Electronic Supplementary Information

Solvothermal Synthesis of Discrete Cages and Extended Networks Comprising {Cr(III)3O(O2CR)3(oxime)3}2- (R = H, CH3, C(CH3)3, C14H9) Building Blocks

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Figure S1 Two adjacent $\{NaCr(III)_6O_2(O_2C-C_{14}H_9)_6(Naphth-sao)_6\}^{3-}$ cages in 1 forming a hydrophobic pocket accommodating a space-fill represented $[NEt_4]^+$ counter anion (grey spheres).



Figure S2 Crystal packing observed in 1 as viewed along the *c* unit cell direction. All hydrogen atoms and solvent molecules of crystallisation have been omitted for clarity. Colour code as main text.



Figure S3 The 2-D sheets in 2 as viewed along the *b* direction of the unit cell. Colour code: Green (Cr), Yellow (Na), Red (O). All other atoms have been omitted for simplification and clarity.



Figure S4 Bonding mode exhibited by the 3,5-di-*tert*-Bu-sao²⁻ ligands in 3.



Figure S5: The bridging arrangements exhibited by the Me-sao²⁻ and acetate anions in the 2D network in 4.



Figure S6 The bridging arrangements exhibited by the Me-sao²⁻ and acetate anions in the 2D network in 5.

	1	2	3
Formula ^a	C ₁₇₂ H ₁₃₆ N ₈ O ₂₆ Na ₁ Cr ₆	$C_{96}H_{88}N_{12}O_{30}Na_2Cr_6$	$C_{132}H_{198}N_{12}O_{26}Na_4Cr_6$
$M_{ m W}$	3065.87	2237.68	2772.98
Crystal System	Monoclinic	Monoclinic	Trigonal
Space group	C2/c	$P2_1/n$	R-3c
a/Å	39.800(2)	17.8061(18)	26.9779(12)
b/Å	17.2760(6)	12.4863(14)	26.9779(12)
$c/{ m \AA}$	30.452(2)	22.4726(18)	71.622(4)
$lpha^{ m /o}$	90	90	90
$eta / ^{\circ}$	121.745(4)	91.553(9)	90
$\gamma^{ m /o}$	90	90	120
$V/Å^3$	17805.9(19)	4994.5(9)	45143(5)
Ζ	4	2	12

 Table S1 Crystal data obtained from complexes 1-3.

<i>T</i> /K	150(2)	150(2)	150(2)
$\lambda^{ m b}/{ m \AA}$	0.7107	0.7107	0.7107
$D_{\rm c}/{ m g~cm^{-3}}$	1.144	1.488	1.224
μ (Mo-Ka)/ mm ⁻¹	0.419	0.724	0.496
Meas./indep.(<i>R</i> _{int}) refl.	72141/21163(0.2071)	9115/4123 (0.1742)	8855/4076(0.2348)
Restraints, Parameters	2210, 1945	0, 646	1286, 616
wR2 (all data) ^c	0.4039	0.3020	0.2509
$R1^{d,e}$	0.1343	0.0967	0.0907
Goodness of fit on F^2	0.955	1.035	1.020

^{*a*} Includes guest molecules (Note: Solvents of crystallisation in 1 and 3 respectively are not counted in formula as were calculated using SQUEEZE program.^{*b*} Mo-K α radiation, graphite monochromator. ^{*c*} $wR2 = [\Sigma w(1F_o^2 - 1F_o^2 - 1F$

	4	5.3MeCN
Formula ^a	C ₆₂ H ₆₆ N ₇ O ₂₇ Na ₃ Cr ₆	C ₃₆ H ₃₉ N ₆ O ₁₉ Na ₂ Cr ₃
$M_{ m W}$	1719.16	1084.70
Crystal System	Monoclinic	Trigonal
Space group	I2/a	R-3
<i>a</i> /Å	26.3193(11)	15.5335(5)
$b/{ m \AA}$	13.1295(5)	15.5335(5)
c/Å	22.7708(12)	36.608(2)
α/º	90	90
β/°	113.736(5)	90
γ/ ⁰	90	120
V/Å ³	7203.0(6)	7649.8(6)
Ζ	4	6
T/K	150(3)	150(3)
$\lambda^{\mathrm{b}}/\mathrm{\AA}$	0.7107	0.7107
$D_{\rm c}/{ m g~cm^{-3}}$	1.585	1.413
μ (Mo-Ka)/ mm ⁻¹	0.977	0.726
Meas./indep.(R_{int}) refl.	6590/4621 (0.0387)	3118/2744 (0.00350)
Restraints, Parameters	0, 475	0, 205
wR2 (all data) ^c	0.1315	0.1304
$R1^{d,e}$	0.0483	0.0473
Goodness of fit on F^2	1.019	1.146

 Table S2 Crystal data obtained from extended networks 4 and 5.

 $[\Sigma w (|F_o^2| - |F_c^2|)^2 / \Sigma w |F_o^2|^2]^{1/2}. \ ^d$ For observed data. $^e R_1 = \Sigma ||F_o| - |F_c| / \Sigma |F_o|$



Figure S7 TGA trace obtained on crystalline samples of and 3 (top) and 5 (bottom) analyzed in the 25-600 °C temperature range in an N₂ atmosphere.