## **Supporting Information**

Three dimensional hierarchically porous crystalline MnO<sub>2</sub> structure design for high rate performance lithium-ion battery anode

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\*Corresponding authors. E-mail: jpzhao@hit.edu.cn; liu88062321@163.com; yaoli@hit.edu.cn **Electrodeposited Processes of Manganese Dioxide:** 

Anode: 
$$\operatorname{Mn}^{2+} + 2\operatorname{H}_2\operatorname{O} \to \operatorname{MnO}_2 + 4\operatorname{H}^+ + 2e^-$$
 (1)

Cathode: 
$$2H^+ + 2e^- \rightarrow H_2$$
 (2)

Overall: 
$$Mn^{2+} + 2H_2O \rightarrow MnO_2 + H_2 + 2H^+$$
 (3)

Water electrolysis: 
$$2H_2O \rightarrow 2H_2 + O_2$$
 (4)



Fig. S1 Optical image of Ni foam substrate, Ni foam/PS template, HPA-Mn and HPC-Mn samples

## directly grown on Ni foam.



Fig. S2 Low resolution SEM images of (a) pure Ni foam and (b) HPC-Mn sample.



Fig. S3 The wide scan XPS spectrum of HPC-Mn sample.



Fig. S4 Two-dimensional SAXS patterns of HPA-Mn anode.



Fig. S5 Schematic representation of the mesoporous formation assisted by oxygen.



Fig. S6 N<sub>2</sub> adsorption-desorption isotherms of (a) HPC-Mn anode and (b) HPA-Mn anode.



Fig. S7 Nyquist plots of HPC-Mn and HPA-Mn anodes.

MnO <sub>2</sub> Based Anodes	Low Current Density (A g <sup>-1</sup> ) Capacity (mAh g <sup>-1</sup> )	High Current Density (A g <sup>-1</sup> ) Capacity (mAh g <sup>-1</sup> )	Current Density Ratio	Capacity Retention (%)	Ref.
HPC-Mn	0.1 (973.8)	(IIIAII g ) 2.0 (798.8)	20	~82.0	Our Work
MnO <sub>2</sub> /3D porous graphene	0.1 (926.0)	1.6 (433.0)	16	~46.8	1
MnO <sub>2</sub> network-Ni/PVDF double shell/core fiber	0.05 (1079.0)	0.6 (544.7)	12	~50.5	2
MnO <sub>2</sub> @N-doped carbon nanotubes	0.05 (1146.0)	1.0 (620.7)	20	~54.2	3
Freestanding MnO <sub>2</sub> /Ni /PVDF coaxial fiber	0.05 (1178.4)	1.0 (415.0)	20	~35.2	4
MnO <sub>2</sub> on 3D N-doped graphene hybrid aerogels	0.1 (1003.0)	1.5 (636.0)	15	~63.4	5
MnO <sub>2</sub> nanoflakes on reduced graphene oxide	0.1 (1430.0)	2.0 (1000.0)	20	~69.9	6
Nanoflaky MnO <sub>2</sub> on carbon microbeads	0.1 (700.0)	1.5 (230.0)	15	~32.9	7
Mesoporous MnO <sub>2</sub> nanosheet arrays	0.1 (-)	1.0 (-)	10	~50.0	8
Nanoflaky MnO <sub>2</sub> /carbon nanotube	0.2 (820.0)	2.0 (420.0)	10	~51.2	9

Table S1 Rate performance comparison of the HPC-Mn and other reported MnO<sub>2</sub>-based anodes for

MnO <sub>2</sub> Based Anodes	Current	Cycling	Specific Capacity (mA	Capacity		
	Density (A g <sup>-1</sup> )	Number	h g <sup>-1</sup> ) After Cycling	Retention (%)	Ref.	
HPC-Mn	0.4	200	778.0	~97.6	Our Work	
MnO <sub>2</sub> /3D porous graphene	0.1	100	836.0	~84.6	1	
MnO <sub>2</sub> network-Ni/PVDF double shell/core fiber	0.2	70	500.2	-	2	
MnO <sub>2</sub> @N-doped carbon nanotubes	0.1	100	1415.0	>100	3	
Freestanding MnO <sub>2</sub> /Ni /PVDF coaxial fiber	0.05	70	1031.2	-	4	
MnO <sub>2</sub> on 3D						
N-doped graphene hybrid	0.4	200	909.0	>100	5	
aerogels MnO <sub>2</sub> nanoflakes on reduced graphene oxide nanosheets	1.0	200	1000.0	~100	6	
Nanoflaky MnO <sub>2</sub> on carbon microbeads	0.1	100	525.0	-	7	
Mesoporous MnO <sub>2</sub> nanosheet arrays	1.0	200	900.0	>100	8	
Nanoflaky MnO <sub>2</sub> /carbon nanotube	0.2	50	620.0	~77.0	9	

Table S2 Cycling performance comparison of the HPC-Mn and other reported MnO<sub>2</sub>-based anodes

for LIBs.

Note: The specific capacities of some  $MnO_2$  based anodes would increase with the increase of cycling number, so the capacity retention is >100%.

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