

Electronic Supplementary information (ESI)

Highly water-dispersible Pd nanoparticles and single atoms magnetite-supported as excellent catalysts for Suzuki and hydrogenation reactions

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Figures:

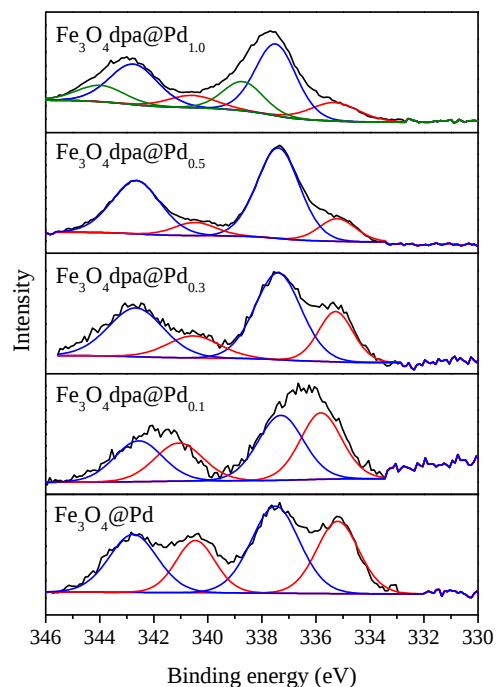


Fig. S1 Pd3d core level XPS of the catalysts. The black lines represent the experimental data, whereas the red, blue and green show the fitted Pd(0), Pd(II) and Pd(IV) components, respectively.

Tables:

Table S1. Temperature effect on Suzuki-Miyaura coupling with Fe₃O₄dpa@Pd_{0.5} as catalyst.

Temperature (°C)	TOF (h ⁻¹)
25	204
65	110000
80	182000

Bromobenzene (3 mmol), phenylboronic acid (3.6 mmol), K₃PO₄ (9 mmol), catalyst (9.4×10⁻⁵ mmol Pd), 65 °C and EtOH:water.

Table S2. Base effect on Suzuki-Miyaura coupling with Fe₃O₄dpa@Pd_{0.5} as catalyst.

Base	TOF (h ⁻¹)
KOH	122
K ₃ PO ₄	110000
K ₂ CO ₃	100000
AcONa	28900
NaHCO ₃	63500

Bromobenzene (3 mmol), phenylboronic acid (3.6 mmol), base (9 mmol), catalyst (9.4×10⁻⁵ mmol Pd), 65 °C, and EtOH:water.

Table S3. Solvent effect on Suzuki-Miyaura coupling with Fe₃O₄dpa@Pd_{0.5} as catalyst.

Solvent	TOF (h ⁻¹)
EtOH	384
Water	1470
EtOH:Water	110000

Bromobenzene (3 mmol), phenylboronic acid (3.6 mmol), K₃PO₄ (9 mmol), catalyst (9.4×10⁻⁵ mmol Pd), 65 °C, and solvent (60 mL).

Table S4. Temperature effect on Suzuki-Miyaura coupling with Fe₃O₄dpa@Pd_{0.5} as catalyst in water.

Temperature (°C)	TOF (h ⁻¹)
65	3150
100	14950

Bromoaryl (0.5 mmol), phenylboronic acid (0.6 mmol), K₃PO₄ (1.5 mmol), 2.0×10⁻⁴ mmol Pd, water, air atmosphere.