## Supporting Information

A Facile Access to Substituted Cationic 12-Azapyrene Salts by Rhodium(III)-Catalyzed C-H Annulation of N Arylpyridinium Salts<br>Boya Feng, Danyang Wan, Lipeng Yan, Vilas D. Kadam, Jingsong You and Ge Gao*<br>Key Laboratory of Green Chemistry and Technology of Ministry of Education,<br>College of Chemistry, Sichuan University, 29 Wangjiang Road, Chengdu 610064, P. R. China<br>E-mail:gg2b@scu.edu.cn

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## I. General remarks

Unless otherwise noted, all reagents were prepared from commercial suppliers and used without further purification. Alkynes ${ }^{1}$ and $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}{ }^{2}$ were prepared according to the literature procedure. DCE, MeCN, DMF, DMSO were dried by refluxing over $\mathrm{CaH}_{2}$ and freshly distilled prior to use. Toluene and 1,4-dioxane were dried by refluxing over sodium and freshly distilled prior to use. NMR spectra were recorded on a Bruker AV II-400 MHz or Agilent 400-MR DD2 spectrometer ( ${ }^{1} \mathrm{H}$ NMR at $400 \mathrm{MHz},{ }^{13} \mathrm{C}$ NMR at 100 MHz and ${ }^{19} \mathrm{~F}$ at 376 MHz ). The ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ) chemical shifts and the ${ }^{13} \mathrm{C}$ NMR ( 100 MHz ) chemical shifts were measured relative to $\mathrm{CDCl}_{3}$ or DMSO- $d_{6}$ as the internal reference $\left(\mathrm{CDCl}_{3}: \delta_{\mathrm{H}}=7.26 \mathrm{ppm}, \delta_{\mathrm{C}}=77.16 \mathrm{ppm} ; \mathrm{DMSO}-d_{6}: \delta_{\mathrm{H}}=2.50\right.$ ppm, $\delta_{\mathrm{C}}=39.52 \mathrm{ppm}$ ). High resolution mass spectra (HRMS) were recorded on a Waters-Q-TOF-Premier (ESI) or a Shimadzu LCMS-IT-TOF (ESI). UV/vis spectra were measured on a HITACHI U-2910. Fluorescence spectra were collected on a Horiba Jobin Yvon-Edison Fluoromax-4 fluorescence spectrometer with a calibrated integrating sphere system.

## II. General procedure for the synthesis of $\mathbf{N}$-arylpyridinium salts



The N -arylpyridinium salts were synthesized from a modified procedure of our previous report. ${ }^{3}$ To a 25 mL round bottom flask, a substituted pyridine ( 2 mmol ), a diaryliodonium tetrafluoroborate ( $3 \mathrm{mmol}, 1.5$ equiv), copper acetate monohydrate ( 10 $\mathrm{mol} \%, 0.2 \mathrm{mmol})$ and DMF ( 8 ml ) were added. The reaction mixture was then heated at $100{ }^{\circ} \mathrm{C}$ for 8 hours. After cooled down to room temperature, DMF was removed under vacuum. The mixture was dissolved in methanol and precipitated using ethyl ether to give the pure product suitable for analysis.

## III. Optimization of the reaction conditions

Table S1 Optimization of the reaction conditions. ${ }^{a}$

|  |  | $\xrightarrow[\text { ase, oxidant }]{\text { Catalyst }}$ |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Catalyst (5 mol \%) | Oxidant <br> (equiv) | Base (equiv) | Solvent | Yield $b$ of 3a | Yield $b$ of $4 a$ |
| 1 | $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{AgBF}_{4}(2)$ | $\mathrm{NaOAc}(2)$ | DCE | Trace | n.d. |
| 2 | $\left[\mathrm{Cp} * \mathrm{Rh}(\mathrm{MeCN})_{3}\left(\mathrm{SbF}_{6}\right)_{2}\right]$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | n.d. | 88 \% |
| 3 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{acac})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | n.d. | n.d. |
| 4 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{CuO}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | n.d. | n.d. |
| 5 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | - | DCE | n.d. | n.d. |
| 6 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | - | $\mathrm{NaOAc}(4)$ | DCE | n.d. | n.d. |
| 7 | - | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | n.d. | n.d. |
| 8 | $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DMF | n.d. | n.d. |
| 9 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DMSO | n.d. | n.d. |
| 10 | $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | Toluene | 28\% | 21\% |
| 11 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | MeCN | 24\% | 45\% |
| 12 | [ $\left.\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | Dioxane | 42\% | Trace |
| 13 | $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | KOAc(4) | DCE | 20\% | 36\% |
| 14 | $\left[\mathrm{Cp} * \mathrm{IrCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | 40\% | Trace |
| 15 | $\left[\mathrm{Cp} * \mathrm{CoCl}_{2}\right]_{2}$ | $\mathrm{Cu}(\mathrm{OAc})_{2}(4)$ | $\mathrm{NaOAc}(4)$ | DCE | n.d. | n.d. |

${ }^{a}$ Reaction conditions: $\mathbf{1 a}(0.1 \mathrm{mmol}, 1$ equiv), $\mathbf{2 a}(0.4 \mathrm{mmol}, 4$ equiv), catalyst, base, oxidants in solvent ( 2 mL ) under $\mathrm{N}_{2}$ at $140{ }^{\circ} \mathrm{C}$ for $16 \mathrm{~h} .{ }^{b}$ Isolated yields.

## IV. General procedure for the synthesis of 12-azapyrene derivatives



A flame-dried Schlenk tube equipped with a magnetic stir bar was charged with an
arylpyridinium salt $\mathbf{1}(0.1 \mathrm{mmol})$, an alkyne $\mathbf{2}(0.4 \mathrm{mmol}),\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}(3.1 \mathrm{mg}, 0.005$ $\mathrm{mmol}), \mathrm{Cu}(\mathrm{OAc})_{2}(73 \mathrm{mg}, 0.4 \mathrm{mmol})$ and $\mathrm{NaOAc}(32.8 \mathrm{mg}, 0.4 \mathrm{mmol})$ under $\mathrm{N}_{2}$. Dry DCE ( 2.0 mL ) was then added and the tube was sealed with a teflon-coated screw cap. The reaction solution was heated at $140{ }^{\circ} \mathrm{C}$ for 16 h . After cooled to ambient temperature, 4.0 mL of saturated $\mathrm{NaBF}_{4}$ (aq.) was added and the mixture was stirred at room temperature for another 0.5 h under air. The organic layer was then separated and the water layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5.0 \mathrm{~mL} \times 3)$. The combined organic phase was concentrated under vacuum and the residue was purified by column chromatography on $\mathrm{Al}_{2} \mathrm{O}_{3}$ (neutral, 200-300 mesh) with $\mathrm{MeCN} / \mathrm{CH}_{2} \mathrm{Cl}_{2}(1 / 10$ to $1 / 3)$ to provide the desired product.

## V. H/D Exchange experiments

a)



a) A flame-dried Schlenk tube equipped with a magnetic stir bar was charged with 1a ( 0.1 mmol ), $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}(3.1 \mathrm{mg}, 0.005 \mathrm{mmol}), \mathrm{Cu}(\mathrm{OAc})_{2}(73 \mathrm{mg}, 0.4 \mathrm{mmol})$ and $\mathrm{NaOAc}(32.8 \mathrm{mg}, 0.4 \mathrm{mmol})$ under $\mathrm{N}_{2}$. Dry DCE $(2.0 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(1.0 \mathrm{~mL})$ was then added and the tube was sealed with a teflon-coated screw cap. The reaction solution was heated at $140{ }^{\circ} \mathrm{C}$ for 3 h . After the mixture was cooled to room temperature, the solvent was removed by rotary evaporation. The crude residue was subjected to the ${ }^{1} \mathrm{H}$ NMR analysis. The ${ }^{1} \mathrm{H}$ NMR spectum shows the incorporation
of deuterium into the labeled protons of the substrate 1a.

b) A flame-dried Schlenk tube equipped with a magnetic stir bar was charged with 1a ( 0.1 mmol ), $\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}(3.1 \mathrm{mg}, 0.005 \mathrm{mmol})$ and $\mathrm{Cu}(\mathrm{OAc})_{2}(73 \mathrm{mg}, 0.4 \mathrm{mmol})$ under $\mathrm{N}_{2}$. Dry DCE $(2.0 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(1.0 \mathrm{~mL})$ was then added and the tube was sealed with a teflon-coated screw cap. The reaction solution was heated at $140^{\circ} \mathrm{C}$ for 3 h . After the mixture was cooled to room temperature, the solvent was removed by rotary evaporation. The crude residue was subjected to the ${ }^{1} \mathrm{H}$ NMR analysis. The ${ }^{1} \mathrm{H}$ NMR spectum shows the incorporation of deuterium into the labeled protons of the substrate $\mathbf{1 a}$.

c) A flame-dried Schlenk tube equipped with a magnetic stir bar was charged with 1a ( 0.1 mmol ), diphenylacetylene $(0.4 \mathrm{mmol}),\left[\mathrm{Cp} * \mathrm{RhCl}_{2}\right]_{2}(3.1 \mathrm{mg}, 0.005 \mathrm{mmol})$, $\mathrm{Cu}(\mathrm{OAc})_{2}(73 \mathrm{mg}, 0.4 \mathrm{mmol})$ and $\mathrm{NaOAc}(32.8 \mathrm{mg}, 0.4 \mathrm{mmol})$ under $\mathrm{N}_{2}$. Dry DCE $(2.0 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(1.0 \mathrm{~mL})$ was then added and the tube was sealed with a tefloncoated screw cap. The reaction solution was heated at $140{ }^{\circ} \mathrm{C}$ for 3 h . After the mixture was cooled to room temperature, the solvent was removed by rotary evaporation. The crude residue was subjected to the ${ }^{1} \mathrm{H}$ NMR analysis. The ${ }^{1} \mathrm{H}$ NMR spectrum shows the incorporation of deuterium into the labeled protons of the single annulated product 3a.

VI. Experimental data for the described substances


4-(tert-Butyl)-1-phenylpyridin-1-ium tetrafluoroborate (1a): A white solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=1.43(\mathrm{~s}, 9 \mathrm{H}), 7.61-7.68(\mathrm{~m}, 5 \mathrm{H}), 8.21(\mathrm{~d}, J=5.6 \mathrm{~Hz}$, $2 \mathrm{H}), 8.88(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=30.0,37.0,124.0$, 126.3, 130.9, 131.7, 142.3, 143.4, $172.8 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-151.92$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$212.1434, found 212.1431.


4-(tert-butyl)-1-phenylpyridin-1-ium trifluoromethanesulfonate (4a'): A white solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.45(\mathrm{~s}, 9 \mathrm{H}), 7.63-7.64(\mathrm{~m}, 3 \mathrm{H}), 7.72-7.74$ (m, 2H), $8.21(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.93(\mathrm{dd}, J=6.4 \mathrm{~Hz}, 1.2 \mathrm{~Hz} 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=30.1,37.1,124.1,126.3,131.0,131.8,142.2,143.6,172.9 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-78.27$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}\right]^{+}[\mathrm{M}-$ $\left.\mathrm{BF}_{4}\right]^{+}$212.1434, found 212.1430.


1-Phenylpyridin-1-ium tetrafluoroborate (1b): An off-white solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $d 6$ ): $\delta=7.70-7.80(\mathrm{~m}, 3 \mathrm{H}), 7.88-7.90(\mathrm{~m}, 2 \mathrm{H}), 8.29-8.33(\mathrm{~m}, 2 \mathrm{H})$, $8.79(\mathrm{tt}, J=8.0 \mathrm{~Hz}, 1.2 \mathrm{~Hz}, 1 \mathrm{H}), 9.34-9.35(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO$\left.d_{6}\right): \delta=124.8,128.1,130.2,131.2,142.8,145.0,146.6 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( 376 MHz , DMSO-d d $^{\prime}: \delta=-148.25$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{~N}\right]^{+}\left[\mathrm{M}_{\left.-\mathrm{BF}_{4}\right]^{+}}\right.$ 156.0808 , found 156.0808 .


4-Methoxy-1-phenylpyridinium tetrafluoroborate (1c): An off-white solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=4.19(\mathrm{~s}, 3 \mathrm{H}), 7.61-7.64(\mathrm{~m}, 7 \mathrm{H}), 8.67-8.70(\mathrm{~m}, 2 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=58.6,114.6,123.9,131.0,131.4,142.0,145.3$, $172.4 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-152.15$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{NO}\right]^{+}\left[\mathrm{M}_{\left.-\mathrm{BF}_{4}\right]^{+}}\right.$186.0913, found 186.0913.


3-Methyl-1-phenylpyridinium tetrafluoroborate (1d): A reddish-brown solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=2.71(\mathrm{~s}, 3 \mathrm{H}), 7.60-7.69(\mathrm{~m}, 5 \mathrm{H}), 8.01(\mathrm{~d}, J=6.4 \mathrm{~Hz}$, $2 \mathrm{H}), 8.77-8.79(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=22.5,124.07,124.09$, 129.7, 130.97, 130.99, 131.8, 142.4, 143.2, $161.2 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-151.93$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$170.0964, found
170.0964.


4-Cyano-1-phenylpyridin-1-ium tetrafluoroborate (1e): A light-brown solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO-d6): $\delta=7.77-7.79(\mathrm{~m}, 3 \mathrm{H}), 7.87-7.91(\mathrm{~m}, 2 \mathrm{H}), 8.90(\mathrm{~d}, J$ $=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 9.67(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{DMSO}-d 6\right): \delta=$ $114.8,124.8,127.6,130.3,131.0,131.9,142.5,146.4 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR (376 MHz, DMSO-d $d$ ): $\delta=-148.26$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$181.0760, found 181.0758 .


4-(tert-Butyl)-1-(4-methoxyphenyl)pyridin-1-ium tetrafluoroborate (1f): A yellowish solid. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.44(\mathrm{~s}, 9 \mathrm{H}), 3.86(\mathrm{~s}, 9 \mathrm{H}), 7.08(\mathrm{~d}, J$ $=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.62(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.15(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 8.81(\mathrm{~d}, J=6.4 \mathrm{~Hz}$, $2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=30.1,36.9,56.0,110.2,116.0,125.3,126.1$, 143.2, 161.9, $172.1 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-152.11$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{NO}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 242.1539$, found 242.1541 .


4-tert-Butyl-1-(4-(methoxycarbonyl)phenyl)pyridinium tetrafluoroborate (1g): A white solid. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.44(\mathrm{~s}, 9 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 7.82(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 2 \mathrm{H}), 8.20(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.25(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 8.90(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H})$
ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=29.99,37.2,52.9,124.4,126.4,132.2,133.2$, 143.3, 145.2, 165.3, 173.6 ppm. ${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-151.70$ (s) ppm. HRMS (ESI) calcd for [ $\left.\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{NO}_{2}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$270.1489, found 270.1487.


3-(tert-Butyl)-5,6-bis(4-chlorophenyl)pyrido[1,2-a]quinolin-11-ium
tetrafluoroborate (3a): Product 3a was prepared according to the general procedure as a white solid ( $39.3 \mathrm{mg}, 83 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.32$ (s, 9H), $7.16-$ $7.20(\mathrm{~m}, 4 \mathrm{H}), 7.32-7.37(\mathrm{~m}, 6 \mathrm{H}), 7.74(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.83(\mathrm{~s}, 1 \mathrm{H}), 8.15(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.38(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 9.19(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H})$, $10.41(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=29.9,36.5,118.5$, 122.7, 123.9, 126.9, 128.6, 128.9, 129.09, 129.11, 129.5, 129.8, 130.5, 130.6, 133.1, 133.8, 133.9, 134.0, 134.6, 134.9, 143.3, 147.3, 166.0 ppm. ${ }^{19}$ F NMR ( 376 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=-152.78$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 388.2060$, found 388.2051.


2-(tert-Butyl)-4,5,9,10-tetraphenylquinolizino $[3,4,5,6-i j a] q u i n o l i n-11-i u m$
tetrafluoroborate (4a): Product $\mathbf{4 a}$ was prepared according to the general procedure as a yellow-green solid ( $58.5 \mathrm{mg}, 90 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d 6$ ): $\delta=1.12$ (s, $9 \mathrm{H}), 7.34-7.36(\mathrm{~m}, 4 \mathrm{H}), 7.39-7.51(\mathrm{~m}, 16 \mathrm{H}), 7.83(\mathrm{~s}, 2 \mathrm{H}), 7.86(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H})$, 8.19 (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO-d ): $\delta=29.4,35.5,120.3$, $126.8,127.4,128.6,128.80,128.83,128.9,129.2,129.8,130.3,131.4,134.5,134.7$, 134.9, 141.9, 144.8, $159.9 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( 376 MHz , DMSO-d ): $\delta=-148.34$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{43} \mathrm{H}_{34} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 564.2686$, found 564.2682 .


2-(tert-Butyl)-4,5,9,10-tetraphenylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate (4a'): Product 4a' was prepared according to the general procedure as a yellow solid ( $48.4 \mathrm{mg}, 68 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d 6$ ): $\delta=1.12(\mathrm{~s}, 9 \mathrm{H})$, $7.32-7.36$ (m, 4H), $7.39-7.51(\mathrm{~m}, 16 \mathrm{H}), 7.83(\mathrm{~s}, 2 \mathrm{H}), 7.876$ (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.19$ (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO- $d 6$ ): $\delta=29.4,35.5,120.3,126.8$, 127.4, 128.6, 128.80, 128.83, 128.9, 129.2, 129.8, 130.3, 131.4, 134.5, 134.7, 134.9, 141.9, 144.8, 159.9 ppm . ${ }^{19} \mathrm{~F}$ NMR ( 376 MHz , DMSO- $d 6$ ): $\delta=-77.77$ (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{43} \mathrm{H}_{34} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 564.2686$, found 564.2684.


2-(tert-Butyl)-4,5,9,10-tetra-o-tolylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate (4b): Product $\mathbf{4 b}$ was prepared according to the general procedure as a green solid ( $58.1 \mathrm{mg}, 82 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d 6 ): $\delta=1.1-1.13$ (m, 9H), $2.03-2.30(\mathrm{~m}, 12 \mathrm{H}), 7.10-7.44(\mathrm{~m}, 16 \mathrm{H}), 7.65-7.67(\mathrm{~m}, 4 \mathrm{H}), 8.14-8.20(\mathrm{~m}$, 1H) ppm. ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO-d 6 ): $\delta=19.2,19.3,29.4,35.3,109.6,112.4$, 119.6, 125.8, 126.0, 126.4, 126.5, 127.0, 128.9, 129.2, 129.3, 130.46, 130.47, 130.54, $130.6,131.9,133.5,133.8,133.9,134.09,134.13,135.5,135.6,136.4,141.36,141.38$, 144.9, 145.0, 160.3, 160.4 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{42} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$620.3312, found 620.3307 .


2-tert-Butyl-5,6,10,11-tetram-tolylquinolizino[3,4,5,6-ija]quinolinium
tetrafluoroborate (4c): Product $\mathbf{4 c}$ was prepared according to the general procedure as a yellow-green solid ( $60.1 \mathrm{mg}, 85 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d 6$ ): $\delta=\delta=1.13$ (s, 9H), 2.30 ( $\mathrm{s}, 6 \mathrm{H}$ ), 2.31 ( $\mathrm{s}, 6 \mathrm{H}$ ), 7.11-7.26 (m, 12H), 7.32-7.40 (m, 4H), 7.84-7.86 (m, $4 \mathrm{H}), 8.17$ (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $\mathrm{d}_{6}$ ): $\delta=20.9,20.9$, 20.98, 21.01, 29.4, 35.5, 120.3, 126.3, 126.4, 126.8, 127.3, 127.37, 127.39, 128.61, $128.63,128.7,129.2,129.25,129.34,129.4,129.5,129.7,130.7,130.8,131.2,134.39$, 134.44, 134.66, 134.69, 134.8, 134.85, 137.91, 137.93, 138.1, 141.86, 141.88, 144.66, 144.72, 159.7 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{42} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$620.3312, found 620.3311 .


2-(tert-Butyl)-4,5,9,10-tetra-p-tolylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate (4d): Product 4d was prepared according to the general procedure as a yellow-green solid ( $61.4 \mathrm{mg}, 87 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d 6$ ): $\delta=1.12$ (s, 9H), 2.34 ( $\mathrm{s}, 6 \mathrm{H}$ ), 2.35 ( $\mathrm{s}, 6 \mathrm{H}$ ), $7.21-7.23$ (m, 4H), $7.27-7.32$ (m, 12H), $7.81-7.82$ (m, 4H), 8.13 (t, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d 6$ ): $\delta=20.90$, $20.93,29.5,35.5,120.2,126.7,127.6,129.1,129.4,129.5,129.6,130.2,131.3,132.0$, 132.1, 134.5, 137.9, 138.0, 142.1, 144.8, 159.7 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{42} \mathrm{~N}\right]^{+}$ $\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$620.3312, found 620.3308 .


2-(tert-Butyl)-4,5,9,10-tetrakis(4-(tert-butyl)phenyl)quinolizino[3,4,5,6-
$\boldsymbol{i j a}$ ]quinolin-11-ium tetrafluoroborate (4e): Product $\mathbf{4 e}$ was prepared according to the general procedure as a yellow solid ( $78.8 \mathrm{mg}, 90 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $\mathrm{d}_{6}$ ): $\delta=1.13(\mathrm{~s}, 9 \mathrm{H}), 1.27(\mathrm{~s}, 36 \mathrm{H}), 7.22(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.28(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.43$
$-7.47(\mathrm{~m}, 8 \mathrm{H}), 7.87(\mathrm{~s}, 2 \mathrm{H}), 7.92(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.20(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO- $d 6$ ): $\delta=29.4,30.92,30.93,34.41,34.44,35.4,120.1,125.3$, 125.4, 126.67, 126.68, 127.4, 129.0, 130.1, 131.3, 131.9, 132.1, 134.7, 141.9, 145.0, $150.9,151.1,159.7 \mathrm{ppm}$. HRMS (ESI) calcd for $\left[\mathrm{C}_{59} \mathrm{H}_{66} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 788.5190$, found 788.5184.


2-(tert-Butyl)-4,5,9,10-tetrakis(4-methoxyphenyl)quinolizino[3,4,5,6-ija]quinolin-11-ium tetrafluoroborate (4f): Product $\mathbf{4 f}$ was prepared according to the general procedure as a green solid ( $71 \mathrm{mg}, 92 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d 6$ ): $\delta=1.16$ (s, $9 \mathrm{H}), 3.79(\mathrm{~s}, 6 \mathrm{H}), 3.80(\mathrm{~s}, 6 \mathrm{H}), 7.02-7.06(\mathrm{~m}, 8 \mathrm{H}), 7.25(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 4 \mathrm{H}), 7.31(\mathrm{~d}$, $J=8.8 \mathrm{~Hz}, 4 \mathrm{H}), 7.87(\mathrm{~s}, 3 \mathrm{H}), 7.89(\mathrm{~s}, 1 \mathrm{H}), 8.15(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO- $d 6$ ): $\delta=29.5,35.5,55.16,55.18,114.25,114.31,120.1,126.7,126.9$, 127.1, 127.8, 129.6, 130.7, 131.3, 131.7, 134.5, 142.3, 144.9, 159.0, 159.1, 159.7 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{42} \mathrm{NO}_{4}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$684.3108, found 684.3099.


2-(tert-Butyl)-4,5,9,10-tetrakis(4-chlorophenyl)quinolizino[3,4,5,6-ija]quinolin-
11-ium tetrafluoroborate (4g): Product $\mathbf{4 g}$ was prepared according to the general procedure as a yellow solid ( $78.8 \mathrm{mg}, 90 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d ): $\delta=1.18$ (s, 9H), 7.38 (d, $J=8.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), $7.46(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.59(\mathrm{t}, J=8.0 \mathrm{~Hz}, 8 \mathrm{H}), 7.84$ (s, 2H), 7.89 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.19(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta=29.5,35.7,120.5,127.0,127.2,129.1,129.2,130.1,131.2,131.4$, 132.3, 133.3, 133.5, 133.59, 133.61, 133.8, 141.6, 144.1, 160.6 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{30} \mathrm{Cl}_{4} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 700.1127$, found 700.1134 .


2-tert-Butyl-5,6,10,11-tetrakis(4-(trifluoromethyl)phenyl)quinolizino[3,4,5,6-
$\boldsymbol{i j a}$ ]quinolinium tetrafluoroborate (4h): Product $\mathbf{4 h}$ was prepared according to the general procedure as a green solid ( $85.8 \mathrm{mg}, 93 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d 6$ ): $\delta$ $=1.16(\mathrm{~s}, 9 \mathrm{H}), 7.62(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.70(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.80(\mathrm{~s}, 2 \mathrm{H}), 7.87-$ $7.91(\mathrm{~m}, 10 \mathrm{H}), 8.22(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d 6$ ): $\delta=$ $29.35,35.71,120.67,123.89\left(\mathrm{q}, J_{C-F}=272.1 \mathrm{~Hz}\right), 123.92\left(\mathrm{q}, J_{C-F}=272.4 \mathrm{~Hz}\right), 125.90$ $\left(\mathrm{q}, J_{C-F}=3.8 \mathrm{~Hz}\right), 125.98\left(\mathrm{q}, J_{C-F}=3.5 \mathrm{~Hz}\right), 126.75,127.43,129.28\left(\mathrm{q}, J_{C-F}=32.1 \mathrm{~Hz}\right)$, $129.45\left(\mathrm{q}, J_{C-F}=32.1 \mathrm{~Hz}\right), 130.40,131.49,131.53,133.44,138.72,141.31,143.90$, $160.89 \mathrm{ppm} .{ }^{19} \mathrm{~F}$ NMR ( 376 MHz , DMSO- $d 6$ ): $\delta=-61.24(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), -148.34 (s) ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{30} \mathrm{NF}_{12}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 836.2181$, found 836.2179.


4,5,9,10-Tetrakis(4-acetylphenyl)-2-(tert-butyl)quinolizino[3,4,5,6-ija]quinolin-
11-ium tetrafluoroborate (4i): Product $4 \mathbf{i}$ was prepared according to the general procedure as a yellow solid ( $68.8 \mathrm{mg}, 84 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d 6$ ): $\delta=1.13$ (s, 9H), 2.60 (s, 6H), $2.61(\mathrm{~s}, 6 \mathrm{H}), 7.55(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.62(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H})$, 7.79 (s, 2H), 7.82 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.05-8.08(\mathrm{~m}, 8 \mathrm{H}), 8.18(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm}$. ${ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO- $d 6$ ): $\delta=26.85,26.88,29.4,35.7,120.6,126.9,127.2$, 128.8, 128.9, 129.8, 130.2, 130.9, 131.4, 133.6, 136.7, 136.8, 139.0, 139.2, 141.4, 144.1, 160.6, 197.5, 197.6 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{51} \mathrm{H}_{42} \mathrm{NO}_{4}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$ 732.3108 , found 732.3093.


2-(tert-Butyl)-4,5,9,10-tetrakis(3-cyanophenyl)quinolizino[3,4,5,6-ija]quinolin-11ium tetrafluoroborate $\mathbf{( 4 j})$ : Product $\mathbf{4 j}$ was prepared according to the general procedure as a yellow solid ( $61.6 \mathrm{mg}, 82 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d 6$ ): $\delta=1.19$ $(\mathrm{s}, 9 \mathrm{H}), 7.61-7.82(\mathrm{~m}, 11 \mathrm{H}), 7.91-8.06(\mathrm{~m}, 9 \mathrm{H}), 8.22(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO- $d 6$ ): $\delta=29.4,35.9,112.06,112.14,118.2,121.0,126.7,127.7$, $130.4,130.47,130.53,131.5,132.9,133.1,133.2,133.95,133.98,134.2,135.1,135.2$, $135.3,135.4,135.6,141.2,143.6 \mathrm{ppm} . \mathrm{HRMS}(\mathrm{ESI})$ calcd for $\left[\mathrm{C}_{47} \mathrm{H}_{30} \mathrm{~N}_{5}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$ 664.2496, found 664.2499.


## 2-(tert-Butyl)-4,5,9,10-tetra(naphthalen-1-yl)quinolizino[3,4,5,6-ija]quinolin-11-

 ium tetrafluoroborate $\mathbf{( 4 k ) : ~ P r o d u c t ~} \mathbf{4 k}$ was prepared according to the general procedure as a yellow solid $(63.8 \mathrm{mg}, 75 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) : $\delta=0.68$ $(\mathrm{s}, 9 \mathrm{H}), 7.14-7.30(\mathrm{~m}, 8 \mathrm{H}), 7.50-7.74(\mathrm{~m}, 11 \mathrm{H}), 7.80-8.02(\mathrm{~m}, 14 \mathrm{H}) \mathrm{ppm} . \mathrm{HRMS}$ (ESI) calcd for $\left[\mathrm{C}_{59} \mathrm{H}_{42} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 764.3312$, found 764.3302.

2-(tert-butyl)-4,5,9,10-tetrapropylquinolizino[3,4,5,6-ija]quinolin-11-ium
Product 41 was prepared according to the general procedure as a yellow solid ( 39.6 mg , $77 \%) .{ }^{1} \mathrm{H}$ NMR (400 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=1.236-1.244(\mathrm{~m}, 12 \mathrm{H}), 1.62(\mathrm{~s}, 9 \mathrm{H}), 1.78-$
$1.84(\mathrm{~m}, 8 \mathrm{H}), 3.23-3.29(\mathrm{~m}, 8 \mathrm{H}), 8.31(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.43-8.47(\mathrm{~m}, 4 \mathrm{H})$, ppm. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=14.7,14.8,22.7,23.0,30.6,31.8,32.0,36.5,117.8$, 124.3, 126.8, 129.9, 130.4, 132.1, 140.9, 144.6, 160.8 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{31} \mathrm{H}_{42} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 428.3312$, found 428.3310 .


4,5,9,10-Tetraphenylquinolizino[3,4,5,6-ija]quinolin-11-ium tetrafluoroborate (4m): Product $\mathbf{4 m}$ was prepared according to the general procedure as a yellow solid ( $48.2 \mathrm{mg}, 81 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, ~ D M S O-d 6$ ): $\delta=7.32-7.34$ (m, 4H), $7.39-7.46$ $(\mathrm{m}, 16 \mathrm{H}), 7.88(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.02(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.21(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $8.55(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d 6$ ): $\delta=124.0,126.6,127.7$, 128.6, 128.7, 128.8, 128.9, 129.2, 129.9, 130.3, 131.3, 131.8, 134.8, 134.9, 137.7, 141.9, 144.8 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{43} \mathrm{H}_{34} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$564.2686, found 564.2682.


2-Methoxy-4,5,9,10-tetraphenylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate ( $\mathbf{4 n}$ ): Product $\mathbf{4 n}$ was prepared according to the general procedure as a grey solid ( $48.1 \mathrm{mg}, 77 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d 6$ ): $\delta=3.34-4.32(\mathrm{~m}$, 3H), $7.21-7.27$ (m, 3H), 7.32 - 7.33 (m, 4H), $7.38-7.45$ (m, 15H), 7.77 (d, $J=8.0$ $\mathrm{Hz}, 2 \mathrm{H}), 8.09(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO- $d 6$ ): $\delta=56.9,109.0$, 126.5, 127.17, 127.18, 128.5, 128.7, 128.8, 128.9, 129.2, 130.3, 131.2, 133.8, 134.6, 134.9, 144.4, 144.5, 163.6 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{40} \mathrm{H}_{28} \mathrm{NO}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$ 548.2165, found 548.2169.


2-(tert-Butyl)-7-methoxy-4,5,9,10-tetraphenylquinolizino[3,4,5,6-ija]quinolin-11ium tetrafluoroborate (40): Product 40 was prepared according to the general procedure as a yellow solid ( $50.4 \mathrm{mg}, 74 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.17$ (s, $9 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 7.28-7.37(\mathrm{~m}, 22 \mathrm{H}), 7.93(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=30.1,35.9,55.9,112.6,121.7,128.65,128.74,128.9,129.0,129.6,130.0$, $130.7,131.8,135.2,135.3,135.5,141.3,144.9,159.2,159.5 \mathrm{ppm}$. HRMS (ESI) calcd for $\left[\mathrm{C}_{44} \mathrm{H}_{36} \mathrm{NO}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$594.2791, found 594.2794.


1-Methyl-4,5,9,10-tetraphenylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate (4p): Product $\mathbf{4 p}$ was prepared according to the general procedure as a green solid $(36.6 \mathrm{mg}, 70 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d 6$ ): $\delta=2.56(\mathrm{~s}, 3 \mathrm{H})$, $7.30-7.47(\mathrm{~m}, 20 \mathrm{H}), 7.76(\mathrm{~s}, 2 \mathrm{H}), 7.82(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.16(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO-d 6 ): $\delta=21.6,124.1,126.7,127.3,128.5,128.69$, $128.72,128.8,129.2,129.6,130.4,131.5,134.3,134.7,135.0,141.5,144.8,149.7 \mathrm{ppm}$. HRMS (ESI) calcd for $\left[\mathrm{C}_{44} \mathrm{H}_{36} \mathrm{~N}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+}$522.2216, found 522.2208.


2-Cyano-4,5,9,10-tetraphenylquinolizino[3,4,5,6-ija]quinolin-11-ium
tetrafluoroborate $(\mathbf{4 q})$ : Product $\mathbf{4 q}$ was prepared according to the general procedure as a reddish-brown solid (37.9 mg, $61 \%) .{ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO-d 6$): \delta=7.32-$
7.33 (m, 4H), 7.37 - 7.46 (m, 16H), 8.02 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 8.33-8.39 (m, 3H) ppm. ${ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO- $d 6$ ): $\delta=115.4,118.9,125.5,127.6,128.0,128.8,128.98$, 129.03, 129.1, 130.5, 131.2, 131.84, 133.84, 134.5, 135.0, 142.2, 146.8 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{40} \mathrm{H}_{25} \mathrm{~N}_{2}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 533.2012$, found 533.2007.

## 2-(tert-Butyl)-7-(methoxycarbonyl)-4,5,9,10-tetra-o-tolylquinolizino[3,4,5,6-

 $\boldsymbol{i j a}$ ]quinolin-11-ium tetrafluoroborate (4r): Product $\mathbf{4 r}$ was prepared according to the general procedure as a greenwish solid ( $49.9 \mathrm{mg}, 64 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO$d 6): \delta=1.13(\mathrm{~s}, 9 \mathrm{H}), 2.03-2.31(\mathrm{~m}, 12 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 7.26-7.43(\mathrm{~m}, 16 \mathrm{H}), 7.72(\mathrm{~s}$, 2 H ), 8.13 (s, 2H) ppm. ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d 6$ ): $\delta=19.2,19.3,29.4,35.5$, $53.2,109.6,112.4,120.1,125.8,126.0,126.1,127.5,129.2,129.4,130.6,130.7,131.8$, 133.2, 133.4, 135.1, 135.47, 135.51, 136.3, 141.8, 144.9, 164.6, 192.5 ppm. HRMS (ESI) calcd for $\left[\mathrm{C}_{49} \mathrm{H}_{44} \mathrm{NO}_{2}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}\right]^{+} 678.3367$, found 678.3360 .

2-(tert-Butyl)-4,5,9,10-tetrakis(4-(diphenylamino)phenyl)quinolizino[3,4,5,6-
ija]quinolin-11-ium tetrafluoroborate (4s): Product 4s was prepared according to the general procedure as a crimson solid ( $121.2 \mathrm{mg}, 92 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d_{6}$ ) $\delta=1.28(\mathrm{~s}, 9 \mathrm{H}), 7.05-7.12(\mathrm{~m}, 32 \mathrm{H}), 7.21-7.25(\mathrm{~m}, 8 \mathrm{H}), 7.32-7.36(\mathrm{~m}, 16 \mathrm{H}), 8.07$ (s, 2H), 8.17 (d, $J=8 \mathrm{~Hz}, 2 \mathrm{H}), 8.29(\mathrm{t}, J=8 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO$\left.d_{6}\right): \delta=29.5,35.6,120.16,122.58,123.22,123.23,123.5,123.6,124.0,124.2,126.9$, $127.3,128.6,128.8,129.69,129.70,130.8,131.5,131.8,134.5,141.9,144.8,146.8$, 146.9, 147.35, 147.38, 159.85 ppm . HRMS (ESI) calcd for $\left[\mathrm{C}_{91} \mathrm{H}_{71} \mathrm{~N}_{5}\right]^{+}\left[\mathrm{M}-\mathrm{BF}_{4}+\mathrm{H}\right]^{+}$ 1233.5704, found 1233.5708 .
VII. Photophysical spectra of 12-azapyrene derivatives in $\mathbf{C H}_{\mathbf{2}} \mathbf{C l}_{\mathbf{2}}$

| Compound | $\lambda_{\text {max }}$ | $\lambda_{\text {ex }}$ | $\lambda_{\text {em }}$ | $\Phi_{\mathrm{F}}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{4 a}$ | 354 | 356 | 468 | 0.76 |
| $\mathbf{4 b}$ | 353 | 354 | 458 | 0.79 |
| $\mathbf{4 c}$ | 355 | 356 | 472 | 0.75 |
| $\mathbf{4 d}$ | 356 | 356 | 475 | 0.62 |
| $\mathbf{4 e}$ | 358 | 359 | 476 | 0.91 |
| $\mathbf{4 f}$ | 357 | 353 | 489 | 0.78 |
| $\mathbf{4 g}$ | 354 | 351 | 471 | 0.61 |
| $\mathbf{4 h}$ | 351 | 352 | 464 | 0.28 |
| $\mathbf{4 i}$ | 353 | 355 | 469 | 0.68 |
| $\mathbf{4 j}$ | 351 | 352 | 463 | 0.68 |
| $\mathbf{4 k}$ | 352 | 353 | 467 | 0.59 |
| $\mathbf{4 l}$ | 354 | 351 | 451 | 0.48 |
| $\mathbf{4 m}$ | 355 | 355 | 481 | 0.58 |
| $\mathbf{4 n}$ | 355 | 355 | 452 | 0.66 |
| $\mathbf{4 o}$ | 316 | 362 | 460 | 0.59 |
| $\mathbf{4 p}$ | 355 | 355 | 468 | 0.70 |
| $\mathbf{4 q}$ | 365 | 360 | 536 | 0.36 |
| $\mathbf{4 r}$ | 354 | 355 | 460 | 0.77 |
| $\mathbf{4 s}$ | 438 | 489 | 614 | 0.26 |

Table S2 Photophysical data of 12-azapyrene derivatives in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$



















## VIII. Single crystal X-Ray structure



Figure S1 X-Ray single crystal structure of 4s

## XI. References

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(2) C. White, A. Yates, and P. M. Maitlis, Inorg. Synth., 1992, 29, 228.
(3) T. Lv, Z. Wang, J. You, J. Lan, and G Gao, J. Org. Chem., 2013, 78, 5723
IX. Copies of NMR Spectra for Compounds
${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 a}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 a}$ :


$\stackrel{\stackrel{\circ}{\text { ® }}}{\stackrel{-}{1}}$
 $\stackrel{\circ}{1}$
$\begin{array}{ll}8 & \vdots \\ \text { ® } \\ \text { ® } \\ \text { i } \\ \text { i }\end{array}$

${ }^{19}$ F NMR spectra of $\mathbf{1 a}$ :


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 a}{ }^{\prime}$ :



${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 a}$ ':
sos)

$\stackrel{\square}{7}$


$\begin{array}{llllllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 & -10\end{array}$
${ }^{19}$ F NMR spectra of $\mathbf{1 a}$ ':


S92 8L-

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 b}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 b}$ :



$\left.\begin{array}{lllllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ \mathrm{fl}(\mathrm{ppm})\end{array}\right)$
${ }^{19}$ F NMR spectra of $\mathbf{1 b}$ :


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 c}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 c}$ :

${ }^{19} \mathrm{~F}$ NMR spectra of $\mathbf{1 c}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 d}$ :

${ }^{13} \mathrm{C}$ NMR spectra of 1d:

$\begin{array}{llllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ \mathrm{fl}(\mathrm{ppm})\end{array}$
${ }^{19}$ F NMR spectra of $\mathbf{1 d}$ :


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 e}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 e}$ :


${ }^{19} \mathrm{~F}$ NMR spectra of $\mathbf{1 e}$ :

$+48.261$
$\square$

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 f}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 f}$ :
$\underset{\substack{0 \\ 0}}{\text { 号 }}$

77.16
-56.02

-36.94
-30.05
${ }^{19}$ F NMR spectra of $\mathbf{1 f}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{1 g}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 g}$ :

${ }^{19}$ F NMR spectra of $\mathbf{1 g}$ :

$\qquad$

${ }^{1} \mathrm{H}$ NMR spectra of 3a：

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{3 a}$ ：

${ }^{19}$ F NMR spectra of $\mathbf{3 a}$ :



${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 a}$ :

${ }^{13} \mathrm{C}$ NMR spectra of 4 a :

${ }^{19} \mathrm{~F}$ NMR spectra of $\mathbf{4 a}$ :

$\qquad$

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 a}$ ':

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 a}$ ':

${ }^{19}$ F NMR spectra of $\mathbf{4} \mathbf{a}^{\prime}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 b}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 b}$ :


| 230 | 220 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | $\begin{gathered} 110 \\ \mathrm{f} 1(\mathrm{ppm} \end{gathered}$ | $100$ | 90 | 8 |  | 70 | 60 | 50 |  | 40 | 30 | 20 | 1 | 10 | 0 | -1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 c}$ :



${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 c}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 d}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 d}$ :


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${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 e}$ :




${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 e}$ :


${ }^{1} H$ NMR spectra of $\mathbf{4 f}$ :





${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 f}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 g}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 g}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 h}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 h}$ :


${ }^{19} \mathrm{~F}$ NMR spectra of $\mathbf{4 h}$ :

$$
\begin{aligned}
& \infty \\
& \underset{\sim}{\infty} \stackrel{\infty}{\sim} \\
& \stackrel{\sim}{\oplus} \\
& \vdots
\end{aligned}
$$


${ }^{1} \mathrm{H}$ NMR spectra of $4 \mathbf{i}$ ：

${ }^{13} \mathrm{C}$ NMR spectra of 4 i ：

| 合志 | J | Nす |
| :---: | :---: | :---: |
| ぶふ | 8 | ¢ |
| V | । | ¢0， |





[^0]${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 j}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 j}$ :


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 k}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 k}$ :


${ }^{1} \mathrm{H}$ NMR spectra of 41:

${ }^{13} \mathrm{C}$ NMR spectra of 41:





${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 m}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 m}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 n}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 n}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 0}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 0}$ :


$n$
$n$
$i$
$i$

$\begin{array}{cc}\underset{m}{\infty} \\ \cdots & \stackrel{m}{\infty} \\ 1 & 1\end{array}$


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 p}$ :



$\qquad$

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 p}$ :
(

[^1]${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 q}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 q}$ :

${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 r}$ :

${ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{4 r}$ :


${ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{4 s}$ :

${ }^{13} \mathrm{C}$ NMR spectra of 4 s :




[^0]:    

[^1]:    

