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A heteropolyacid-based ionic liquid immobilized onto magnetic fibrous nano-silica as robust and recyclable heterogeneous catalysts for the synthesis of tetrahydrodipyrazolopyridines in water

Seyed Mohsen Sadeghzadeh

Department of Chemistry, College of Sciences, Sina Masihabadi Student Research, P.O. Box 97175-615, Neishabour, Iran. Fax/Tel: +98 561 2502065; E-mail: seyedmohsen.sadeghzadeh@gmail.com

Table S1 Comparison of the catalytic efficiency of Fe₃O₄/SiO₂/KCC-1/SO₄H MNPs with various catalysts.

Entry	Catalyst	Solvent	Amount catalyst (mol%)	T (°C)	Time (h)	Yield (%) ^a
1	$ZnCl_2$	EtOH	10	Reflux	5	88 [51]
2	FeCl ₃	EtOH	10	Reflux	5	85 [51]
3	$MgCl_2$	EtOH	10	Reflux	5	84 [51]
4	$Yb(OTf)_3$	EtOH	5	Reflux	3	90 [51]
5	$Cu(ClO_4)_2_6H_2O$	EtOH	5	Reflux	3	90 [51]
6	Vitamin B1	EtOH	5	Reflux	4	88 [51]
7	TsOH	EtOH	10	Reflux	4	88 [51]
8	NH_2SO_3H	EtOH	10	Reflux	4	86 [51]
9	-	EtOH	-	Reflux	5	79 [52]
10	-	H ₂ O/EtOH	-	Reflux	6	40 [52]
11	-	[Hmim]TFA	-	Reflux	12	60 [52]
12	p-TSA	H_2O	_b	Reflux	6	36 [52]
13	K_2CO_3	H_2O	_ b	Reflux	9	68 [52]
14	Piperidine	H_2O	_ b	Reflux	9	43 [52]
15	Fe ₃ O ₄ /KCC-1/IL/HPW	H_2O	0.4	r.t.	0.5	96 [52]

^a Reaction conditions: benzaldehyde (1 mmol), phenylacetylene (1 mmol), 1H-benzo[d]imidazol-2-amine (1 mmol) in different amounts of catalyst, temperature, and time.

^b The amount of catalyst is not mentioned in the article.