

Supplementary Information

Fast response ammonia sensors based on TiO₂ and NiO nanostructured bilayer thin films

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FIGURE S1

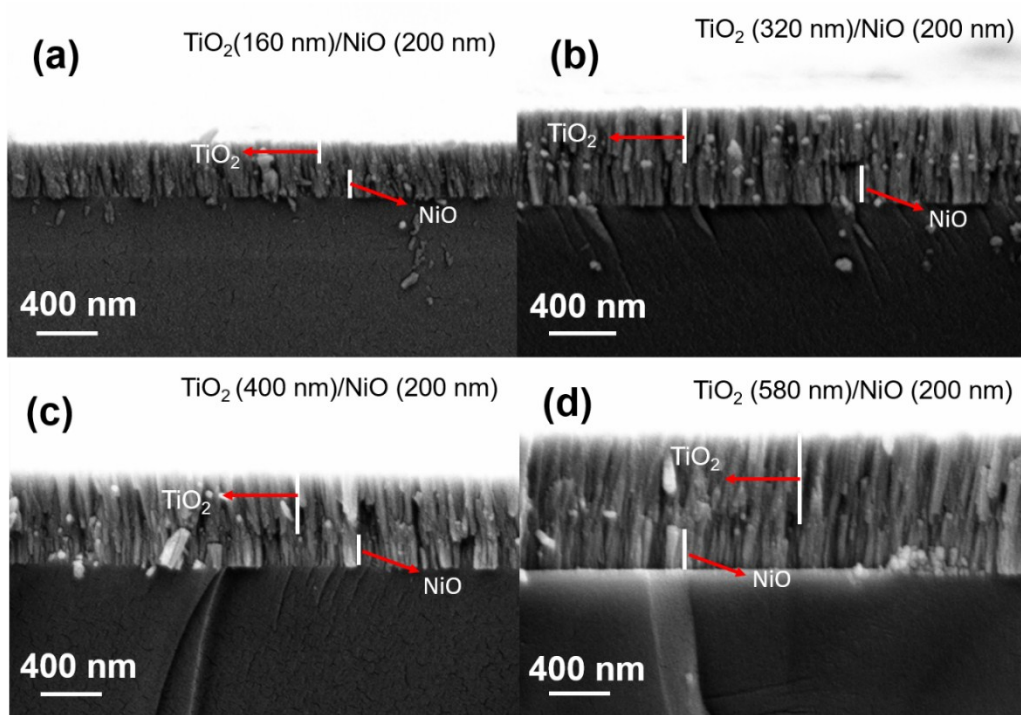


Figure S1. The cross-section micrographs of TiO₂/NiO bilayer sensor with different thickness of top TiO₂ layer

Figure S1 depicts the FE-SEM cross-sectional micrographs of four TiO₂/NiO bilayer sensing films. The results indicates that the bottom NiO film thickness was found to be fixed as 200 nm and the top TiO₂ film thickness was varied from 160 nm to 580 nm.

FIGURE S2

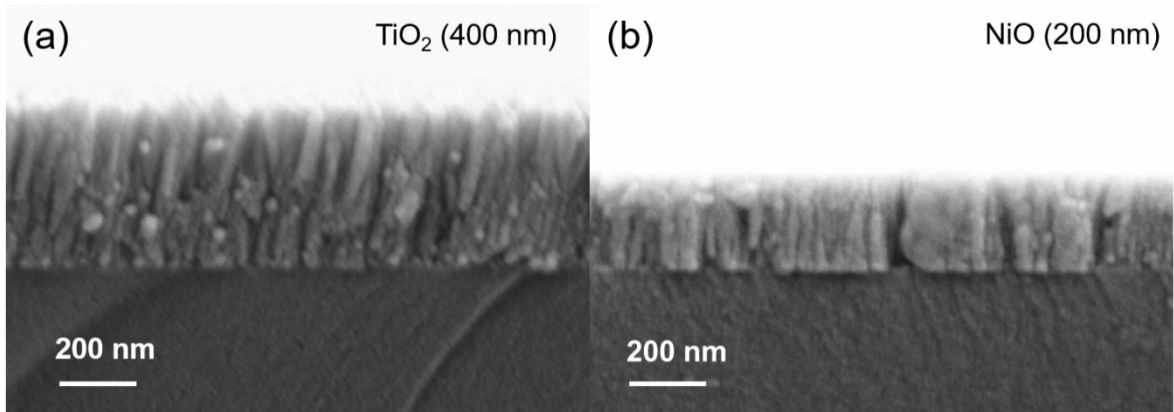


Figure S2. The cross-section micrographs of (a) pure TiO₂ of 400 nm and (b) pure NiO layer of 200 nm.

Figure S2 depicts the individual thickness of TiO₂ and NiO thin films to be nearly 400 nm and 200 nm respectively.

FIGURE S3

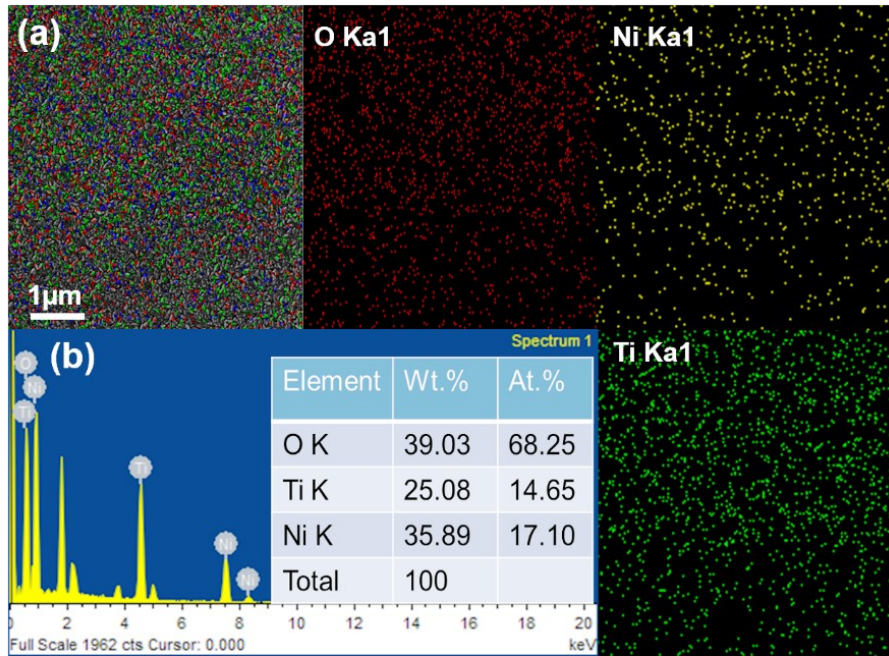


Figure S3. (a) Elemental mapping, and (b) EDS spectra of NiO/TiO₂ bilayer thin film.

Figure S3 shows the elemental mapping and EDS spectrum of NiO/TiO₂ bilayer sensor which indicate the uniform element distribution of titanium, oxygen and nickel elements. The red region of O, green region of Ti and yellow regions of Ni are typically overlapped.

FIGURE S4

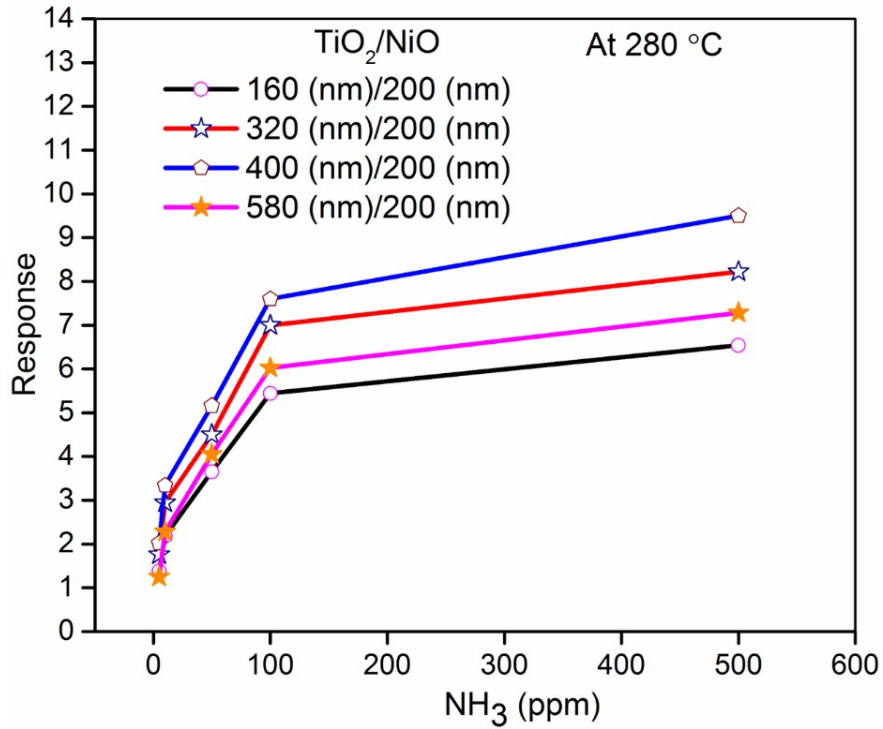


Figure S4. Variation of the response of TiO₂/NiO sensor with thickness of top TiO₂ layer using different ammonia concentrations at 280°C.

Figure S4 shows that during the sensing measurements, the sensor response increases as the TiO₂ layer thickness increases from 160 nm to 400 nm but it decreases when the TiO₂ thickness further increases to 580 nm. It means the TiO₂/NiO bilayer sensor with 400 nm TiO₂ layer (at top) and 200 nm NiO layer (at bottom) exhibited the best sensing response towards ammonia at 280 °C.

FIGURE S5

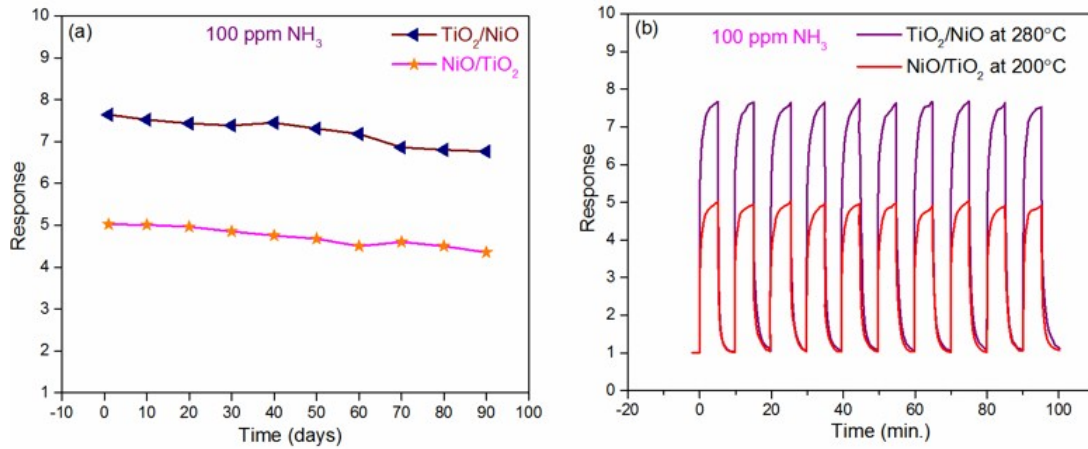


Figure S5. (a) Stability test for 90 days and (b) Stable response up to 10th cycles of TiO₂/NiO and NiO/TiO₂ bilayer sensors towards 100 ppm NH₃ at 280°C and 200°C, respectively.

Figure S5 (a) and (b) shows the stability characteristics of TiO₂/NiO and NiO/TiO₂ bilayer sensors towards 100 ppm NH₃ at 280°C and 200°C respectively. During the measurements, the TiO₂/NiO and NiO/TiO₂ bilayer sensors exhibits nearly constant response signal (~10% changes) and (~12% changes) respectively, indicating the significant long term stability of the sensor. Therefore, we concluded that the above-mentioned stability test will contribute to establish both the bilayer sensors as potential candidate for NH₃ gas sensing application.

FIGURE S6

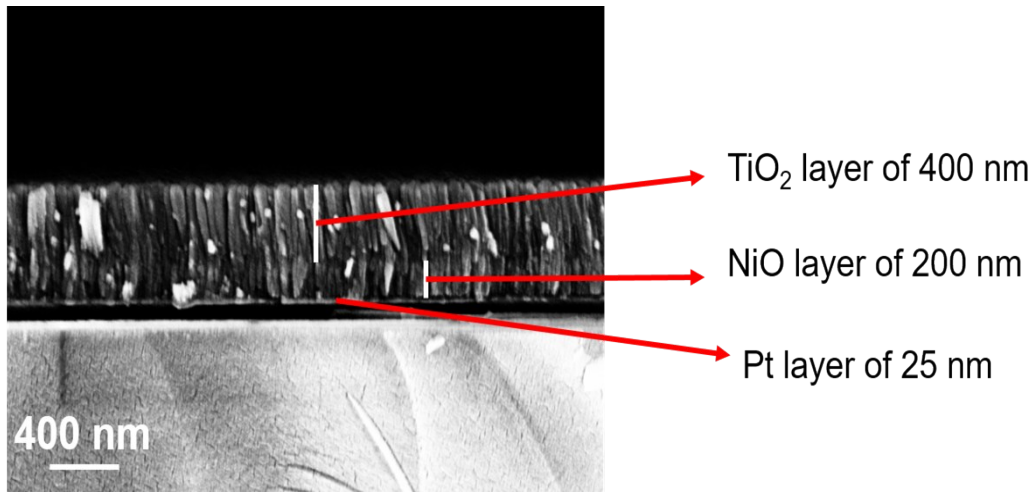


Figure S6. The cross-section micrographs of TiO₂/NiO bilayer sensor with Pt layer of 25 nm thick as bottom electrode.

Figure S6 shows the required FE-SEM cross section view of TiO₂/NiO bilayer with the Pt bottom contact (electrode) thickness of about 25 nm.

FIGURE S7

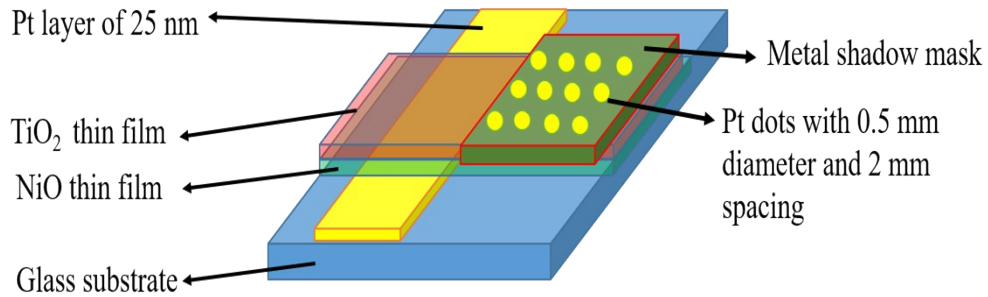


Figure S7. The schematic view of the TiO₂/NiO bilayer sensor with the platinum (Pt) as bottom and top electrodes

Figure S7 shows the schematic view of TiO₂/NiO bilayer sensor with the top and bottom Pt contacts (electrodes). For electrical measurements, the bottom contact layer and top dots of platinum (Pt) of 25 nm thickness were prepared using metal shadow mask on glass substrates via DC sputtering technique. The diameter of Pt dot is 0.5 mm and space between the two dots is 2 mm, approximately.

Table S1: Brief summary of previous reported literature on several semiconductor oxide based

Sensor (metal oxide)	Fabrication technique	NH ₃ ppm	Operating Temp.(°C)	Sensing response	Response/recovery time	Ref.
Co ₃ O ₄ /SnO ₂ composite	Hydrothermal route	100	200	13.6	4 sec/--	1
Pd/WO ₃	hydrothermal	1000	300	2.2	15 sec/76 sec	2
α-Fe ₂ O ₃	Sol-gel	100	RT	39%	27 sec/46 sec	3
NiO nanoparticles	Hydrothermal process	30	RT	20%	29 sec/150 sec	4
TeO ₂ thin films	Sputtering	500	170	58%	3.1 min/5.6 min	5
Hierarchical SnO ₂	Thermal evaporation	300	200	6	16 sec/25 sec	6
Pd doped WO ₃	Screen printing	200	350	10	----	7
ZnO	Hydrothermal	4.6	100	3.96%	-----	8
h-WO ₃	Annealing	50	30	6	1.3 min/3.8 min	9
NiO-ZnO	Hydrothermal route	50	RT	40%	27 sec/155 sec	10
TiO ₂ /NiO bilayer thin film	DC sputtering	100	280	7.6	24 sec/45 sec	Present work

ammonia gas sensor prepared by various deposition techniques.

References

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