Supporting information

A rhodamine-based sensor for Hg²⁺ and resultant complex as a fluorescence sensor for I⁻

Haichang Ding^{a,b}, Chunhong Zheng^b, Baoqiang Li^{a,*}, Gang Liu^b, Shouzhi Pu^{b,*}, Dechang Jia^a, Yu Zhou^a

^aInstitute for Advanced Ceramics; State Key Laboratory of Urban Water Resource and Environment, Harbin Institute of Technology, Harbin 150001, PR China

^bJiangxi Key Laboratory of Organic Chemistry, Jiangxi Science and Technology Normal

University, Nanchang 330013, PR China

*Corresponding author: libq@hit.edu.cn (B. Li), pushouzhi@tsinghua.org.cn (S. Pu)

Figures captions:

Fig. S1. ¹H NMR (in CDCl₃) and ¹³C NMR (in CDCl₃) spectrums of 1.

Fig. S2. ESI–MS spectrum of 1.

Fig. S3. The relationship between the absorbance and Hg^{2+} concentration, taken from the inset of Fig. 4A.

Fig. S4. The LOD for Hg²⁺ by absorbance.

Fig. S5. The relationship between the emission intensity and Hg^{2+} concentration, taken from the inset of Fig. 4B.

Fig. S6. The LOD for Hg²⁺ by fluorescence.

Fig. S7. The relationship between the emission intensity and I⁻ concentration, taken from the inset of Fig. 7.



Fig. S1. ¹H NMR (in CDCl₃) and ¹³C NMR (in CDCl₃) spectrums of 1.



Fig. S2. ESI–MS spectrum of 1.



Fig. S3. The relationship between the absorbance and Hg²⁺ concentration, taken from the inset of Fig. 4A.



Fig. S4. The LOD for Hg^{2+} by absorbance.



Fig. S5. The relationship between the emission intensity and Hg^{2+} concentration, taken from the inset of Fig. 4B.



Fig. S6. The LOD for Hg^{2+} by fluorescence.



Fig. S7. The relationship between the emission intensity and I⁻ concentration, taken from the inset of Fig. 7.