Electronic Supplementary Information

Surface and interface interplay on the oxidizing temperature of iron oxide and Au-Iron oxide core-shell nanoparticles

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(a) X-ray diffraction for the fresh and one year old sample of iron oxide nanoparticles:



Figure S1 X-ray diffraction of the fresh and one year old sample of iron oxide nanoparticles, (a) IO 125, (b) IO150, (c) IO200, and IO250.



(b) Small angle x-ray scattering curves for the fresh iron oxide nanoparticles:

Figure S2. (*a*) Double-logarithmic representation of colloid scattering curves. (b) fitted Gaussian size distribution for iron oxide nanoparticles.

The figure S2 (a) shows the double logarithmic representation of small angle x-ray scattering curves for samples IO125, IO150, IO200 and IO250. Due to the diversity in the nanoparticle size, dispersity and possible cluster effects, one can see differences among the scattering profiles. i.e., at low *q* region all scattering patterns reveals different, but a well-marked power law behaviors instead of the typical Guinier behavior of a single particle systems, which is a distinctive signature of systems composed by polydisperse and agglomerate particles. At the intermediate *q* region ($0.5 - 1 \text{ nm}^{-1}$), the scattering profile of sample IO200 presents a finger form, which is typical of nanoparticle systems with finite-size distribution [1]. Such feature is not observed for the others IO samples, thereby indicating that the polydispersity and the effects associated to the scattering interference between the neighboring particles are more

relevant in samples IO125, IO150 and IO250. For q values larger than 1.5 nm⁻¹, the scattering profile behavior is masked by an incoherent background [2].

In order to take into account the single or core-shell (revealed by TEM for IO125) nanoparticle aggregation, each SAXS experimental pattern was fitted assuming that the scattered intensity *I* (q) is composed by one or two different scattering contributions (depending on the sample). In this sense, the intensity *I*(q) from a collection of particles can be described by:

 $I(q) = I_1(q) + I_2(q) + bkg,$

where $I_1(q)$ and $I_2(q)$ are the scattering contributions, which each one can be rewritten as the product of a form factor (P(q)) and a structure factor (S(q)). An invariant q term was added in order to consider the incoherent background (*bkg*). Because of the complexity and differences of the multiphase systems under study, we have chosen deferments form factors and structure factors to fit the experimental SAXS data (see Table 1).

Sample	<i>l</i> ₁ (<i>q</i>)	I ₂ (q)		
	<i>P</i> ₁ (<i>q</i>)	<i>S</i> ₁ (<i>q</i>)	P ₂ (q)	S ₂ (q)
IO125	$\int_{0}^{\infty} [K(q,r + \Delta R, \Delta \eta_{1}) - K(q,r, \Delta \eta_{2} - \Delta \eta_{1})]^{2} f(r) dr$ Core-shell form factor	$D_{\rm e}\Gamma(D_{\rm e}=1)\sin\left[(D_{\rm e}=1)\tan^{-1}(qE)\right]$	$\int_{0}^{\infty} K^{2}(q,r,\Delta\eta)f(r)dr$ polydisperse spherical particles	$\frac{1}{1+24f_p\frac{G(f_p,qr)}{qr}}$ hard sphere structure factor
IO150		$1 + \frac{v_{F} (q_{F})^{D_{F}} (1 + 1/(q_{F})^{2})^{(D_{F}-1)/2}}{(q_{F})^{D_{F}} [1 + 1/(q_{F})^{2}]^{(D_{F}-1)/2}}$		
	$\int\limits_{-\infty}^{\infty}K^{2}(q,r,\Delta\eta)f(r)dr$	factor	$\int_{0}^{\infty} K^{2}(q,r,\Delta\eta)f(r)dr$	$\frac{1}{1+24f_p\frac{G(f_p,qr)}{qr}}$
10200	Polydisperse spherical particles		polydisperse spherical particles	hard sphere structure factor
10250				

Table 1 contains the form and structure factors used to fit each SAXS data

For samples IO150, IO200 and IO250, the scattering of diluted polydisperse spherical particles is

given by $K = \frac{4}{3}\pi r^3 \Delta \eta 3 \frac{\sin (qr) - qr\cos (qr)}{(qr)^3}$, being $\Delta \eta$ is the scattering length density difference between nanoparticles and the solvent. For sample, *IO125* was chosen the core-shell form factor for polydisperse spherically symmetric particles with mean radius size *R* and shell thickness ΔR (according with TEM information), in which $\Delta \eta_1$ and $\Delta \eta_2$ are the scattering length density difference between the core or shell and the solvent, respectively. As can be noted, for the first scattering contribution (*I*₁ (*q*)) was chosen the same structure factor (*S*₁ (*q*)), which was postulated by Chen and Teixeira in the framework of the fractal aggregate model. Here,

 $\Gamma(D_F - 1)$ is the gamma function and D_F is the fractal dimension. The parameter ξ is the finite cluster size that appears in the $h(r_0,\xi)$ cut-off function. Such function describes the perimeter of the aggregate (in our case, $h(r_0,\xi) = \exp[-r_0/\xi]$). For samples IO150 and IO250 a well fit was achieved by using only one scattering contribution ($I_1(q)$) with the form and structure factors already mentioned, while for samples IO125 and IO200 was necessary to use one more scattering contribution ($I_2(q)$). In these two samples, we use the same form factor (polydisperse spherical particles) than in first scattering contribution, but this time we used a hard sphere structure factor ($S_2(q)$) to consider the interparticle interference effects [3]. In this, f_P is the local volume fraction of the spherical particles within the clusters, which is related to the probability of finding particles in the vicinity of each other. In all cases, it was considered a Gaussian radii

 $f(r) = \frac{1}{\sigma\sqrt{2\pi}}e^{-(r-R)^2/2\sigma^2}$, being *R* the mean particle radius and σ the standard deviation. Obtained fitting parameters are summarized in Table 2.

Table 2 SAXS fitted parameters. *D*, ΔR , ξ , D_F , and f_P (defined in the text) obtained from the curve fitting according to form and structure factors registered on Table 1.

Sample	<i>D</i> (nm)	ΔR	<i>ξ</i> (nm)	D _F	f _P
IO125	29	1.3	53	2.93	0.27
IO150	20		108	2.90	
10200	8		20	1.30	0.44
IO250	6		20	1.92	

The primary particle size distribution is shown in Figure S2(b) and the corresponding values are reported in table 2. It is observed that the average particle diameters are 29 nm (core), 20 nm 8.0 nm and 6.0 nm for samples IO125, IO150, IO200 and IO250, respectively. These values are consistent with the nanoparticles diameters calculated by TEM and XRD. The layer thickness of

the sample IO125 also matches well with the information obtained by TEM. On the other hand, values of the cluster size (ξ) and the fractal dimension (D_F) for samples IO125 and IO150 are representative of system with larger and compact aggregates.

(c) TEM images and Particle Size distribution:

Iron Oxide nanoparticles:



Figure S3 (a-f) HRTEM images of iron-oxide nanoparticles (IONPs) oxidized at 200°C for 30 minutes during cooling from 315°C.



Figure S4 (a-d) HRTEM images of the iron oxide nanoparticles (IONPs) oxidized at 250°C for 30 minutes during cooling from 315°C.



Figure S5 Particle size histograms derived by counting at least 100 particles for the Iron oxide nanoparticles.

Au Nano seeds:



Figure S6 Particle size histograms for the Au nanoparticles.





Figure S7 (a-f) HRTEM images of Au-iron oxide nanoparticles oxidized at 125°C for 30 minutes during cooling from 315°C.



Figure S8 (a-f) HRTEM images of iron oxide nanoparticles oxidized at 150°C for 30 minutes during cooling from 315° C



Figure S9 (a-c) HRTEM images of iron oxide nanoparticles oxidized at 200°C for 30 minutes during cooling from 315°C



Figure S10 HRTEM images for iron oxide nanoparticles oxidized at 250°C for 30 minutes during cooling from 315°C.



Figure S11 Particle size histograms derived by counting at least 100 particles for the Iron oxide nanoparticles.

TABLE III Nanocomposites sizes calculated from the Scherrer equation (XRD) and TEM imaging, (a) iron oxide, and (b) Au-iron oxide

Sample	XRD	TEM
	Scherrer's formula (nm)	Mean (nm)/Std. dev.
IO125	35.6	39 ± 3.0
IO150	36.7	38 ± 2.5
IO200	26.6	12 ± 2.0
IO250	12.6	10 ± 2.0

(a)

(b)

Sample	XRD Scherrer formula (nm)	TEM Mean (nm)/Std. dev.
AI0125	8.5	9.4 ± 3.0
AI0150	27.3	30 ± 2.5
AIO200	30.6	32 ± 2.3
AIO250	15.4	16 ± 2.0
Au NPs	13.4	14 ± 1.0

Sample	(311)	(111)	(220)	(311)	(422)	(511)	(440)
IO125	8.369 Fe(110)	4.845	2.960	2.522	1.706	1.608	1.477
	2.855						
	FeO(200)						
	4.181						
IO150	8.364	4.835	2.957	2.521	1.706	1.608	1.477
	Fe(110)						
	2.854						
10200	8.361	4.832	2.956	2.520	1.705	1.607	1.476
	FeO(OH)(110)						
	5.838						
10250	8.352	4.844	2.953	2.518	1.704	1.605	1.474

Table IV Lattice Parameters (in Å) for Samples IO125-IO250

Sample	Subspectrum	I.S	Q.S	H(T)	Area
		(mm/s)	(mm/s)		
	1	0.98±0.02	0.59±0	-	3.85%
	2	0.28±0.003	-0.02	49.20±0.03	33.33%
10125	3	0.67±0.003	-0.02±0.005	46.02±0.02	40.79%
	4	0.006±0.005	0	33.24±0.04	13.99%
	5	0.38±0.02	0	43.09±0.4	7.71%
	1	0.38±0.02	0.73±0.04	-	2.31%
10150	2	0.28±0.002	-0.02±0.004	49.39±0.02	36.07%
	3	0.67±0.002	0.0002±.004	46.22±0.02	41.0%
	4	-0.005±0.003	0	33.24±0.02	20.62%
	1	0.29±0.005	- 0.004±0.008	49.35±0.04	41.81%
10200	2	0.67±0.004	- 0.003±0.007	46.25±0.04	45.48%
	3	0.63±0	0	42.28±0.9	12.71%
	1	0.35±0.003	-0.01±0.005	48.89±0.05	46.15%
10250	2	1.11±0	0	40.88±0.6	27.69%
	3	0.37±0.01	-0.06±0.02	45.56±0.4	16.93%

Table V (a) Mossbauer hyperfine parameters iron oxide nanoparticles

Table V ((b)	Mossbauer	hy	perfine	parameters of	f Au-iron	oxide	nanoparticles

Sample	Subspectrum	I.S	Q.S	Н(Т)	Area
		(mm/s)	(mm/s)		
	1	0.35±0.01	-	-	42.66%
AIO125	2	0.334±0.008	0.67±0.012	-	6.05%
	3	0.34±0.05	0	-	49.03%
				0.048±10797.001	
	4	0.34±0.07	0	47.842±0	2.26%
	1	0.278±0.05	-	-	26.29%
AIO150	2	0.31±0.004	-0.03±0.008	49.425±0.03	47.02%
	3	0.64±0.01	0.05±0.017	45.65±0.08	26.68%
AIO200	1	0.65±0.01	0.04±0.02	45.84±0.08	44.37%
	2	0.31±0.007	0±0.08	49.19±0	55.62%
	1	0.29±0.004	-0.02±0.005	48.69±0.02	32.20%
	2	0	0	43.86±0.6	13.42%
AIO250	3	0.63±0.007	-0.006 ±0	45.68±0.07	20.99%
	4	0	0	39.57±1.67	7.31%
	5	0.29±0.08	0	27.48±1.97	25.33%

Sample	TEM	Magneti te	Fe(II)	Fe(III)	α-Fe	Collective excitation	A (Fe ³⁺)	B (Fe ³⁺ /Fe ² ⁺)
10125	39 ± 3.0	82%	4%	-	14%	7%	0.45	0.55
IO150	38 ± 2.5	77%	-	2%	21%	-	0.47	0.53
10200	29 ± 2.0	100%	-	-	-	13%	0.48	0.52
IO250	15 ± 2.0	100%	-	-	-	53%	0.52	0.48
AIO125	9.4 ± 3.0	94%	-	6%	-	50% 42% SPM	-	-
AIO150	30 ± 2.5	100%	-	-	-	26%	0.66	0.33
AIO200	32 ± 2.3	100%	-	-	-	-	0.55	0.45
AIO250	16 ± 2.0	100%	-	-	-	47%	0.62	0.38

Table V (c) Mössbauer hyperfine parameters iron oxide and Au-iron oxide nanoparticles

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