

Photoresponsive Nanostructured Membranes

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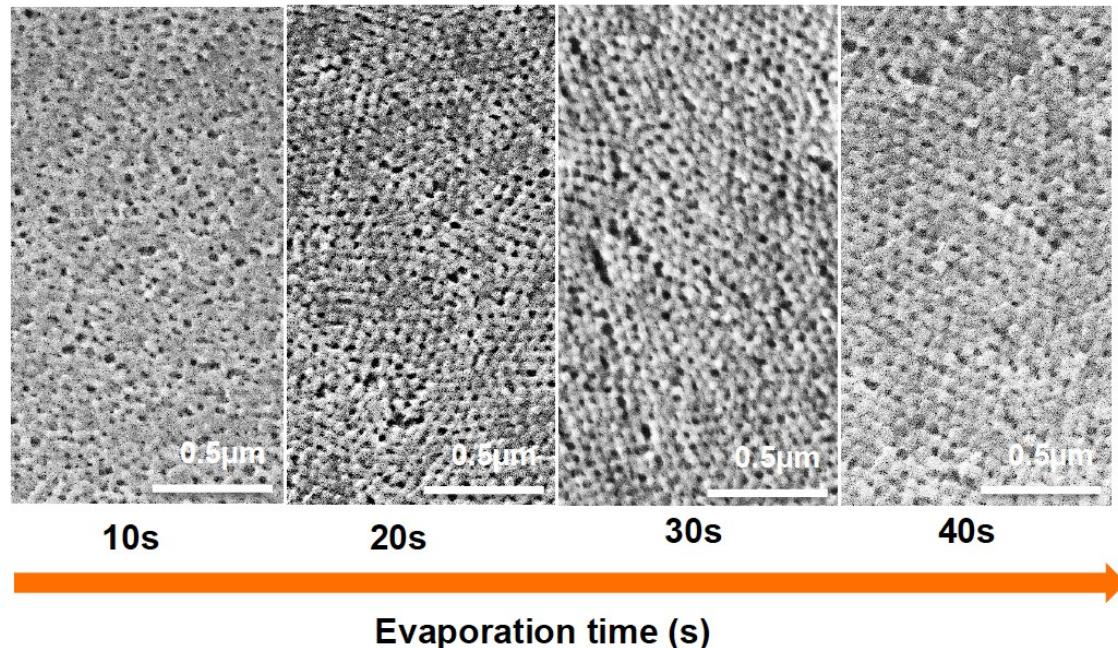
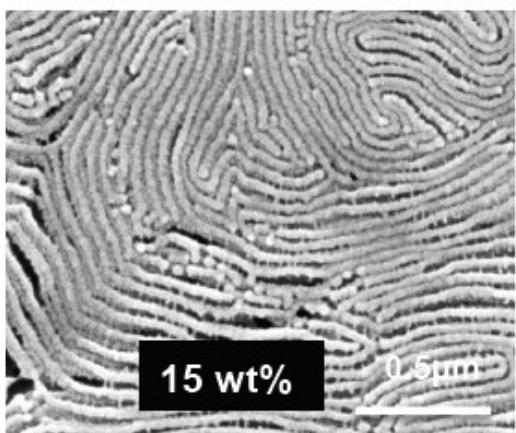


Figure S1. SEM images of membranes surfaces cast from 18 wt% $\text{PS}_{43\text{K}}\text{-}b\text{-PAnMMA}_{1.5\text{K}}\text{-}b\text{-PMMA}_{105\text{K}}$ copolymer solutions in 35/29/18 DMAc/Acetonitrile/THF with different evaporation times.

PS_{44K}-b-PAnMMA_{1.2K}-b-PMMA_{45K}



PS_{43K}-b-PAnMMA_{1.5K}-b-PMMA_{105K}

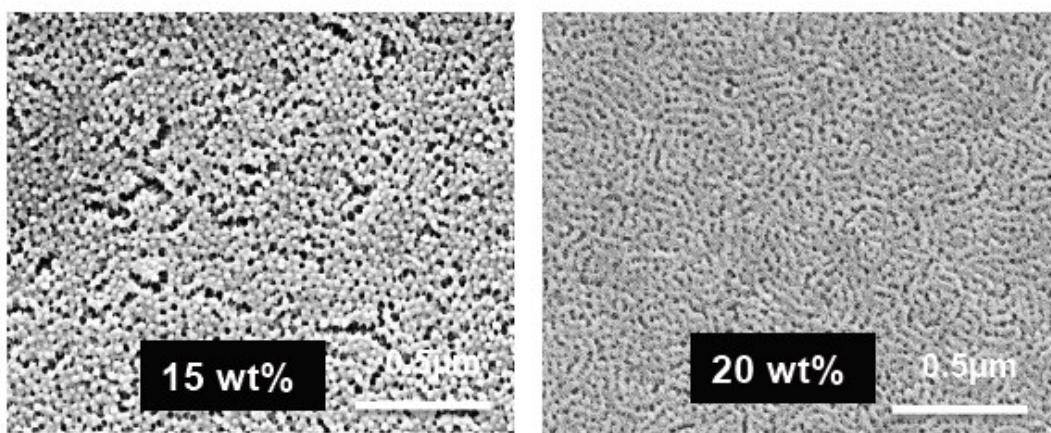


Figure S2. SEM images of membranes surfaces cast from 15 wt% PS_{44K}-b-PAnMMA_{1.2K}-b-PMMA_{45K} copolymer solution in DMAc/Acetone/THF with 10 s evaporation time; membranes surfaces cast from PS_{43K}-b-PAnMMA_{1.5K}-b-PMMA_{105K} copolymer solutions with different concentrations in DMAc/Acetonitrile/THF with 20 s evaporation time membranes.

Table 1 Hansen solubility parameters for polymer and solvents³¹

	δ_D	δ_P	δ_H	$\delta = (\delta_D^2 + \delta_P^2 + \delta_H^2)^{1/2}$
PS	18.6	1.0	4.1	19.1
PMMA	18.6	10.5	7.5	22.6
Water	15.6	16.0	42.3	47.8
DMAc	16.8	11.5	10.2	22.3
Acetone	15.5	10.4	7.0	19.9
THF	16.8	5.7	8.0	19.5
Acetonitrile	15.3	18.0	6.1	24.4

δ_D : Dispersion forces (van der Waals), δ_P : Polarity and δ_H : hydrogen bonding