**Supporting Information for** 

## Direct conversion mechanism from BiOCl nanosheets to BiOF, Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub> and BiF<sub>3</sub> in the presence of fluorine resource

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Table S1 The elemental composition of the  $Bi_7F_{11}O_5/BiOCl$  sample from EDS analysis.

wt%	at%
7.18	30.41
9.85	35.14
4.77	9.11
78.20	25.35
	9.85 4.77



Fig. S1 The adsorption curves of the samples.



**Fig. S2** (a) XRD patterns of the samples prepared at different reaction time with NH<sub>4</sub>F ( $R_F=2$ ,  $V_{Eth}:V_w=18:2$ ); (b) The intensity ratios of the strongest peaks for Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub> and BiOCl.



Fig. S3 The EDS of the  $Bi_7F_{11}O_5/BiOCl$  sample.





**Fig. S4** SEM images of the untreated BiOCl and treated samples by NH<sub>4</sub>F at different molar ratios of F to Bi ( $R_F$ ): (a) untreated BiOCl; (b)  $R_F=0$  (BiOCl); (c)  $R_F=0.5$  (BiOF/BiOCl); (d)  $R_F=1$  (BiOF/Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub>/BiOCl); (e)  $R_F=2$  (Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub>/BiOCl); (f)  $R_F=4$  (Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub>); (g)  $R_F=6$  (Bi<sub>7</sub>F<sub>11</sub>O<sub>5</sub>/BiF<sub>3</sub>); (h)  $R_F=8$  (BiF<sub>3</sub>). Preparation conditions:  $V_{Eth}/V_w=18/2$  (volumetric ratio of ethanol to water)



Fig. S5 (a) Degradation curves of MO over commercial P25; (b) UV-vis absorption spectra of 2-nitrophenol (10 mg/L) over  $Bi_7F_{11}O_5/BiOCl$  sample at different reaction times.



Fig. S6 SEM images of the samples treated using different fluorine sources: (a) NH<sub>4</sub>F; (b) NaF; (c) HF. Preparation conditions:  $R_F = 2$ ;  $V_{Eth}/V_w = 18/2$ .





Fig. S7 SEM images of the samples prepared with NH<sub>4</sub>F ( $R_F$ =2) at different volumetric ratios of ethanol to water ( $V_{Eth}/V_w$ ): (a) 0/20; (b) 2/18; (c) 10/10; (d) 18/2; (e) 20/0.