

Supporting Information

Visible light assisted improved photocatalytic activity of combustion synthesized spongy-ZnO towards dye degradation and bacterial inactivation

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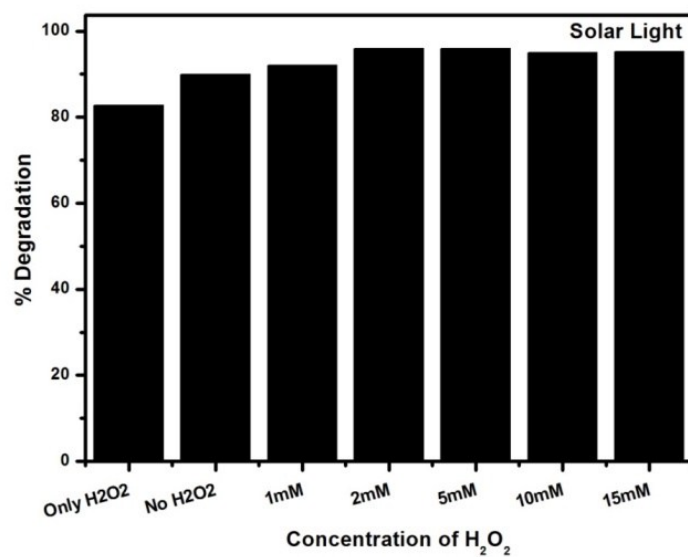


Figure S1: Effect of H_2O_2 on the degradation of Crystal Violet dye (Initial dye concentration = 20 mg L^{-1} , catalyst loading = 50 mg L^{-1} , temperature = $25 \pm 2^\circ\text{C}$, pH ~ 7 , and solar intensity = 0.768 kW m^{-2})

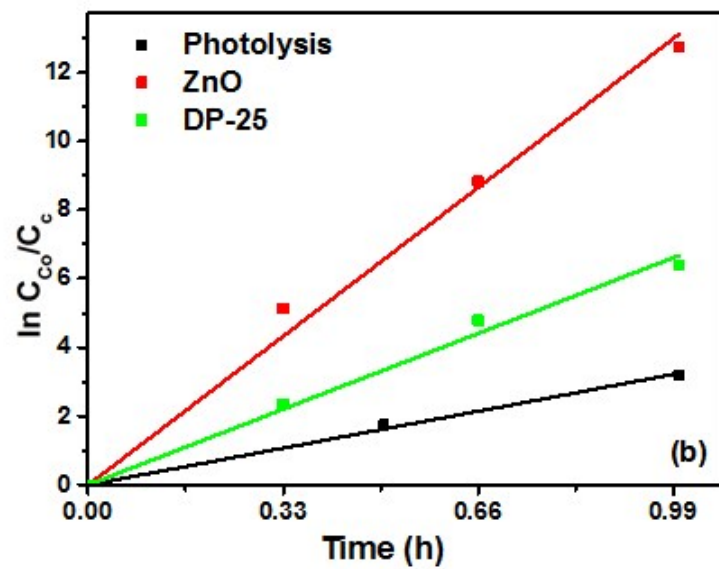


Figure S2. Pseudo first order kinetics for photolysis, spongy-ZnO and P-25 TiO₂. (Initial concentration $\sim 10^8$ CFU/ml, catalyst loading = 0.25 g/L, temperature = $25 \pm 2^\circ\text{C}$, and visible light intensity = 42800 lux)

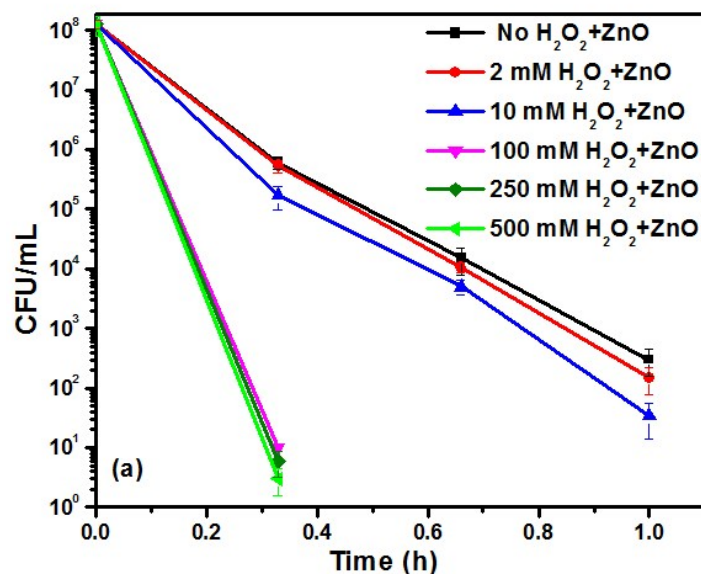


Figure S3. Effect of hydrogen peroxide on inactivation of *E.coli* under visible irradiation with initial concentration 10^8 CFU/mL for 0, 2.0, 10.0, 100.0, 250.0 and 500.0 mM concentration of H_2O_2 in presence of 0.25 g/L ZnO.

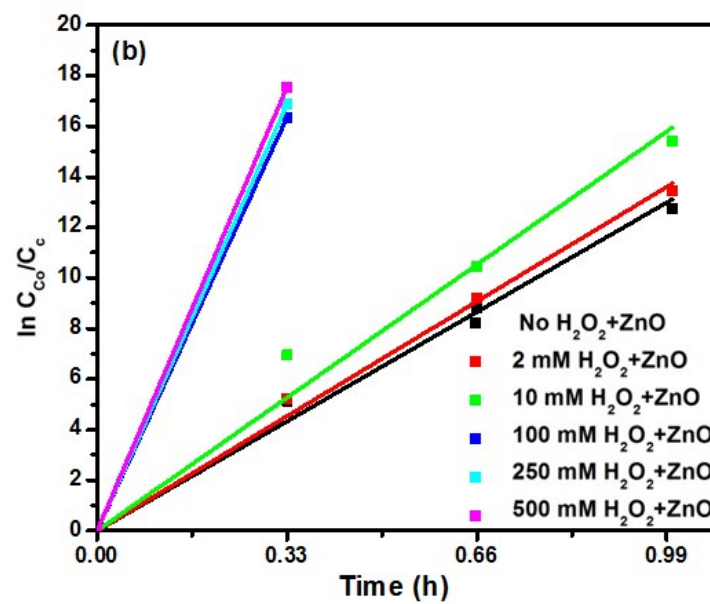


Figure S4. Pseudo first order kinetics 0, 2.0, 10.0, 100.0, 250.0 and 500.0 mM concentration of H_2O_2 in presence of 0.25 g/L ZnO.

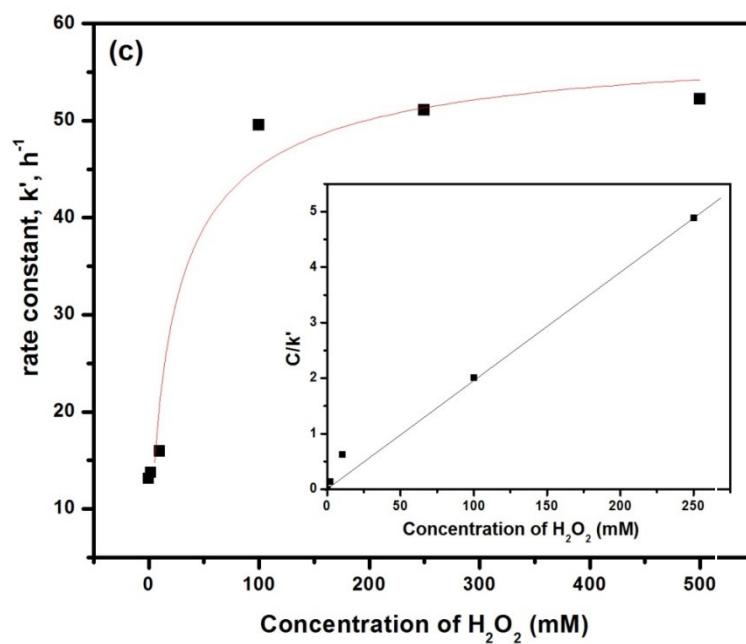


Figure S5. Variation of rate constant with H_2O_2 concentration. (The inset represents the relationship between concentration/rate constant of the reactions performed at 0, 2.0, 10.0, 100.0, 250.0 and 500.0 mM concentration of H_2O_2)