

Bis(diphenylamino)naphthalene Host Materials: Careful Selection of the Substitution Pattern for the Design of Fully Solution-Processed Triple-Layered Electroluminescent Devices

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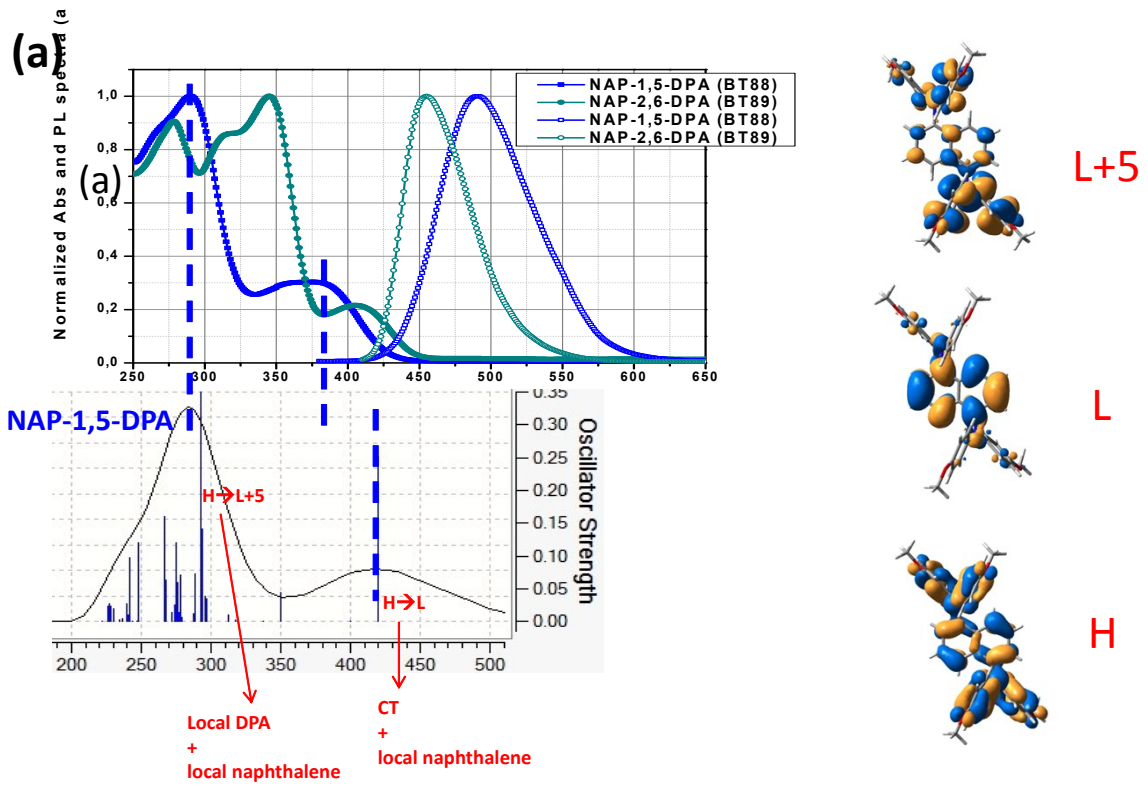
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Supporting information



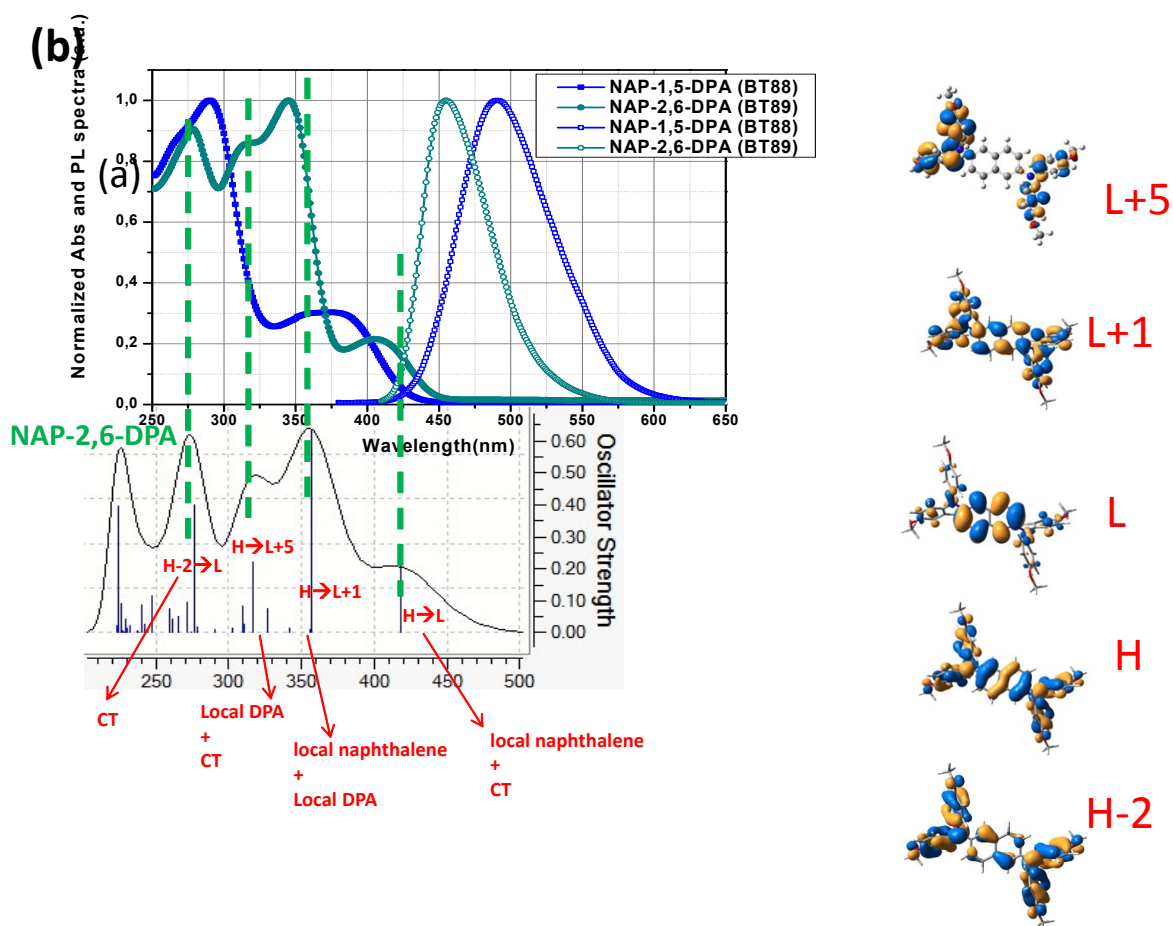


Figure S1. Experimental and theoretical absorption spectra of compounds **NAP-1,5-DPA** and **NAP-2,6-DPA**. The red shift of the theoretical low energy band for both compounds is due to the well-known bad performance of B3LYP (and other functionals containing low % of Hartree-Fock exchange) to correctly describe the CT transitions. Note the larger redshift (~40 nm) in the case of **NAP-1,5-DPA** (dominantly CT character), as compared to 13 nm in the case of **NAP-2,6-DPA** (dominantly local-naphthalene excitation character).

Table S1. Theoretical dipole moments (Debye) of compounds **NAP-1,5-DPA** and **NAP-2,6-DPA** and of the corresponding “half-compounds” (**NAP-1-DPA** and **NAP-2-DPA**, containing only one DPA substituent) in the ground state and the 1st excited state (vertical values). The dipole moments of the S₁ state in the relaxed geometry are given in parentheses). All values are obtained at the B3LYP/6-31G(d,p) level in gas phase.

	GS	S ₁
NAP-1,5-DPA	0.5	1.2 (15)
NAP-2,6-DPA	0.1	0.1 (0.2)
NAP-1-DPA	2.1	12.8
NAP-2-DPA	3.0	11.9