#### **Supplementary Information**

# Title: Photolysis and photocatalytic decomposition of sulfamethazine antibiotics in an aqueous solution with TiO<sub>2</sub>.

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Fig. S1 The kinetics of photo-catalytic decomposition of 0.072 mM SMT (PCDS) on  $TiO_2$  at pH 3, 5.5, and 10. The insert was the simulation of PCDS using the PFO model.



**Fig. S2** Adsorption of 0.072 mM sulfamethazine (SMT) on 0.5 g/L TiO<sub>2</sub>, and the contributions of self-photolysis and the photo-catalytic decomposition of SMT (PCDS) to the overall SMT photo-decomposition at pH 3, 5.5, and 10.



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**Fig. S4** Effects of (a) TiO2 loading on dissolved organic carbon (DOC) contents, and (b) SMT initial concentration on sulfamethazine (SMT) photo-decomposition. The inset in (b) was the simulation results of SMT photo-decomposition using the PFO model.



**Fig. S5** Effects of sodium iodide (NaI) and isopropanol (ISP) on 0.072 mM sulfamethazine (SMT) photo-decomposition in the presence of 0.5 g/L TiO<sub>2</sub> at pH 5.5. The inset was the simulation results of SMT photo-decomposition with NaI and ISP using the PFO model. The "normal system" represented the system without the addition of any radical scavengers.



Fig. S6 HPLC spectra of the photo-decomposition of sulfamethazine (SMT) and the major intermediates. Sample was extracted after photo-decomposition of 0.072 mM SMT in a system with 0.5 g/L TiO<sub>2</sub> at pH 5.5.

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Compounda	Retention time	ESI+	Ε	ESI-	
Compounds	(min)	(m/z)	(m/z)		
SMT	15.565	279	277		
<sup>a</sup> Int. 1	14 667	295	293		
Int. 2	14.007	<sup>b</sup> n.d.		156	
Int. 3	14 221	295	293		
Let 4	14.221			122	
IIIt. 4	12.400	124			
Int. 5		311	309		
Int. 6	13.483	n.d.		172	
Int. 7		n.d.		108	
Int. 8	4.317	140	n.d.		

Table S1 HPLC analyses of 0.072 mM sulfamethazine (SMT) and its major photodecomposition products.

 $<sup>^{\</sup>rm a}$  Int. indicated the intermediates of sulfamethazine photo-decomposition  $^{\rm b}$  n.d. indicated the not detected