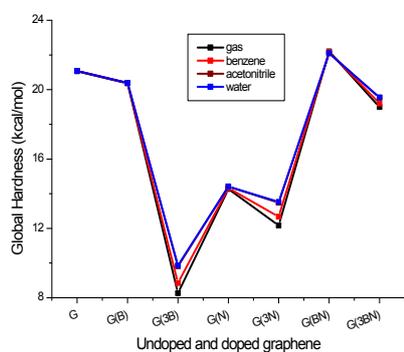
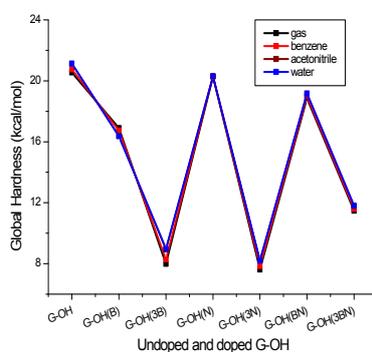


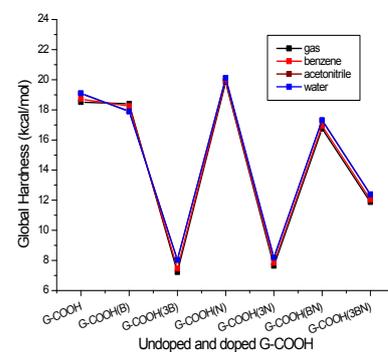
Electronic supplementary information



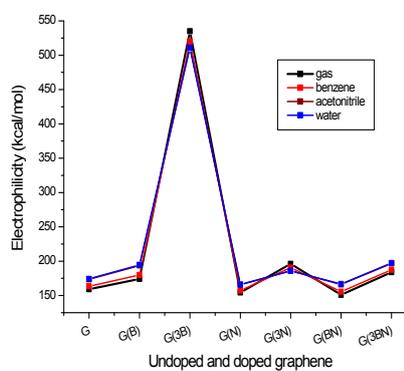
(a)



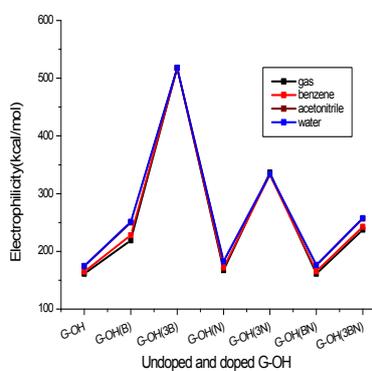
(b)



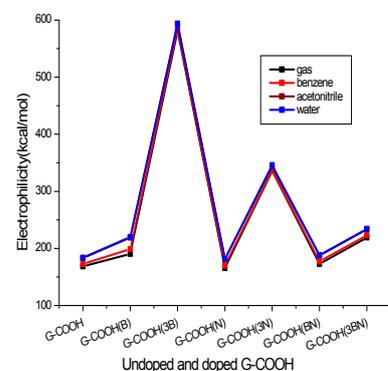
(c)



(d)

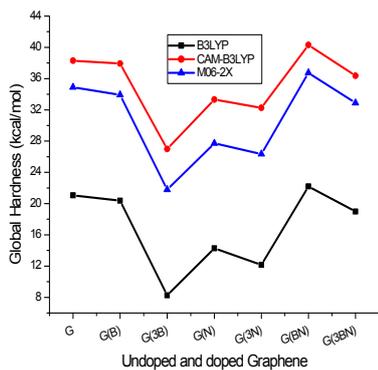


(e)

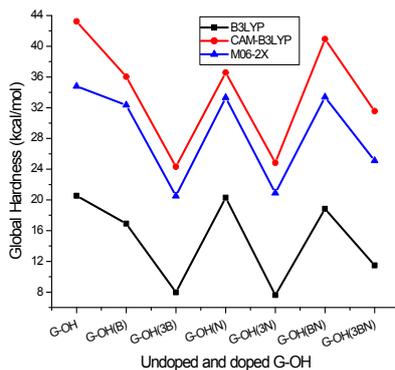


(f)

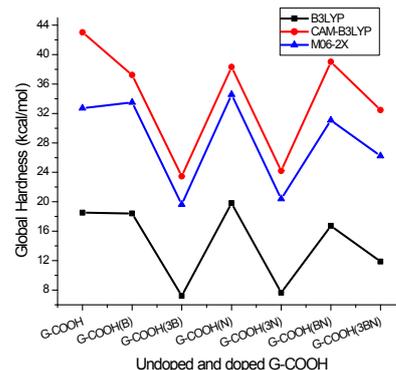
Supplementary Fig. S1 Variation in global hardness (a-c) and electrophilicity (d-f) of undoped and doped graphene, G-OH and G-COOH in gas and solvent phase at B3LYP/6-31G(d) level of theory



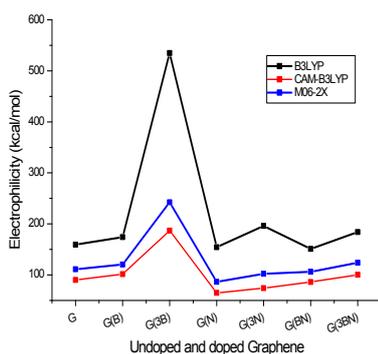
(a)



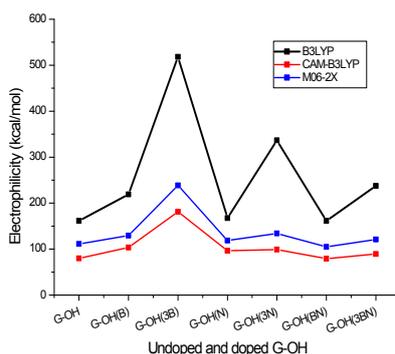
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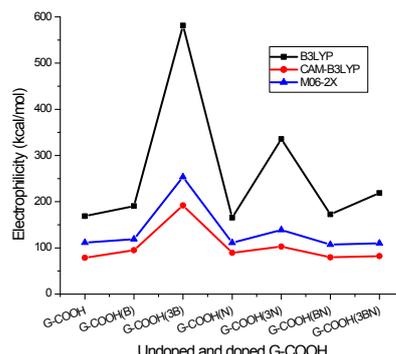
(c)



(d)

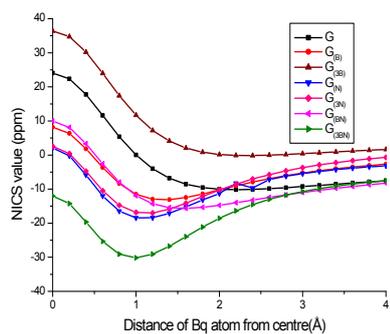


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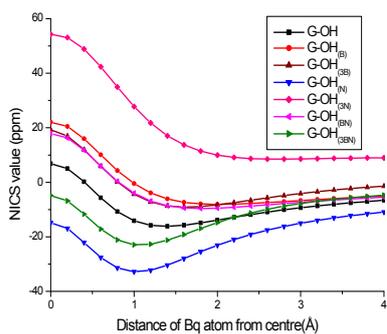


(f)

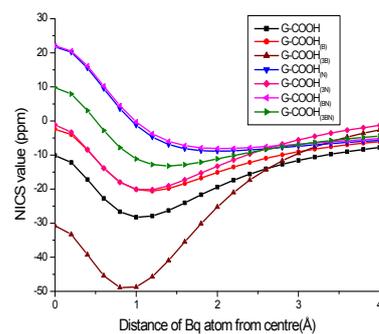
Supplementary Fig. S2 Variation in global hardness (a-c) and electrophilicity (d-f) of undoped and doped graphene, G-OH and G-COOH in gas phase using B3LYP, CAMB3LYP and M062X functional with 6-31G(d) basis set



(a)

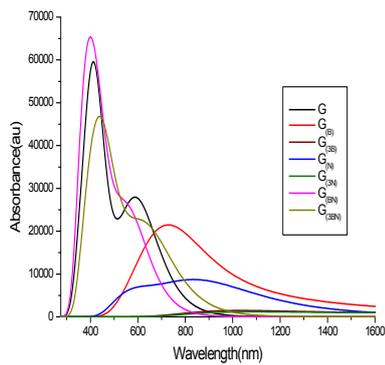


(b)

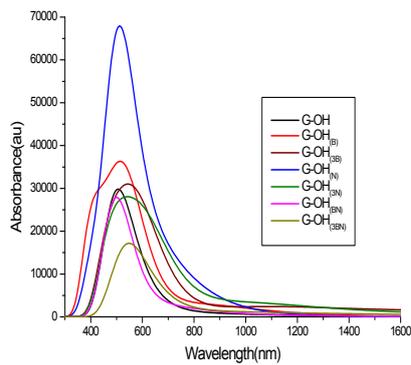


(c)

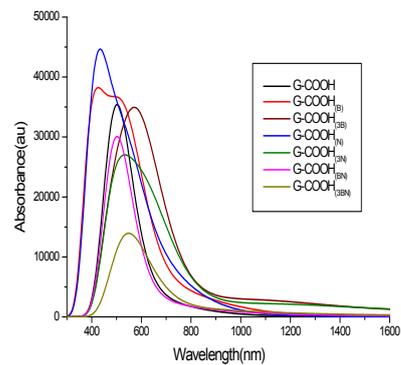
Supplementary Fig. S3 Variation in aromaticity ($NICS_{zz}$) at the edge of the undoped and doped (a) graphene (b) G–OH and (c) G–COOH at B3LYP/6-31G(d) level of theory.



(a)

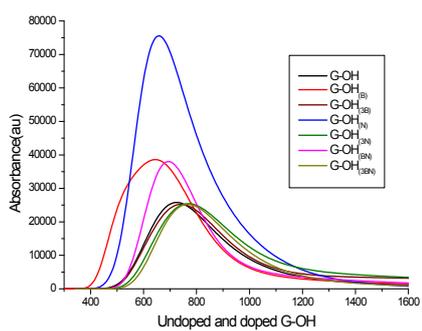


(b)

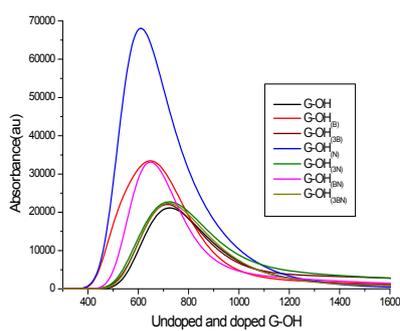


(c)

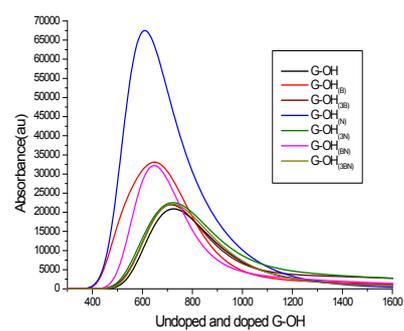
Supplementary Fig. S4 UV-Visible absorption spectra of undoped and doped (a) graphene, (b) G–OH and (c) G–COOH in gas phase at CAM-B3LYP/6-31G(d) level of theory



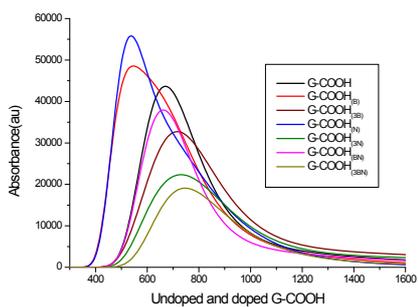
(a)



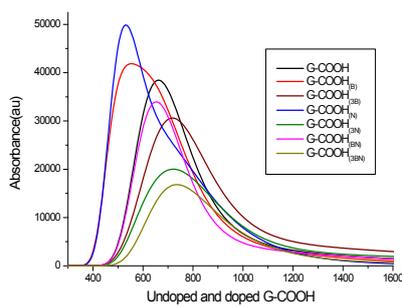
(b)



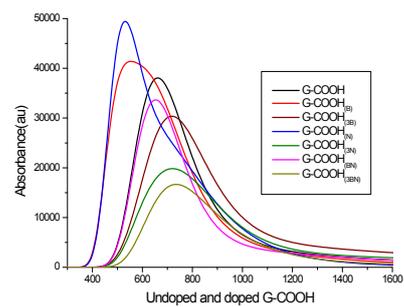
(a)



(d)



(e)



(f)

Supplementary Fig. S5 UV-Visible absorption spectra of undoped and doped G-OH in (a) benzene ($\epsilon=2.27$) (b) acetonitrile ($\epsilon=35.69$) (c) water ($\epsilon=78.35$) and G-COOH in (d) benzene ($\epsilon=2.27$) (e) acetonitrile ($\epsilon=35.69$) (f) water ($\epsilon=78.35$) at B3LYP/6-31G(d) level of theory