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Supporting materials for:

## Facile preparation of novel quaternary g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> nanocomposites: Magnetically separable visible-light-driven photocatalysts with significantly enhanced activity

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Fig. 1S. FT-IR spectra for the g-C<sub>3</sub>N<sub>4</sub>, g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>, and g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> (30%) samples.



Fig. 2S. TGA curves for the g-C<sub>3</sub>N<sub>4</sub>, g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>, and g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> nanocomposites with different weight percents of  $Bi_2S_3$  along with the pure AgI and  $Bi_2S_3$  samples.









Fig. 3S. (a) The degradation rate constants of RhB over the g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> (30%) nanocomposite prepared at different refluxing times. (b) SEM image of the g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> (30%) nanocomposite prepared by refluxing for 4 h. (c) The degradation rate constants of RhB over the g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> (30%) nanocomposite calcined at different temperatures. (d) SEM image of the g-C<sub>3</sub>N<sub>4</sub>/Fe<sub>3</sub>O<sub>4</sub>/AgI/Bi<sub>2</sub>S<sub>3</sub> (30%) nanocomposite calcined at 500 °C for 2 h.



Fig. 4S. The degradation rate constants of RhB over the  $g-C_3N_4/Fe_3O_4/AgI/Bi_2S_3$  (30%) nanocomposite in presence of various scavengers.



Fig. 5S. Effect of solution pH on the degradation rate constant of RhB over the g-

 $C_3N_4/Fe_3O_4/AgI/Bi_2S_3\,(30\%)$  nanocomposite.