

Supporting Information

TiO₂-SiO₂ Janus Particles with highly enhanced photocatalytic activity

Kamlesh Panwar, Manjeet Jassal*, Ashwini K. Agrawal*

SMITA Research Lab, Indian Institute of Technology,

Hauz Khas, New Delhi 110016, India

*Corresponding Author. Tel.: +91 11 2659 1426; E-mail: ashwini@smita-iitd.com, manjeetjassal@gmail.com.

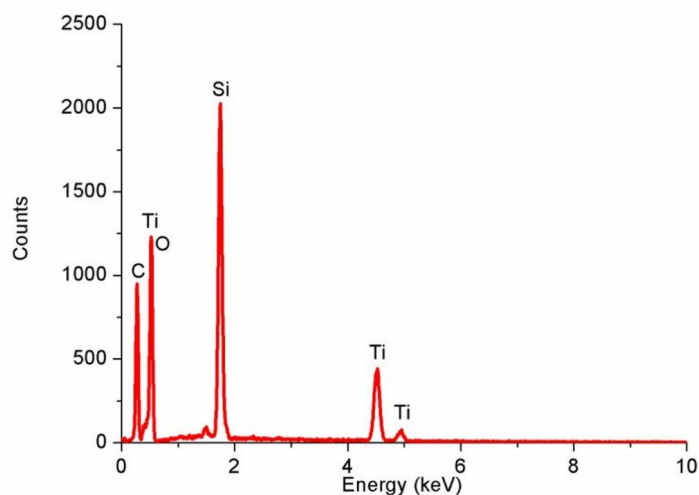


Figure S1. EDX analysis of SiO₂-TiO₂ core-shell particles.

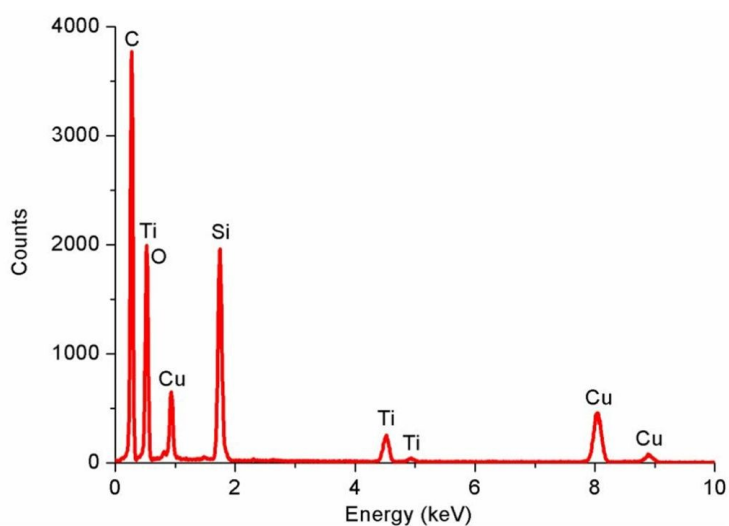


Figure S2. EDX analysis of TiO₂-SiO₂ Janus particles.

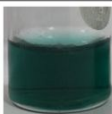
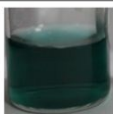
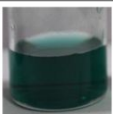
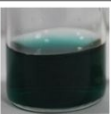









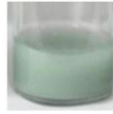
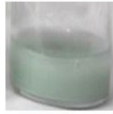





Samples	t=0 h	t=1 h	t=2 h	t=4.5 h
Control SG dye				
P25 TiO ₂				
ST- CS@450				
Janus@ 450				
Janus@ 750				

Figure S3. Images of Solophenyl green dye solutions containing different TiO₂ nanoparticles with varying duration of UV exposure.

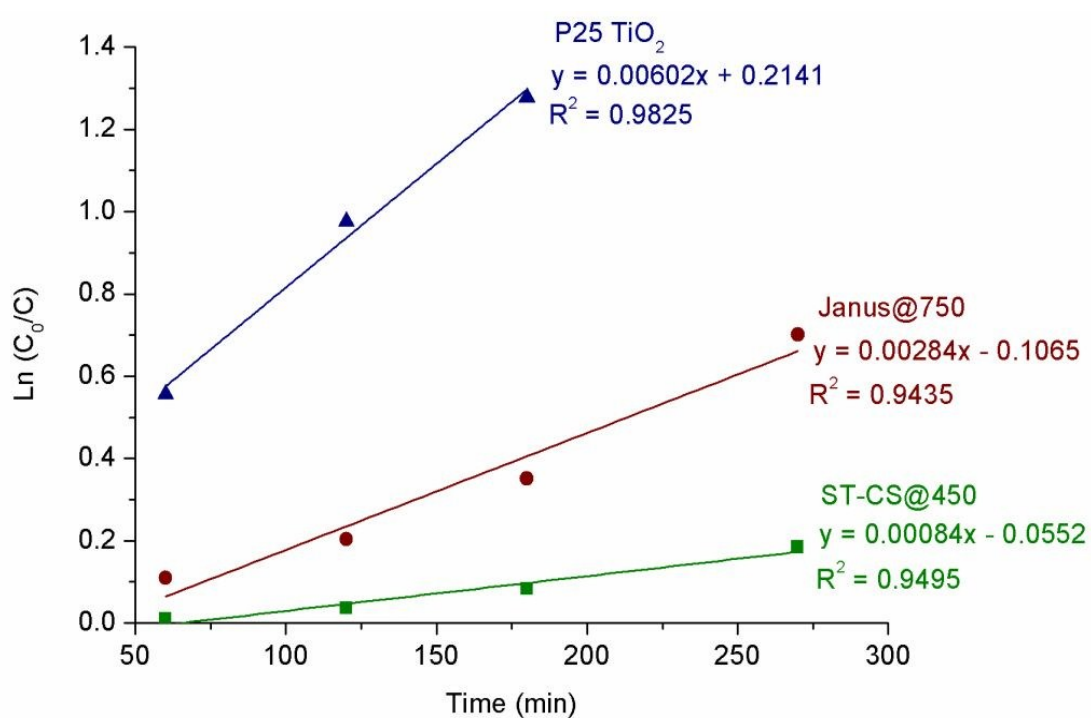


Figure S4. The kinetic curves for degradation of Solophenyl green dye using P25 TiO₂, ST-CS@450 and Janus@750 at the TiO₂ concentration of 50 mg/L.

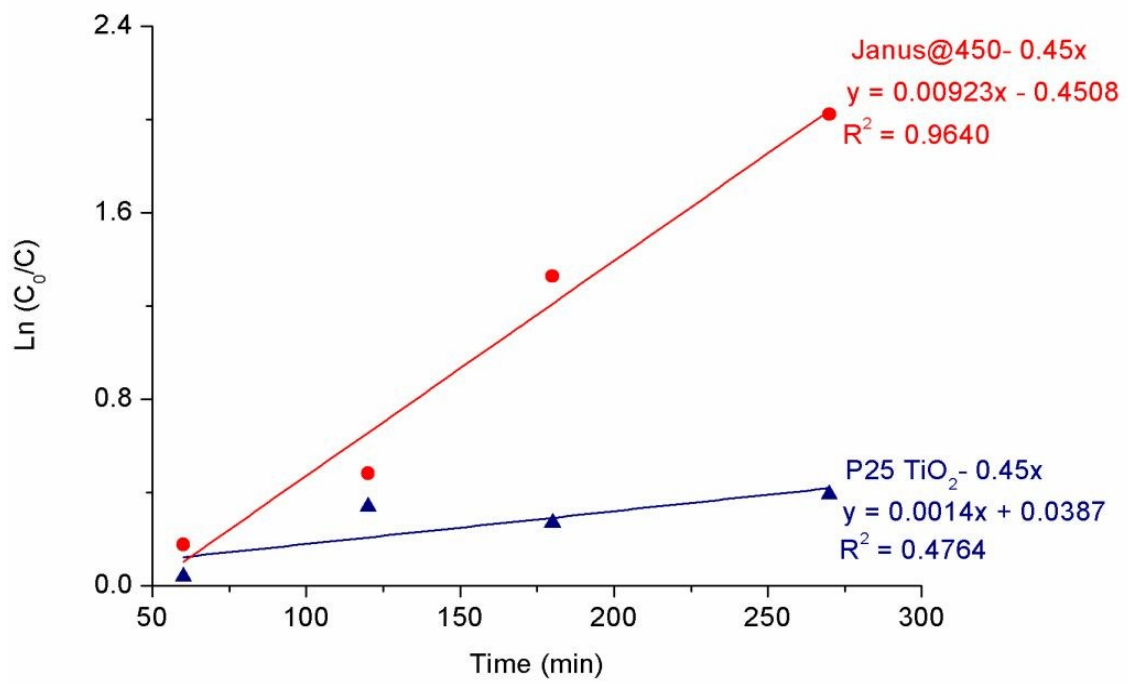


Figure S5. The kinetic curves for degradation of Solophenyl green dye using P25 TiO₂ and Janus@450 at TiO₂ concentration of 22.5 mg/L (0.45 times).