

Supplementary Information

Wet-chemical Preparation of Barium Magnesium Orthophosphate, $\text{Ba}_2\text{Mg}(\text{PO}_4)_2:\text{Eu}^{2+}$, Nanorod Phosphor with Enhanced Optical and Photoluminescence Properties

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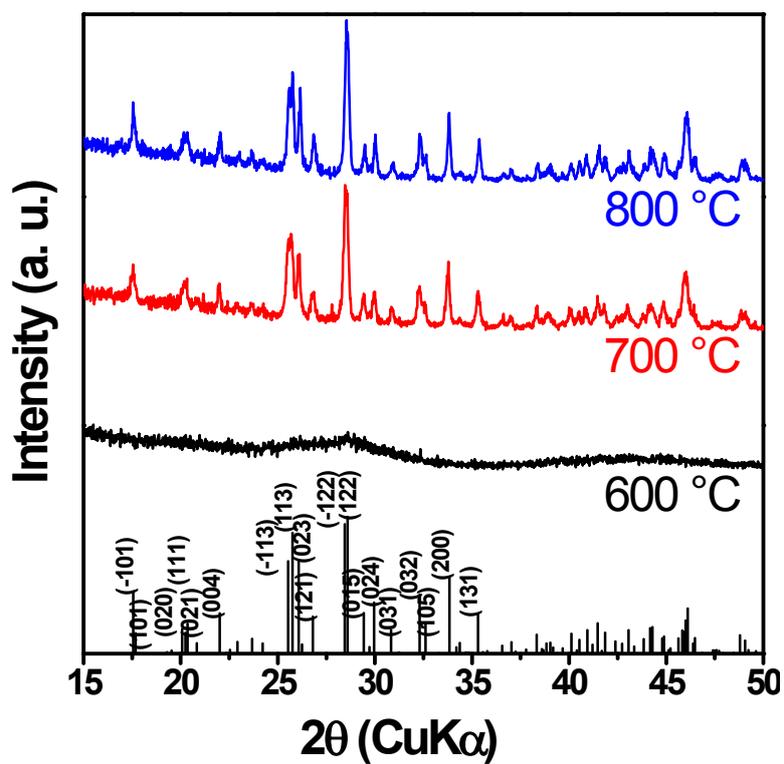


Figure S1. XRD patterns of BMP09 samples annealed at different temperatures (600, 700, and 800 °C) for 2 h, showing that a phase-pure $\text{Ba}_2\text{Mg}(\text{PO}_4)_2$ orthophosphate compound are formed at above 700 °C.

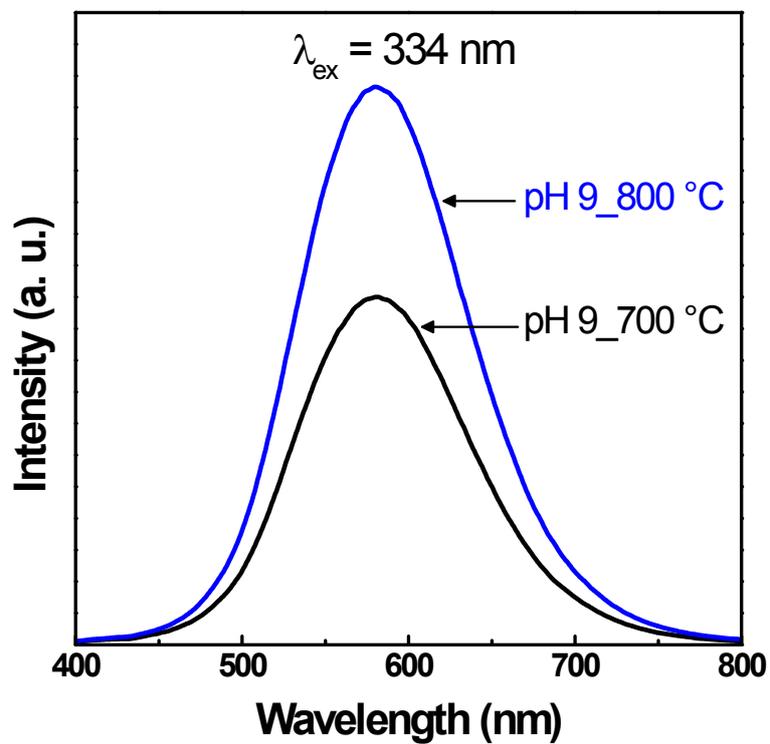


Figure S2. Photoluminescence emission spectra of BMP09 nanophosphors post-annealed at 700 °C and 800 °C in reducing atmosphere. The BMP09_800 sample shows higher PL intensity than BMP09_700 sample, which is due to its high crystallinity.

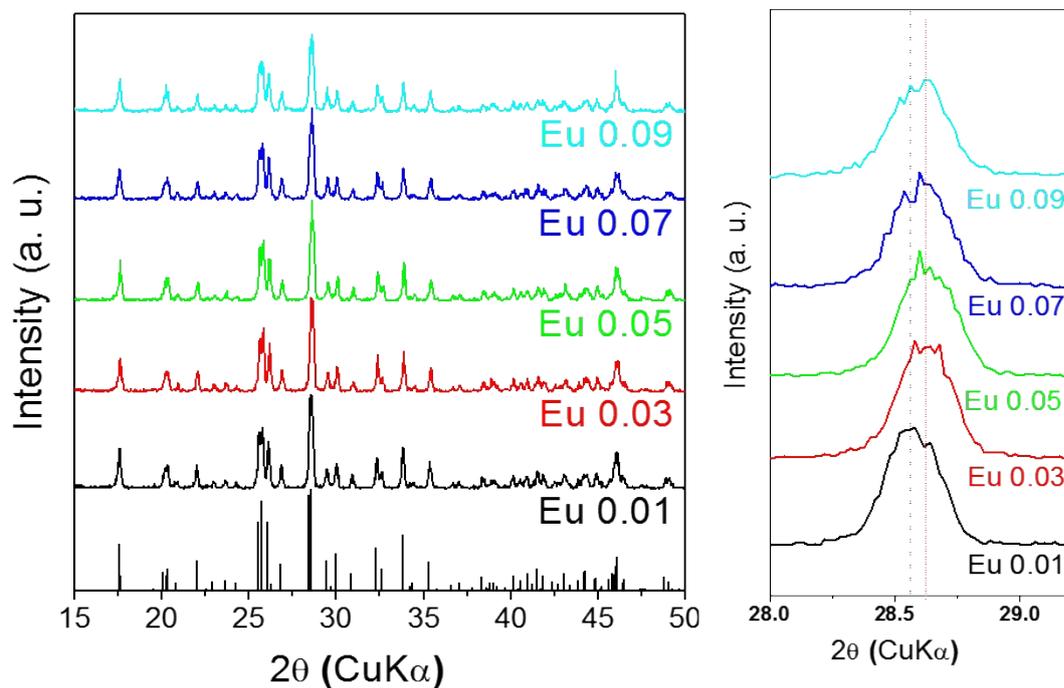


Figure S3. XRD patterns of $\text{Ba}_{2-2x}\text{Mg}(\text{PO}_4)_2:\text{Eu}^{2+}_x$; $x=0.01, 0.03, 0.05, 0.07, 0.09$, annealed at $800\text{ }^\circ\text{C}$. As increasing Eu^{2+} doping amount, the main XRD peak near 28.50 shifted toward a high angle, implying a successful incorporation of Eu^{2+} ions into BMP host matrix.

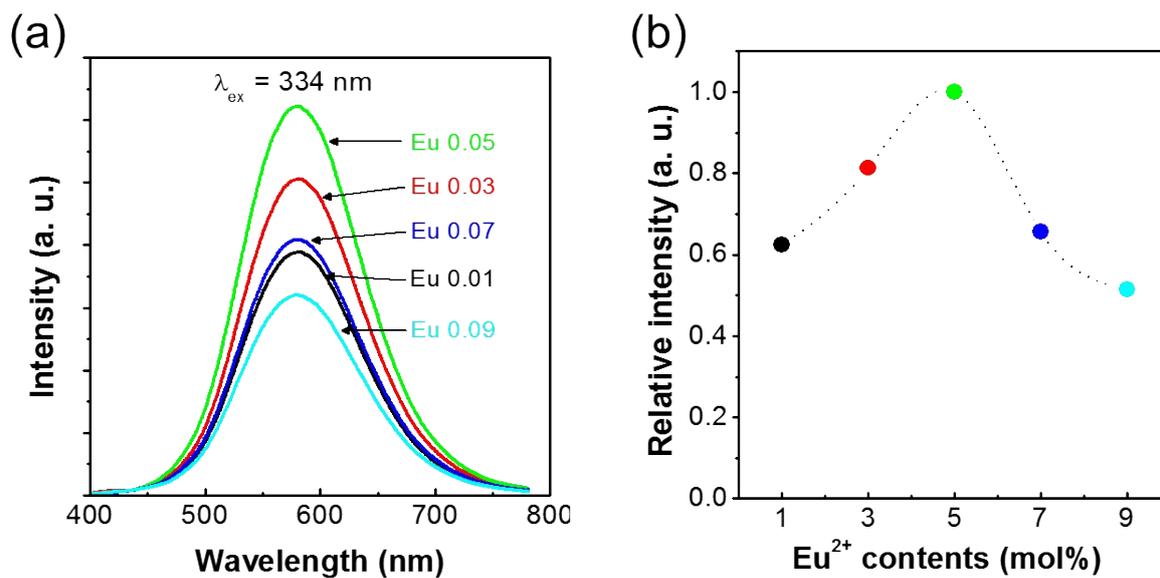


Figure S4. Photoluminescence emission spectra of micro-sized bulk BMP:Eu phosphors synthesized by a solid-state reaction method (1200 °C/2h) as a function of Eu²⁺ concentration.

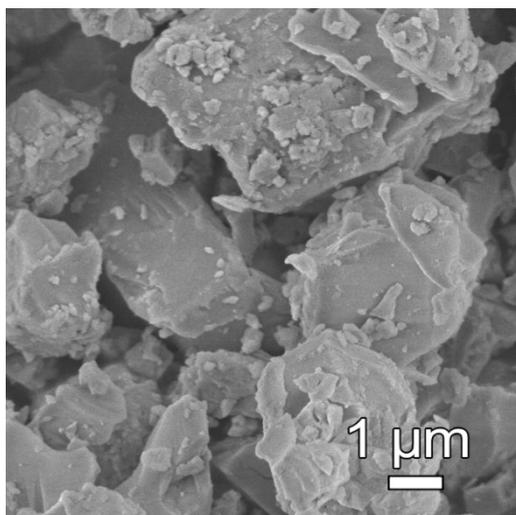


Figure S5. SEM image of micro-sized bulk BMP:Eu phosphors synthesized by a solid-state reaction method (5 mol.% of Eu^{2+} , 1200 °C/2h).

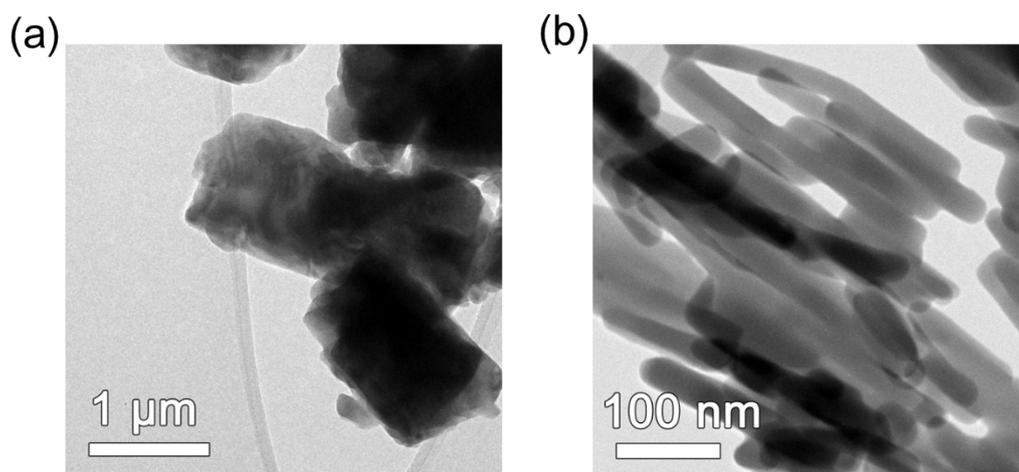


Figure S6. Comparison of TEM image of the bulk and nanorod BMP:Eu phosphors, showing irregular/rough surface for the bulk BMP:Eu and smooth/uniform surface for the nanorod BMP:Eu phosphors. This different surface roughness may affect the light absorption as well as PL intensity.