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# Modified Structure of Two-Dimensional Polythiophene Derivatives by

## Incorporating Electron-Deficient Units into Terthiophene-Vinylene Conjugated

## Side Chains and Polymer Backbone: Synthesis, Optoelectronic and Self-

## Assembly Properties, and Photovoltaic Application

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- $\bigcirc$  Dark *J-V* curves of polymer/PC<sub>61</sub>BM devices.
- ◎ Mobility of polymer with/without PC<sub>61</sub>BM
- ◎ TGA and DSC plots



Figure S1. <sup>1</sup>H-NMR spectrum of compound M2



Figure S2. <sup>13</sup>C-NMR spectrum of compound M2

Entry	Solvent	Concentration (M)	Catalyst <sup>a</sup>	Condition <sup>b</sup>	Mn (kDa) <sup>c</sup>	PDI
1	Toluene	0.01	A, 5 mol%	1	5.2 <sup>e</sup>	1.85
2	Toluene	0.01	A, 5 mol%	2	9.5	2.32
3	Toluene	0.05	A, 5 mol%	1	11.0	2.61
4	Toluene	0.05	A, 3 mol%	1	12.0	2.58
5	Toluene	0.05	A, 1 mol%	1	13.2	2.59
6	Toluene	0.05	B, 5 mol%	1	10.0	4.33
7	<i>p</i> -xylene	0.05	A, 3 mol%	2	14.0	2.36
8	<i>p</i> -xylene	0.05	B, 3 mol%	2	14.0	4.61
9	<i>p</i> -xylene	0.05	A, 3 mol%	3	12.1	2.60
10 <sup>d</sup>	Toluene	0.01	A, 5 mol%	4	12.0	3.61

<sup>a</sup>A : Pd(PPh<sub>3</sub>)<sub>4</sub>; B : Pd<sub>2</sub>dba<sub>3</sub>, P(*o*-tol)<sub>3</sub>; <sup>b</sup>Condition (1) raise temperature from r.t. to 200 °C as fast as possible; hold the temperature 30 min; cool down to 55 °C. Condition (2) raise temperature from r.t. to 200 °C as fast as possible; hold the temperature 60 min; cool down to 55 °C. Condition (3) raise temperature from r.t. to 250 °C as fast as possible; hold the temperature 30 min; cool down to 55 °C. Condition (4) reflux 2 days. °Soxhlet extractions by using methanol and hexane quickly to remove the small molecules and oligomers and finally chloroform to obtain the target compounds for optimizing polymerization conditions.  $M_n$  and PDI of the polymers were estimated by GPC using polystyrene as standards in THF. <sup>d</sup>conventional heating. °Soxhlet extractions by only using methanol to remove impurity and chloroform to obtain the target polymer.

### O PESA spectrum of polymer films



**Figure S3.** PESA spectrum of (a) **P3HT**; (b) **P1**; (c) **P2**; (d) **P3** film prepared by spin-coating followed by thermal annealing at 120°C for 15 min and measured under identical condition.

(c) P3 (a) P1 (b) P2  $q_z(nm^{-1})$  $q_z(nm^{-1})$  $q_z(nm^{-1})$  $q_{xy}(nm^{-1})$ 20 20 15 10 15 20  $q_{xy}(nm^{-1})$  $q_{xy}(nm^{-1})$ 200 1200 1400 400 800 1000 Intensity

<sup>(C)</sup> Two-dimensional grazing incidence X-ray diffraction (GIXRD) of pristine polymer thin film.

Figure S4. Two-dimensional grazing incidence X-ray diffraction (GIXRD) from thin films of (a) P1; (b)P2; (c) P3 prepared by drop-cast followed by thermal annealing at 120°C for 15 min and measured under identical condition.

 $\bigcirc$  Dark *J-V* curves of polymer/PC<sub>61</sub>BM devices.



**Figure S5.** Dark J-V curves of polymer/PC<sub>61</sub>BM devices.

 $\bigcirc$  Mobility of polymer with/without PC<sub>61</sub>BM.



**Figure S6.**  $\ln(J_{dark}L^3V^{-2})$  versus  $(VL^{-1})^{0.5}$  plots of (a) the pristine polymers for the measurement of hole mobility; the polymers blend PC<sub>61</sub>BM for the measurement of (b) hole and (c) electron mobility by the SCLC method.

Table S2.	Mobility of <b>P1</b> ,	P2 and P3	with/without PC <sub>6</sub>	<sub>1</sub> BM by	the SCLC method.
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	Pristine polymer hole mobility (cm^2/(V*sec))	Blend with PC <sub>61</sub> BM hole mobility (cm^2/(V*sec))	Blend with PC <sub>61</sub> BM electron mobility (cm^2/(V*sec))	h+/e-
P1	3.9×10 <sup>-4</sup>	2.1*10-4	2.5*10-4	0.84
P2	3.8×10 <sup>-5</sup>	1.1x10-6	3.5x10-4	0.003
Р3	1.0×10 <sup>-4</sup>	8.4x10-5	3.3x10-4	0.25

 $\bigcirc$  TGA and DSC.



Figure S7. (a) TGA and (b-d) DSC second heating profiles of P1, P2, and P3 with a heating rate of 10  $^{\circ}$ C/min under N<sub>2</sub> atmosphere and a cooling rate of 10  $^{\circ}$ C/min.