

# Electronic Supplementary Information (ESI)

## Aryl-Fused Tetrathianaphthalene (TTN): Synthesis, Structures, Properties, and Cocrystals with Fullerenes

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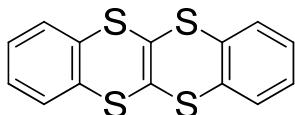
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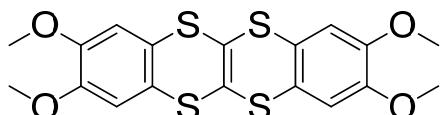
## I. Characterization of the compounds

The following are the appearance, yield, melting point (Mp),  $^1\text{H}$  NMR,  $^{19}\text{F}$  NMR,  $^{13}\text{C}$  NMR, and high-resolution mass spectra (HRMS) for the compounds reported in this work.



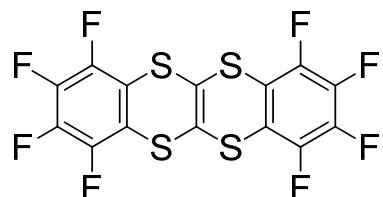
**TTN-1**

White solid(Yield = 63%); Mp: 249.5-251.0 °C.  $^1\text{H}$  NMR (400 MHz, CDCl<sub>3</sub>) δ 7.41-7.38 (m, 4H), 7.30-7.28 (m, 4H).  $^{13}\text{C}$  NMR (100 MHz, CDCl<sub>3</sub>) δ 135.65, 128.64, 128.38, 124.48. HRMS (C<sub>14</sub>H<sub>8</sub>S<sub>4</sub> + H): Calc. 304.9582, found 304.9583.



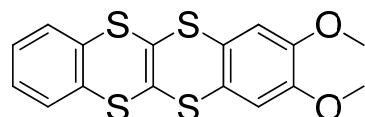
**TTN-2**

White solid( Yield = 73% ); Mp: 309.7-311.2 °C.  $^1\text{H}$  NMR (400 MHz, CDCl<sub>3</sub>) δ 6.87 (s, 4H), 3.87 (s, 12H). HRMS (C<sub>18</sub>H<sub>16</sub>O<sub>4</sub> S<sub>4</sub>): Calc. 423.9926, found 423.9916.



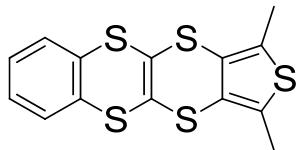
**TTN-4**

White solid ( Yield = 70%); Mp: 226.8-228.6 °C.  $^{19}\text{F}$  NMR (376 MHz, CDCl<sub>3</sub>) δ -132.60--132.65 (m,4F), -152.19--152.24 (m, 4F). HRMS (C<sub>14</sub>F<sub>8</sub>S<sub>4</sub>): Calc. 447.8750, found 447.8749.



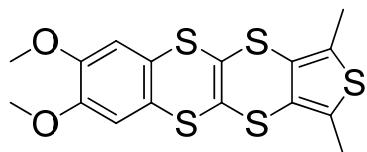
**TTN-5**

White solid( Yield = 78%). Mp: 222.0-224.0 °C.  $^1\text{H}$  NMR (400 MHz, CDCl<sub>3</sub>) δ 7.38 (dd, J = 5.8, 3.4 Hz, 1H), 7.28 (dd, J = 5.8, 3.3 Hz, 1H), 6.90 (s, 1H), 3.89 (s, 3H).  $^{13}\text{C}$  NMR (100 MHz, CDCl<sub>3</sub>) δ 149.37 (s), 135.72 (s), 128.61 (s), 128.36 (s), 126.96 (s), 125.46 (s), 111.53 (s), 56.23 (s). HRMS (C<sub>16</sub>H<sub>12</sub>O<sub>2</sub>S<sub>4</sub>): Calc. 363.9715, found 363.9706.



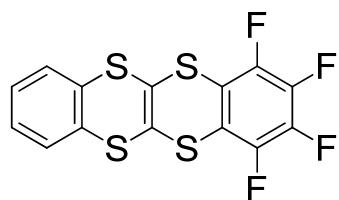
**TTN-6**

White solid( Yield = 81%); Mp: 251.0-252.8 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.39 (dd,  $J$  = 5.8, 3.4 Hz, 2H), 7.29 (dd,  $J$  = 5.8, 3.3 Hz, 4H), 2.33 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  136.12, 129.72, 128.54, 128.34, 127.41, 124.78, , 13.66. HRMS ( $\text{C}_{14}\text{H}_{10}\text{S}_5$ ): Calc. 337.9381, found 337.9372.



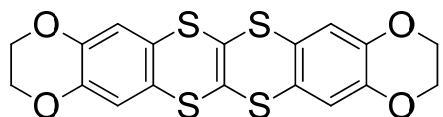
**TTN-7**

White solid( Yield = 61%). Decompositon point: 273°C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  6.88 (s, 1H), 3.87 (s, 3H), 2.31 (s, 3H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  149.36, 129.62, 127.47, 127.45, 125.66, 111.39, 56.23, 13.65. HRMS ( $\text{C}_{16}\text{H}_{14}\text{O}_2\text{S}_5$ ): Calc. 397.9592, found 397.9581.



**TTN-8**

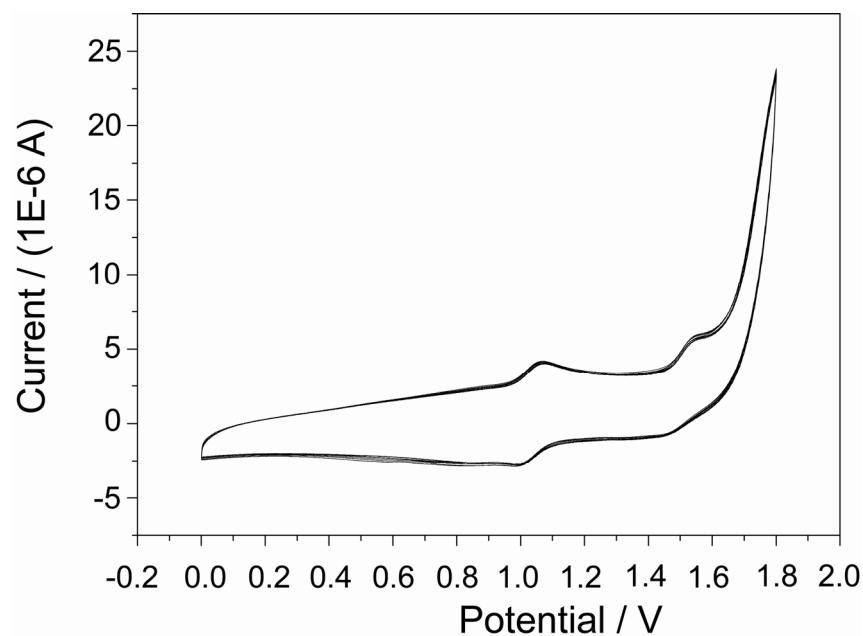
White solid ( Yield = 71% ); Mp 237.7-238.5 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40-7.38 (m, 2H), 7.32-7.30 (m, 2H).  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -133.27--133.32 (m, 2F), -153.33--153.38(m, 2F). HRMS ( $\text{C}_{14}\text{H}_4\text{F}_4\text{S}_4$ ): Calc. 375.9126, found 375.9119.



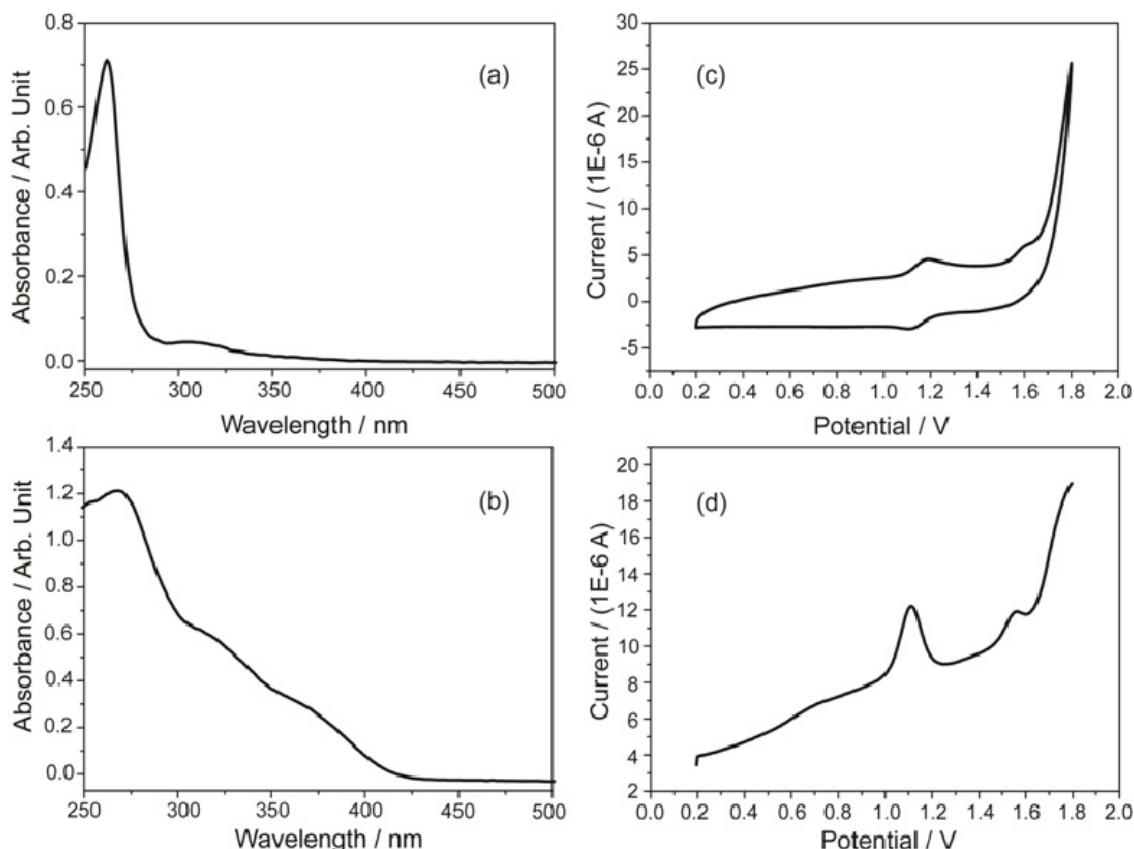
**TTN-9**

White solid ( Yield = 71% ); Mp > 310 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) 6.89 (s, 1H), 4.24 (s, 2H). HRMS ( $\text{C}_{18}\text{H}_{12}\text{O}_4\text{S}_4$ ): Calc. 419.9613, found 419.9623.

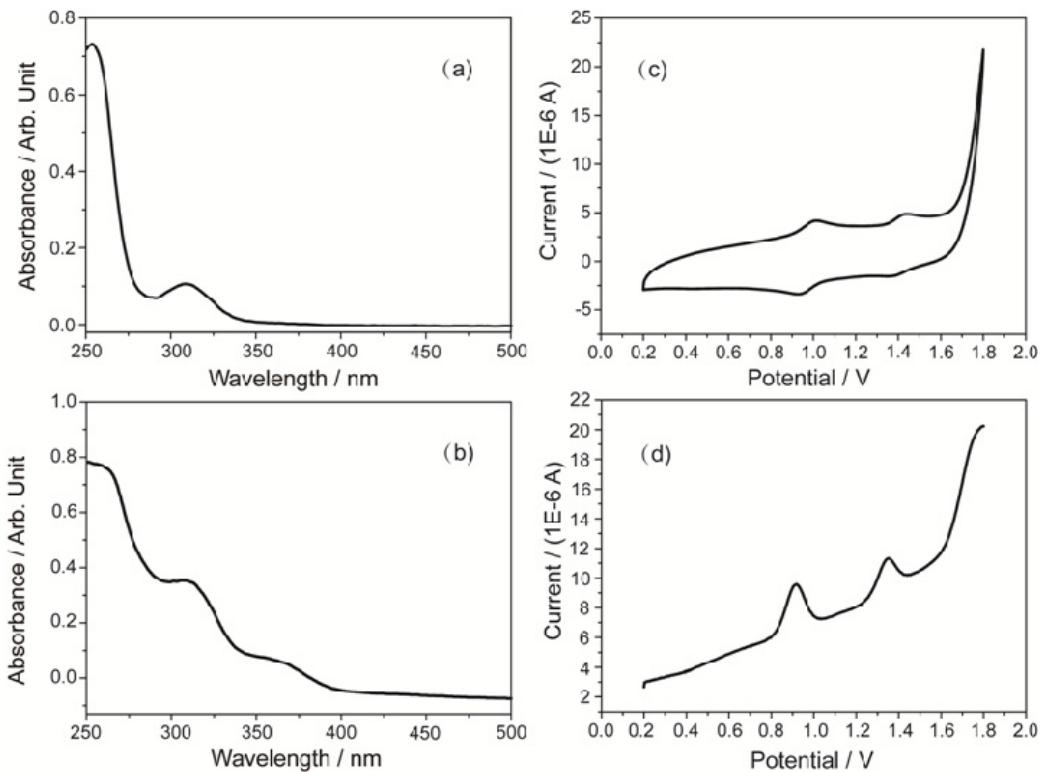
## II. UV/Vis absorption spectra and cyclic voltammograms (CV) of TTNs



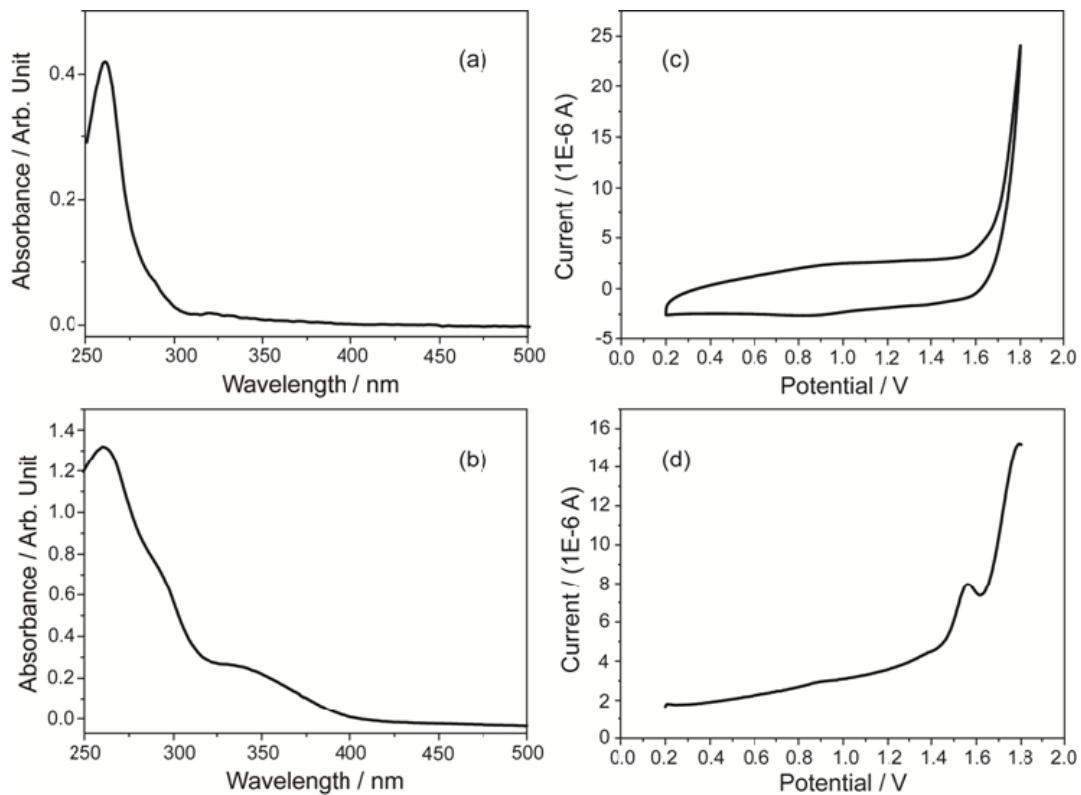
**Fig. S1** Repeated cyclic voltammograms (20 times) of TTN3



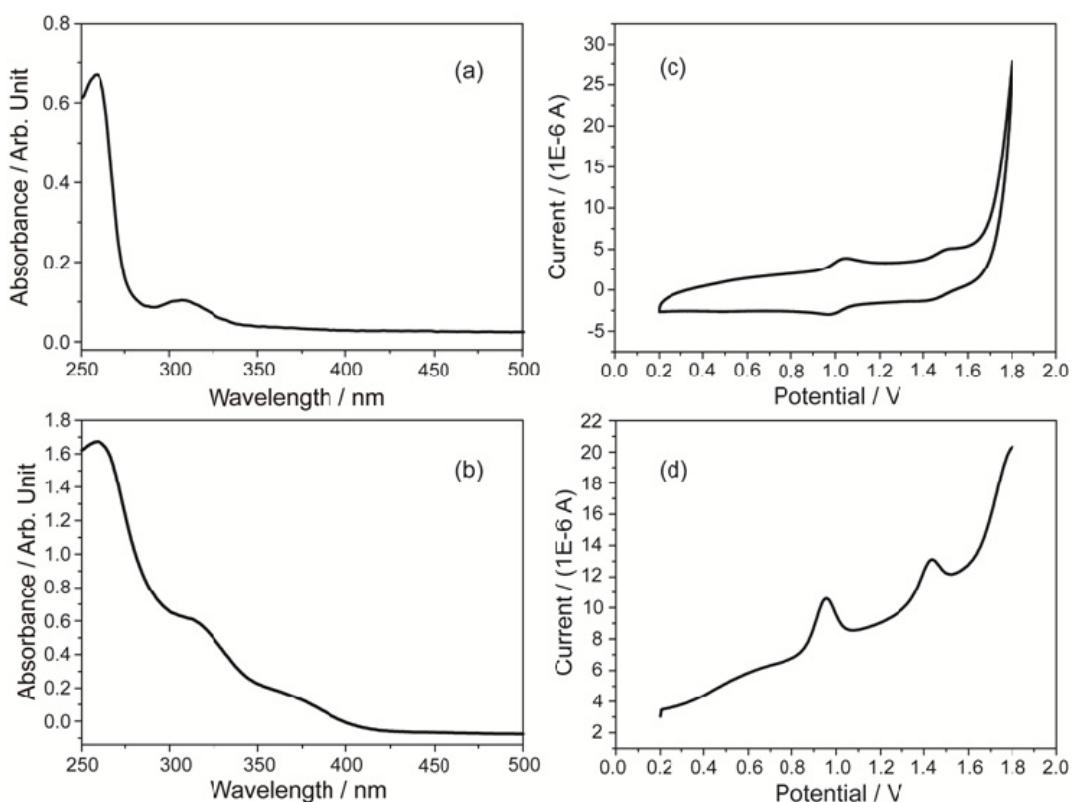
**Fig. S2** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN1



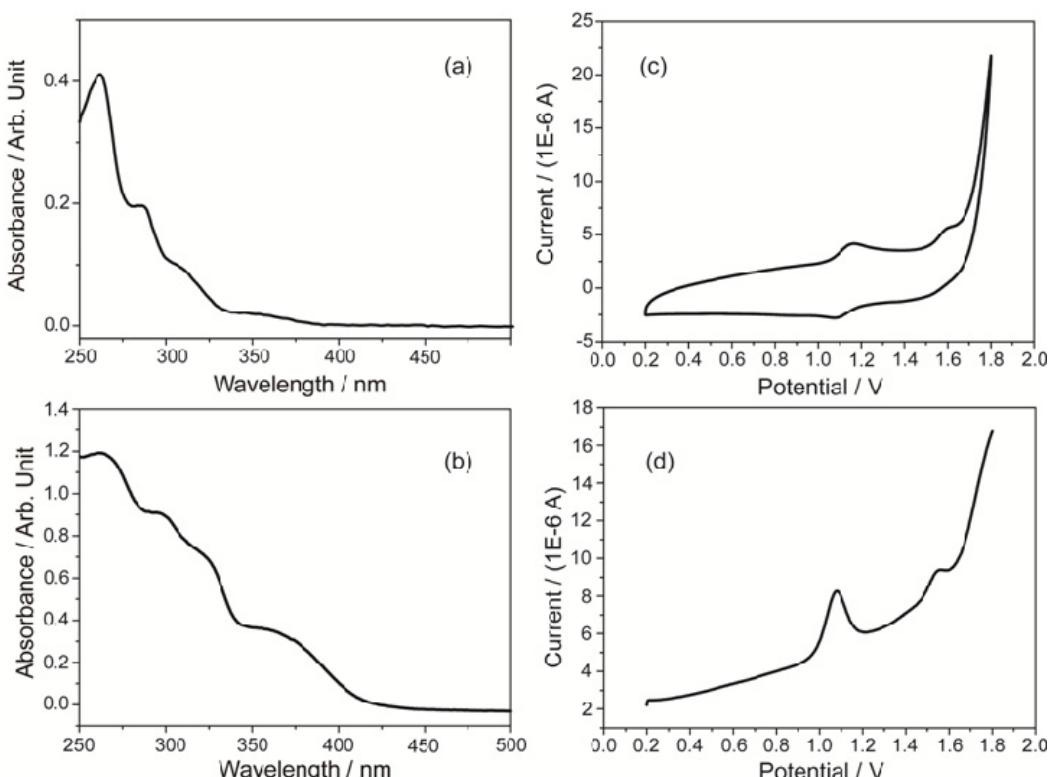
**Fig. S3** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN2



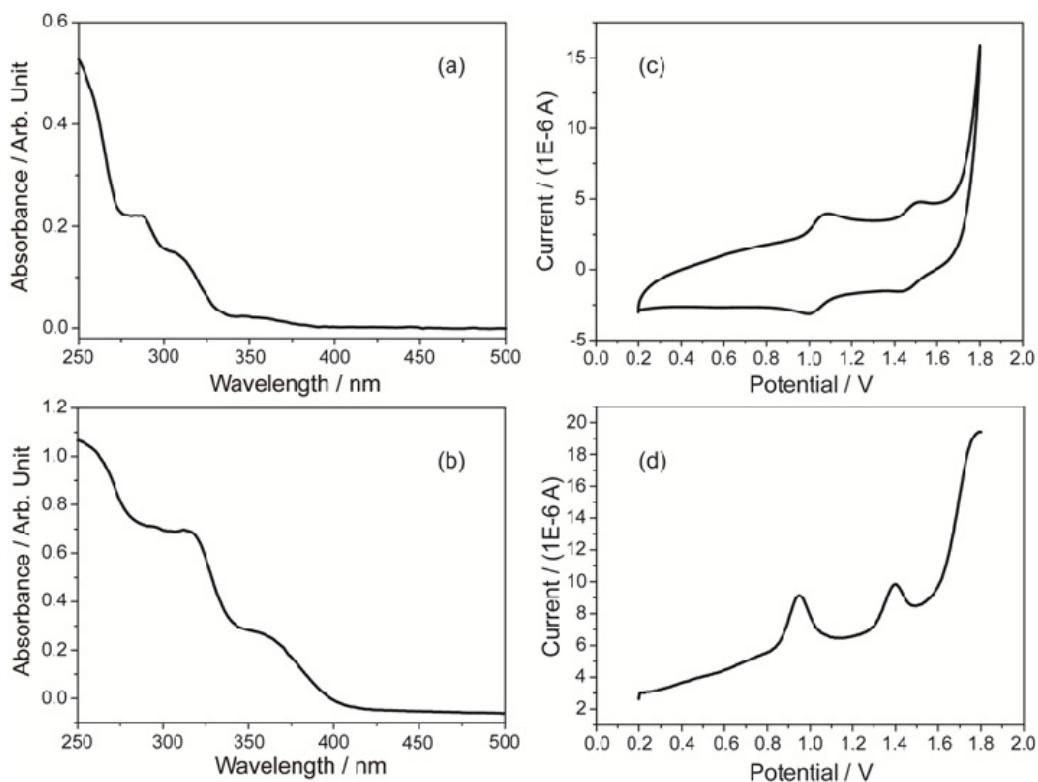
**Fig. S4** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN4



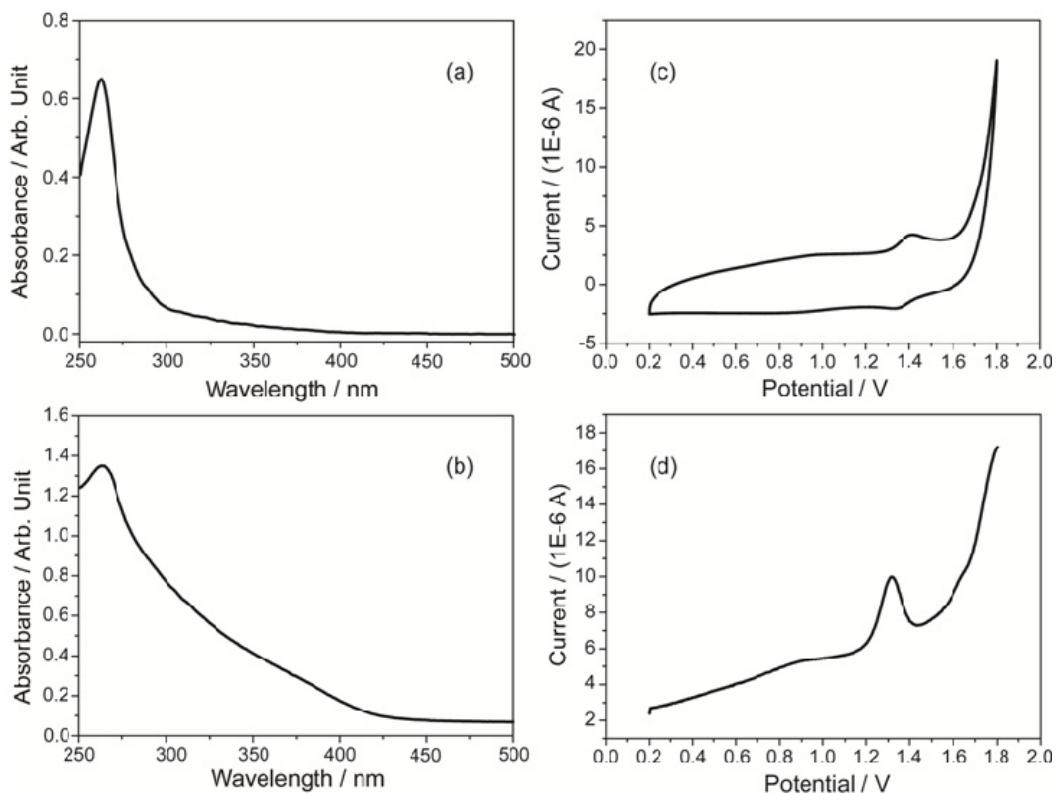
**Fig. S5** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of **TTN5**



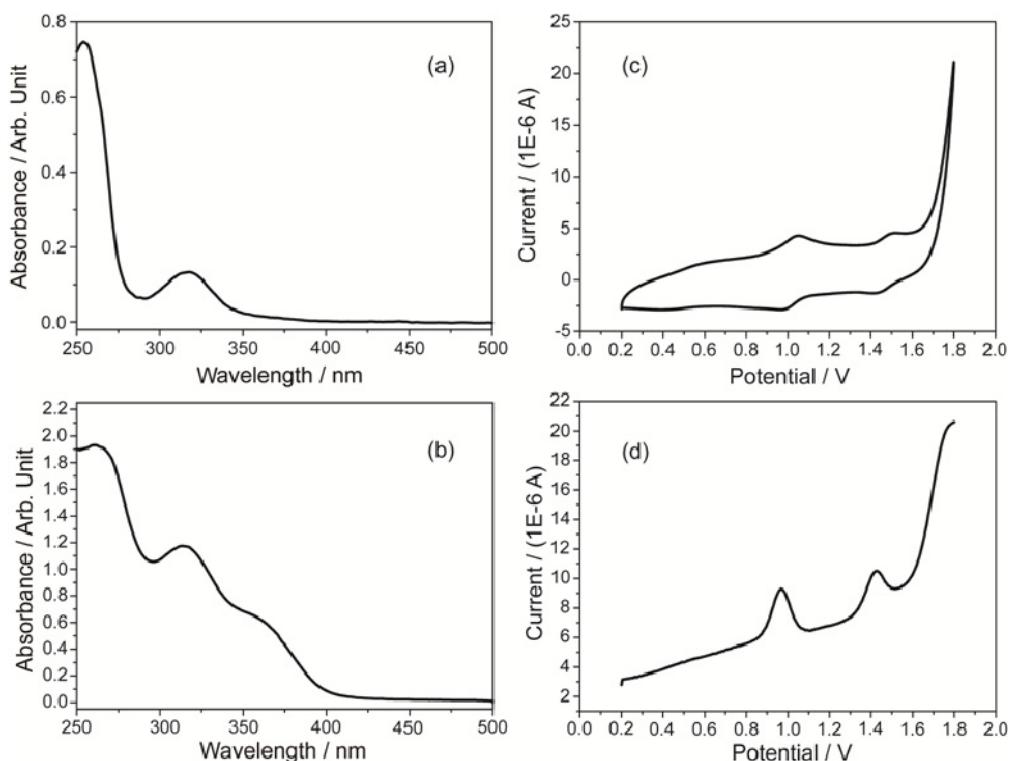
**Fig. S6** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of **TTN6**



**Fig. S7** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN7

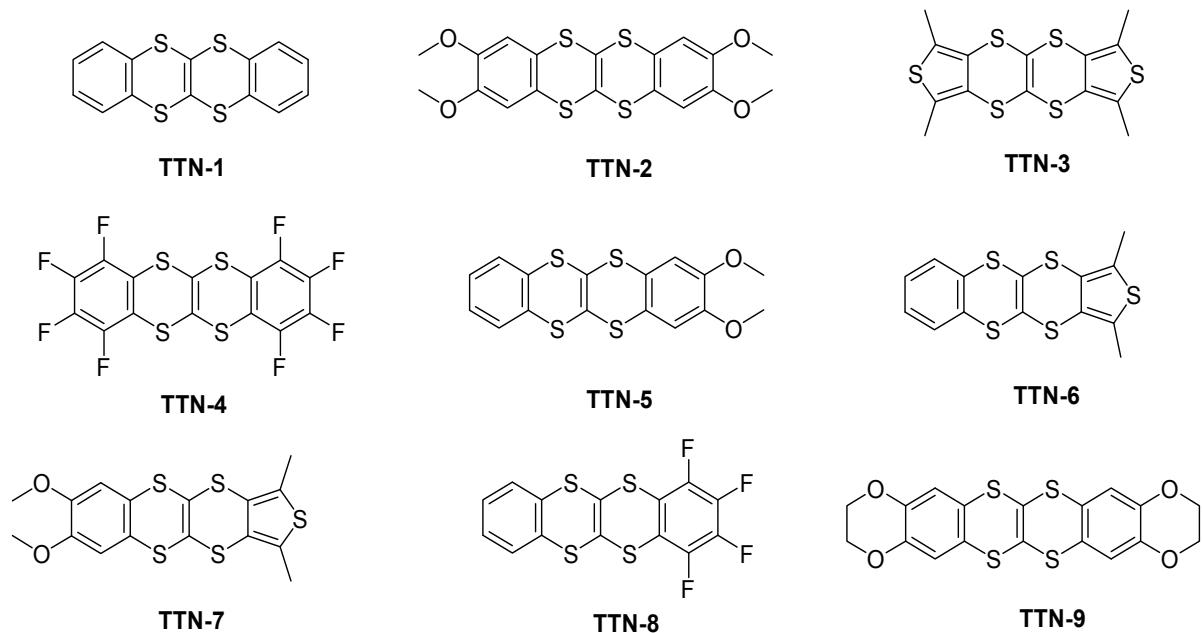


**Fig. S8** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN8



**Fig. S9** UV-Vis spectra in DCM solution (a), solid state absorption spectra (b), cyclic voltammogram (c) and differential pulse voltammetry (d) of TTN9

### III. Calculated frontier orbitals for TTNs



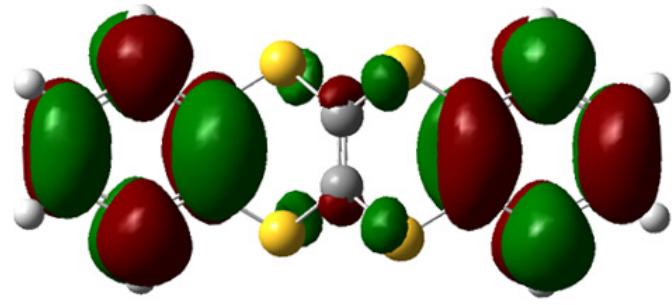
**Scheme S1.** Chemical structures of TTNs for theoretical calculation.

**Table S1** Calculated energy levels

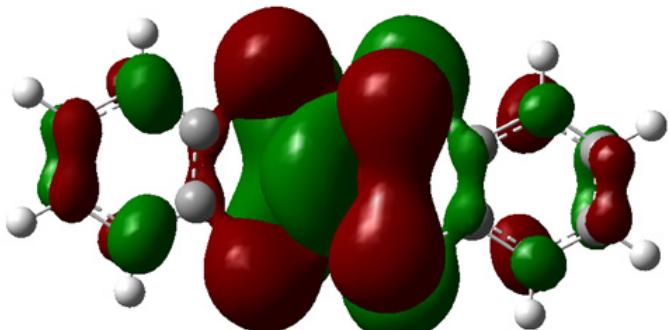
	Energy / eV		
	LUMO	HOMO	$\Delta E^{[b]}$
<b>TTN1<sup>[a]</sup></b>	-0.955	-5.349	4.394
<b>TTN2<sup>[a]</sup></b>	-0.607	-4.949	4.342
<b>TTN3<sup>[a]</sup></b>	-0.868	-5.264	4.396
<b>TTN4<sup>[a]</sup></b>	-1.724	-6.061	4.337
<b>TTN5<sup>[c]</sup></b>	-0.853 <sup>[c]</sup>	-5.120 <sup>[c]</sup>	4.267
<b>TTN6<sup>[a]</sup></b>	-0.975	-5.304	4.329
<b>TTN7<sup>[c]</sup></b>	-0.828 <sup>[c]</sup>	-5.183 <sup>[c]</sup>	4.355
<b>TTN8<sup>[c]</sup></b>	-1.417 <sup>[c]</sup>	-5.702 <sup>[c]</sup>	4.285
<b>TTN9<sup>[a]</sup></b>	-0.646	-5.046	4.400
<b>TTF3<sup>[a]</sup></b>	-1.000	-4.811	3.811

<sup>[a]</sup> Calculation was conducted on the basis of the molecular geometry in the crystal structures.

<sup>[b]</sup> HOMO–LUMO energy gap  $\Delta E = E_{\text{LUMO}} - E_{\text{HOMO}}$ . <sup>[c]</sup> Calculation was conducted on the basis of optimized molecular geometry.

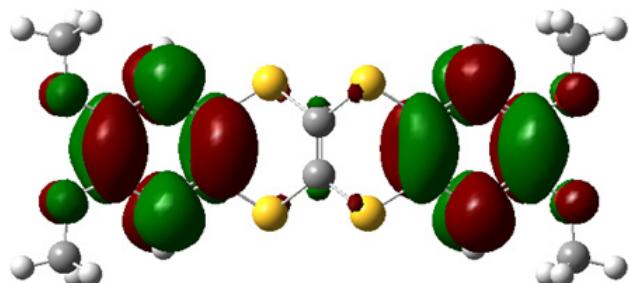


LUMO: -0.955 eV

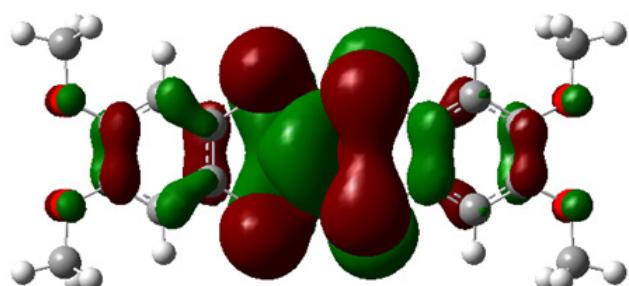


HOMO: -5.349 eV

**Fig. S10** HOMO and LUMO orbitals of TTN1

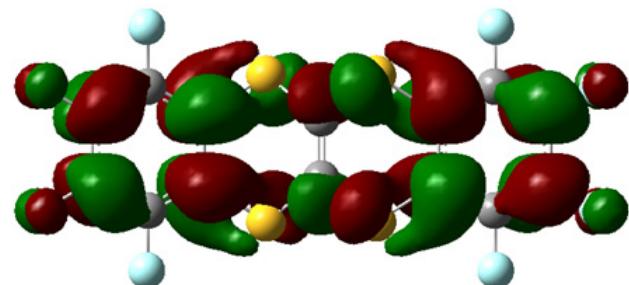


LUMO: -0.607 eV

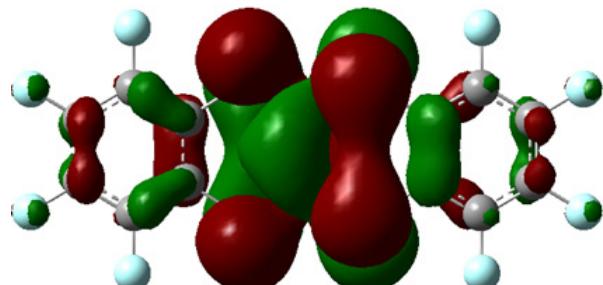


HOMO: -4.949 eV

**Fig. S11** HOMO and LUMO orbitals of TTN2

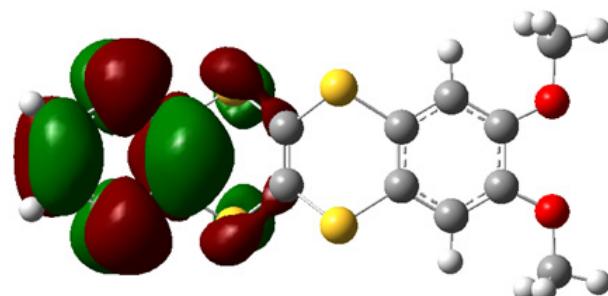


LUMO: -1.724 eV

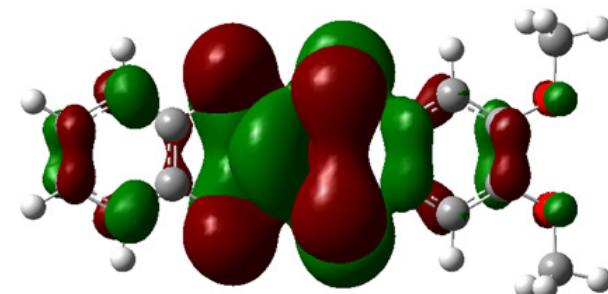


HOMO: -6.061 eV

**Fig. S12** HOMO and LUMO orbitals of TTN4

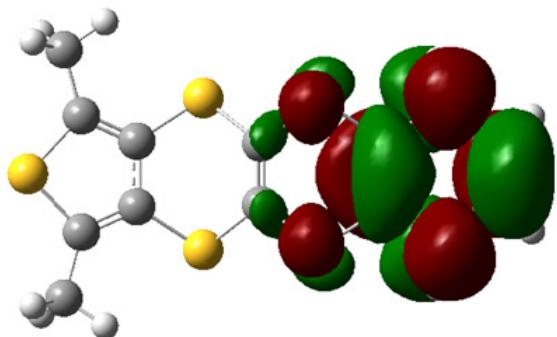


LUMO: -0.853 eV

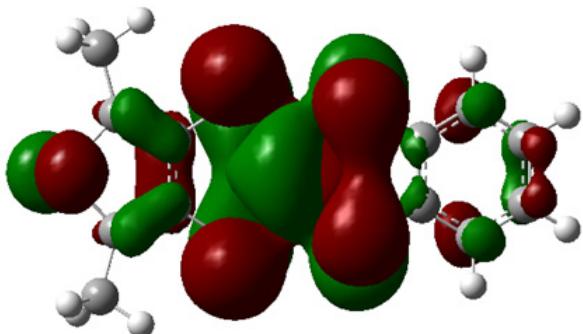


HOMO: -5.120 eV

**Fig. S13** HOMO and LUMO orbitals of TTN5

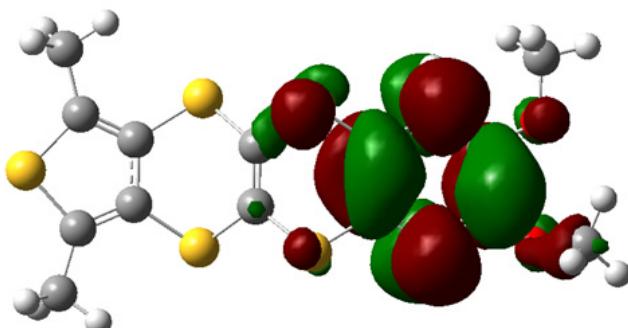


LUMO: -0.975 eV

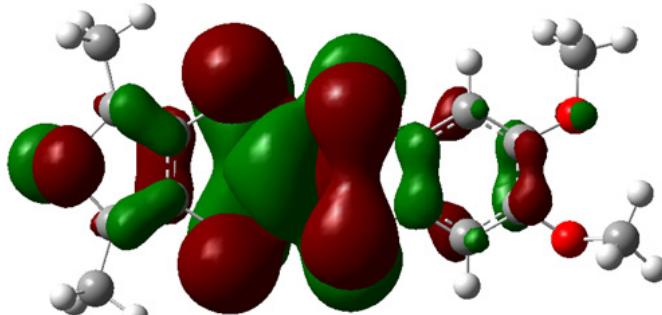


HOMO: -5.304 eV

**Fig. S14** HOMO and LUMO orbitals of TTN6

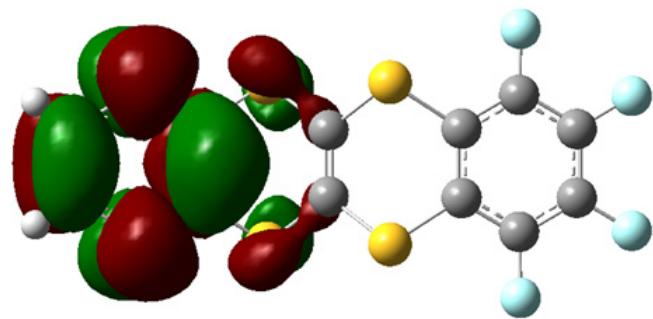


LUMO: -0.828 eV

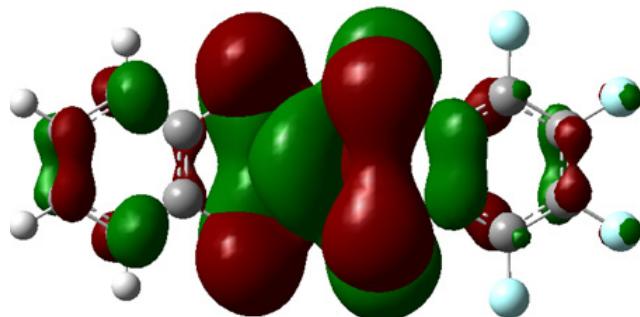


HOMO: -5.183 eV

**Fig. S15** HOMO and LUMO orbitals of TTN7

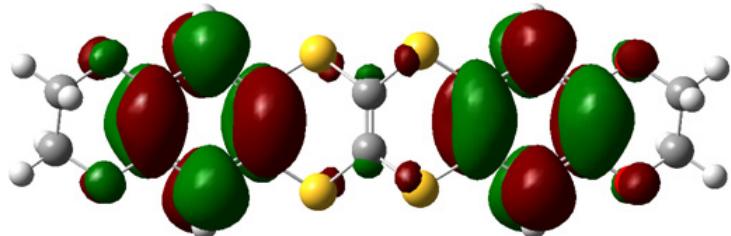


LUMO: -1.417 eV

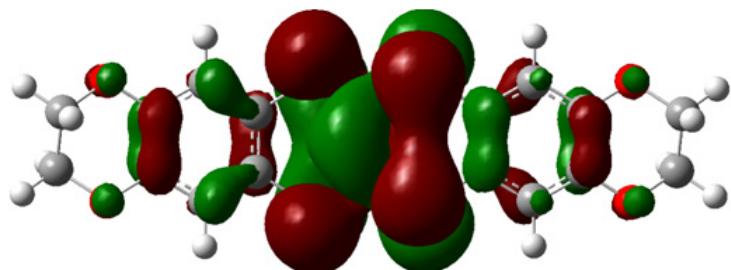


HOMO: -5.702 eV

**Fig. S16** HOMO and LUMO orbitals of TTN8

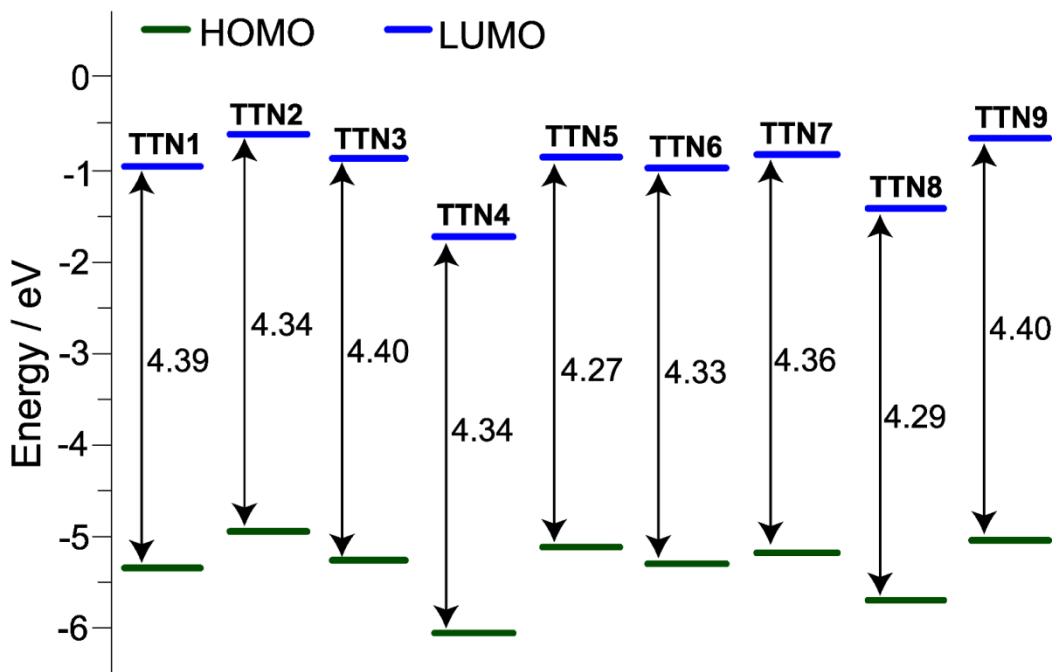


LUMO: -0.646 eV



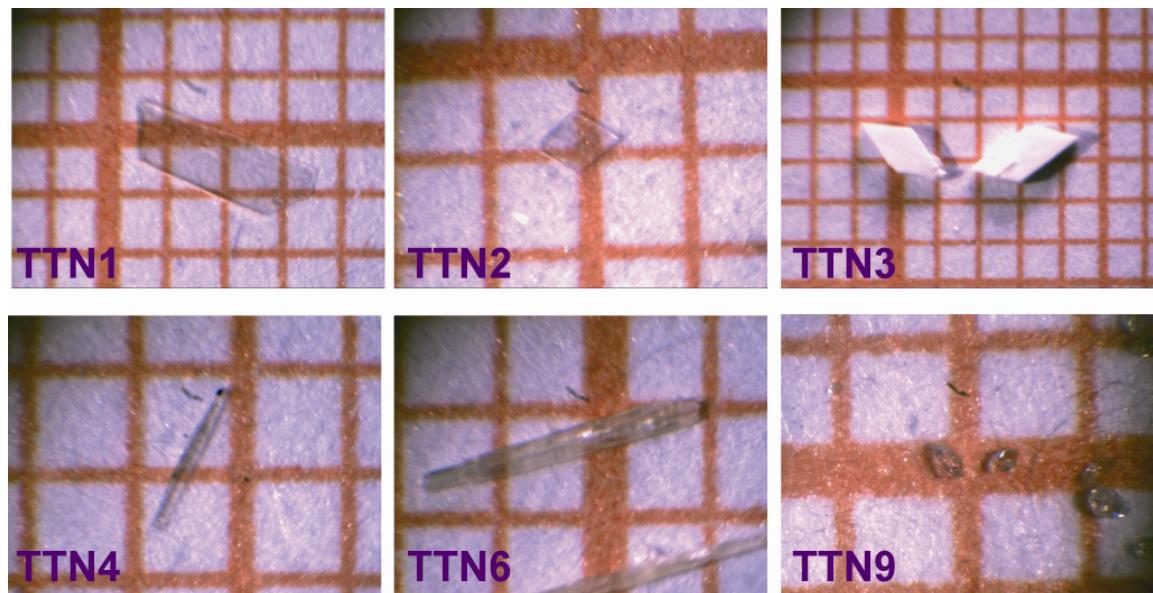
HOMO: -5.046 eV

**Fig. S17** HOMO and LUMO orbitals of TTN9



**Fig. S18** Comparison of energy levels for TTN1–TTN9.

#### IV. Selected crystallographic data of TTNs

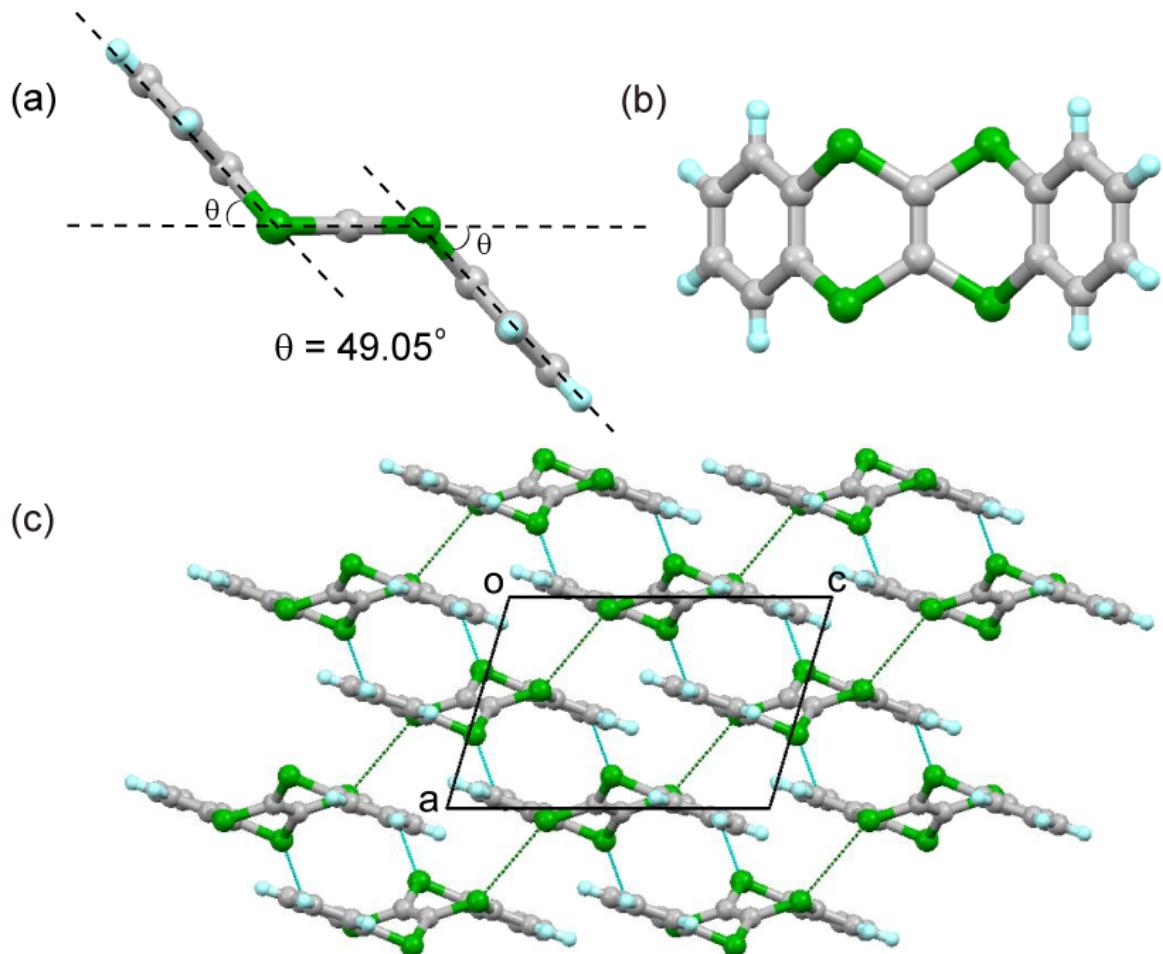


**Fig. S19** Photographs for the single crystals of TTNs. The grid for the background of each figure is 1 mm × 1mm.

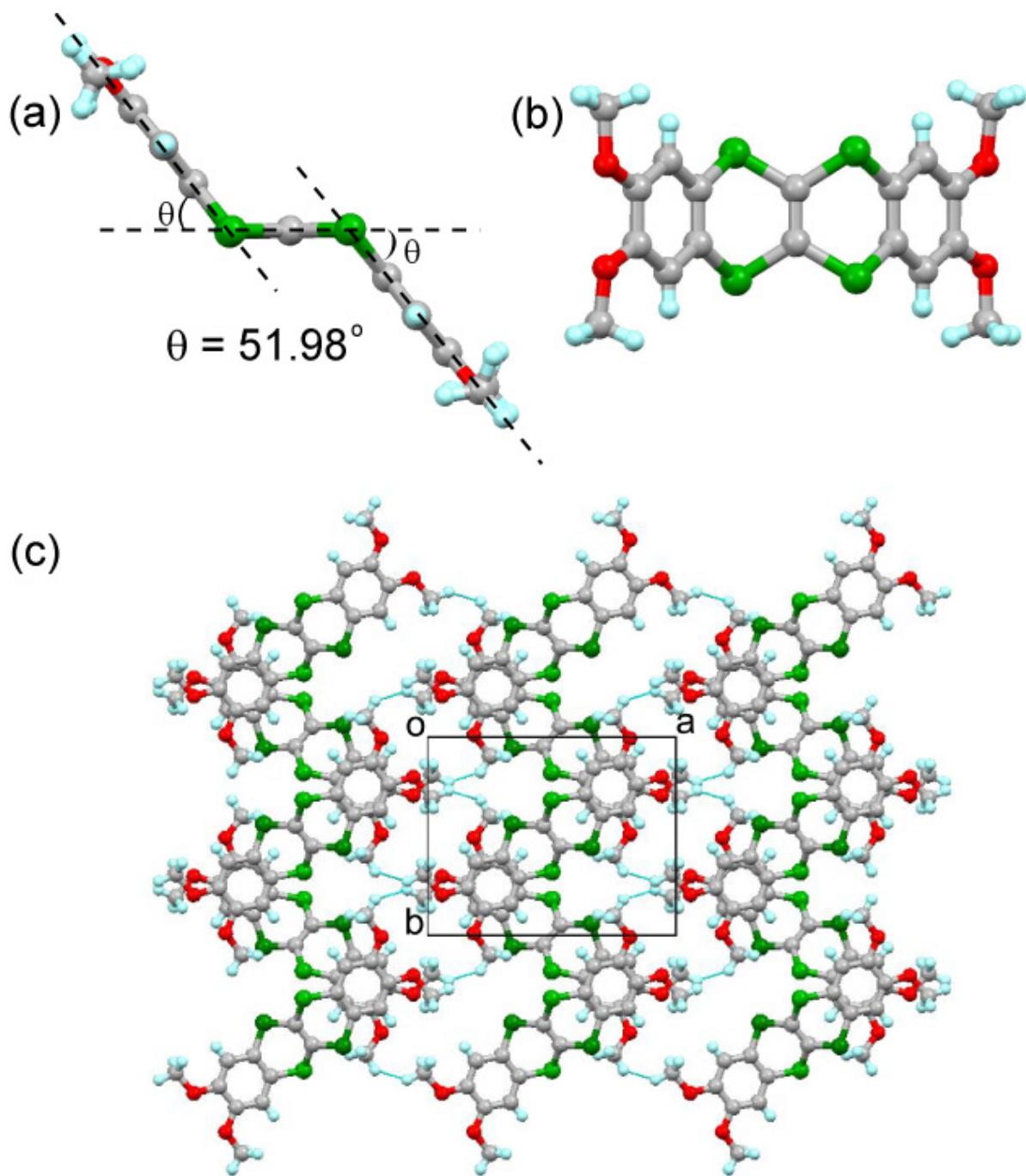
**Table 2** Selected crystallographic data for TTNs and TTF3

	<b>TTN1</b>	<b>TTN2</b>	<b>TTN3•CS<sub>2</sub></b>	<b>TTN4</b>	<b>TTN6</b>	<b>TTN9</b>	<b>TTF3</b>
CCDC number	1492704	1492705	1492706	1492707	1492708	1492710	1492709
Empirical formula	C <sub>14</sub> H <sub>8</sub> S <sub>4</sub>	C <sub>18</sub> H <sub>16</sub> O <sub>4</sub> S <sub>4</sub>	C <sub>15</sub> H <sub>12</sub> S <sub>8</sub>	C <sub>14</sub> F <sub>8</sub> S <sub>4</sub>	C <sub>14</sub> H <sub>10</sub> S <sub>5</sub>	C <sub>18</sub> H <sub>12</sub> O <sub>4</sub> S <sub>4</sub>	C <sub>14</sub> H <sub>12</sub> S <sub>6</sub>
Formula weight	304.44	424.54	448.72	448.38	338.52	420.52	372.60
Temperature / K	293(2)	293(2)	293(2)	296(2)	293(2)	293(2)	289.9
$\lambda$ / Å	0.71073	0.71073	1.54184	0.71073	0.71073	0.71073	1.5418
Crystal size / mm <sup>3</sup>	0.37×0.34×0.23	0.36×0.26×0.24	0.33×0.24×0.11	0.28×0.16×0.07	0.27×0.08×0.06	0.14×0.11×0.05	0.27×0.09×0.07
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic	Triclinic	Monoclinic	Monoclinic
space group	<i>P</i> 2/n	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> -1	<i>P</i> 2 <sub>1</sub> /n	<i>I</i> 2/a
<i>a</i> / Å	7.4504(9)	12.6610(5)	8.8305(6)	12.361(19)	7.4016(9)	7.8871(15)	17.1106(5)
<i>b</i> / Å	8.4459(9)	10.1047(5)	7.1569(6)	3.954(6)	7.7951(9)	11.518(2)	7.1272(2)
<i>c</i> / Å	10.8382(13)	7.3307(3)	15.0030(9)	15.050(2)	12.6647(11)	18.586(3)	13.5589(4)
$\alpha$ / °	90	90	90	90	83.801(8)	90	90
$\beta$ / °	106.345(13)	96.481(4)	91.764(6)	97.159(15)	83.450(9)	90	104.115(3)
$\gamma$ / °	90	90	90	90	77.253(10)	90	90
<i>V</i> / Å <sup>3</sup>	654.43(13)	928.18(7)	947.73(11)	729.9(19)	705.88(14)	1688.5(5)	1603.58(8)
<i>Z</i>	2	2	2	2	2	4	4
<i>d</i> <sub>calc</sub> / g·cm <sup>-3</sup>	1.545	1.519	1.572	2.040	2.389	1.654	1.543
$\mu$ / mm <sup>-1</sup>	0.701	0.533	8.677	0.740	1.202	0.586	0.838
2θ <sub>max</sub> / °	57.16	56.7	141.56	51.3	56.9	57.26	51.22
Data / restraints / parameters	1484 / 0 / 82	2093 / 0 / 120	1803 / 0 / 108	1373 / 0 / 118	3189 / 0 / 174	3737 / 0 / 235	1495/0/94
<i>GooF</i>	1.076	1.044	1.036	1.024	1.071	1.029	1.094
<i>R</i> [ <i>I</i> >2σ( <i>I</i> )]	0.0505	0.0369	0.0547	0.0428	0.0472	0.0598	0.0528
<i>wR</i> <sub>2</sub>	0.1097	0.0874	0.1416	0.0976	0.1016	0.1292	0.1535

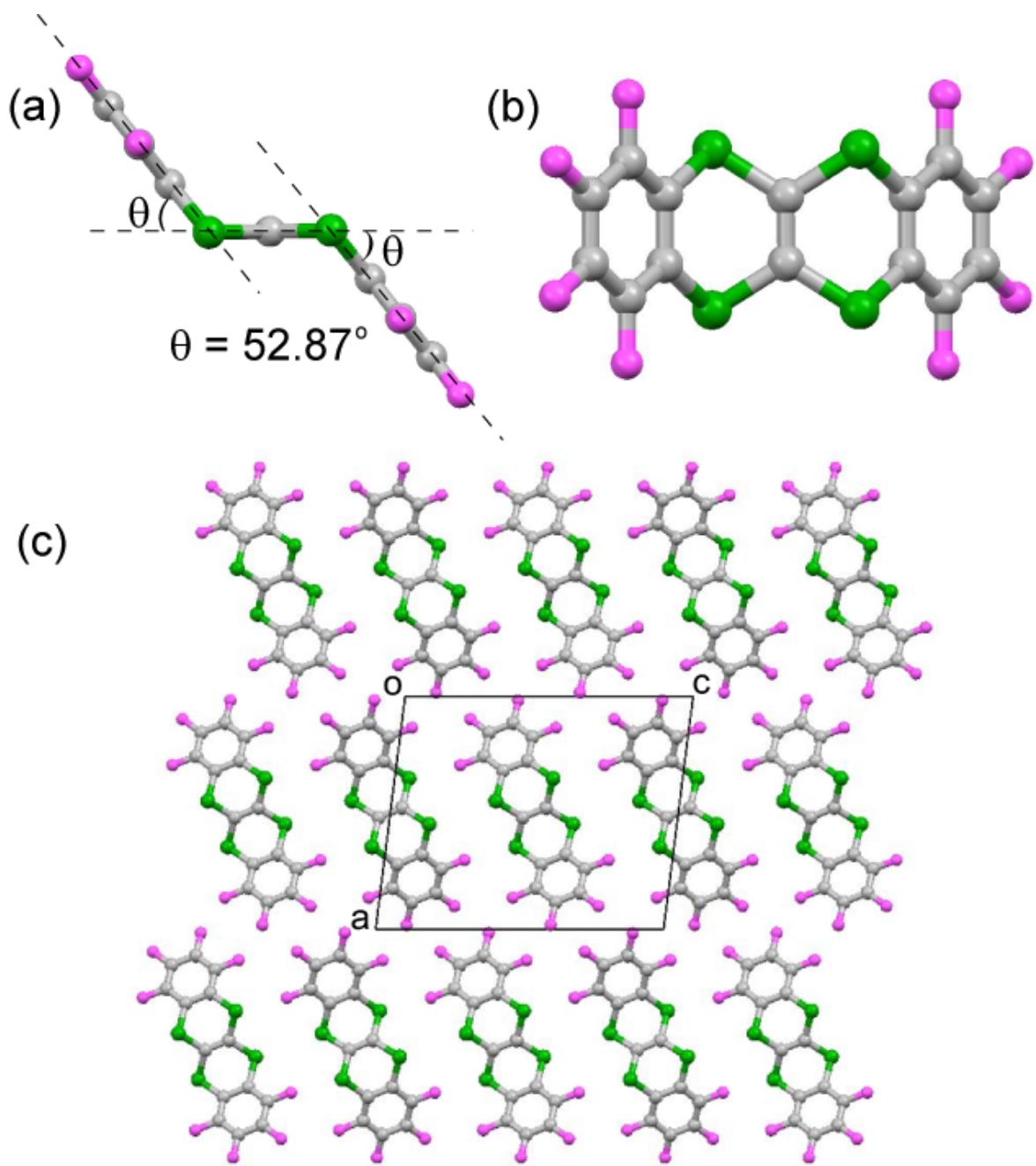
The following are the crystal structures of **TTN1**, **TTN2**, **TTN4**, **TTN6**, and **TTN9**. In these figures, panels (a), (b), and (c) depict the side view, top view, and packing structure of the corresponding TTN, respectively. In general, the grey, green, pink, and pale cyan balls represent carbon, sulfur, fluorine, and hydrogen atoms, respectively.



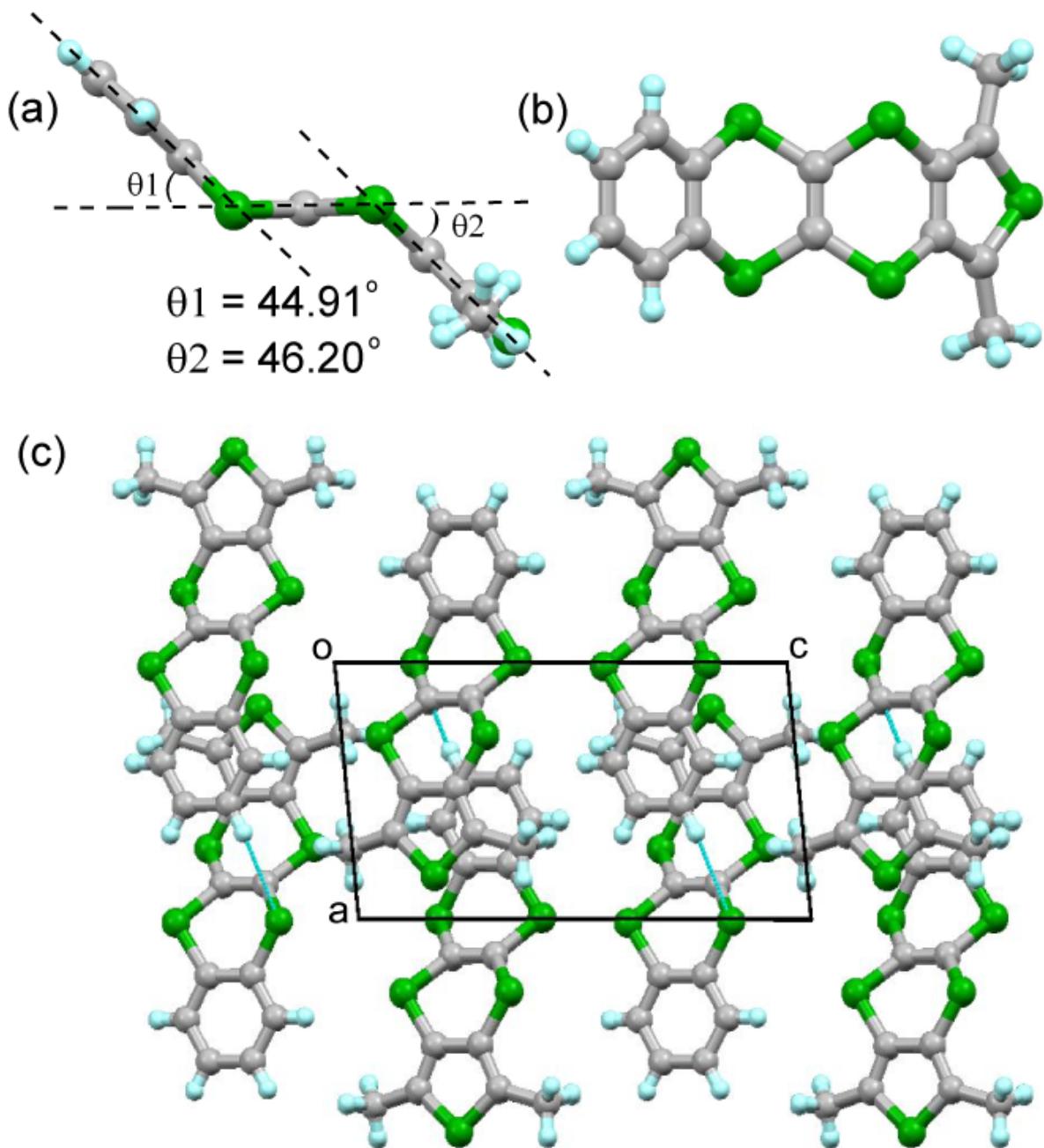
**Fig. S20** Crystal structure of **TTN1**: (a), (b), and (c) depict the side view, top view, and packing structures, respectively.



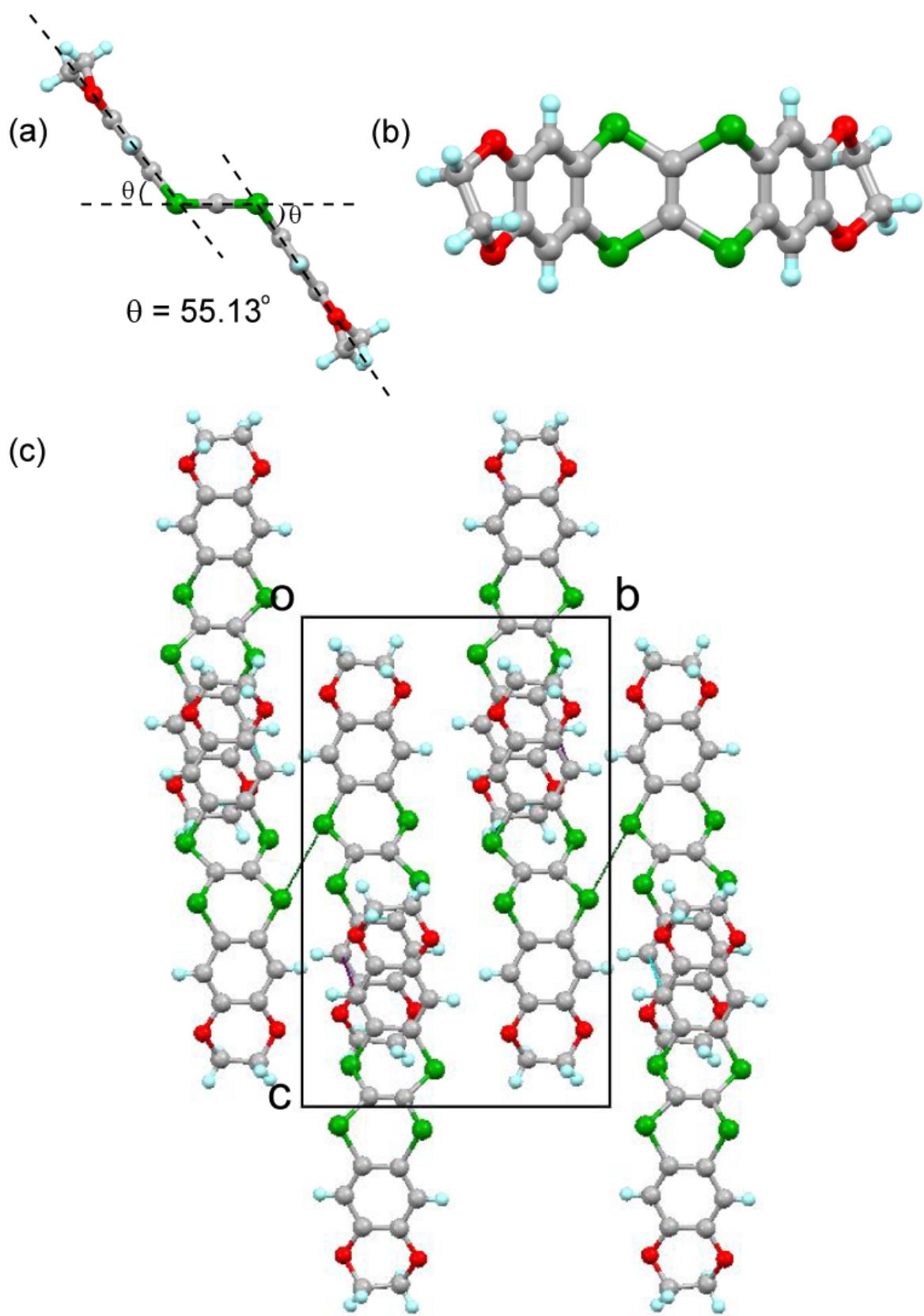
**Fig. S21** Crystal structure of TTN2: (a), (b), and (c) depict the side view, top view, and packing structures, respectively.



**Fig. S22** Crystal structure of TTN4: (a), (b), and (c) depict the side view, top view, and packing structures, respectively.



**Fig. S23** Crystal structure of TTN6: (a), (b), and (c) depict the side view, top view, and packing structures, respectively.



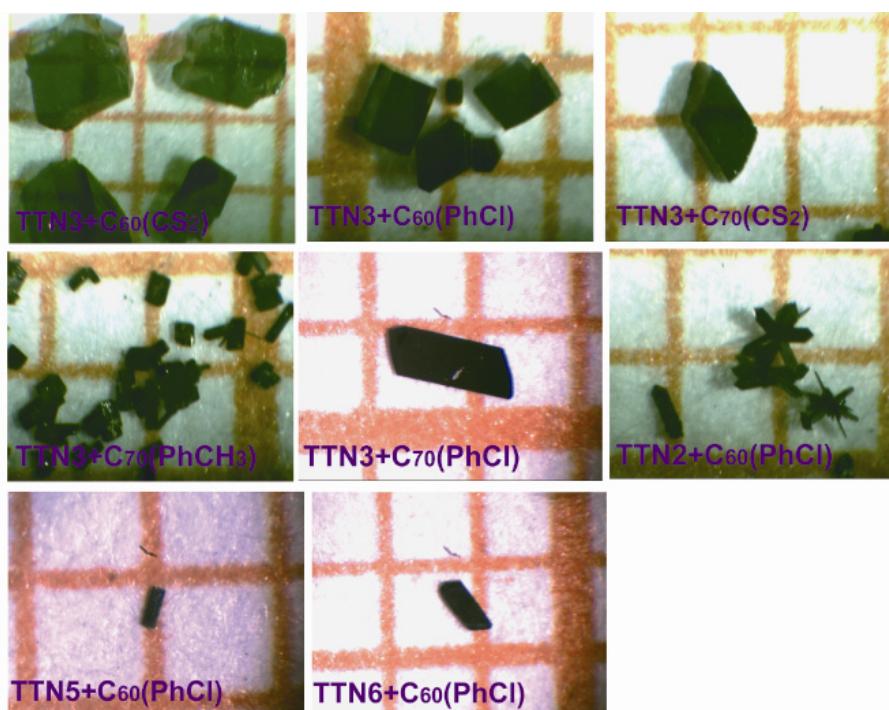
**Fig. S24** Crystal structure of TTN9: (a), (b), and (c) depict the side view, top view, and packing structures, respectively.

## V. Crystallographic data of “TTN•fullerene” cocrystals

The samples for the crystallographic analyses of the complexes were obtained by slow evaporation of the solvent. **TTN3•C<sub>60</sub>•CS<sub>2</sub>** was selected as an example. It was prepared by slow evaporation of the mixed solution of **1** (7.4 mg) and C<sub>60</sub> (7.2 mg) in CS<sub>2</sub> (10 ml) at room temperature (RT). After 5 days, the black block single crystals were formed.

**Table S3** Experimental conditions for the preparation of “TTN•fullerene” cocrystals

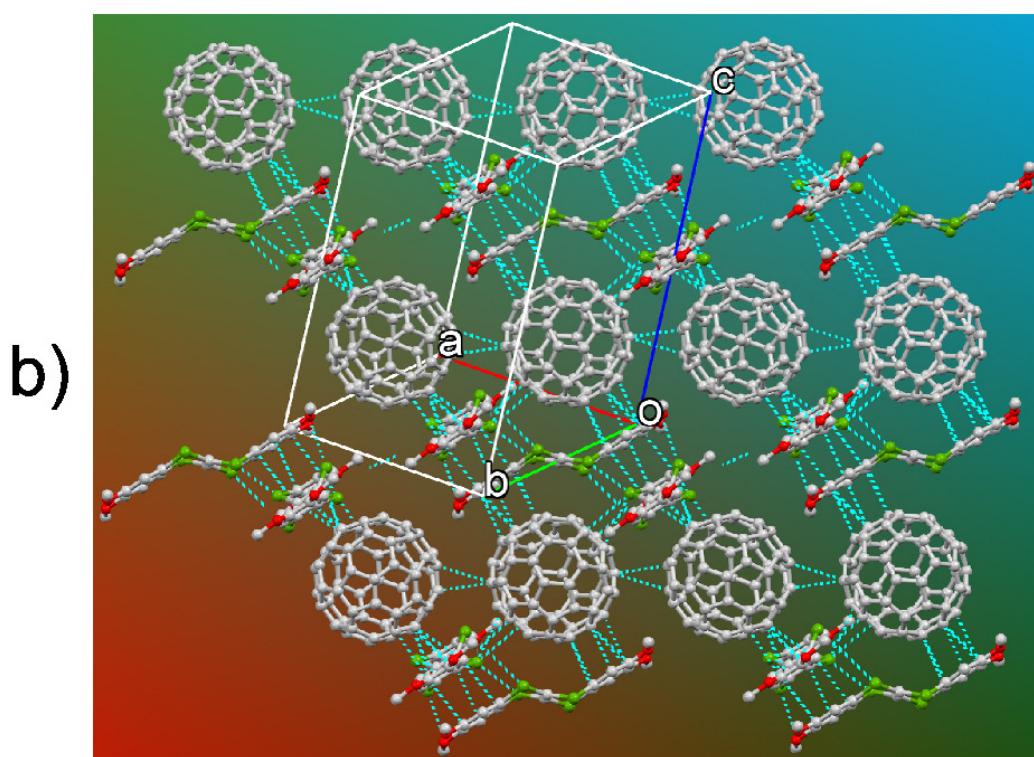
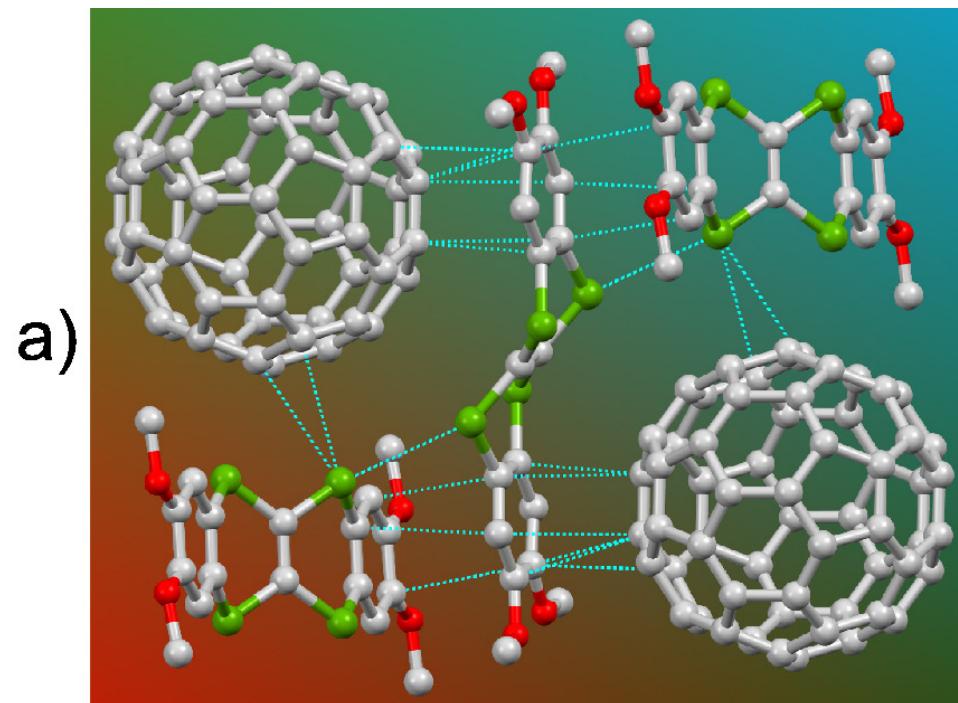
TTN	Fullerene	Solvent	Appearance	Composition
<b>TTN3</b> , 7.4 mg	C <sub>60</sub> , 7.2 mg	CS <sub>2</sub> , 10 mL	Black block	(TTN3)•(C <sub>60</sub> )•(CS <sub>2</sub> )
<b>TTN3</b> , 7.4 mg	C <sub>60</sub> , 7.2 mg	PhCl, 10 mL	Black block	(TTN3)•(C <sub>60</sub> )
<b>TTN3</b> , 7.4 mg	C <sub>70</sub> , 8.4 mg	CS <sub>2</sub> , 10 mL	Black block	(TTN3)•(C <sub>70</sub> )•(CS <sub>2</sub> )
<b>TTN3</b> , 7.4 mg	C <sub>70</sub> , 8.4 mg	PhCH <sub>3</sub> , 10 mL	Black block	Not determined
<b>TTN3</b> , 14.4 mg	C <sub>70</sub> , 16.8 mg	PhCl, 16 mL	Black block	(TTN3)•(C <sub>70</sub> )
<b>TTN2</b> , 8.5 mg	C <sub>60</sub> , 7.2 mg	PhCl, 10 mL	Black block	(TTN2)•(C <sub>60</sub> )
<b>TTN4</b> , 8.8 mg	C <sub>60</sub> , 7.2 mg	PhCl, 10 mL	Black block	Not determined
<b>TTN5</b> , 7.2 mg	C <sub>60</sub> , 7.2 mg	PhCl, 10 mL	Black block	Not determined
<b>TTN6</b> , 6.8 mg	C <sub>60</sub> , 7.2 mg	PhCl, 10 mL	Black block	(TTN6) <sub>2</sub> •(C <sub>60</sub> )



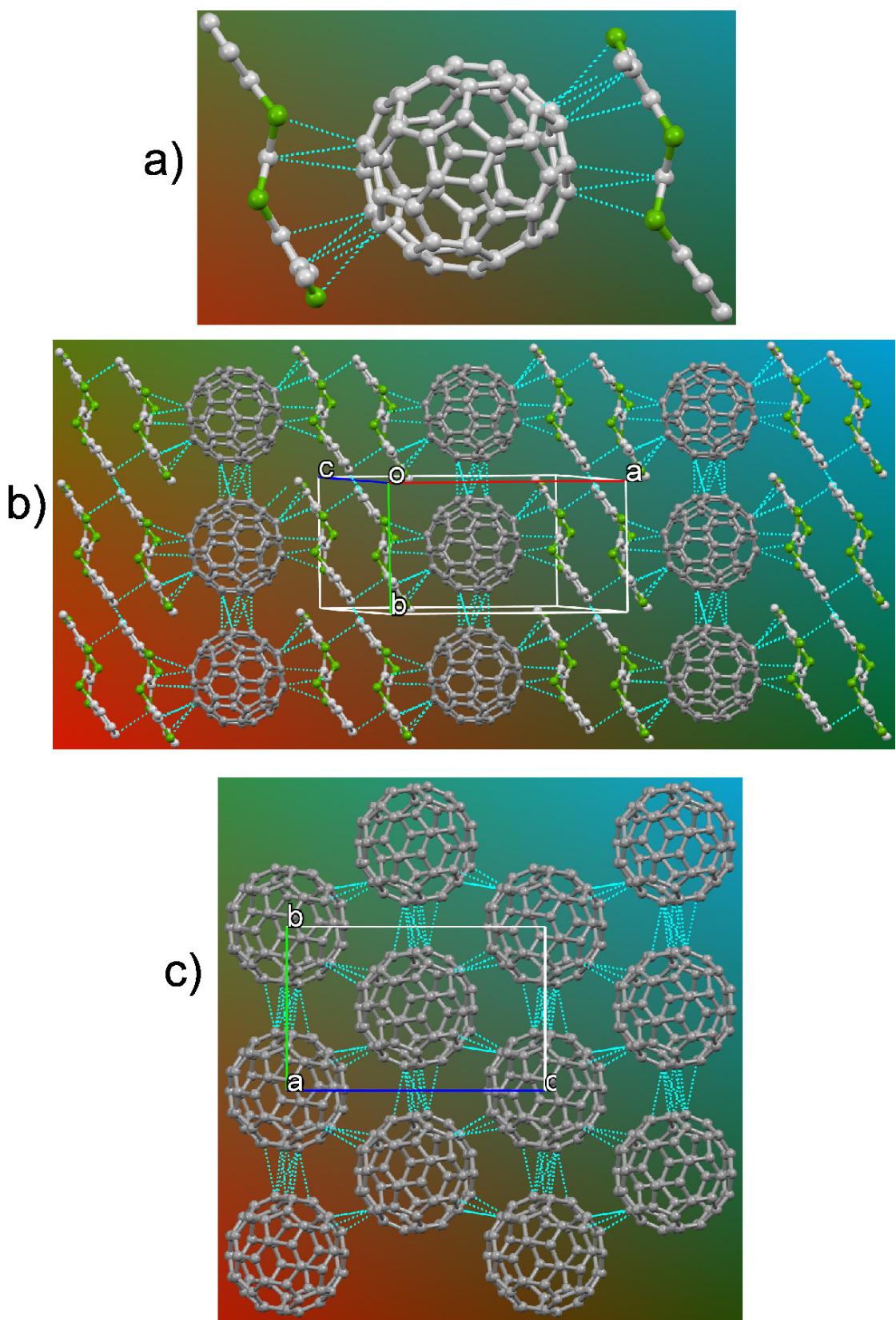
**Fig. S25** Photographs for the single crystals of “TTN•fullerene” cocrystals. The grid for the background of each figure is 1 mm × 1mm.

**Table 4** Selected crystallographic data for “TTN•fullerene” cocrystals

	(TTN3)•(C <sub>60</sub> )•(CS <sub>2</sub> )	(TTN3)•(C <sub>60</sub> )	(TTN3)•(C <sub>70</sub> )•(CS <sub>2</sub> )	(TTN3)•(C <sub>70</sub> )	(TTN2)•(C <sub>60</sub> )	(TTN6) <sub>2</sub> •(C <sub>60</sub> )
CCDC number	1492711	1492712	1492713	1492714	1492715	1492716
Empirical formula	C <sub>75</sub> H <sub>12</sub> S <sub>8</sub>	C <sub>74</sub> H <sub>12</sub> S <sub>6</sub>	C <sub>85</sub> H <sub>12</sub> S <sub>8</sub>	C <sub>84</sub> H <sub>12</sub> S <sub>6</sub>	C <sub>78</sub> H <sub>16</sub> O <sub>4</sub> S <sub>4</sub>	C <sub>88</sub> H <sub>20</sub> S <sub>10</sub>
Formula weight	1169.32	1093.20	1289.43	1213.30	1145.15	1397.64
Temperature / K	110(2)	110(2)	110(2)	77(1)	110(2)	110(2)
$\lambda$ / Å	0.71073	0.71073	1.5418	0.95382	1.5418	1.5418
Crystal size / mm <sup>3</sup>	0.38×0.35×0.34	0.19×0.17×0.14	0.42×0.35×0.32	0.25×0.19×0.17	0.30×0.20×0.07	0.33×0.26×0.16
Crystal system	Triclinic	Orthorhombic	Triclinic	Triclinic	Monoclinic	Monoclinic
space group	<i>P</i> -1	<i>Pnma</i>	<i>P</i> -1	<i>P</i> -1	<i>I</i> 2/ <i>a</i>	<i>P</i> 2 <sub>1</sub> / <i>c</i>
<i>a</i> / Å	9.9766(4)	17.5533(19)	10.0905(3)	10.1580(9)	14.7684(5)	17.97(4)
<i>b</i> / Å	10.6457(6)	24.604(3)	13.6277(3)	12.6980(9)	13.5962(4)	9.81(2)
<i>c</i> / Å	10.9297(6)	9.7837 (6)	18.3146(6)	19.1600(13)	22.7333(10)	17.28(6)
$\alpha$ / °	77.433(5)	90	86.280(2)	93.18	90	90
$\beta$ / °	81.220(4)	90	77.027(3)	99.488(12)	103.105(4)	116.4(3)
$\gamma$ / °	80.005(4)	90	87.191(2)	104.98	90	90
<i>V</i> / Å <sup>3</sup>	1107.82(10)	4225.3 (8)	2447.46(12)	2342.(3)	4445.8(3)	2729(13)
<i>Z</i>	1	4	2	2	4	2
<i>d</i> <sub>calc</sub> / g·cm <sup>-3</sup>	1.753	1.718	1.750	1.720	1.711	1.701
$\mu$ / mm <sup>-1</sup>	0.463	0.383	3.874	0.409	2.53	4.224
2 <i>θ</i> <sub>max</sub> / °	57.2	56.92	141.52	59.6	141.38	142.3
Data / restraints / parameters	4986/0/378	4835/0/369	9358/0/845	10169/0/816	4201/0/390	4291/222/384
<i>GooF</i>	1.107	1.046	1.038	1.112	1.138	1.074
<i>R</i> [ <i>I</i> >2σ( <i>I</i> )]	0.0526	0.0559	0.0516	0.0688	0.0681	0.1064
<i>wR</i> <sub>2</sub>	0.1243	0.1136	0.1380	0.2102	0.1877	0.2869

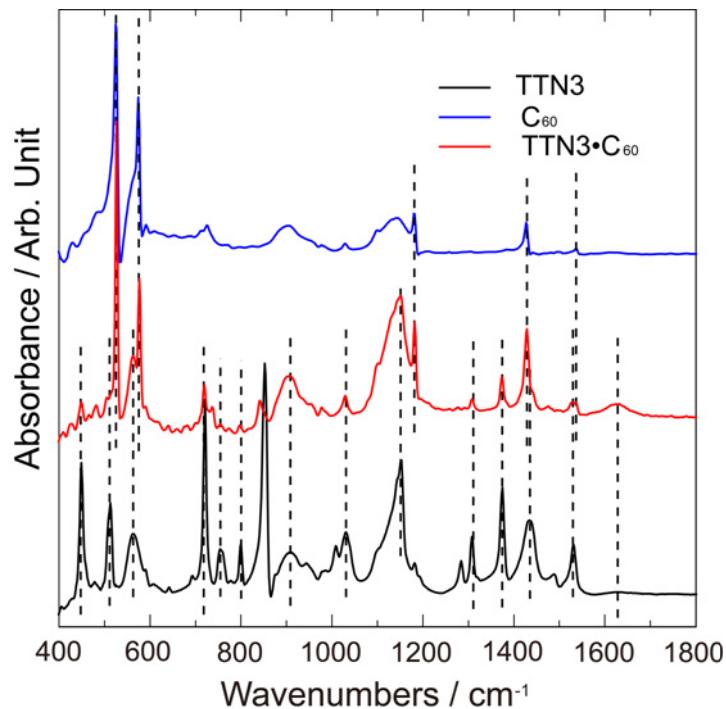


**Fig. S26** Crystal structure of  $(\text{TTN}2)\bullet(\text{C}_{60})$ : a) intermolecular interactions between **TTN2** and  $\text{C}_{60}$ ; b) packing structures. The dashed lines shows the intermolecular interactions and the hydrogen atoms are omitted for clarity

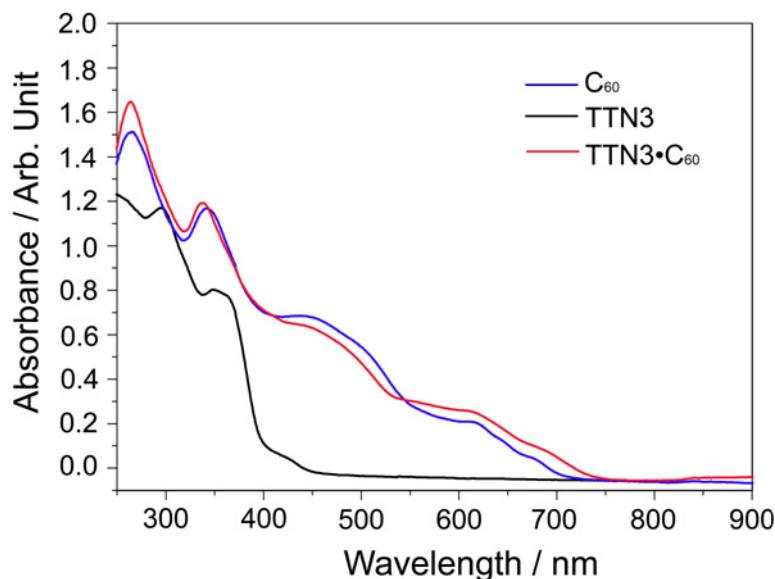


**Fig. S27** Crystal structure of  $(\text{TTN6})_2 \bullet (\text{C}_{60})$ : a) intermolecular interactions between **TTN6** and  $\text{C}_{60}$ ; b) packing structures; c) packing structure of  $\text{C}_{60}$  molecules. The dashed lines shows the intermolecular interactions and the hydrogen atoms are omitted for clarity

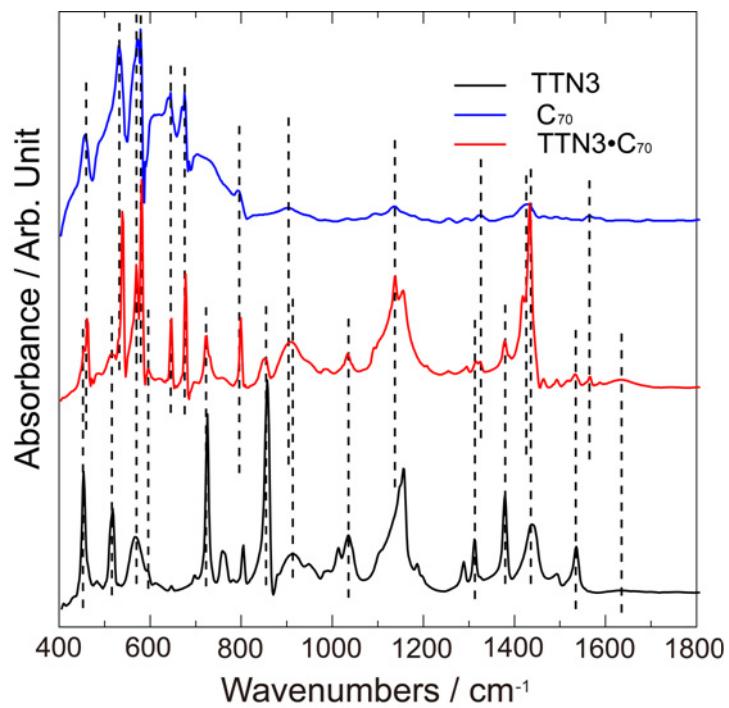
## VI. Optical spectra of the representative “TTN<sub>3</sub>•fullerene” cocrystals



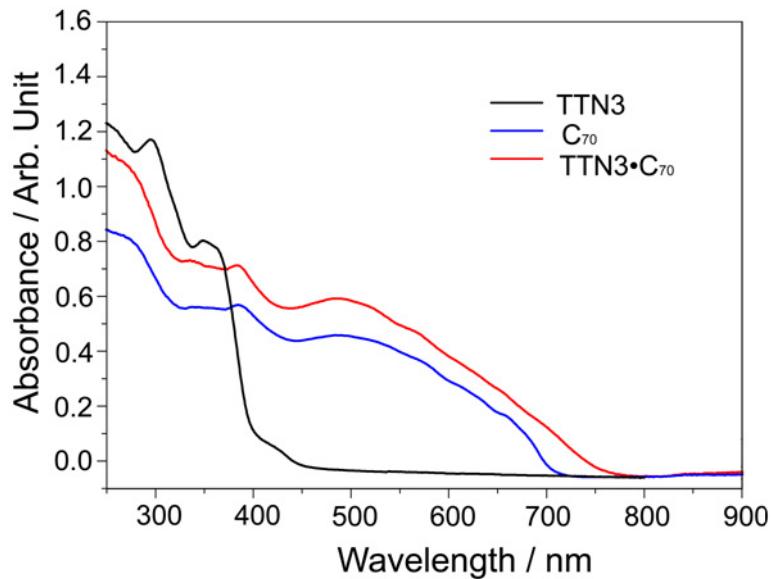
**Fig. S28** IR spectra of (TTN<sub>3</sub>)•(C<sub>60</sub>) cocrystal, along with those of TTN<sub>3</sub> and C<sub>60</sub> for comparison. The spectra were measured on the dispersed samples in KBr pellet, and the black dashed lines are guides for the eye.



**Fig. S29** Solid state absorption spectra of (TTN<sub>3</sub>)•(C<sub>60</sub>) cocrystal, along with those of TTN<sub>3</sub> and C<sub>60</sub> for comparison. The spectra were measured on the dispersed samples in KBr pellet.

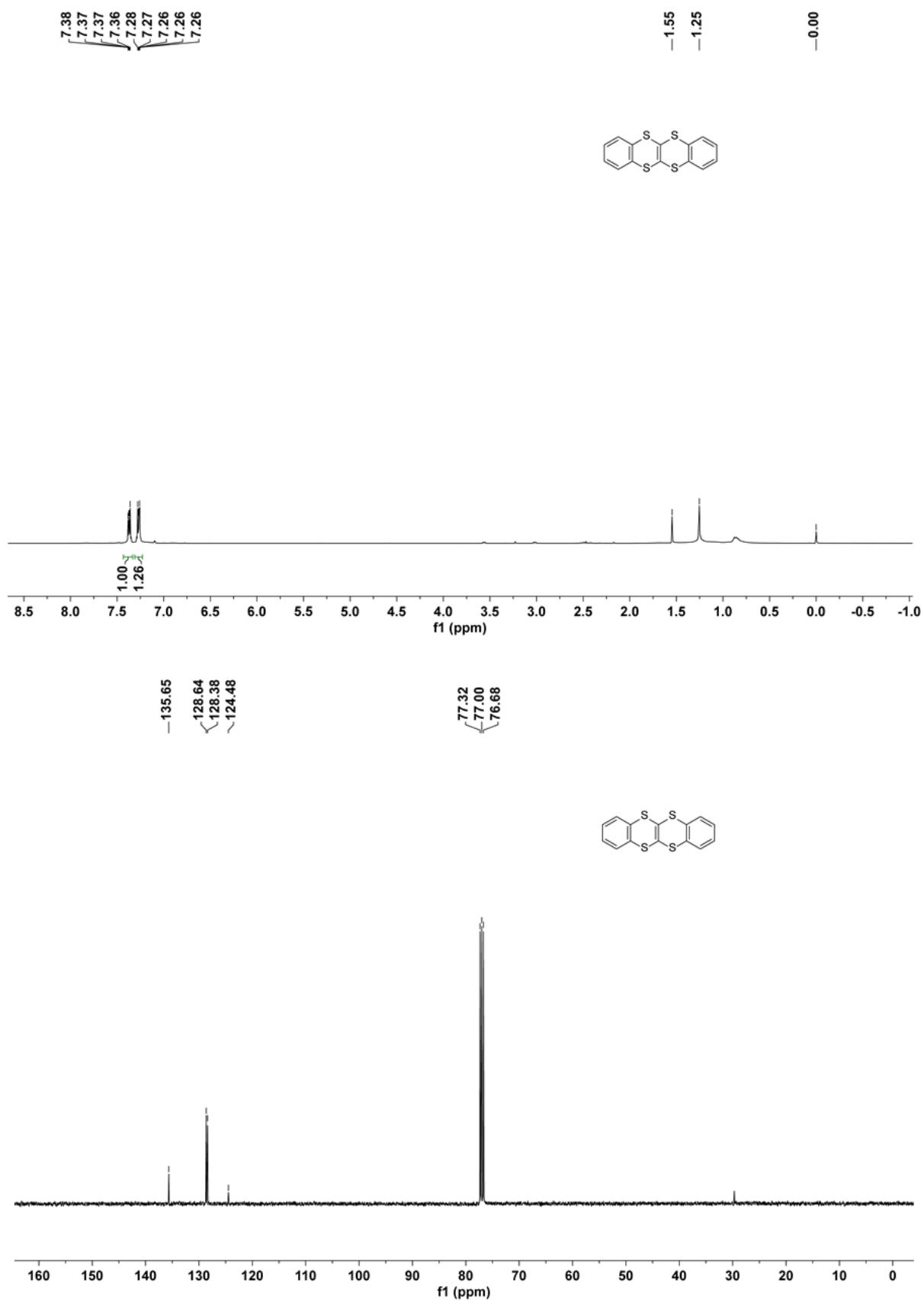


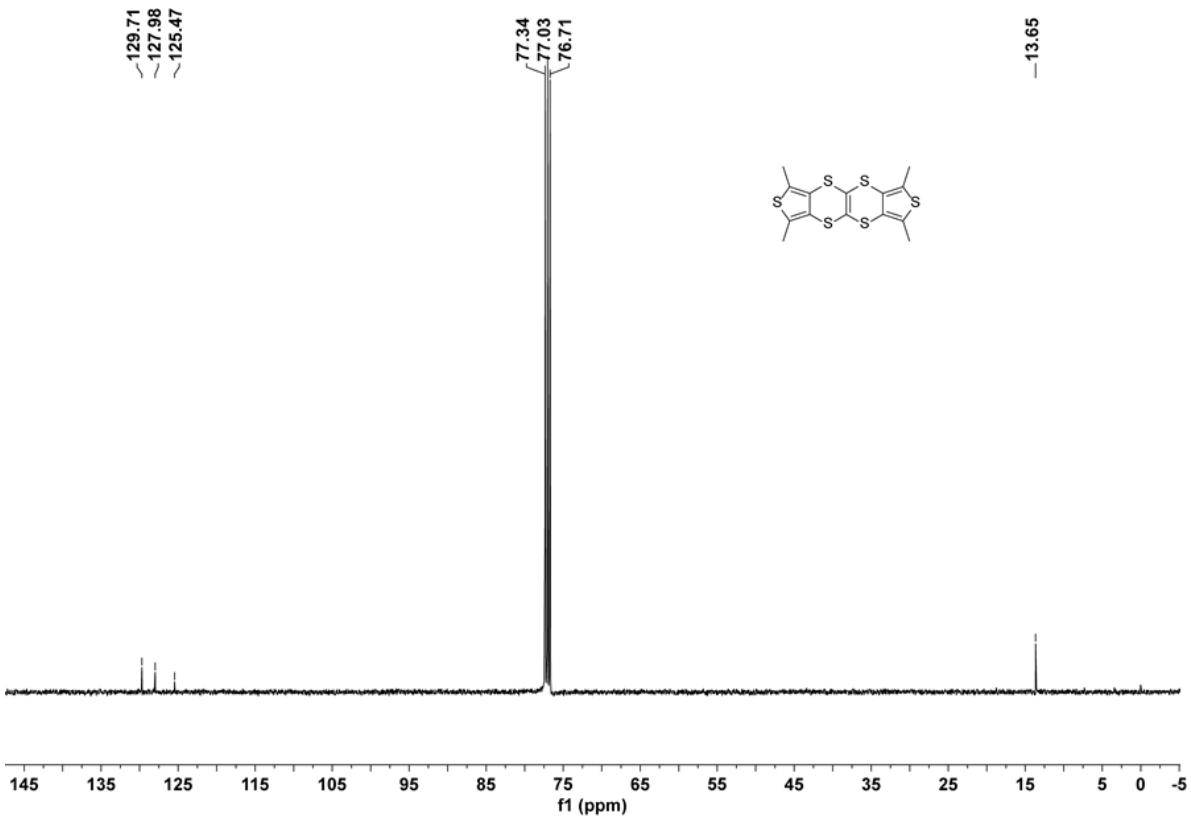
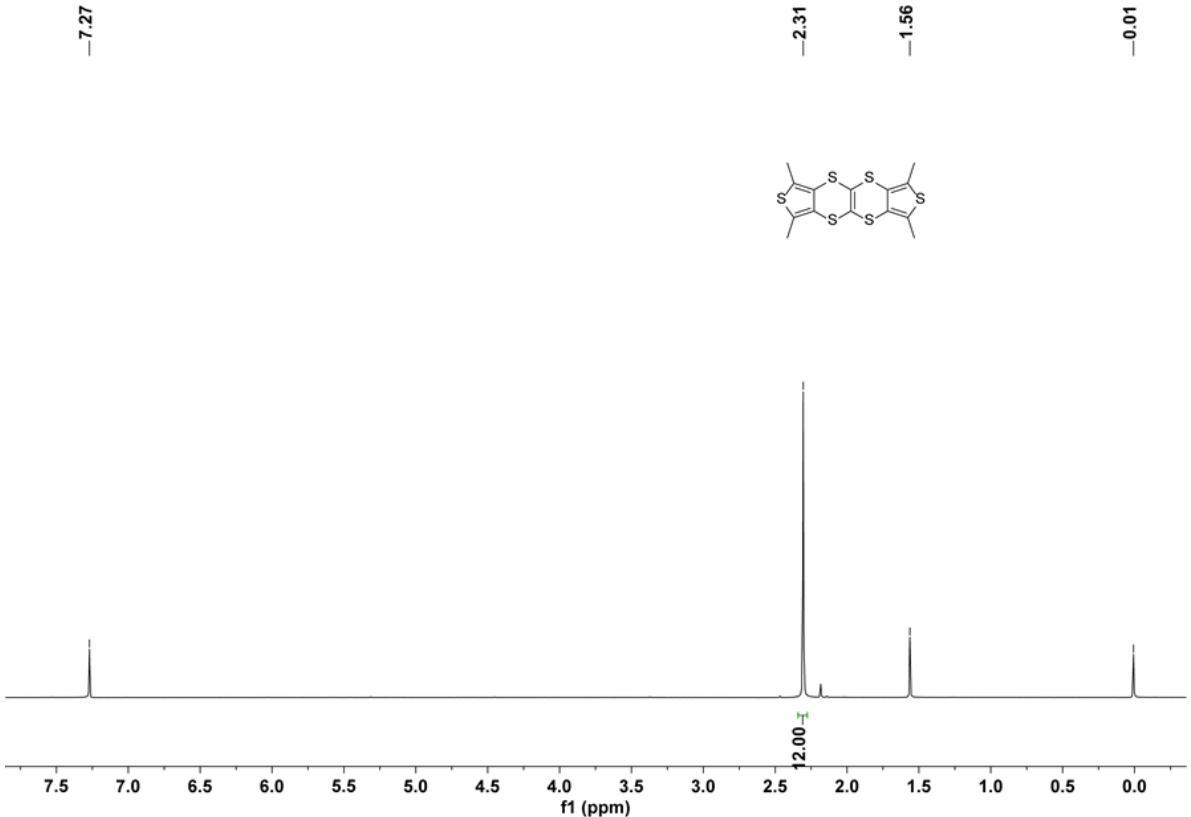
**Fig. S30** IR spectra of (TTN3)•(C<sub>70</sub>) cocrystal, along with those of TTN3 and C<sub>70</sub> for comparison. The spectra were measured on the dispersed samples in KBr pellet, and the black dashed lines are guides for the eye.

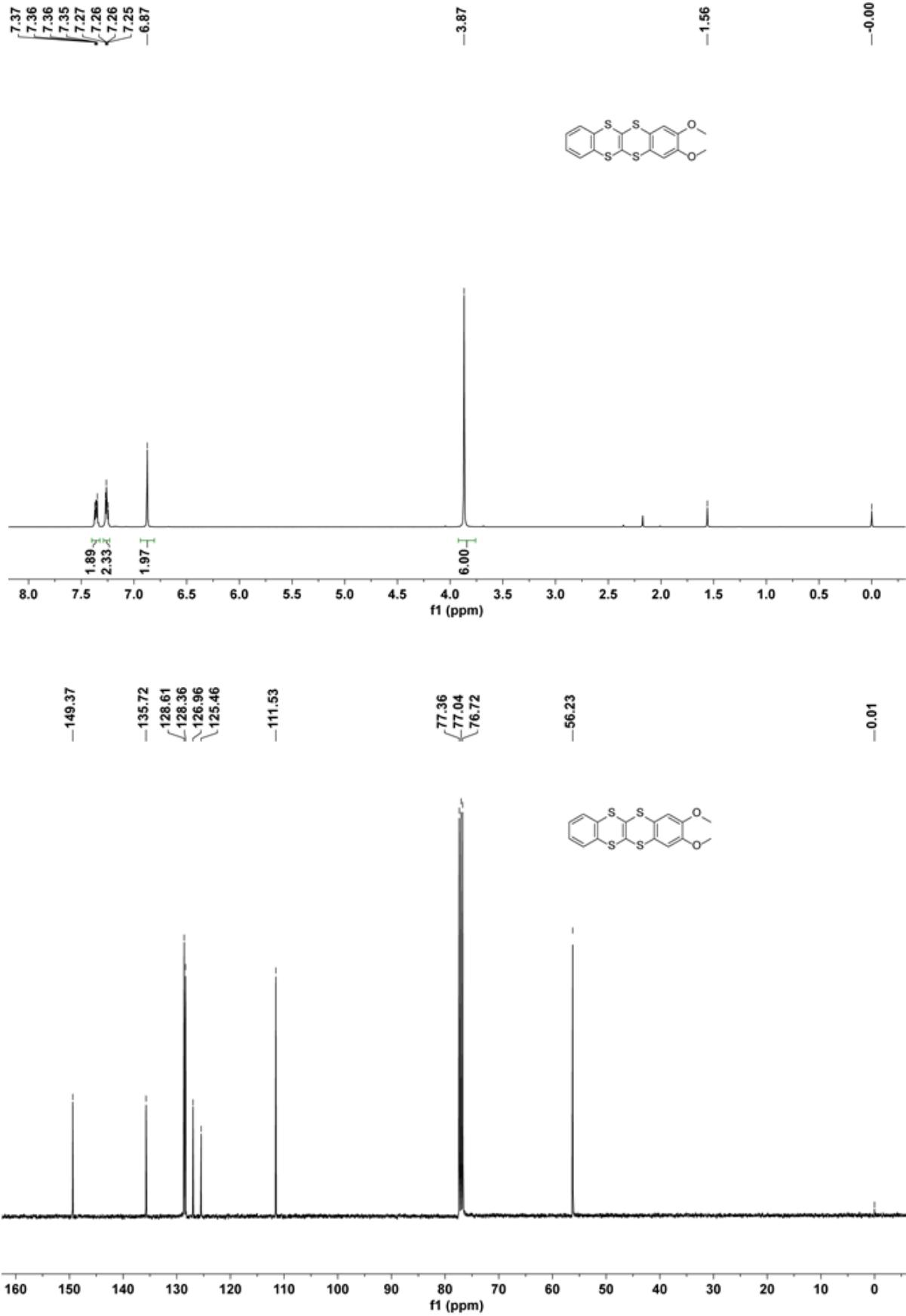


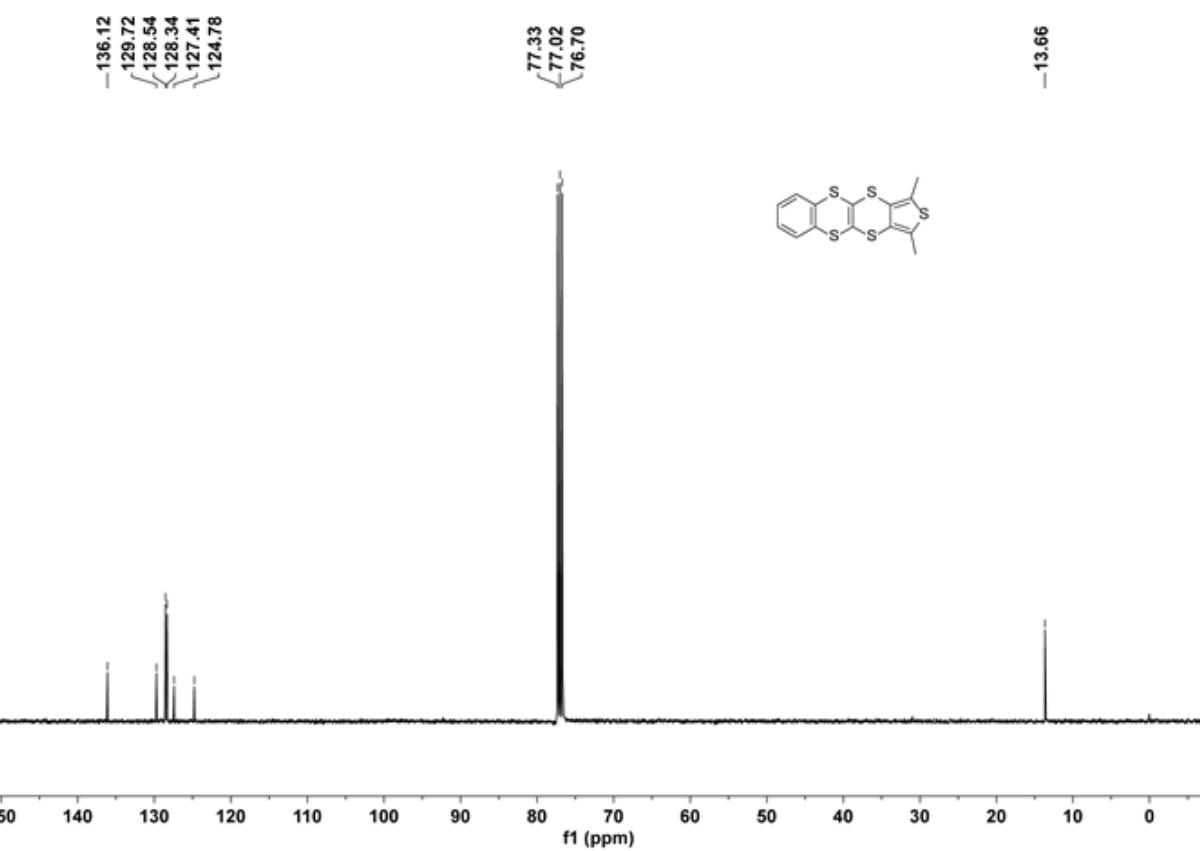
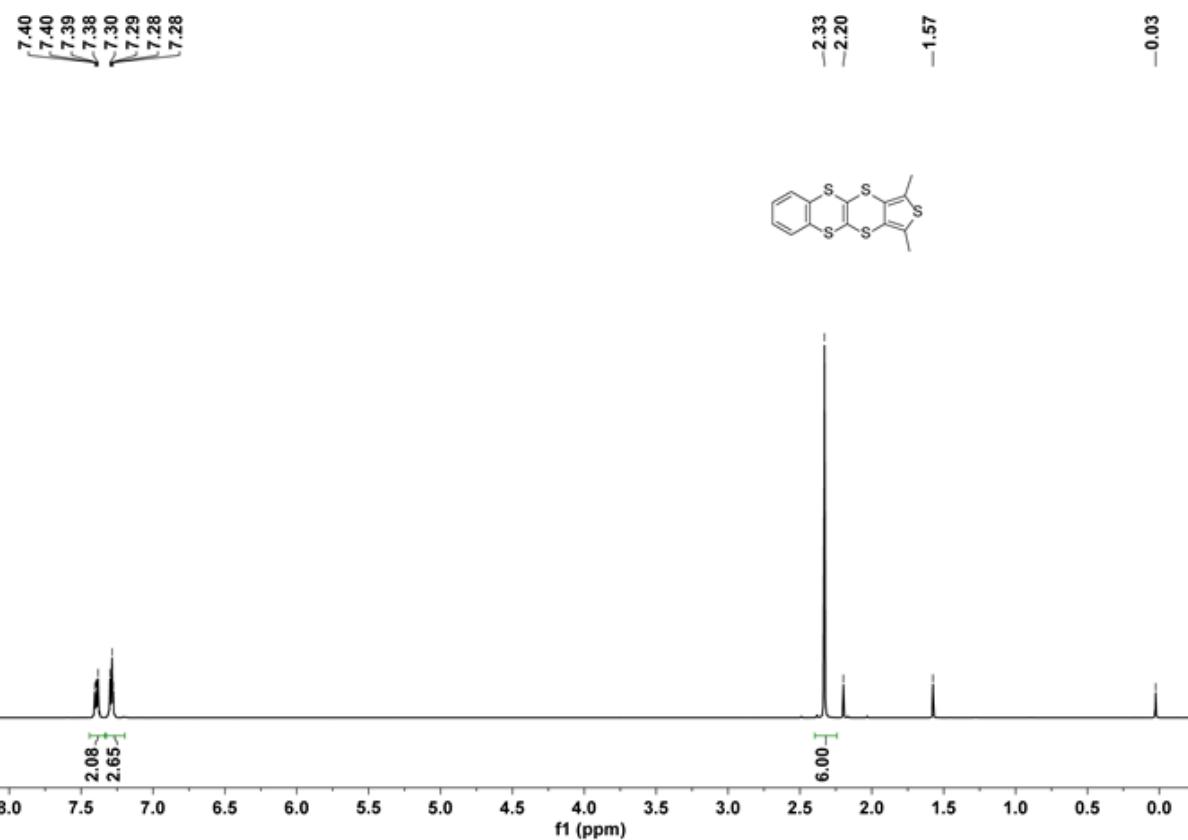
**Fig. S31** Solid state absorption spectra of (TTN3)•(C<sub>70</sub>) cocrystal, along with those of TTN3 and C<sub>70</sub> for comparison. The spectra were measured on the dispersed samples in KBr pellet.

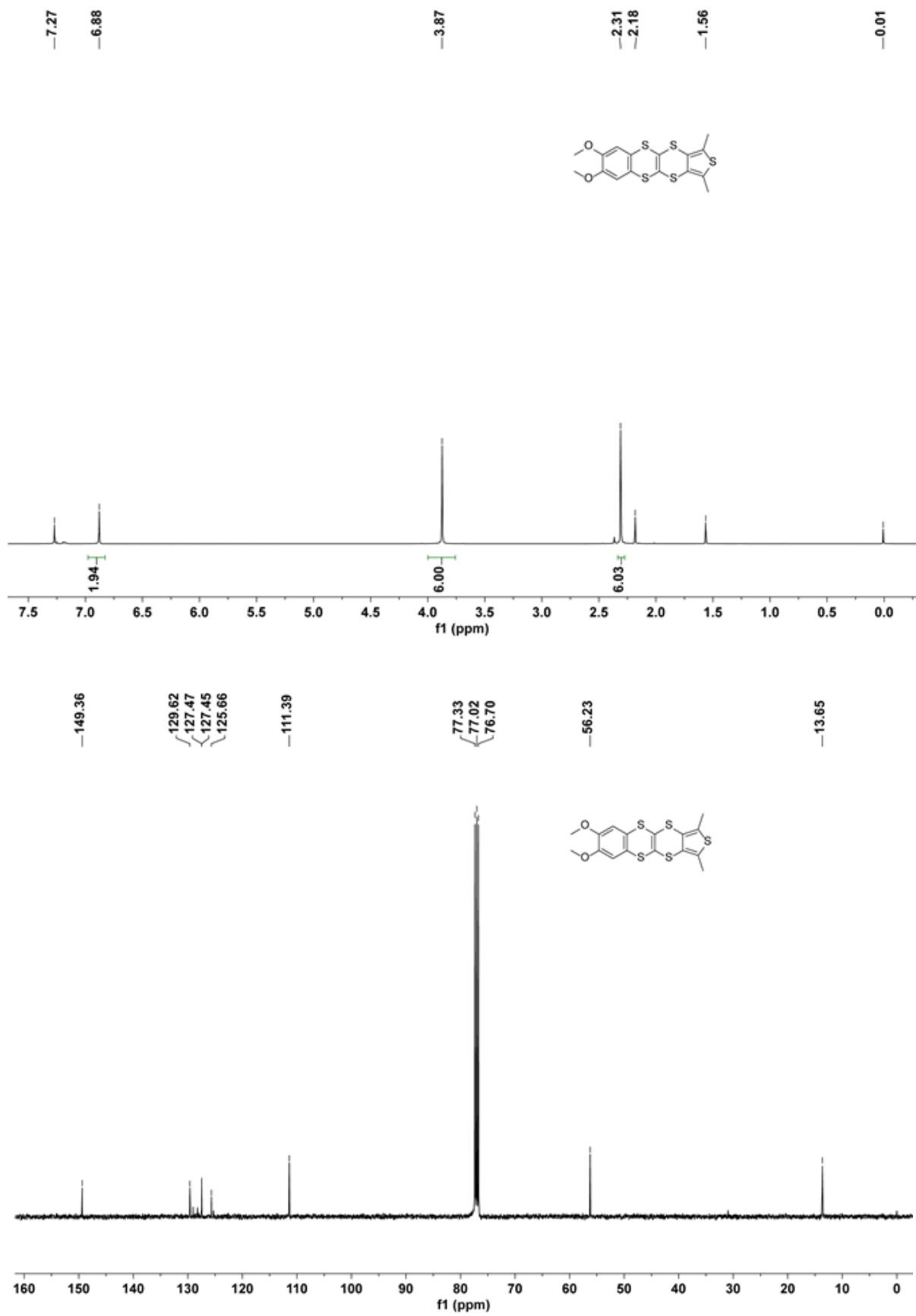
## VII. Original NMR and Mass Spectra for TTNs

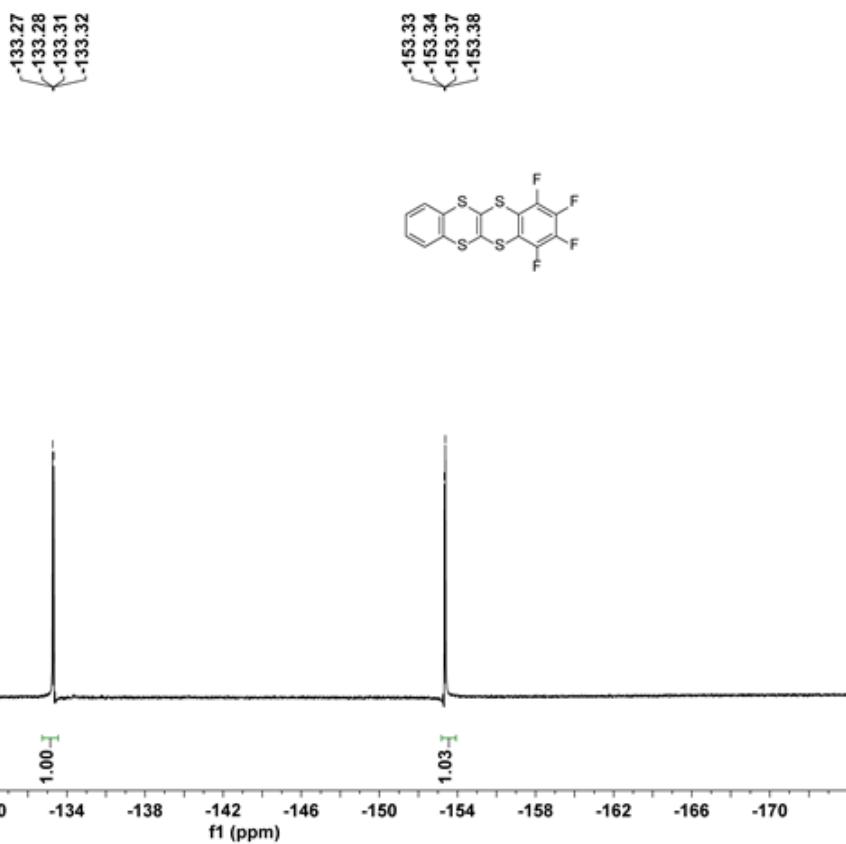
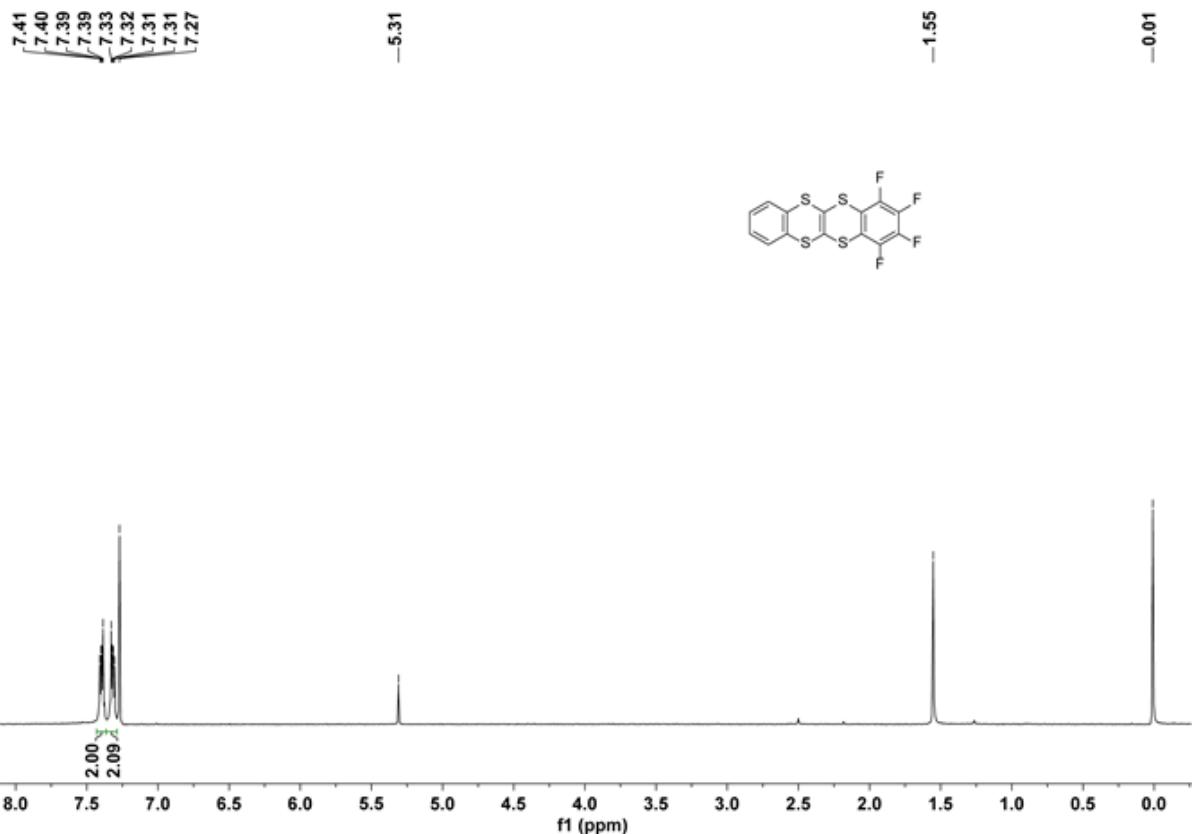


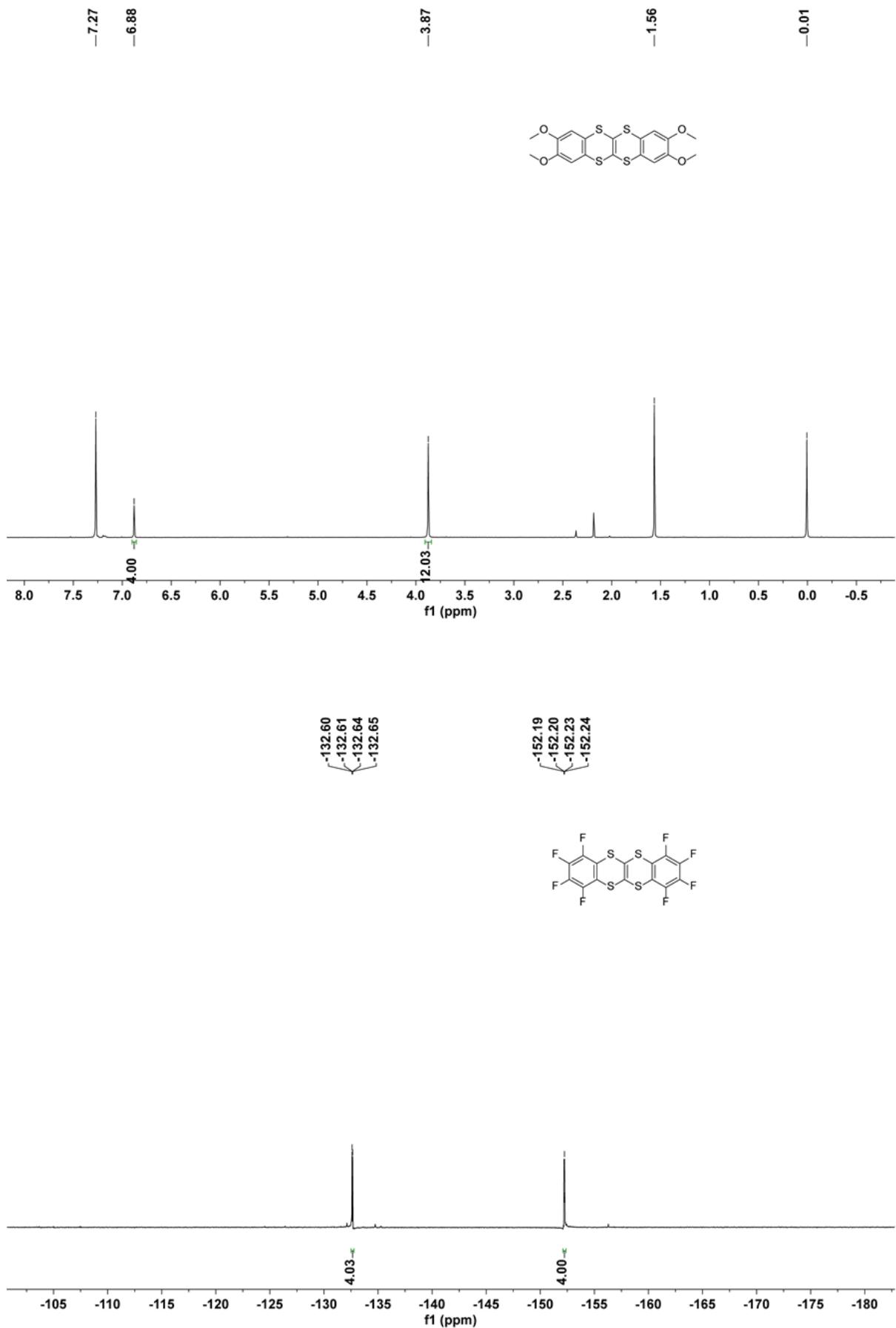












**Figure S39**

