

An in-situ self-catalytic hybrid cyanate ester resin and its self- catalytic polymerization behavior

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Polymerization of Si-NCE-1 and Si-NCE-2

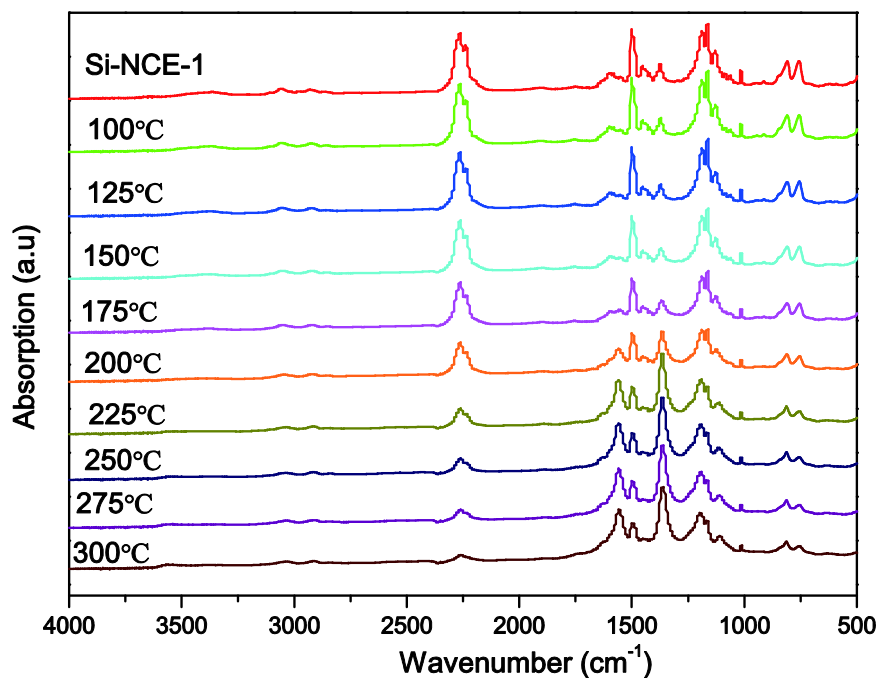


Figure 1S. FT-IR spectra of Si-NCE-1 at different temperature

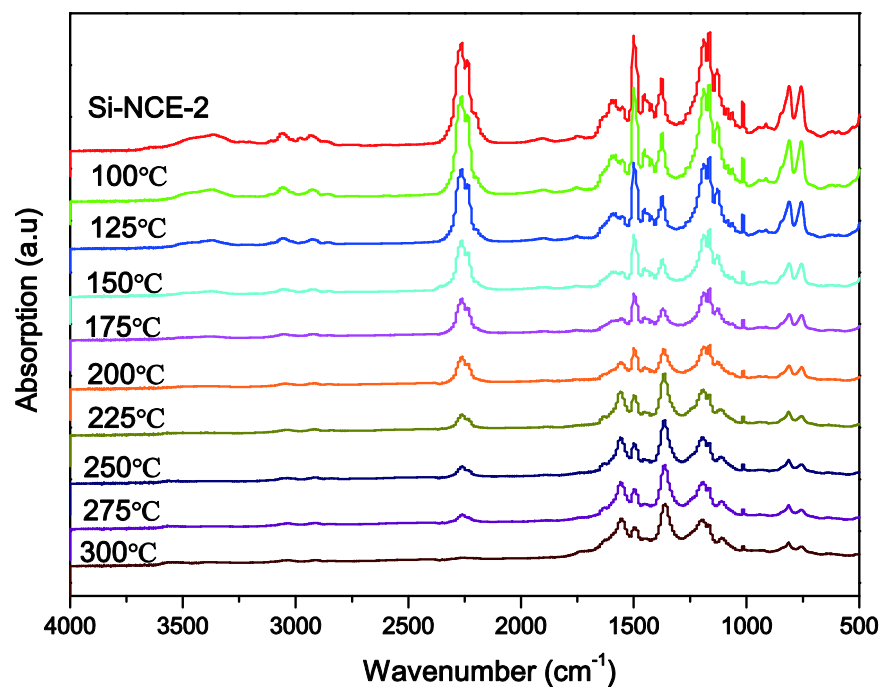


Figure 2S. FT-IR spectra of Si-NCE-2 at different temperature

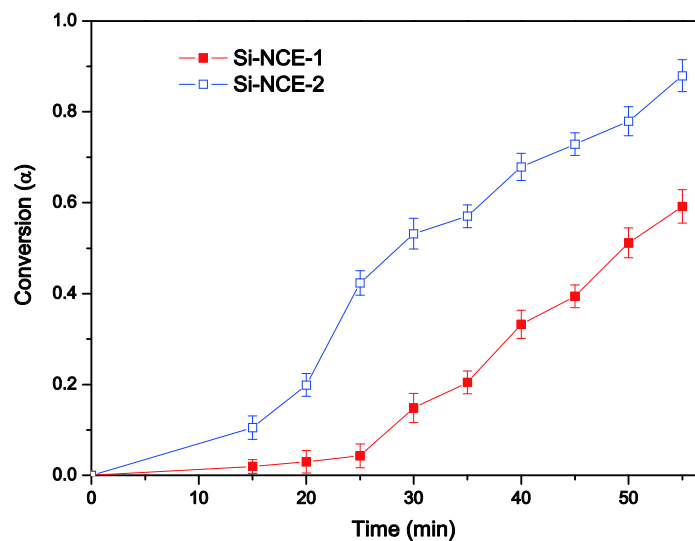


Figure 3S. The curve of conversion of group (-OCN) versus time for Si-NCE-1 and Si-NCE-2.

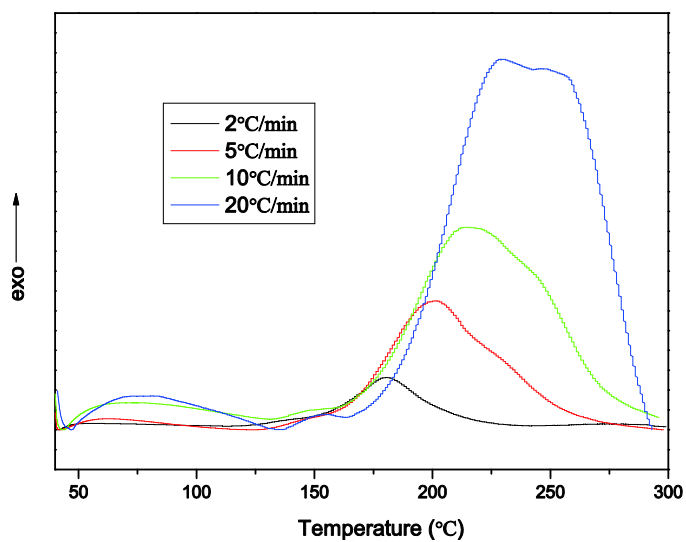


Figure 4S. DSC thermogram of Si-NCE-1 at different heating rate under N₂.

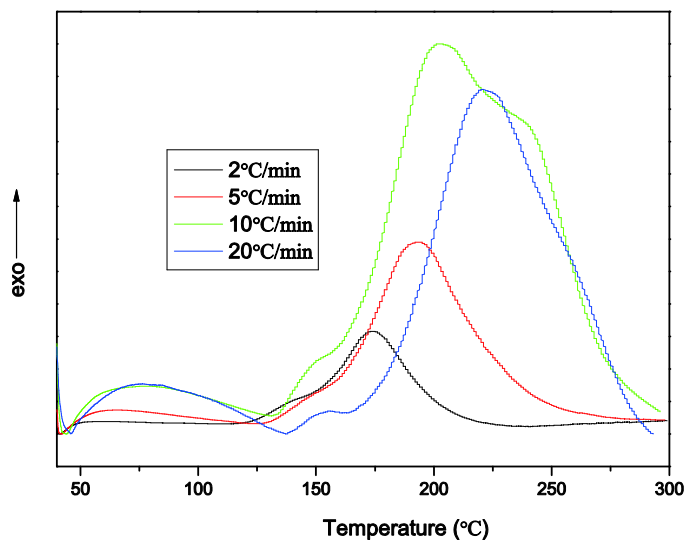


Figure 5S. DSC thermogram of Si-NCE-2 at different heating rate under N₂.

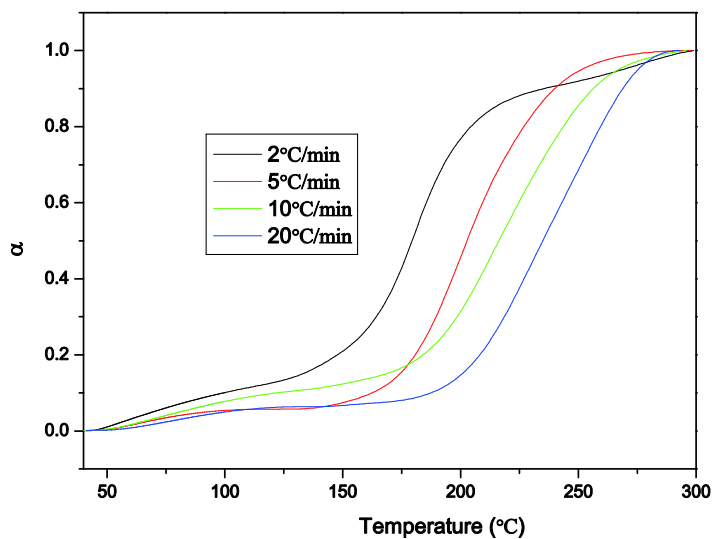


Figure 6S. The conversion of Si-NCE-1 versus temperature at different heating rate.

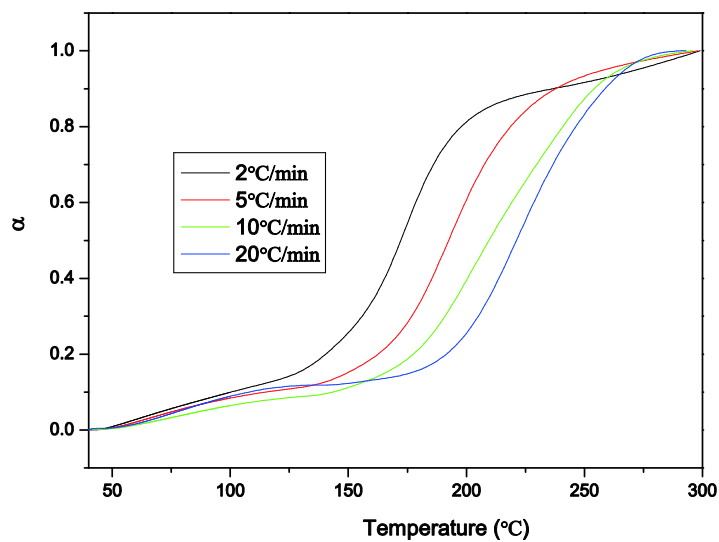


Figure 7S. The conversion of Si-NCE-2 versus temperature at different heating rate.

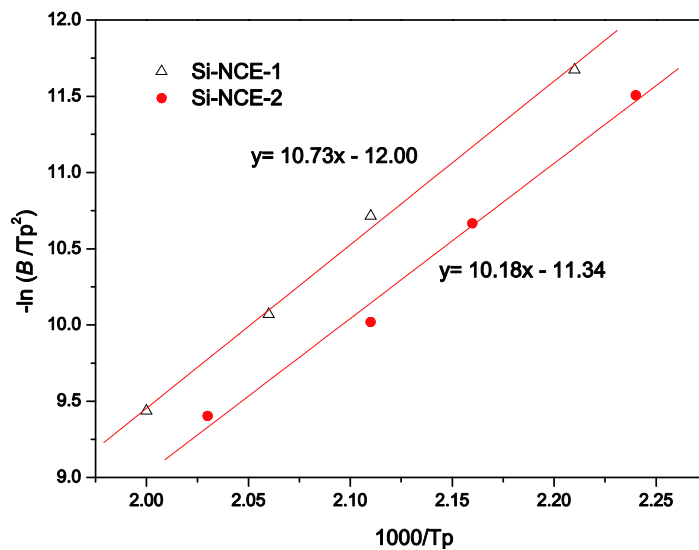


Figure 8S. Kissinger plot for determination of the activation energy of Si-NCE-1 and Si-NCE-2.

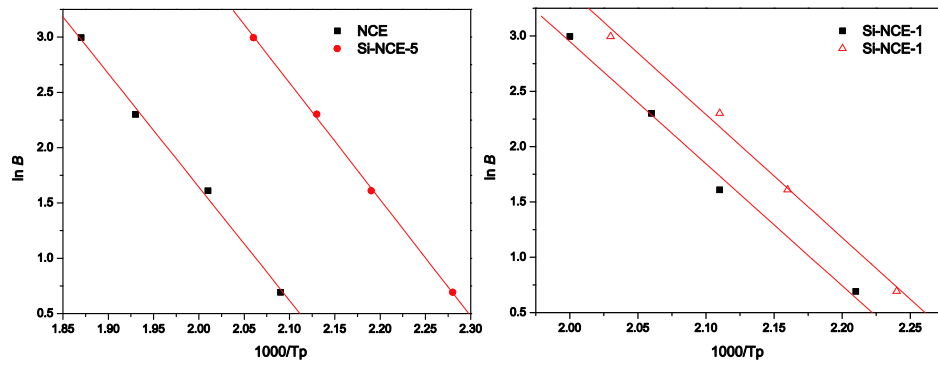


Figure 9S. Plots of $\ln B$ vs. $1000/T_p$ for hybrid NCE resins.

Table 1S. The kinetic parameter of hybrid NCE resins.

Hybrid NCE	E_a (activity energy) (KJ/mol)	n (order of curing reaction)	m (order of curing reaction)	Pre-exponential factor (A) (min^{-1})
NCE	93.83	2.42	1.01	1.45×10^9
Si-NCE-1	89.21	2.37	0.99	1.77×10^9
Si-NCE-2	84.64	2.31	0.99	8.58×10^8
Si-NCE-5	80.31	2.29	0.98	3.61×10^8

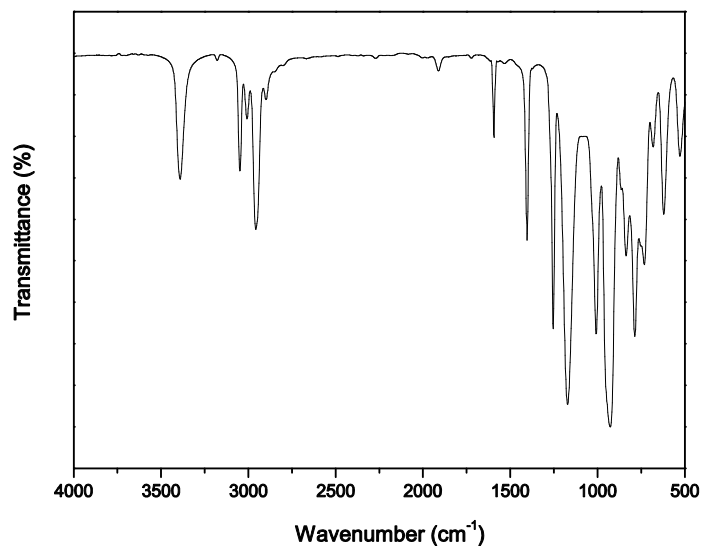


Figure 10S. FT-IR spectrum of the MVSZ.

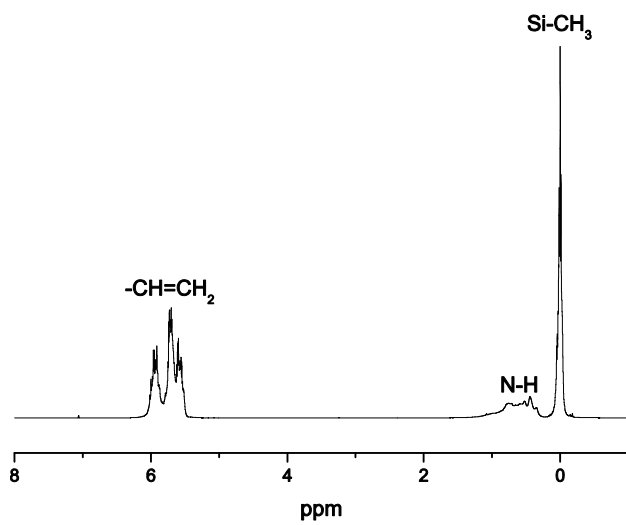


Figure 11S. ¹H-NMR spectrum of the MVSZ.