

Supporting Information

Lanthanide MOFs Constructed Based on a Difunctional Ligand with Bimodal Emission and Eu³⁺ doped Dy³⁺ Materials: White Emission and Color Tuning

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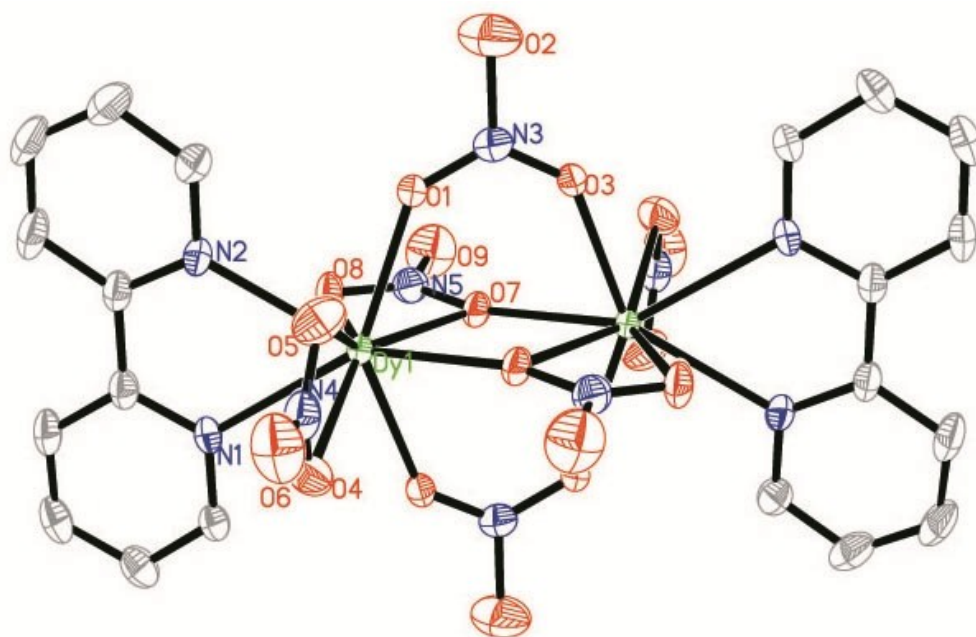


Fig. S1 The structure unit of complex **1·Dy**.

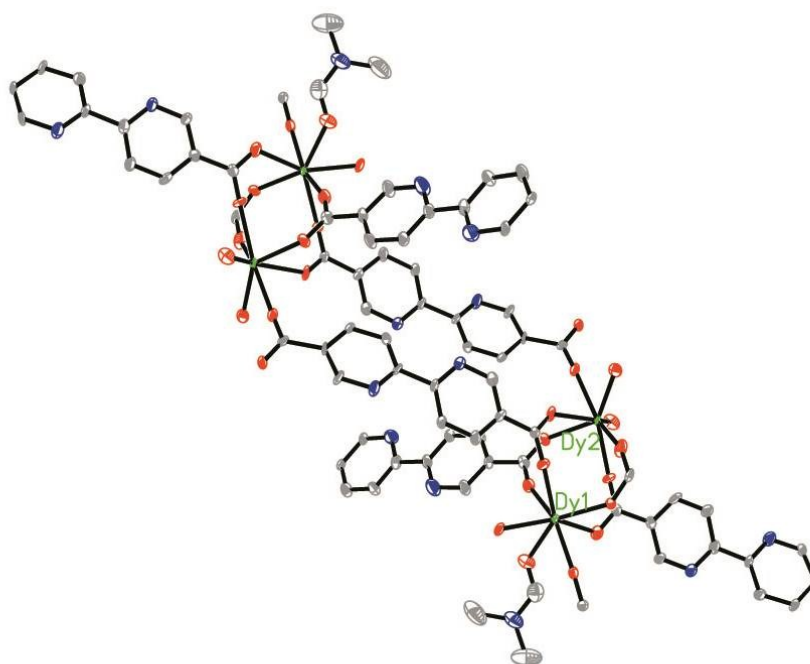


Fig. S2 The structure unit of complex **3·Dy**.

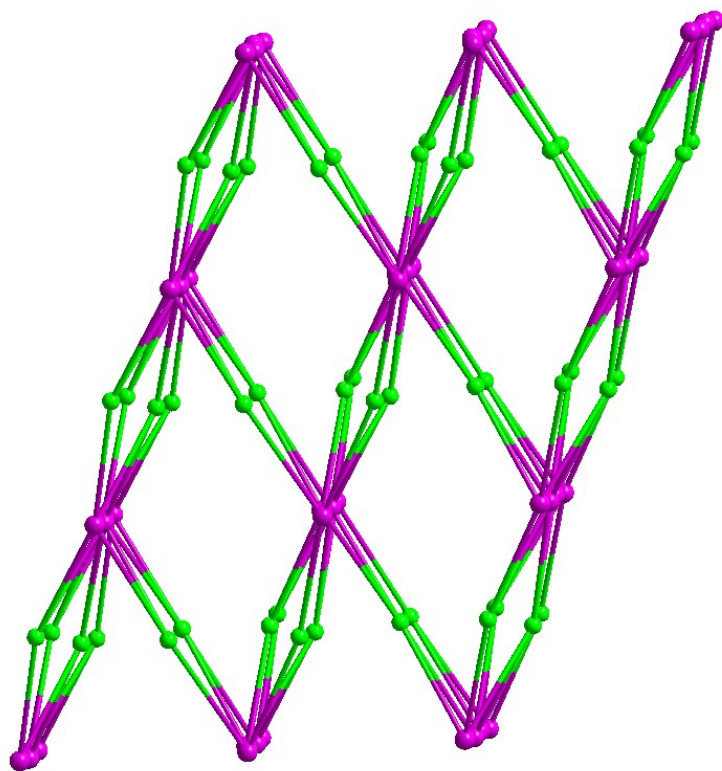


Fig. S3 The topology structure of **3·Dy**.

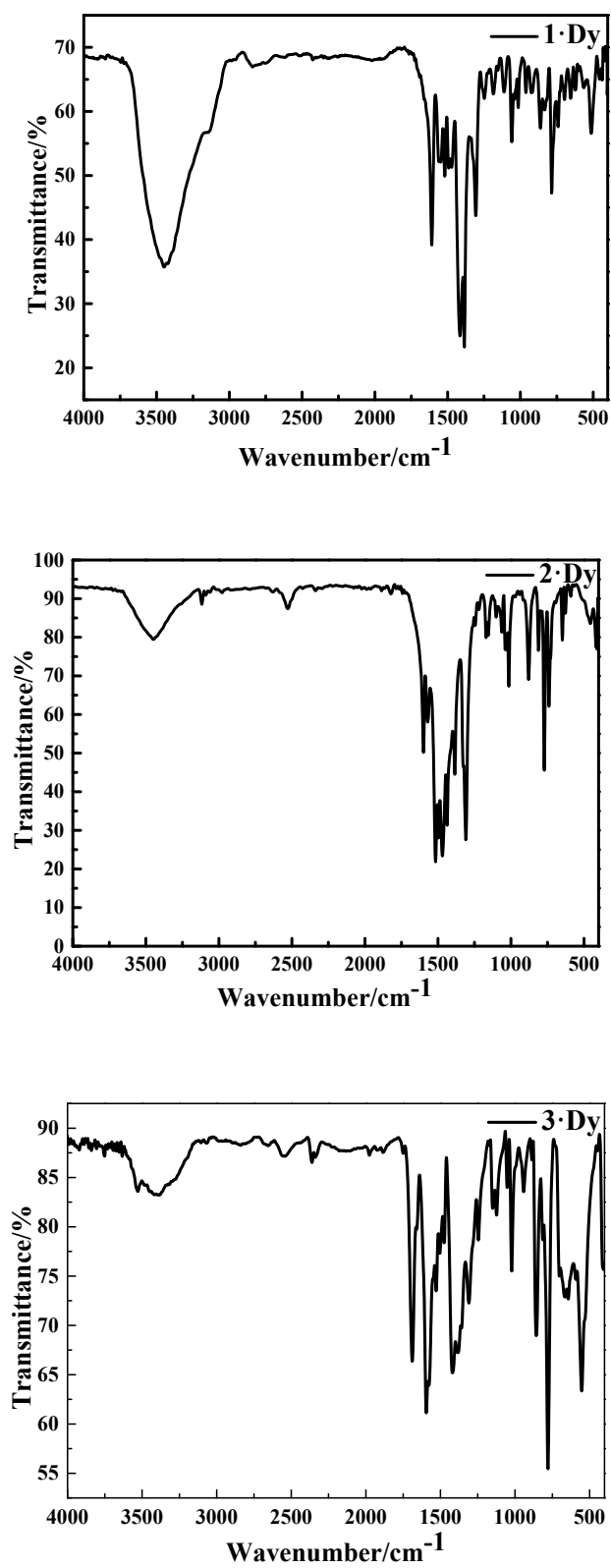


Fig. S4 Infrared spectra of complex **1·Dy**~**3·Dy** recorded from a KBr pellet.

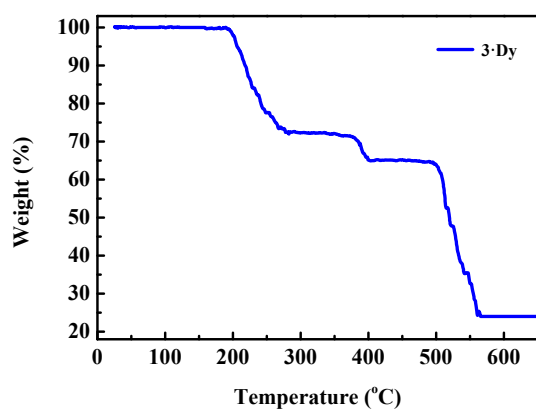
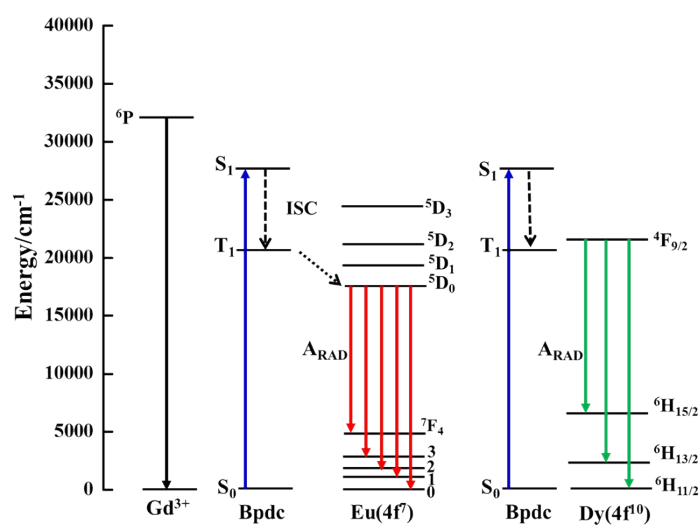


Fig. S5 Thermal gravimetric curves of complex **3·Dy**.



Scheme S1. Schematic of the energy absorption, migration and emission process of lanthanide complexes.

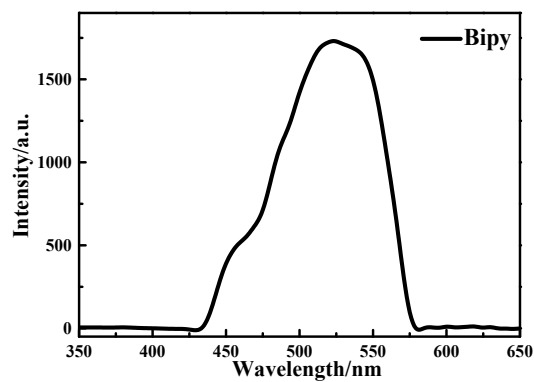


Fig. S6 The luminescence property of 2,2-bipyridine in solid state at 298 K.

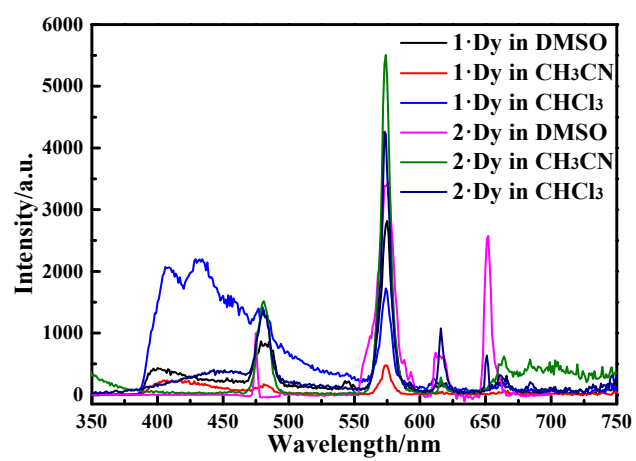


Fig. S7 The luminescence property of **1-Dy** and **2-Dy** in different solvents.

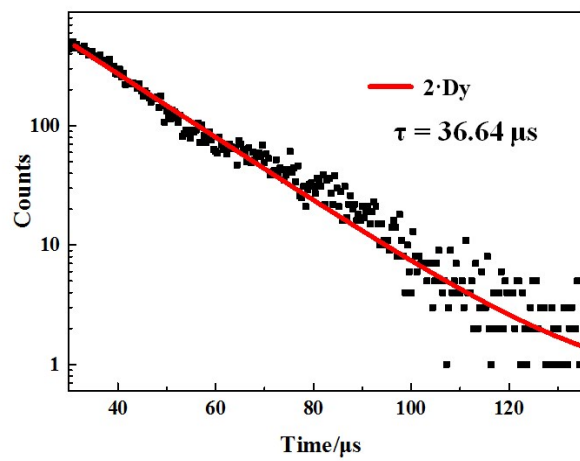
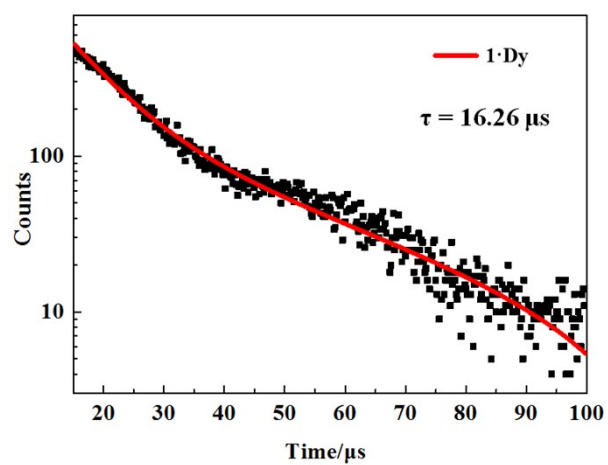


Fig. S8 Luminescence decay curves of **1-Dy** and **2-Dy** in the solid state at 298 K.

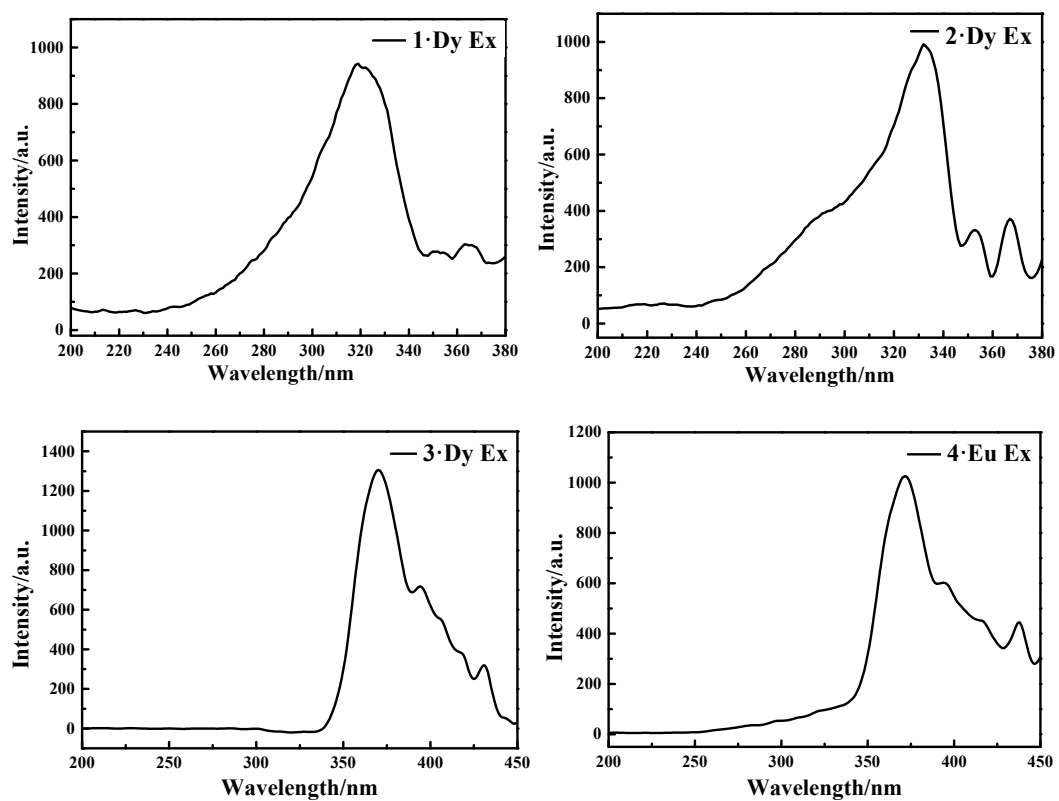


Fig. S9 Excitation spectra of complex 1·Dy, 2·Dy, 3·Dy and 4·Eu.

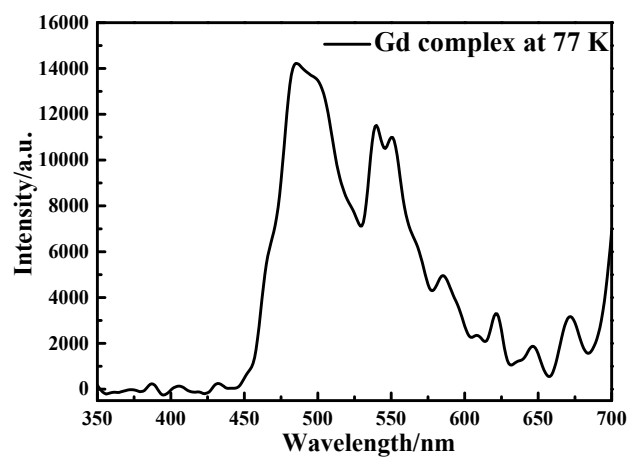


Fig. S10 Phosphorescence spectra of Gd complex at 77 K.

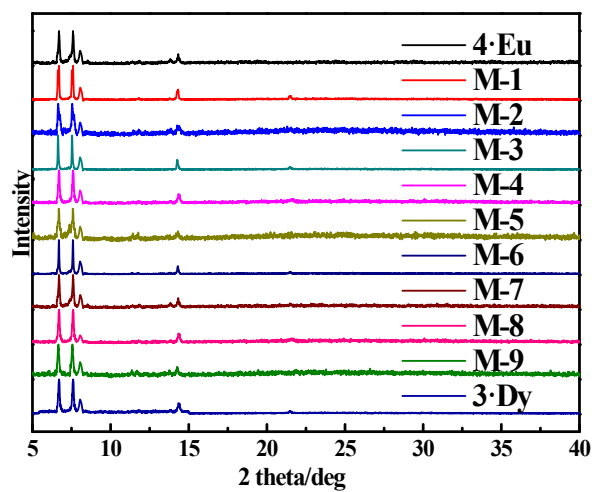


Fig. S11 Experimental and simulated PXRD patterns of complex **3·Dy**, **4·Eu** and **M-1~M-9**.

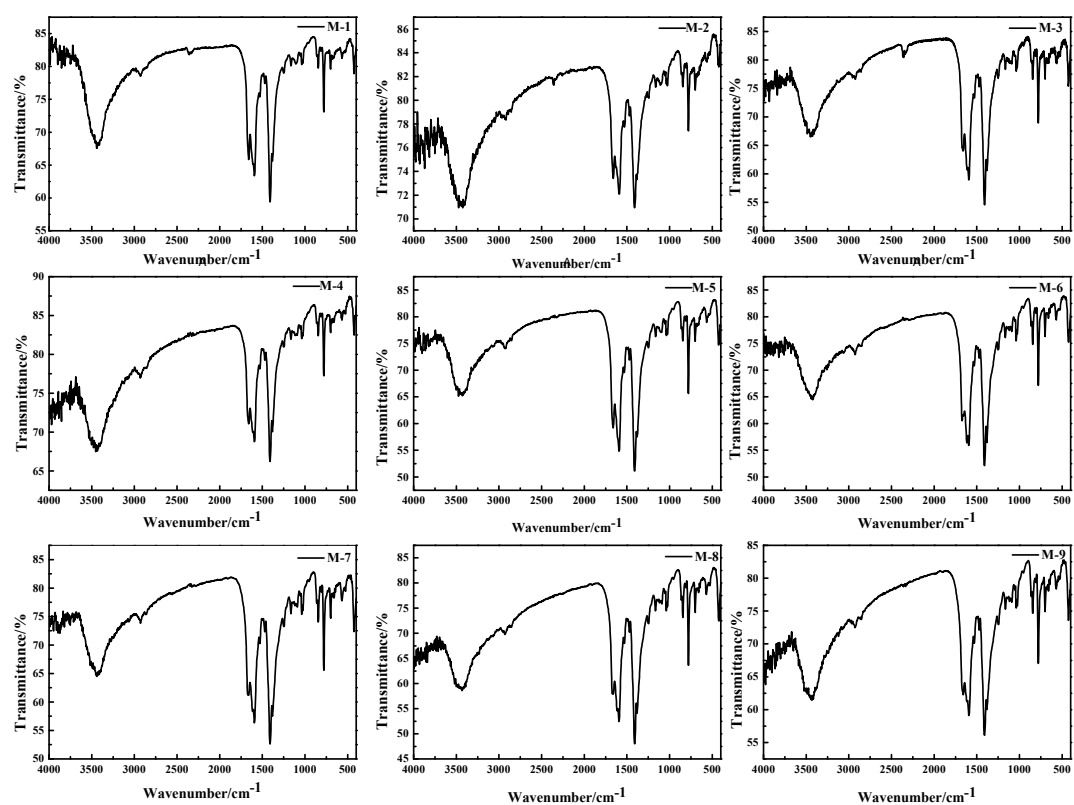


Fig. S12 Infrared spectra of complex **M-1~M-9** recorded from a KBr pellet.

Table S1. Summary of the tunable color luminescent properties based on RGB primary colors.

Empirical formula	Tunable color	Metal and Ligand	References
$[\text{Eu}(\text{tta})_2(\text{CR1})_2]_n$	White	Eu CR1 = 7-diethylamino-2-oxo-2H-chromen-3-carboxylic chloride + N-(Rhodamine-6G)lactamethylenediamine tta = 1,1,1-trifluoro-3-(2-thenoyl)acetone	Angew. Chem. Int. Ed. 2009, 48, 6132–6135
$\text{Na}_3[\text{Ln}(\text{PDA})_3](\text{H}_2\text{O})_x$ Ln = Eu (x = 10), Tb (x = 9)	White	Eu / Tb PDA = pyridine-2,6-dicarboxylate	J. Mater. Chem., 2012, 22, 3210–3214
$\{\text{EuOH}(\text{H}_2\text{O})_6[\text{Zn}_2\text{Eu}_4(4\text{-Htbca})_2\text{-}(4\text{-tbca})_8(\text{H}_2\text{O})_{12}]\}_n \cdot 6n\text{H}_2\text{O}$ $\{\text{TbOH}(\text{H}_2\text{O})_6[\text{Zn}_2\text{Tb}_4(4\text{-Htbca})_2\text{-}(4\text{-tbca})_8(\text{H}_2\text{O})_{12}]\}_n \cdot 6n\text{H}_2\text{O}$	White	Gd / Tb / Eu 4-H ₂ tbca = 4-(1H-tetrazol-5-yl)-biphenyl-3-carboxylic acid	J. Mater. Chem. C, 2013, 1, 4634–4639
$[\text{H}(\text{H}_2\text{O})_8]\text{-}[\text{LnZn}_4(\text{imdc})_4(\text{Him})_4]$ [Ln = Eu (3), Gd (4), Tb (5)]	White	Gd / Tb / Eu H ₃ imdc = 4,5-imidazoledicarboxylic acid; Him = imidazole	Inorg. Chem. 2012, 51, 1201–1203
$\text{La}(1,3,5\text{-BTC})(\text{H}_2\text{O})_6\text{:Eu}^{3+}, \text{Tb}^{3+}$	White	Eu / Tb	Crystal Growth & Design, Vol. 10, No. 1, 2010,

		1,3,5-BTC = 1,3,5-benzenetricarboxylate	16–19
$\{[\text{Ln}_2(\text{L})_2] \cdot (\text{H}_2\text{O})_3 \cdot (\text{Me}_2\text{NH}_2)_2\}_n$ (Ln = Eu (6), Gd (7), Tb (8))	White	Gd / Tb / Eu H ₄ L = 5-(3,5-dicarboxybenzyloxy)-isophthalic acid	Dalton Trans., 2013, 42, 10579–10586
$[\text{Ln}(\text{dpdc})_{1.5}(\text{IP})(\text{H}_2\text{O})]_n$ (Ln = Eu 2, Gd 3)	White	Eu / Gd dpdc = 2,2'-diphenyldicarboxylate and IP = 1H-imidazo[4,5-f][1,10]-phenanthroline	Chem. Commun., 2013, 49 , 10397–10399
$[\text{Ln}(\text{TZI})(\text{H}_2\text{O})_5]_n$ [Ln = Eu (2), Gd (3), Tb (4)]	White	Gd / Tb / Eu H ₃ TZI = 5-(1H-tetrazol-5-yl)isophthalic acid	CrystEngComm, 2015, 17, 6030–6036
$[\text{NaLn}-(\text{pztc})(\text{H}_2\text{O})_3] \cdot \text{H}_2\text{O}$ [Ln = Eu (3), Gd (4) and Tb (5)]	White	Gd / Eu / Tb H ₄ pztc = pyrazine-2,3,5,6-tetracarboxylic acid	Dalton Trans., 2014, 43, 12574–12581

Table S2. Selected bond distances (Å) and angles (°) for **1·Dy**.

Dy(1)-O(1)	2.301(3)	Dy(1)-O(8)	2.412(3)	Dy(1)-O(7)	2.533(3)
Dy(1)-O(3)	2.306(3)	Dy(1)-O(4)	2.471(4)	Dy(1)-N(2)	2.547(3)
Dy(1)-O(7)#1	2.320(3)	Dy(1)-O(5)	2.513(4)	Dy(1)-N(1)	2.579(3)
O(1)-Dy(1)-O(3)	138.59(10)	O(7)#1-Dy(1)-O(5)	77.44(13)	O(8)-Dy(1)-N(2)	75.63(12)
O(1)-Dy(1)-O(7)#1	75.30(11)	O(8)-Dy(1)-O(5)	143.41(14)	O(4)-Dy(1)-N(2)	92.11(14)
O(3)-Dy(1)-O(7)#1	75.21(11)	O(4)-Dy(1)-O(5)	49.99(16)	O(5)-Dy(1)-N(2)	70.33(13)
O(1)-Dy(1)-O(8)	83.18(13)	O(1)-Dy(1)-O(7)	72.30(11)	O(7)-Dy(1)-N(2)	121.19(11)
O(3)-Dy(1)-O(8)	91.69(13)	O(3)-Dy(1)-O(7)	72.40(11)	O(1)-Dy(1)-N(1)	137.77(12)
O(7)#1-Dy(1)-O(8)	126.53(11)	O(7)#1-Dy(1)-O(7)	74.89(11)	O(3)-Dy(1)-N(1)	77.57(11)
O(1)-Dy(1)-O(4)	125.73(14)	O(8)-Dy(1)-O(7)	51.91(10)	O(7)#1-Dy(1)-N(1)	146.60(12)
O(3)-Dy(1)-O(4)	77.78(14)	O(4)-Dy(1)-O(7)	146.16(13)	O(8)-Dy(1)-N(1)	72.91(13)
O(7)#1-Dy(1)-O(4)	82.42(14)	O(5)-Dy(1)-O(7)	142.70(13)	O(4)-Dy(1)-N(1)	73.17(14)
O(8)-Dy(1)-O(4)	145.91(14)	O(1)-Dy(1)-N(2)	77.53(11)	O(5)-Dy(1)-N(1)	102.38(15)
O(1)-Dy(1)-O(5)	76.86(15)	O(3)-Dy(1)-N(2)	140.76(12)	O(7)-Dy(1)-N(1)	114.50(11)
O(3)-Dy(1)-O(5)	123.35(14)	O(7)#1-Dy(1)-N(2)	141.62(12)	N(2)-Dy(1)-N(1)	63.24(12)

Table S3. Selected bond distances (Å) and angles (°) for **2·Dy**.

Dy(1)-O(4)#1	2.4620(18)	Dy(1)-N(1)	2.5169(18)	Dy(1)-O(2)	2.557(2)
Dy(1)-O(4)	2.4620(18)	Dy(1)-N(1)#1	2.5169(18)	Dy(1)-O(2)#1	2.557(2)
Dy(1)-O(1)#1	2.4676(16)	Dy(1)-N(2)#1	2.5243(17)		
Dy(1)-O(1)	2.4676(16)	Dy(1)-N(2)	2.5243(17)		
O(4)#1-Dy(1)-O(4)	51.97(10)	O(4)#1-Dy(1)-N(2)	82.64(7)	O(1)#1-Dy(1)-O(2)#1	50.65(6)
O(4)#1-Dy(1)-O(1)#1	70.25(6)	O(4)-Dy(1)-N(2)	128.94(7)	O(1)-Dy(1)-O(2)#1	124.74(6)
O(4)-Dy(1)-O(1)#1	73.07(6)	O(1)#1-Dy(1)-N(2)	69.99(6)	N(1)#1-Dy(1)-O(2)#1	120.93(6)
O(4)#1-Dy(1)-O(1)	73.07(6)	O(1)-Dy(1)-N(2)	122.47(6)	N(1)-Dy(1)-O(2)#1	68.49(6)
O(4)-Dy(1)-O(1)	70.25(6)	N(1)#1-Dy(1)-N(2)	89.15(6)	N(2)-Dy(1)-O(2)#1	111.20(6)
O(1)#1-Dy(1)-O(1)	139.04(9)	N(1)-Dy(1)-N(2)	64.38(6)	N(2)#1-Dy(1)-O(2)#1	71.91(6)
O(4)#1-Dy(1)-N(1)#1	134.57(7)	O(4)#1-Dy(1)-N(2)#1	128.93(7)	O(4)#1-Dy(1)-O(2)	66.48(7)
O(4)-Dy(1)-N(1)#1	138.00(6)	O(4)-Dy(1)-N(2)#1	82.64(7)	O(4)-Dy(1)-O(2)	103.55(7)
O(1)#1-Dy(1)-N(1)#1	146.45(6)	O(1)#1-Dy(1)-N(2)#1	122.47(6)	O(1)#1-Dy(1)-O(2)	124.75(6)
O(1)-Dy(1)-N(1)#1	74.28(6)	O(1)-Dy(1)-N(2)#1	69.99(6)	O(1)-Dy(1)-O(2)	50.65(6)
O(4)#1-Dy(1)-N(1)	138.01(6)	N(1)#1-Dy(1)-N(2)#1	64.37(6)	N(1)#1-Dy(1)-O(2)	68.49(6)
O(4)-Dy(1)-N(1)	134.57(7)	N(1)-Dy(1)-N(2)#1	89.15(6)	N(1)-Dy(1)-O(2)	120.93(6)
O(1)#1-Dy(1)-N(1)	74.28(6)	N(2)-Dy(1)-N(2)#1	147.68(9)	N(2)-Dy(1)-O(2)	71.91(6)
O(1)-Dy(1)-N(1)	146.45(6)	O(4)#1-Dy(1)-O(2)#1	103.55(7)	N(2)#1-Dy(1)-O(2)	111.19(6)
N(1)#1-Dy(1)-N(1)	73.02(9)	O(4)-Dy(1)-O(2)#1	66.48(7)	O(2)#1-Dy(1)-O(2)	169.48(9)

Table S4. Selected bond distances (Å) and angles (°) for **3·Dy**.

Dy(1)-O(1)	2.301(3)	Dy(1)-O(8)	2.412(3)	Dy(1)-O(4)	2.471(4)
Dy(1)-O(5)	2.513(4)	Dy(1)-O(7)	2.533(3)	Dy(1)-N(2)	2.547(3)
Dy(1)-O(3)#1	2.306(3)	Dy(1)-O(7)#1	2.320(3)	Dy(1)-N(1)	2.579(3)
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O(8)-Dy(1)-O(4)	145.91(14)	O(1)-Dy(1)-O(5)	76.86(15)	O(3)#1-Dy(1)-O(5)	123.35(14)
O(7)#1-Dy(1)-O(5)	77.44(13)	O(8)-Dy(1)-O(5)	143.41(14)	O(4)-Dy(1)-O(5)	49.99(16)
O(1)-Dy(1)-O(7)	72.30(11)	O(3)#1-Dy(1)-O(7)	72.40(11)	O(7)#1-Dy(1)-O(7)	74.89(11)
O(8)-Dy(1)-O(7)	51.91(10)	O(4)-Dy(1)-O(7)	146.16(13)	O(5)#1-Dy(1)-N(1)	142.70(13)
O(1)-Dy(1)-N(2)	77.53(11)	O(3)#1-Dy(1)-N(2)	140.76(12)	O(7)#1-Dy(1)-N(2)	141.62(12)
O(8)-Dy(1)-N(2)	75.63(12)	O(4)-Dy(1)-N(2)	92.11(14)	O(5)-Dy(1)-N(2)	70.33(13)
O(7)-Dy(1)-N(2)	121.19(11)	O(1)-Dy(1)-N(1)	137.77(12)	O(3)#1-Dy(1)-N(1)	77.57(11)
O(7)#1-Dy(1)-N(1)	146.60(12)	O(8)-Dy(1)-N(1)	72.91(13)	O(4)-Dy(1)-N(1)	73.17(14)
O(5)-Dy(1)-N(1)	102.38(15)	O(7)-Dy(1)-N(1)	114.50(11)	N(2)-Dy(1)-N(1)	63.24(12)