

Electronic supplementary information

From mononuclear to linear one-dimensional coordination species of copper(II)-chloranilate: design and characterization

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Table S1 Geometric parameters of copper(II) coordination spheres (\AA , $^\circ$) in compounds **1–6**. Symmetry operators: *i*) $-x, -y, 1-z$; *ii*) $-1-x, -y, 1-z$; *iii*) $1+x, y, z$; *iv*) $1-x, y, 1/2-z$; *v*) $x, 1+y, z$; *vi*) $1-x, 1+y, 1/2-z$; *vii*) $-1+x, y, z$; *viii*) $1-x, -y, 1-z$; *ix*) $1-x, y, 1/2-z$; *x*) $1-x, 1+y, 1/2-z$; *xi*) $1/2-x, 3/2-y, -z$.

1	2	3	
Cu1–O1	1.9999(17)	Cu1–O1	1.9673(17)
Cu1–O1 ^{viii}	1.9999(17)	Cu1–O2	1.9652(16)
Cu1–O2	1.9444(19)	Cu1–O3	1.9333(17)
Cu1–O2 ^{viii}	1.9444(19)	Cu1–O4	1.9453(18)
Cu1–O5	2.418(2)	Cu1–O7	2.2799(18)
O2 ^{viii} –Cu1–O2	180	O2–Cu1–O1	82.69(7)
O2 ^{viii} –Cu1–O1 ^{viii}	83.23(7)	O3–Cu1–O1	96.66(7)
O2 ^{viii} –Cu1–O1	96.77(7)	O4–Cu1–O1	169.92(7)
O2 ^{viii} –Cu1–O5 ^{viii}	91.85(9)	O1–Cu1–O7	91.03(7)
O2 ^{viii} –Cu1–O5	88.15(9)	O3–Cu1–O2	172.01(7)
O2–Cu1–O1 ^{viii}	96.77(7)	O4–Cu1–O2	94.62(7)
O2–Cu1–O1	83.23(7)	O2–Cu1–O7	97.19(7)
O2–Cu1–O5 ^{viii}	88.15(9)	O3–Cu1–O4	84.65(7)
O2–Cu1–O5	91.85(9)	O3–Cu1–O7	90.78(7)
O1 ^{viii} –Cu1–O1	180	O4–Cu1–O7	98.95(7)
O1 ^{viii} –Cu1–O5 ^{viii}	89.51(8)		
O1 ^{viii} –Cu1–O5	90.49(8)		
O1–Cu1–O5 ^{viii}	90.49(8)		
O1–Cu1–O5	89.51(8)		
O5 ^{viii} –Cu1–O5	180		
4	5		
Cu1–O1	2.281(5)	Cu1–O1	1.978(2)
Cu1–O1 ⁱ	2.281(5)	Cu1–O1 ^{iv}	1.978(2)
Cu1–O2 ⁱⁱ	2.048(5)	Cu1–O2 ^v	1.979(2)
Cu1–O2 ⁱⁱⁱ	2.048(5)	Cu1–O2 ^{vi}	1.979(2)
Cu1–O3	2.024(7)	Cu1–O3	2.314(4)
Cu1–O3 ⁱ	2.024(7)	Cu1–O3 ^{iv}	2.314(4)
O3–Cu1–O3 ⁱ	179.9(4)	O1–Cu1–O1 ^{iv}	82.85(9)
O3–Cu1–O2 ⁱⁱ	90.0(3)	O1–Cu1–O2 ^v	97.24(9)
O3–Cu1–O2 ⁱⁱⁱ	90.0(3)	O1–Cu1–O2 ^{vi}	179.36(11)
O3–Cu1–O1	88.7(2)	O1–Cu1–O3	89.97(12)

O3–Cu1–O1 ⁱ	91.3(2)	O1–Cu1–O3 ^{iv}	90.22(12)
O3 ⁱ –Cu1–O2 ⁱⁱ	90.0(3)	O1 ^{iv} –Cu1–O2 ^v	179.36(11)
O3 ⁱ –Cu1–O2 ⁱⁱⁱ	90.0(3)	O1 ^{iv} –Cu1–O2 ^{vi}	97.24(9)
O3 ⁱ –Cu1–O1	91.3(2)	O1 ^{iv} –Cu1–O3	90.22(12)
O3 ⁱ –Cu1–O1 ⁱ	88.7(2)	O1 ^{iv} –Cu1–O3 ^{iv}	89.97(12)
O2 ⁱⁱ –Cu1–O2 ⁱⁱⁱ	180.0(3)	O2 ^v –Cu1–O2 ^{vi}	
O2 ⁱⁱ –Cu1–O1	75.8(2)	O2 ^v –Cu1–O3	89.15(12)
O2 ⁱⁱ –Cu1–O1 ⁱ	104.2(2)	O2 ^v –Cu1–O3 ^{iv}	90.67(12)
O2 ⁱⁱⁱ –Cu1–O1	104.2(2)	O2 ^{vi} –Cu1–O3	90.67(12)
O2 ⁱⁱⁱ –Cu1–O1 ⁱ	75.8(2)	O2 ^{vi} –Cu1–O3 ^{iv}	89.15(12)
O1–Cu1–O1 ⁱ	180.0(2)	O3–Cu1–O3 ^{iv}	179.76(14)

6a		6b	
Cu1a–O1a	2.1654(14)	Cu1b–O1b	2.0469(13)
Cu1a–O1a ^{ix}	2.1654(14)	Cu1b–O1b ^{xi}	2.0469(13)
Cu1a–O2a ^v	2.1703(14)	Cu1b–O2b	2.2995(13)
Cu1a–O2a ^x	2.1703(14)	Cu1b–O2b ^{xi}	2.2995(13)
Cu1a–N1a	2.000(2)	Cu1b–N1b	2.004(2)
Cu1a–N1a ^{xi}	2.000(2)	Cu1b–N1b ^{ix}	2.004(2)
N1a ^{ix} –Cu1a–N1a	177.70(7)	N1b–Cu1b–N1b ^{xi}	180
N1a ^{ix} –Cu1a–O1a	89.17(7)	N1b–Cu1b–O1b	88.51(6)
N1a ^{ix} –Cu1a–O1a ^{ix}	92.65(7)	N1b–Cu1b–O1b ^{xi}	91.49(6)
N1a ^{ix} –Cu1a–O2a ^v	86.58(7)	N1b–Cu1b–O2b ^{xi}	87.09(6)
N1a ^{ix} –Cu1a–O2a ^x	91.59(7)	N1b–Cu1b–O2b	92.91(6)
N1a–Cu1a–O1a	92.65(7)	N1b ^{xi} –Cu1b–O1b	91.49(6)
N1a–Cu1a–O1a ^{ix}	89.17(7)	N1b ^{xi} –Cu1b–O1b ^{xi}	88.51(6)
N1a–Cu1a–O2a ^v	91.59(7)	N1b ^{xi} –Cu1b–O2b ^{xi}	92.91(6)
N1a–Cu1a–O2a ^x	86.58(7)	N1b ^{xi} –Cu1b–O2b	87.09(6)
O1a–Cu1a–O1a ^{ix}	75.84(6)	O1b–Cu1b–O1b ^{xi}	180
O1a–Cu1a–O2a ^v	104.48(5)	O1b–Cu1b–O2b ^v	104.82(5)
O1a–Cu1a–O2a ^x	179.16(6)	O1b–Cu1b–O2b	75.18(5)
O1a ^{ix} –Cu1a–O2a ^v	179.16(6)	O1b ^{xi} –Cu1b–O2b ^{xi}	75.18(5)
O1a ^{ix} –Cu1a–O2a ^x	104.48(5)	O1b ^{xi} –Cu1b–O2b	104.82(5)
O2a ^v –Cu1a–O2a ^x	75.22(6)	O2b ^{xi} –Cu1b–O2b	180

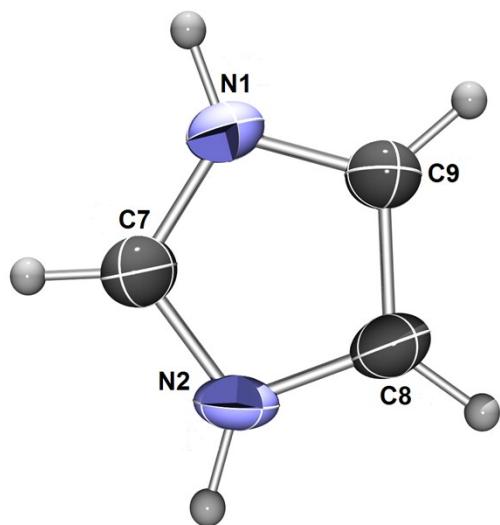


Fig. S1 ORTEP-3 drawing of the imidazolium cation in **1**. Displacement ellipsoids are drawn at the probability of 50% and hydrogen atoms are shown as spheres of arbitrary radii.

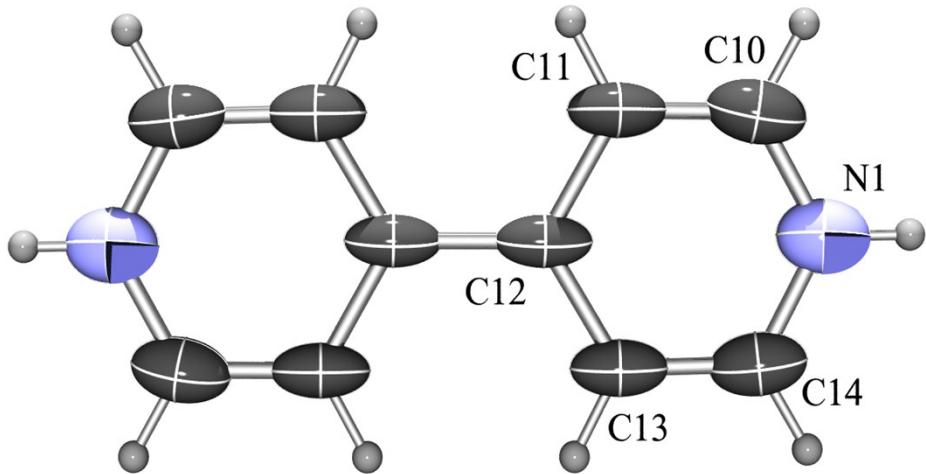


Fig. S2 ORTEP-3 drawing of the 4,4'-bipyridinium cation in **2**. Displacement ellipsoids are drawn at the probability of 50% and hydrogen atoms are shown as spheres of arbitrary radii.