

## SUPPORTING INFORMATION

**Title: Synthesis, molecular and electronic structure of a stacked half-sandwich dititanium complex incorporating a cyclic  $\pi$ -faced bridging ligand**

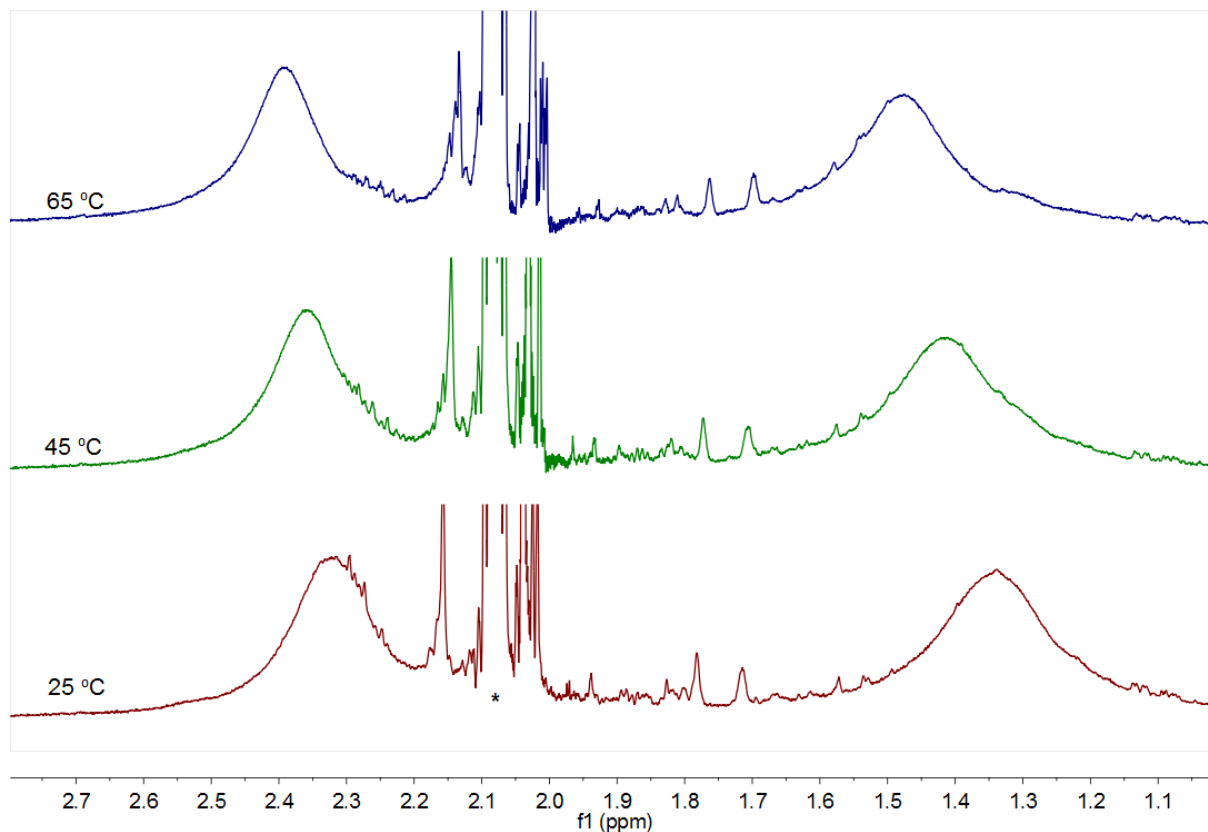
**Authors:** Róbert Gyepes,<sup>b</sup> Jiří Pinkas,<sup>a</sup> Ivana Císařová,<sup>c</sup> Jiří Kubišta,<sup>a</sup> Michal Horáček,<sup>a</sup> and Karel Mach<sup>\*a</sup>

### Contents:

- <sup>1</sup>H NMR spectra of solution of **3** in toluene-*d*<sub>8</sub> measured at variable temperatures.
- Experimental data for 1,2,4,5-tetra(trimethylsilyl)benzene.
- Additional analytical data for **2**.
- Table of crystallographic data and data collection and structure refinement details for **3**.
- CASPT2 energies (in atomic units) for the three lowest roots.

**$^1\text{H}$  NMR spectra of solution of 3 in toluene- $d_8$  measured at variable temperatures.**

Signals at 1.3 ppm and 2.3 ppm were tentatively assigned to  $\text{SiMe}_3$  and  $\text{C}_5\text{Me}_5$  groups. The asterisk denotes signal of the solvent.



**Experimental data for 1,2,4,5-tetra(trimethylsilyl)benzene.**

1,2,4,5-tetra(trimethylsilyl)benzene were prepared accordingly to literature. [H. Bock, M. Ansari, N. Nagel and Z. Havlas, *J. Organomet. Chem.* 1995, **499**, 63–71.]

$^1\text{H}$  NMR (toluene- $d_8$ ): 0.38 (s, 36 H,  $\text{SiMe}_3$ ); 8.11 (s, 2H,  $\text{C}_6\text{H}_2$ ).

GC-MS ( $m/z$ , abundance): 366 ( $[\text{M}]^+$ , 30); 351 ( $[\text{M} - \text{Me}]^+$ , 100); 335 (93); 263 (66); 152 (18); 73 ( $[\text{SiMe}_3]^+$ , 67).

**Additional analytical data for 2.**

Mp. > 360 °C. EI-MS (380 °C):  $m/z$  (relative abundance) 789 (15), 788 (24), 787 (37), 786 (38), 785 (40), 784 ( $M^+$ ; 39), 783 (23), 782 (17), 781 (12), 392 ( $M^{2+}$ ; 31), 183 (9), 169 (13), 136 (10), 135 ( $[Cp^*]^+$ ; 16), 121 (22), 119 (36), 105 (23), 91 (31), 85 (16), 83 (12), 81 (10), 77 (12), 71 (16), 69 (20), 67 (13), 57 (31), 55 (34). The isotope peaks distribution pattern for  $m/z$  784 corresponded to theory for  $C_{44}H_{64}Ti_4$ . IR (KBr,  $cm^{-1}$ ): 2972 (m), 2904 (vs), 2855 (s), 2719 (vw), 1490 (w), 1432 (m,b), 1373 (s), 1064 (vw), 1024 (m), 801 (vw), 597 (s), 584 (vs), 506 (m), 476 (w), 450 (w), 416 (m). The reported in ref.<sup>5</sup> strong absorption band at  $790\text{ cm}^{-1}$  is to be attributed to an oxygenation impurity since it appeared to grow in intensity after exposition of the KBr pellet to air while intensities of a very strong band at  $584\text{ cm}^{-1}$  and a strong band at  $597\text{ cm}^{-1}$  were decreasing. UV-near IR (m-xylene,  $d = 1.0\text{ cm}$ ): 310 > 530 >> 950 nm.

### Crystal data and structure refinement for 3.

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Empirical formula	C38 H68 Si4 Ti2
Formula weight	733.08
Temperature	150(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P 21/n
Unit cell dimensions	a = 11.8817(3) Å    alpha = 90 deg. b = 13.3685(4) Å    beta = 99.0810(10) deg. c = 12.8886(3) Å    gamma = 90 deg.
Volume	2021.57(9) Å <sup>3</sup>
Z, Calculated density	2, 1.204 Mg/m <sup>3</sup>
Absorption coefficient	0.538 mm <sup>-1</sup>
F(000)	792
Crystal size	0.70 x 0.63 x 0.46 mm
Theta range for data collection	2.21 to 27.50 deg.
Limiting indices	-15<=h<=14, -17<=k<=17, -12<=l<=16
Reflections collected / unique	33636 / 4649 [R(int) = 0.0173]
Completeness to theta = 27.50	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7918 and 0.7061
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	4649 / 0 / 210
Goodness-of-fit on F <sup>2</sup>	0.947
Final R indices [I>2sigma(I)]	R1 = 0.0235, wR2 = 0.0652
R indices (all data)	R1 = 0.0271, wR2 = 0.0693
Largest diff. peak and hole	0.322 and -0.265 e.Å <sup>-3</sup>

Bond distances (Å) for **3**.

Atoms	distances
Ti(1)-C(13)	2.2229(11)
Ti(1)-C(13)#1	2.2621(11)
Ti(1)-C(11)	2.2809(11)
Ti(1)-C(12)#1	2.2851(10)
Ti(1)-C(12)	2.3155(11)
Ti(1)-C(11)#1	2.3184(10)
Ti(1)-C(5)	2.3946(11)
Ti(1)-C(4)	2.4234(12)
Ti(1)-C(1)	2.4313(11)
Ti(1)-C(3)	2.4428(11)
Ti(1)-C(2)	2.4500(11)
Si(1)-C(15)	1.8673(13)
Si(1)-C(16)	1.8747(14)
Si(1)-C(14)	1.8800(13)
Si(1)-C(11)	1.8889(11)
Si(2)-C(18)	1.8681(13)
Si(2)-C(17)	1.8727(14)
Si(2)-C(19)	1.8779(14)
Si(2)-C(12)	1.8871(11)
C(1)-C(5)	1.4179(17)
C(1)-C(2)	1.4182(16)
C(1)-C(6)	1.5061(17)
C(2)-C(3)	1.4136(18)
C(2)-C(7)	1.5062(17)
C(3)-C(4)	1.4190(17)
C(3)-C(8)	1.5028(17)
C(4)-C(5)	1.4240(17)
C(4)-C(9)	1.5024(17)
C(5)-C(10)	1.5011(16)
C(6)-H(6A)	0.9800
C(6)-H(6B)	0.9800
C(6)-H(6C)	0.9800
C(7)-H(7A)	0.9800
C(7)-H(7B)	0.9800
C(7)-H(7C)	0.9800
C(8)-H(8A)	0.9800
C(8)-H(8B)	0.9800
C(8)-H(8C)	0.9800
C(9)-H(9A)	0.9800
C(9)-H(9B)	0.9800
C(9)-H(9C)	0.9800
C(10)-H(10A)	0.9800
C(10)-H(10B)	0.9800
C(10)-H(10C)	0.9800
C(11)-C(13)#1	1.4735(15)
C(11)-C(12)	1.4896(15)
C(11)-Ti(1)#1	2.3184(10)
C(12)-C(13)	1.4720(14)
C(12)-Ti(1)#1	2.2851(10)
C(13)-C(11)#1	1.4735(15)
C(13)-Ti(1)#1	2.2621(11)
C(13)-H(13)	0.9500
C(14)-H(14A)	0.9800
C(14)-H(14B)	0.9800

Bond distances (Å) for **3**.

Atoms	distances
C(14)-H(14C)	0.9800
C(15)-H(15A)	0.9800
C(15)-H(15B)	0.9800
C(15)-H(15C)	0.9800
C(16)-H(16A)	0.9800
C(16)-H(16B)	0.9800
C(16)-H(16C)	0.9800
C(17)-H(17A)	0.9800
C(17)-H(17B)	0.9800
C(17)-H(17C)	0.9800
C(18)-H(18A)	0.9800
C(18)-H(18B)	0.9800
C(18)-H(18C)	0.9800
C(19)-H(19A)	0.9800
C(19)-H(19B)	0.9800
C(19)-H(19C)	0.9800
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Ti(1)-Cg(1)	2.1076(6)
Ti(1)-RING(1)	2.1068(2)
Ti(1)-Cg(2)	1.7381(5)
Ti(2)-RING(2)	1.7381(2)

Cg(2) = C(11), C(12), C(13), C(11)a, C(12)a, C(13)a

Cg(1) = C(1), C(2), C(3), C(4), C(5)

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Symmetry transformations used to generate equivalent atoms:  
#1 -x+2,-y,-z+1

Valence angles (°) for **3**.

Atoms	angle
C(13)-Ti(1)-C(13)#1	78.38(4)
C(13)-Ti(1)-C(11)	68.22(4)
C(13)#1-Ti(1)-C(11)	37.85(4)
C(13)-Ti(1)-C(12)#1	68.21(4)
C(13)#1-Ti(1)-C(12)#1	37.77(4)
C(11)-Ti(1)-C(12)#1	69.99(4)
C(13)-Ti(1)-C(12)	37.78(4)
C(13)#1-Ti(1)-C(12)	67.03(4)
C(11)-Ti(1)-C(12)	37.81(4)
C(12)#1-Ti(1)-C(12)	81.85(4)
C(13)-Ti(1)-C(11)#1	37.79(4)
C(13)#1-Ti(1)-C(11)#1	66.93(4)
C(11)-Ti(1)-C(11)#1	81.81(4)
C(12)#1-Ti(1)-C(11)#1	37.75(4)
C(12)-Ti(1)-C(11)#1	68.82(4)
C(13)-Ti(1)-C(5)	106.99(4)
C(13)#1-Ti(1)-C(5)	174.61(4)
C(11)-Ti(1)-C(5)	144.06(4)
C(12)#1-Ti(1)-C(5)	143.52(4)
C(12)-Ti(1)-C(5)	117.37(4)
C(11)#1-Ti(1)-C(5)	117.07(4)
C(13)-Ti(1)-C(4)	122.77(4)
C(13)#1-Ti(1)-C(4)	142.15(4)
C(11)-Ti(1)-C(4)	168.47(4)
C(12)#1-Ti(1)-C(4)	115.94(4)
C(12)-Ti(1)-C(4)	149.25(4)
C(11)#1-Ti(1)-C(4)	108.99(4)
C(5)-Ti(1)-C(4)	34.38(4)
C(13)-Ti(1)-C(1)	121.66(4)
C(13)#1-Ti(1)-C(1)	143.01(4)
C(11)-Ti(1)-C(1)	116.08(4)
C(12)#1-Ti(1)-C(1)	169.40(4)
C(12)-Ti(1)-C(1)	108.32(4)
C(11)#1-Ti(1)-C(1)	148.23(4)
C(5)-Ti(1)-C(1)	34.16(4)
C(4)-Ti(1)-C(1)	56.51(4)
C(13)-Ti(1)-C(3)	156.25(4)
C(13)#1-Ti(1)-C(3)	118.16(4)
C(11)-Ti(1)-C(3)	135.41(4)
C(12)#1-Ti(1)-C(3)	113.35(4)
C(12)-Ti(1)-C(3)	161.54(4)
C(11)#1-Ti(1)-C(3)	129.60(4)
C(5)-Ti(1)-C(3)	56.65(4)
C(4)-Ti(1)-C(3)	33.90(4)
C(1)-Ti(1)-C(3)	56.10(4)
C(13)-Ti(1)-C(2)	154.72(4)
C(13)#1-Ti(1)-C(2)	118.55(4)
C(11)-Ti(1)-C(2)	112.49(4)
C(12)#1-Ti(1)-C(2)	136.83(4)
C(12)-Ti(1)-C(2)	127.96(4)
C(11)#1-Ti(1)-C(2)	163.12(4)
C(5)-Ti(1)-C(2)	56.54(4)
C(4)-Ti(1)-C(2)	56.20(4)
C(1)-Ti(1)-C(2)	33.78(4)

Valence angles (°) for **3**.

Atoms	angle
C(3)-Ti(1)-C(2)	33.59(4)
C(15)-Si(1)-C(16)	107.12(7)
C(15)-Si(1)-C(14)	108.39(6)
C(16)-Si(1)-C(14)	104.09(7)
C(15)-Si(1)-C(11)	112.65(5)
C(16)-Si(1)-C(11)	116.55(6)
C(14)-Si(1)-C(11)	107.48(5)
C(18)-Si(2)-C(17)	106.92(7)
C(18)-Si(2)-C(19)	108.86(6)
C(17)-Si(2)-C(19)	103.10(7)
C(18)-Si(2)-C(12)	111.27(6)
C(17)-Si(2)-C(12)	117.96(6)
C(19)-Si(2)-C(12)	108.18(5)
C(5)-C(1)-C(2)	108.03(10)
C(5)-C(1)-C(6)	126.04(11)
C(2)-C(1)-C(6)	124.35(11)
C(5)-C(1)-Ti(1)	71.50(6)
C(2)-C(1)-Ti(1)	73.83(6)
C(6)-C(1)-Ti(1)	131.68(8)
C(3)-C(2)-C(1)	108.06(10)
C(3)-C(2)-C(7)	126.01(12)
C(1)-C(2)-C(7)	124.35(12)
C(3)-C(2)-Ti(1)	72.93(7)
C(1)-C(2)-Ti(1)	72.39(6)
C(7)-C(2)-Ti(1)	131.68(8)
C(2)-C(3)-C(4)	108.27(10)
C(2)-C(3)-C(8)	125.15(12)
C(4)-C(3)-C(8)	125.17(13)
C(2)-C(3)-Ti(1)	73.49(7)
C(4)-C(3)-Ti(1)	72.30(6)
C(8)-C(3)-Ti(1)	130.74(8)
C(3)-C(4)-C(5)	107.68(11)
C(3)-C(4)-C(9)	124.84(12)
C(5)-C(4)-C(9)	125.67(12)
C(3)-C(4)-Ti(1)	73.79(7)
C(5)-C(4)-Ti(1)	71.70(6)
C(9)-C(4)-Ti(1)	132.12(8)
C(1)-C(5)-C(4)	107.93(10)
C(1)-C(5)-C(10)	125.45(11)
C(4)-C(5)-C(10)	126.33(12)
C(1)-C(5)-Ti(1)	74.34(6)
C(4)-C(5)-Ti(1)	73.92(6)
C(10)-C(5)-Ti(1)	122.45(8)
C(1)-C(6)-H(6A)	109.5
C(1)-C(6)-H(6B)	109.5
H(6A)-C(6)-H(6B)	109.5
C(1)-C(6)-H(6C)	109.5
H(6A)-C(6)-H(6C)	109.5
H(6B)-C(6)-H(6C)	109.5
C(2)-C(7)-H(7A)	109.5
C(2)-C(7)-H(7B)	109.5
H(7A)-C(7)-H(7B)	109.5
C(2)-C(7)-H(7C)	109.5
H(7A)-C(7)-H(7C)	109.5



Valence angles (°) for **3**.

Atoms	angle
H(7B)-C(7)-H(7C)	109.5
C(3)-C(8)-H(8A)	109.5
C(3)-C(8)-H(8B)	109.5
H(8A)-C(8)-H(8B)	109.5
C(3)-C(8)-H(8C)	109.5
H(8A)-C(8)-H(8C)	109.5
H(8B)-C(8)-H(8C)	109.5
C(4)-C(9)-H(9A)	109.5
C(4)-C(9)-H(9B)	109.5
H(9A)-C(9)-H(9B)	109.5
C(4)-C(9)-H(9C)	109.5
H(9A)-C(9)-H(9C)	109.5
H(9B)-C(9)-H(9C)	109.5
C(5)-C(10)-H(10A)	109.5
C(5)-C(10)-H(10B)	109.5
H(10A)-C(10)-H(10B)	109.5
C(5)-C(10)-H(10C)	109.5
H(10A)-C(10)-H(10C)	109.5
H(10B)-C(10)-H(10C)	109.5
C(13)#1-C(11)-C(12)	117.12(9)
C(13)#1-C(11)-Si(1)	112.96(8)
C(12)-C(11)-Si(1)	129.73(8)
C(13)#1-C(11)-Ti(1)	70.38(6)
C(12)-C(11)-Ti(1)	72.36(6)
Si(1)-C(11)-Ti(1)	122.99(5)
C(13)#1-C(11)-Ti(1)#1	67.59(6)
C(12)-C(11)-Ti(1)#1	69.91(6)
Si(1)-C(11)-Ti(1)#1	137.23(6)
Ti(1)-C(11)-Ti(1)#1	98.19(4)
C(13)-C(12)-C(11)	117.06(9)
C(13)-C(12)-Si(2)	113.49(8)
C(11)-C(12)-Si(2)	129.29(8)
C(13)-C(12)-Ti(1)#1	70.27(6)
C(11)-C(12)-Ti(1)#1	72.34(6)
Si(2)-C(12)-Ti(1)#1	123.48(5)
C(13)-C(12)-Ti(1)	67.70(6)
C(11)-C(12)-Ti(1)	69.83(6)
Si(2)-C(12)-Ti(1)	136.93(5)
Ti(1)#1-C(12)-Ti(1)	98.15(4)
C(12)-C(13)-C(11)#1	125.52(9)
C(12)-C(13)-Ti(1)	74.52(6)
C(11)#1-C(13)-Ti(1)	74.62(6)
C(12)-C(13)-Ti(1)#1	71.96(6)
C(11)#1-C(13)-Ti(1)#1	71.77(6)
Ti(1)-C(13)-Ti(1)#1	101.62(4)
C(12)-C(13)-H(13)	117.2
C(11)#1-C(13)-H(13)	117.2
Ti(1)-C(13)-H(13)	125.5
Ti(1)#1-C(13)-H(13)	132.8
Si(1)-C(14)-H(14A)	109.5
Si(1)-C(14)-H(14B)	109.5
H(14A)-C(14)-H(14B)	109.5
Si(1)-C(14)-H(14C)	109.5
H(14A)-C(14)-H(14C)	109.5

Valence angles (°) for **3**.

Atoms	angle
H(14B)-C(14)-H(14C)	109.5
Si(1)-C(15)-H(15A)	109.5
Si(1)-C(15)-H(15B)	109.5
H(15A)-C(15)-H(15B)	109.5
Si(1)-C(15)-H(15C)	109.5
H(15A)-C(15)-H(15C)	109.5
H(15B)-C(15)-H(15C)	109.5
Si(1)-C(16)-H(16A)	109.5
Si(1)-C(16)-H(16B)	109.5
H(16A)-C(16)-H(16B)	109.5
Si(1)-C(16)-H(16C)	109.5
H(16A)-C(16)-H(16C)	109.5
H(16B)-C(16)-H(16C)	109.5
Si(2)-C(17)-H(17A)	109.5
Si(2)-C(17)-H(17B)	109.5
H(17A)-C(17)-H(17B)	109.5
Si(2)-C(17)-H(17C)	109.5
H(17A)-C(17)-H(17C)	109.5
H(17B)-C(17)-H(17C)	109.5
Si(2)-C(18)-H(18A)	109.5
Si(2)-C(18)-H(18B)	109.5
H(18A)-C(18)-H(18B)	109.5
Si(2)-C(18)-H(18C)	109.5
H(18A)-C(18)-H(18C)	109.5
H(18B)-C(18)-H(18C)	109.5
Si(2)-C(19)-H(19A)	109.5
Si(2)-C(19)-H(19B)	109.5
H(19A)-C(19)-H(19B)	109.5
Si(2)-C(19)-H(19C)	109.5
H(19A)-C(19)-H(19C)	109.5
H(19B)-C(19)-H(19C)	109.5
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Cg(1)-Ti(1)-Cg(2)	176.85(2)
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Least-squares planes and atomic deviations therefrom for **3**.

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Plane No.      Formed by \* atoms and distances (Å) of some atoms from the plane

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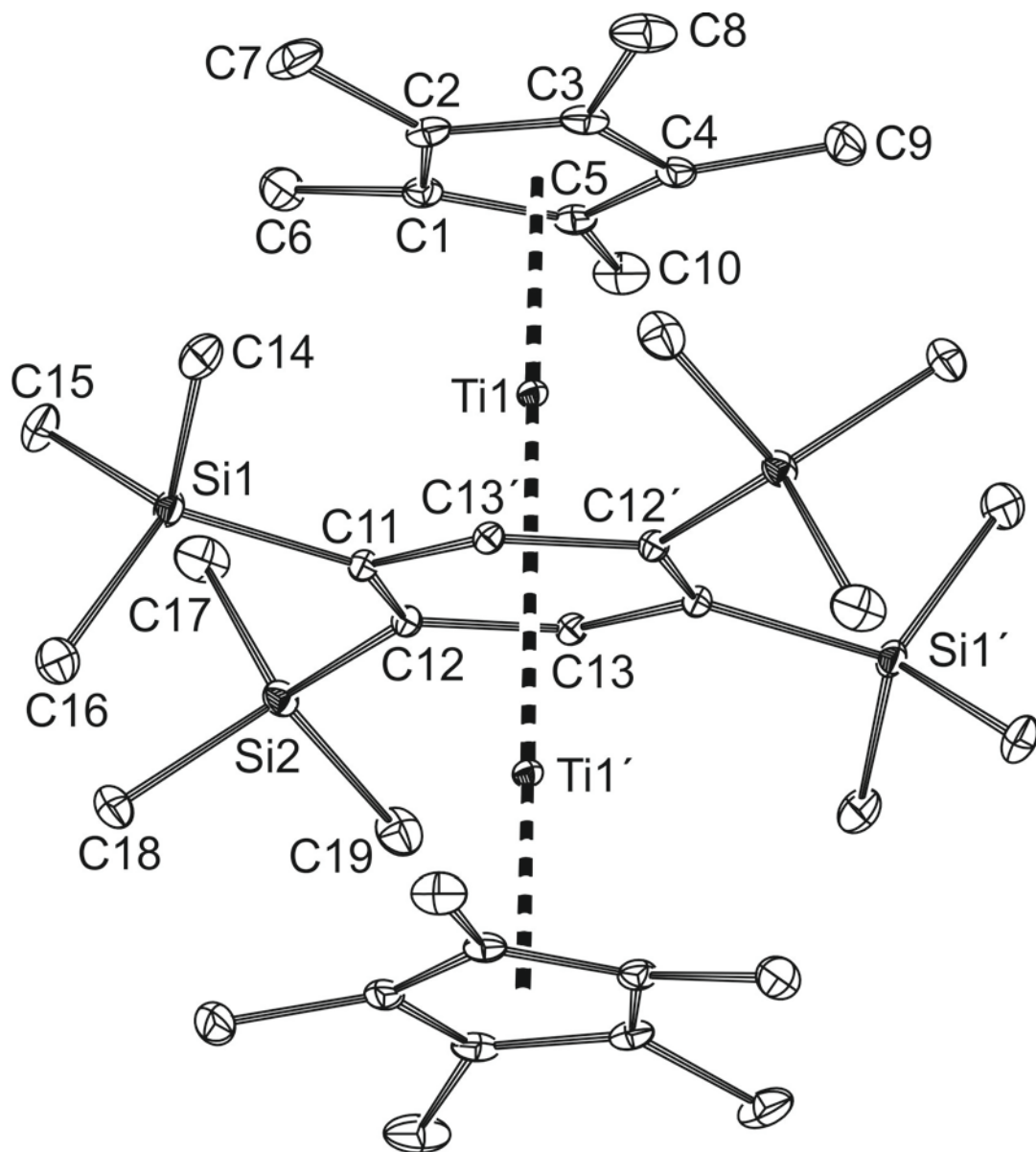
1. Ring	*	-0.0072	(0.0007)	C1
	*	0.0024	(0.0007)	C2
	*	0.0033	(0.0007)	C3
	*	-0.0076	(0.0007)	C4
	*	0.0092	(0.0007)	C5
		2.1068	(0.0005)	Ti1
		-0.3228	(0.0021)	C6
		-0.2823	(0.0021)	C7
		-0.2629	(0.0022)	C8
		-0.3427	(0.0021)	C9
	-0.0855	(0.0021)	C10	
2. Ring	*	-0.0230	(0.0006)	C11
	*	0.0230	(0.0006)	C12
	*	-0.0242	(0.0006)	C13
	*	0.0230	(0.0006)	C11_\$1
	*	-0.0230	(0.0006)	C12_\$1
	*	0.0242	(0.0006)	C13_\$1
		-1.7381	(0.0002)	Ti1
		-0.2713	(0.0020)	Si1
		0.2606	(0.0020)	Si2
		0.2713	(0.0020)	Si1_\$1
	-0.2606	(0.0020)	Si2_\$1	
3. Plane	*	-0.0260	(0.0003)	Si1
	*	0.0686	(0.0007)	C11
	*	-0.0685	(0.0007)	C12
	*	0.0259	(0.0003)	Si2

The angles between the planes are:

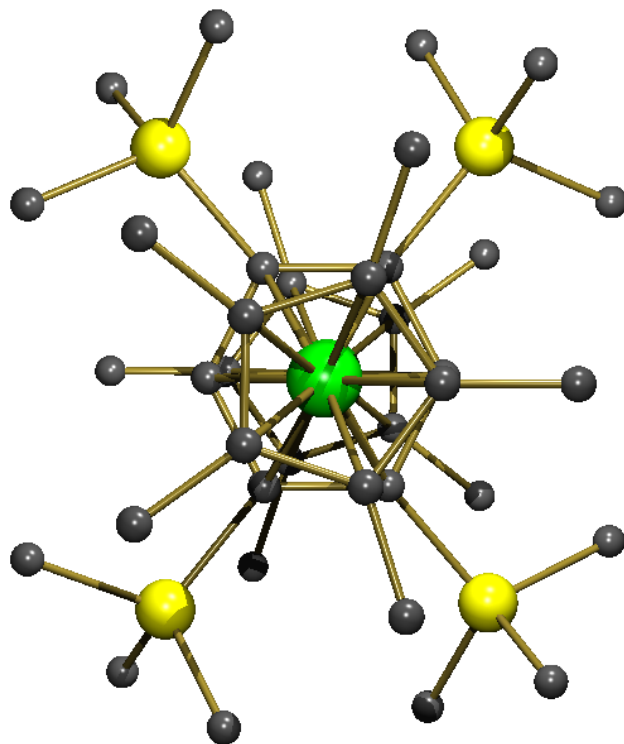
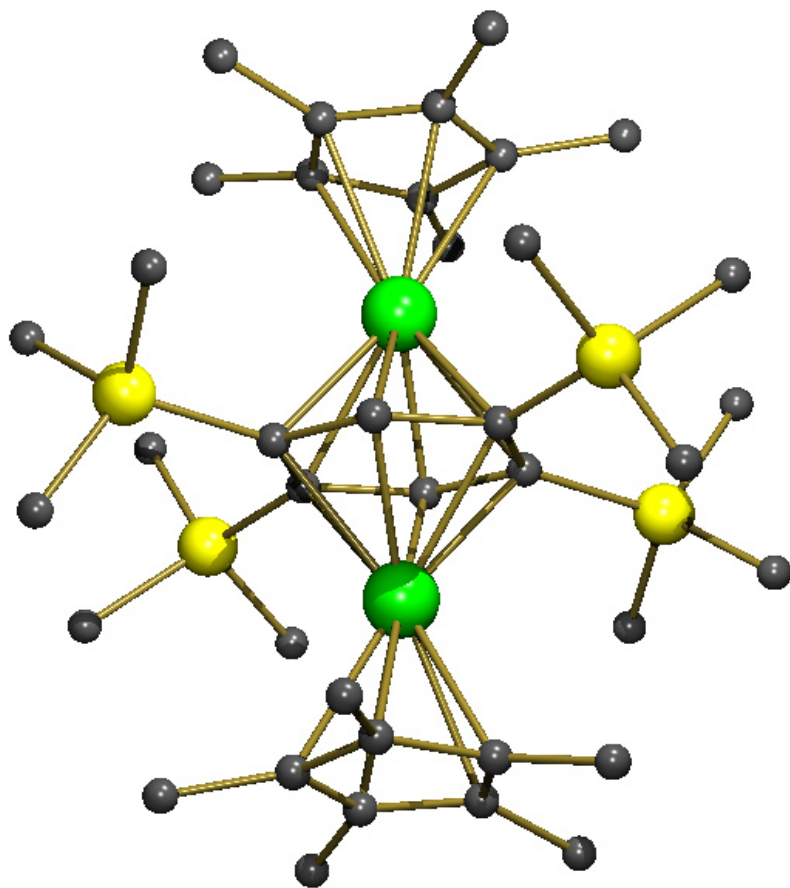
$$1 / 2 = 4.62 ( 0.09 )$$

$$2 / 3 = 7.05 ( 0.07 )$$

PLATON drawing with 30% probability for 3.



PLUTON drawing for 3.



**CASPT2 energies (in atomic units) for the three lowest roots.**

CASPT2 Root No.	Singlet Multiplicity	Triplet Multiplicity
1	-4329.09271924	-4329.05107072
2	-4329.03115037	-4329.05069110
3	-4329.02916731	-4329.01893572

CASPT2 energies (in atomic units) for the three lowest roots for 3. Energy values were devised on the solid state geometry and assuming singlet and triplet state for the molecule.