## Supplemental Information

"Rigid versus flexible anilines or anilides confirm the bicyclic ring as the hydrophobic portion for optimal $\sigma_{\mathbf{2}}$ receptor binding and provide novel tools for the development of future $\sigma_{\mathbf{2}}$ receptor PET radiotracer"

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## General procedure for the synthesis of 3-Chloro- $N$-Phenylpropanamides (3a-c).

One of the appropriate anilines ( 2.5 mmol ) was dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and after cooling at $0{ }^{\circ} \mathrm{C}, \mathrm{Et}_{3} \mathrm{~N}(1 \mathrm{mmol}, 0.13 \mathrm{~mL})$ was added under a stream of $\mathrm{N}_{2}$. 3-Chloropropionyl chloride was then added to the solution in a dropwise manner, and the resulting mixture was stirred for 2 h at $0^{\circ} \mathrm{C}$. The solution was then quenched with $\mathrm{H}_{2} \mathrm{O}$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10 \mathrm{~mL})$. The collected organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure to afford a crude solid, which was used for the next step without purification ( $75 \%$ yield).

3-Chloro- $N$-phenylpropanamides (3a): GC-MS $m / z 185\left(\mathrm{M}^{+}+2,10\right), 183\left(\mathrm{M}^{+}, 30\right), 93(100)$.
3-Chloro-N-3-methoxyphenylpropanamides (3b): GC-MS $m / z 215\left(\mathrm{M}^{+}+2,10\right), 213\left(\mathrm{M}^{+}, 35\right)$, 123 (100). IR cm ${ }^{-1}: 3300,3090,1664,1610$.

3-Chloro-N-4-fluorophenylpropanamides (3c): GC-MS m/z $203\left(\mathrm{M}^{+}+2,10\right), 201\left(\mathrm{M}^{+}, 30\right), 111$ (100).

## General procedures for the synthesis of alkyl chloride derivatives (9a-c)

To a suspension of $\mathrm{NaH}(0.16 \mathrm{~g}, 6.80 \mathrm{mmol})$ in dry DMF $(15 \mathrm{~mL})$, a solution in DMF $(5 \mathrm{~mL})$ of the appropriate 3,4-dihydroquinolin-2(1H)-one (8a-c) ( 2.7 mmol ), was added in a dropwise manner, at $0{ }^{\circ} \mathrm{C}$ under a stream of $\mathrm{N}_{2}$. After $30 \mathrm{~min}, 1$-bromo-3-chloropropane ( $3 \mathrm{mmol}, 0.29 \mathrm{~mL}$ ) was added in a dropwise manner and the mixture was allowed to warm to room temperature and was stirred for 4 h . After cooling to $0{ }^{\circ} \mathrm{C} \mathrm{H}_{2} \mathrm{O}$ was added and the solvent was removed under reduced pressure. The residue was taken up with water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The organic layers were collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure to afford a crude residue as a brown solid.

1-(3-Chloropropyl)-3,4-dihydroquinolin-2(1H)-one (9a) was purified by column chromatography with $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{AcOEt}(8: 2)$ as eluent to provide a yellow oil ( $0.33 \mathrm{~g}, 55 \%$ ); GC-MS $m / z 225\left(\mathrm{M}^{+}+2\right.$, 15), $223\left(\mathrm{M}^{+}, 45\right), 188(55), 132(100)$.

1-(3-Chloropropyl)-5-methoxy-3,4-dihydroquinolin-2(1H)-one (9b) the crude residue was used for the next step with no purification ( $0.39 \mathrm{~g}, 58 \%$ ); GC-MS $m / z 255\left(\mathrm{M}^{+}+2,25\right), 253\left(\mathrm{M}^{+}, 75\right)$, 218 (85), 162 (100).

1-(3-Chloropropyl)-6-fluoro-3,4-dihydroquinolin-2(1H)-one (9c) was purified by column chromatography with $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{AcOEt}(9: 1)$ as eluent to provide a colorless oil ( $0.42 \mathrm{~g}, 65 \%$ ); GCMS m/z $243\left(\mathrm{M}^{+}+2,15\right), 243\left(\mathrm{M}^{+}, 45\right), 206(60), 150(100)$.

## Table of Elemental Analyses

|  | Calculated |  |  | Found |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Compound | $\mathbf{C}$ | $\mathbf{H}$ | $\mathbf{N}$ | $\mathbf{C}$ | $\mathbf{H}$ | $\mathbf{N}$ |
| $\mathbf{4 a}$ | 63.74 | 6.69 | 7.43 | 63.56 | 6.70 | 7.34 |
| $\mathbf{4 b}$ | 61.99 | 6.69 | 6.88 | 62.10 | 6.73 | 7.00 |
| $\mathbf{4 c}$ | 60.83 | 6.13 | 7.09 | 61.10 | 6.30 | 7.12 |
| $\mathbf{5 a}$ | 70.15 | 7.07 | 7.01 | 70.30 | 7.20 | 7.11 |
| $\mathbf{5 b}$ | 58.13 | 7.09 | 6.46 | 58.02 | 6.98 | 6.46 |
| $\mathbf{5 c}$ | 56.94 | 6.57 | 6.64 | 57.08 | 6.51 | 6.91 |
| $\mathbf{6 a}$ | 61.82 | 7.43 | 6.55 | 61.25 | 7.06 | 6.73 |
| $\mathbf{6 b}$ | 59.81 | 6.73 | 5.58 | 59.81 | 6.43 | 5.66 |
| $\mathbf{6 c}$ | 62.48 | 6.67 | 6.62 | 62.30 | 6.56 | 6.70 |
| $\mathbf{7 a}$ | 61.82 | 7.55 | 6.55 | 62.03 | 7.43 | 6.60 |
| $\mathbf{7 b}$ | 60.39 | 7.49 | 6.12 | 60.56 | 7.40 | 6.21 |
| $\mathbf{7 c}$ | 59.33 | 7.02 | 6.29 | 59.24 | 7.13 | 6.40 |
| $\mathbf{1 0 a}$ | 66.26 | 7.01 | 6.72 | 66.06 | 6.97 | 6.81 |
| $\mathbf{1 0 b}$ | 60.58 | 7.25 | 5.89 | 60.32 | 6.97 | 5.95 |
| $\mathbf{1 0} \mathbf{c}$ | 61.60 | 6.63 | 6.25 | 61.56 | 6.44 | 6.26 |
| $\mathbf{1 1 a}$ | 62.23 | 7.38 | 6.31 | 62.13 | 7.29 | 6.32 |
| $\mathbf{1 1 b}$ | 60.82 | 7.34 | 5.91 | 60.82 | 7.33 | 6.00 |
| $\mathbf{1 1 c}$ | 59.80 | 6.87 | 6.06 | 59.56 | 6.85 | 6.08 |

Table of Physical Properties of Novel Compounds

| Compound | Formula ${ }^{\text {a }}$ | mp, ${ }^{\circ} \mathrm{C}$ |
| :---: | :---: | :---: |
| 4 a | $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 232-235 |
| 4b | $\mathrm{C}_{21} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{4} \cdot \mathrm{HCl} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 195-197 |
| 4 c | $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 244-246 |
| 5 a | $\mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 2 \mathrm{HCl} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 230-232 |
| 5b | $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 235-237 |
| 5c | $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~F} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 248-250 |
| 6 a | $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{HCl} \cdot 1.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 129-132 |
| 6b | $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{4} \cdot(\mathrm{COOH})_{2} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 119-121 |
| 6 c | $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~F} \cdot \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 126-129 |
| 7 a | $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 2 \mathrm{HCl} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 187-190 |
| 7b | $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 2 \mathrm{HCl} \cdot 1.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 174-176 |
| 7c | $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~F} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 210-213 |
| 10a | $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{HCl}^{\text {b }}$ | 191-193 |
| 10b | $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{4} \cdot \mathrm{HCl} \cdot 1.6 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 188-190 |
| 10c | $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~F} \cdot \mathrm{HCl} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 225-227 |
| 11a | $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 211-213 |
| 11b | $\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\text {b }}$ | 181-183 |
| 11c | $\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~F} \cdot 2 \mathrm{HCl} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}^{\mathrm{b}}$ | 230-232 |

${ }^{\text {a }}$ Elemental analyses for $\mathrm{C}, \mathrm{H}, \mathrm{N}$ were within $\pm 0.4 \%$ of the theoretical values for the formulas given. ${ }^{\mathrm{b}}$ Recrystallized from MeOH.

