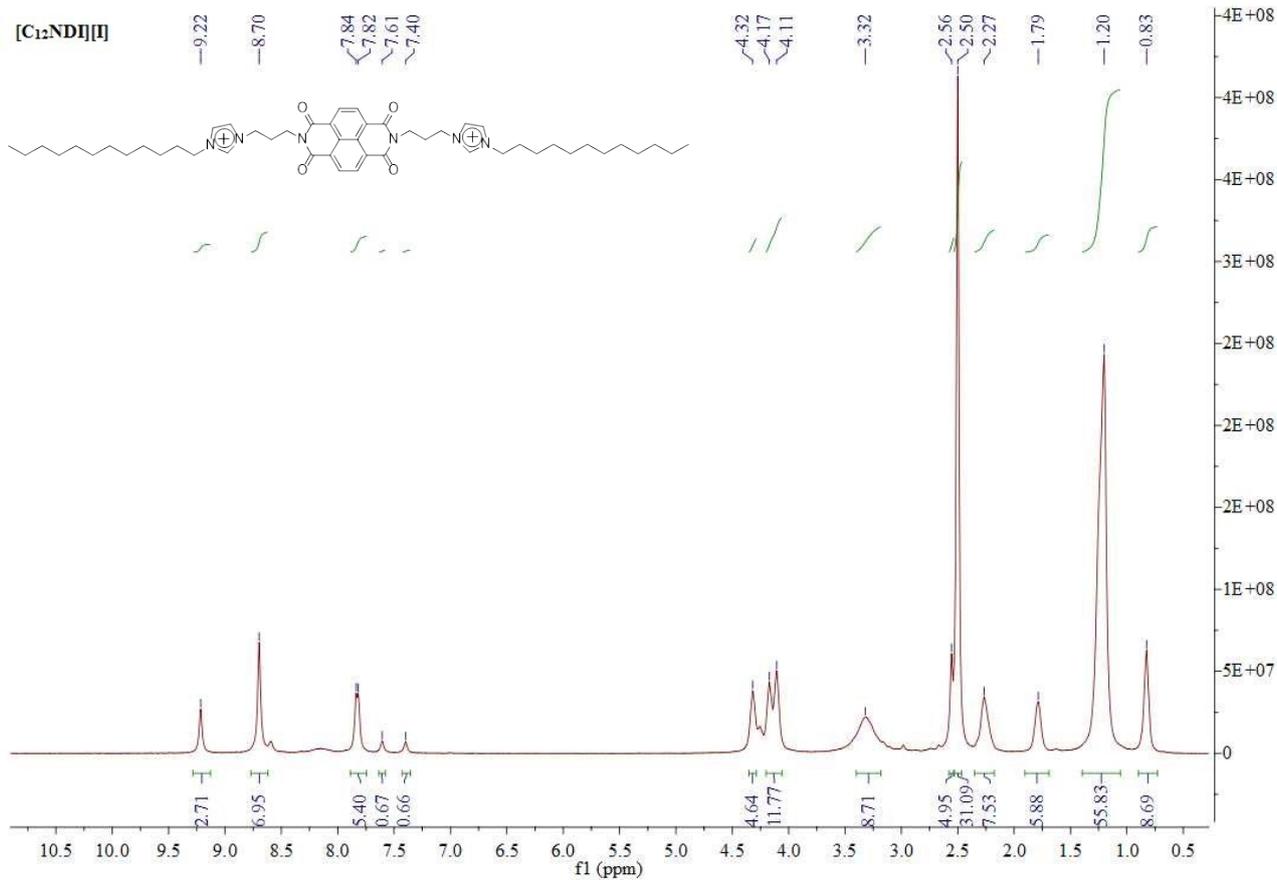












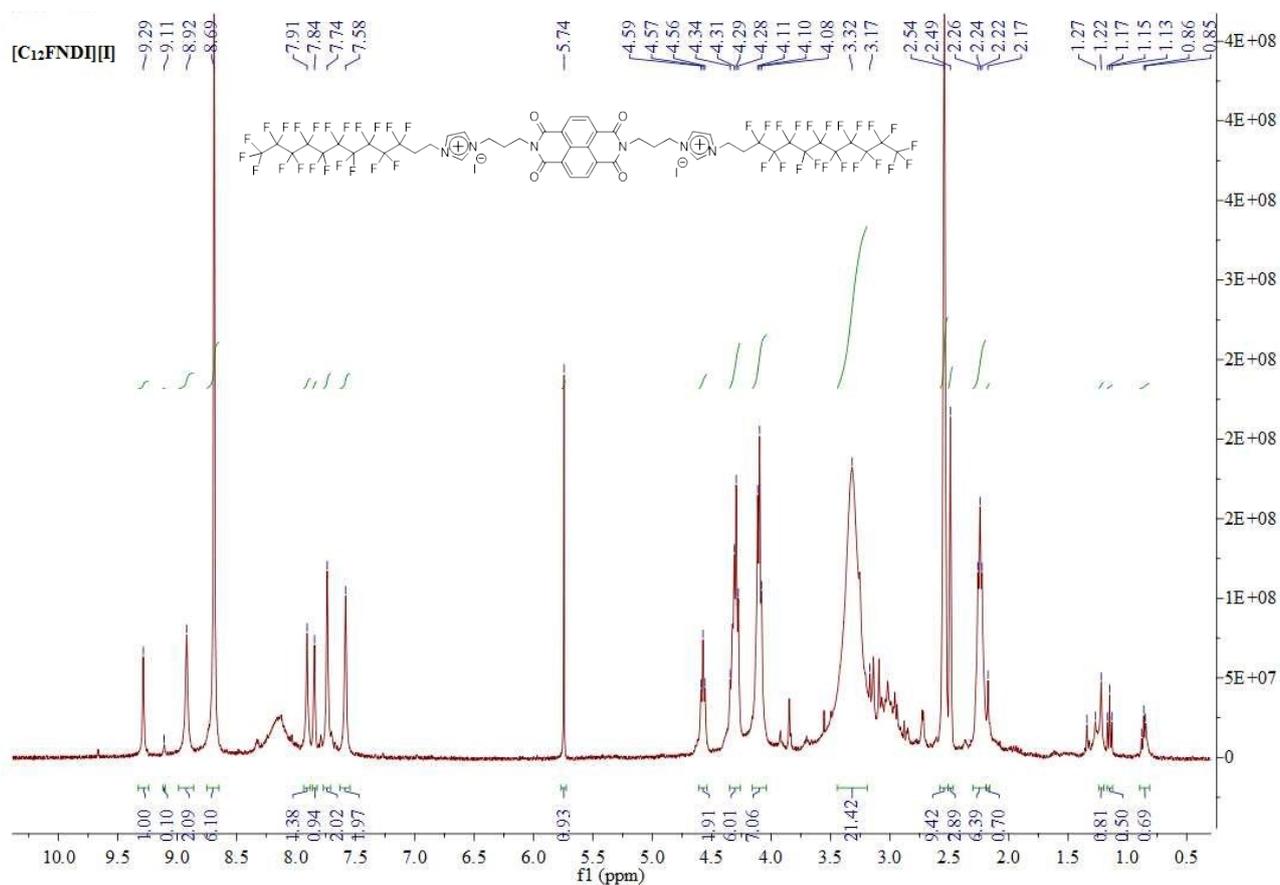
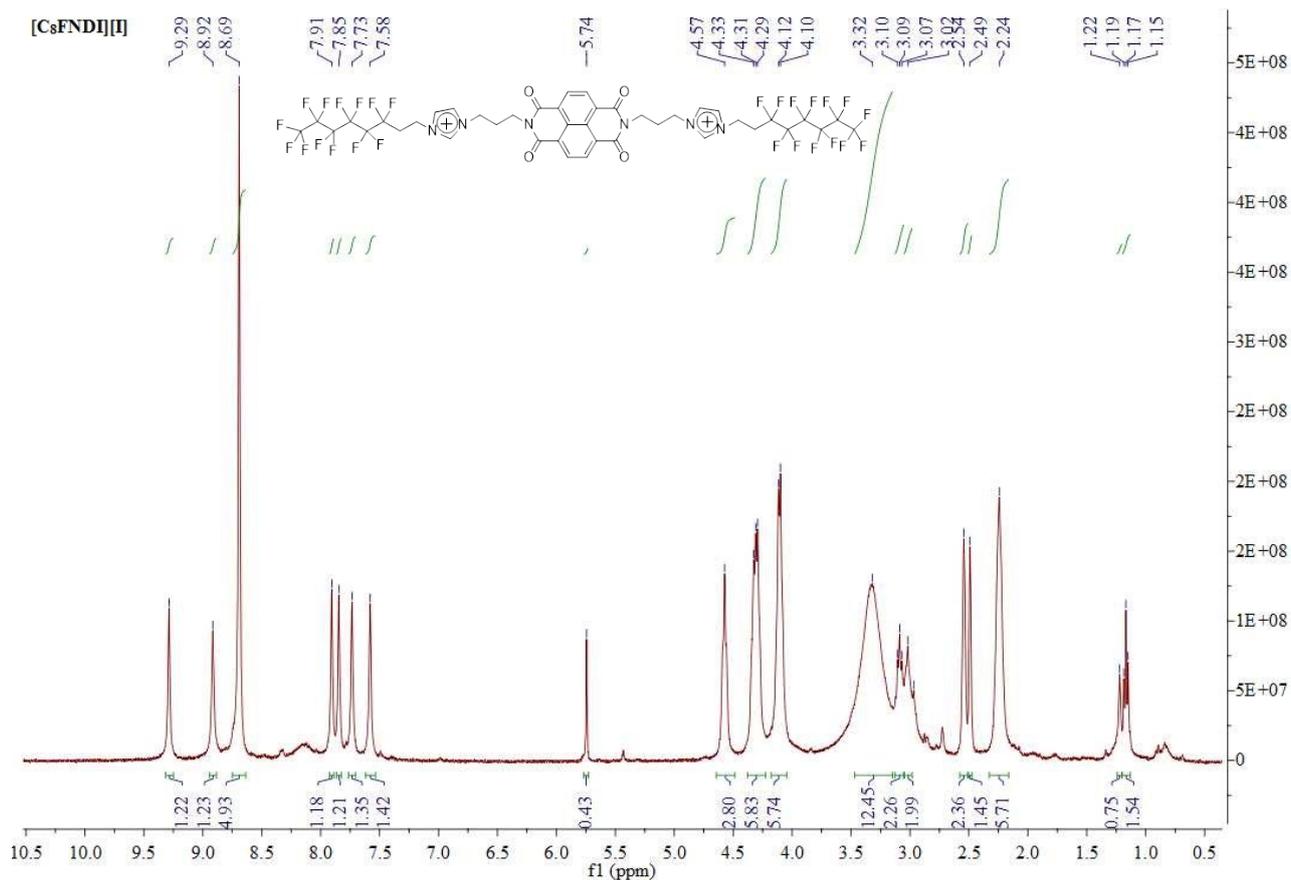
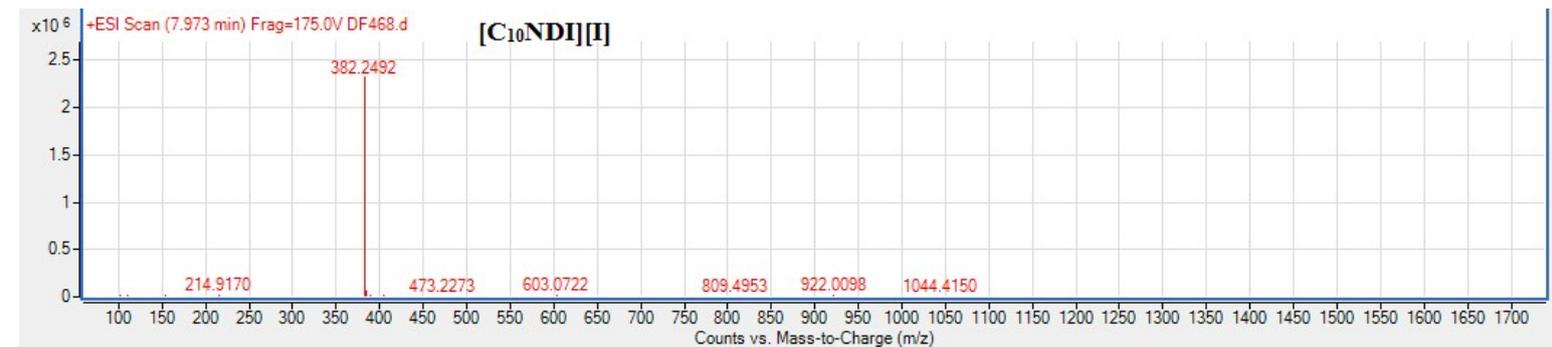
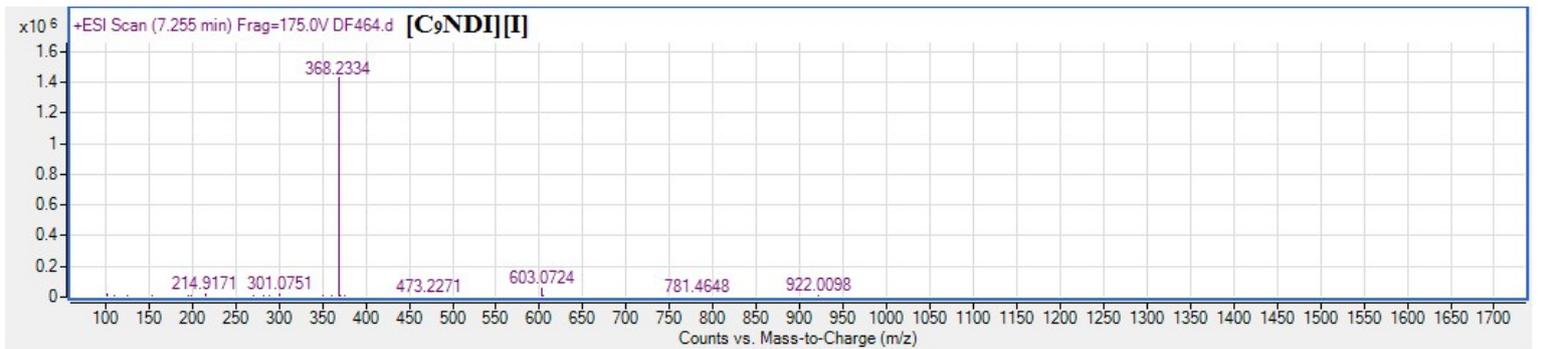
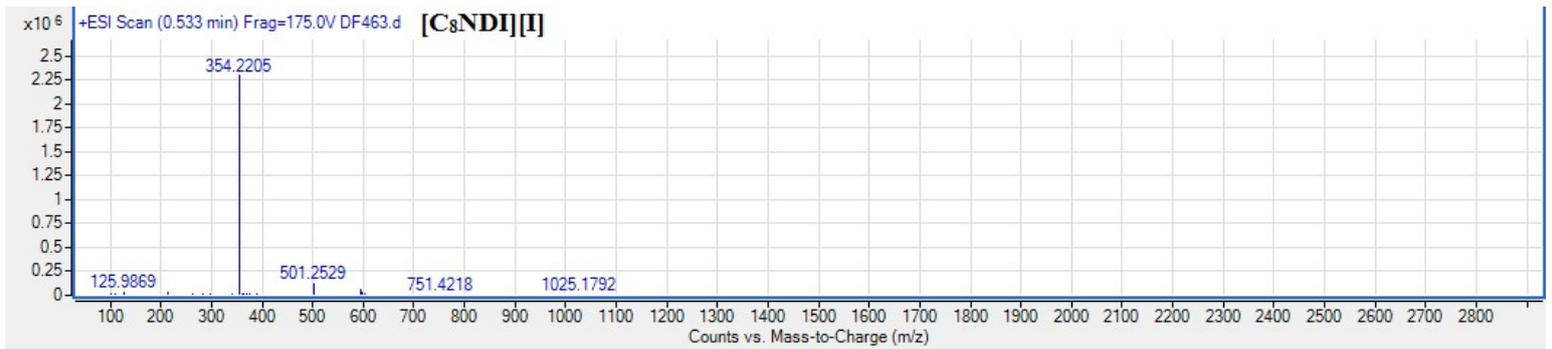
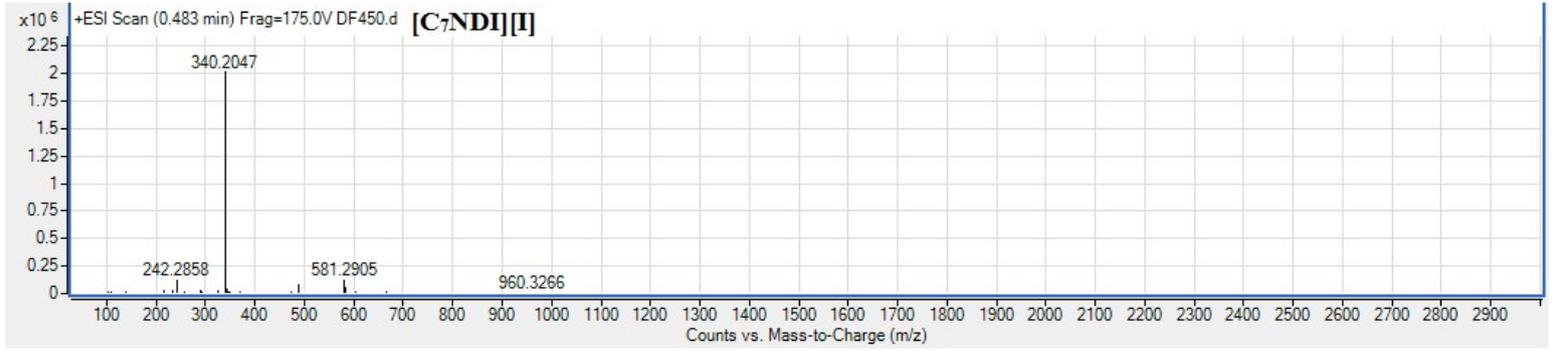
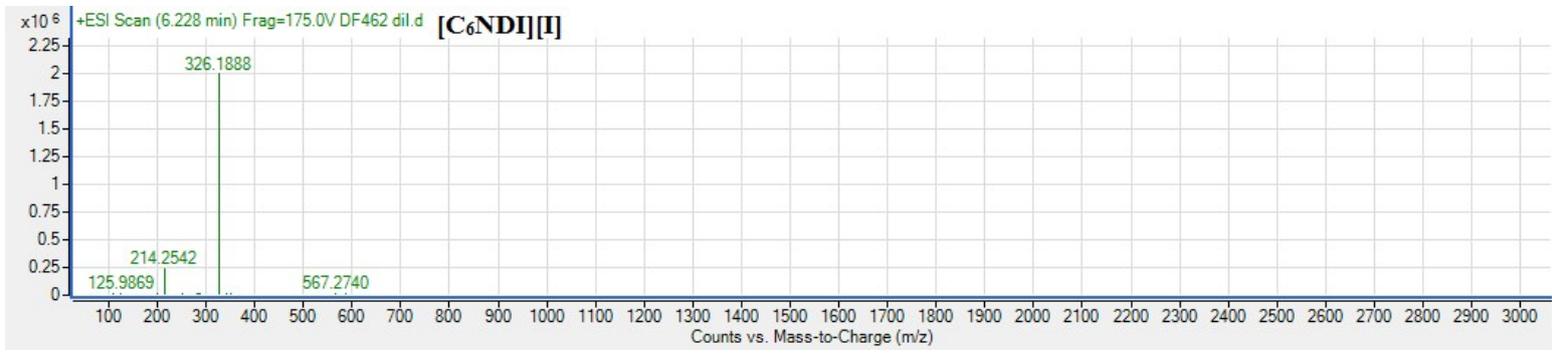


Figure S13. ¹H NMR and ¹³C NMR spectra of synthesized salts.



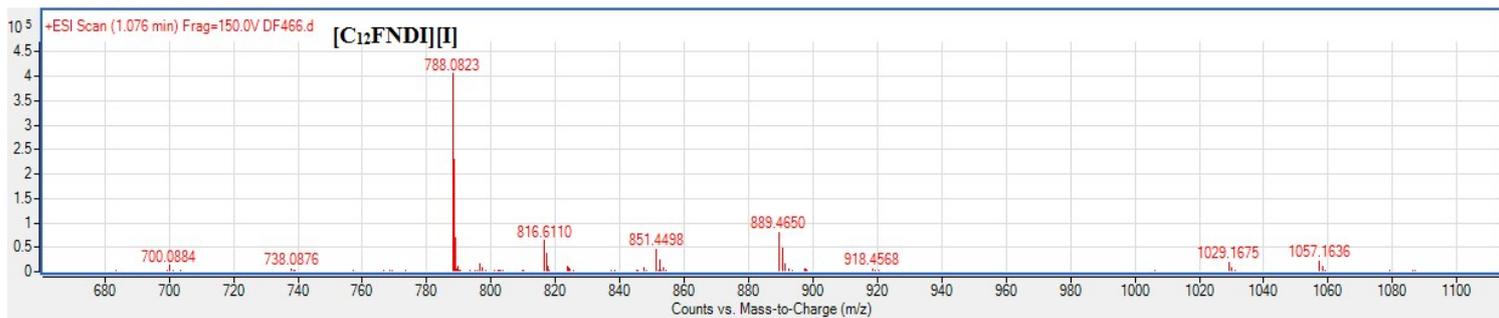
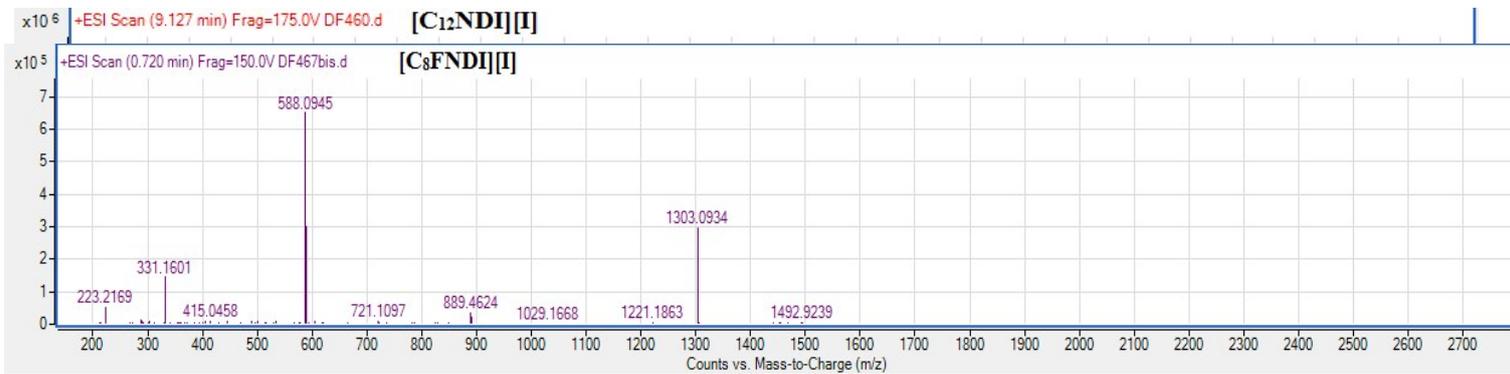
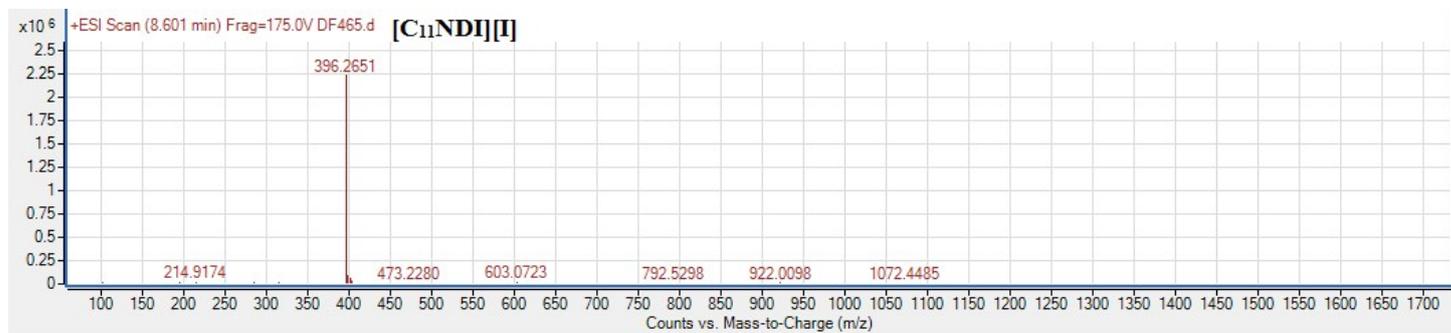


Figure S14.ESI Mass spectra of synthesized salts.