## **Supporting Information**

## Kinetics and Mechanisms of Gas Phase Reactions of Hexenols with Ozone

Xiaoxiao Lin,<sup>1</sup> Qiao Ma,<sup>1,3</sup> Chengqiang Yang,<sup>1,2</sup> Xiaofeng Tang, <sup>1</sup> Weixiong Zhao,<sup>1</sup>

Changjin Hu,<sup>1</sup> Xuejun Gu,<sup>1</sup> Bo Fang,<sup>1</sup> Yanbo Gai,<sup>1,\*</sup> Weijun Zhang,<sup>1,2,\*</sup>

- <sup>1</sup> Laboratory of Atmospheric Physico-Chemistry, Anhui Institute of Optics and Fine Mechanics, Chinese Academy of Sciences, Hefei, 230031, Anhui, China
- <sup>2</sup> School of Environmental Science and Optoelectronic Technology, University of Science and Technology of China, Hefei, 230026, Anhui, China

<sup>3</sup>University of Science and Technology of China, Hefei 230026, China

\* Corresponding author:

- Prof. Weijun Zhang: wjzhang@aiofm.ac.cn
- Dr. Yanbo Gai: gaiyanbo@aiofm.ac.cn

## **SUMMARY:**

11 pages, including 9 Figures.

## **Figure Captions:**

- **Fig. S1** Concentration measurement of (E)-2-hexen-1-ol by GC/FID in absence of O<sub>3</sub> (values used here were from the integrated peak areas, which were proportional to the concentration of the measured hexenol).
- Fig. S2 Pseudo-first-order plots for  $O_3$  reactions with different concentrations of (E)-3-hexen-1-ol (in molecule cm<sup>-3</sup>).
- Fig. S3 Pseudo-first-order plots for  $O_3$  reactions with different concentrations of (E)-4-hexen-1-ol (in molecule cm<sup>-3</sup>).
- Fig. S4 Pseudo-first-order plots for O<sub>3</sub> reactions with different concentrations of (Z)-2-hexen-1-ol (in molecule cm<sup>-3</sup>).
- Fig. S5 Pseudo-first-order plots for O<sub>3</sub> reactions with different concentrations of (Z)-3-hexen-1-ol (in molecule cm<sup>-3</sup>).
- Fig. S6 Pseudo-first-order plots for  $O_3$  reactions with different concentrations of (Z)-4-hexen-1-ol (in molecule cm<sup>-3</sup>).
- **Fig. S7** Structures of the hexenols optimized at the BH&HLYP/6-31+G(d,p) level of theory.
- Fig. S8 Schematic potential energy surface for the reactions of (E)-3-hexen-1-ol and (Z)-3-hexen-1-ol with O<sub>3</sub> at the BH&HLYP/6-31+G(d,p) level of theory (in kcal/mol).
- Fig. S9 Schematic potential energy surface for the reactions of (E)-4-hexen-1-ol and (Z)-4-hexen-1-ol with O<sub>3</sub> at the BH&HLYP/6-31+G(d,p) level of theory (in kcal/mol).

Fig. S1



Wall loss experiments were studied by measuring the decrease of the integrated peak areas using a GC/FID (GC7820A, Agilent Technologies). As can be seen from Fig.S1, within the typical time span of the  $O_3$  reaction experiments (less than half an hour) in this work, the decrease of the integrated peak areas measured by GC/FID was below 0.2% of their initial values, which could be neglected. Even after 2 hours, the decrease was still smaller than 3% of their initial values. So, the wall loss of (E)-2-hexen-1-ol in our experiments could be negligible.

Fig. S2



Fig. S3



```
Fig. S4
```



```
Fig. S5
```



```
Fig. S6
```







