Supporting Information

Cobalt/Nitrogen co-Doped Porous Carbon Nanosheets as Highly Efficient Catalysts for Oxygen Reduction Reaction in Both Basic and Acidic Media

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Figure S1. XRD analysis of the GMC-CoPor-800

Table S1 The content of Co and N element in the composite GMP-CoPor and CMP-

Sample	Name	%At Conc	N/Co ratio
GMP-CoPor	N 1s	3.58	6 1
	Co 2p	0.59	0.1
GMP-CoPor-700	N 1s	3.13	7.5
	Co 2p	0.42	
GMP-CoPor-800	N 1s	2.50	7.1
	Co 2p	0.35	
GMP-CoPor-900	N 1s	1.73	9.1
	Co 2p	0.19	

CoPor-T (T=700, 800 and 900)

Materials	Half-wave Potential (V vs Ag/AgCl)	Diffusion-limited	
		Current Density	Reference
		(mA cm ⁻²)	
Co/N co-doped porous carbon nanosheets	-0.147	4.64	This Work
Co-N doping of carbon/graphene sheets	-0.18	5.21 (-0.3 V)	[1]
Fe-N co-doped mesoporous 2D Carbon	-0.156	5.64 (-0.4 V)	[2]
Co-N co-doped WC Carbide	-0.194	4.8 (-0.4 V)	[3]
Co-N co-doped graphene-like carbon nanosheets	-0.151	~5.7	[4]
N-doped graphene aerogel supported Co nanoparticles	~-0.25	4.5	[5]
Co and N co-doped carbon	-0.115	~4.6	[6]
Co and P co-doped reduced graphene	~-0.19	~-4.9 (-0.4 V)	[7]
Fe, Co, N-doped 3D porous carbon foams	~-0.17	~4.3 (-0.4 V)	[8]

Table S2ORR performance of GMC-CoPor-700 and some recently reported Metal/Nco-doped carbon materials. All the electrocatalysts were tested in 0.1M KOH

Materials	Half-wave Potential (V vs Ag/AgCl)	Diffusion-limited	
		Current Density	Reference
		(mA cm ⁻²)	
Co/N co-doped porous	0.54	5.10	This Work
Fa N doned vertically			
aligned carbon	0.54	-6	[0]
nanotubes catalysts	0.34		
Fe-polypyrrole			
derived carbon	0.50	5.7	[10]
electrocatalysts			
Cobalt tripyridyl			
triazine derived Co-N	~0.4	~4.3	[11]
doped carbon			
Fe, Co-N doping of			
carbon/graphene	0.63	~4.8	[1]
sheets			
Mesoporous Co-N co-			
doped carbon derived	0.52 ± 0.02	4.5	[12]
from VB12			

Table S3ORR performance of GMC-CoPor-700 and some recently reportedMetal/N co-doped carbon materials. All the electrocatalysts were tested in 0.5M H2SO4

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