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## **Supporting Information**

ZnO rods rooted on manifold carbon nanofiber paper as scalable photocatalyst platform: *Effects of ZnO morphology* 

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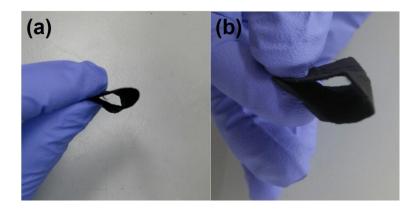
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**Table S1.** Adsorption of phenol in CNF/ZnO samples. The samples ( $\sim$ 4.6 mg) were immersed in aqueous phenol solution (200  $\mu$ M; 50 mL) for 30 min prior to irradiation.

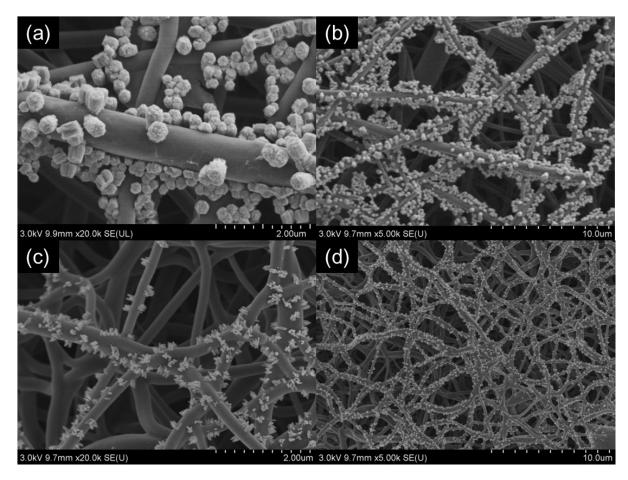
Samples	CZ-a-0.25	CZ-a-0.5	CZ-0.25	CZ-0.5
Adsorption amount (µmol/g)	53.2	51.5	54.9	54.5
Adsorption capacity (%)	2.45	2.37	2.53	2.51

**Table S2.** Charge transfer resistance (R<sub>ct</sub>) of the samples.

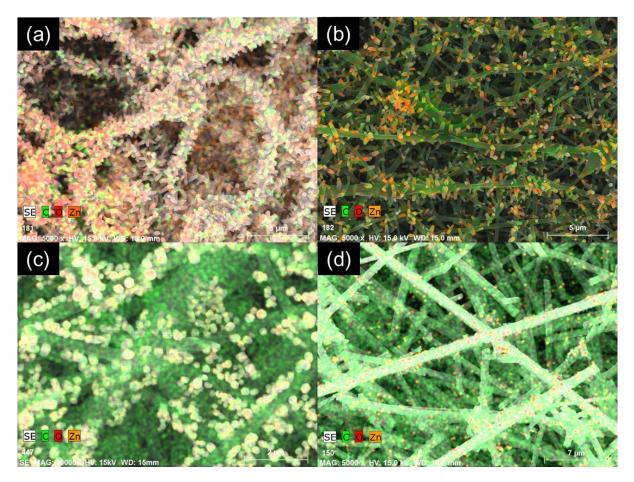
Samples	CZ-a-0.25	CZ-a-0.5	CZ-0.25	CZ-0.5	NP
$R_{ct}\left(\Omega\right)$	57.07	81.83	143.1	113.9	278.1



**Fig. S1**. Photographs of (a) an as-synthesized CNF film and (b) a CNF manifold film (fold five times). Both were annealed at  $1400\,^{\circ}$ C.



**Fig. S2**. SEM images of (a, b) CZ-0.5 and (c, d) CZ-0.25. They were synthesized via the electrodeposition of ZnO on carbon nanofibers-woven webs placed in the aqueous solutions of zinc acetate at 0.5 and 0.25 mM, respectively.



**Fig. S3**. EDX mapping of (a) CZ-a-0.5 and (b) CZ-a-0.25 (c) CZ-0.5 and (d) CZ-0.25.

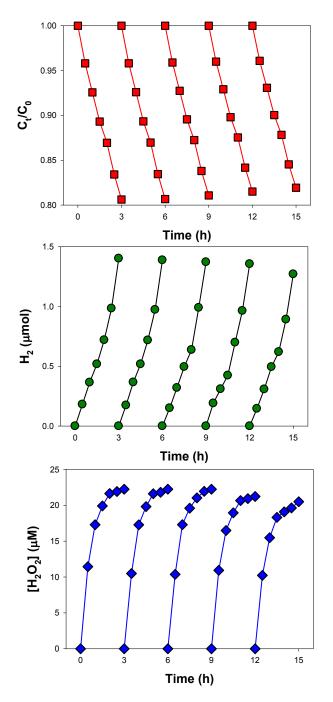
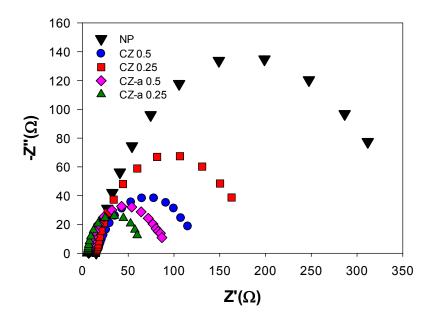


Fig. S4. Reusability tests of CZ-a-0.25 sample for phenol (PhOH, 200  $\mu M)$  degradation, and  $H_2$  and  $H_2O_2$  productions.



**Fig. S5**. EIS analysis of the samples. They were held at a bias potential of -0.2 V vs. Ag/AgCl in 1 M KOH solution with a frequency range of 100 kHz to 0.1 Hz with an AC perturbation of 5 mV.

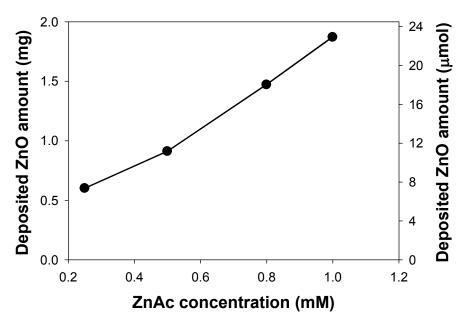
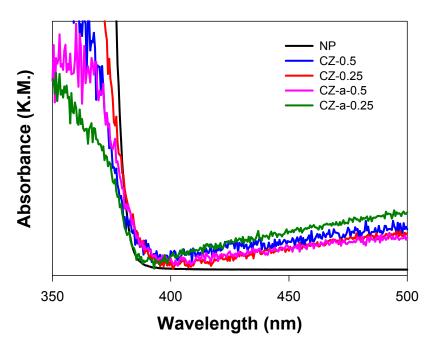


Fig. S6. Effect of ZnAc concentration on the amount of ZnO loaded in CNF.



**Fig. S7**. UV-vis diffuse reflectance absorption spectra of samples. The absorbance was expressed as a Kubelka-Munk unit.

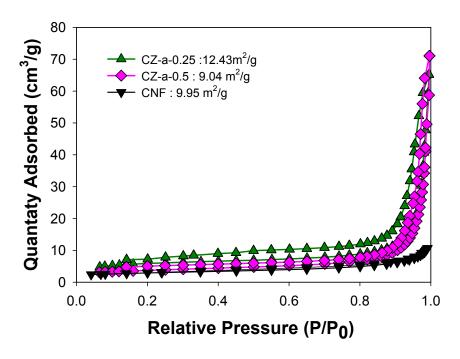


Fig. S8. N<sub>2</sub> adsorption-desorption BET isotherms of CZ-a-0.25, CZ-a-0.5, and CNF.