

**Table S1-** The calculated activation enthalpies ( $\Delta H^\#/\text{kcalmol}^{-1}$ ), reaction enthalpies ( $\Delta H_r/\text{kcalmol}^{-1}$ ) and entropy ( $\Delta S/\text{calmol}^{-1}\text{k}^{-1}$ ) of the HDA reactions between 1,2-diaza-1,3-butadiene and 1,2-oxaza-1,3-butadienes derivatives (**3** and **4**) with some olefins (**5**, **10** and **15**) at the B3LYP/cc-pVDZ level of theory.

Species	TSs	$\Delta H^\#$	$\Delta S^\#$	$\Delta H_r$	$\Delta S_r$
<b>3a+5a→6aa</b>	<b>TS1aa</b>	10.60	-49.81	-37.65	-57.21
<b>3a+5a→7aa</b>	<b>TS2aa</b>	18.95	-50.09	-42.48	-56.01
<b>3a+5a→8aa</b>	<b>TS3aa</b>	15.75	-47.96	-42.98	-58.62
<b>3a+5a→9aa</b>	<b>TS4aa</b>	16.82	-50.78	-42.35	-55.20
<b>3b+5a→6ba</b>	<b>TS1ba</b>	10.29	-40.98	-44.54	-45.01
<b>3b+5a→7ba</b>	<b>TS2ba</b>	11.29	-43.87	-44.50	-47.02
<b>3b+5a→8ba</b>	<b>TS3ba</b>	13.62	-45.42	-44.99	-48.91
<b>3b+5a→9ba</b>	<b>TS4ba</b>	12.61	-49.55	-44.49	-46.35
<b>3a+5b→6ab</b>	<b>TS1ab</b>	12.29	-51.76	-32.63	-58.97
<b>3a+5b→7ab</b>	<b>TS2ab</b>	16.94	-53.25	-37.96	-57.24
<b>3a+5b→8ab</b>	<b>TS3ab</b>	10.85	-53.05	-32.86	-59.34
<b>3a+5b→9ab</b>	<b>TS4ab</b>	15.93	-53.51	-37.45	-57.40
<b>3b+5b→6bb</b>	<b>TS1bb</b>	12.12	-44.13	-40.32	-46.37
<b>3b+5b→7bb</b>	<b>TS2bb</b>	13.93	-45.77	-40.31	-48.98
<b>3b+5b→8bb</b>	<b>TS3bb</b>	10.52	-43.42	-40.18	-49.56
<b>3b+5b→9bb</b>	<b>TS4bb</b>	13.04	-44.87	-39.74	-49.89
<b>3a+10→11a</b>	<b>TS5a</b>	19.84	-54.10	-36.77	-58.19
<b>3a+10→12a</b>	<b>TS6a</b>	14.67	-53.52	-38.40	-59.25
<b>3a+10→13a</b>	<b>TS7a</b>	19.57	-54.78	-34.64	-63.58
<b>3a+10→14a</b>	<b>TS8a</b>	23.84	-55.52	-38.59	-59.84
<b>3b+10→11b</b>	<b>TS5b</b>	11.88	-46.30	-43.79	-50.61
<b>3b+10→12b</b>	<b>TS6b</b>	12.72	-49.94	-40.47	-50.88
<b>3b+10→13b</b>	<b>TS7b</b>	13.48	-44.61	-40.85	-52.36
<b>3b+10→14b</b>	<b>TS8b</b>	16.50	-48.37	-43.99	-50.06
<b>4a+15→16a</b>	<b>TS9a</b>	6.83	-58.77	-41.32	-61.82
<b>4a+15→17a</b>	<b>TS10a</b>	7.73	-57.69	-42.78	-62.46
<b>4a+15→18a</b>	<b>TS11a</b>	17.14	-59.82	-36.07	-62.17
<b>4a+15→19a</b>	<b>TS12a</b>	22.18	-61.38	-32.03	-61.71
<b>4b+15→16b</b>	<b>TS9b</b>	7.04	-56.30	-42.74	-58.70
<b>4b+15→17b</b>	<b>TS10b</b>	5.27	-53.07	-46.52	-57.35
<b>4b+15→18b</b>	<b>TS11b</b>	21.09	-56.53	-38.09	-58.19
<b>4b+15→19b</b>	<b>TS12b</b>	19.86	-58.52	-37.69	-60.88
<b>4c+15→16c</b>	<b>TS9c</b>	10.40	-53.58	-43.88	-56.27
<b>4c+15→17c</b>	<b>TS10c</b>	5.95	-52.42	-45.28	-58.86
<b>4c+15→18c</b>	<b>TS11c</b>	18.62	-52.36	-34.65	-56.18
<b>4c+15→19c</b>	<b>TS12c</b>	21.74	-52.77	-36.61	-58.43

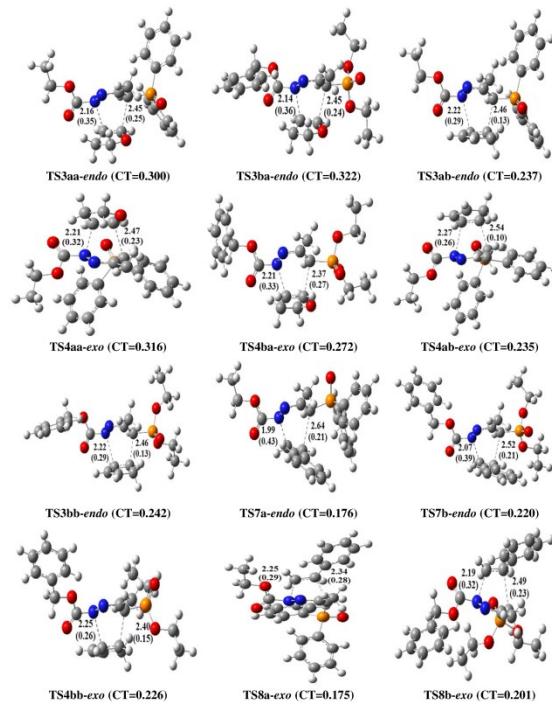
**Table S2-** Calculated activation energies ( $\Delta E^\#/ \text{kcalmol}^{-1}$ ) and reaction energies ( $\Delta E_r/\text{kcalmol}^{-1}$ ) of *ortho* pathways of the HDA reactions between 1,2-diaza-1,3-butadiene and 1,2-oxaza-1,3-butadienes derivatives (**3** and **4**) with some olefins (**5**, **10** and **15**) at the MPWB1K level of theory.

Species	TSs	$\Delta E^\#$	$\Delta E_r$
<b>3a+5a→6aa</b>	<b>TS1aa</b>	6.67	-60.27
<b>3a+5a→7aa</b>	<b>TS2aa</b>	10.82	-64.44
<b>3b+5a→8ba</b>	<b>TS1ba</b>	6.13	-67.01
<b>3b+5a→9ba</b>	<b>TS2ba</b>	7.85	-67.18
<b>3a+5b→6ab</b>	<b>TS1ab</b>	7.66	-53.61
<b>3a+5b→7ab</b>	<b>TS2ab</b>	9.68	-58.78
<b>3b+5b→8bb</b>	<b>TS1bb</b>	7.21	-61.68
<b>3b+5b→9bb</b>	<b>TS2bb</b>	10.46	-61.80
<b>3a+10→11a</b>	<b>TS5a</b>	12.34	-57.73
<b>3a+10→12a</b>	<b>TS6a</b>	7.29	-61.36
<b>3b+10→11b</b>	<b>TS5b</b>	7.27	-62.11
<b>3b+10→12b</b>	<b>TS6b</b>	7.05	-66.66
<b>4a+15→16a</b>	<b>TS9a</b>	2.98	-59.38
<b>4a+15→17a</b>	<b>TS10a</b>	5.52	-60.89
<b>4b+15→16b</b>	<b>TS9b</b>	4.52	-59.66
<b>4b+15→17b</b>	<b>TS10b</b>	0.97	-63.94
<b>4c+15→16c</b>	<b>TS9c</b>	7.62	-60.63
<b>4c+15→17c</b>	<b>TS10c</b>	6.21	-63.09

**Table S3-** Calculated activation energies ( $\Delta E^\#/ \text{kcalmol}^{-1}$ ) and reaction energies ( $\Delta E_r/\text{kcalmol}^{-1}$ ) of *meta* pathways of the HDA reactions 1,2-diaza-1,3-butadiene (**3**) and 1,2-oxaza-1,3-butadienes derivatives (**4**) with some olefins (**5**, **10** and **15**) at the MPWB1K level of theory.

Species	TS	$\Delta E^\#$	$\Delta E_r$
<b>3a+5a→8aa</b>	<b>TS3aa</b>	12.42	-64.62
<b>3a+5a→9aa</b>	<b>TS4aa</b>	13.56	-63.90
<b>3b+5a→8ba</b>	<b>TS3ba</b>	9.61	-66.92
<b>3b+5a→9ba</b>	<b>TS4ba</b>	9.58	-66.35
<b>3a+5b→8ab</b>	<b>TS3ab</b>	10.05	-54.25
<b>3a+5b→9ab</b>	<b>TS4ab</b>	11.58	-57.53
<b>3b+5b→8bb</b>	<b>TS3bb</b>	9.69	-61.72
<b>3b+5b→9bb</b>	<b>TS4bb</b>	12.92	-61.53
<b>3a+10→13a</b>	<b>TS7a</b>	15.65	-59.09
<b>3a+10→14a</b>	<b>TS8a</b>	17.85	-62.02
<b>3b+10→13b</b>	<b>TS7b</b>	8.58	-64.75
<b>3b+10→14b</b>	<b>TS8b</b>	10.35	-66.95
<b>4a+15→18a</b>	<b>TS11a</b>	13.99	-53.61
<b>4a+15→19a</b>	<b>TS12a</b>	19.58	-50.46
<b>4b+15→18b</b>	<b>TS11b</b>	19.69	-54.67
<b>4b+15→19b</b>	<b>TS12b</b>	17.39	-54.88
<b>4c+15→18c</b>	<b>TS11c</b>	17.41	-51.24
<b>4c+15→19c</b>	<b>TS12c</b>	23.16	-53.61

**Figure S1-** The geometrical optimized of transition states for *meta* pathways of the HDA reactions between 1,2-diaza-1,3-butadiene derivatives (**3a** and **3**) and some olefines (**5a** and **5**) at the B3LYP/cc-pVDZ level of theory. Bond distances are given in Å, wiberg bond indices are given in parenthesis and the natural charges (CT) of TSs are also given.



**Figure S2-** The geometrical optimized of transition states for *meta* pathways of the HDA reactions between 1,2-oxaza-1,3-butadiene derivatives (**4a** and **4**) and ethoxyethen (**15**) at the B3LYP/cc-pVDZ level of theory. Bond distances are given in Å, wiberg bond indices are given in parenthesis and the natural charges (CT) of TSs are also given.

