

SUPPORTING INFORMATION

Mild base-promoted benzannulation for the construction of polyfunctionalized *m*-terphenyl derivatives and their application as potent UV-filters

Ramuel John Inductivo Tamargo, Tej Narayan Poudel and Yong Rok Lee*

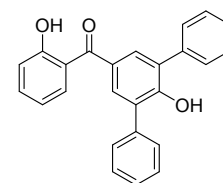
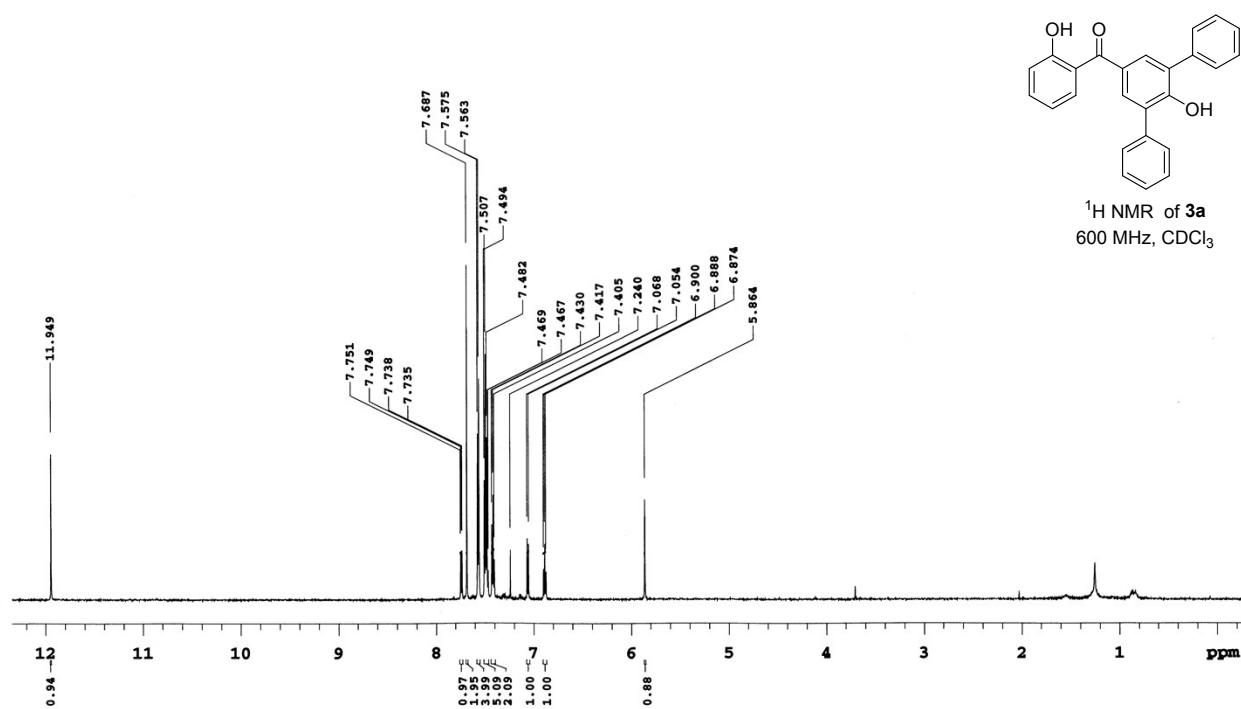
School of Chemical Engineering, Yeungnam University, Gyeongsan 712-749, Republic of Korea

Email: yrlee@yu.ac.kr, Phone: +82-53-810-2529; Fax: +82-53-810-4631

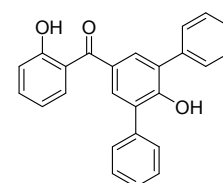
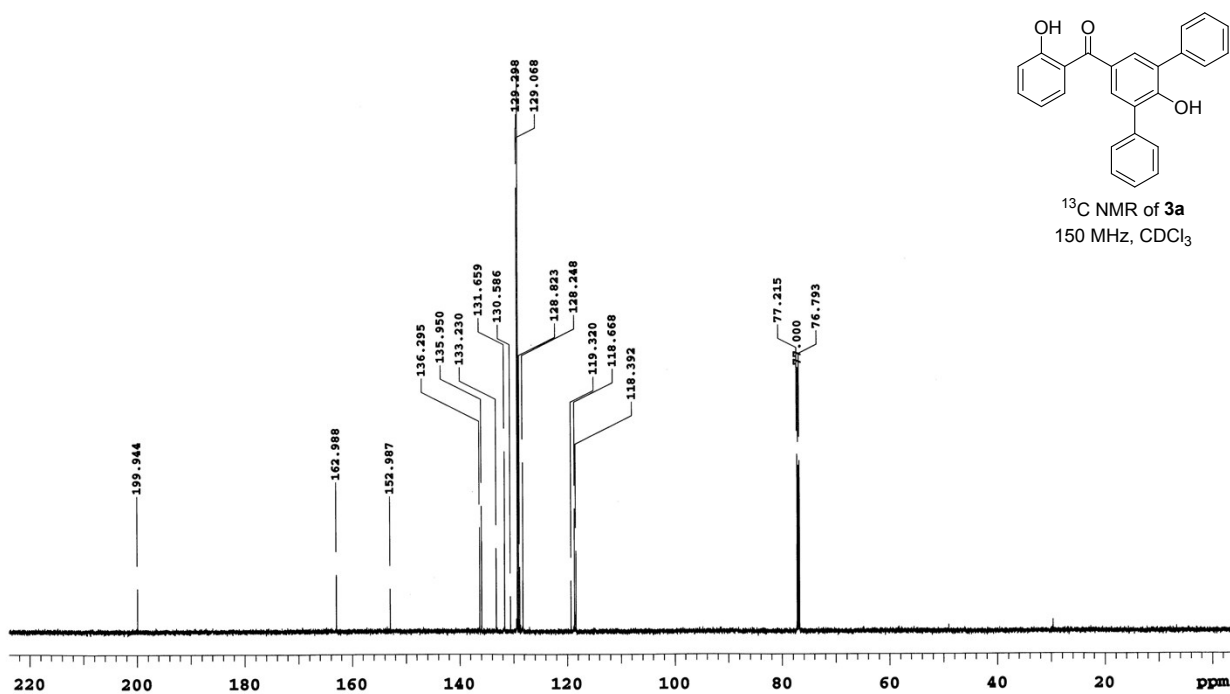
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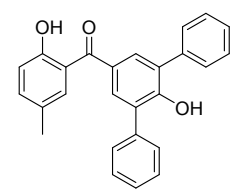
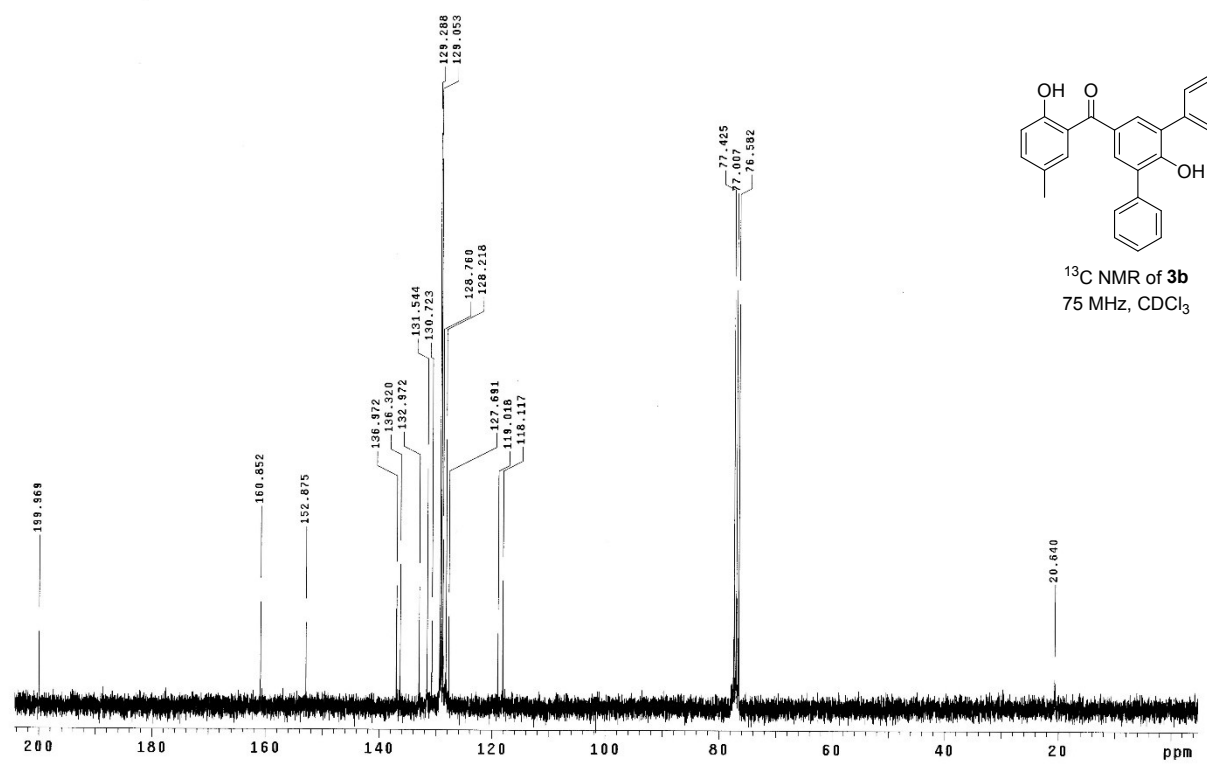
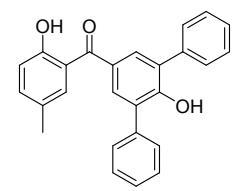
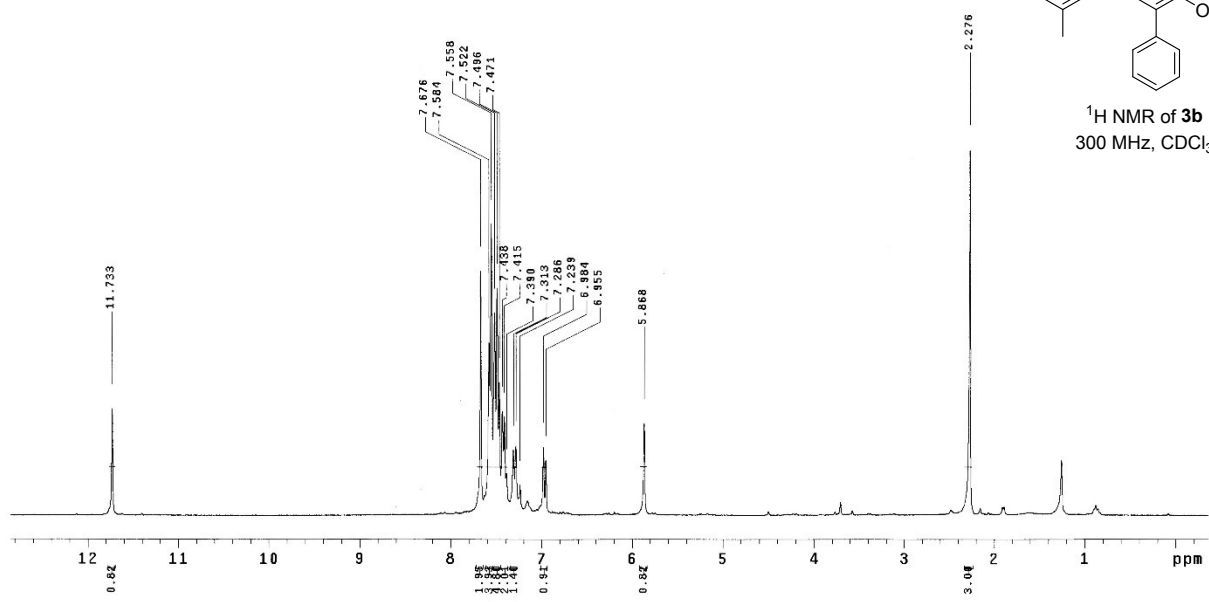
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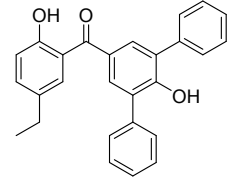


¹H NMR of 3a
600 MHz, CDCl₃

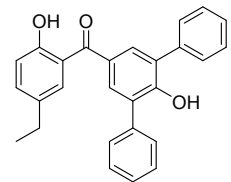
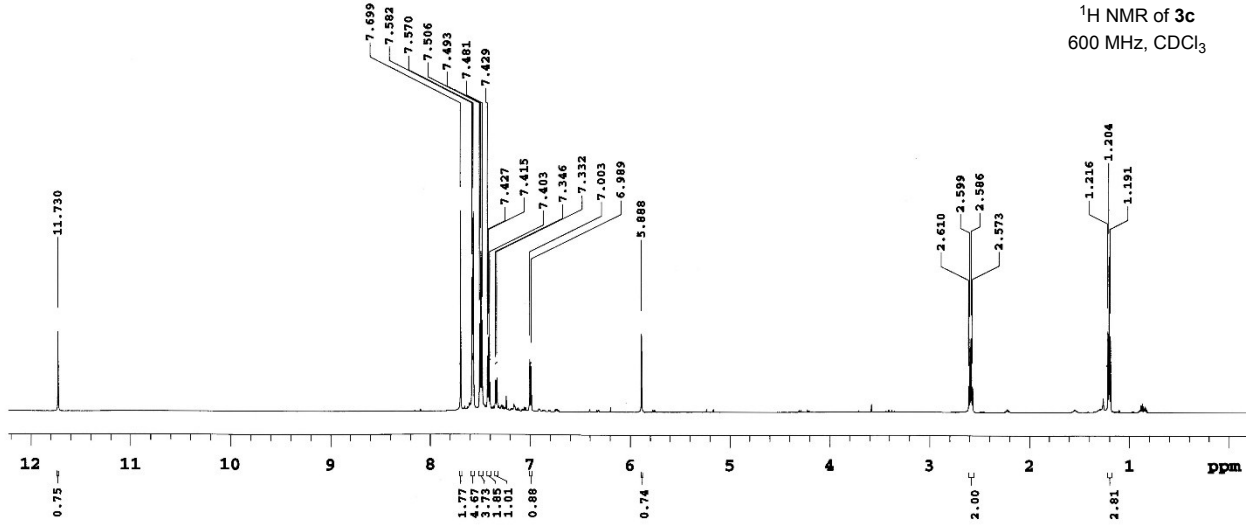


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150 MHz, CDCl₃

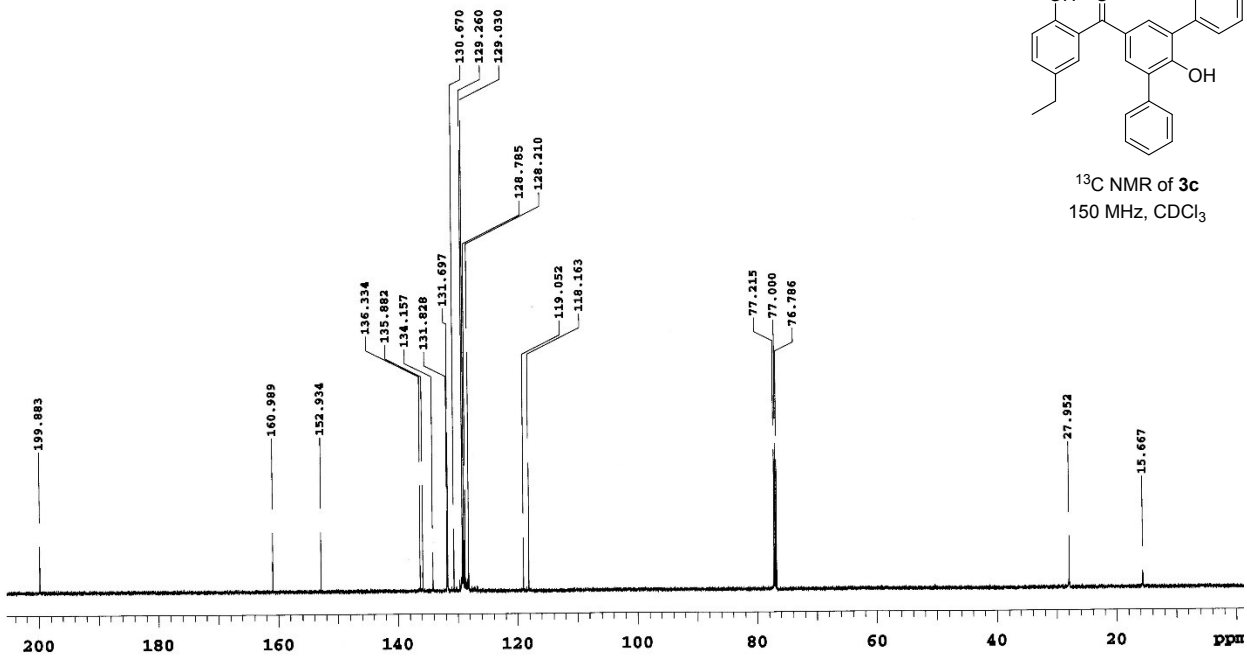


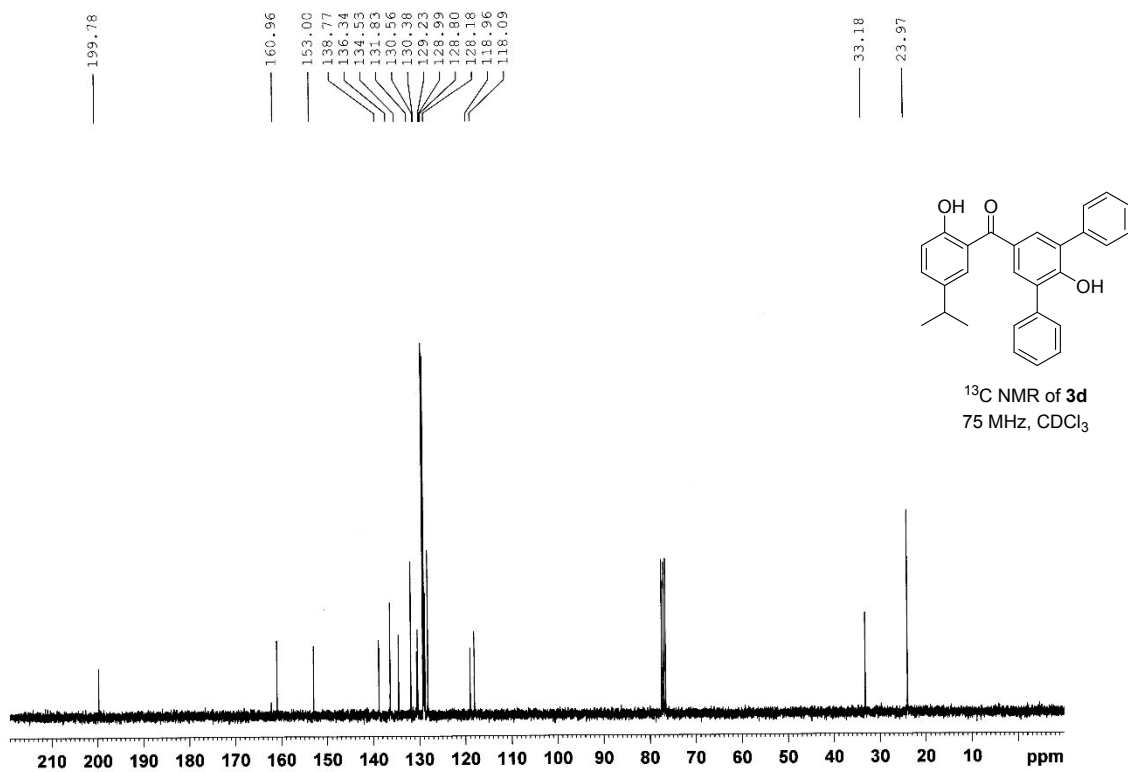
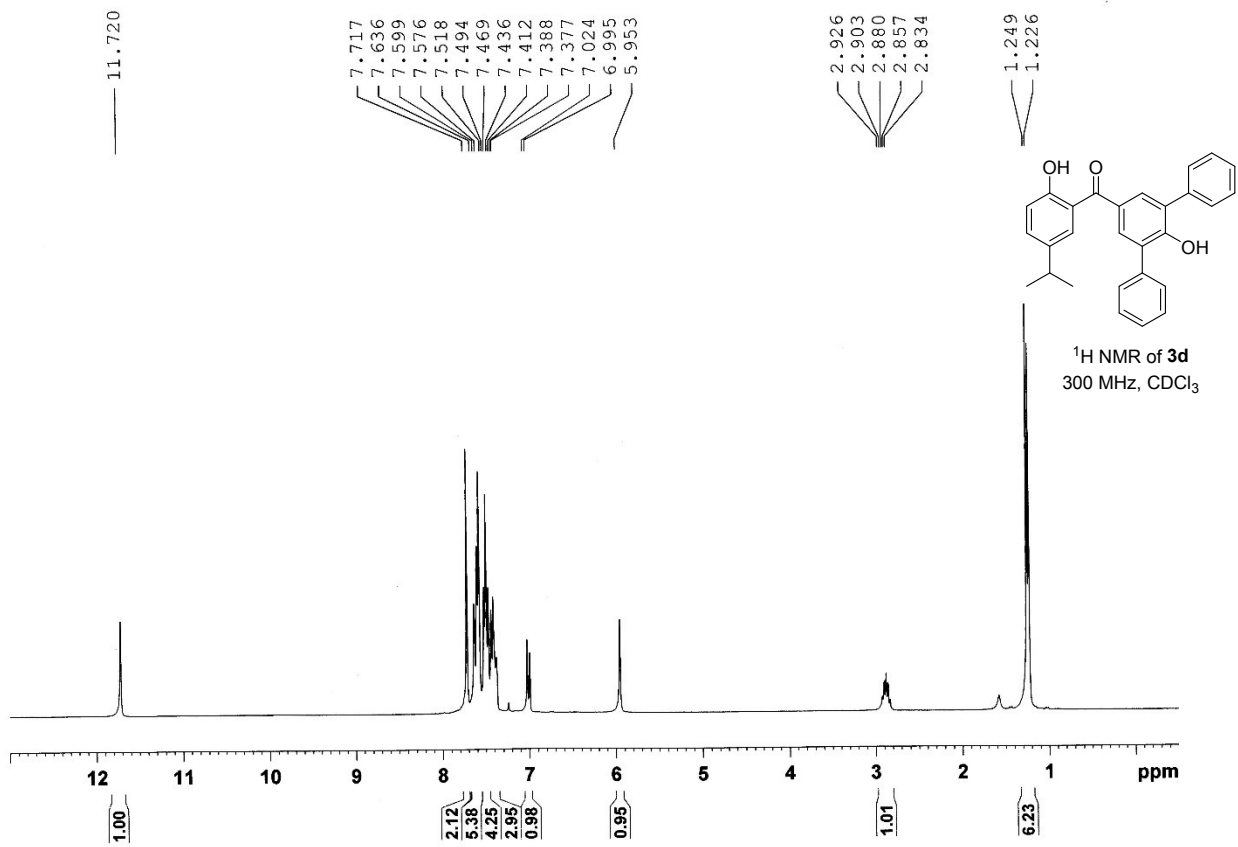


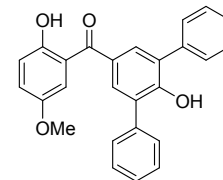
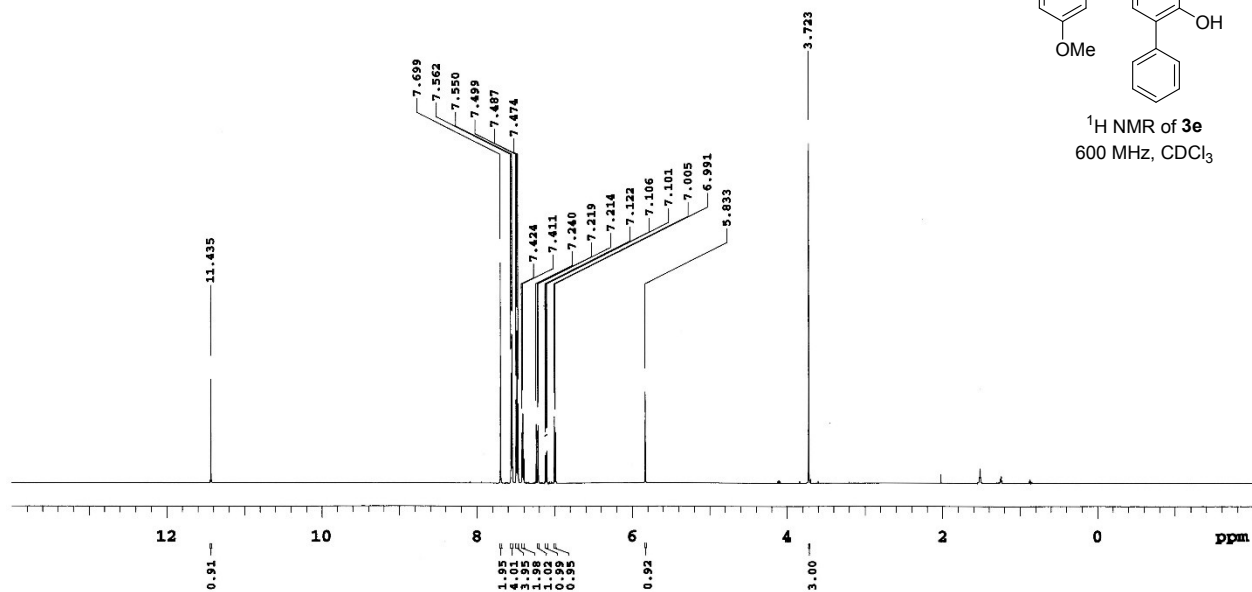
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600 MHz, CDCl₃



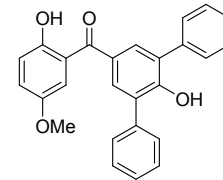
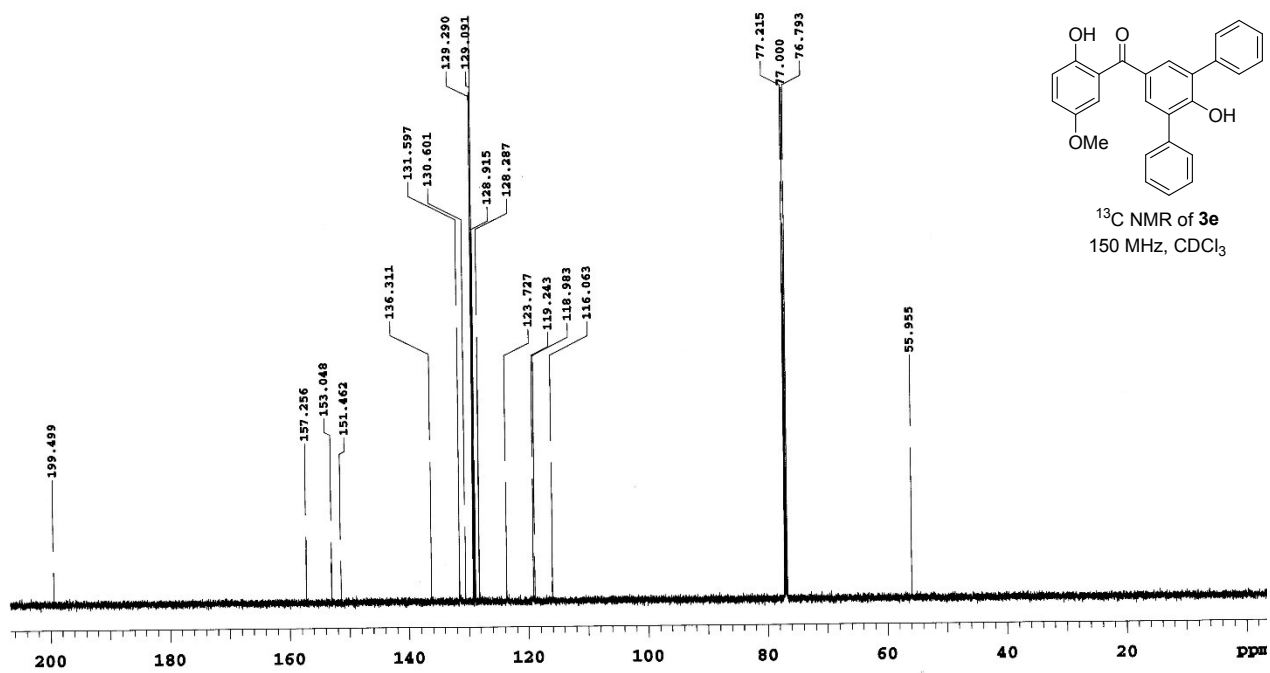
¹³C NMR of **3c**
150 MHz, CDCl₃



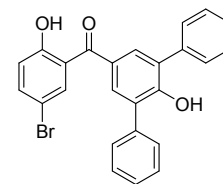
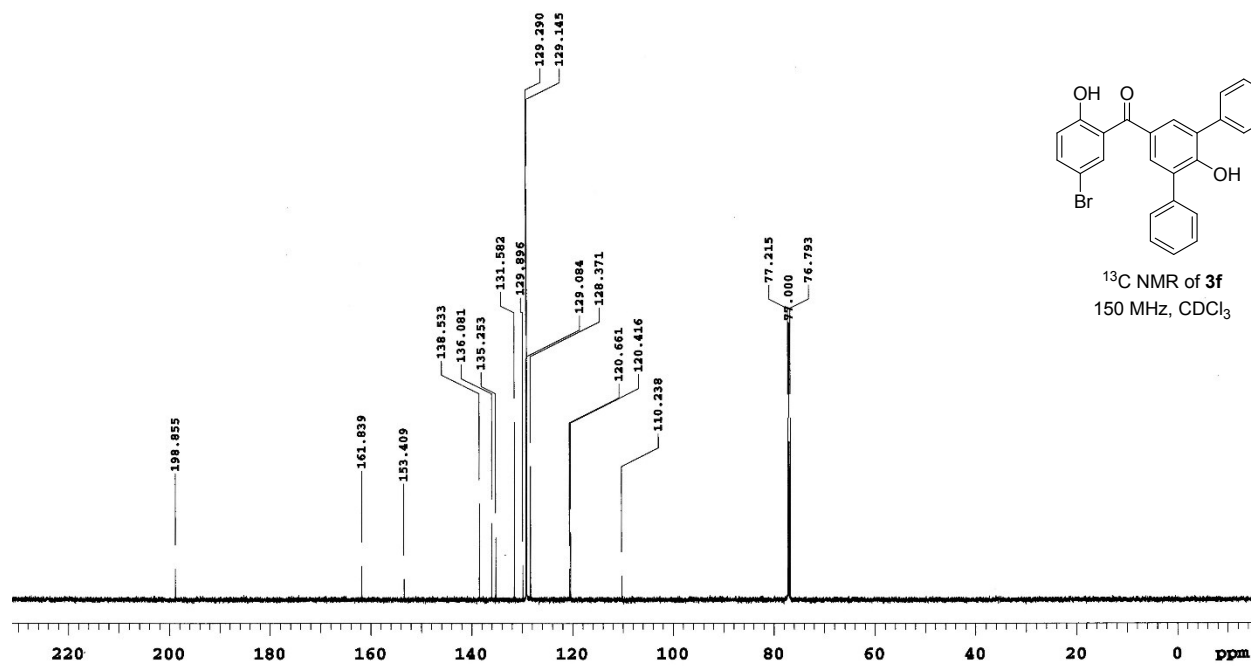
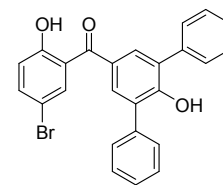
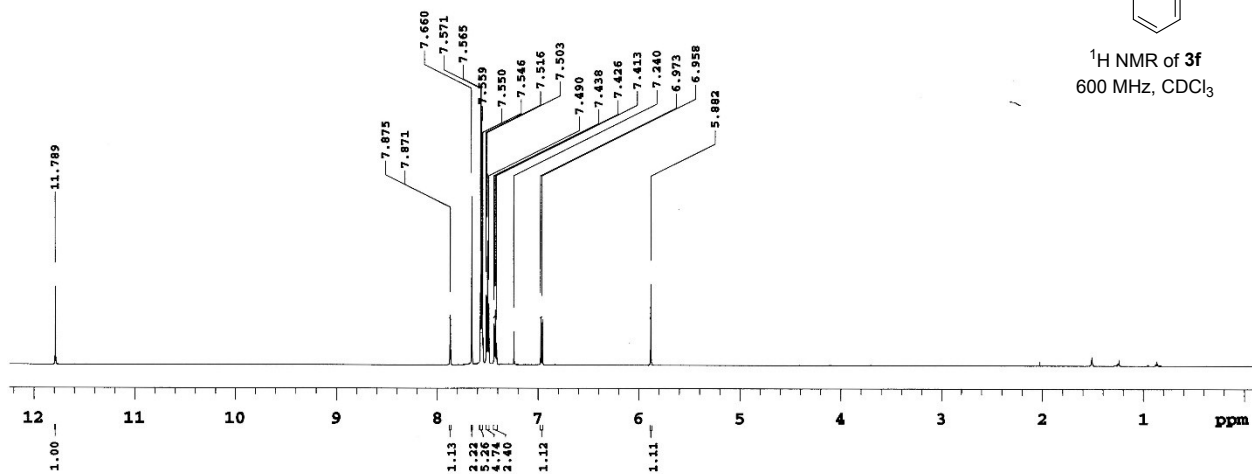


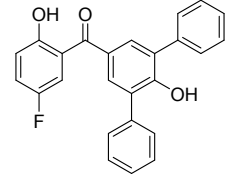


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600 MHz, CDCl₃

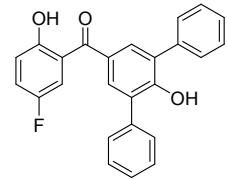
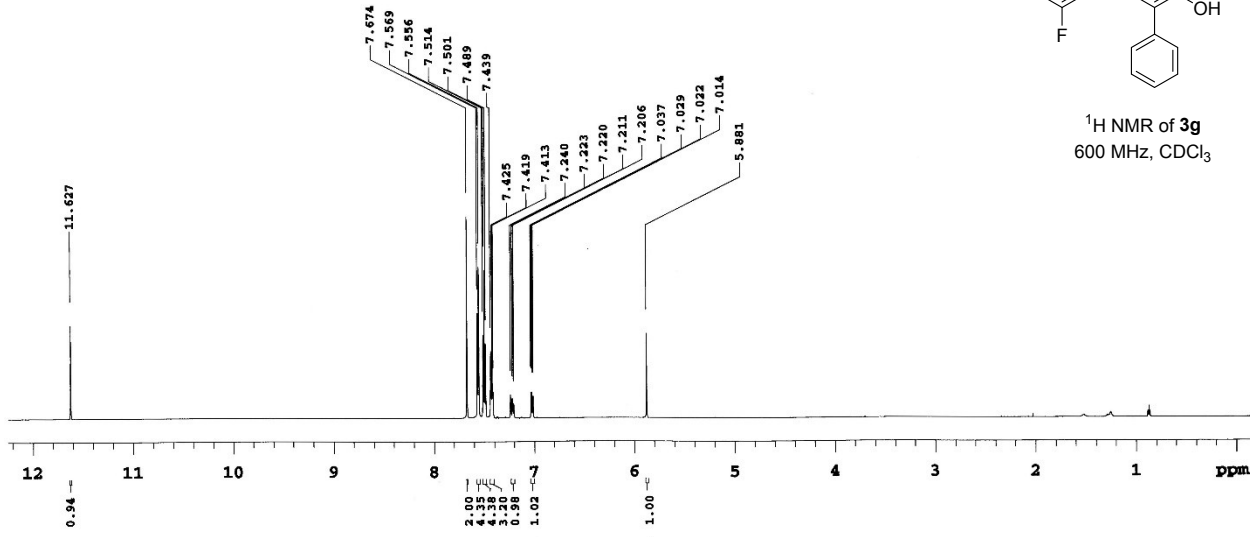


¹³C NMR of 3e
150 MHz, CDCl₃

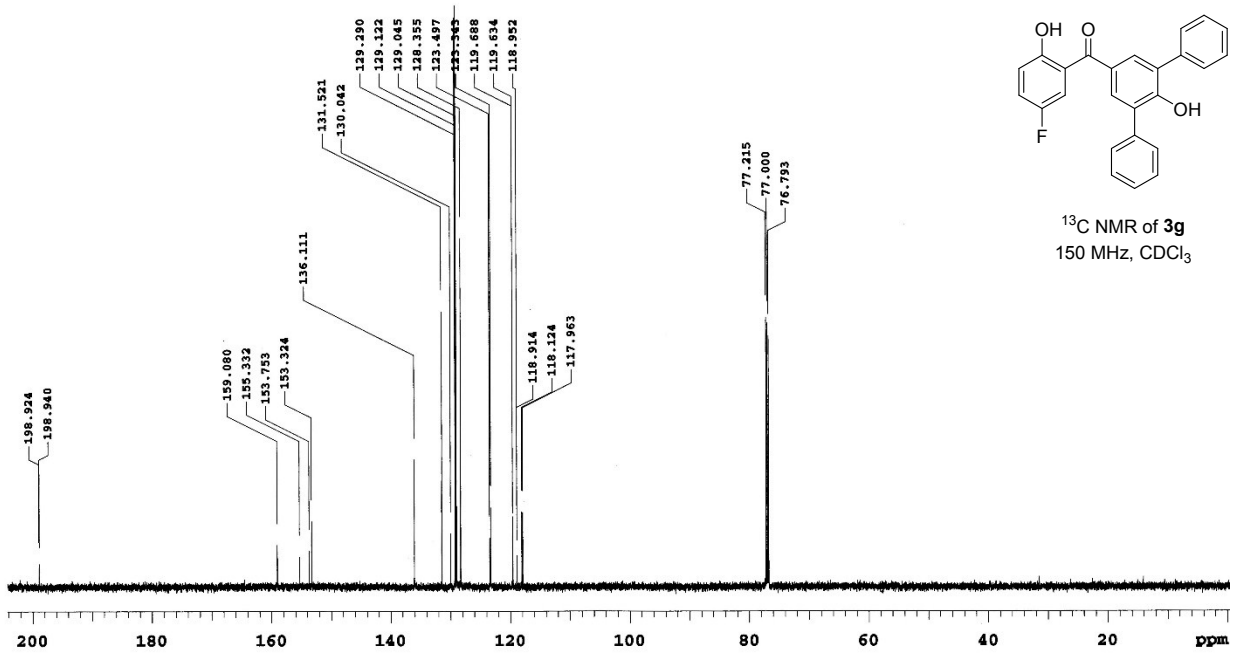


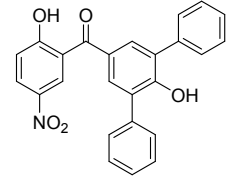


¹H NMR of **3g**
600 MHz, CDCl₃

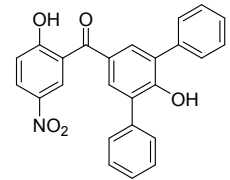
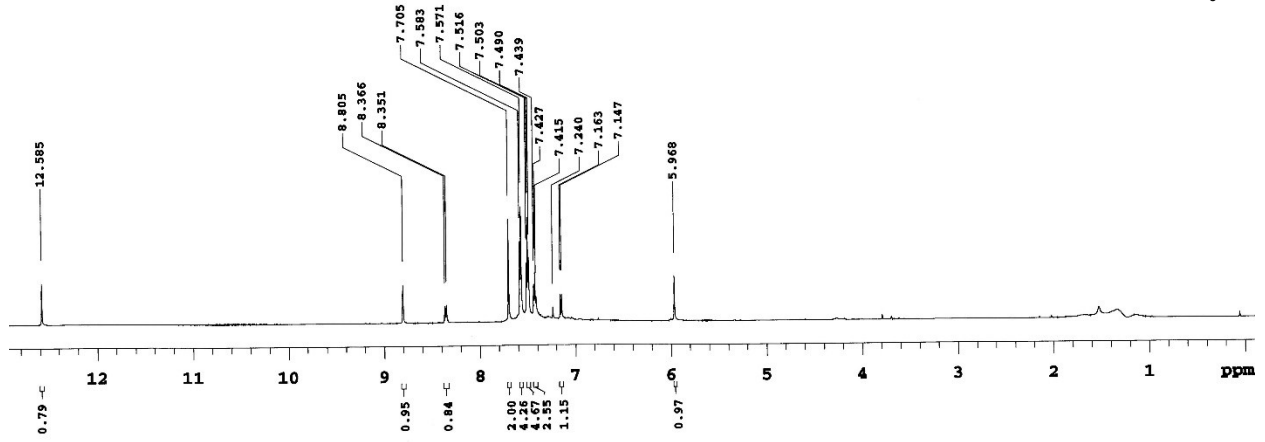


¹³C NMR of **3g**
150 MHz, CDCl₃

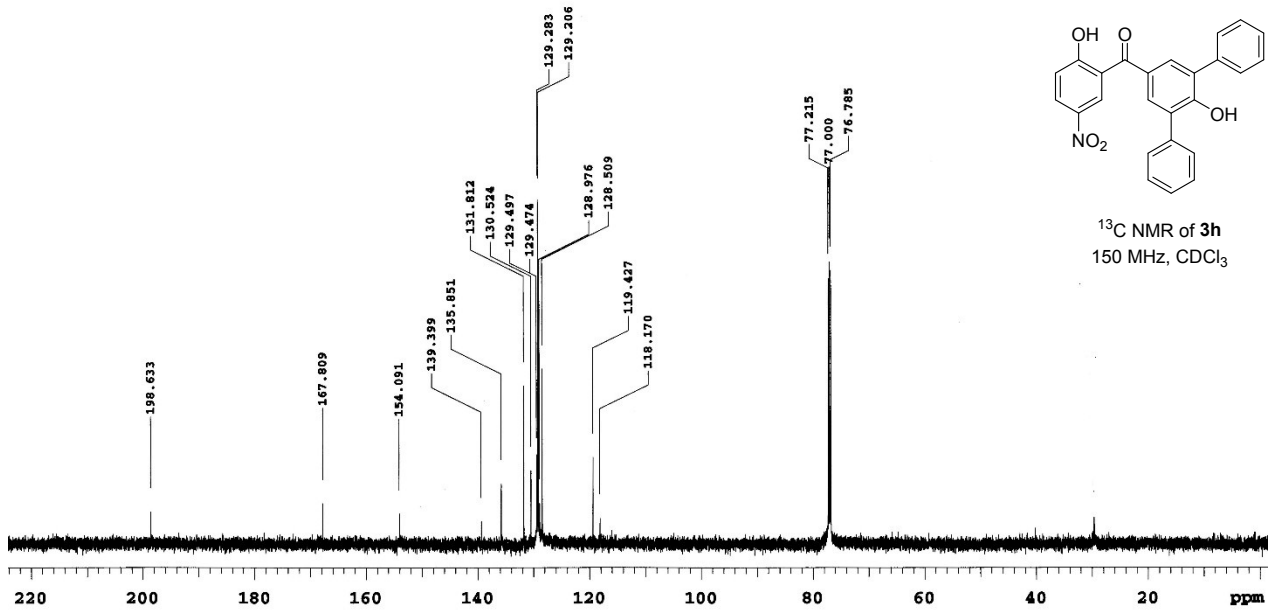


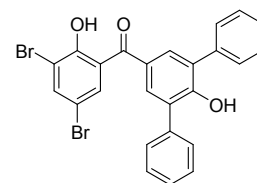
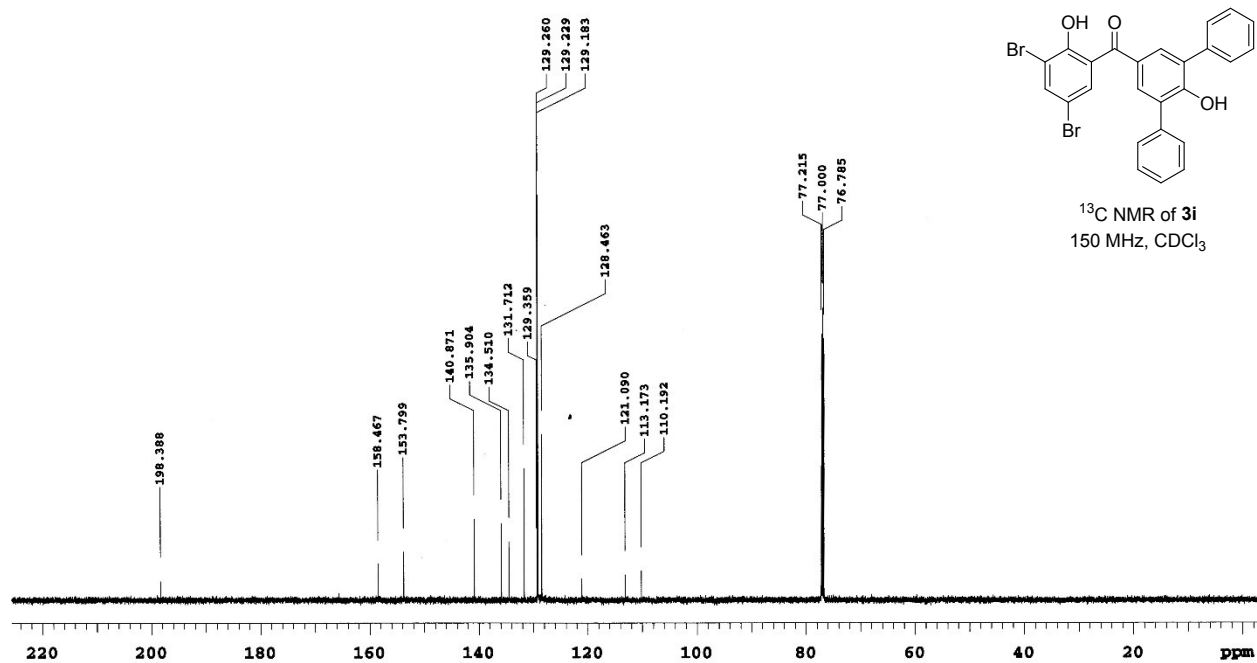
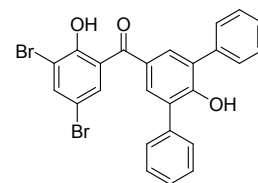
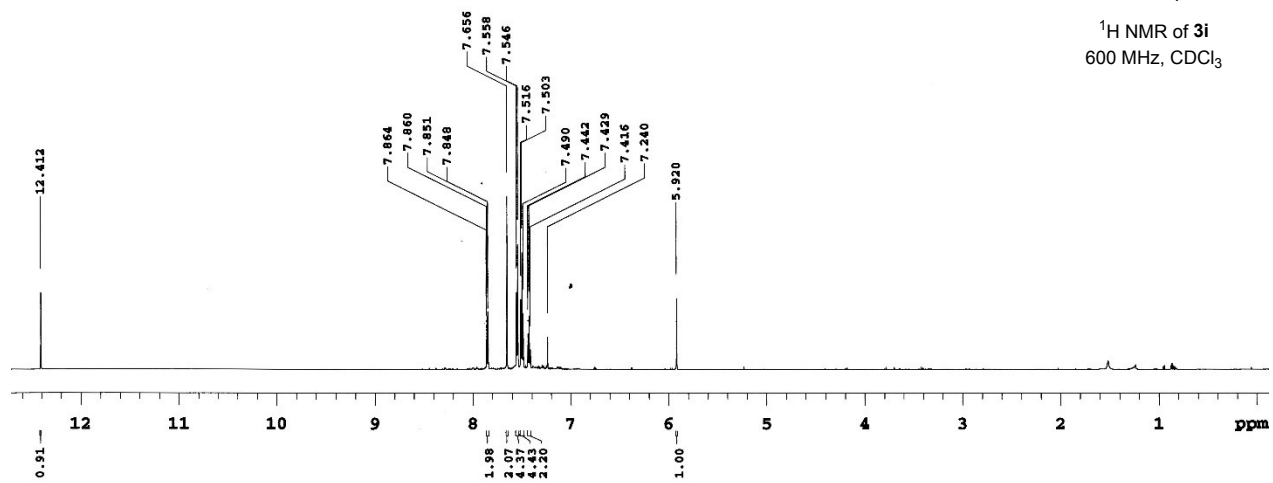


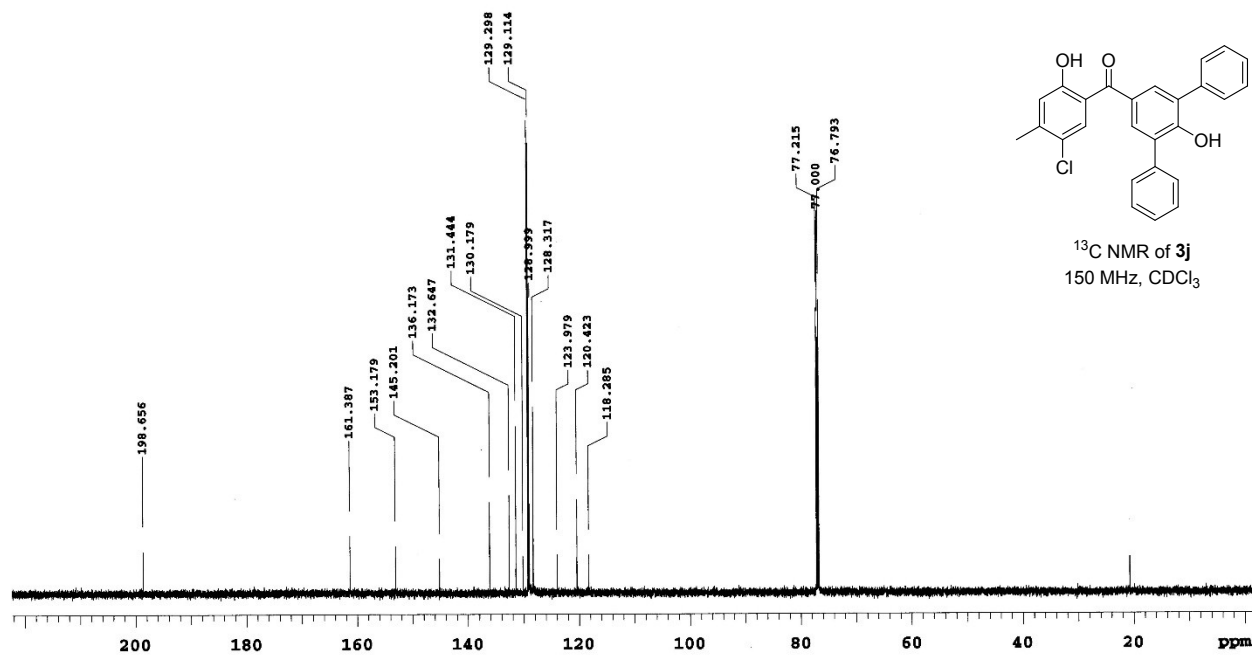
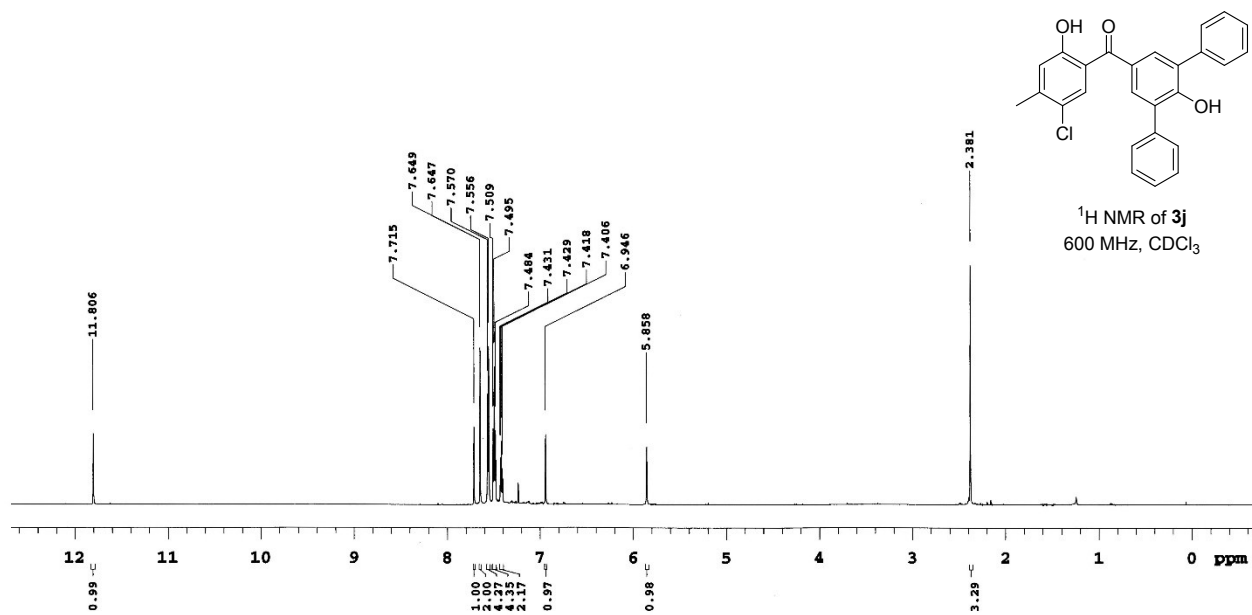
¹H NMR of **3h**
600 MHz, CDCl₃

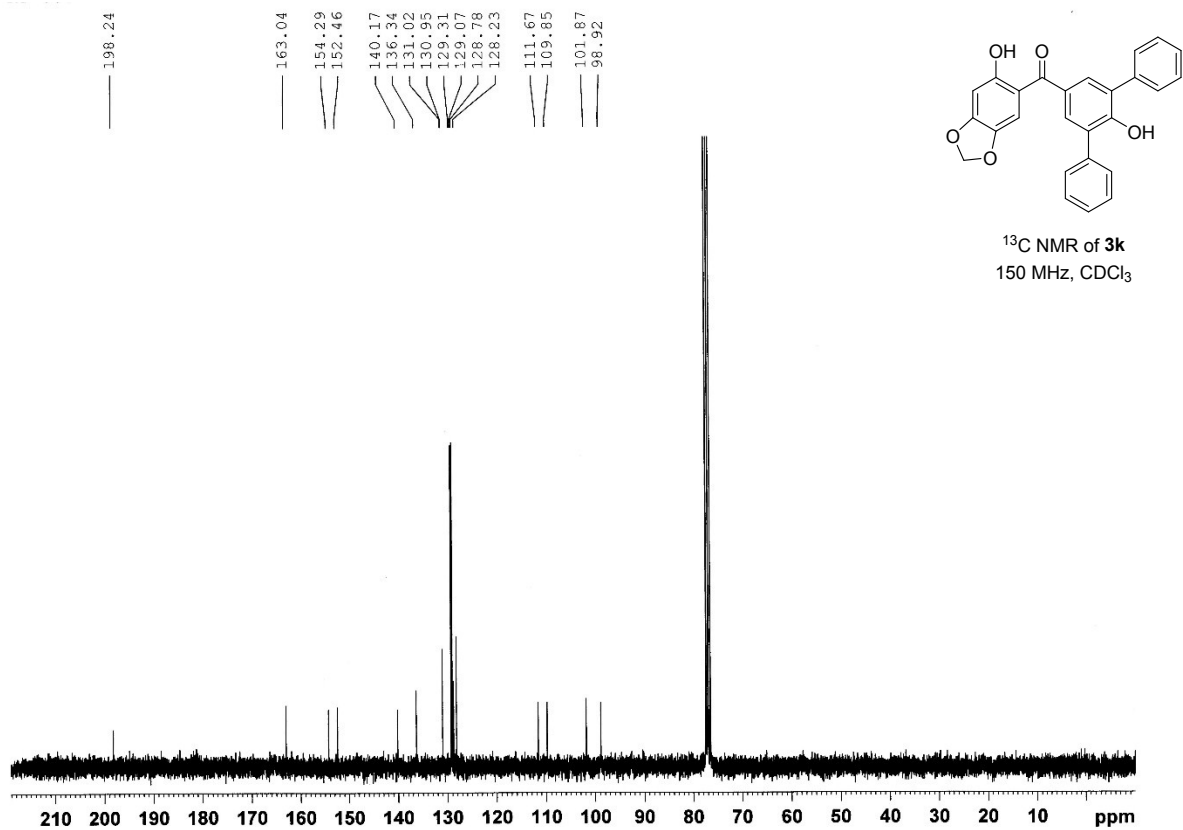
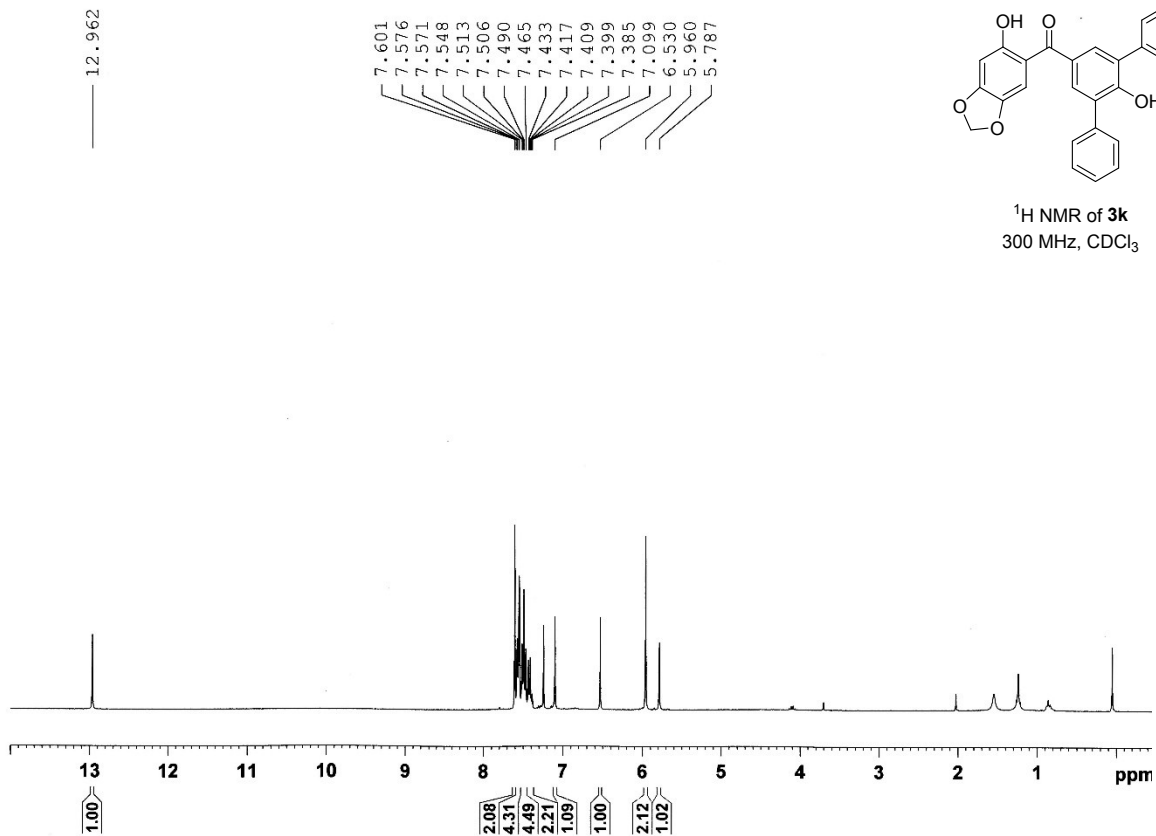


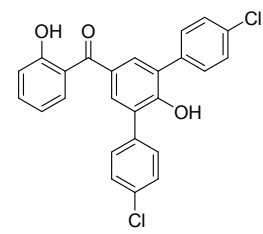
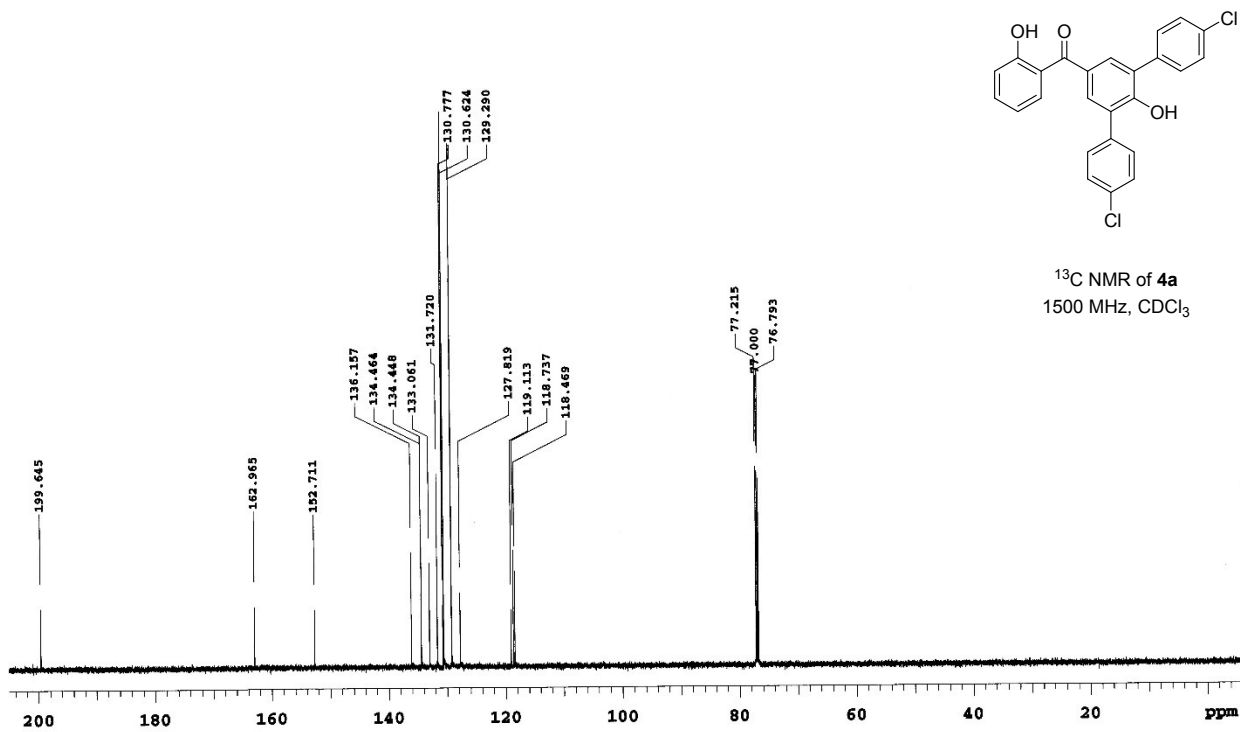
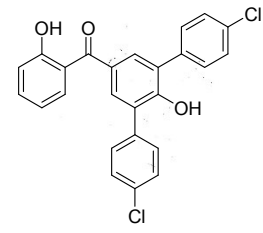
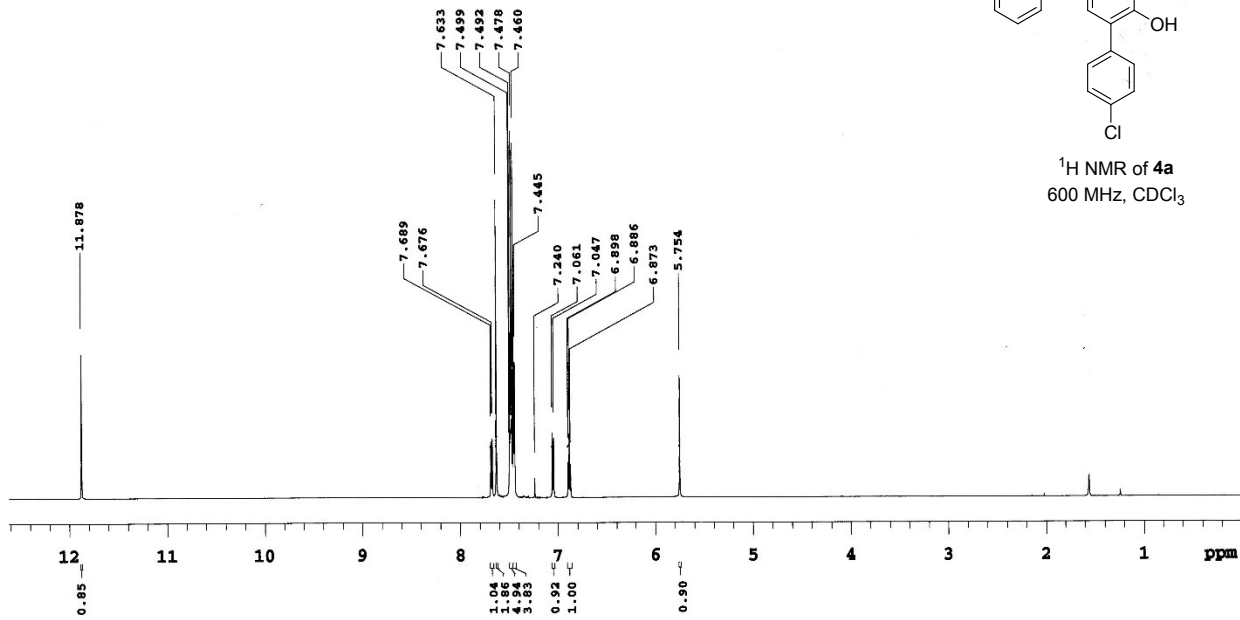
¹³C NMR of **3h**
150 MHz, CDCl₃

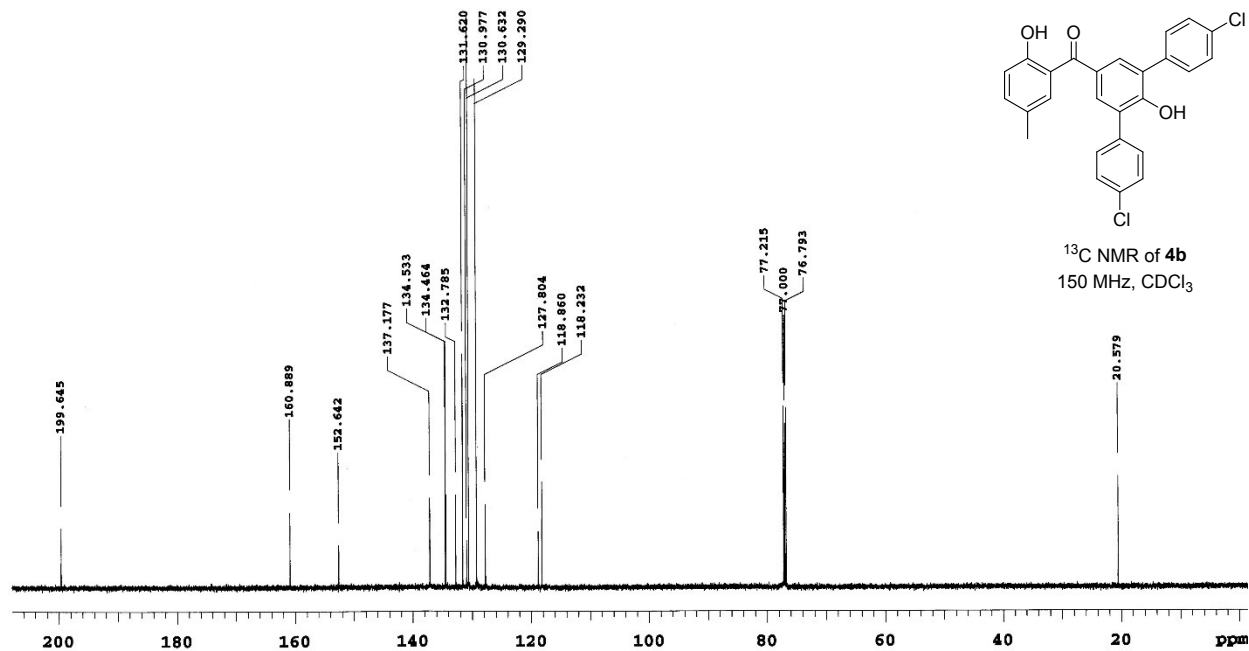
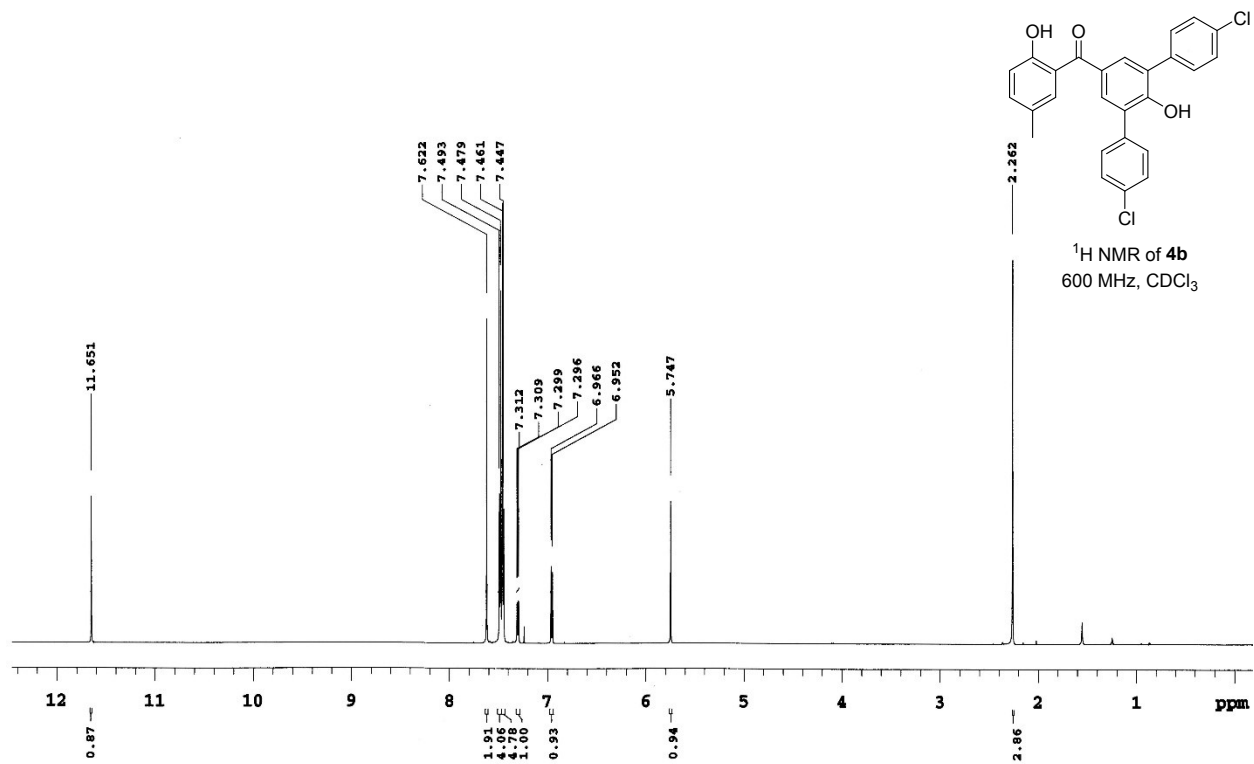


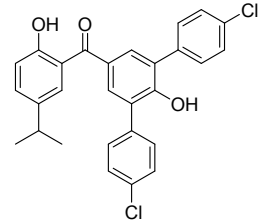




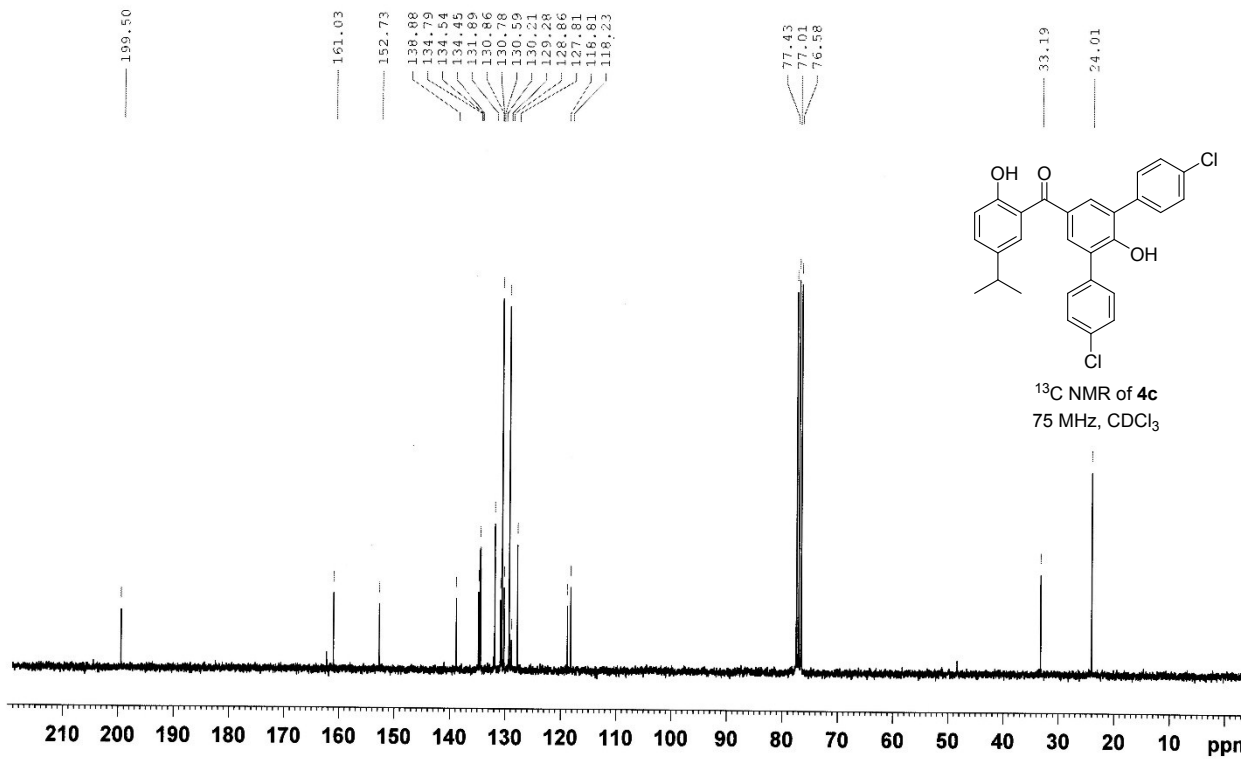
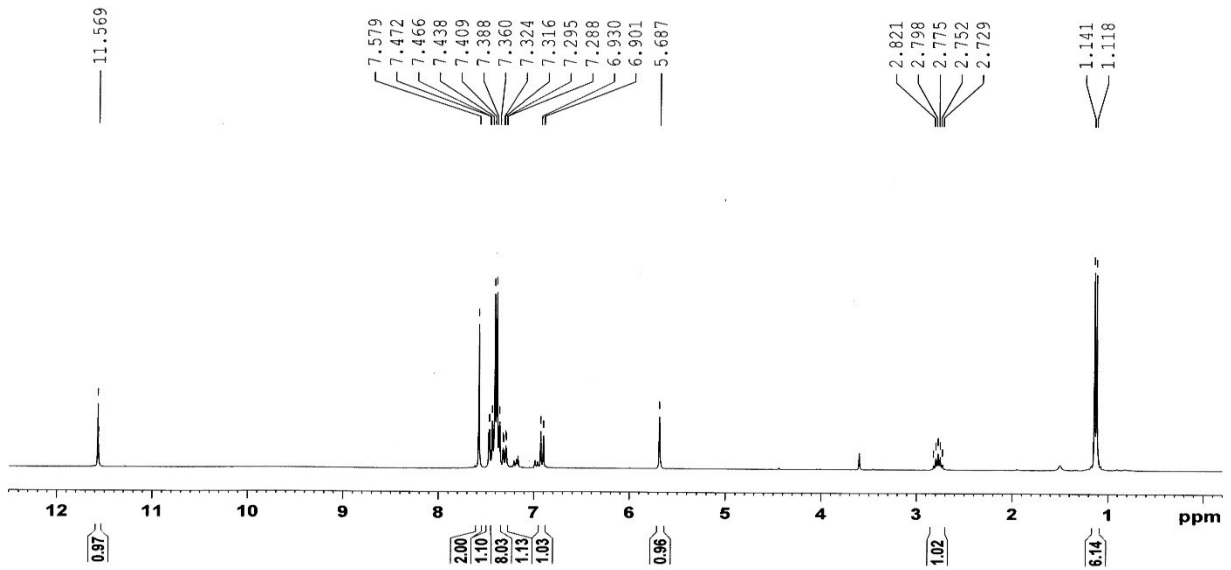


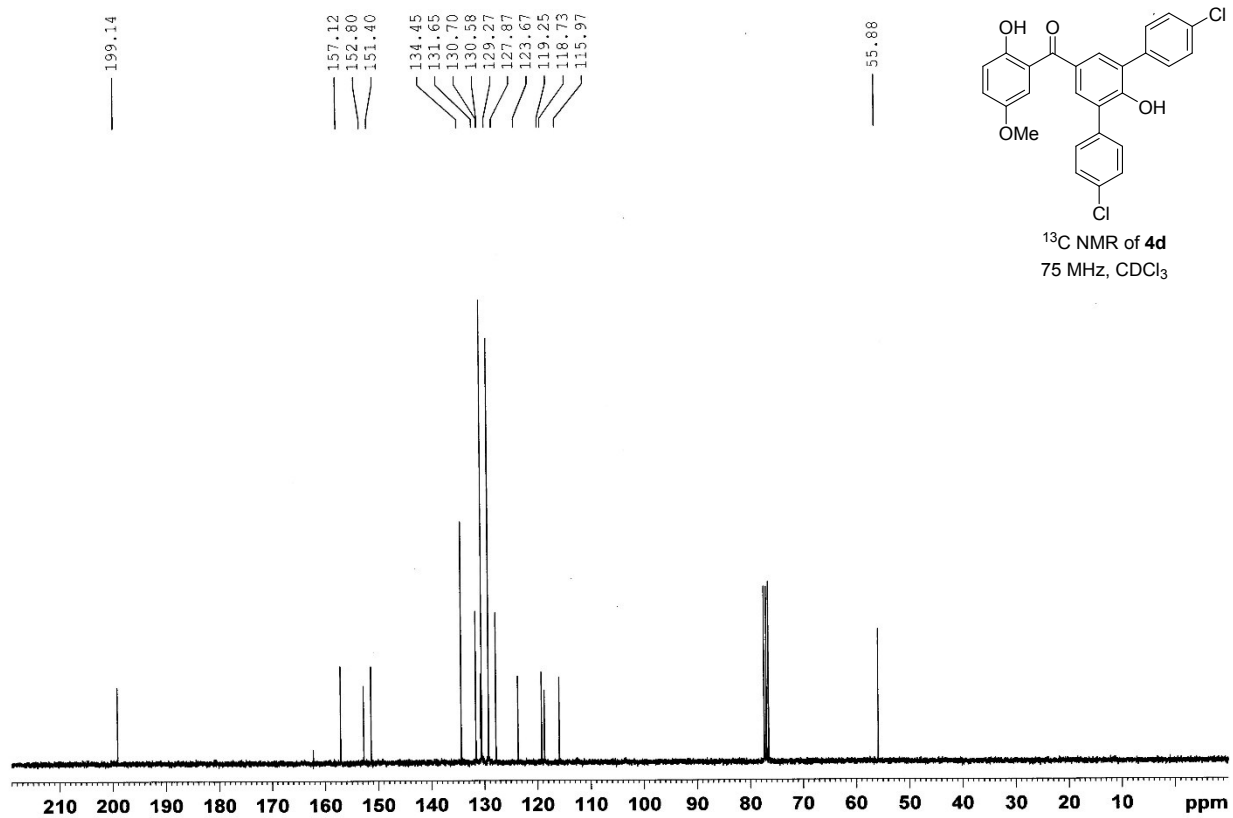
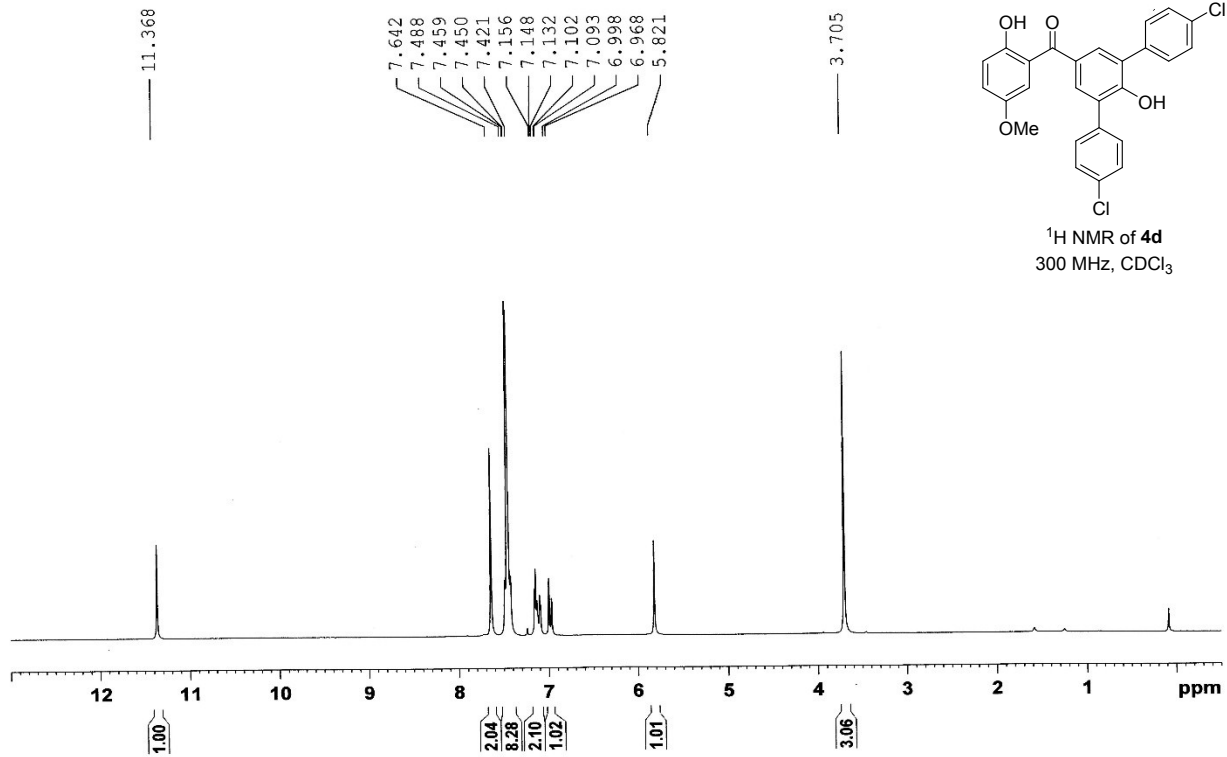


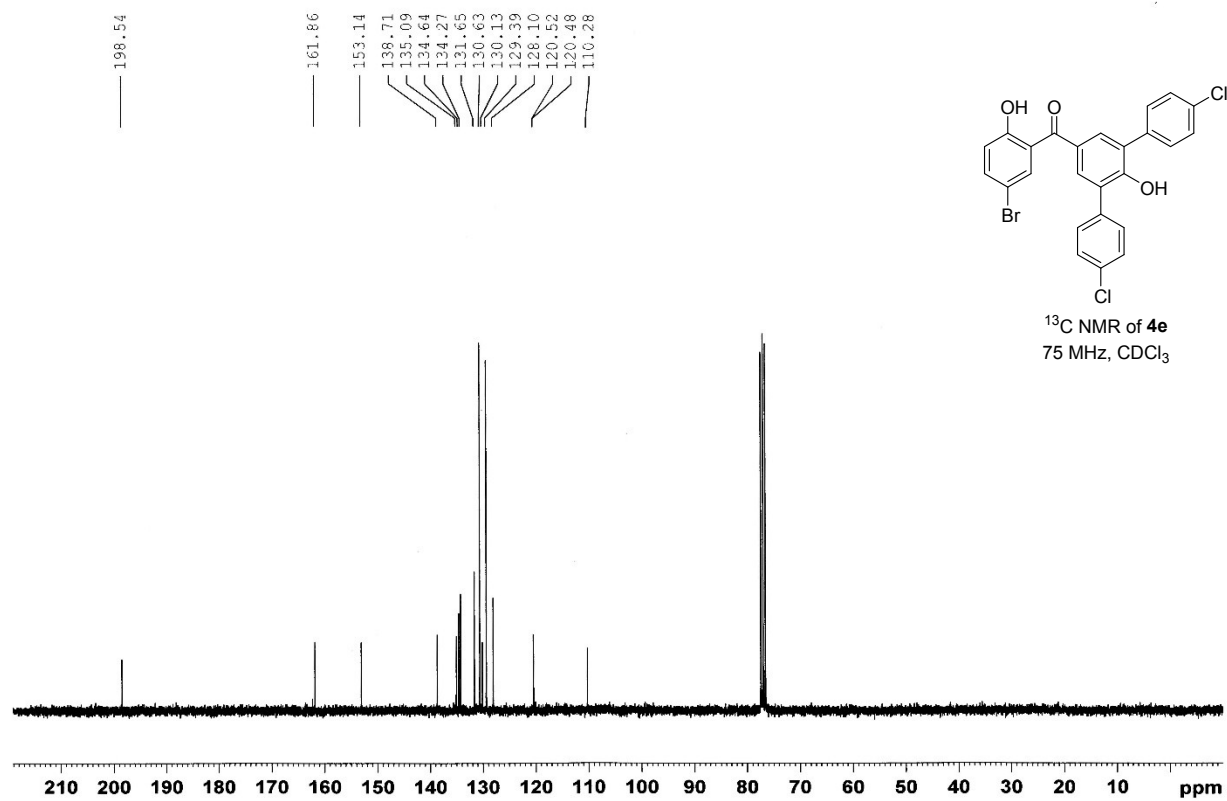
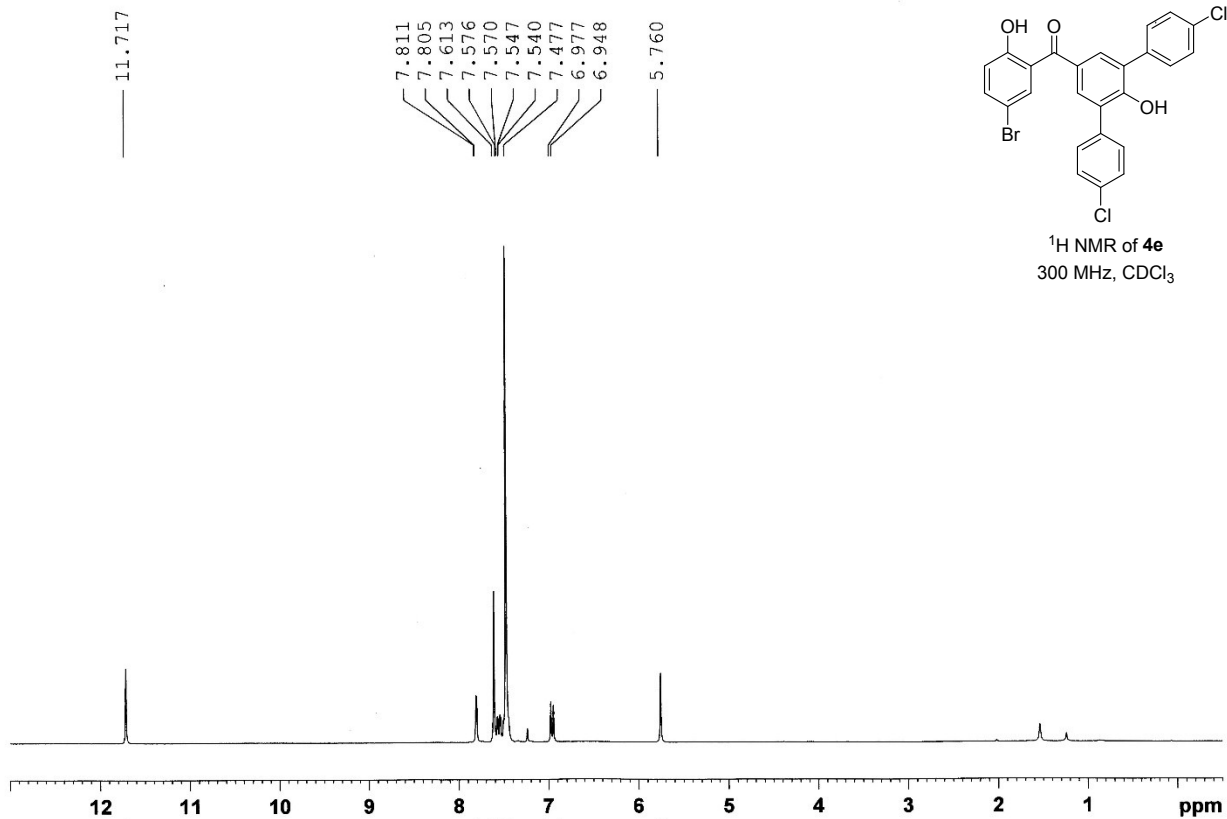


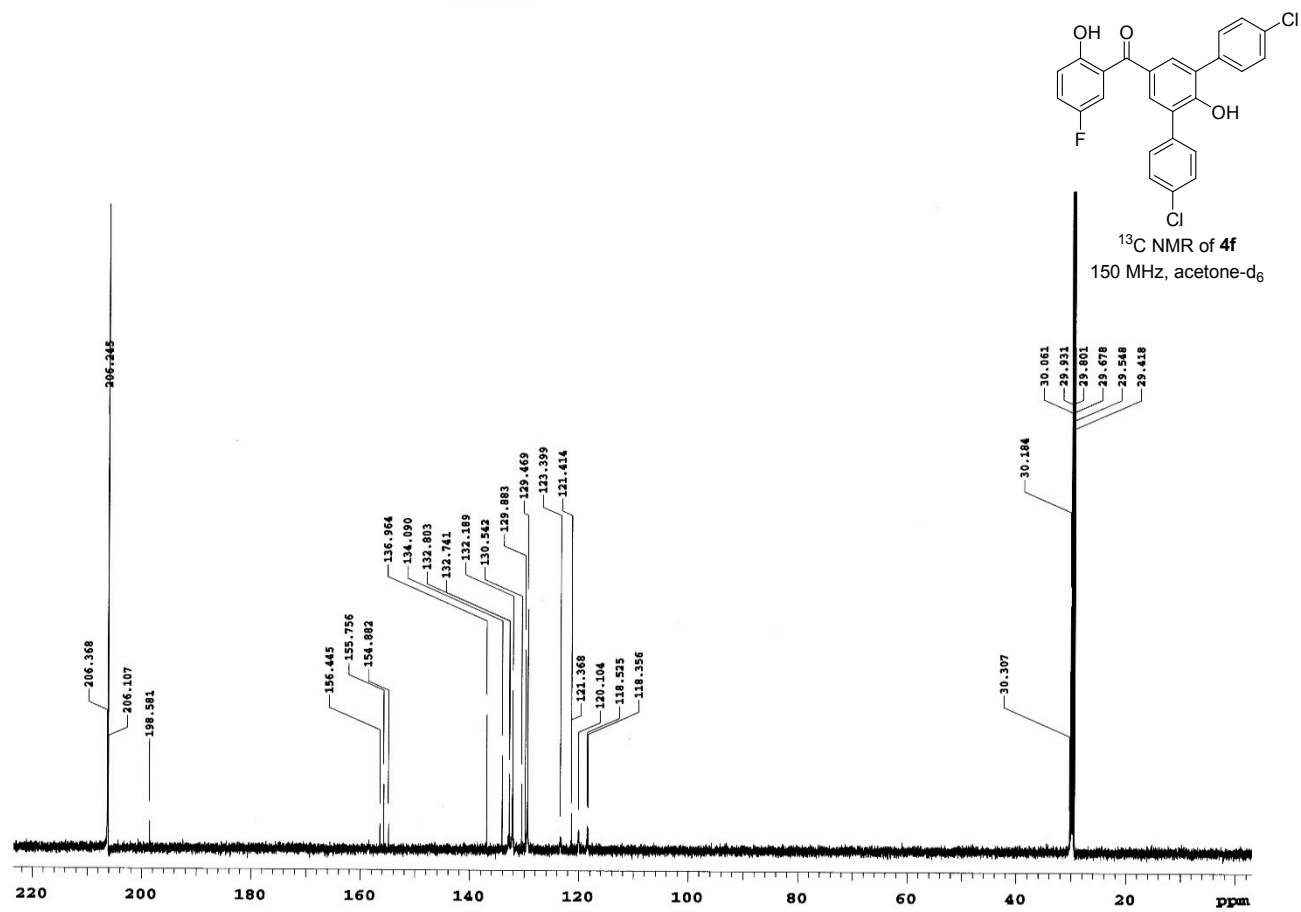
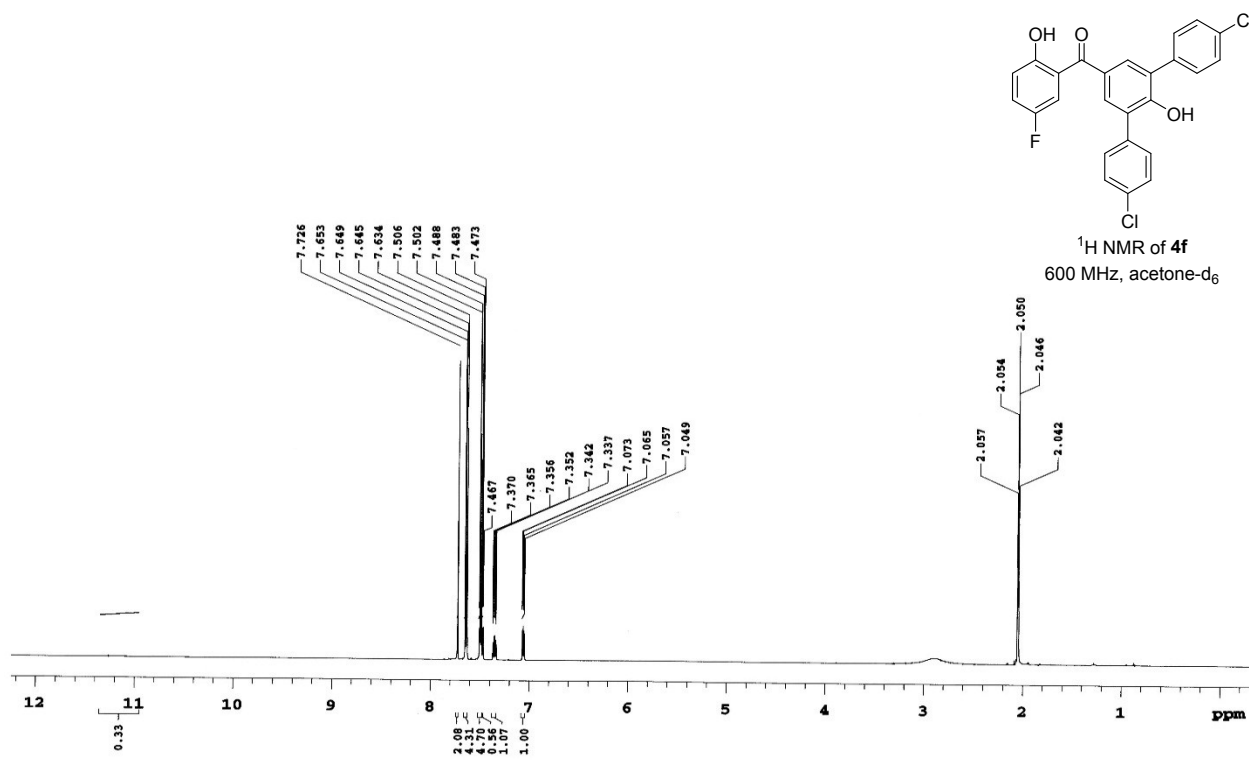


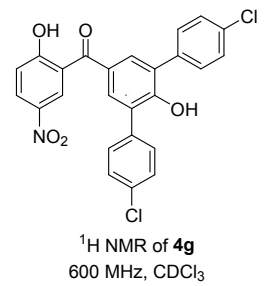
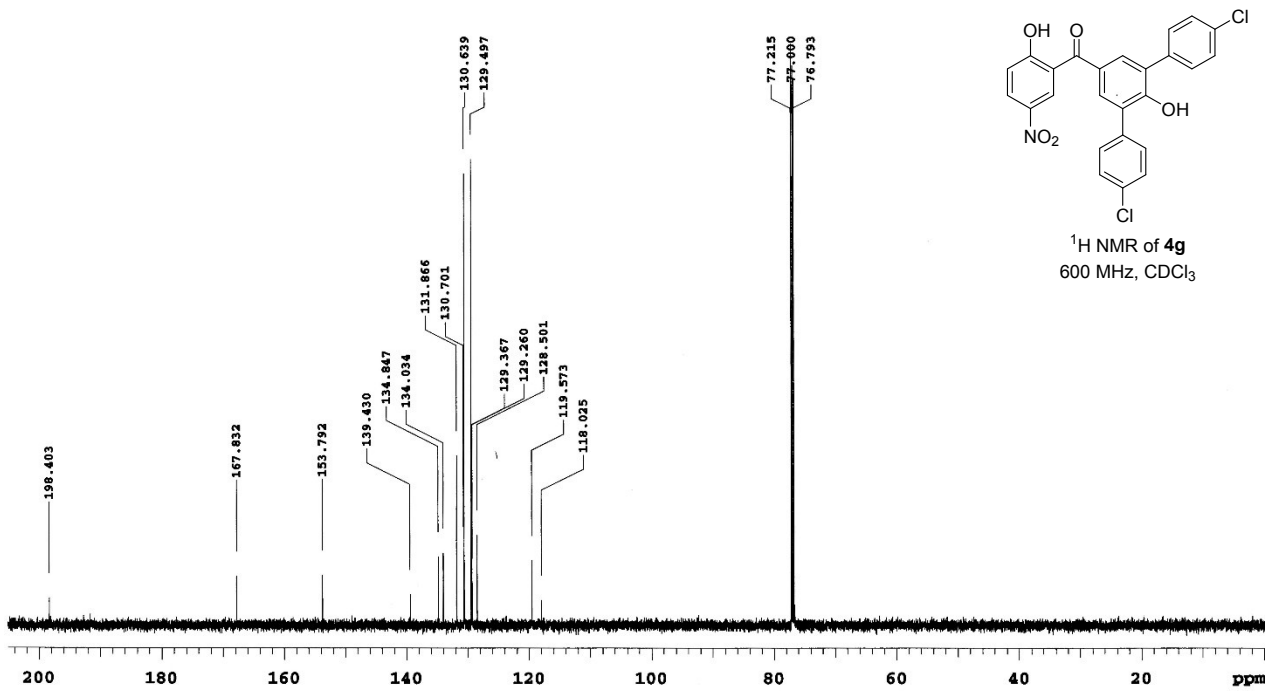
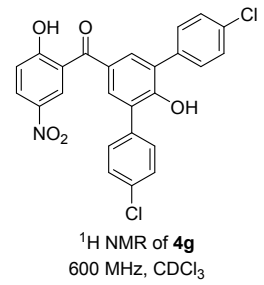
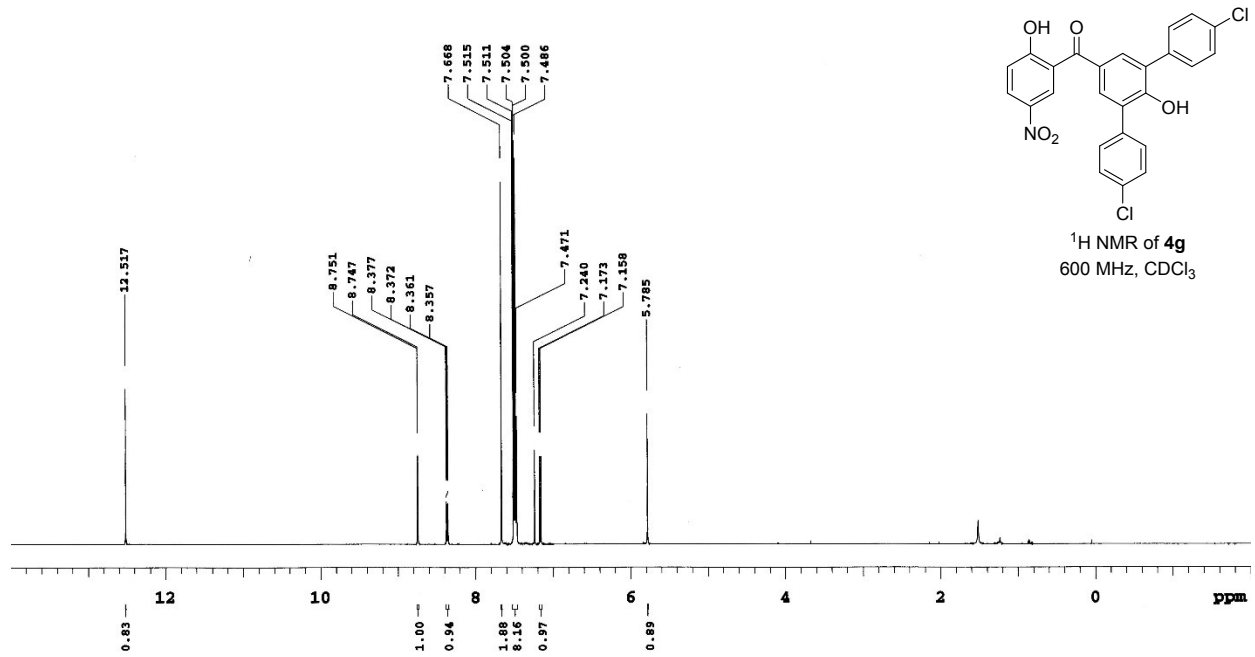
¹H NMR of 4c
300 MHz, CDCl₃

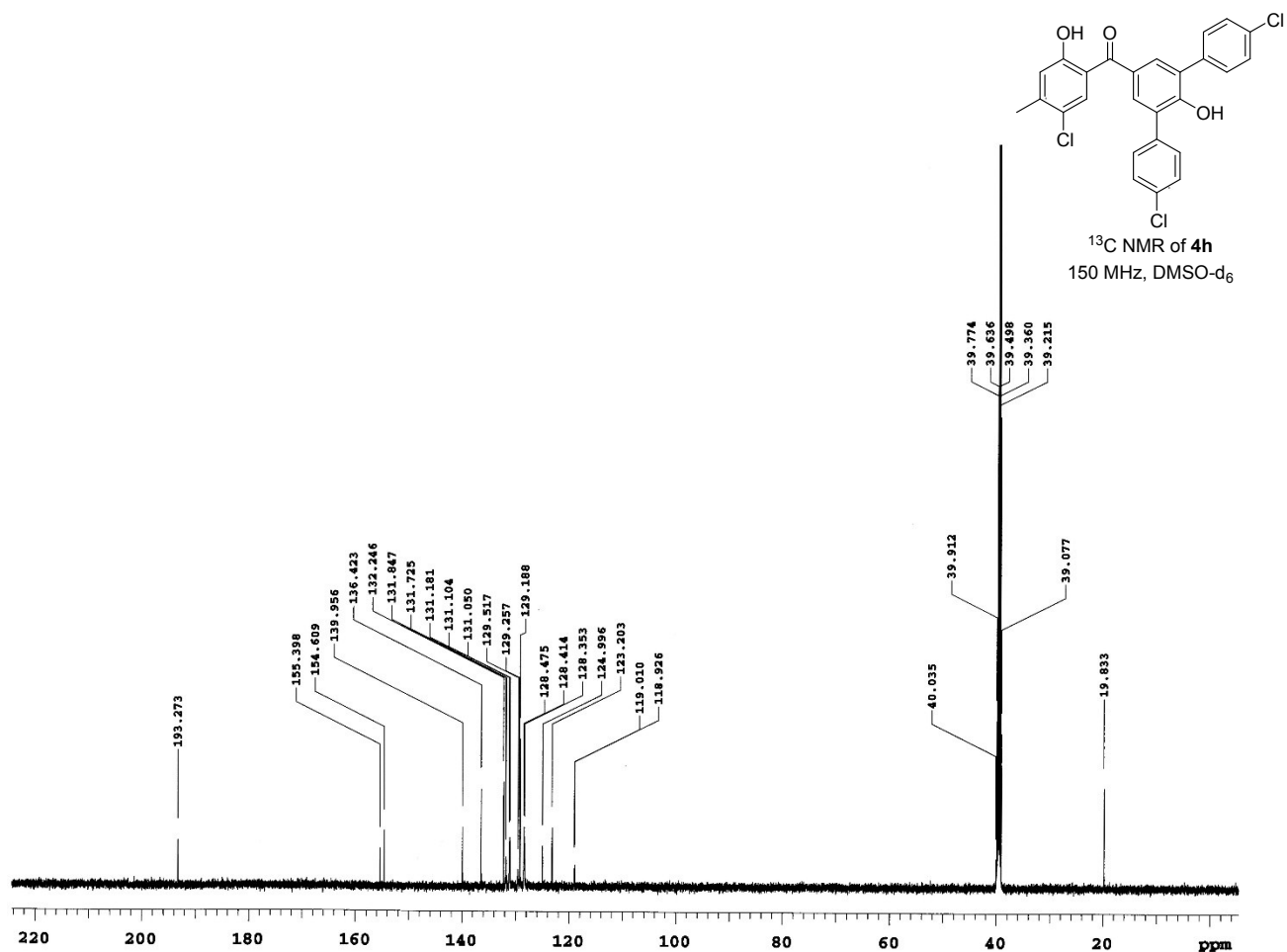
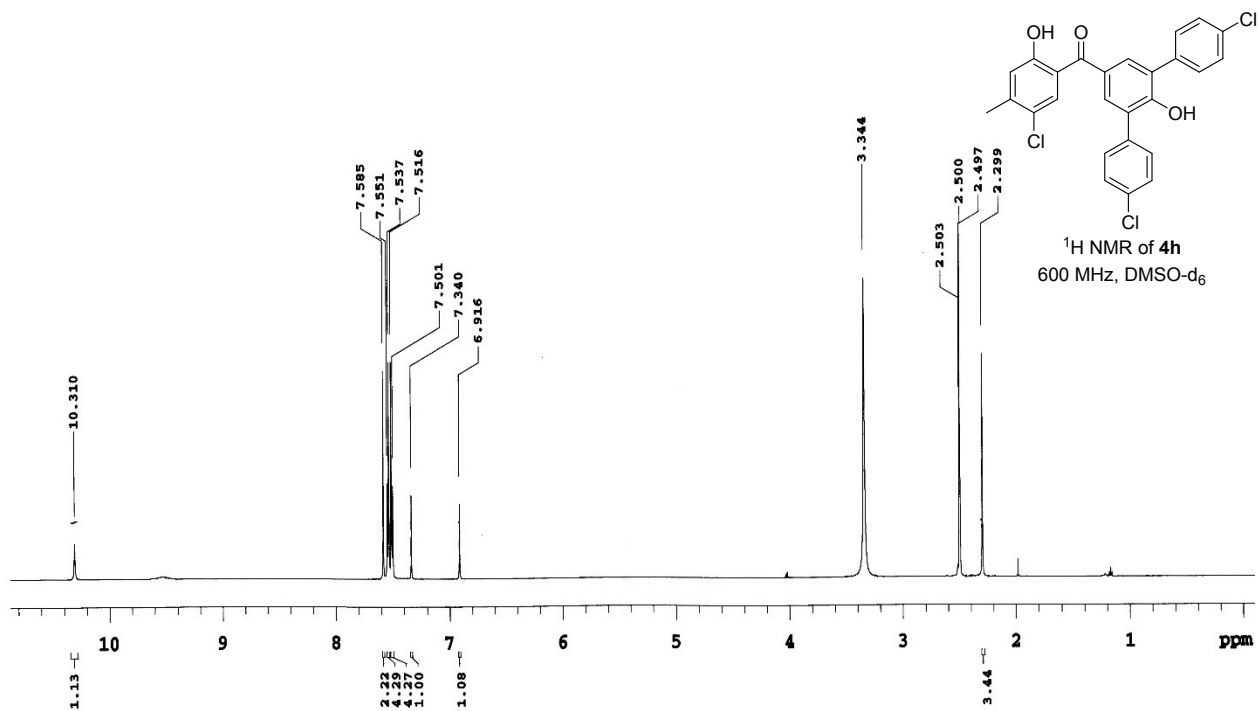


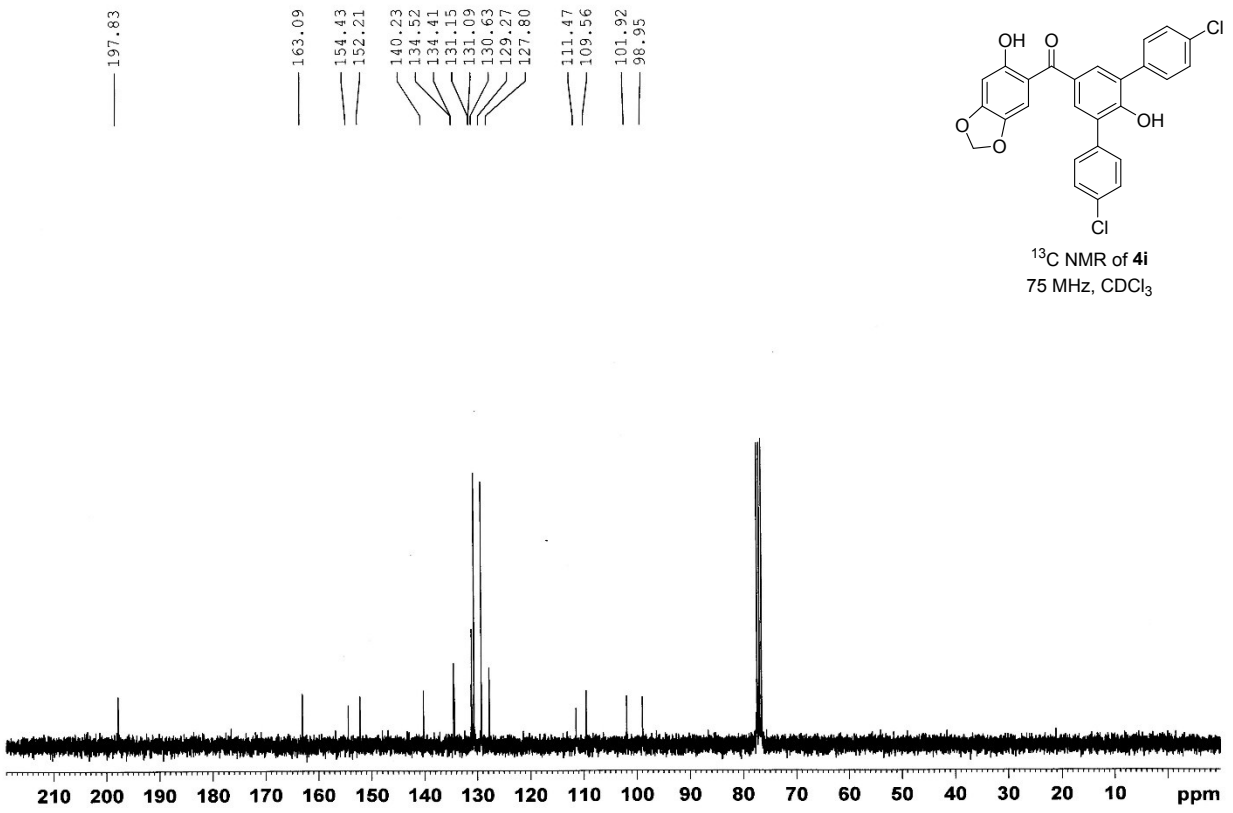
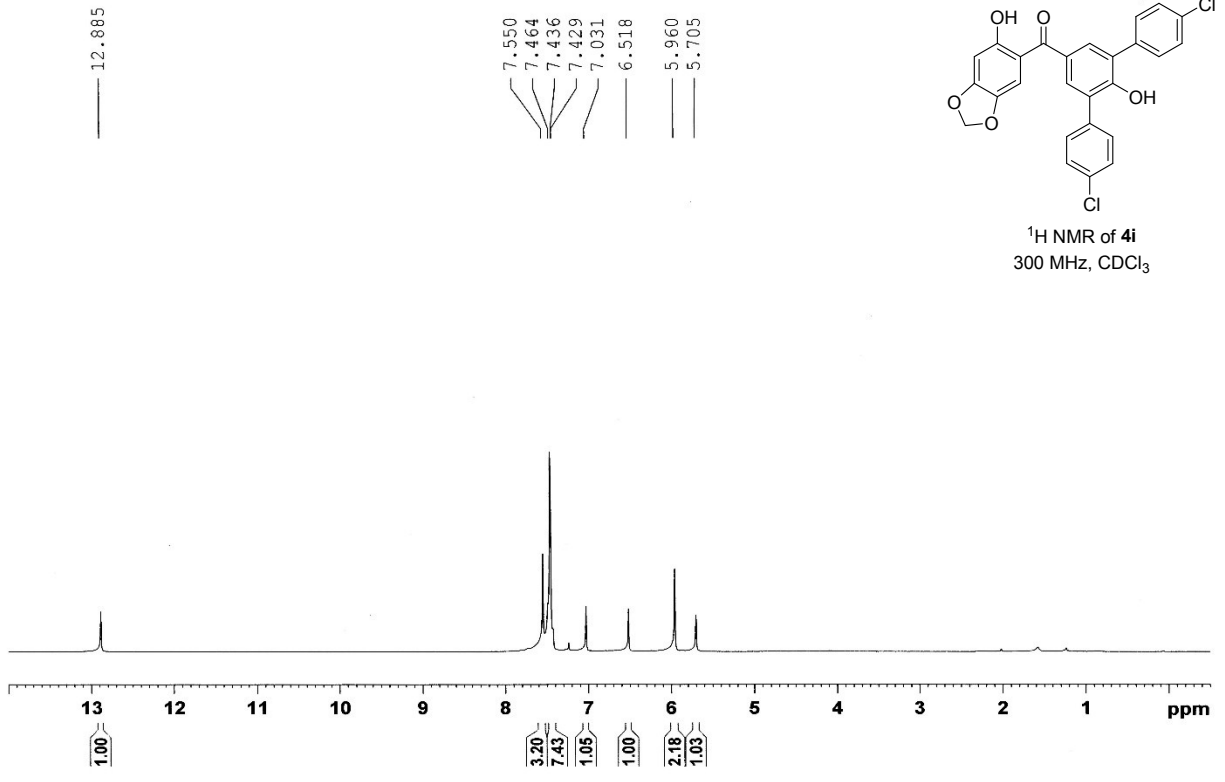


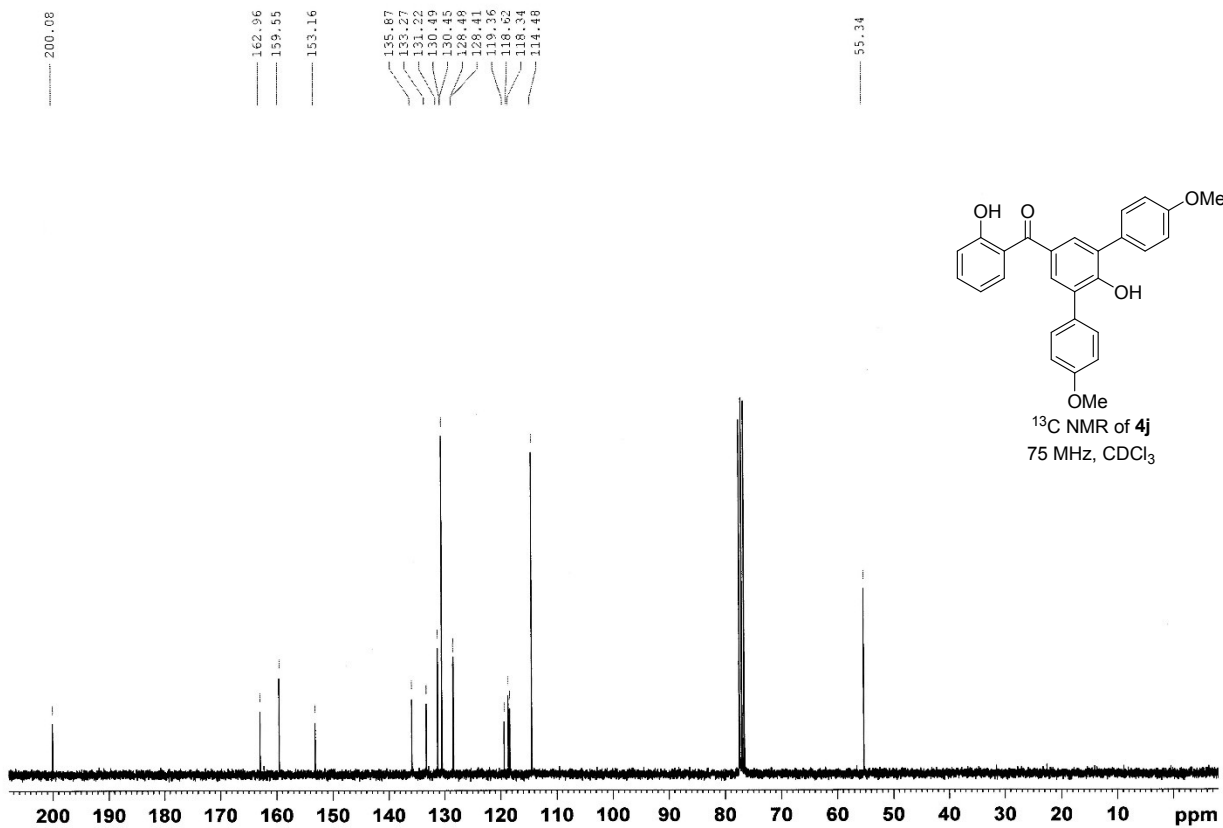
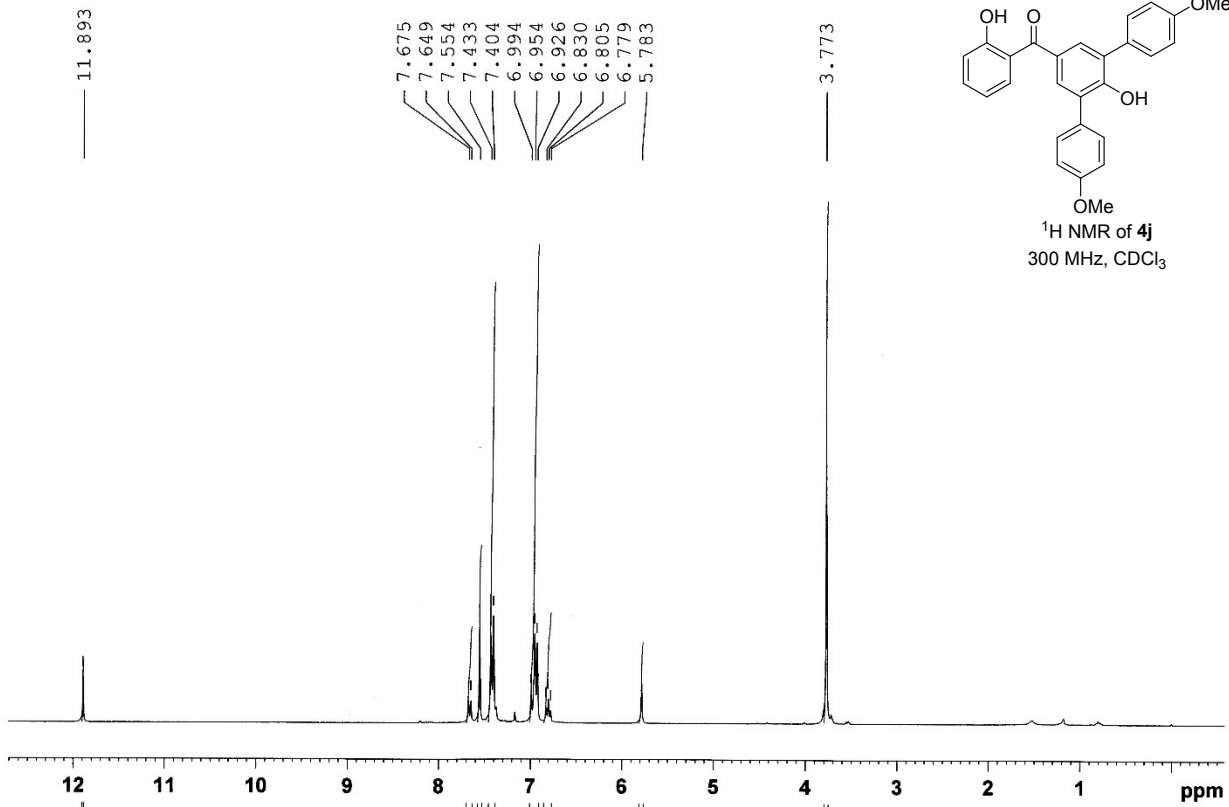


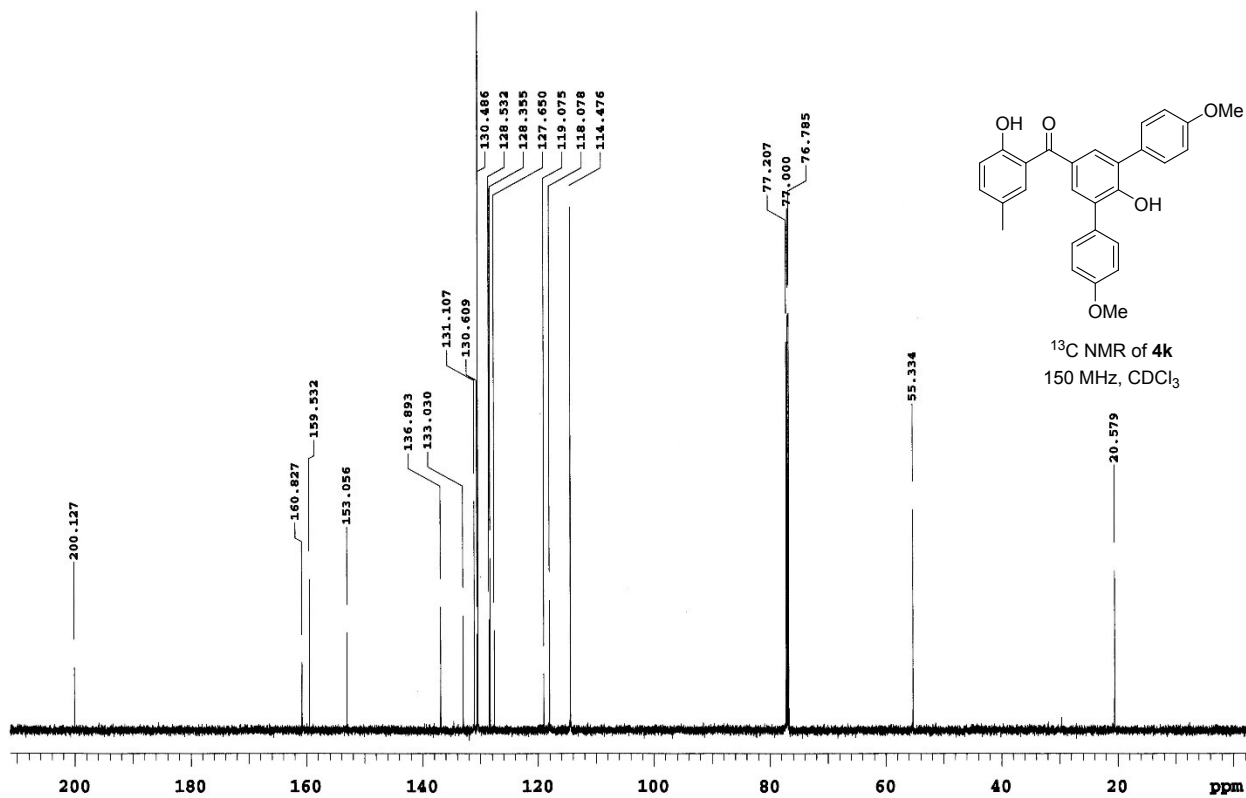
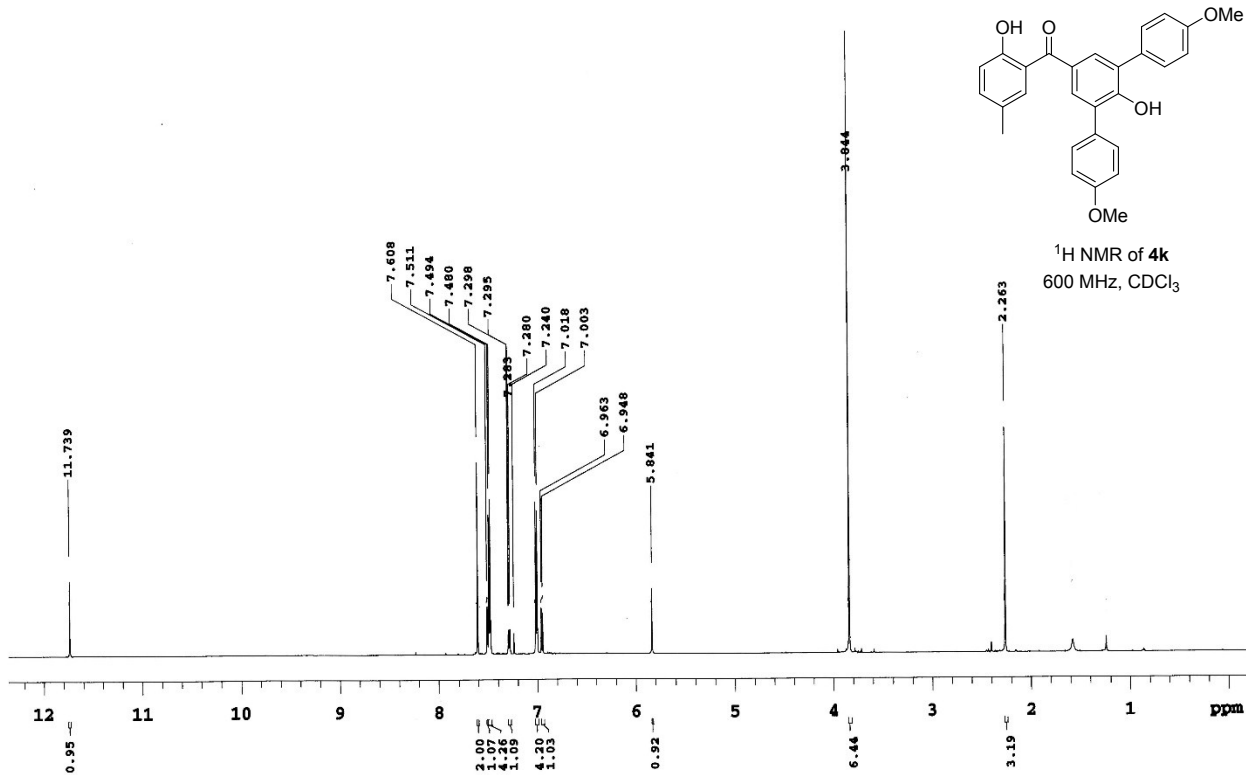


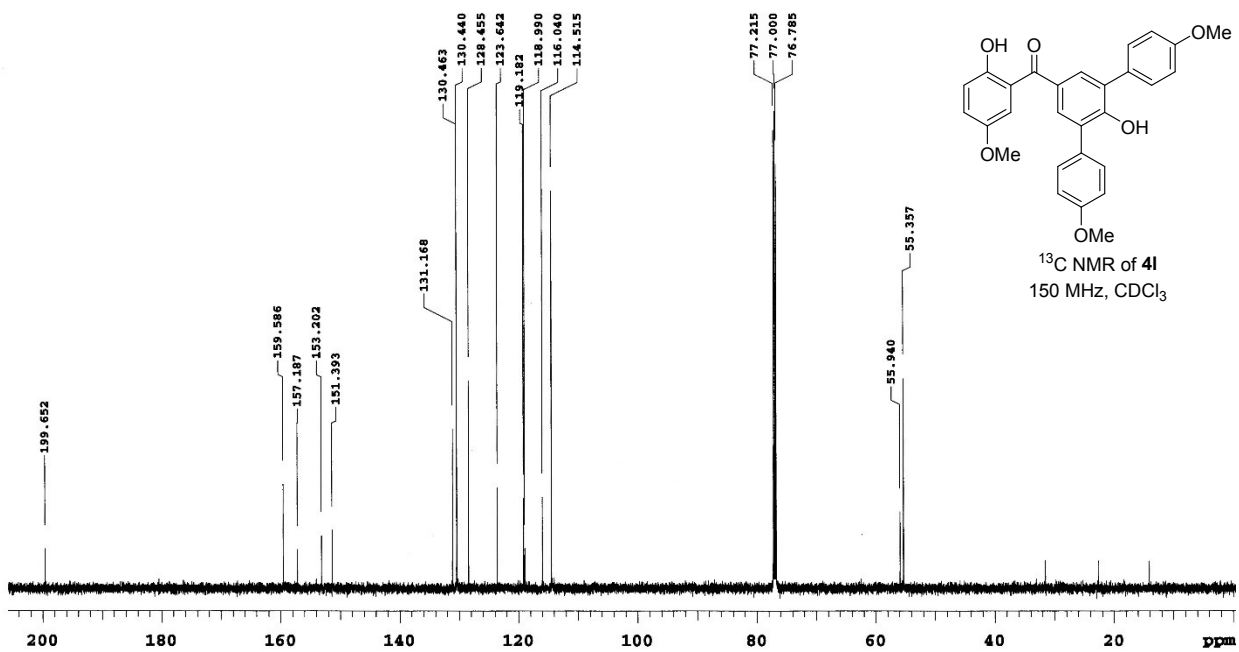
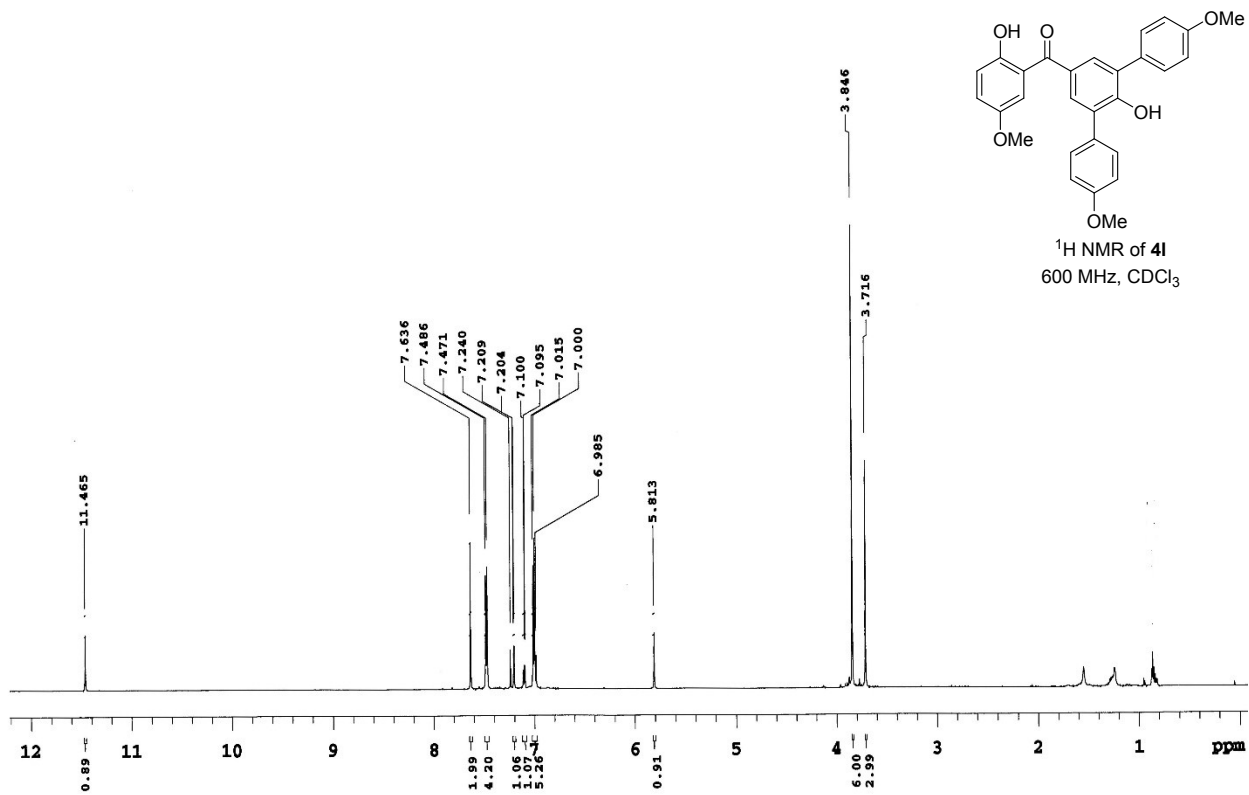


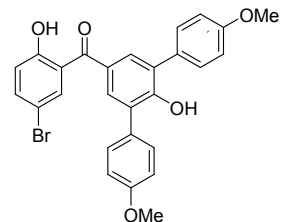
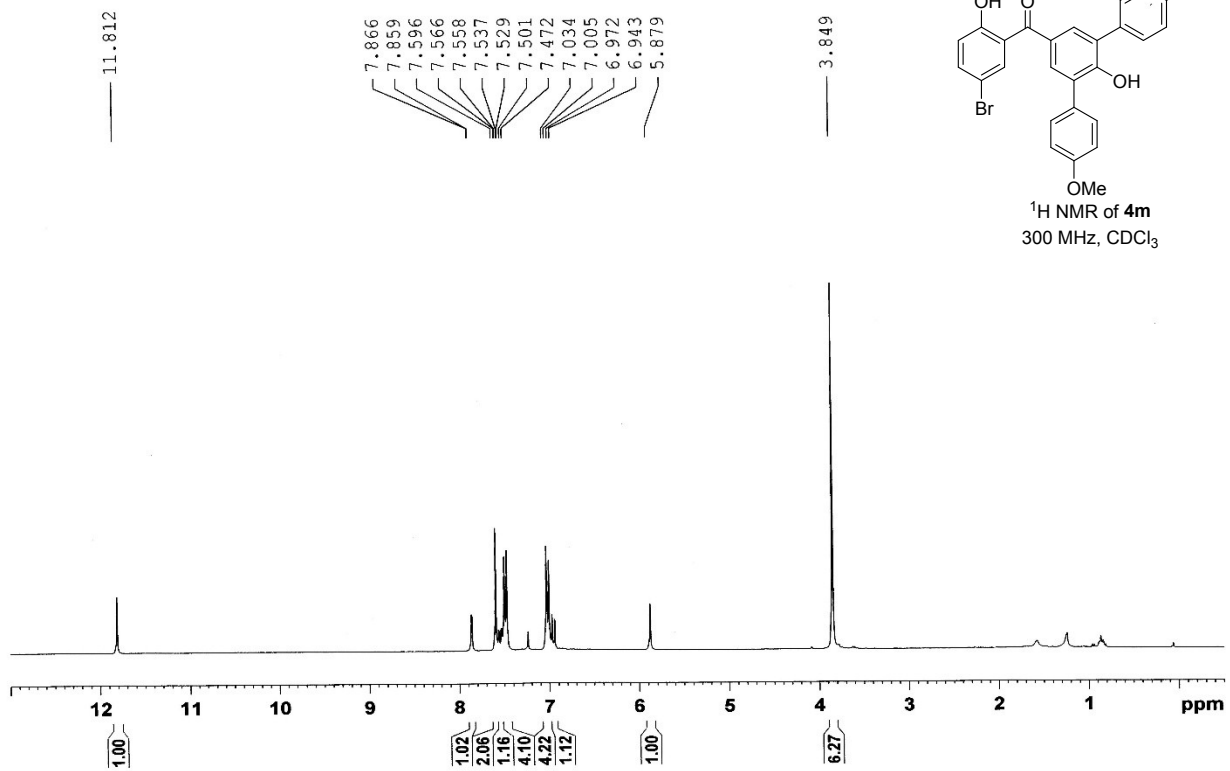




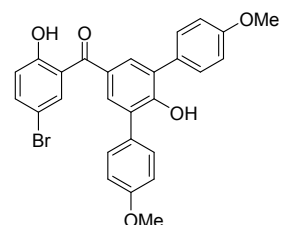
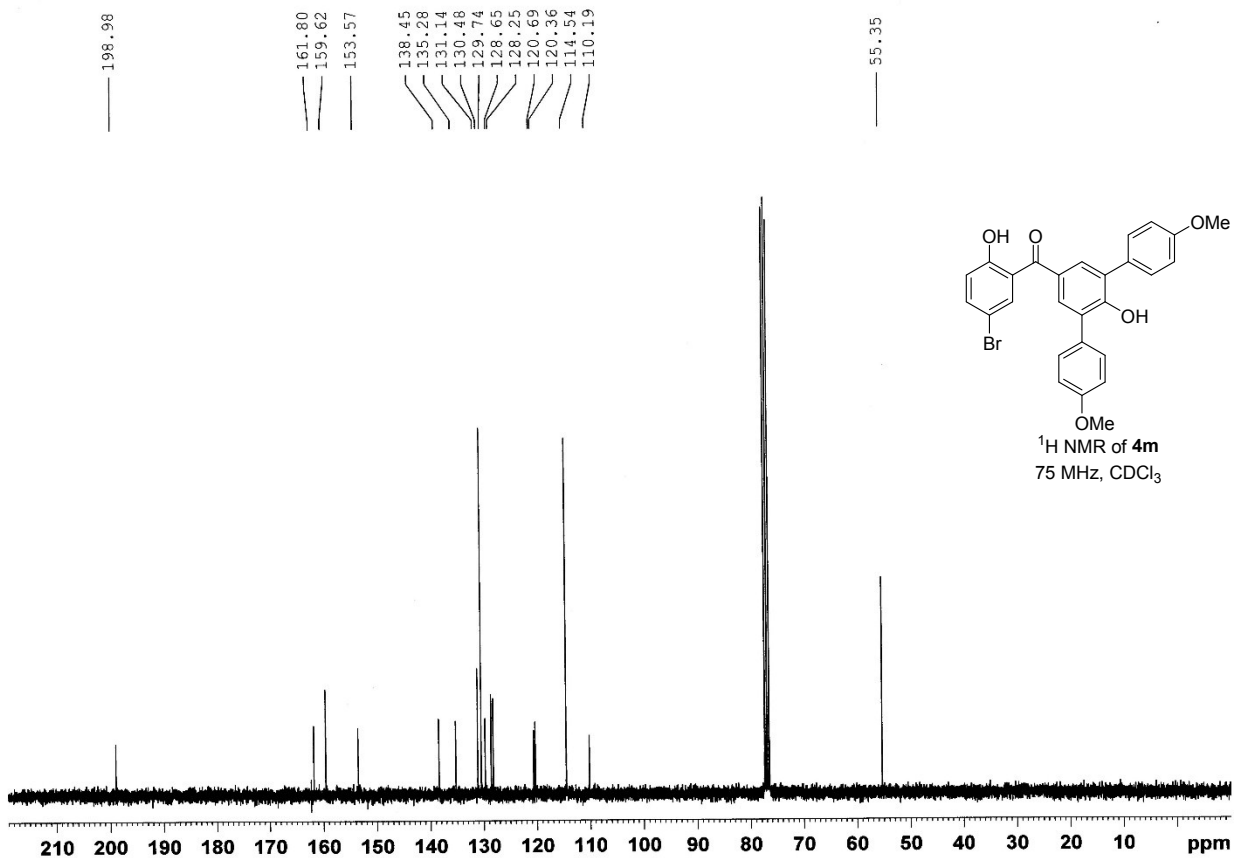




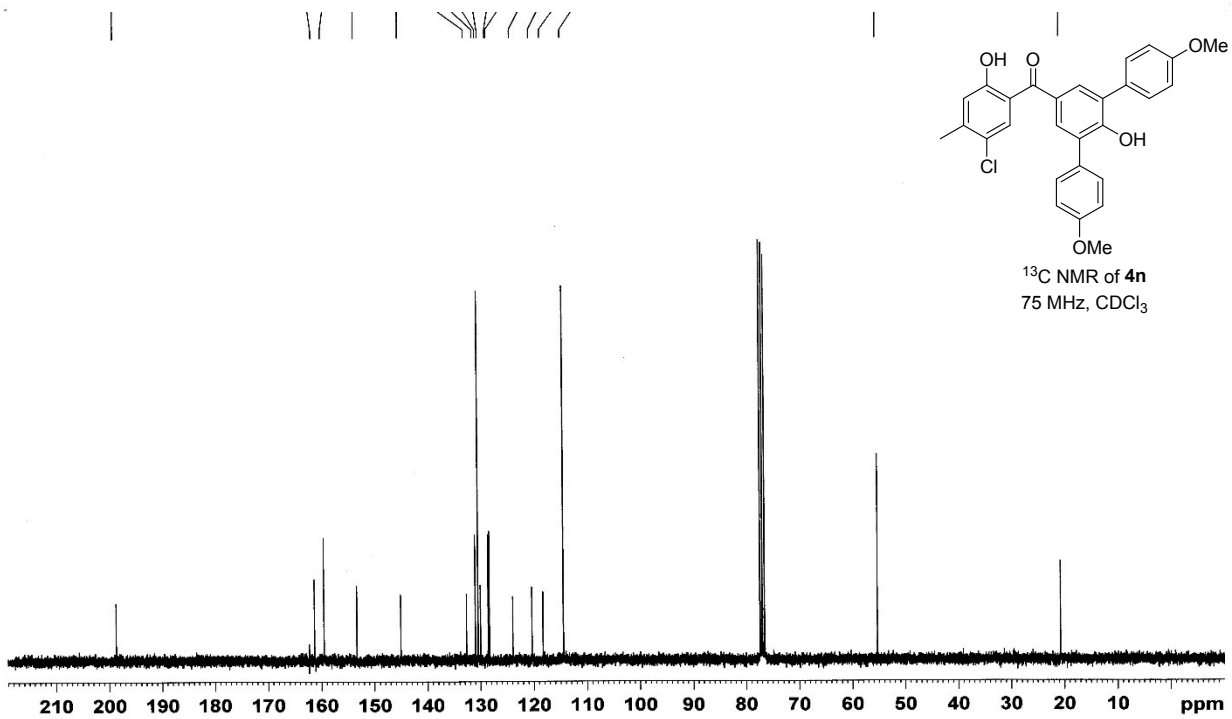
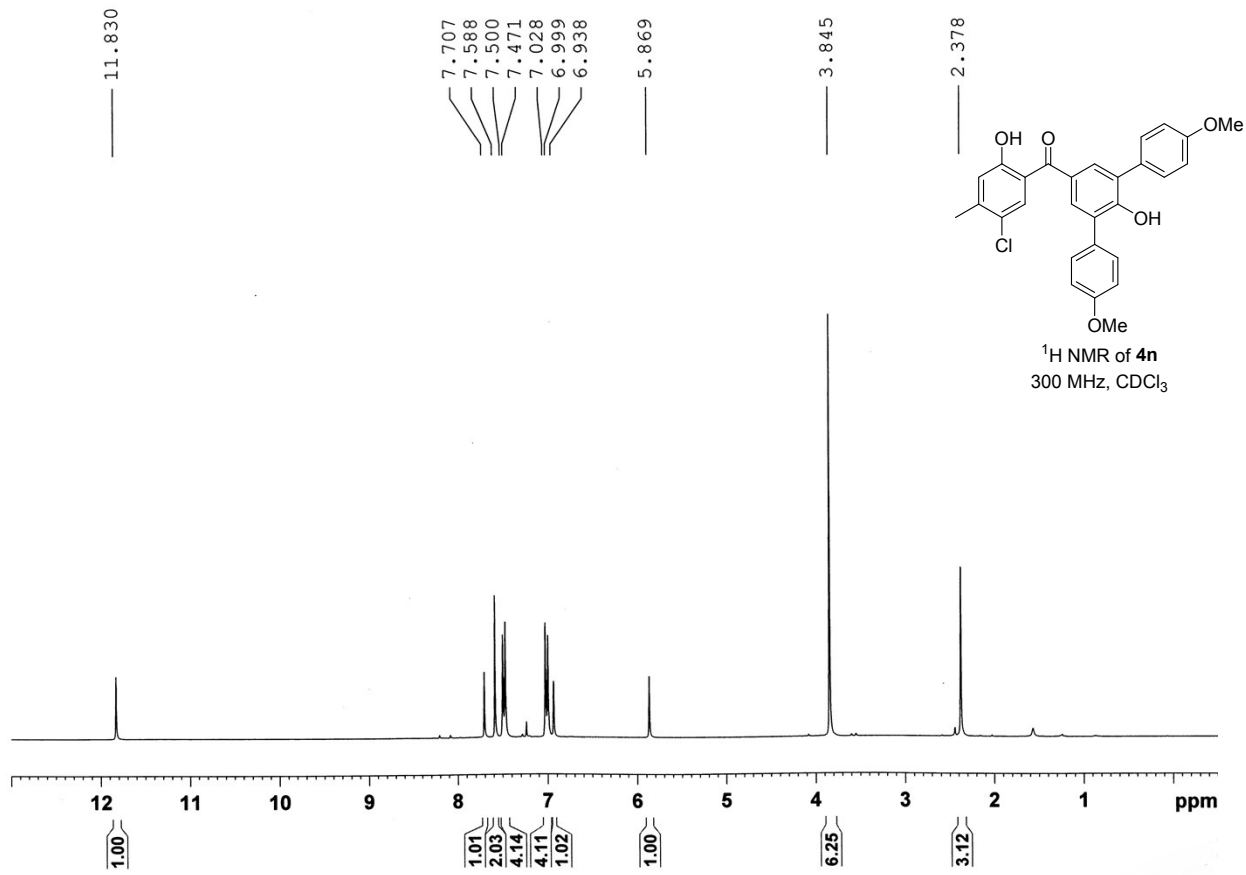


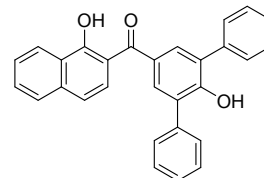
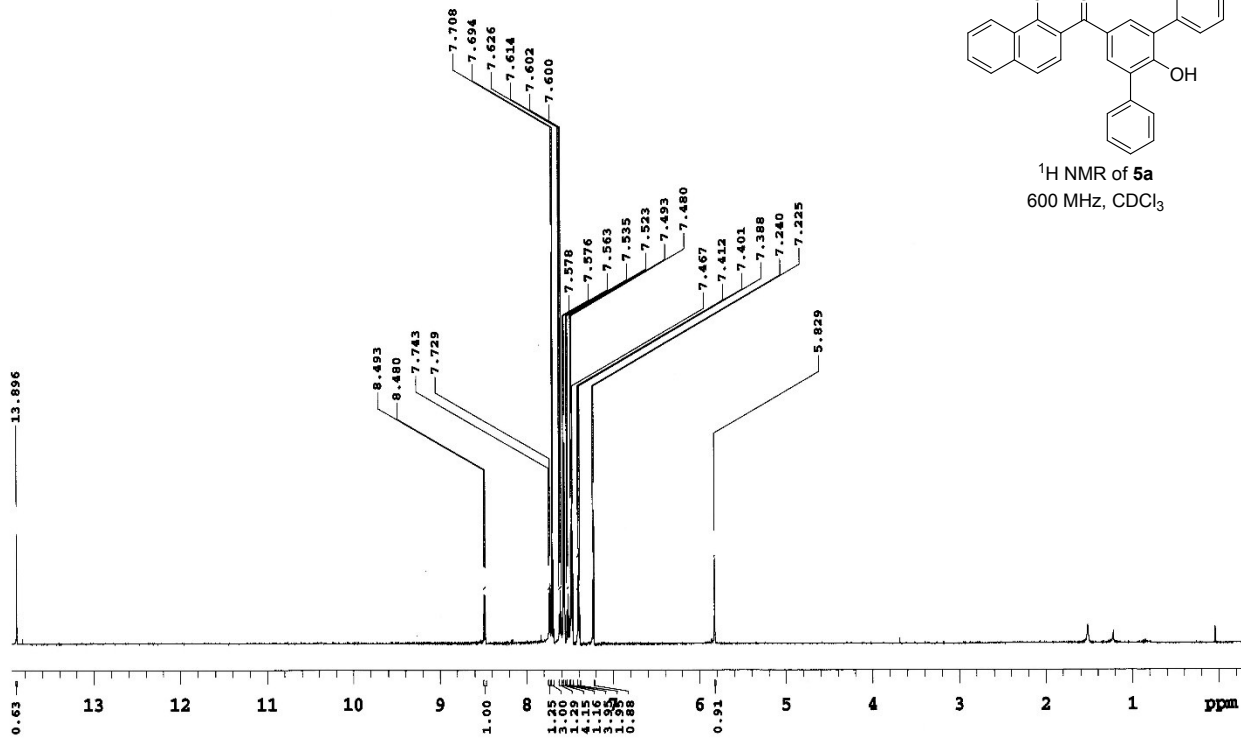


¹H NMR of 4m
300 MHz, CDCl₃

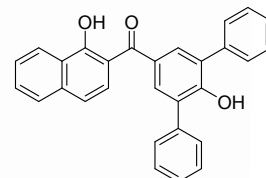
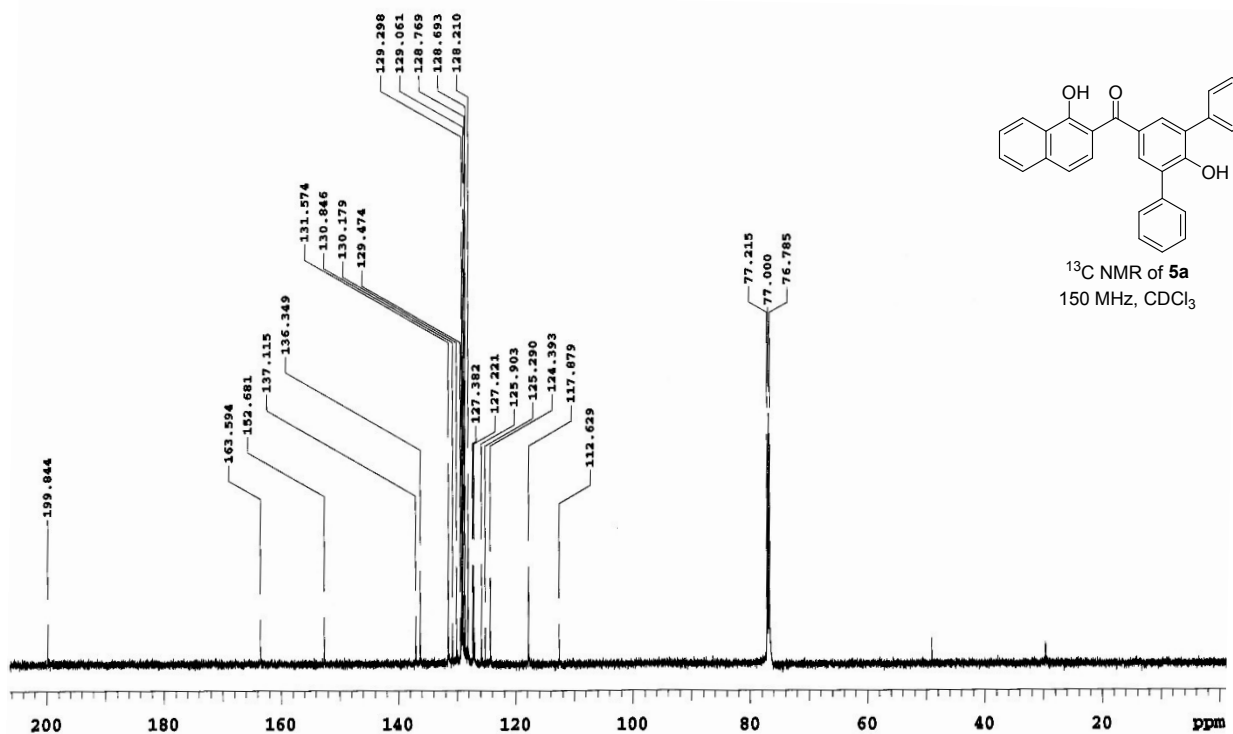


¹³C NMR of 4m
75 MHz, CDCl₃

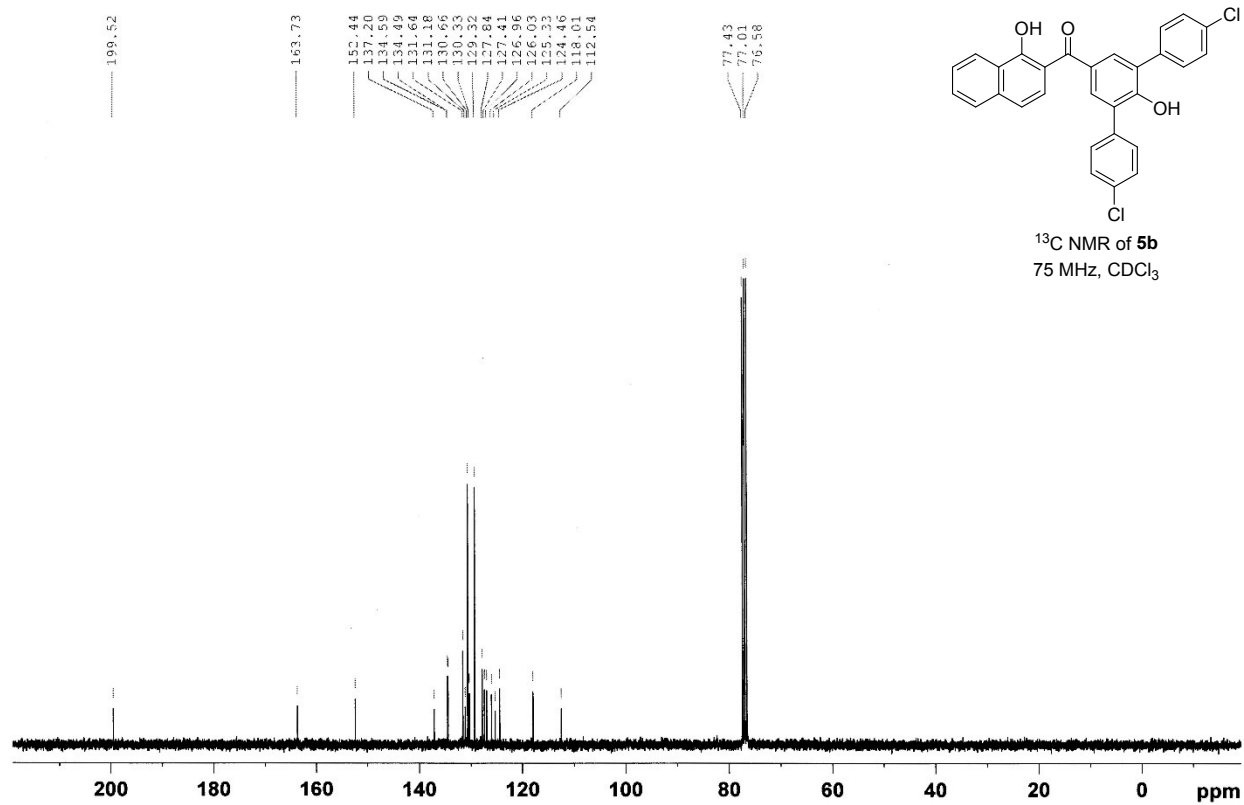
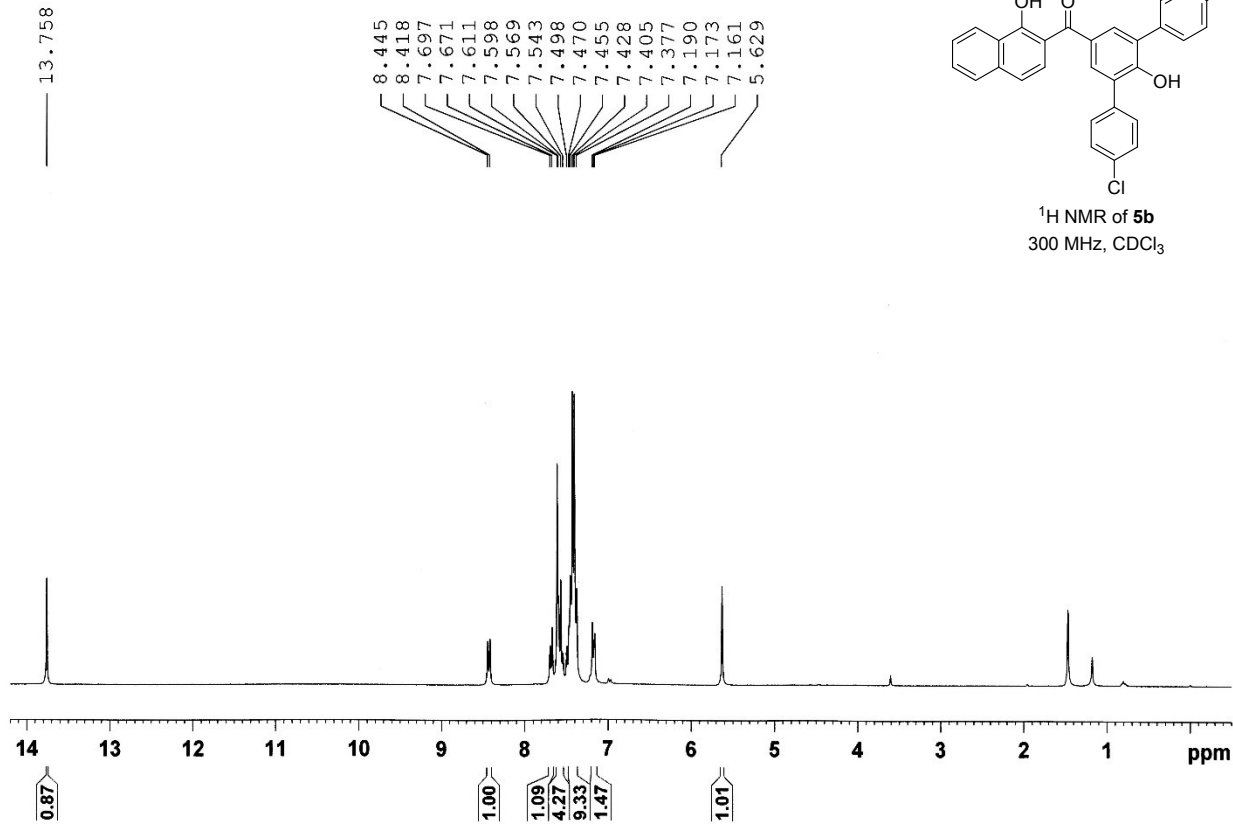


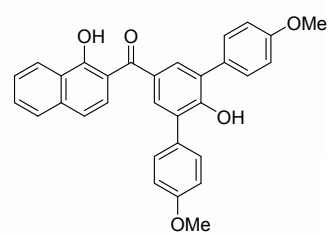
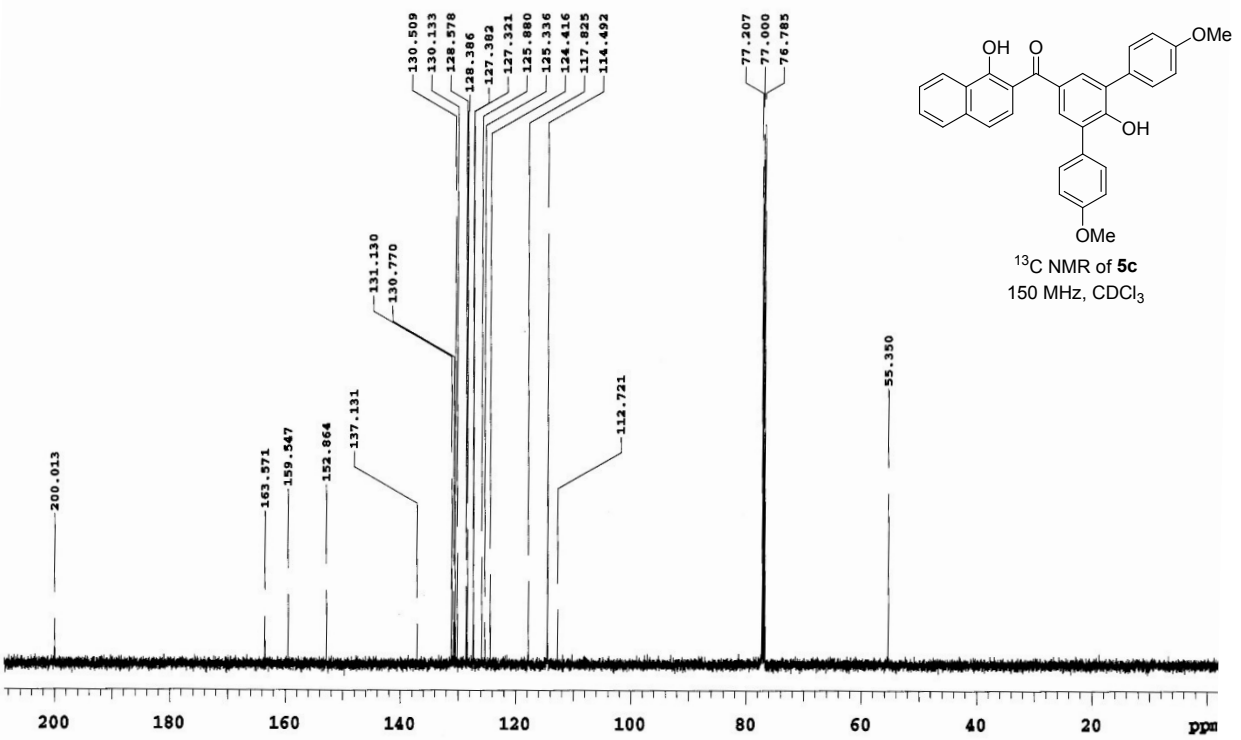
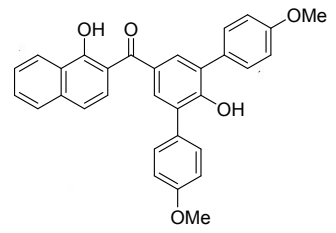
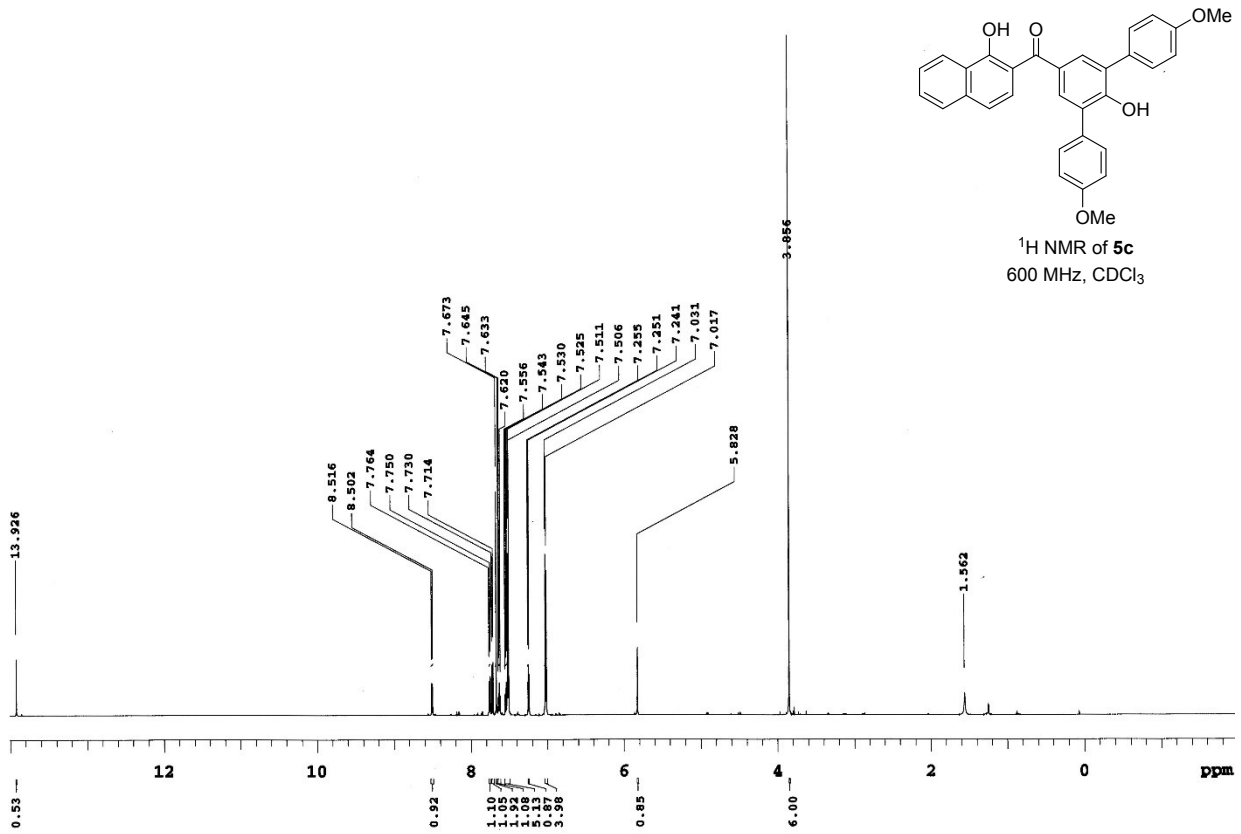


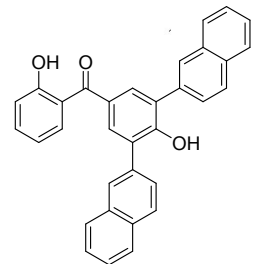
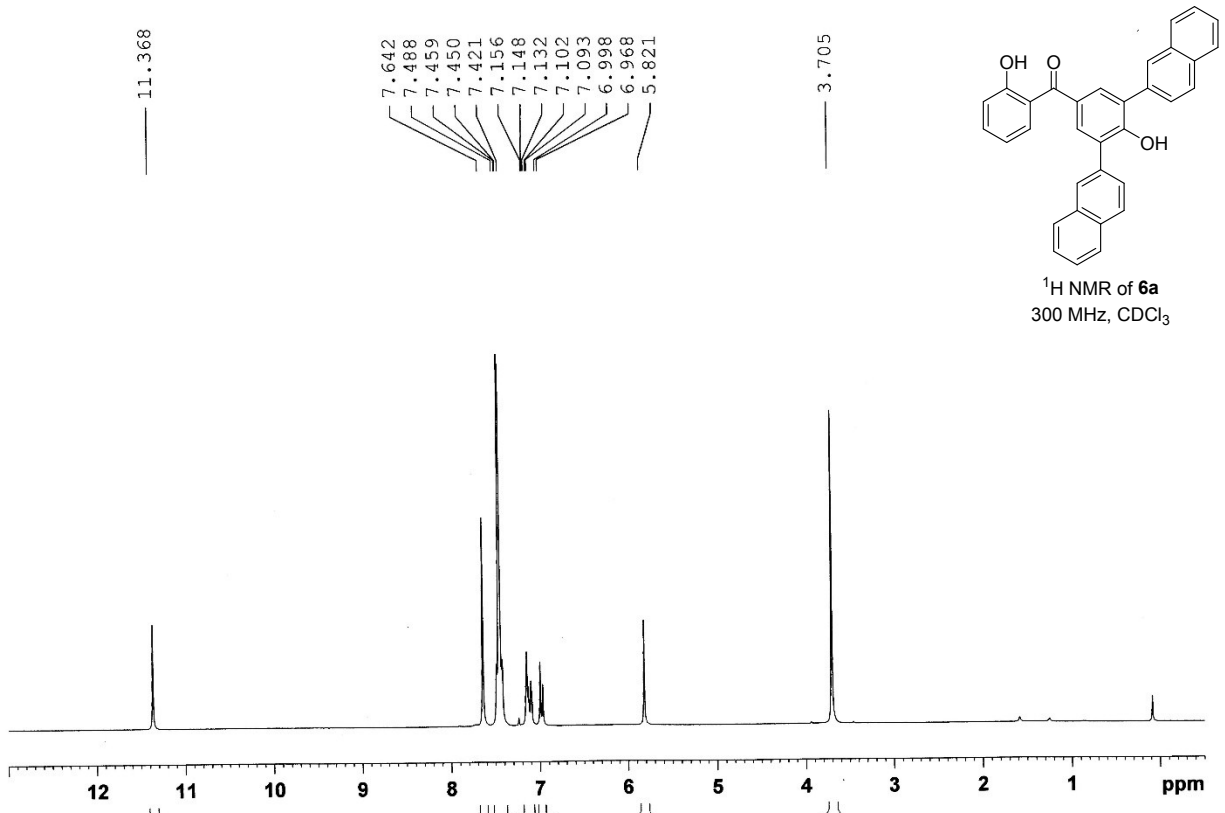
¹H NMR of 5a
600 MHz, CDCl₃



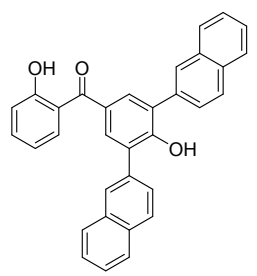
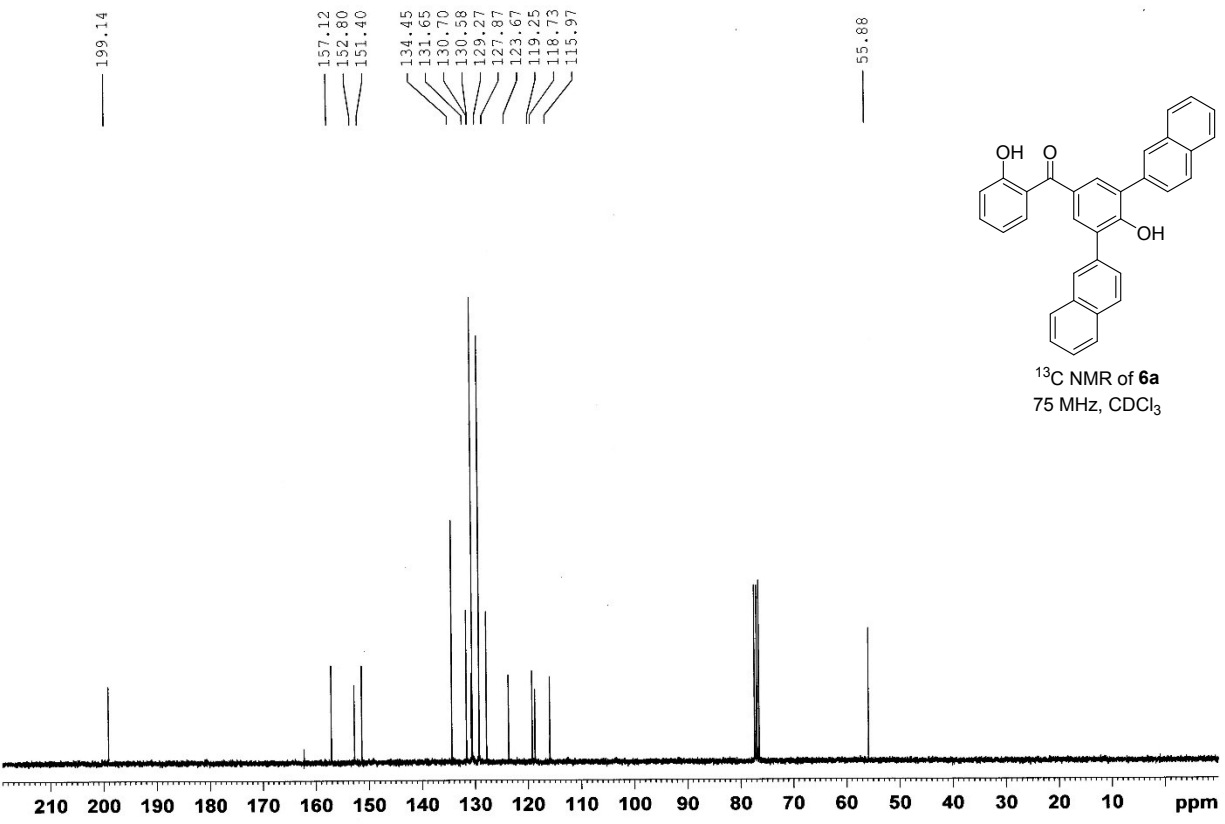
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150 MHz, CDCl₃



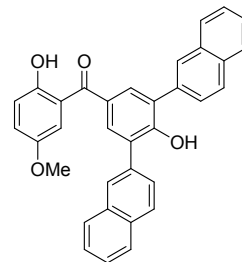
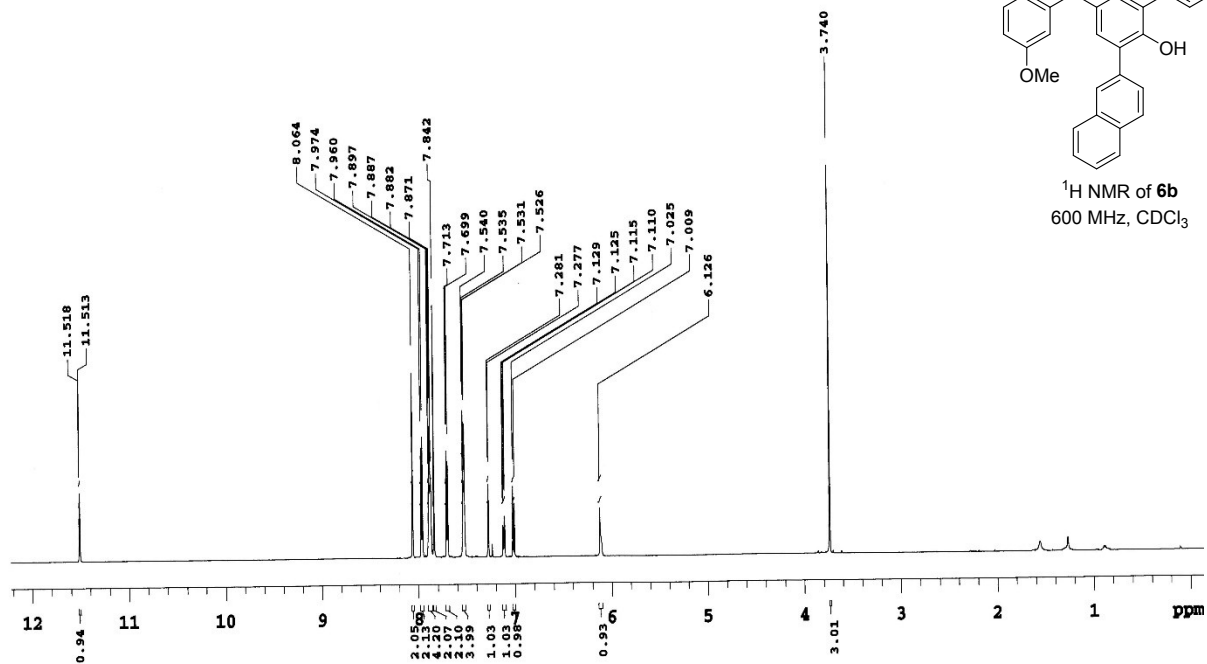




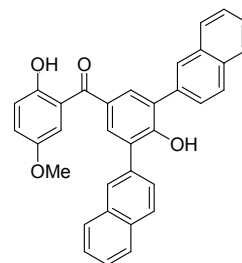
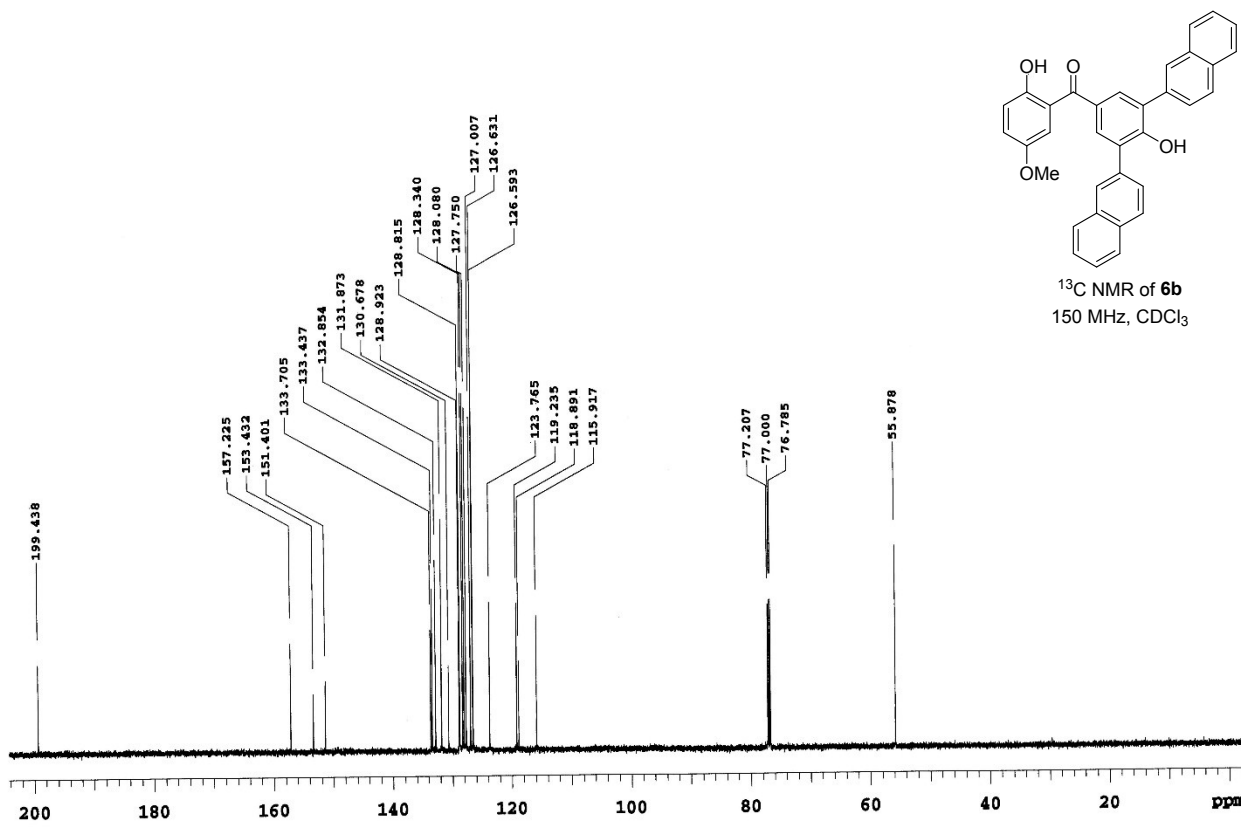
¹H NMR of 6a
300 MHz, CDCl₃



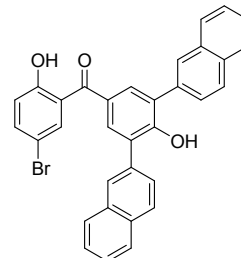
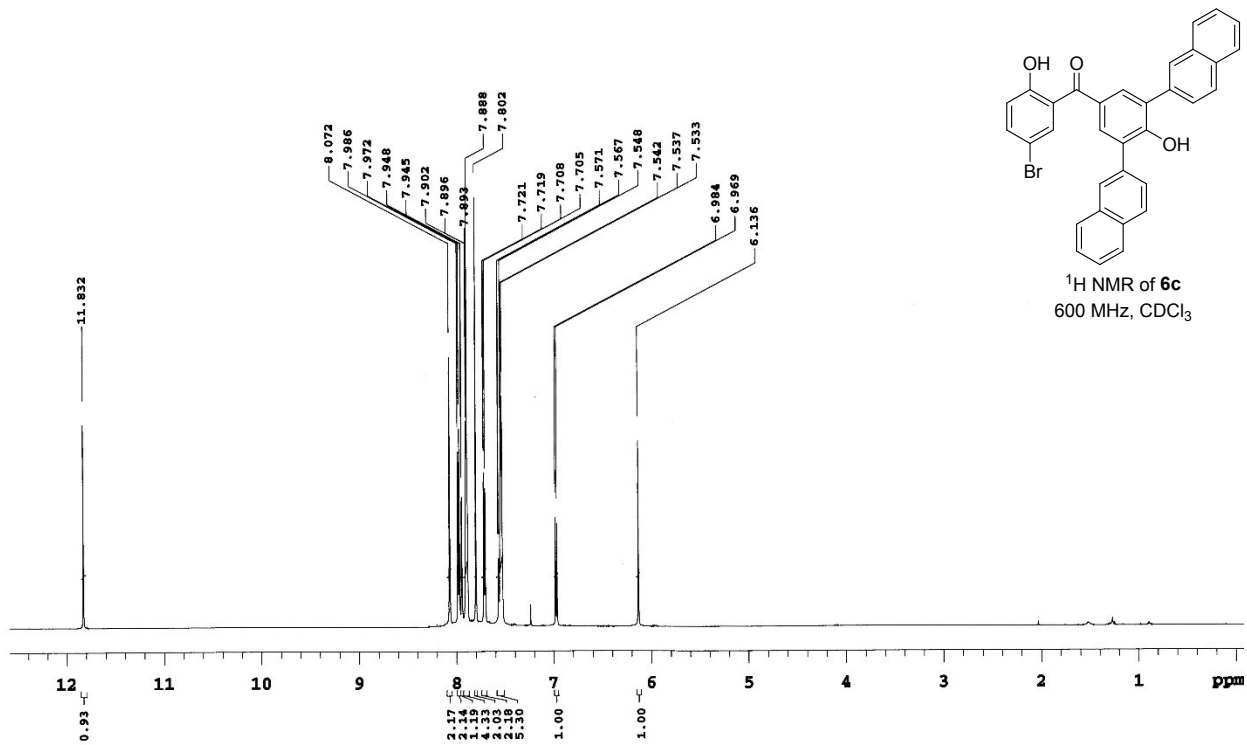
¹³C NMR of 6a
75 MHz, CDCl₃



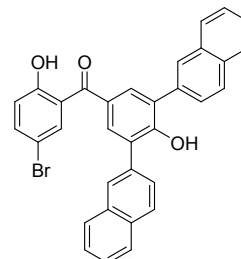
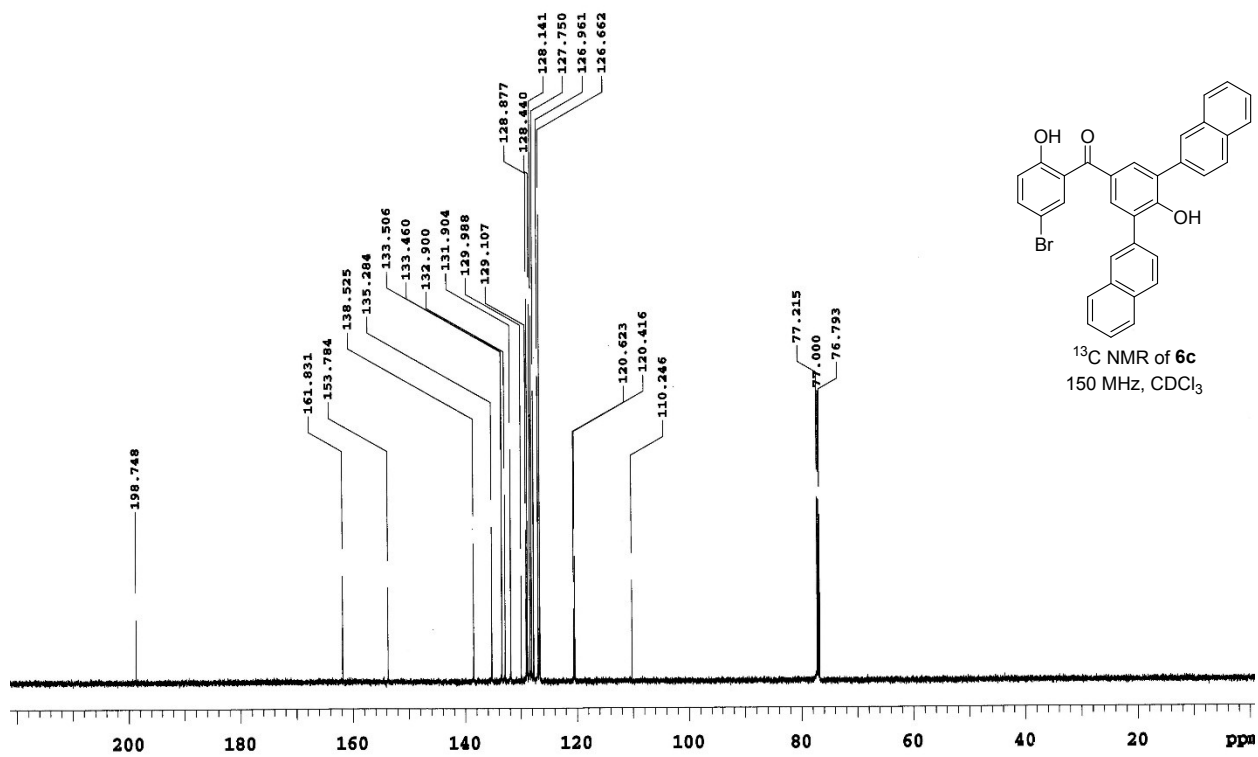
¹H NMR of 6b
600 MHz, CDCl₃



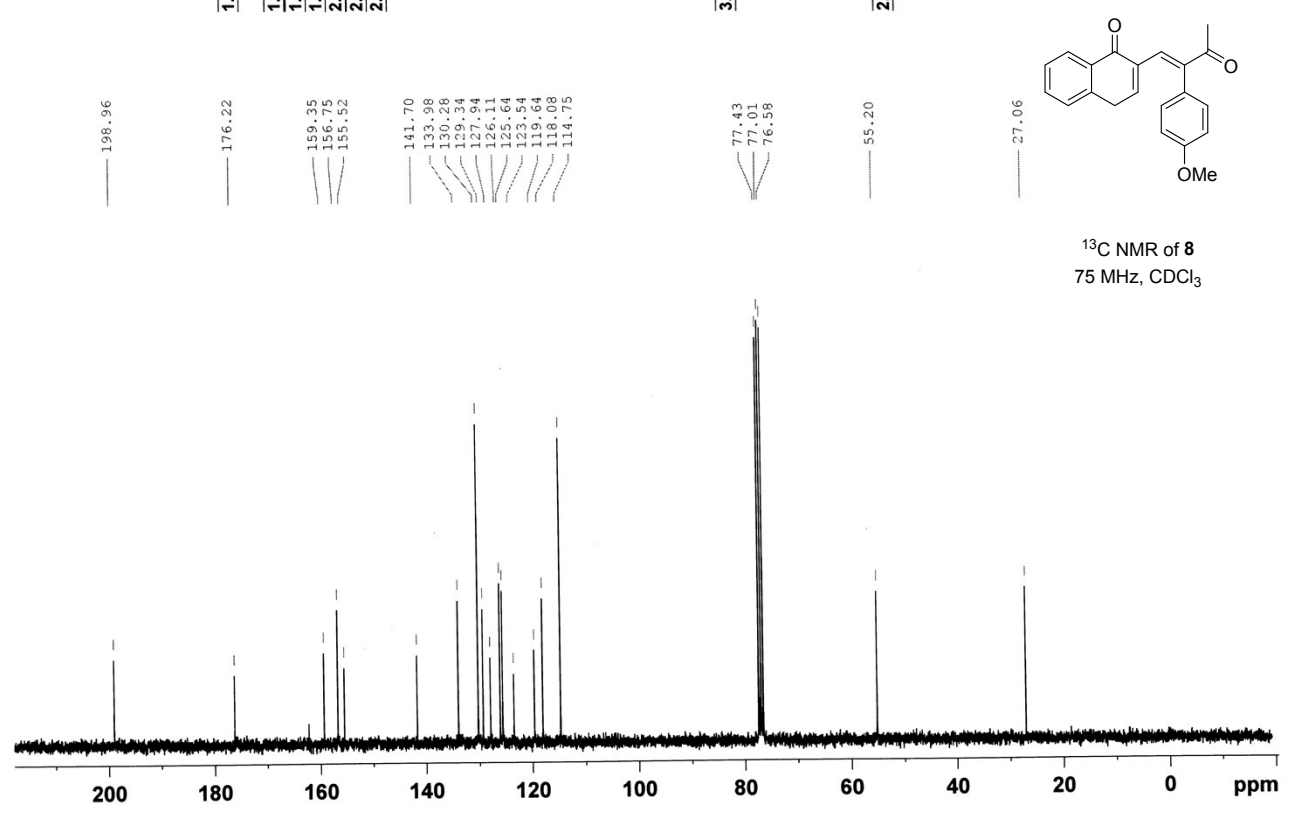
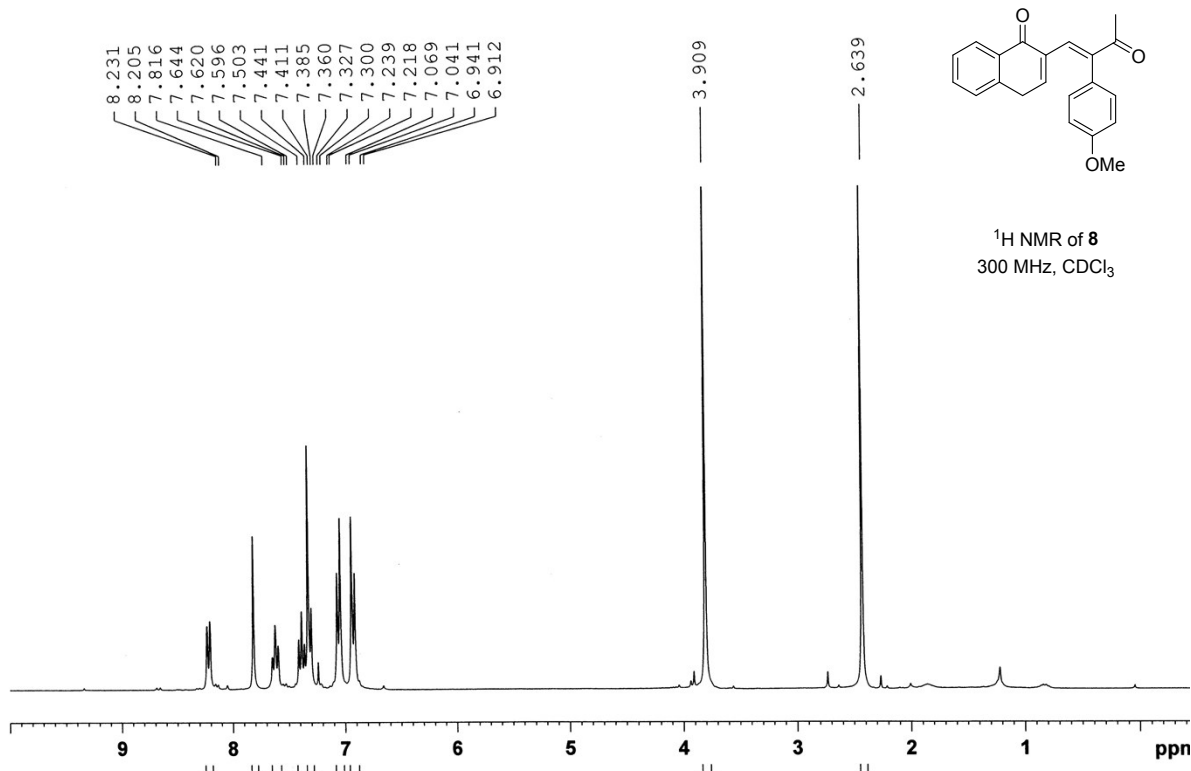
¹³C NMR of 6b
150 MHz, CDCl₃



¹H NMR of 6c
600 MHz, CDCl₃



¹³C NMR of 6c
150 MHz, CDCl₃



Procedure for the UV absorption experiment

Absorbance spectra of the tested compounds were recorded at room temperature (298 K) using UV/Vis spectrophotometer (Optizen UV-3200). The samples were prepared in an ethanol solvent (Sigma-Aldrich, $\geq 99.8\%$ pure) at a concentration of 50 μM . The data were corrected for solvent background by the instrument's calibration using the ethanol as a blank. The absorption spectra of samples in solution were obtained in the range of 220-600 nm at 1 nm interval in 3 determination using three trial samples. The critical wavelength (λ_c) and UVA/UVB ratio were calculated using equation 1 and 2, respectively, as shown below:

$$\int_{290}^{\lambda_c} A(\lambda)d\lambda = 0.9 \int_{290}^{400} A(\lambda)d\lambda \quad (1)$$
$$\frac{\int_{320}^{400} A(\lambda)d\lambda / \int_{320}^{400} d\lambda}{\int_{290}^{320} A(\lambda)d\lambda / \int_{290}^{320} d\lambda} \quad (2)$$

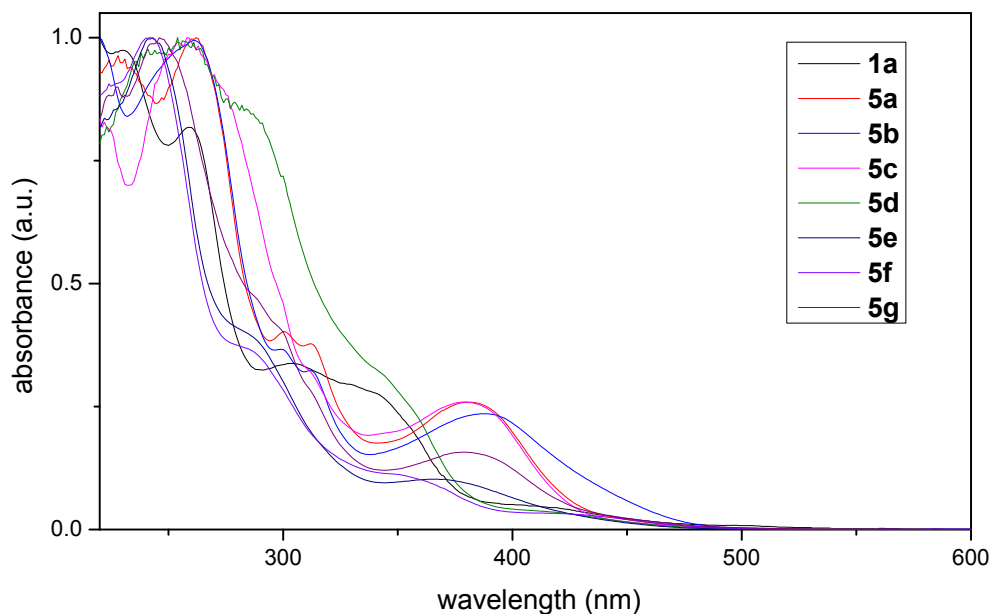


Fig. S1 Normalized UV absorbance spectra of compounds **1a**, **5a-5g** (220-600 nm)

X-Ray crystallographic structure and data of compound 3c: Empirical Formula- $C_{27}H_{22}O_3$, $M = 394.45$, Monoclinic, Space group P_{bca} , $a = 12.4924(9) \text{ \AA}$, $b = 8.6494(5) \text{ \AA}$, $c = 20.0211(14) \text{ \AA}$, $V = 2086.5(2) \text{ \AA}^3$, $Z = 4$, $T = 223(2) \text{ K}$, $\rho_{\text{calcd}} = 1.256 \text{ Mg/m}^3$, $2\theta_{\text{max.}} = 26.07^\circ$, Refinement of 274 parameters on 4126 independent reflections out of 58124 collected reflections ($R_{\text{int}} = 0.1692$) led to $R_1 = 0.0487$ [$I > 2\sigma(I)$], $wR_2 = 0.1181$ (all data) and $S = 1.029$ with the largest difference peak and hole of 0.218 and -0.211 $e. \text{ \AA}^{-3}$ respectively. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre (CCDC 1484692). The data can be obtained free of charge via the Internet at www.ccdc.cam.ac.uk/data_request/cif.

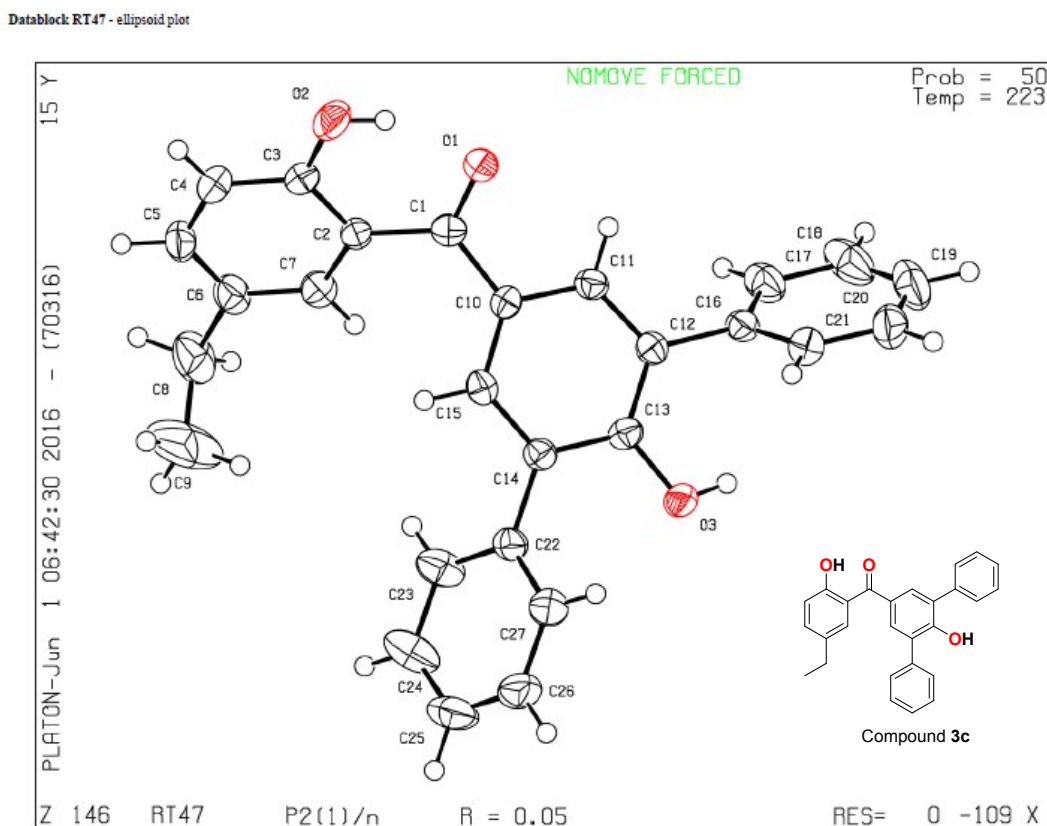


Fig. S2 X-Ray crystallographic structure of compound 3c.

Table 1. Crystal data and structure refinement for **3c**.

Identification code	RT47	
Empirical formula	C ₂₇ H ₂₂ O ₃	
Formula weight	394.45	
Temperature	223(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2(1)/n	
Unit cell dimensions	a = 12.4924(9) Å	α = 90°.
	b = 8.6494(5) Å	β = 105.311(4)°.
	c = 20.0211(14) Å	γ = 90°.
Volume	2086.5(2) Å ³	
Z	4	
Density (calculated)	1.256 Mg/m ³	
Absorption coefficient	0.081 mm ⁻¹	
F(000)	832	
Crystal size	0.30 x 0.22 x 0.20 mm ³	
Theta range for data collection	2.11 to 26.07°.	
Index ranges	-15 ≤ h ≤ 15, -10 ≤ k ≤ 10, -24 ≤ l ≤ 24	
Reflections collected	58124	
Independent reflections	4126 [R(int) = 0.1692]	
Completeness to theta = 26.07°	99.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9840 and 0.9762	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	4126 / 0 / 274	
Goodness-of-fit on F ²	1.029	
Final R indices [I > 2σ(I)]	R1 = 0.0487, wR2 = 0.0968	
R indices (all data)	R1 = 0.1085, wR2 = 0.1181	
Largest diff. peak and hole	0.218 and -0.211 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3c**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	$U(\text{eq})$
C(1)	4707(2)	6601(2)	4064(1)	27(1)
O(1)	4123(1)	7646(2)	4211(1)	34(1)
C(2)	5818(2)	6310(2)	4525(1)	27(1)
C(3)	6344(2)	7461(2)	4993(1)	34(1)
C(4)	7408(2)	7212(3)	5406(1)	44(1)
C(5)	7927(2)	5822(3)	5381(1)	45(1)
C(6)	7421(2)	4631(3)	4945(1)	39(1)
C(7)	6374(2)	4907(2)	4526(1)	33(1)
O(2)	5861(1)	8833(2)	5051(1)	50(1)
C(8)	8010(2)	3125(3)	4911(1)	60(1)
C(9)	8572(3)	3109(4)	4337(2)	111(1)
C(10)	4267(2)	5743(2)	3409(1)	27(1)
C(11)	3123(2)	5625(2)	3151(1)	28(1)
C(12)	2637(2)	5044(2)	2498(1)	28(1)
C(13)	3331(2)	4548(2)	2089(1)	28(1)
C(14)	4488(2)	4617(2)	2339(1)	27(1)
C(15)	4935(2)	5221(2)	2998(1)	28(1)
C(16)	1408(2)	4954(2)	2243(1)	31(1)
C(17)	801(2)	4061(3)	2585(1)	43(1)
C(18)	-349(2)	3987(3)	2342(2)	60(1)
C(19)	-880(2)	4792(3)	1757(2)	63(1)
C(20)	-291(2)	5686(3)	1419(1)	54(1)
C(21)	847(2)	5782(3)	1661(1)	42(1)
O(3)	2925(1)	3970(2)	1443(1)	37(1)
C(22)	5229(2)	3979(2)	1936(1)	30(1)
C(23)	6067(2)	2962(3)	2260(1)	47(1)
C(24)	6756(2)	2283(3)	1906(2)	61(1)
C(25)	6618(2)	2617(3)	1217(2)	55(1)
C(26)	5823(2)	3642(3)	896(1)	55(1)
C(27)	5129(2)	4310(3)	1249(1)	46(1)

Table 3. Bond lengths [Å] and angles [°] for **3c**.

C(1)-O(1)#1	1.245(2)
C(1)-O(1)	1.245(2)
C(1)-C(2)	1.471(3)
C(1)-C(10)	1.480(3)
O(1)-O(1)#1	0.000(4)
C(2)-C(7)	1.397(3)
C(2)-C(3)	1.406(3)
C(3)-O(2)	1.350(2)
C(3)-C(4)	1.384(3)
C(4)-C(5)	1.374(3)
C(4)-H(4)	0.9400
C(5)-C(6)	1.389(3)
C(5)-H(5)	0.9400
C(6)-C(7)	1.376(3)
C(6)-C(8)	1.507(3)
C(7)-H(7)	0.9400
O(2)-H(2)	0.8300
C(8)-C(9)	1.496(4)
C(8)-H(8A)	0.9800
C(8)-H(8B)	0.9800
C(9)-H(9A)	0.9700
C(9)-H(9B)	0.9700
C(9)-H(9C)	0.9700
C(10)-C(11)	1.389(3)
C(10)-C(15)	1.393(3)
C(11)-C(12)	1.382(3)
C(11)-H(11)	0.9400
C(12)-C(13)	1.407(3)
C(12)-C(16)	1.486(3)
C(13)-O(3)	1.356(2)
C(13)-C(14)	1.401(3)
C(14)-C(15)	1.391(3)
C(14)-C(22)	1.485(3)
C(15)-H(15)	0.9400

C(16)-C(17)	1.385(3)
C(16)-C(21)	1.389(3)
C(17)-C(18)	1.391(3)
C(17)-H(17)	0.9400
C(18)-C(19)	1.374(4)
C(18)-H(18)	0.9400
C(19)-C(20)	1.365(4)
C(19)-H(19)	0.9400
C(20)-C(21)	1.378(3)
C(20)-H(20)	0.9400
C(21)-H(21)	0.9400
O(3)-H(3)	0.8300
C(22)-C(27)	1.379(3)
C(22)-C(23)	1.390(3)
C(23)-C(24)	1.381(3)
C(23)-H(23)	0.9400
C(24)-C(25)	1.375(4)
C(24)-H(24)	0.9400
C(25)-C(26)	1.360(4)
C(25)-H(25)	0.9400
C(26)-C(27)	1.382(3)
C(26)-H(26)	0.9400
C(27)-H(27)	0.9400
O(1)#1-C(1)-O(1)	0.00(13)
O(1)#1-C(1)-C(2)	119.35(17)
O(1)-C(1)-C(2)	119.35(17)
O(1)#1-C(1)-C(10)	118.03(18)
O(1)-C(1)-C(10)	118.03(18)
C(2)-C(1)-C(10)	122.57(18)
O(1)#1-O(1)-C(1)	0(10)
C(7)-C(2)-C(3)	117.99(18)
C(7)-C(2)-C(1)	122.38(18)
C(3)-C(2)-C(1)	119.61(18)
O(2)-C(3)-C(4)	117.75(19)
O(2)-C(3)-C(2)	122.64(18)

C(4)-C(3)-C(2)	119.6(2)
C(5)-C(4)-C(3)	120.2(2)
C(5)-C(4)-H(4)	119.9
C(3)-C(4)-H(4)	119.9
C(4)-C(5)-C(6)	122.0(2)
C(4)-C(5)-H(5)	119.0
C(6)-C(5)-H(5)	119.0
C(7)-C(6)-C(5)	117.2(2)
C(7)-C(6)-C(8)	121.3(2)
C(5)-C(6)-C(8)	121.5(2)
C(6)-C(7)-C(2)	122.9(2)
C(6)-C(7)-H(7)	118.5
C(2)-C(7)-H(7)	118.5
C(3)-O(2)-H(2)	109.5
C(9)-C(8)-C(6)	112.1(2)
C(9)-C(8)-H(8A)	109.2
C(6)-C(8)-H(8A)	109.2
C(9)-C(8)-H(8B)	109.2
C(6)-C(8)-H(8B)	109.2
H(8A)-C(8)-H(8B)	107.9
C(8)-C(9)-H(9A)	109.5
C(8)-C(9)-H(9B)	109.5
H(9A)-C(9)-H(9B)	109.5
C(8)-C(9)-H(9C)	109.5
H(9A)-C(9)-H(9C)	109.5
H(9B)-C(9)-H(9C)	109.5
C(11)-C(10)-C(15)	118.36(18)
C(11)-C(10)-C(1)	118.12(18)
C(15)-C(10)-C(1)	123.01(18)
C(12)-C(11)-C(10)	121.99(18)
C(12)-C(11)-H(11)	119.0
C(10)-C(11)-H(11)	119.0
C(11)-C(12)-C(13)	118.49(18)
C(11)-C(12)-C(16)	120.03(18)
C(13)-C(12)-C(16)	121.48(18)
O(3)-C(13)-C(14)	116.54(17)

O(3)-C(13)-C(12)	122.43(18)
C(14)-C(13)-C(12)	121.02(18)
C(15)-C(14)-C(13)	118.20(18)
C(15)-C(14)-C(22)	120.18(18)
C(13)-C(14)-C(22)	121.51(17)
C(14)-C(15)-C(10)	121.91(18)
C(14)-C(15)-H(15)	119.0
C(10)-C(15)-H(15)	119.0
C(17)-C(16)-C(21)	118.7(2)
C(17)-C(16)-C(12)	120.67(19)
C(21)-C(16)-C(12)	120.58(19)
C(16)-C(17)-C(18)	120.2(2)
C(16)-C(17)-H(17)	119.9
C(18)-C(17)-H(17)	119.9
C(19)-C(18)-C(17)	119.7(3)
C(19)-C(18)-H(18)	120.1
C(17)-C(18)-H(18)	120.1
C(20)-C(19)-C(18)	120.6(2)
C(20)-C(19)-H(19)	119.7
C(18)-C(19)-H(19)	119.7
C(19)-C(20)-C(21)	120.0(3)
C(19)-C(20)-H(20)	120.0
C(21)-C(20)-H(20)	120.0
C(20)-C(21)-C(16)	120.7(2)
C(20)-C(21)-H(21)	119.7
C(16)-C(21)-H(21)	119.7
C(13)-O(3)-H(3)	109.5
C(27)-C(22)-C(23)	117.0(2)
C(27)-C(22)-C(14)	124.06(19)
C(23)-C(22)-C(14)	118.89(18)
C(24)-C(23)-C(22)	121.7(2)
C(24)-C(23)-H(23)	119.2
C(22)-C(23)-H(23)	119.2
C(25)-C(24)-C(23)	119.7(3)
C(25)-C(24)-H(24)	120.1
C(23)-C(24)-H(24)	120.1

C(26)-C(25)-C(24)	119.6(2)
C(26)-C(25)-H(25)	120.2
C(24)-C(25)-H(25)	120.2
C(25)-C(26)-C(27)	120.6(2)
C(25)-C(26)-H(26)	119.7
C(27)-C(26)-H(26)	119.7
C(22)-C(27)-C(26)	121.3(2)
C(22)-C(27)-H(27)	119.3
C(26)-C(27)-H(27)	119.3

Symmetry transformations used to generate equivalent atoms:

#1 x,y,z

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3c**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
C(1)	31(1)	26(1)	27(1)	2(1)	11(1)	-1(1)
O(1)	32(1)	38(1)	31(1)	-5(1)	6(1)	7(1)
C(2)	25(1)	35(1)	21(1)	4(1)	7(1)	2(1)
C(3)	34(1)	44(1)	23(1)	-3(1)	7(1)	4(1)
C(4)	36(1)	62(2)	30(1)	-10(1)	2(1)	-1(1)
C(5)	26(1)	74(2)	31(1)	5(1)	3(1)	6(1)
C(6)	35(1)	48(1)	34(1)	13(1)	10(1)	10(1)
C(7)	34(1)	34(1)	31(1)	5(1)	10(1)	2(1)
O(2)	49(1)	51(1)	41(1)	-21(1)	-6(1)	12(1)
C(8)	50(2)	57(2)	68(2)	14(1)	8(1)	22(1)
C(9)	138(3)	97(3)	119(3)	17(2)	71(3)	73(3)
C(10)	27(1)	28(1)	24(1)	1(1)	4(1)	0(1)
C(11)	29(1)	32(1)	27(1)	0(1)	11(1)	2(1)
C(12)	25(1)	30(1)	29(1)	0(1)	8(1)	-2(1)
C(13)	31(1)	29(1)	25(1)	-2(1)	9(1)	-5(1)
C(14)	27(1)	29(1)	27(1)	0(1)	10(1)	-3(1)
C(15)	24(1)	28(1)	31(1)	3(1)	7(1)	0(1)
C(16)	26(1)	33(1)	34(1)	-10(1)	9(1)	-4(1)
C(17)	35(1)	41(1)	57(2)	-8(1)	18(1)	-5(1)
C(18)	42(2)	47(2)	101(2)	-19(2)	36(2)	-11(1)
C(19)	28(2)	57(2)	97(2)	-33(2)	5(2)	1(1)
C(20)	38(2)	58(2)	56(2)	-17(1)	-4(1)	13(1)
C(21)	35(1)	50(1)	38(1)	-7(1)	5(1)	2(1)
O(3)	28(1)	54(1)	31(1)	-13(1)	9(1)	-9(1)
C(22)	26(1)	32(1)	34(1)	-6(1)	13(1)	-8(1)
C(23)	52(2)	45(1)	53(2)	6(1)	30(1)	7(1)
C(24)	60(2)	50(2)	86(2)	3(2)	42(2)	15(1)
C(25)	50(2)	60(2)	69(2)	-27(2)	38(2)	-13(1)
C(26)	45(2)	89(2)	38(1)	-15(1)	22(1)	-11(2)
C(27)	37(1)	67(2)	35(1)	-3(1)	13(1)	-1(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3c**.

	x	y	z	U(eq)
H(4)	7776	7997	5704	52
H(5)	8648	5671	5668	54
H(7)	6015	4116	4228	40
H(2)	5215	8833	4800	75
H(8A)	8565	2952	5353	71
H(8B)	7472	2276	4842	71
H(9A)	8025	3275	3899	166
H(9B)	8930	2117	4329	166
H(9C)	9124	3925	4412	166
H(11)	2665	5951	3429	34
H(15)	5710	5279	3170	33
H(17)	1166	3503	2983	52
H(18)	-762	3388	2578	72
H(19)	-1655	4728	1589	75
H(20)	-662	6237	1020	64
H(21)	1247	6414	1431	50
H(3)	2254	3777	1376	56
H(23)	6168	2730	2731	56
H(24)	7315	1596	2135	73
H(25)	7072	2141	970	66
H(26)	5745	3898	429	66
H(27)	4578	5005	1015	55

Table 6. Torsion angles [°] for **3c**.

C(2)-C(1)-O(1)-O(1)#1	0.0(2)
C(10)-C(1)-O(1)-O(1)#1	0.00(17)
O(1)#1-C(1)-C(2)-C(7)	159.33(18)
O(1)-C(1)-C(2)-C(7)	159.33(18)
C(10)-C(1)-C(2)-C(7)	-23.4(3)
O(1)#1-C(1)-C(2)-C(3)	-19.4(3)
O(1)-C(1)-C(2)-C(3)	-19.4(3)
C(10)-C(1)-C(2)-C(3)	157.93(18)
C(7)-C(2)-C(3)-O(2)	-177.45(19)
C(1)-C(2)-C(3)-O(2)	1.3(3)
C(7)-C(2)-C(3)-C(4)	4.0(3)
C(1)-C(2)-C(3)-C(4)	-177.25(19)
O(2)-C(3)-C(4)-C(5)	178.3(2)
C(2)-C(3)-C(4)-C(5)	-3.1(3)
C(3)-C(4)-C(5)-C(6)	0.5(4)
C(4)-C(5)-C(6)-C(7)	1.0(3)
C(4)-C(5)-C(6)-C(8)	178.6(2)
C(5)-C(6)-C(7)-C(2)	0.1(3)
C(8)-C(6)-C(7)-C(2)	-177.5(2)
C(3)-C(2)-C(7)-C(6)	-2.6(3)
C(1)-C(2)-C(7)-C(6)	178.73(19)
C(7)-C(6)-C(8)-C(9)	82.8(3)
C(5)-C(6)-C(8)-C(9)	-94.7(3)
O(1)#1-C(1)-C(10)-C(11)	-27.5(3)
O(1)-C(1)-C(10)-C(11)	-27.5(3)
C(2)-C(1)-C(10)-C(11)	155.14(18)
O(1)#1-C(1)-C(10)-C(15)	144.08(19)
O(1)-C(1)-C(10)-C(15)	144.08(19)
C(2)-C(1)-C(10)-C(15)	-33.2(3)
C(15)-C(10)-C(11)-C(12)	-1.5(3)
C(1)-C(10)-C(11)-C(12)	170.50(18)
C(10)-C(11)-C(12)-C(13)	0.6(3)
C(10)-C(11)-C(12)-C(16)	-179.47(18)
C(11)-C(12)-C(13)-O(3)	-179.70(18)

C(16)-C(12)-C(13)-O(3)	0.3(3)
C(11)-C(12)-C(13)-C(14)	0.9(3)
C(16)-C(12)-C(13)-C(14)	-179.06(18)
O(3)-C(13)-C(14)-C(15)	179.22(17)
C(12)-C(13)-C(14)-C(15)	-1.4(3)
O(3)-C(13)-C(14)-C(22)	-4.5(3)
C(12)-C(13)-C(14)-C(22)	174.91(18)
C(13)-C(14)-C(15)-C(10)	0.4(3)
C(22)-C(14)-C(15)-C(10)	-175.96(17)
C(11)-C(10)-C(15)-C(14)	1.0(3)
C(1)-C(10)-C(15)-C(14)	-170.56(18)
C(11)-C(12)-C(16)-C(17)	-61.1(3)
C(13)-C(12)-C(16)-C(17)	118.8(2)
C(11)-C(12)-C(16)-C(21)	118.0(2)
C(13)-C(12)-C(16)-C(21)	-62.1(3)
C(21)-C(16)-C(17)-C(18)	0.8(3)
C(12)-C(16)-C(17)-C(18)	179.9(2)
C(16)-C(17)-C(18)-C(19)	0.6(3)
C(17)-C(18)-C(19)-C(20)	-1.1(4)
C(18)-C(19)-C(20)-C(21)	0.3(4)
C(19)-C(20)-C(21)-C(16)	1.1(3)
C(17)-C(16)-C(21)-C(20)	-1.6(3)
C(12)-C(16)-C(21)-C(20)	179.3(2)
C(15)-C(14)-C(22)-C(27)	-133.7(2)
C(13)-C(14)-C(22)-C(27)	50.1(3)
C(15)-C(14)-C(22)-C(23)	47.5(3)
C(13)-C(14)-C(22)-C(23)	-128.7(2)
C(27)-C(22)-C(23)-C(24)	-1.5(3)
C(14)-C(22)-C(23)-C(24)	177.5(2)
C(22)-C(23)-C(24)-C(25)	0.2(4)
C(23)-C(24)-C(25)-C(26)	1.6(4)
C(24)-C(25)-C(26)-C(27)	-2.1(4)
C(23)-C(22)-C(27)-C(26)	0.9(3)
C(14)-C(22)-C(27)-C(26)	-178.0(2)
C(25)-C(26)-C(27)-C(22)	0.9(4)

Symmetry transformations used to generate equivalent atoms:

#1 x,y,z

Table 7. Hydrogen bonds for **3c** [\AA and $^\circ$].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(2)-H(2)...O(1)#1	0.83	1.86	2.582(2)	144.7
O(3)-H(3)...O(1)#2	0.83	2.06	2.7934(19)	147.5
O(2)-H(2)...O(2)#3	0.83	2.49	2.918(3)	113.4

Symmetry transformations used to generate equivalent atoms:

#1 x,y,z #2 -x+1/2,y-1/2,-z+1/2 #3 -x+1,-y+2,-z+1