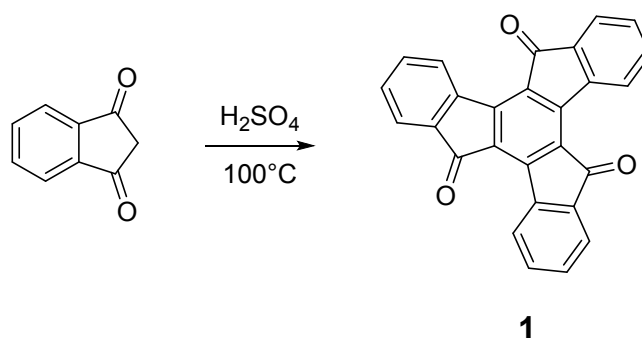


Supporting Information for “The effect of fluorination on the surface structure of truxenones”

L. A. Rochford, A. J. Ramadan, S. Holliday, T. S. Jones and C. Nielsen

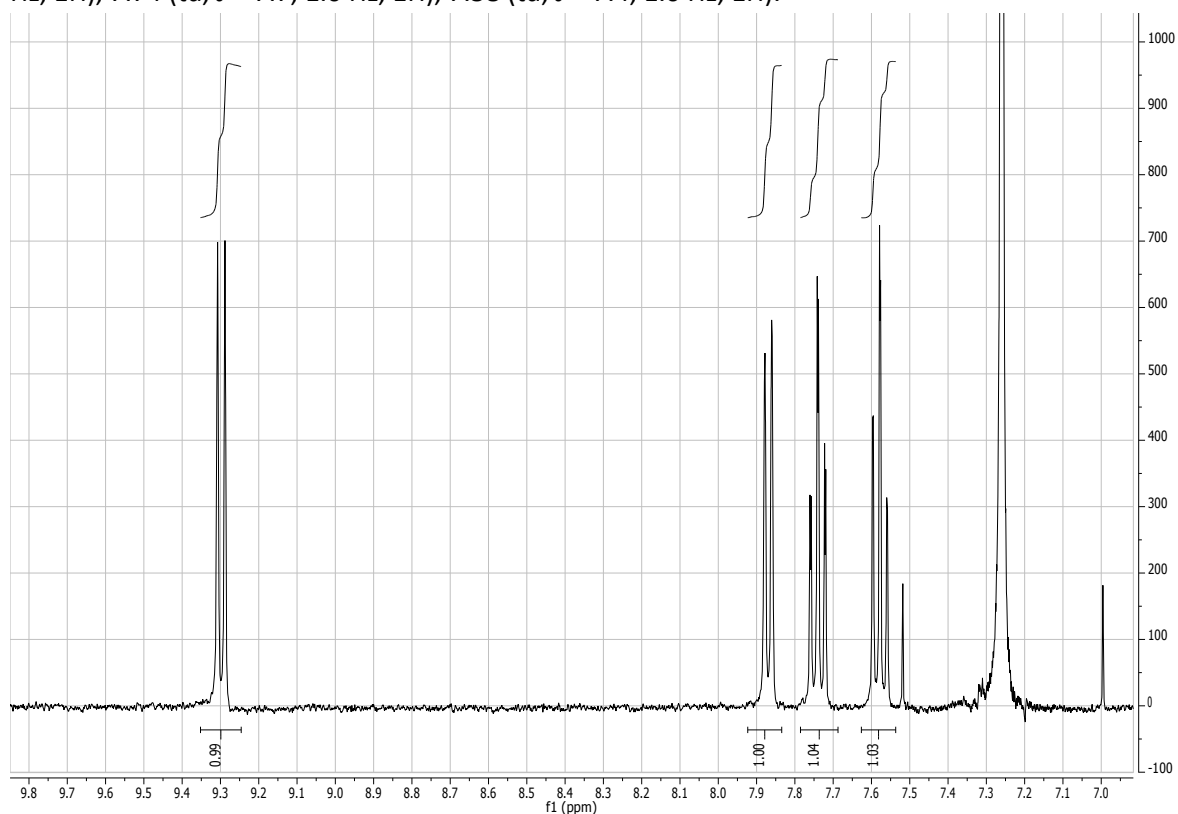
S1 – Synthesis and NMR analysis of truxenone



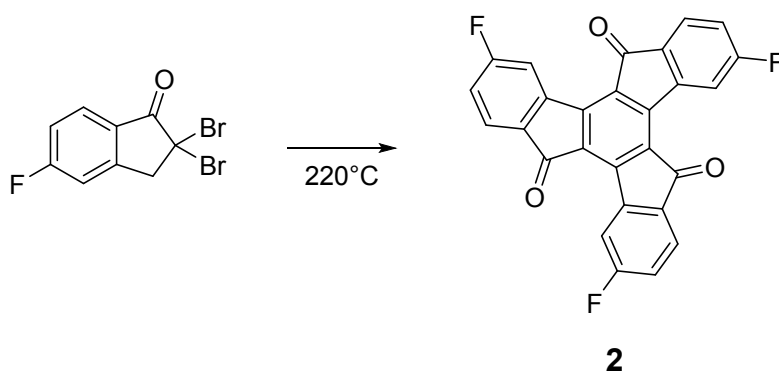
Truxenone (1)

1,3-Indanedione (20.00g, 137 mmol) was added portion-wise to concentrated sulphuric acid (200 ml) under stirring. The reaction mixture was heated to 100°C for 3 hours and subsequently poured onto ice (1L). The crude product was isolated by filtration and washed with copious amounts of water and acetone. The title compound was obtained as a yellow solid (13.81 g, 35.9 mmol, 79% yield) after trituration with dichloromethane.

^1H NMR (400 MHz, chloroform- d , 293 K) δ 9.30 (dt, $J = 7.7$, 1.0 Hz, 1H), 7.90 – 7.83 (dt, $J = 7.4$, 1.0 Hz, 1H), 7.74 (td, $J = 7.7$, 1.0 Hz, 1H), 7.58 (td, $J = 7.4$, 1.0 Hz, 1H).



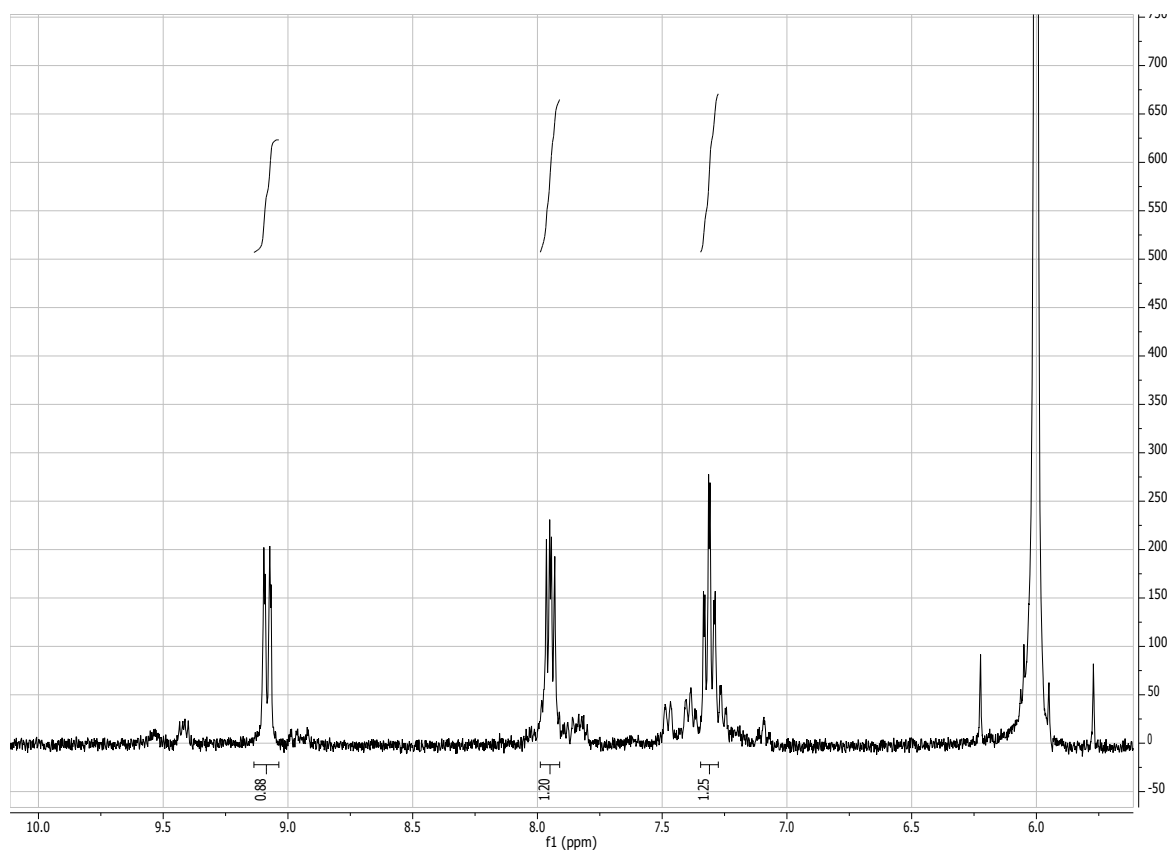
S2 – Synthesis and NMR analysis of 4,9,14-trifluorotruxenone



4,9,14-Trifluorotruxenone (2)

2,2-Dibromo-5-fluoroindan-1-one (1.40 g, 4.55 mmol) was placed in a round-bottom flask and under argon flow heated to 220°C with stirring for 1 hour. The crude product was washed with chloroform and dichloromethane to afford the title compound as a yellow solid (172 mg, 0.392 mmol, 26% yield).

¹H NMR (400 MHz, 1,1,2,2-tetrachloroethane-d₂, 393 K) δ 9.08 (dd, J = 9.5, 2.2 Hz, 1H), 7.95 (dd, J = 8.2, 5.3 Hz, 1H), 7.31 (td, J = 8.2, 2.2 Hz, 1H).



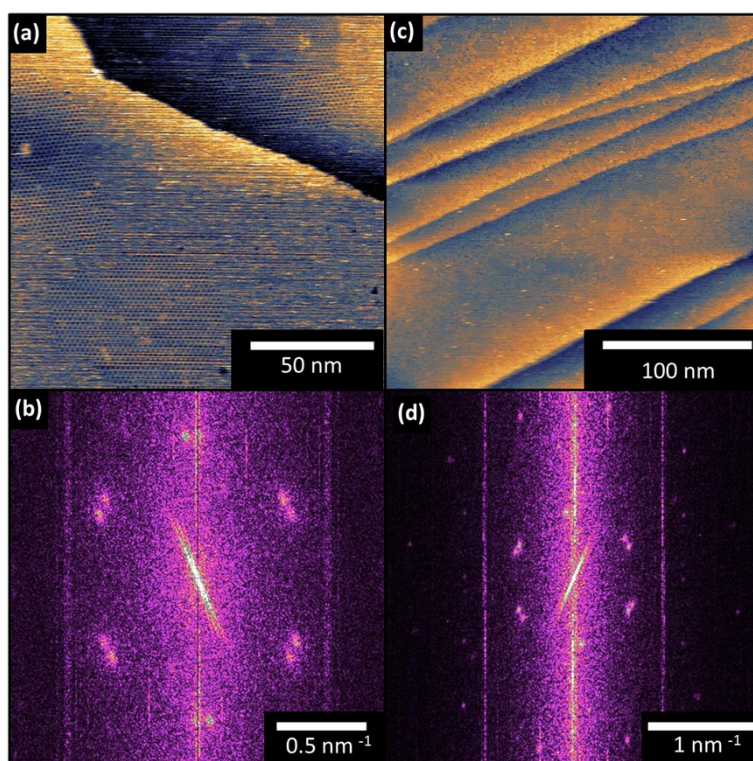


Figure S3 - STM images of F₃-truxenone / Cu (111) with corresponding 2D-FFT images at (a) high ($V_s = -2$ V, $I_T = 65$ pA) / (b) and lower (c) ($V_s = -1.5$ V, $I_T = 125$ pA) / (d) magnification.