

## Supporting Information

### **Triphenylamine based Lab-on-a-molecule for the highly selective and sensitive detection of $Zn^{2+}$ and $CN^-$ in aqueous solution**

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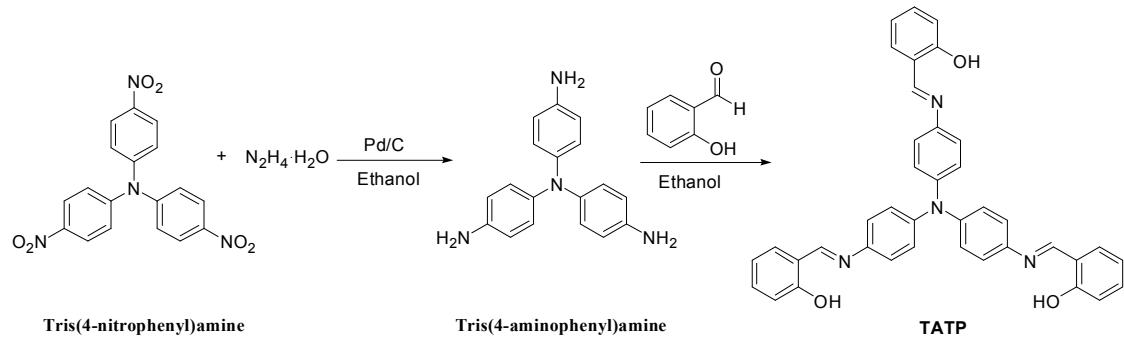
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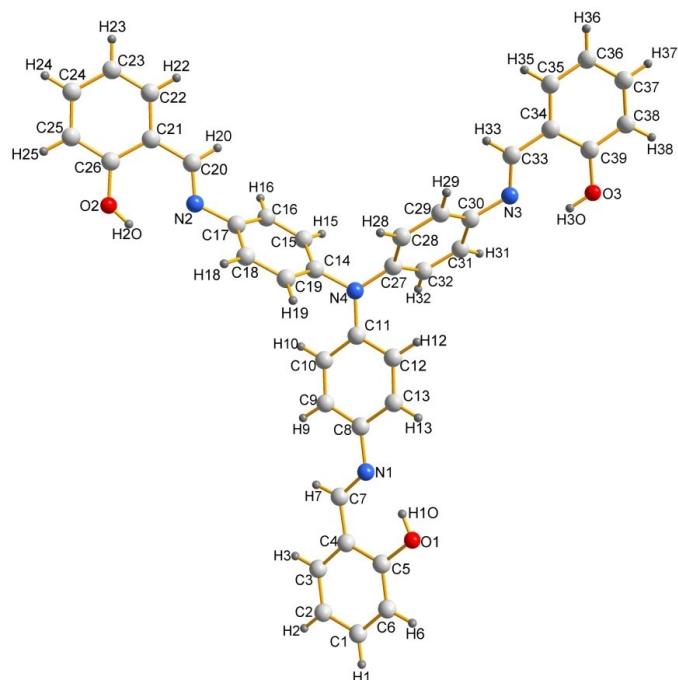
Email: chenxin@sysucc.org.cn.

<sup>c</sup> Institute of Theoretical Chemistry, Jilin University, Changchun, China.

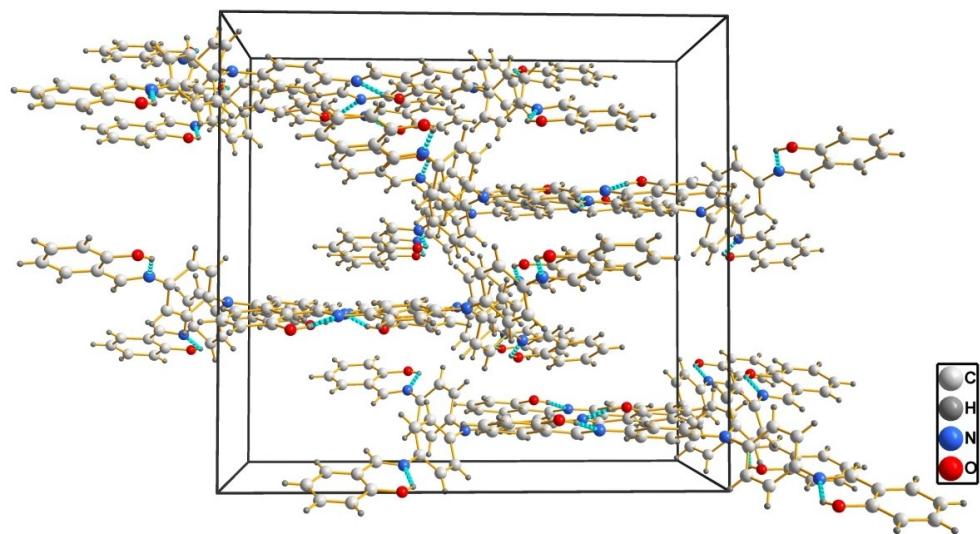
Email: dequan\_wang@jlu.edu.cn.



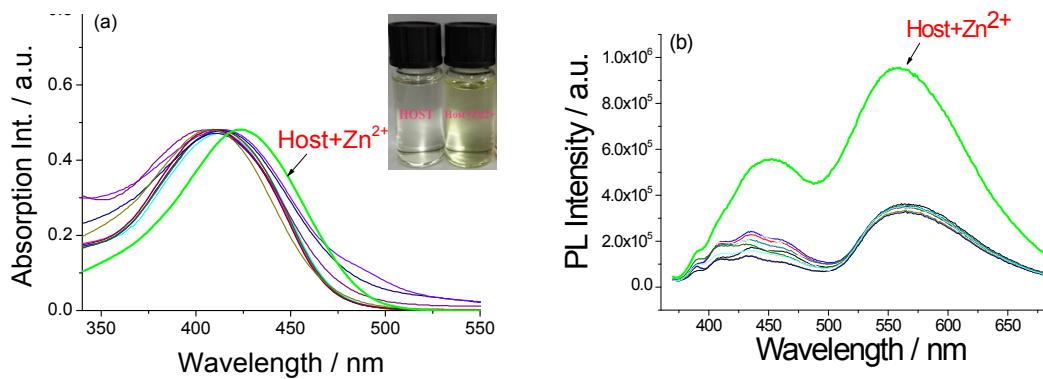
**Fig. S1** The synthesis route of TATP.



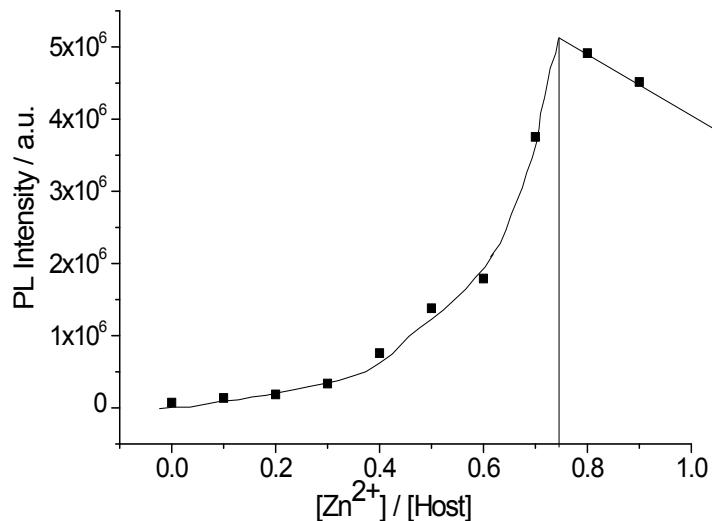
**Fig. S2** Crystal structure of TATP and its unit cell. The deposit CCDC number: 1476312.



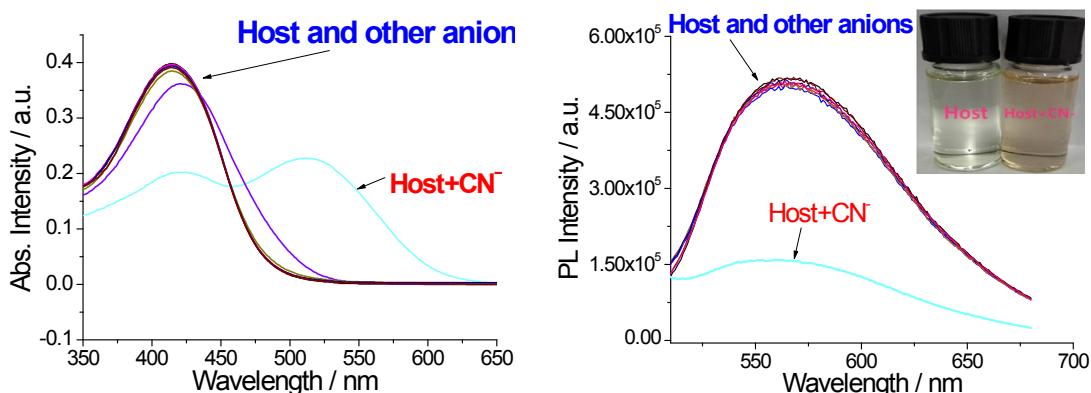
**Fig. S3** Packing cell of the crystal structure of **TATP**.



**Fig. S4** (a) UV-vis spectra of **TATP** (10  $\mu$ M) in 0.1 M Tris-ClO<sub>4</sub> buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v) after addition of 50.0 equiv of various metal ions. Inset: The color changes of **TATP** (10  $\mu$ M) upon addition of Zn<sup>2+</sup>; (b) Fluorescence emission spectra ( $\lambda_{\text{ex}} = 350$  nm) of **TATP** (10  $\mu$ M) in the presence of 50.0 equiv of various metal ions in 0.1 M Tris-ClO<sub>4</sub> buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v).

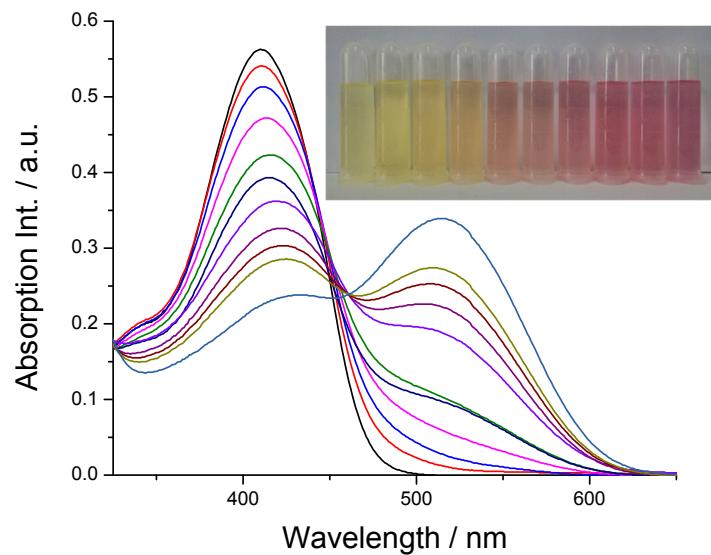


**Fig. S5** Jobs plot of **TATP** and Zn<sup>2+</sup> in 0.1 M Tris-ClO<sub>4</sub> buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v). The total concentration of Zn<sup>2+</sup> and **TATP** is  $1 \times 10^{-4}$  M.

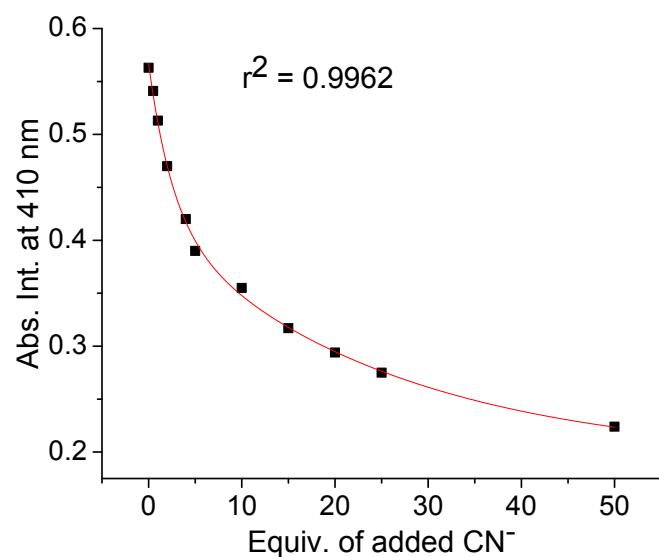


**Fig. S6** UV-vis (a) and emission spectra (b,  $\lambda_{\text{ex}} = 350$  nm) of **TATP** (10  $\mu$ M) in the presence of 50.0 equiv of various metal ions in 0.1 M Tris-ClO<sub>4</sub> buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v); Inset: The solution

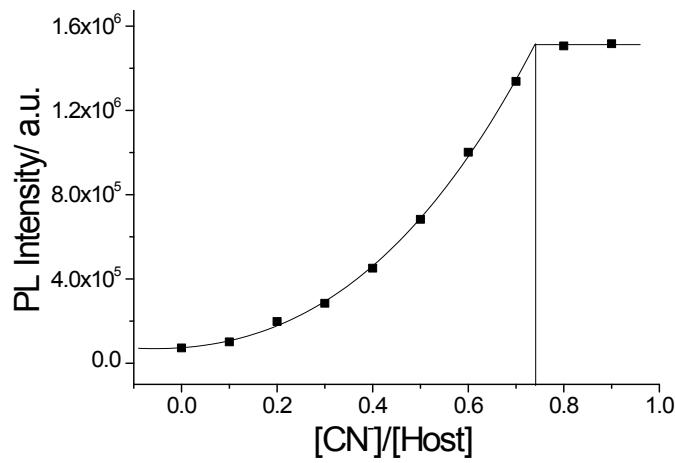
color changes of **TATP** (10  $\mu$ M) upon addition of  $\text{CN}^-$ .



**Fig. S7** UV/vis spectra of **TATP** (10  $\mu$ M) in 0.1 M Tris- $\text{ClO}_4$  buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v) after addition of 0.5-50.0 equiv of  $\text{CN}^-$ .

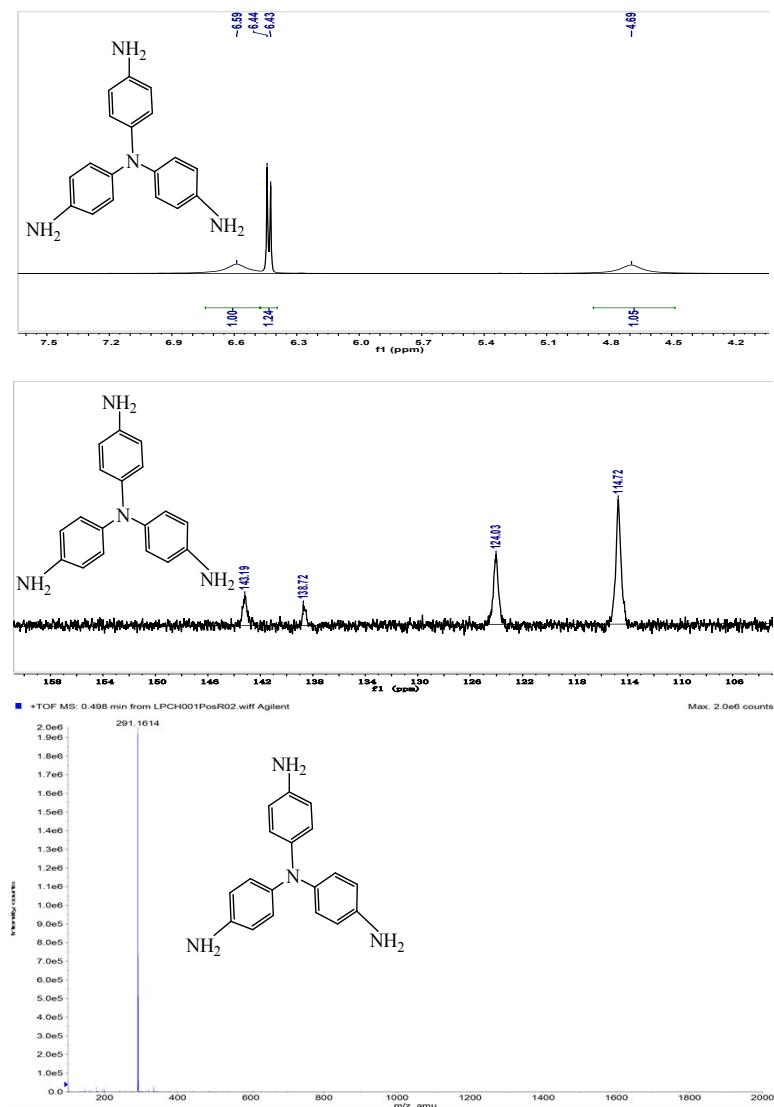


**Fig. S8** The absorption intensity of **TATP** at 410 nm as a function of  $\text{CN}^-$  equivalent in 0.1 M Tris- $\text{ClO}_4$  buffer solution (pH = 7.24, DMF: buffer = 1 / 2, v / v).

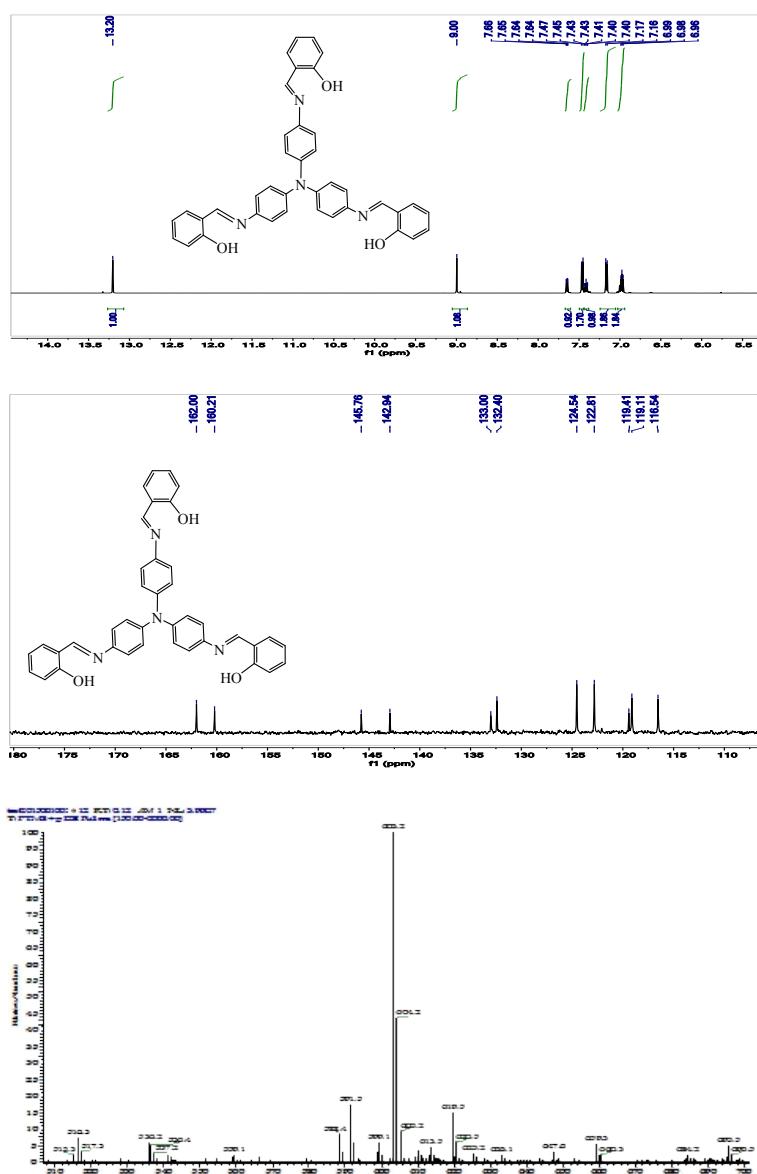


**Figure S9** Jobs plot of TATP and  $CN^-$  in 0.1 M Tris- $ClO_4$  buffer solution ( $pH = 7.24$ , DMF: buffer = 1 / 2, v / v).

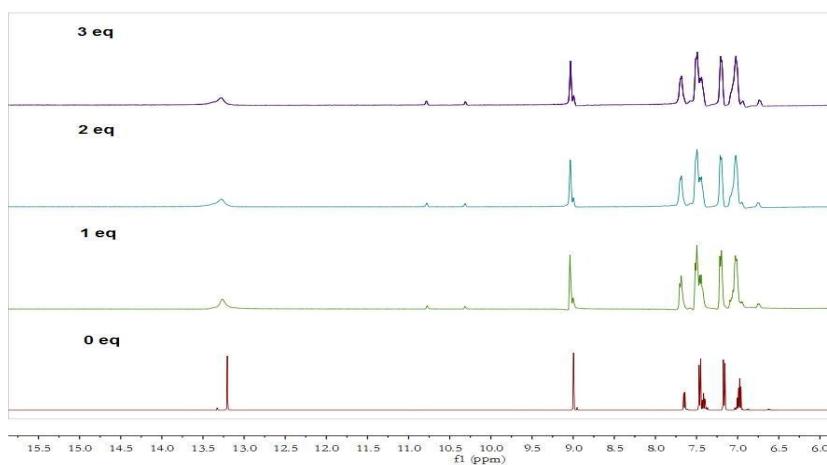
The total concentration of  $CN^-$  and TATP is  $1 \times 10^{-4}$  M.



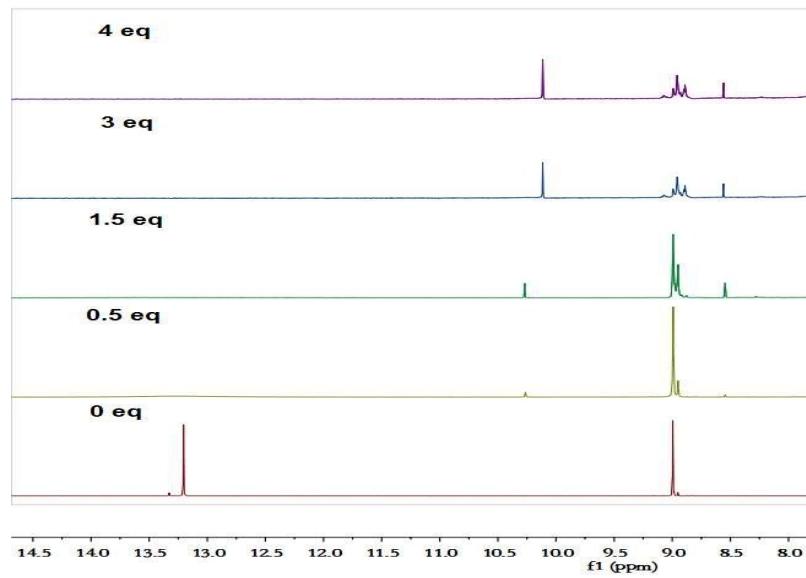
**Fig. S10**  $^1H$ -NMR,  $^{13}C$ -NMR and MS spectrogram (top to bottom) of Tris(4-aminophenyl)amine.



**Fig. S11**  $^1\text{H}$ -NMR,  $^{13}\text{C}$ -NMR and MS spectrogram (top to bottom) of TATP.



**Fig. S12**  $^1\text{H}$  NMR titration of TATP by adding different equivalents of  $\text{Zn}^{2+}$  into TATP solution in  $\text{DMSO}-\text{d}_6$ .



**Fig. S13** <sup>1</sup>H NMR titration of TATP by adding different equivalents of CN<sup>-</sup> into TATP solution in DMSO-d<sub>6</sub>.