

## Supplementary Information

### Solution and solid-state fluorescence of 2-(2'-Hydroxyphenyl) 1,5-benzodiazepin-2-one (HBD) borate complexes

Hasan Mtiraoui,<sup>a</sup> Rafik Gharbi,<sup>a</sup> Moncef Msaddek,<sup>a,\*</sup> Yann Bretonnière,<sup>b</sup> Chantal Andraud,<sup>b</sup> Pierre-Yves Renard,<sup>c</sup> and Cyrille Sabot,<sup>c,\*</sup>

<sup>a</sup>. Université Monastir, Laboratory of Heterocyclic Chemistry Natural Products and Reactivity/CHPNR, Department of Chemistry, Faculty of Science of Monastir, 5000 Monastir (Tunisia).

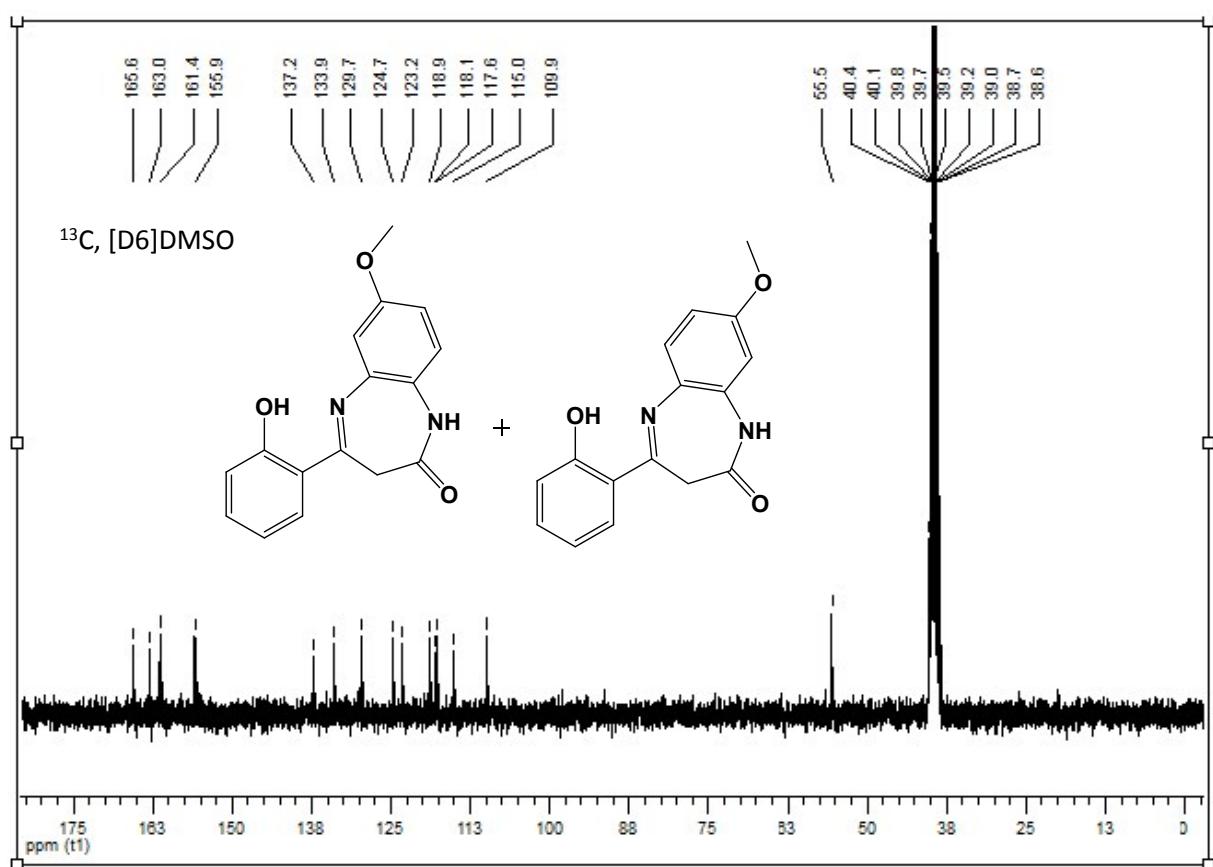
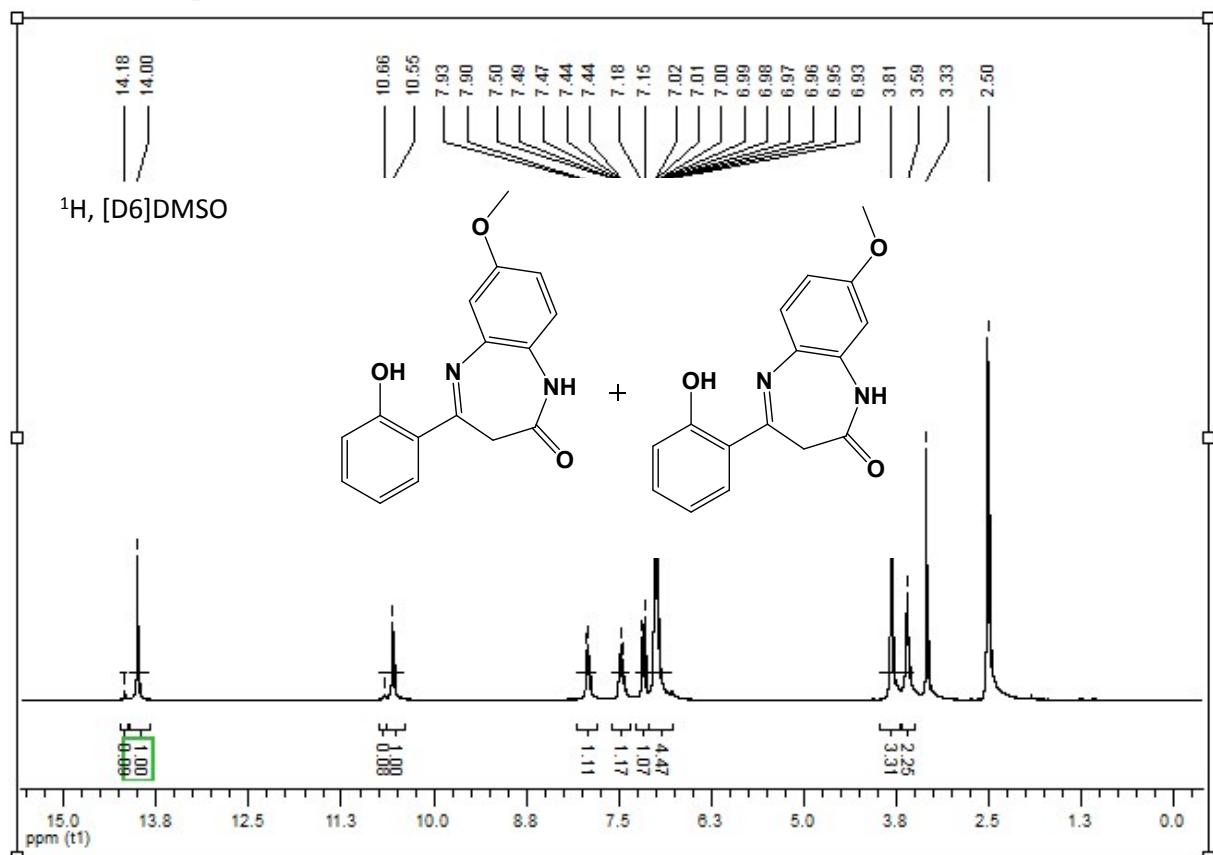
<sup>b</sup>. Université Lyon, ENS de Lyon, CNRS, Université Lyon 1, Laboratoire de Chimie, 69342 Lyon, (France).

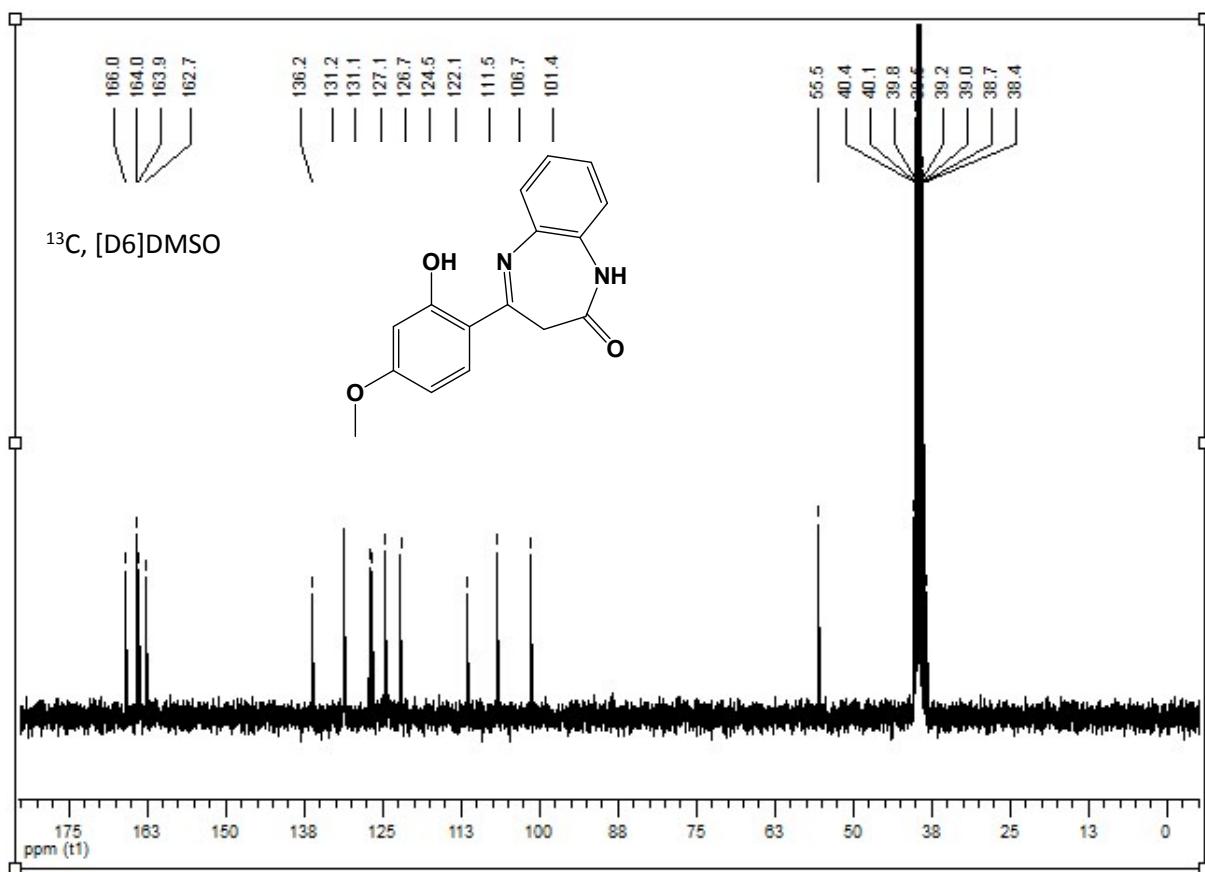
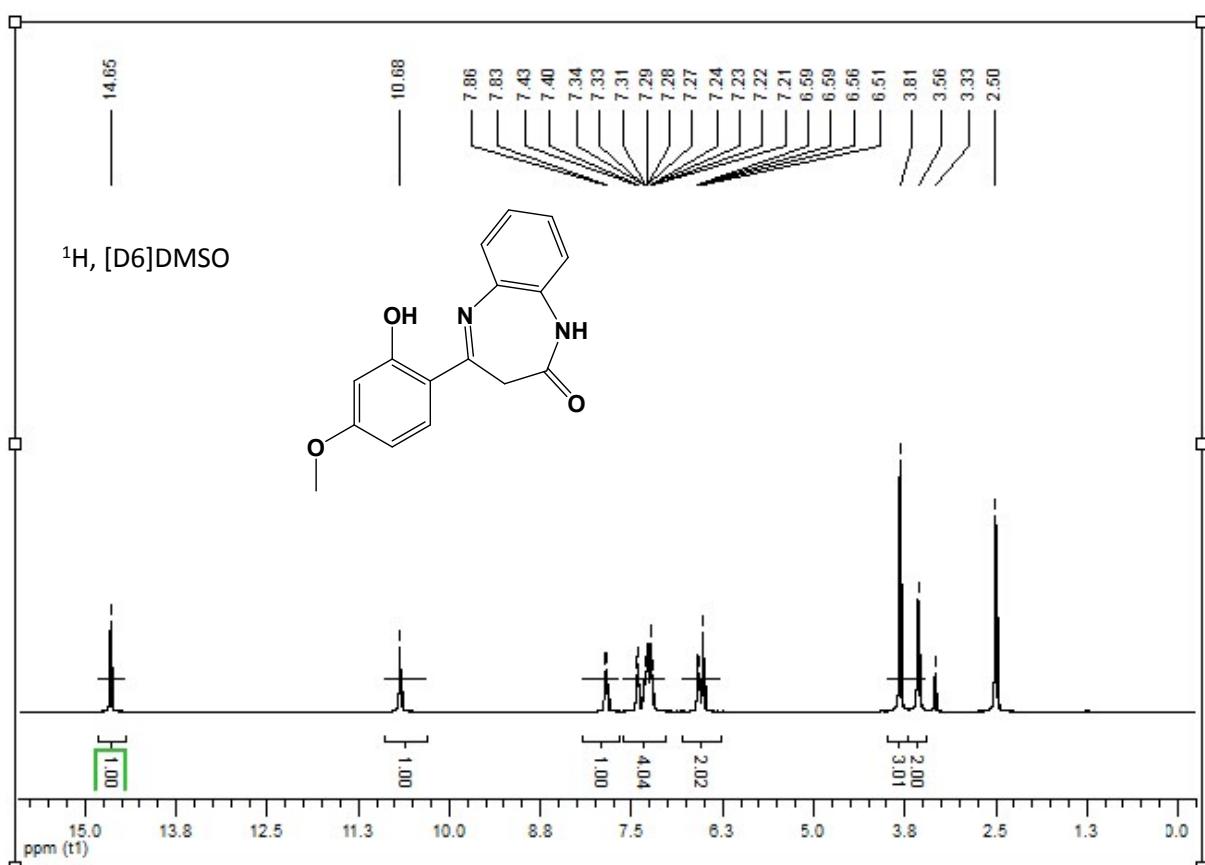
<sup>c</sup>. Normandie Univ, CNRS, UNIROUEN, INSA Rouen, COBRA (UMR 6014), 76000 Rouen, (France).

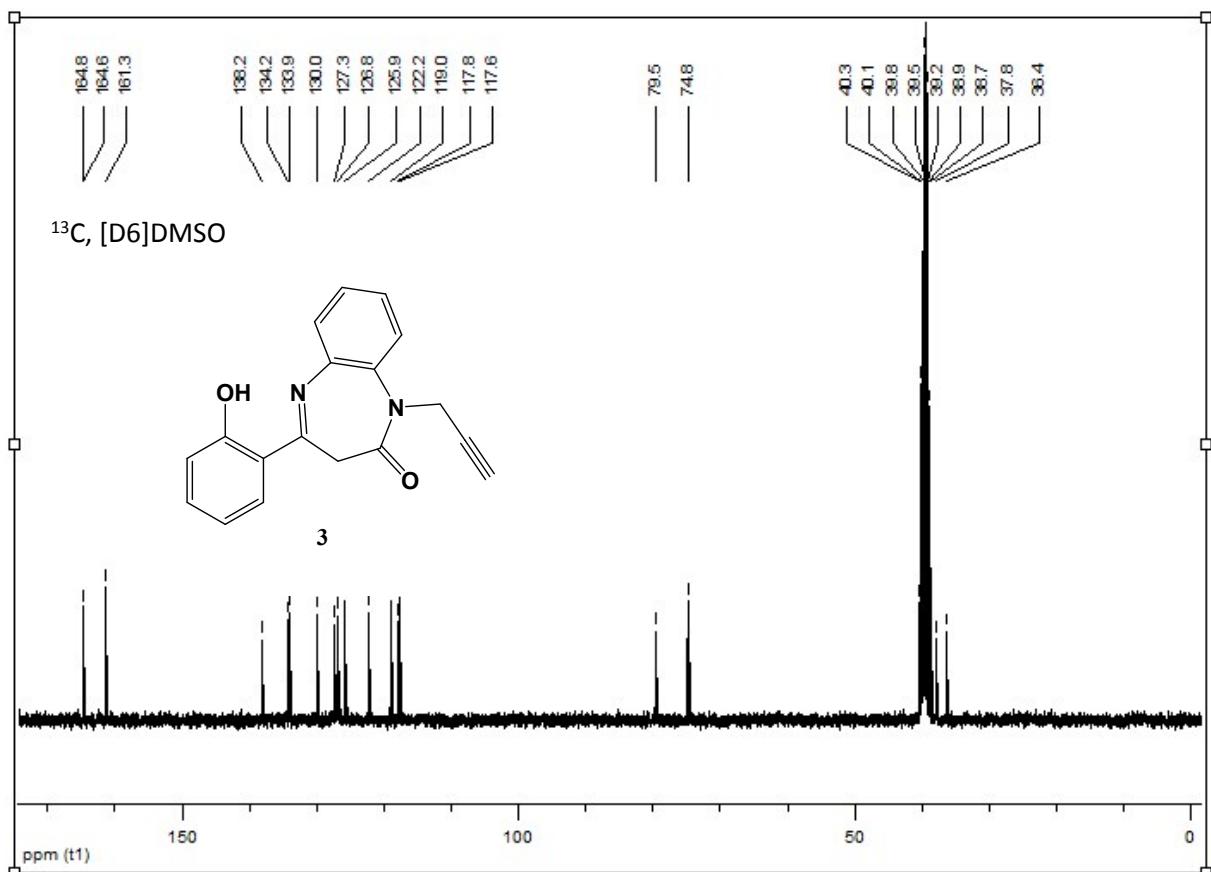
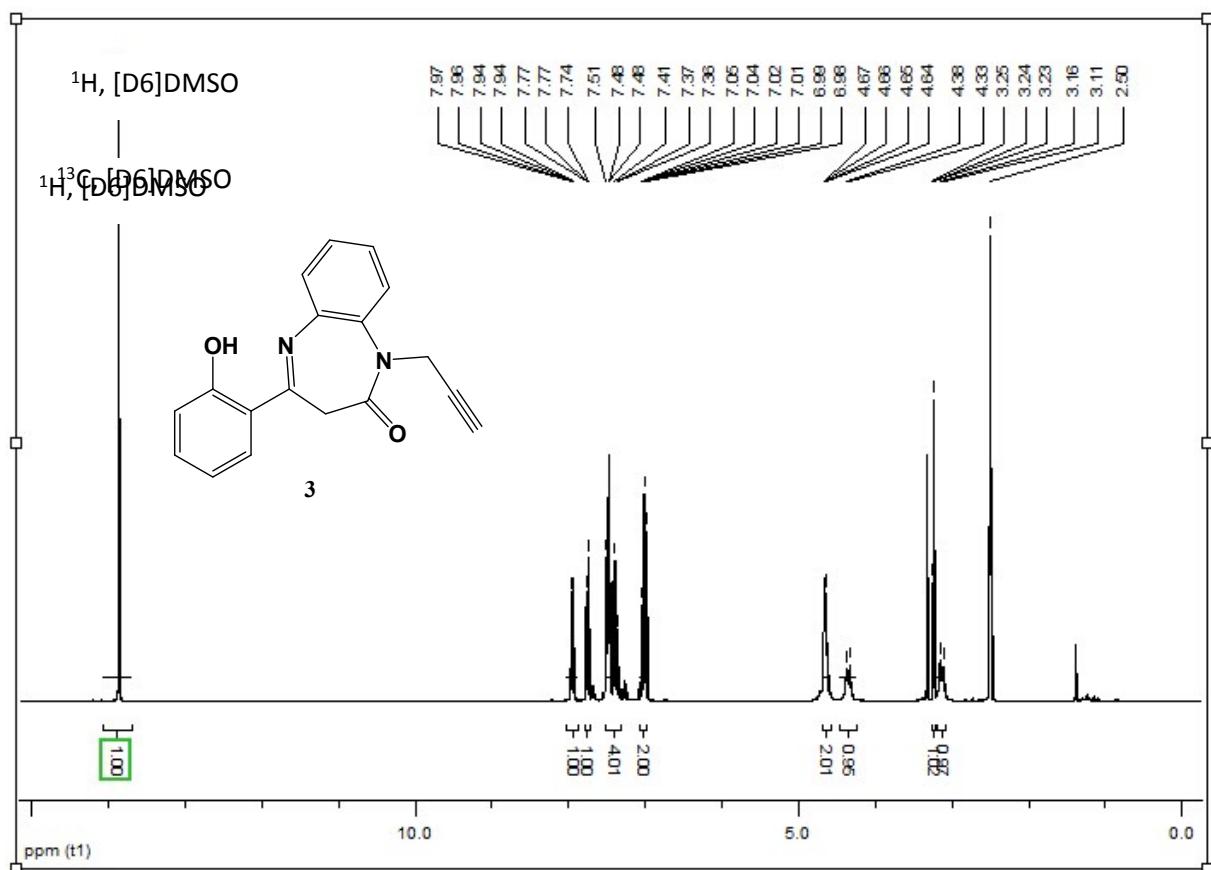
I. Copies of <sup>1</sup>H and <sup>13</sup>C NMR.....S2-S32

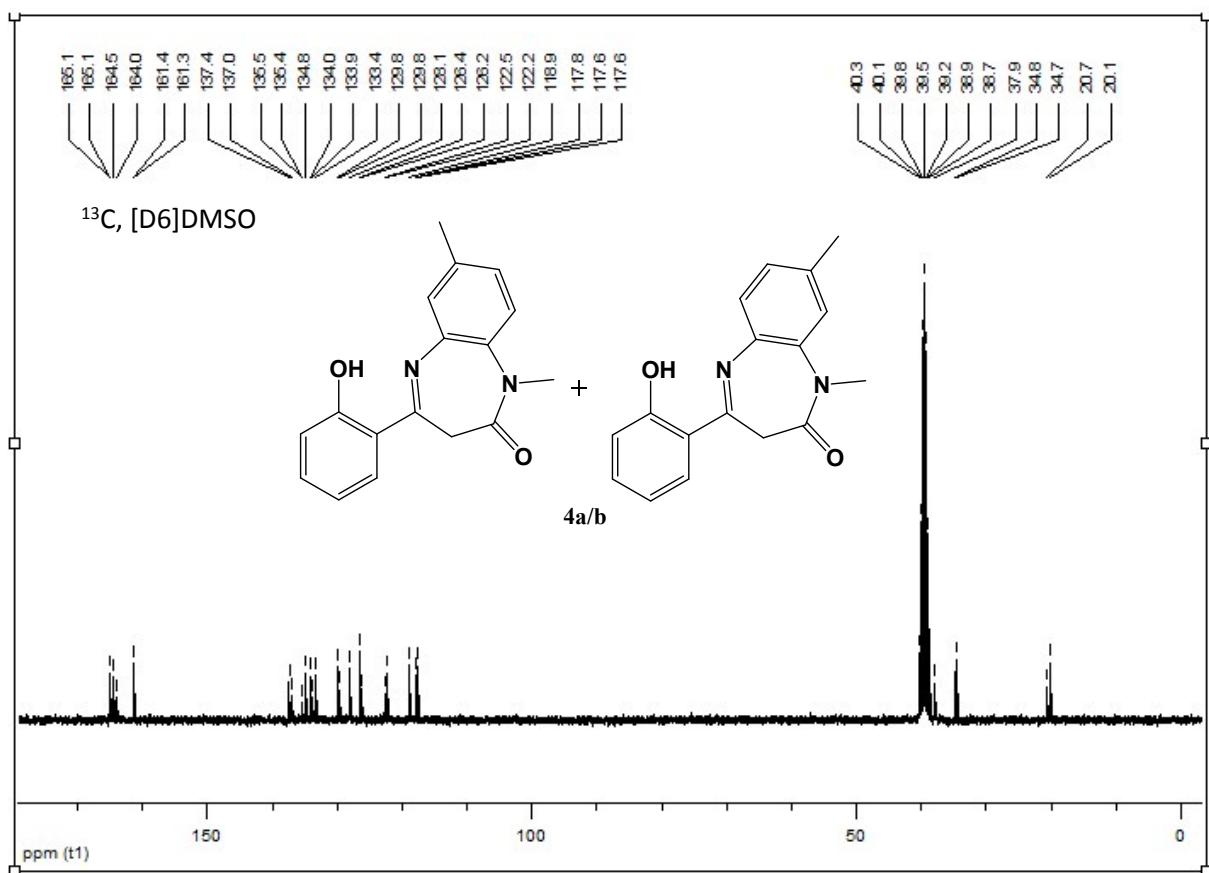
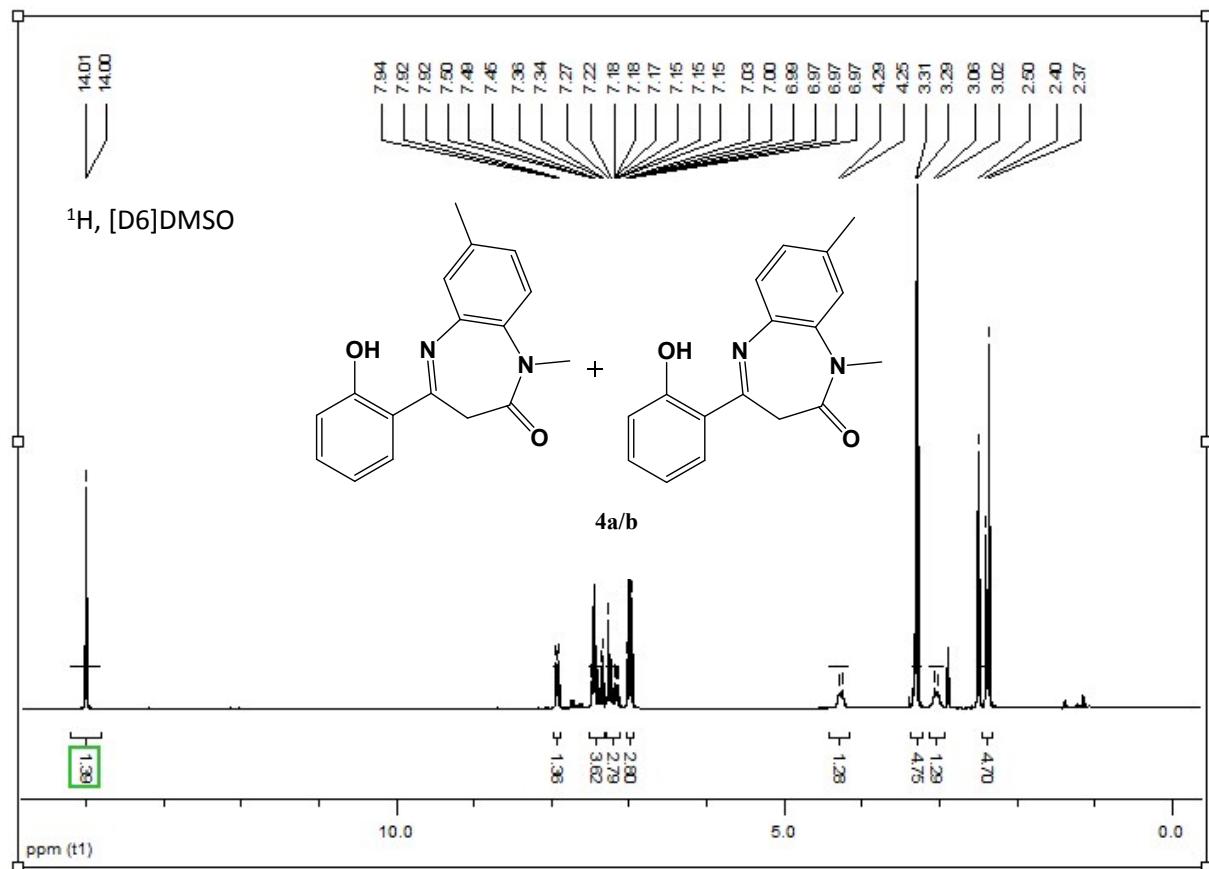
II. Crystallographic data.....S33-S43

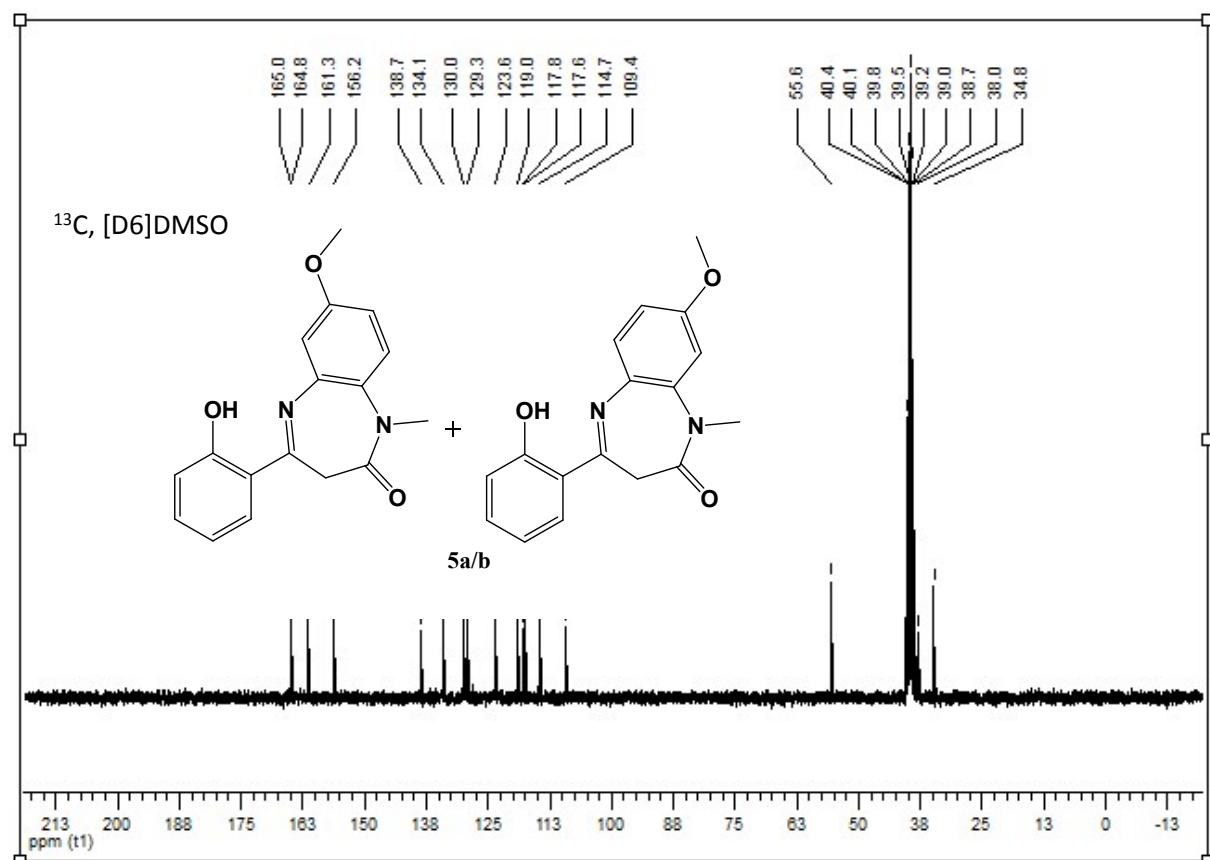
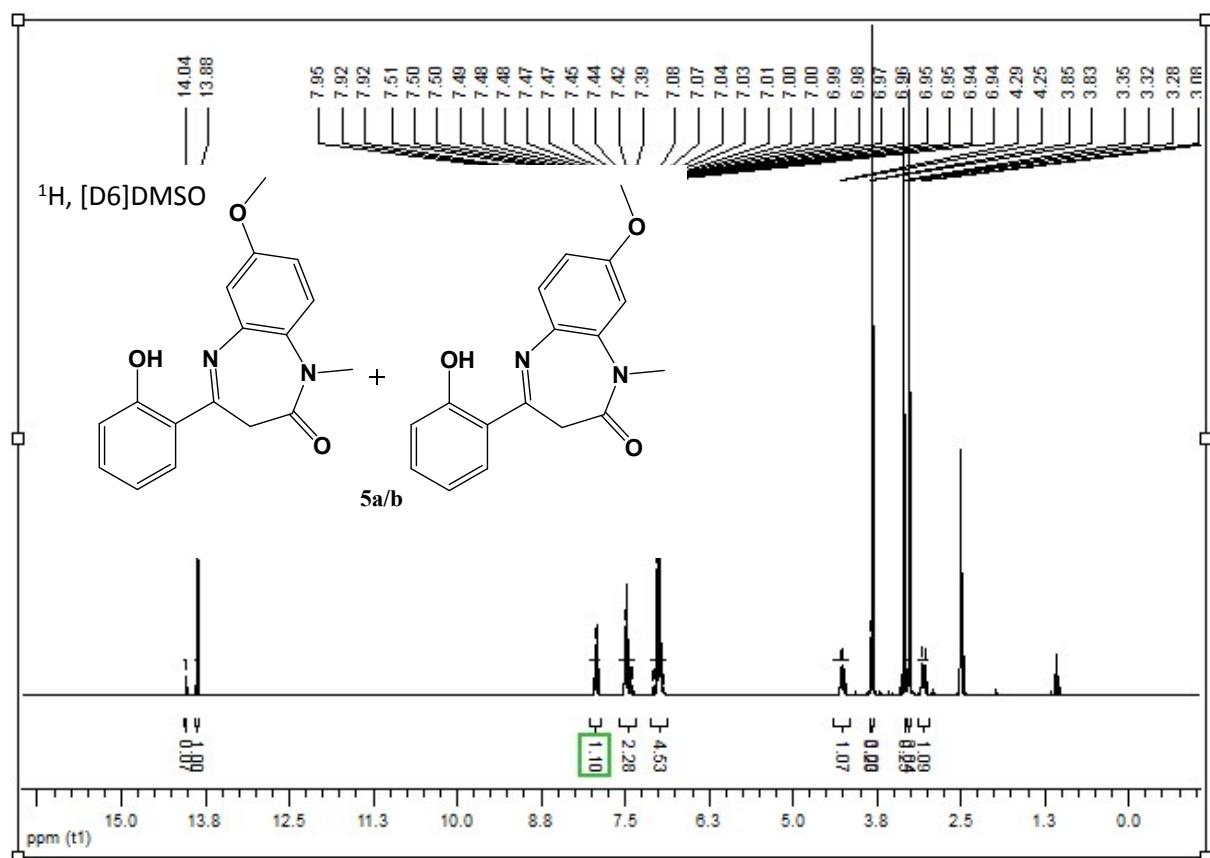
I. Copies of  $^1\text{H}$ , and  $^{13}\text{C}$  NMR

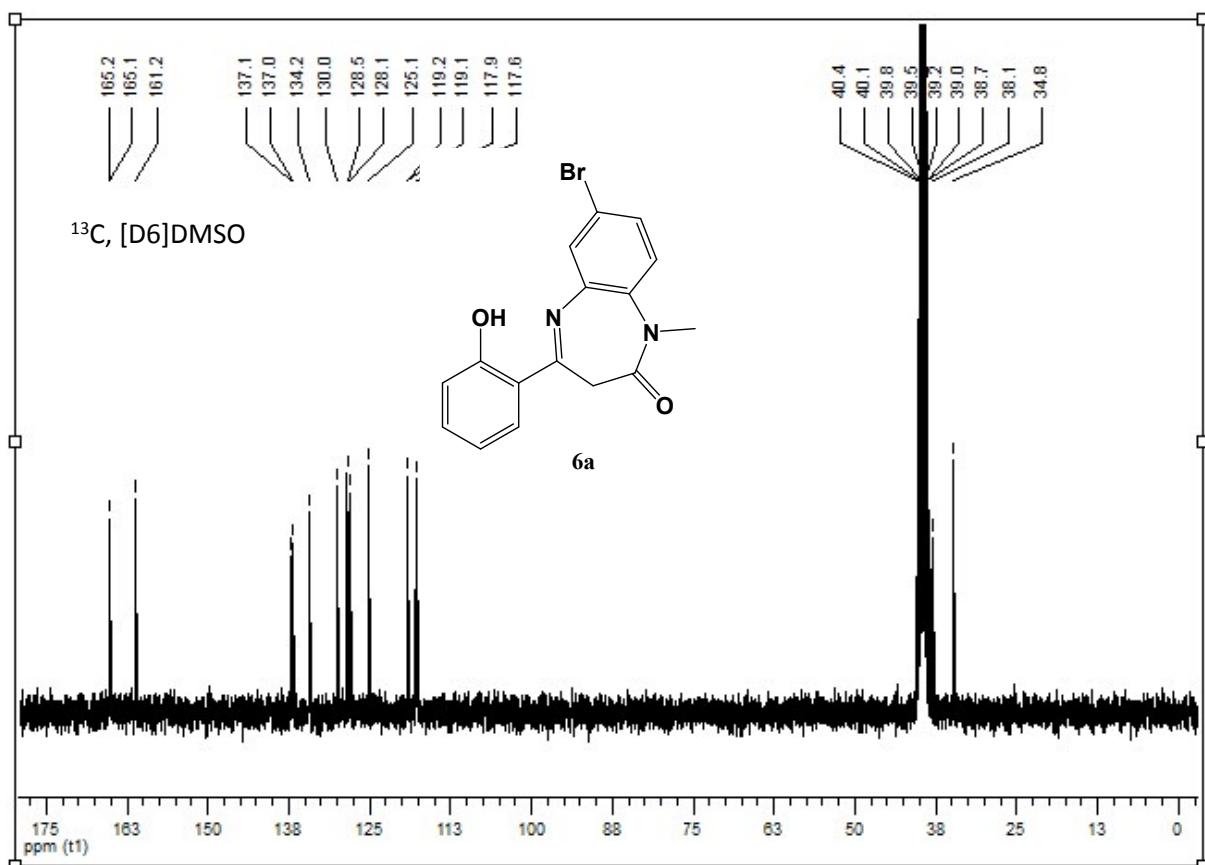
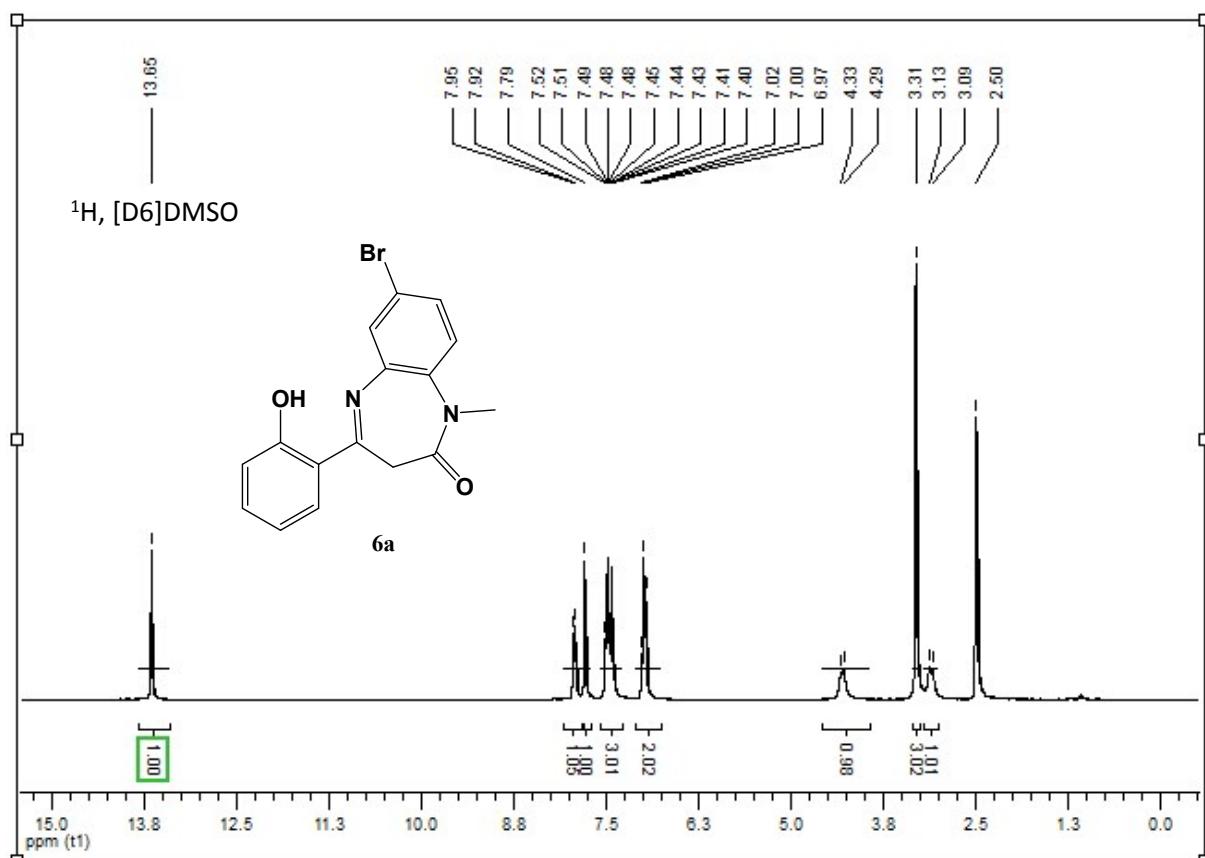


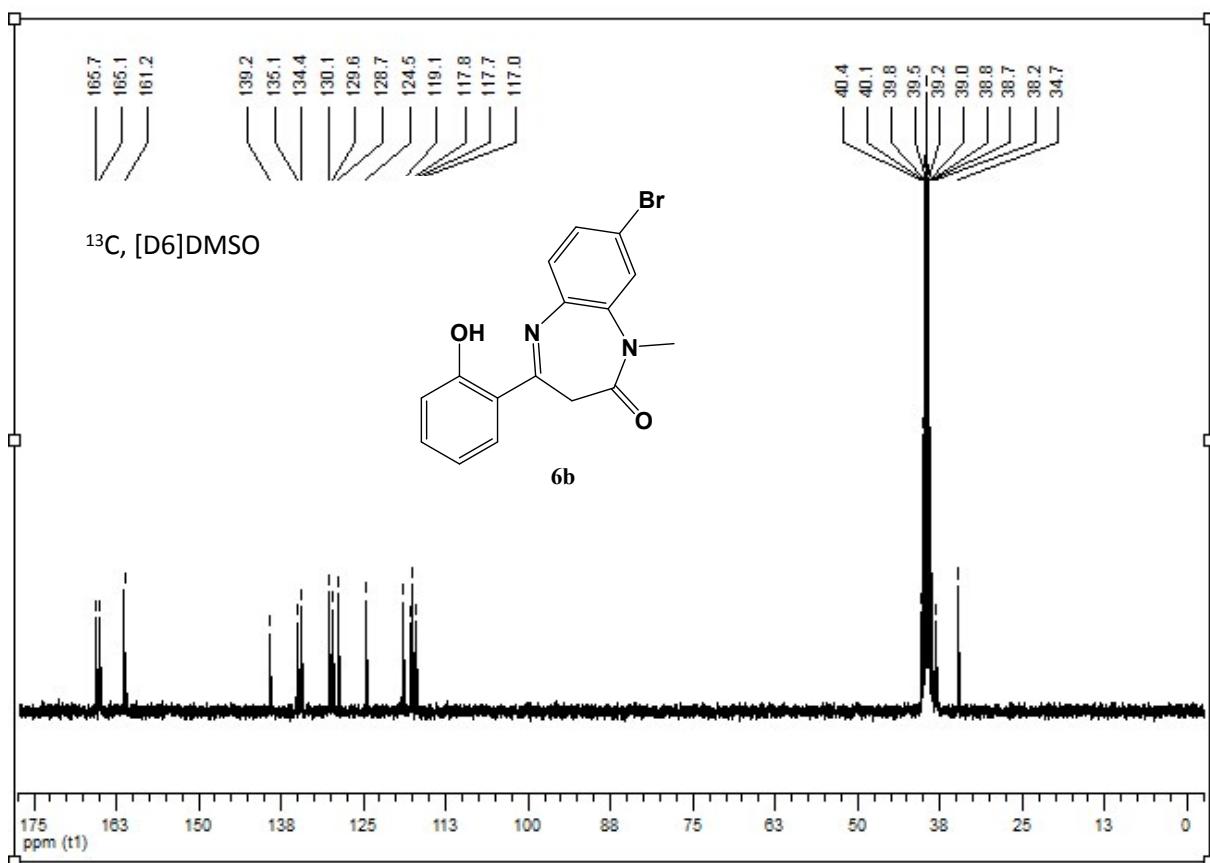
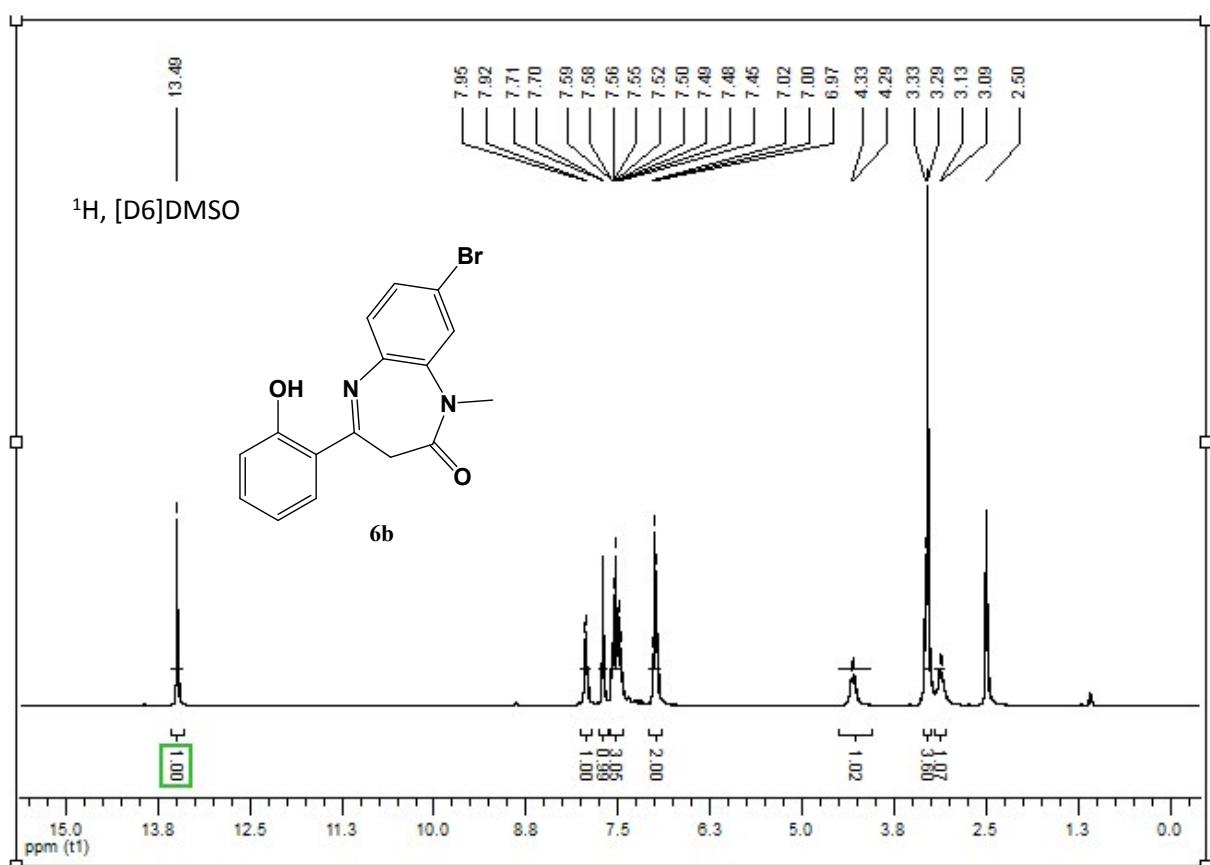


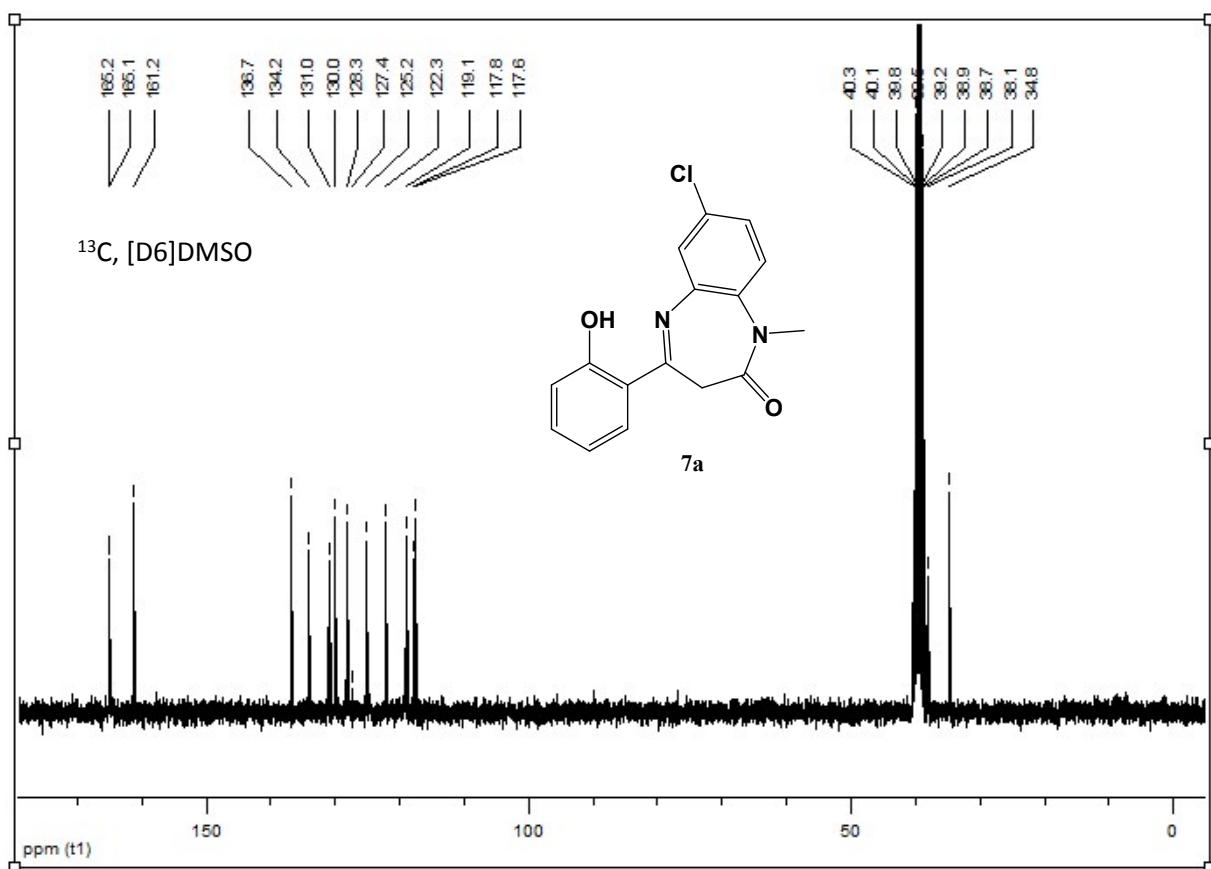
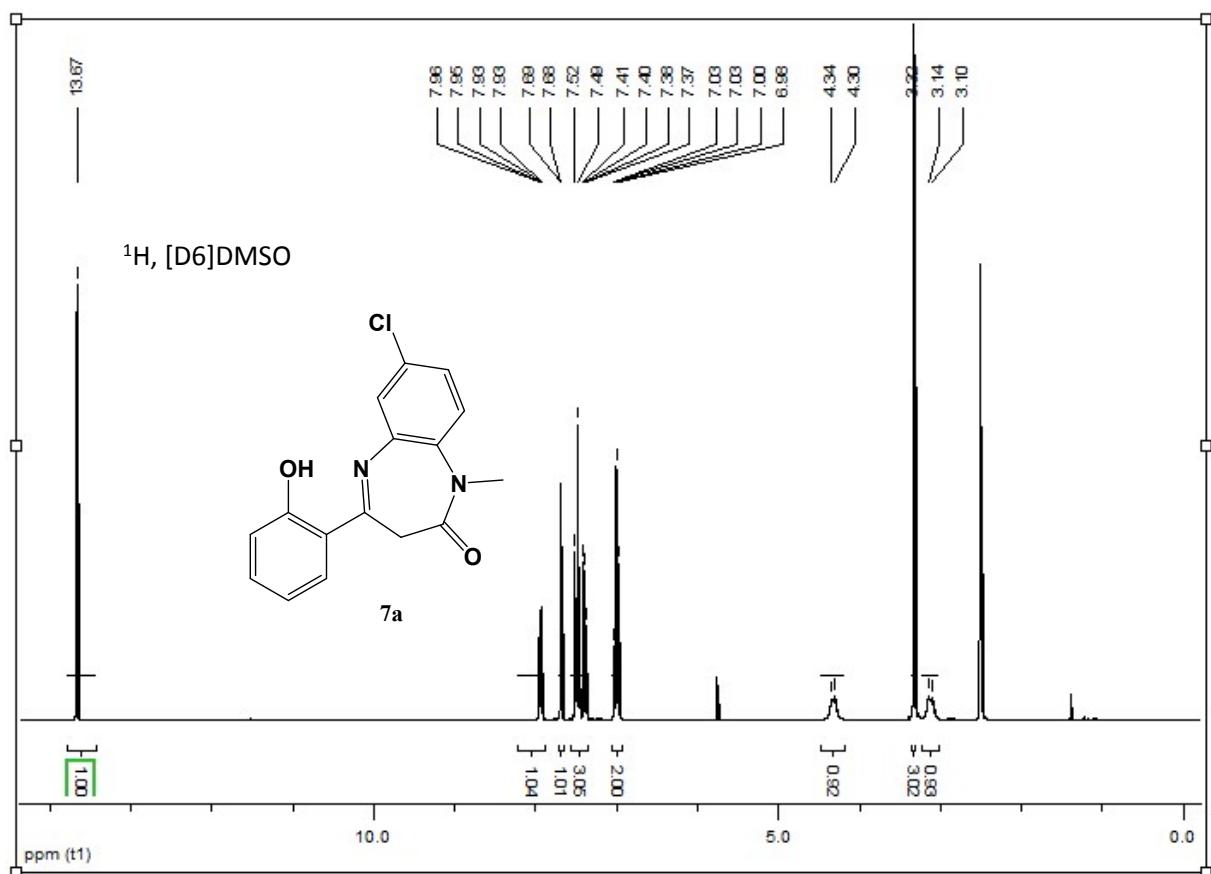


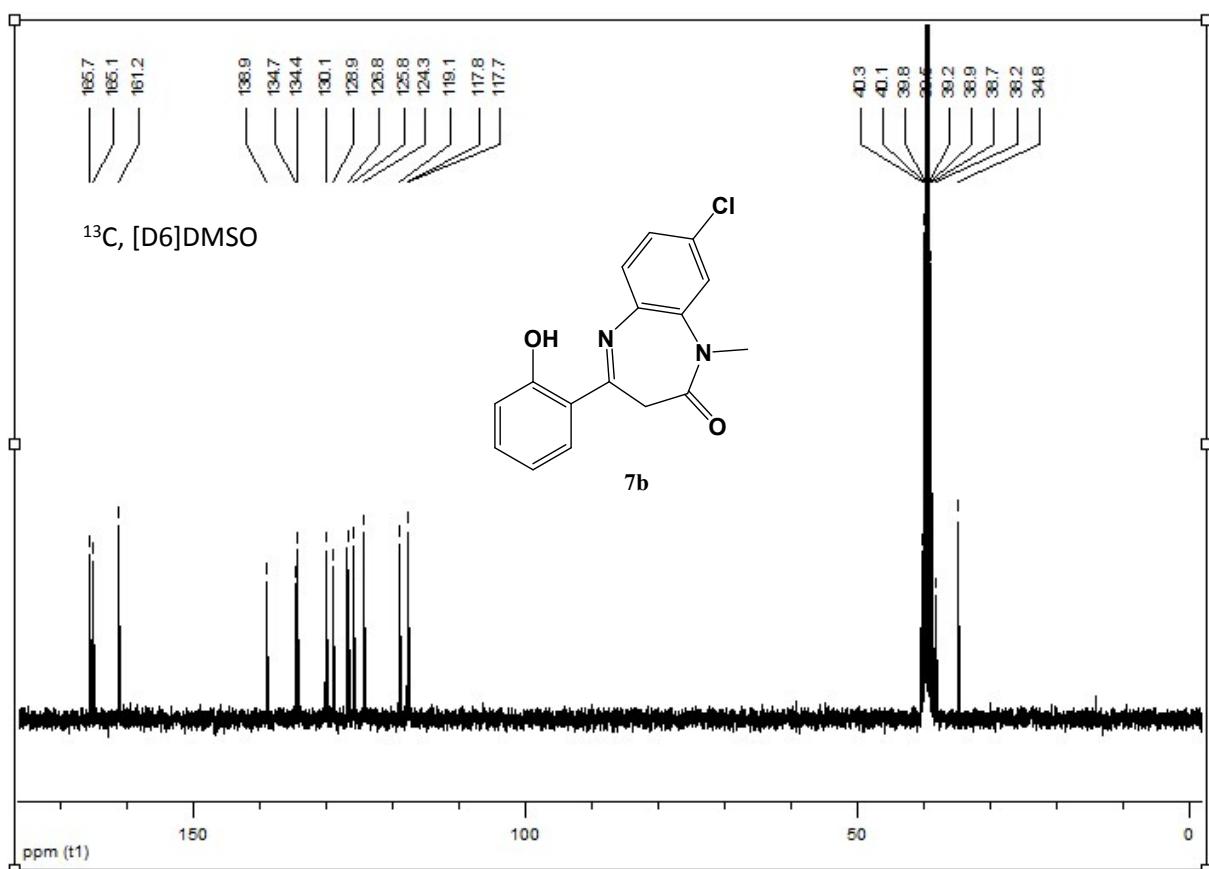
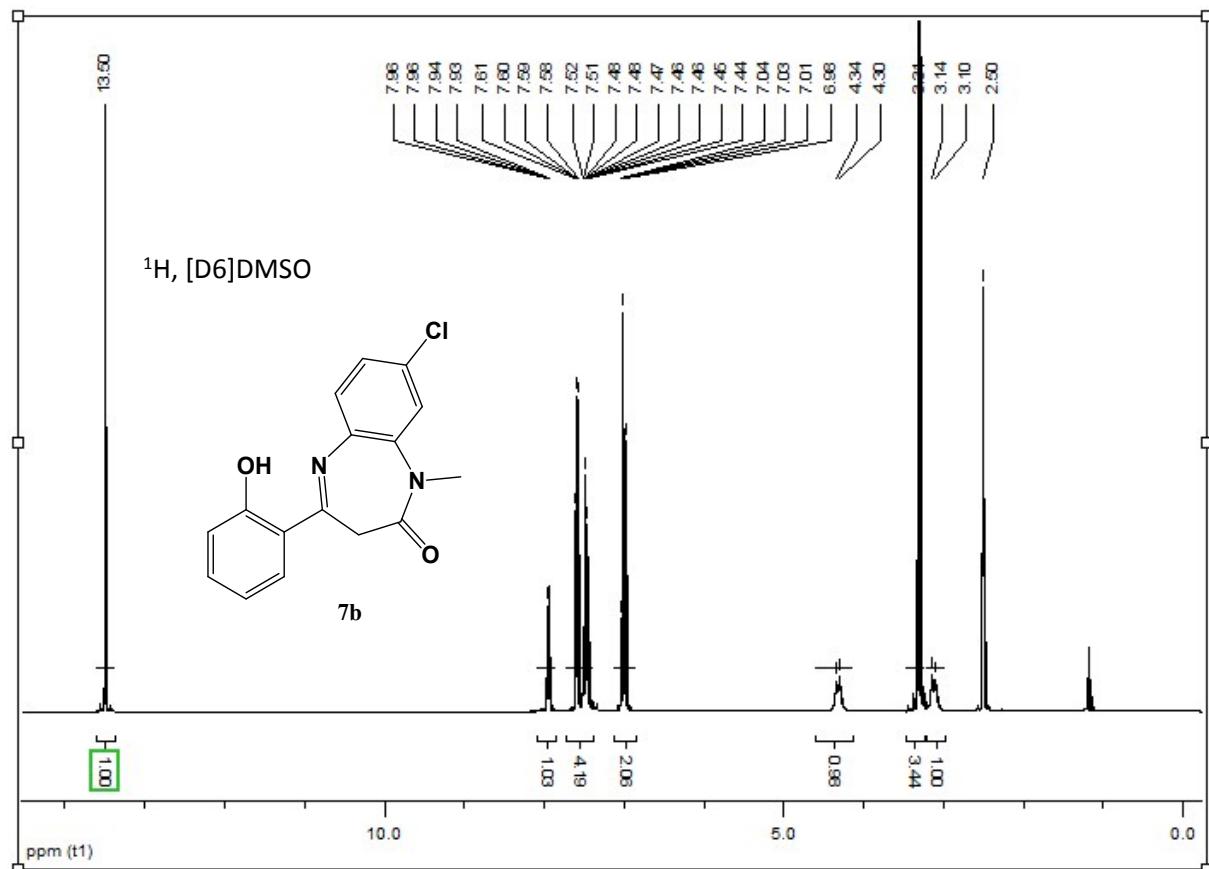


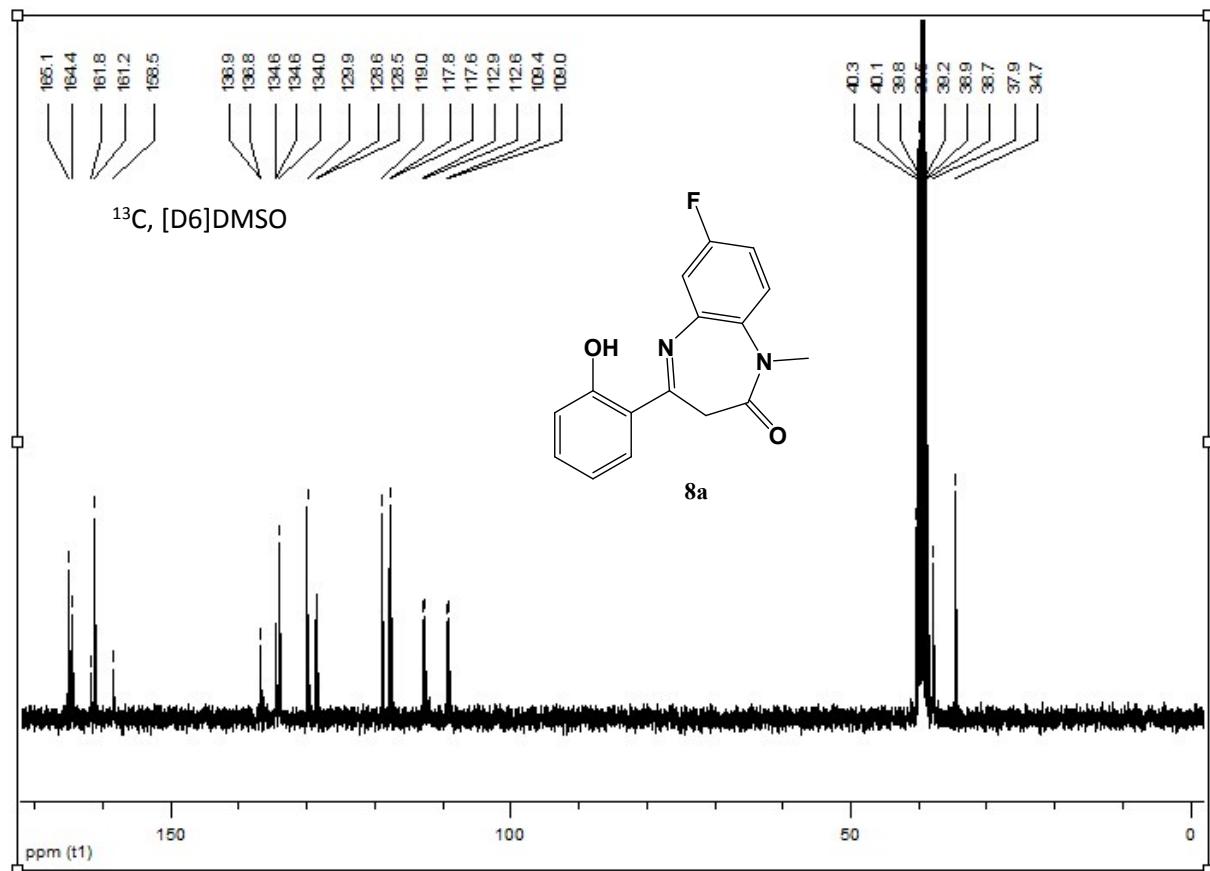
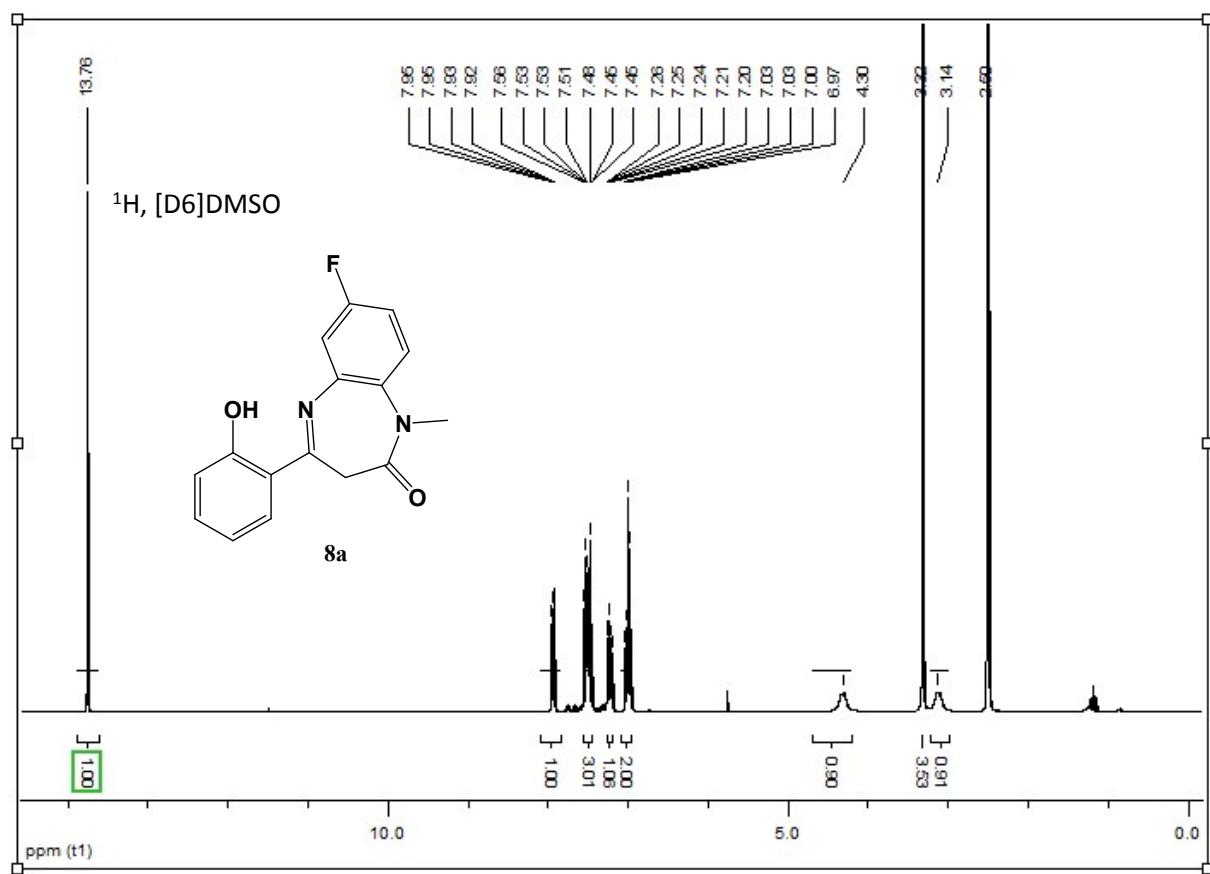


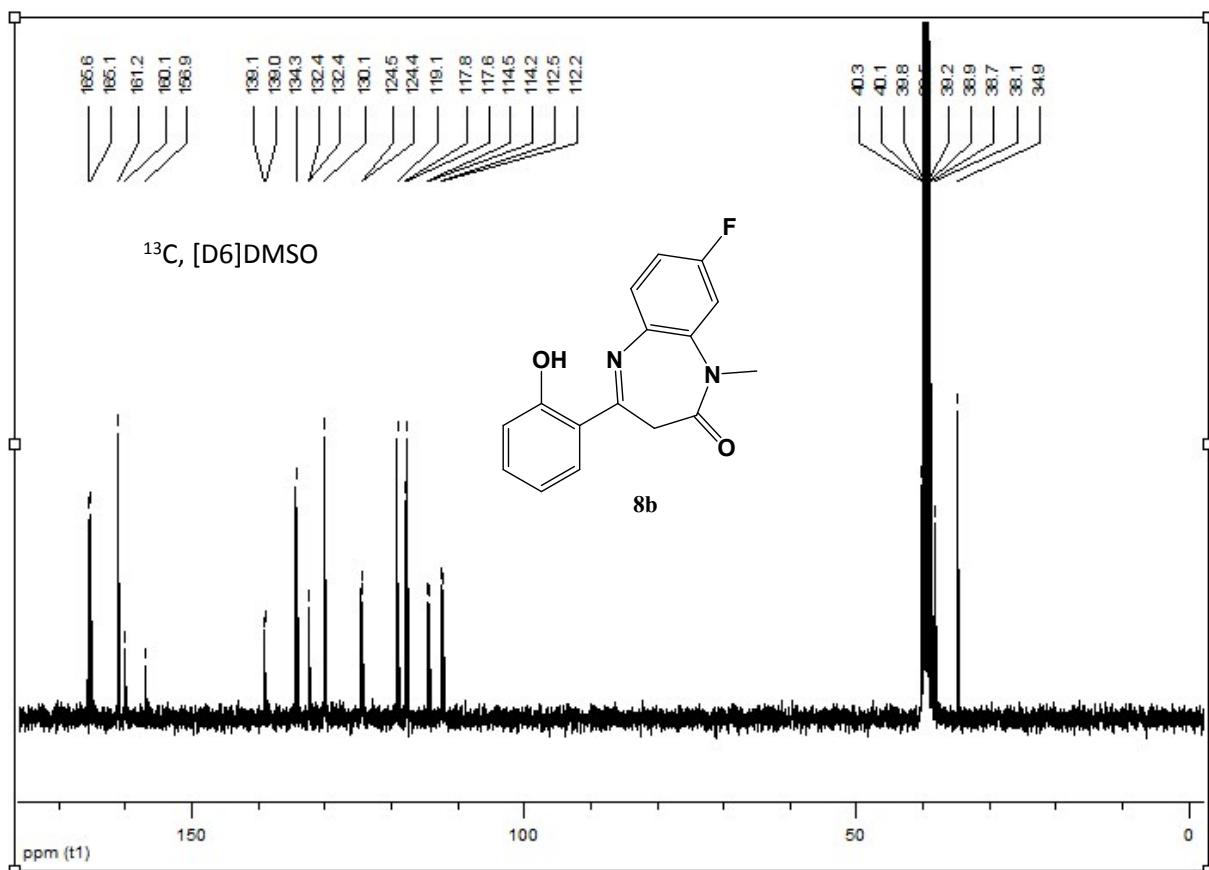
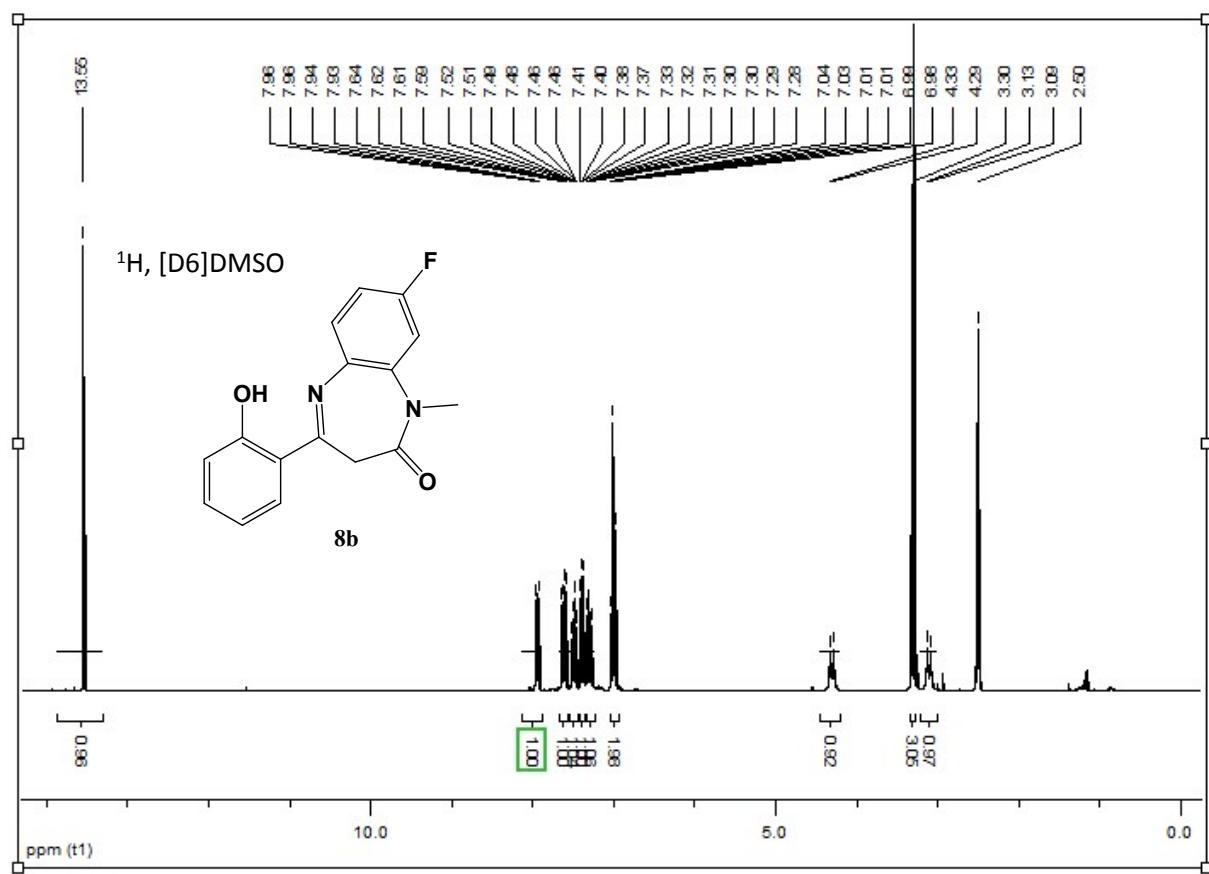


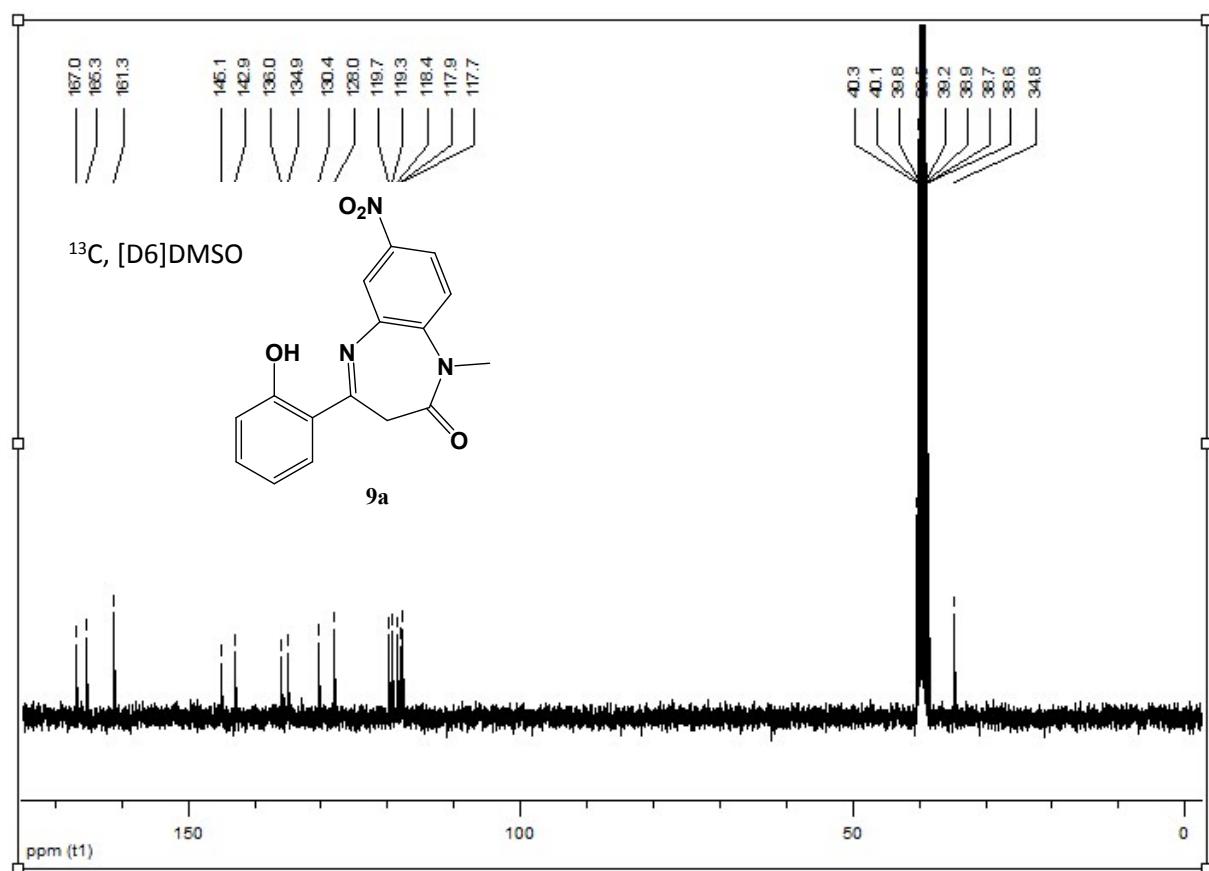
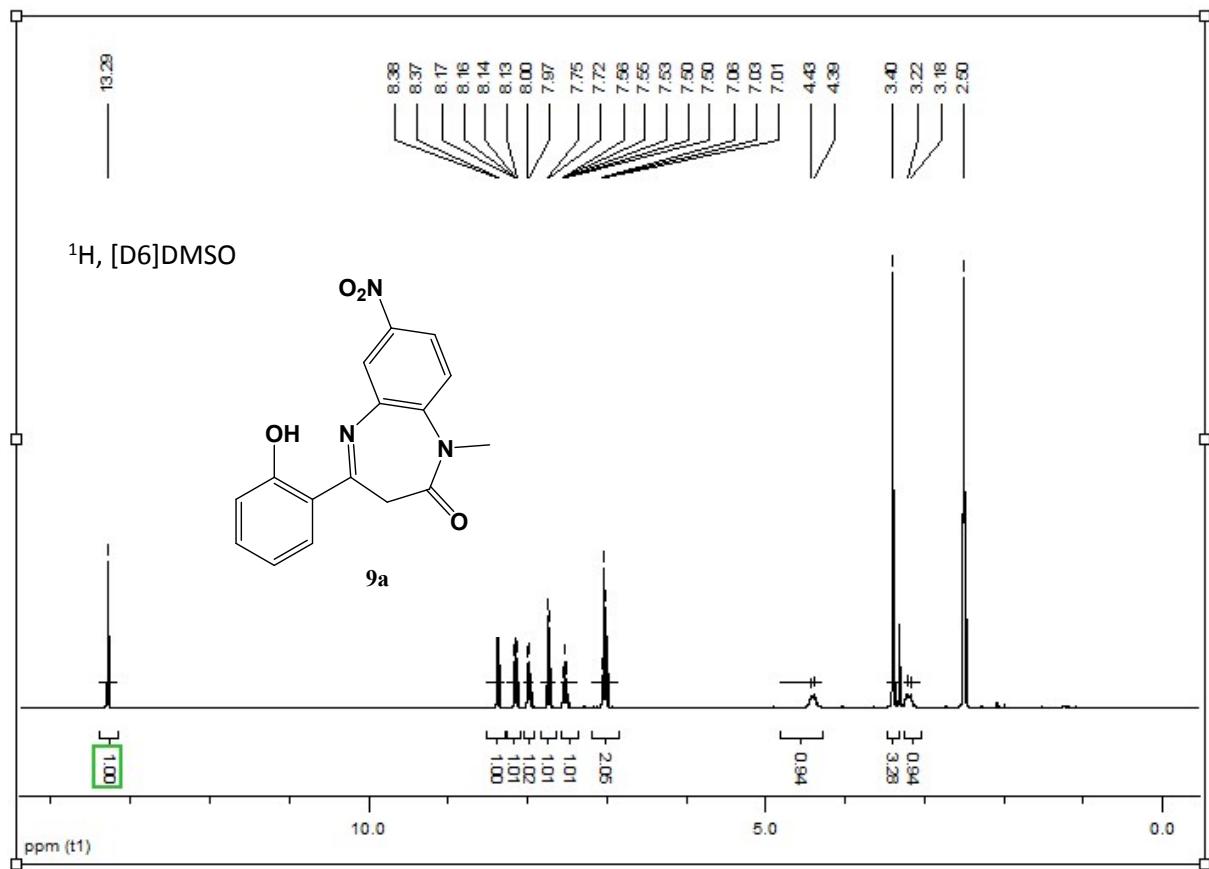


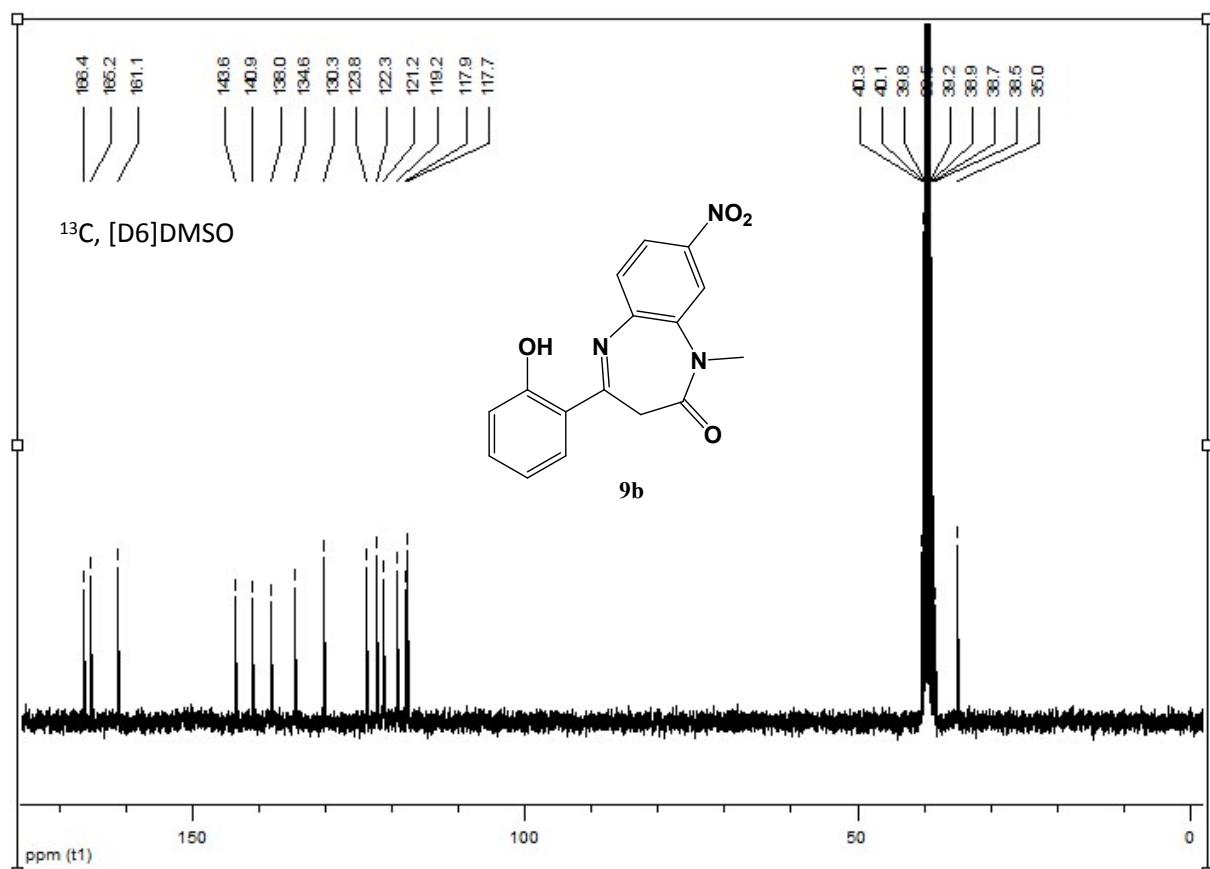
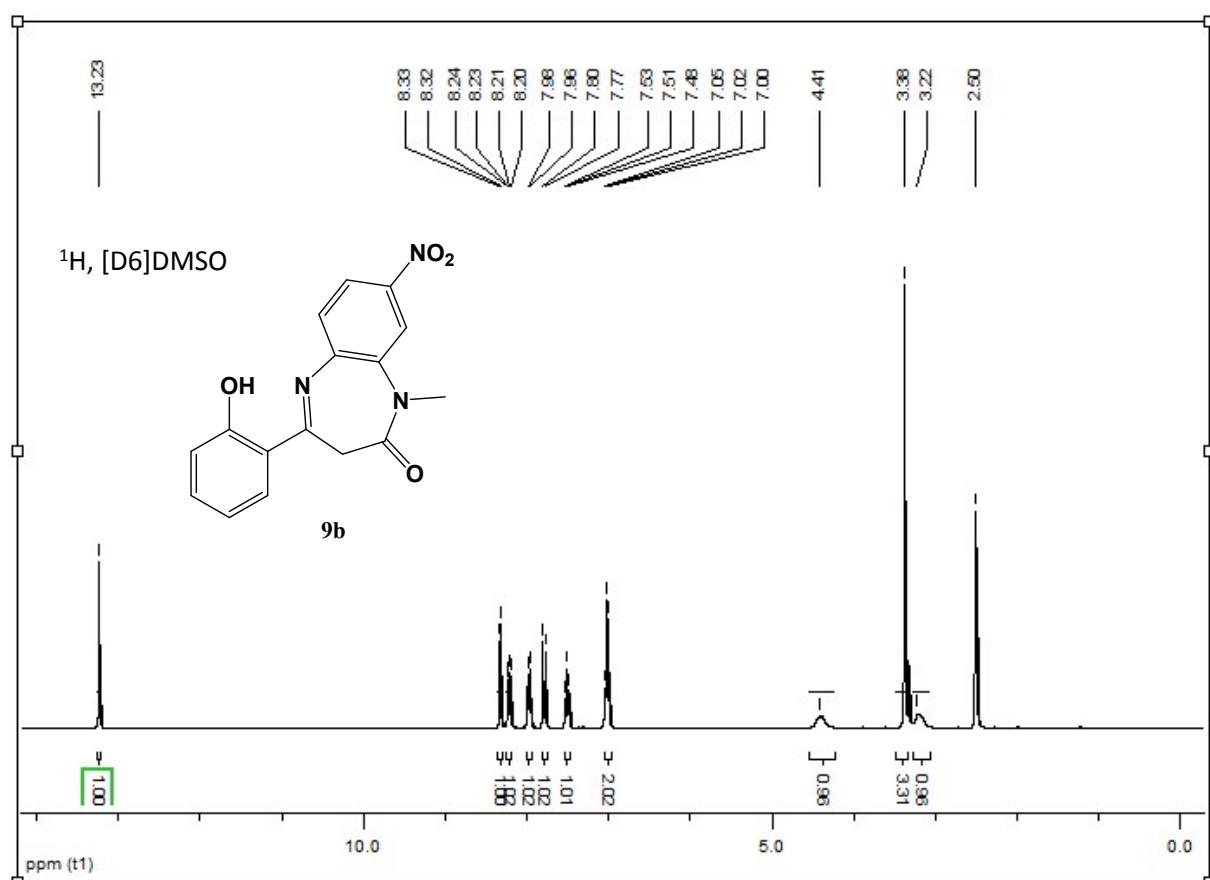


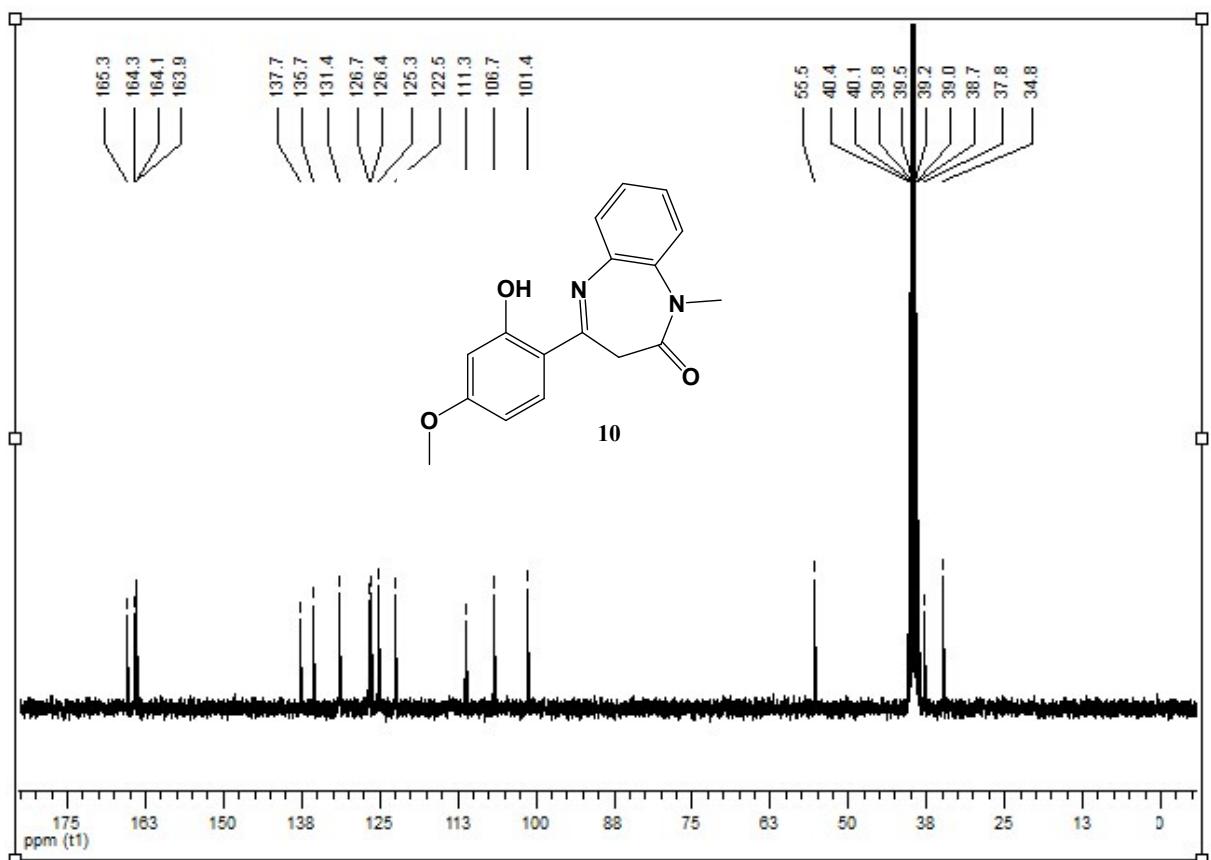
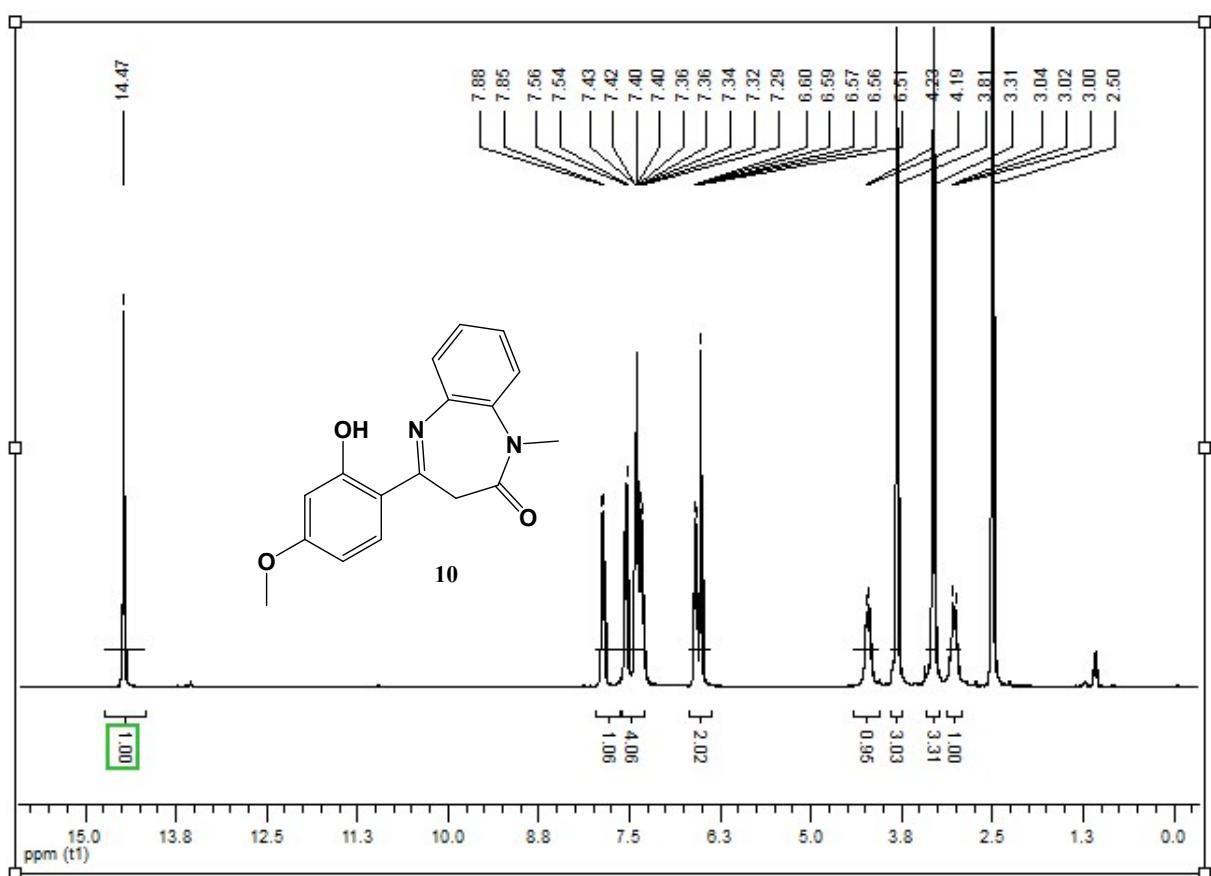


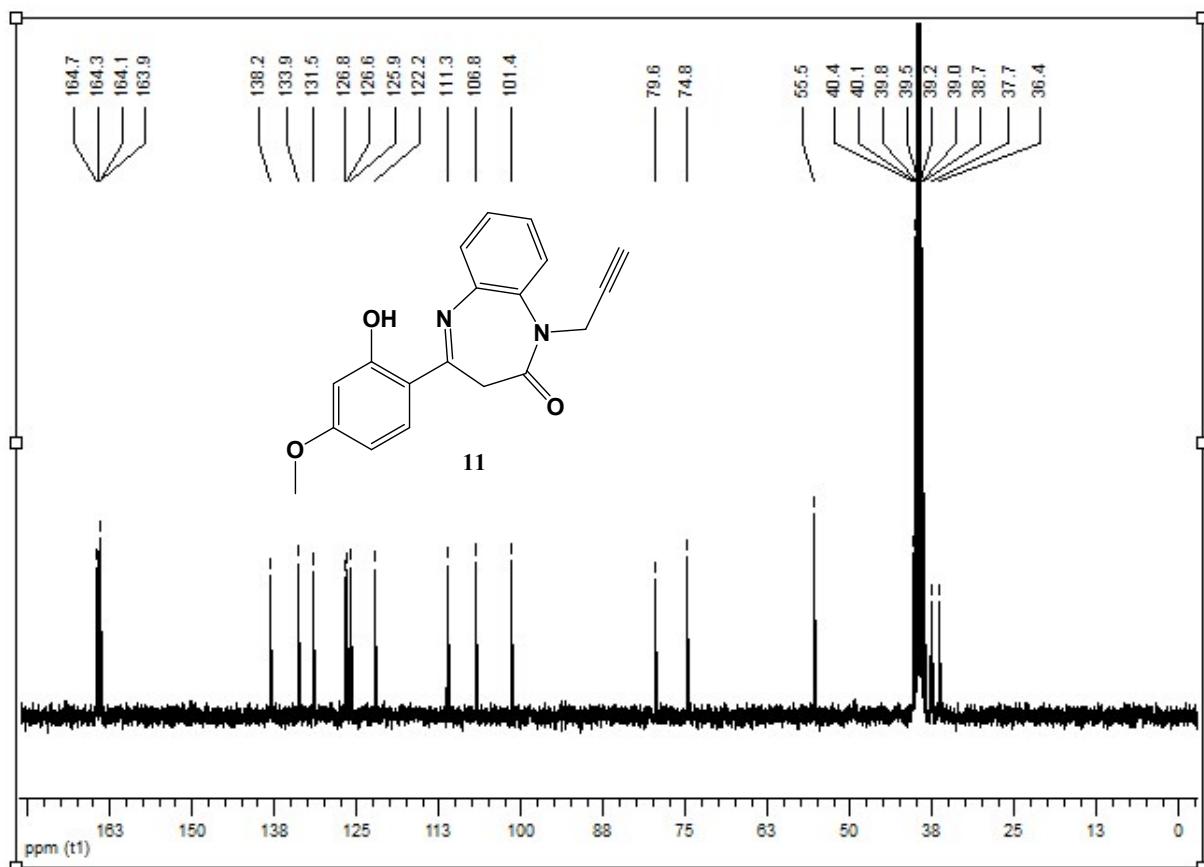
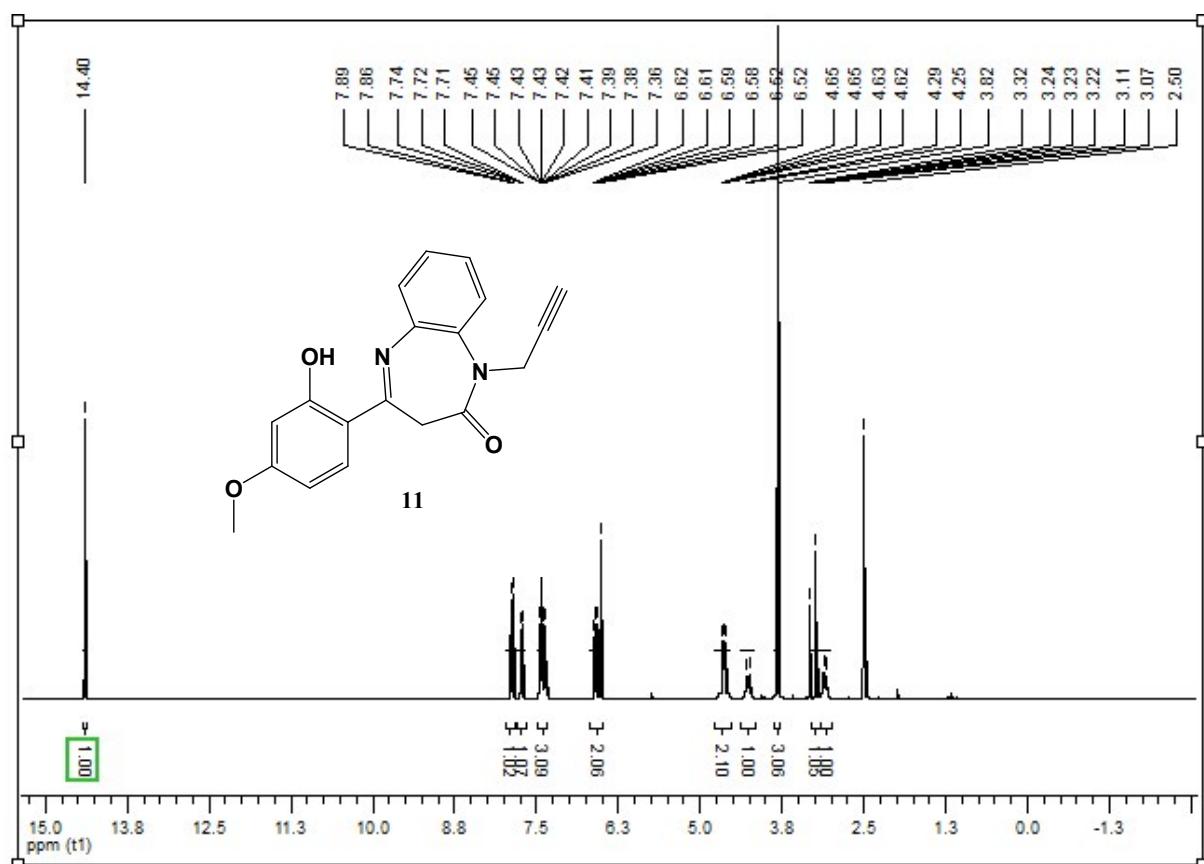


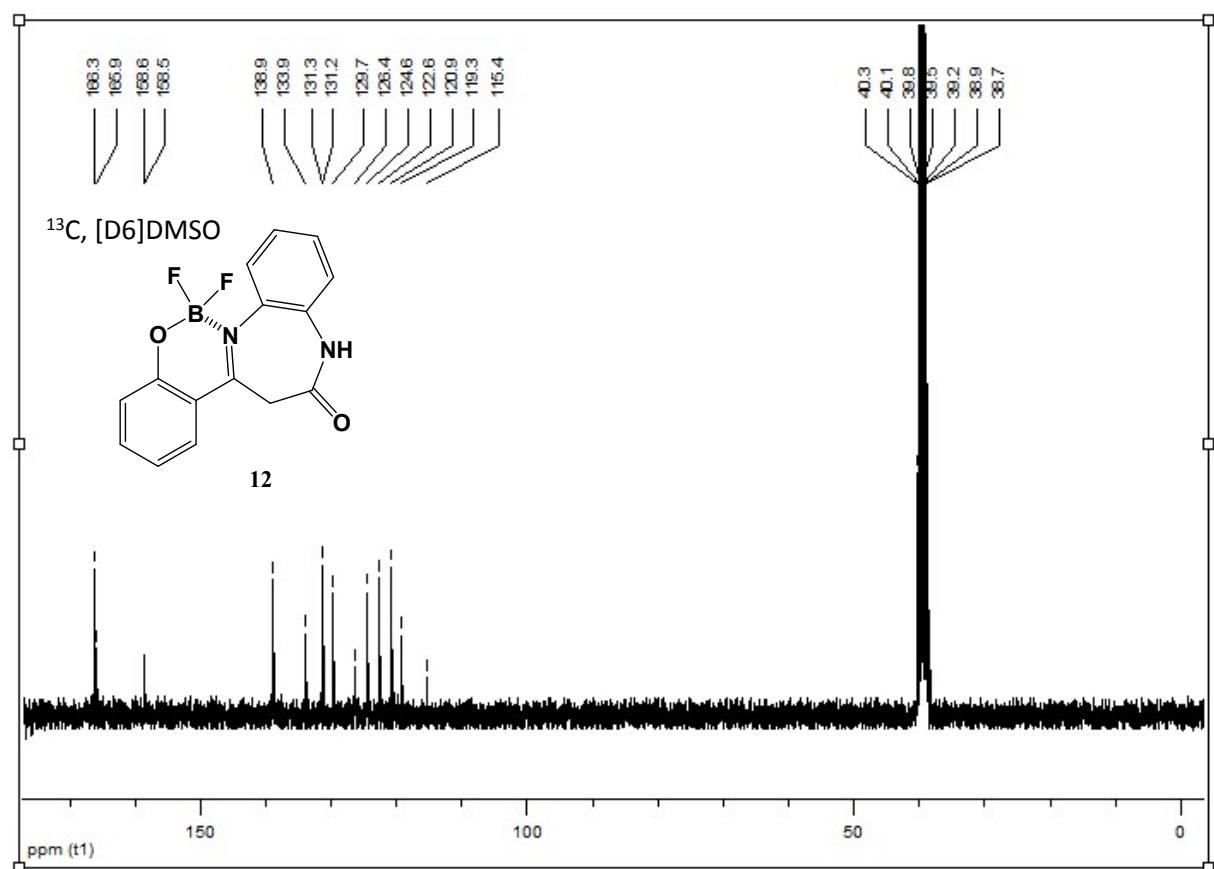
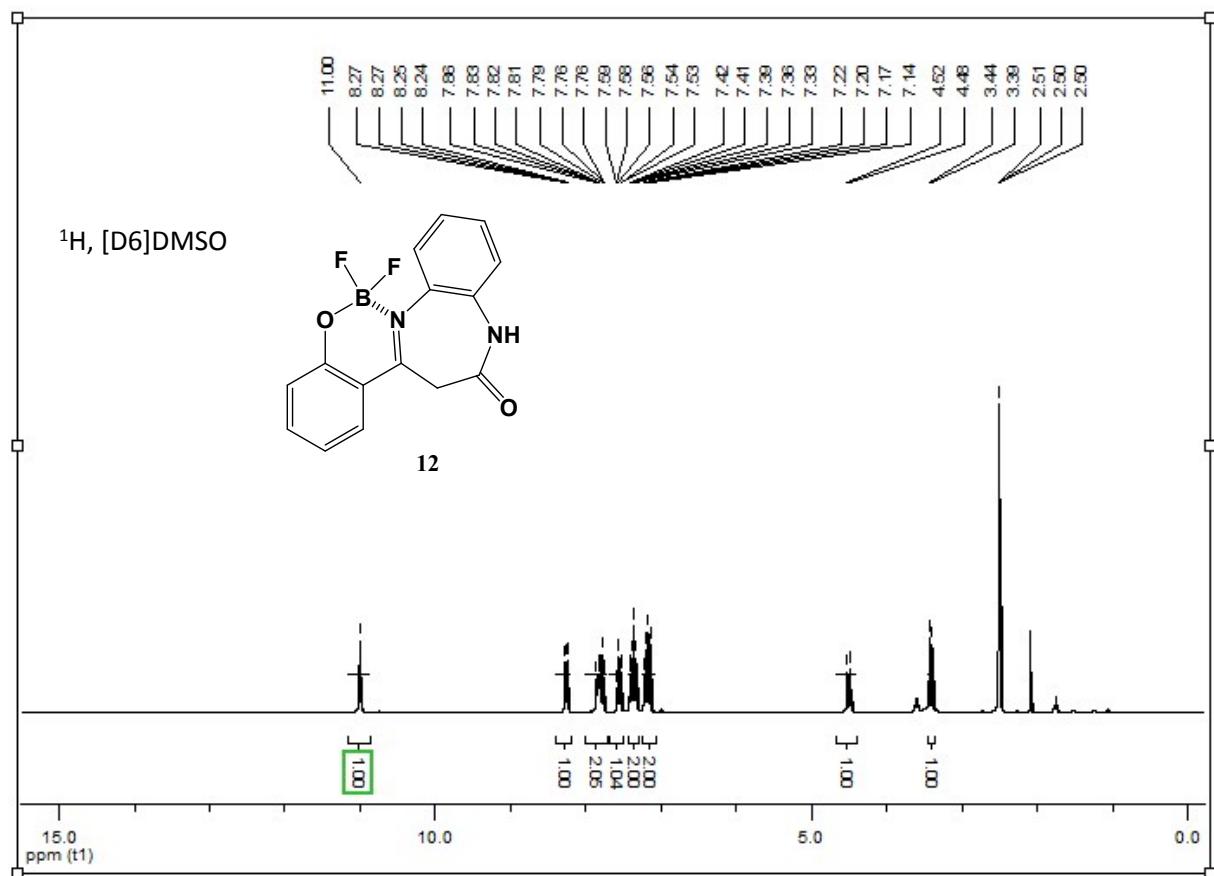


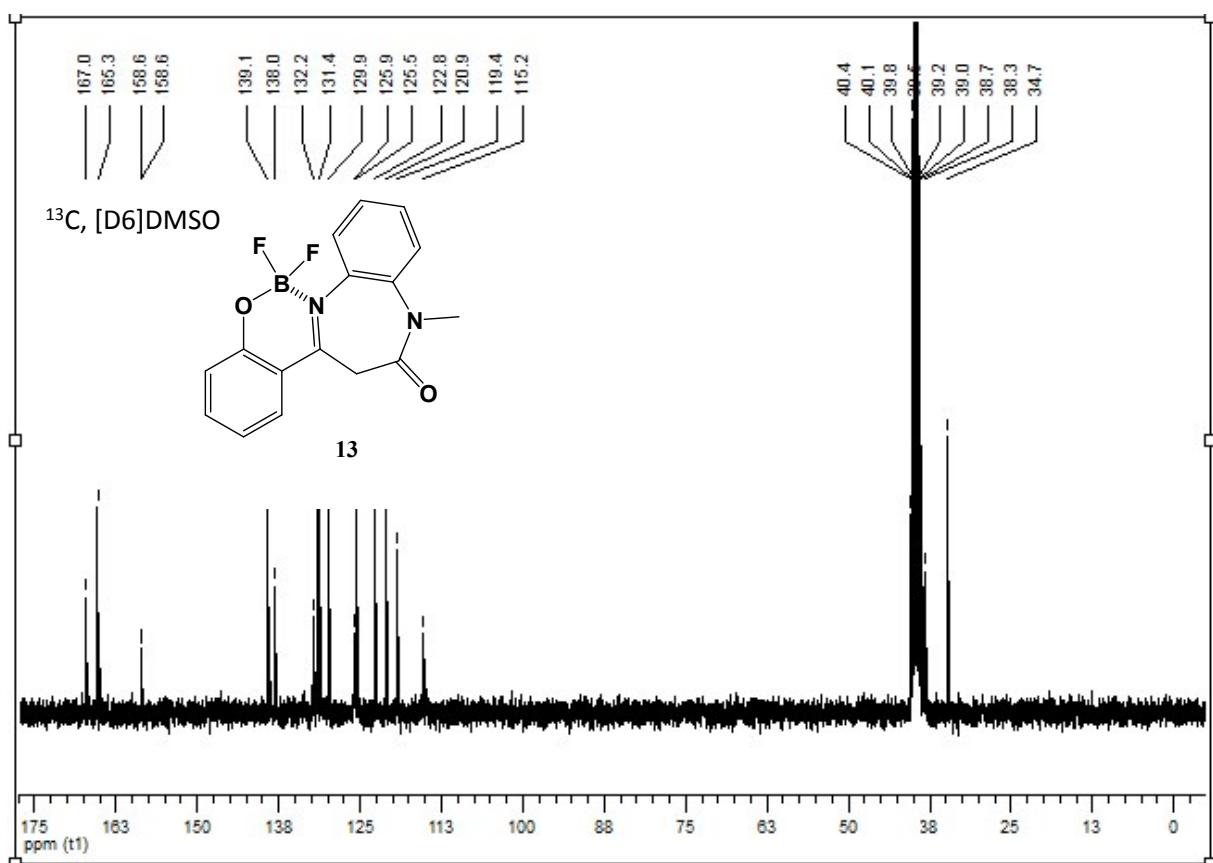
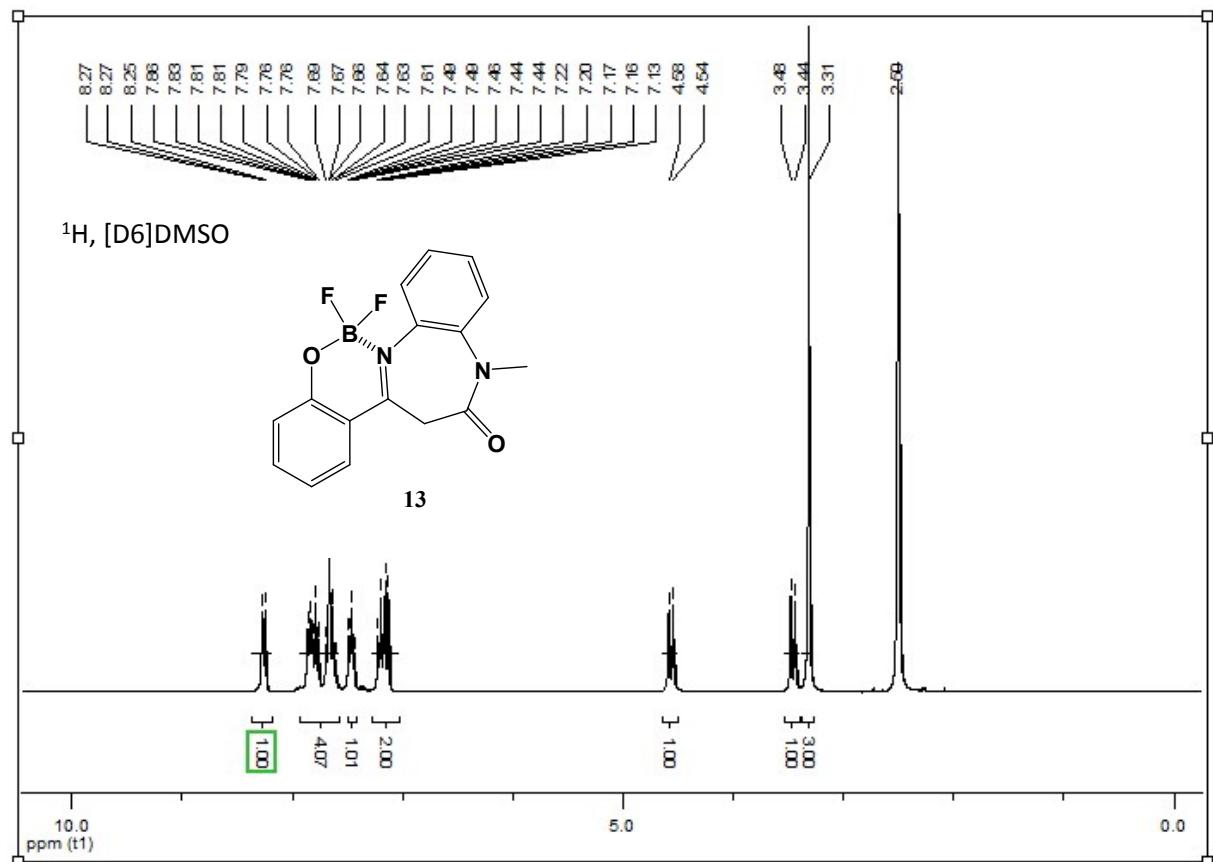


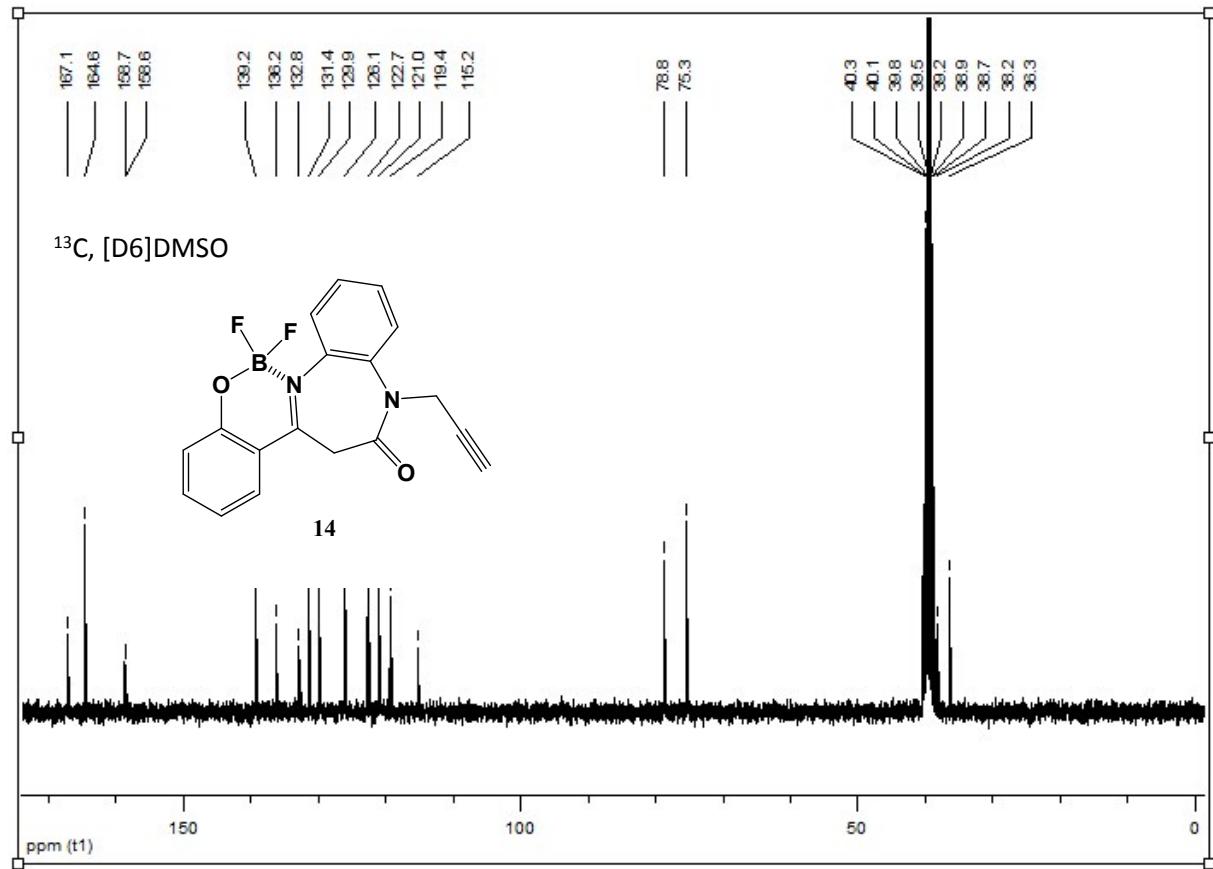
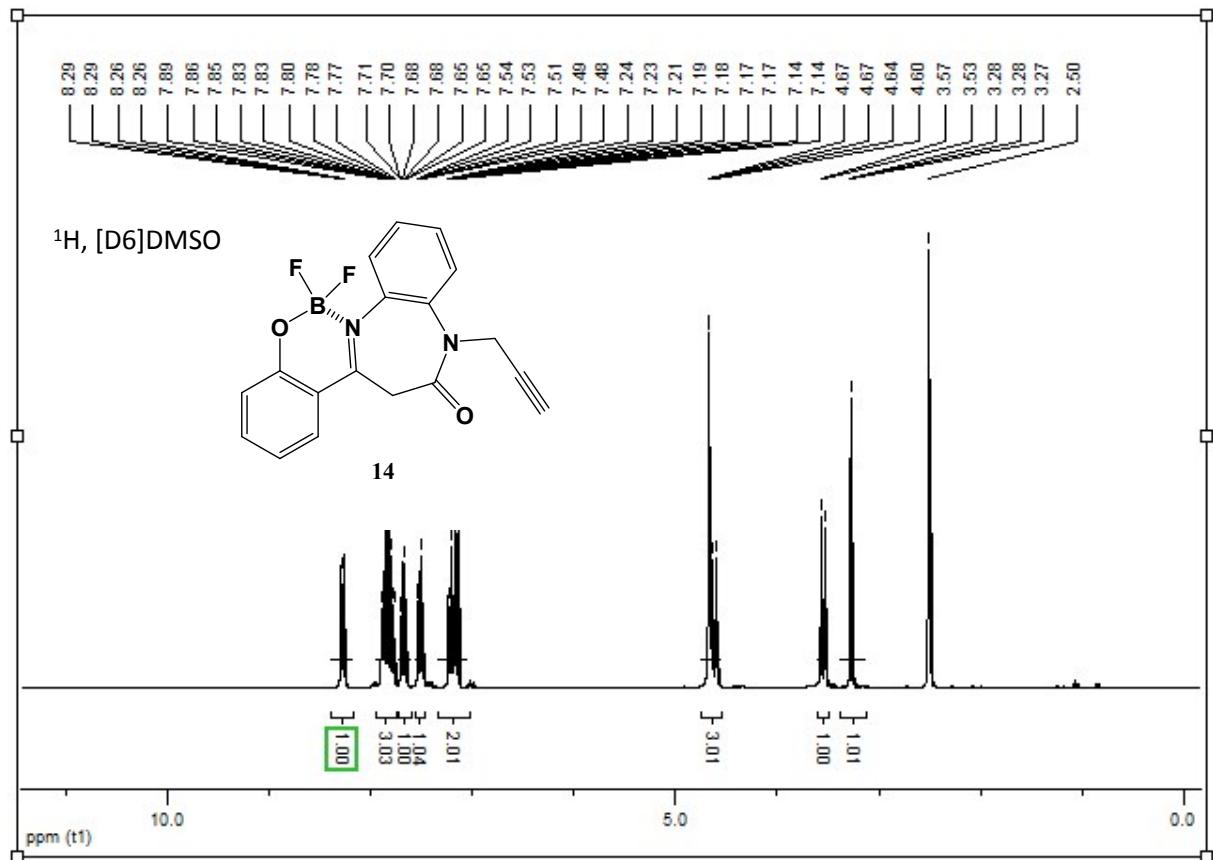


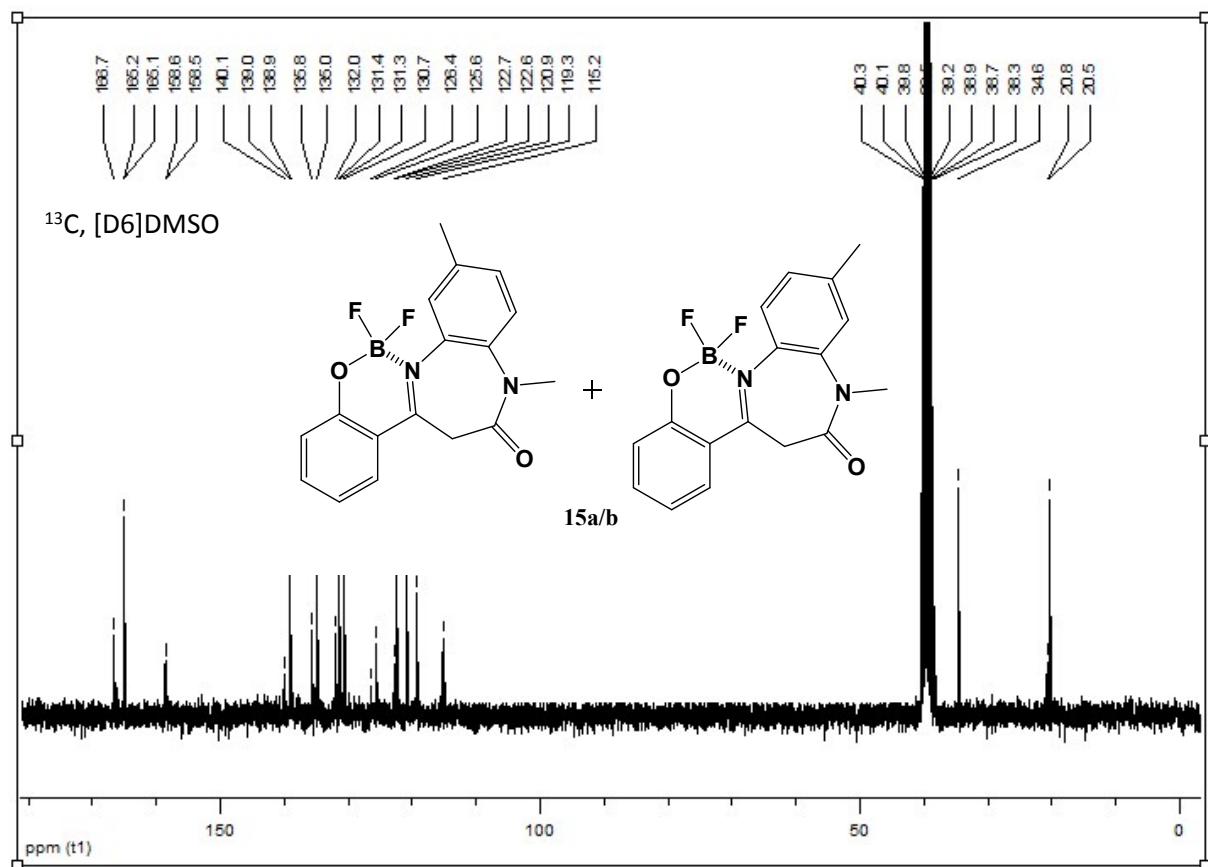
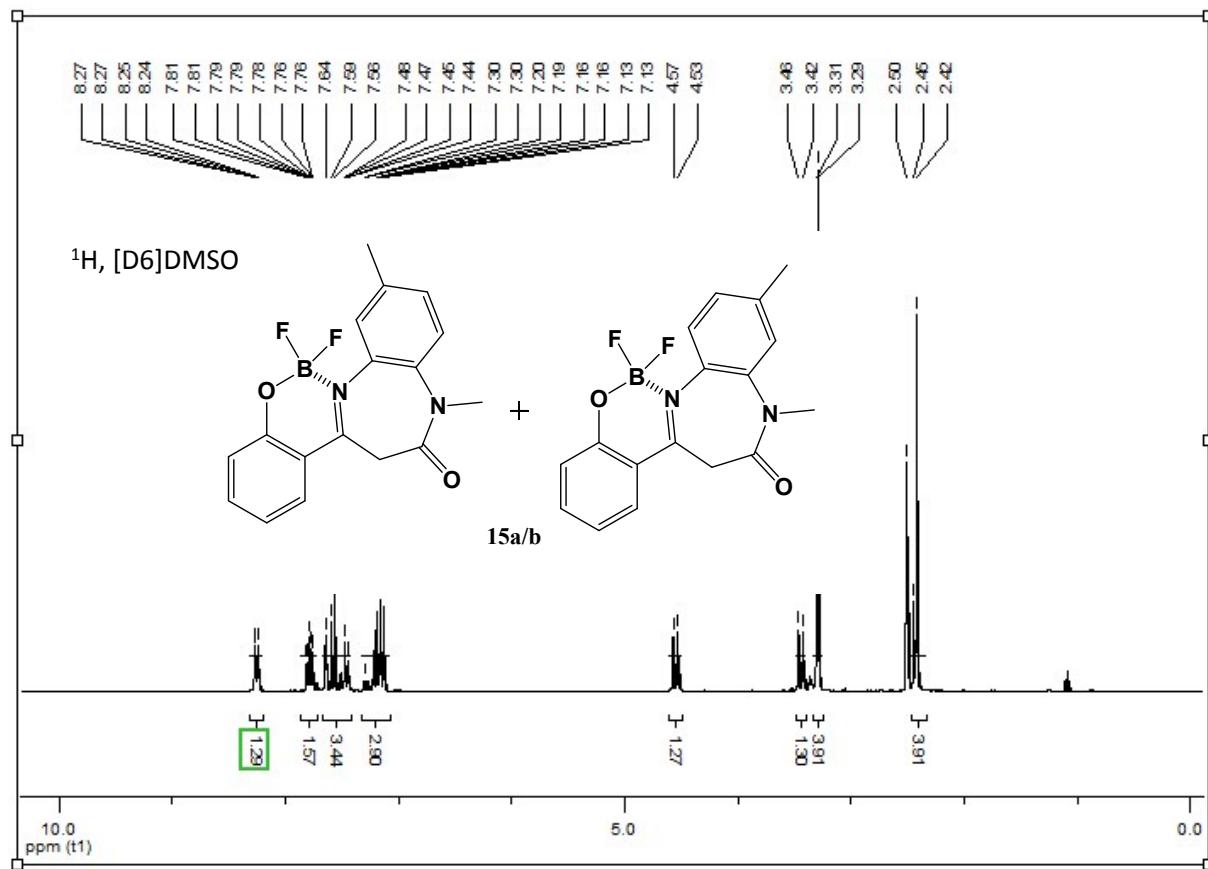


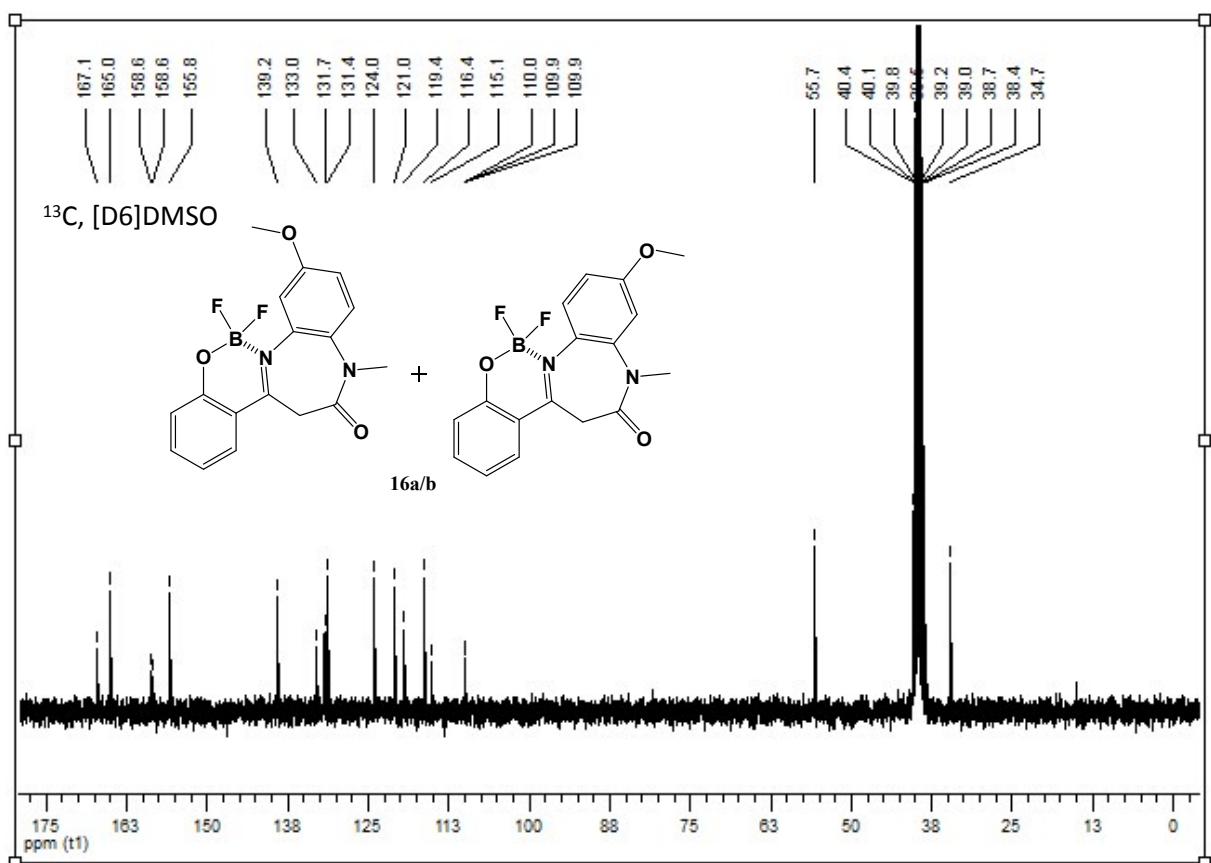
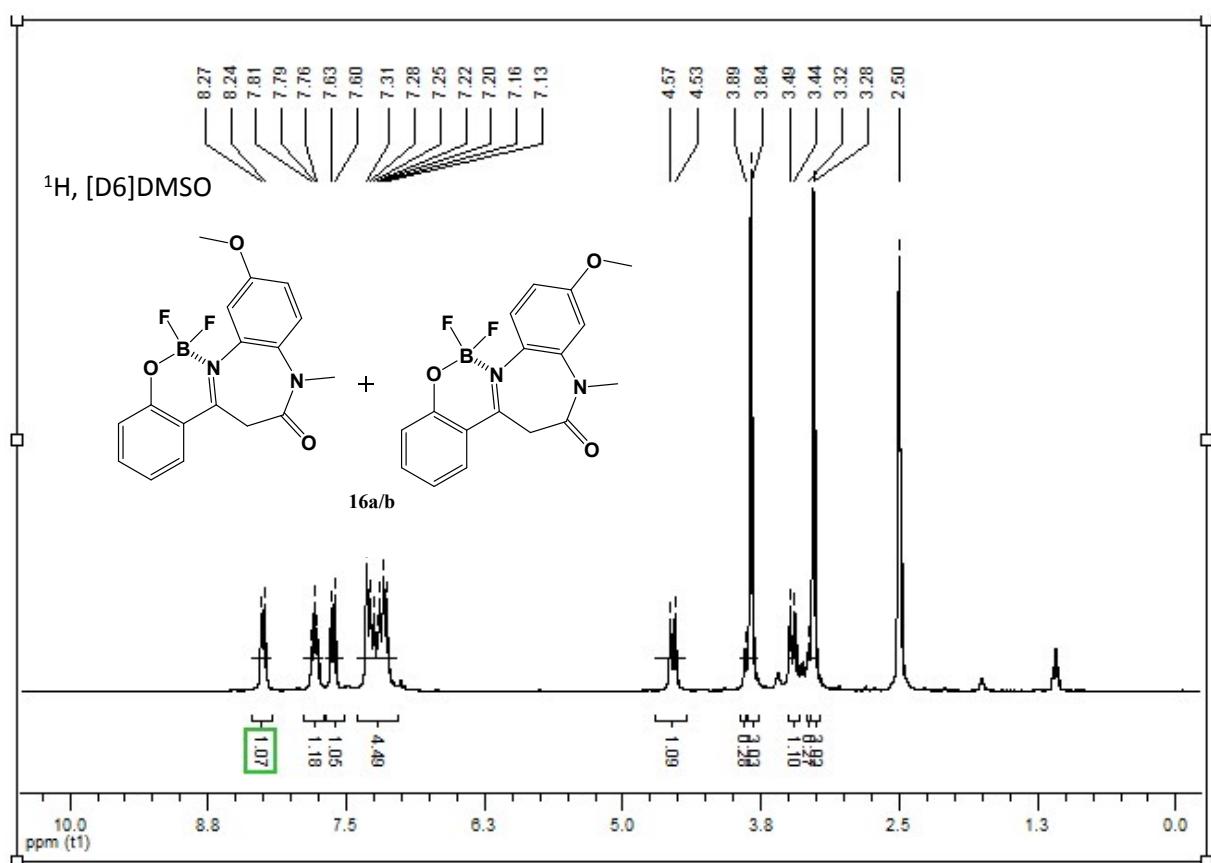


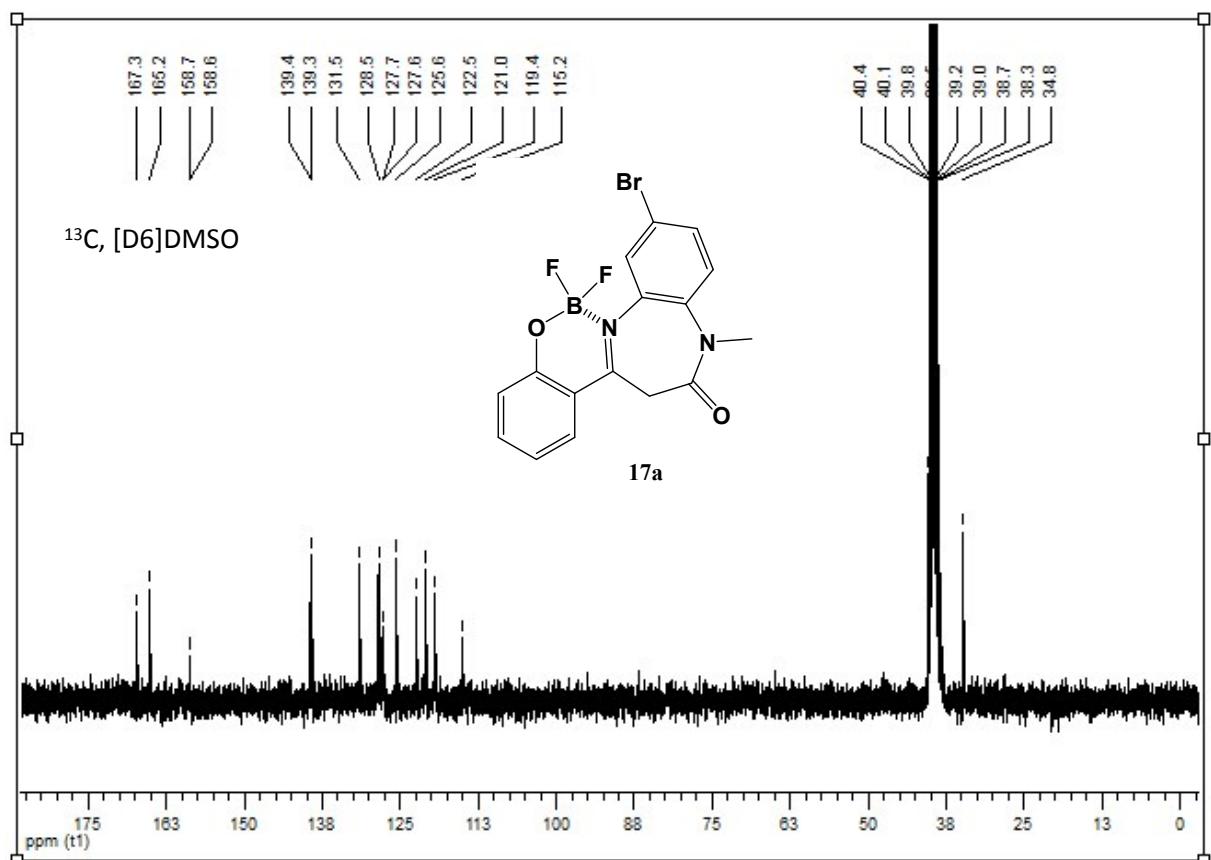
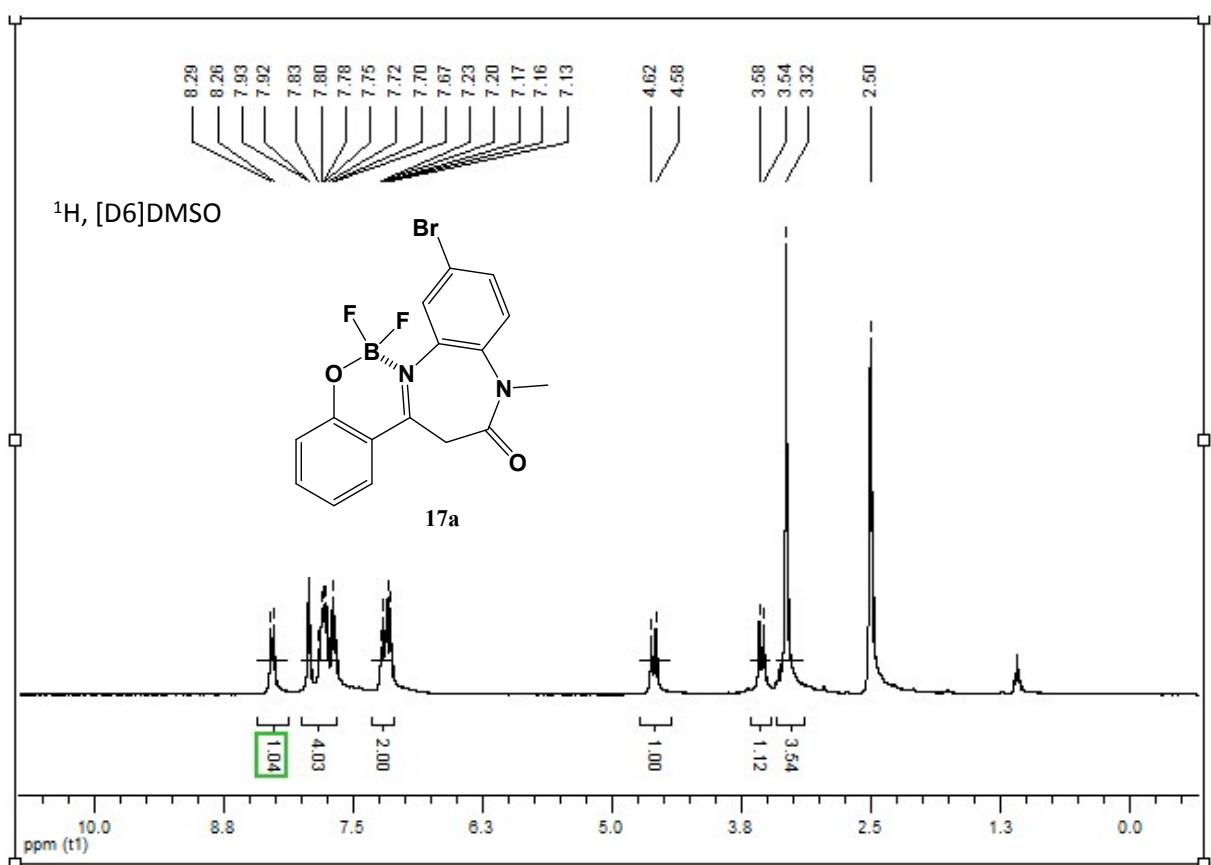


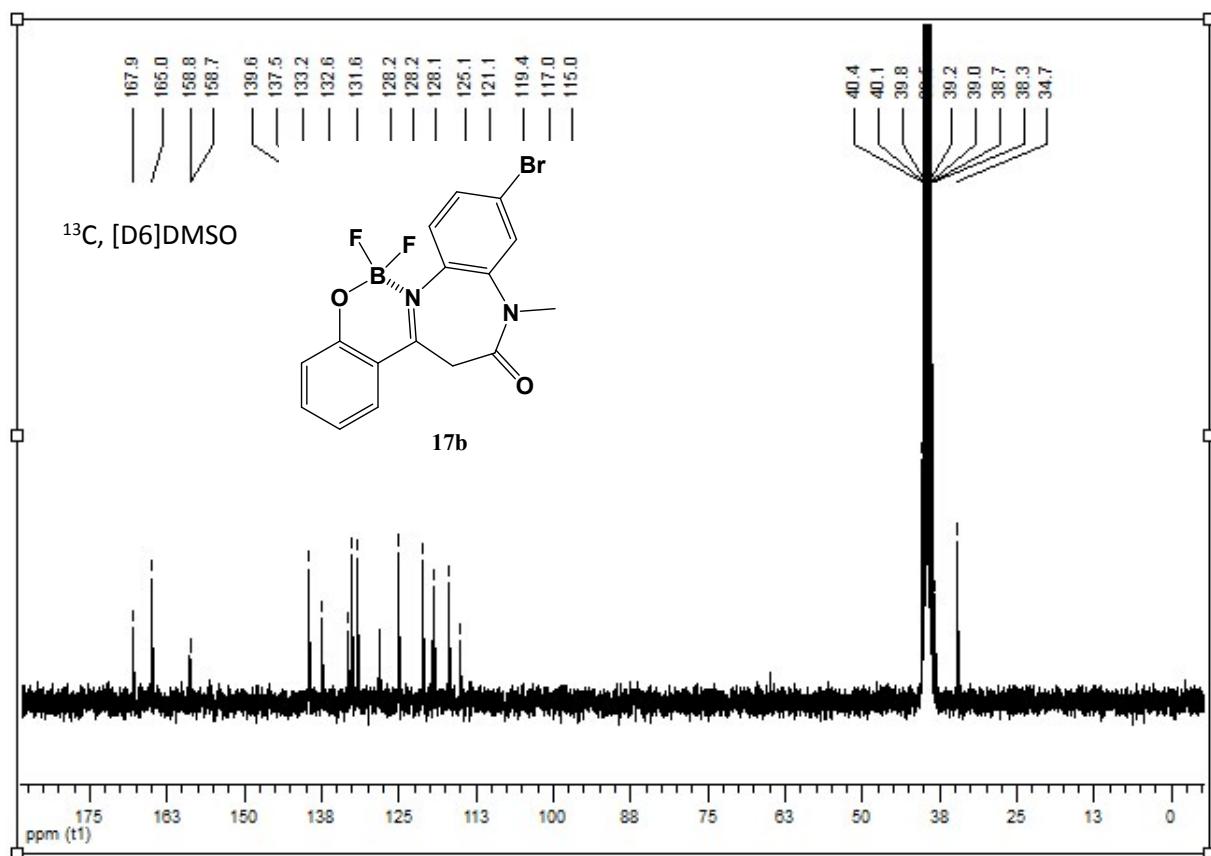
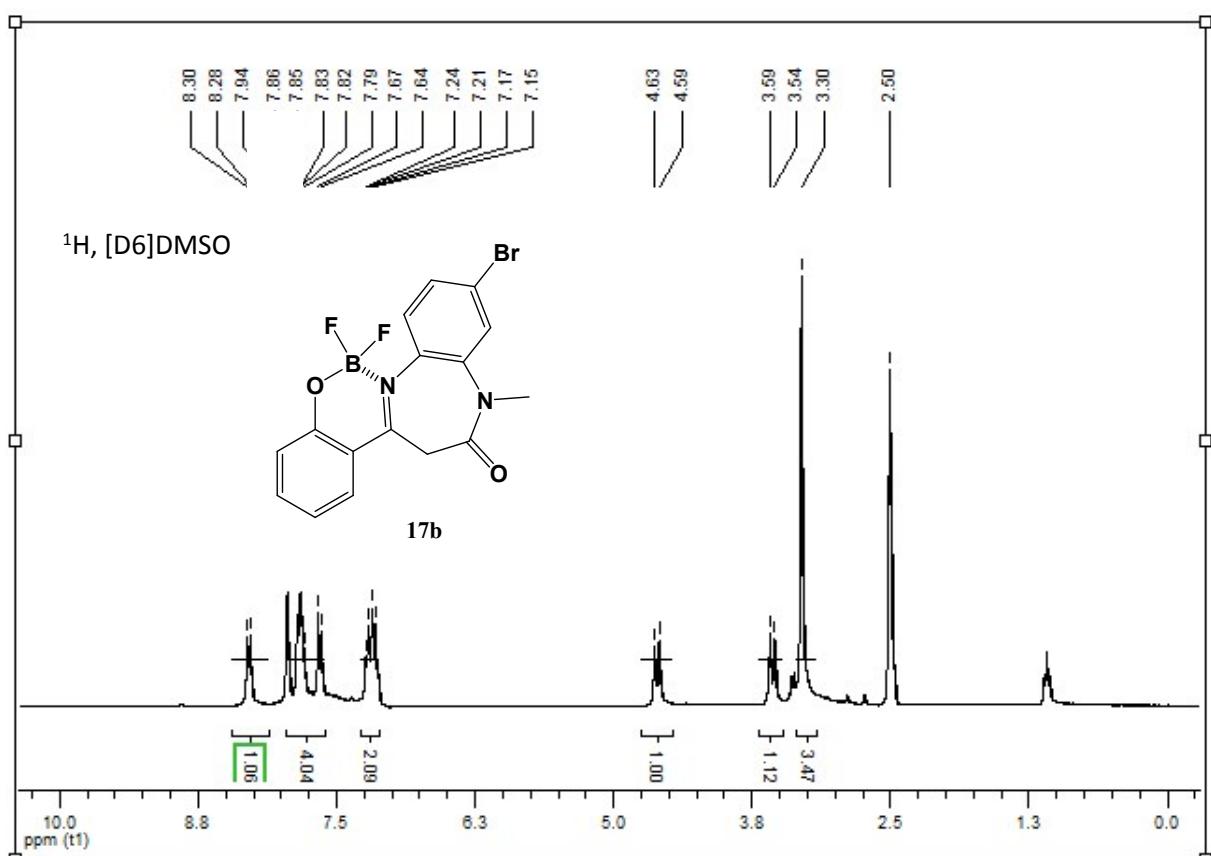


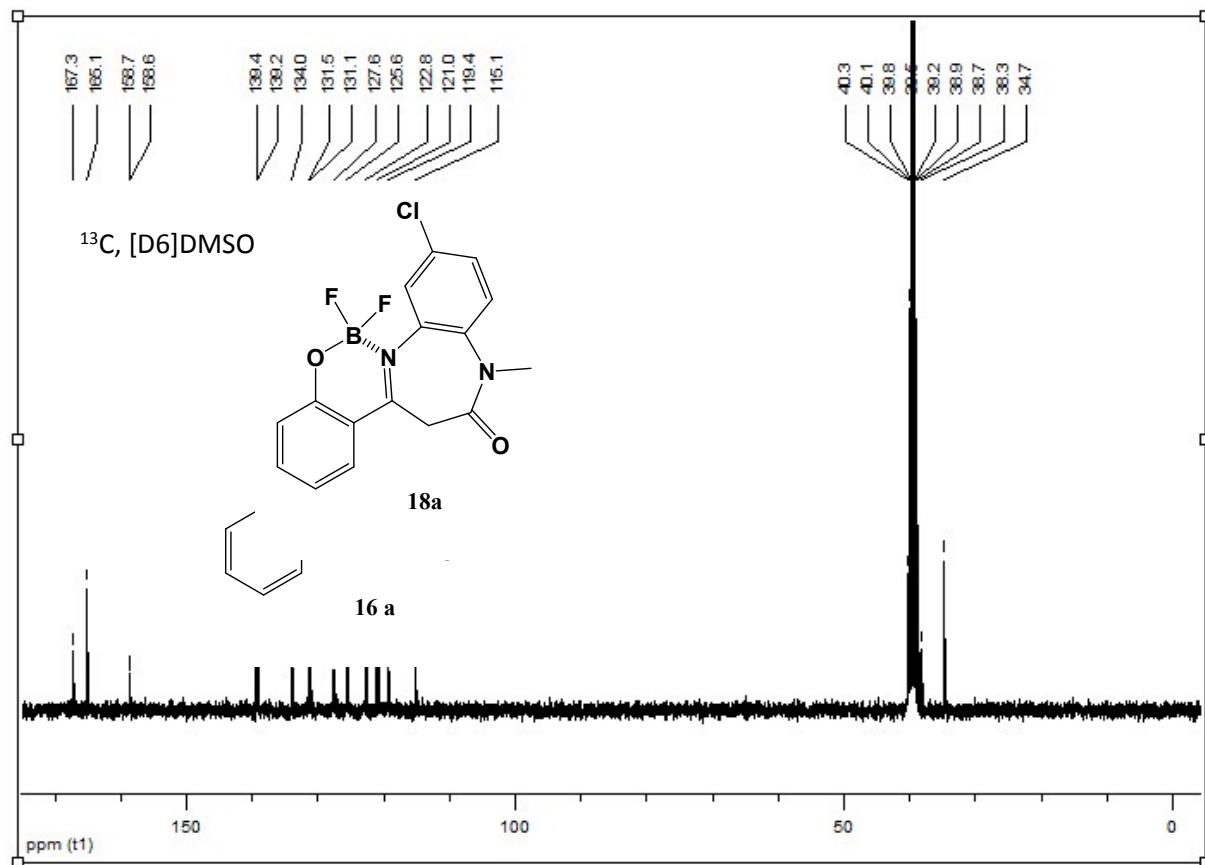
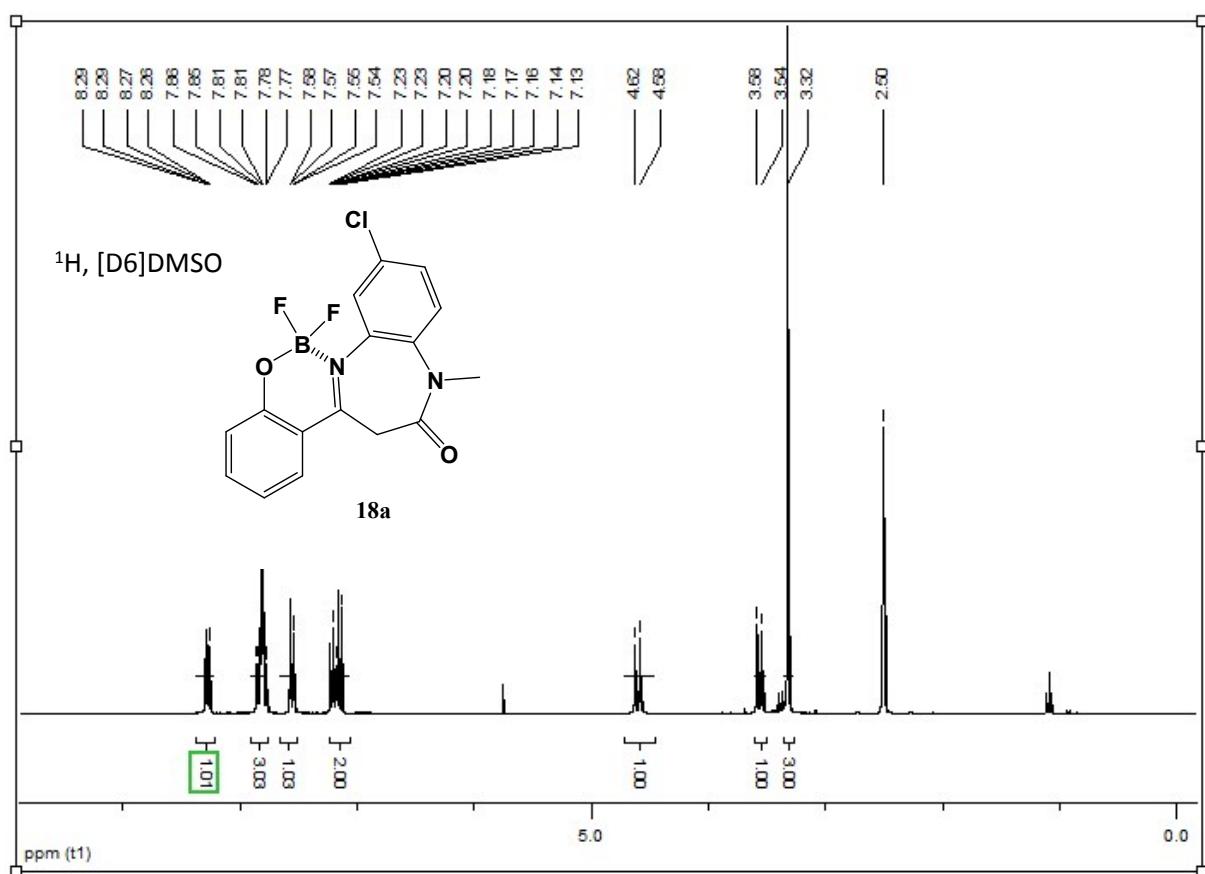


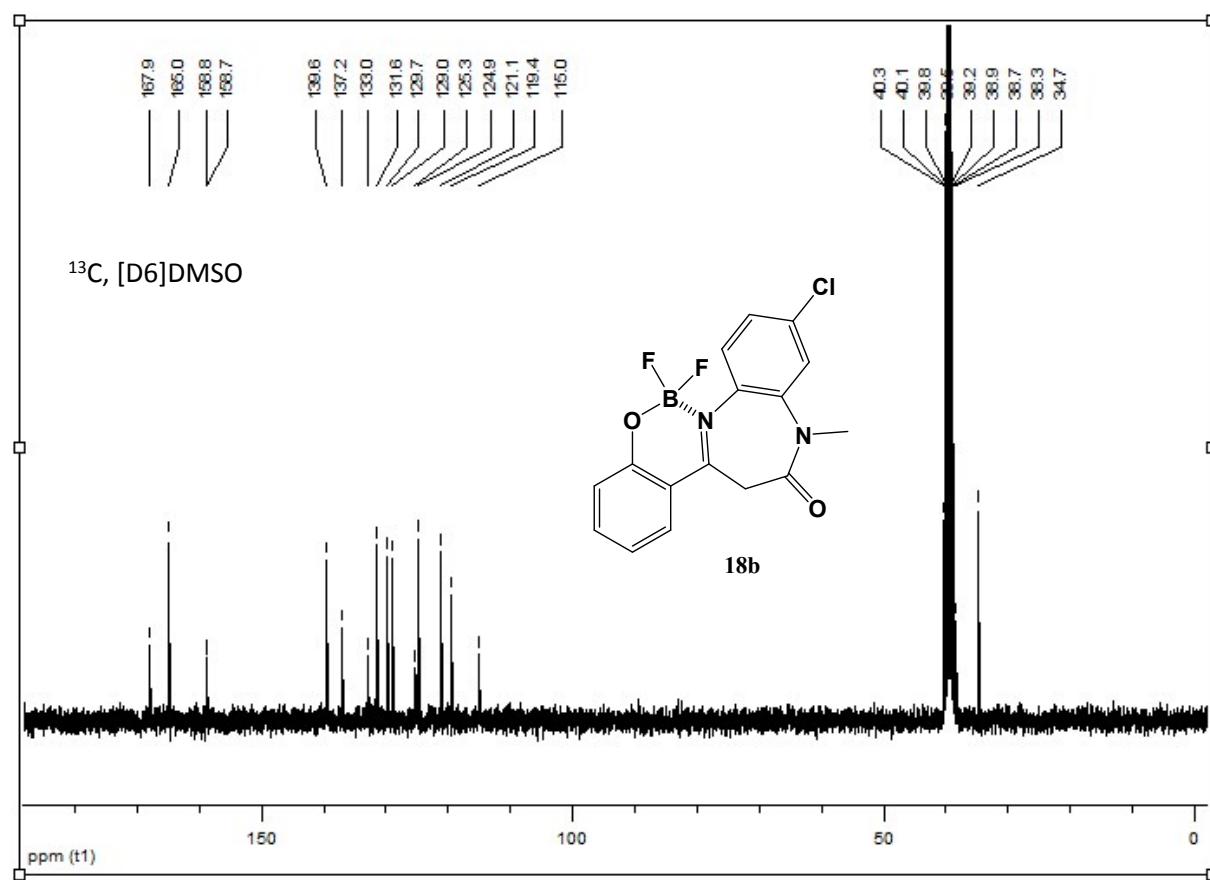
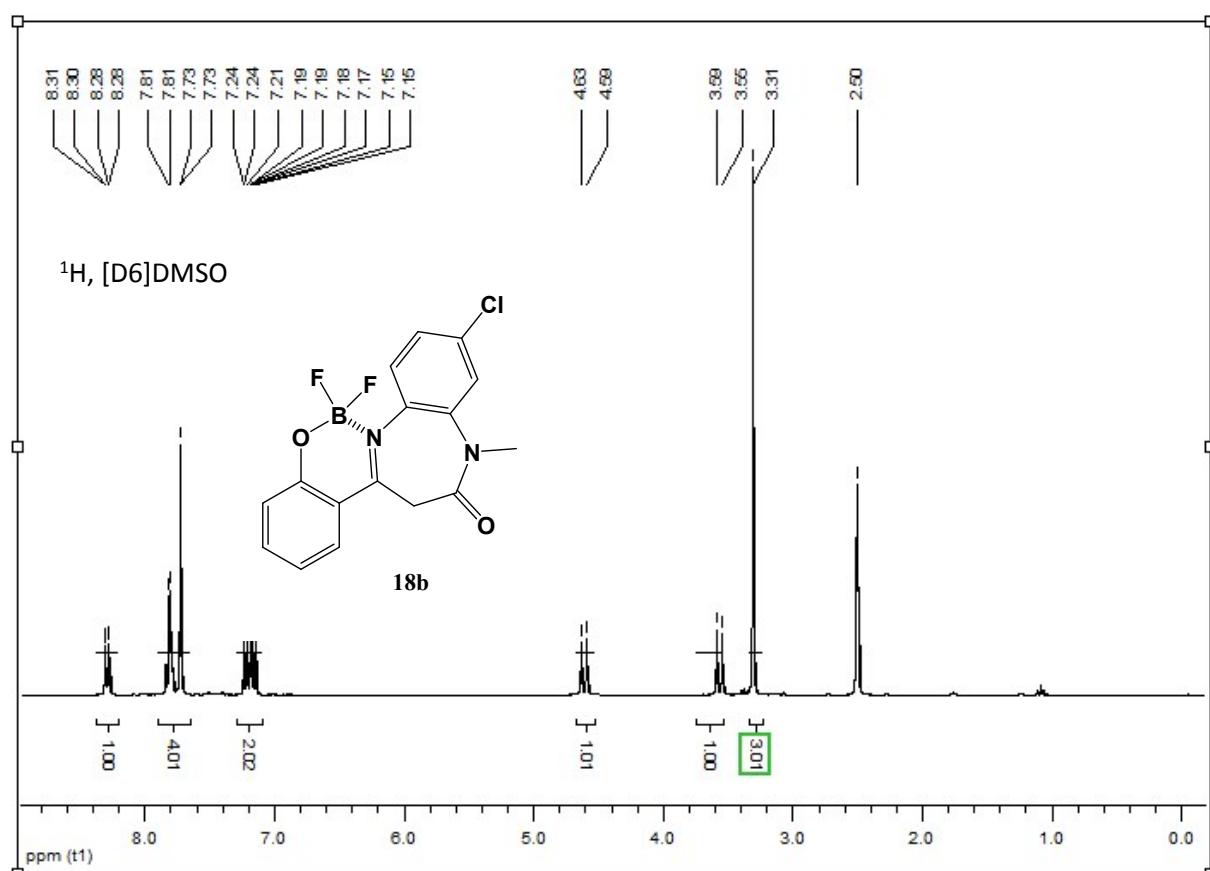


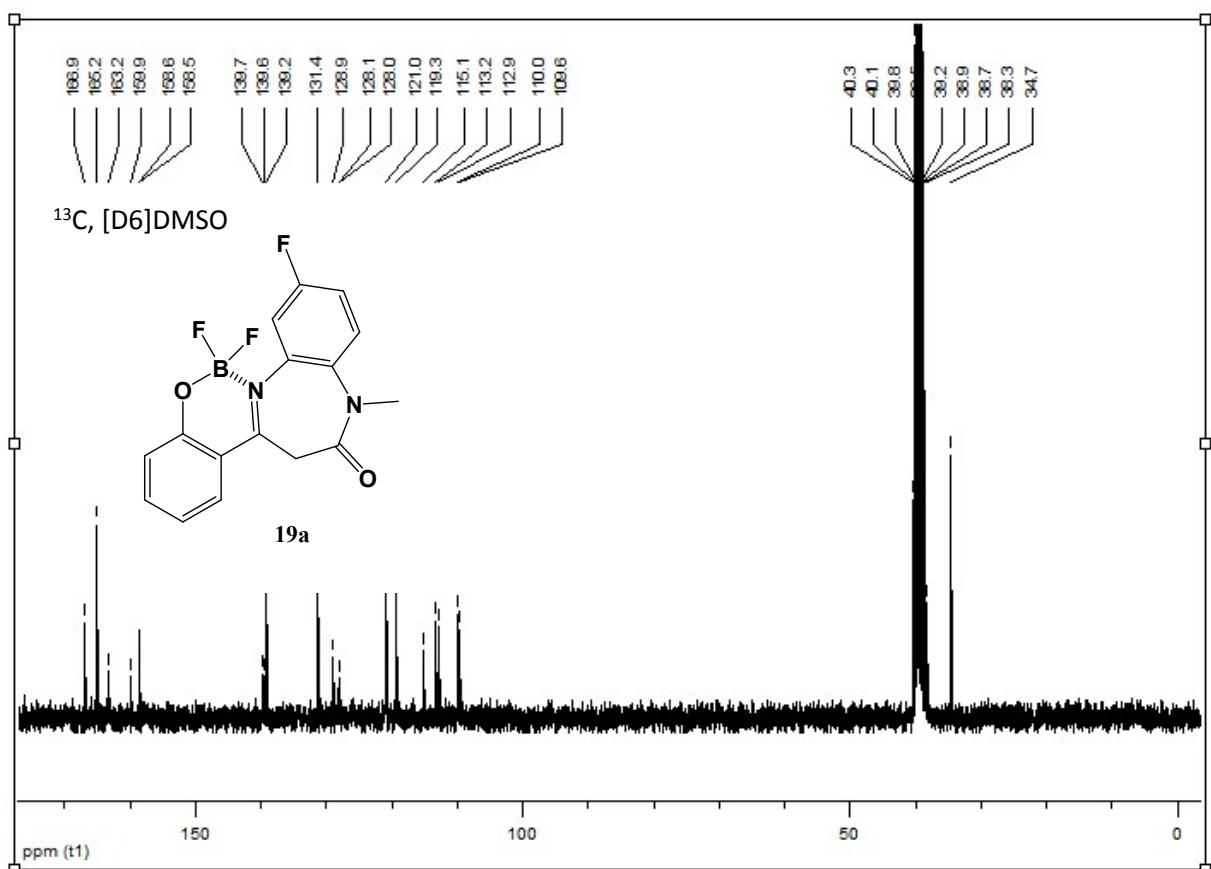
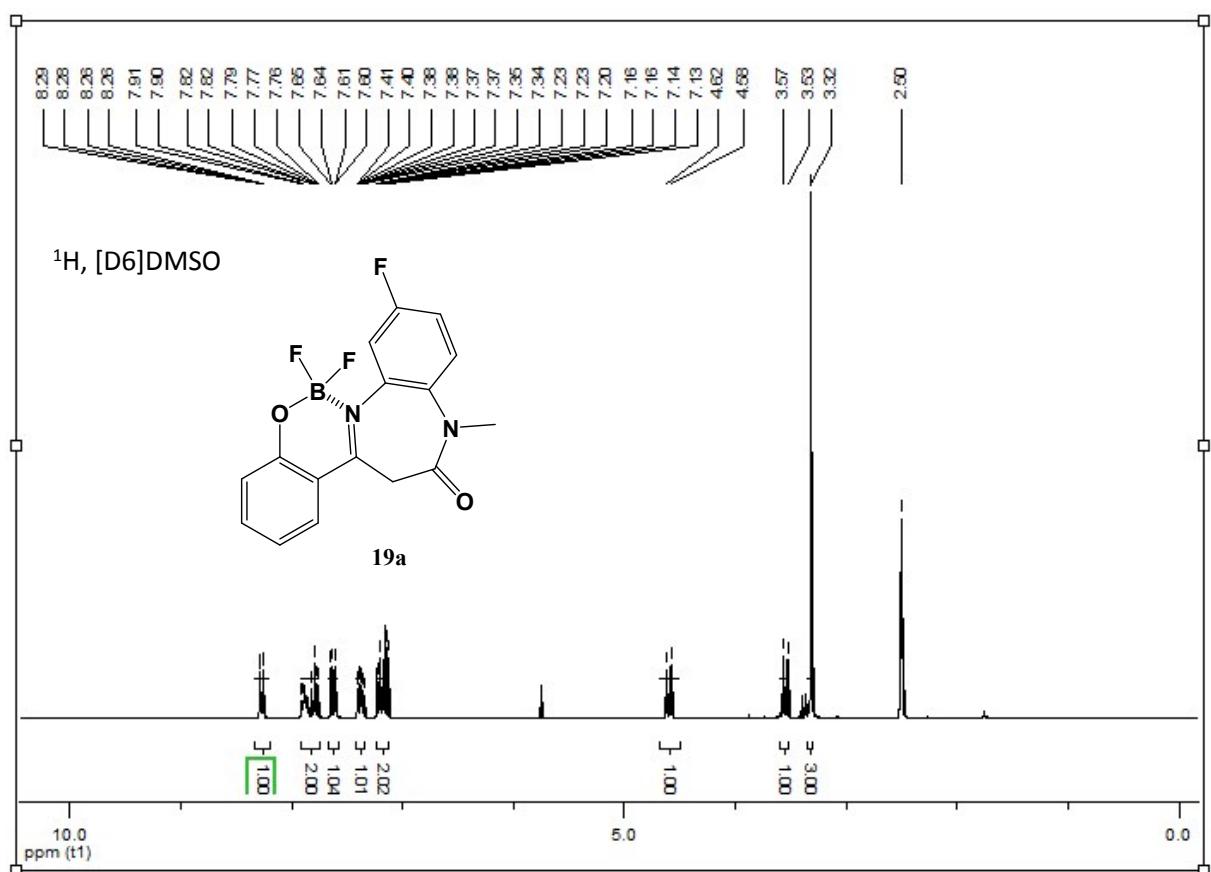


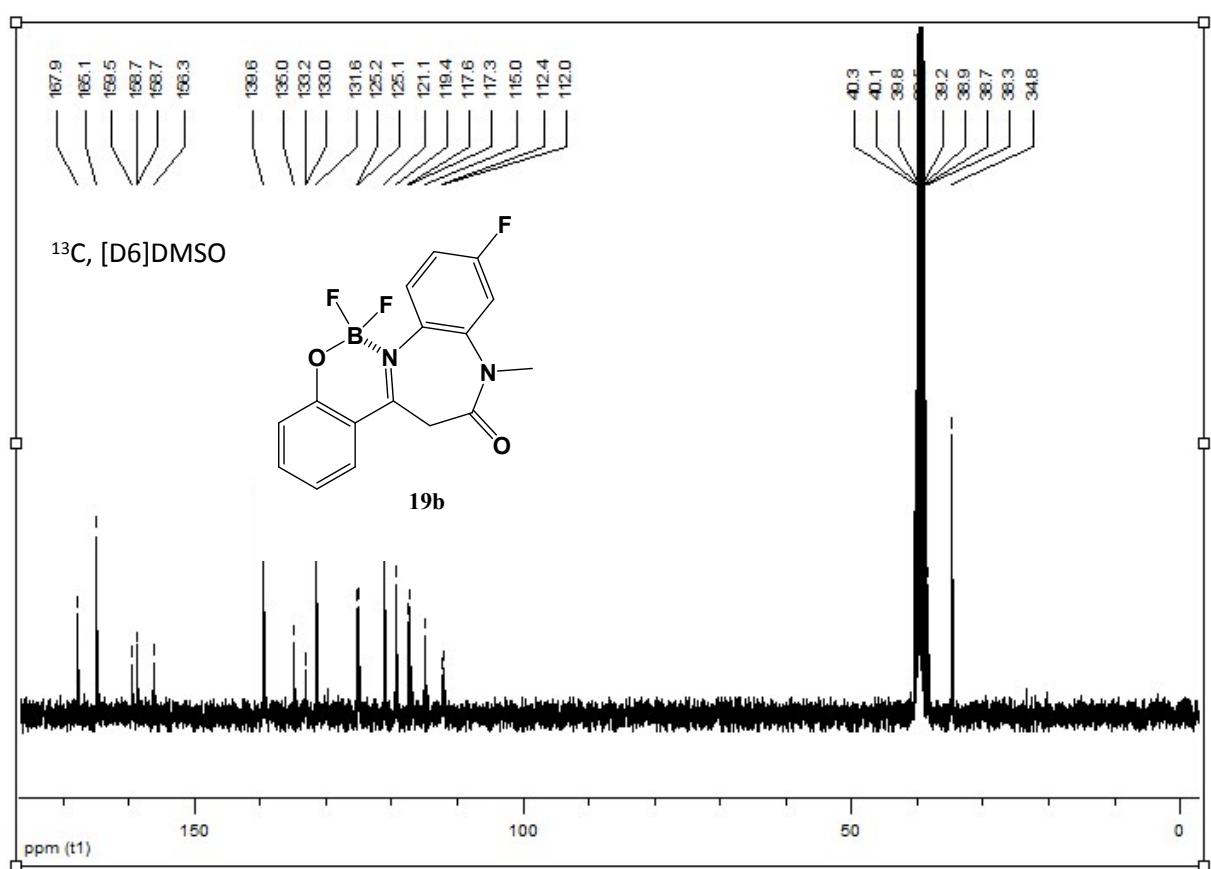
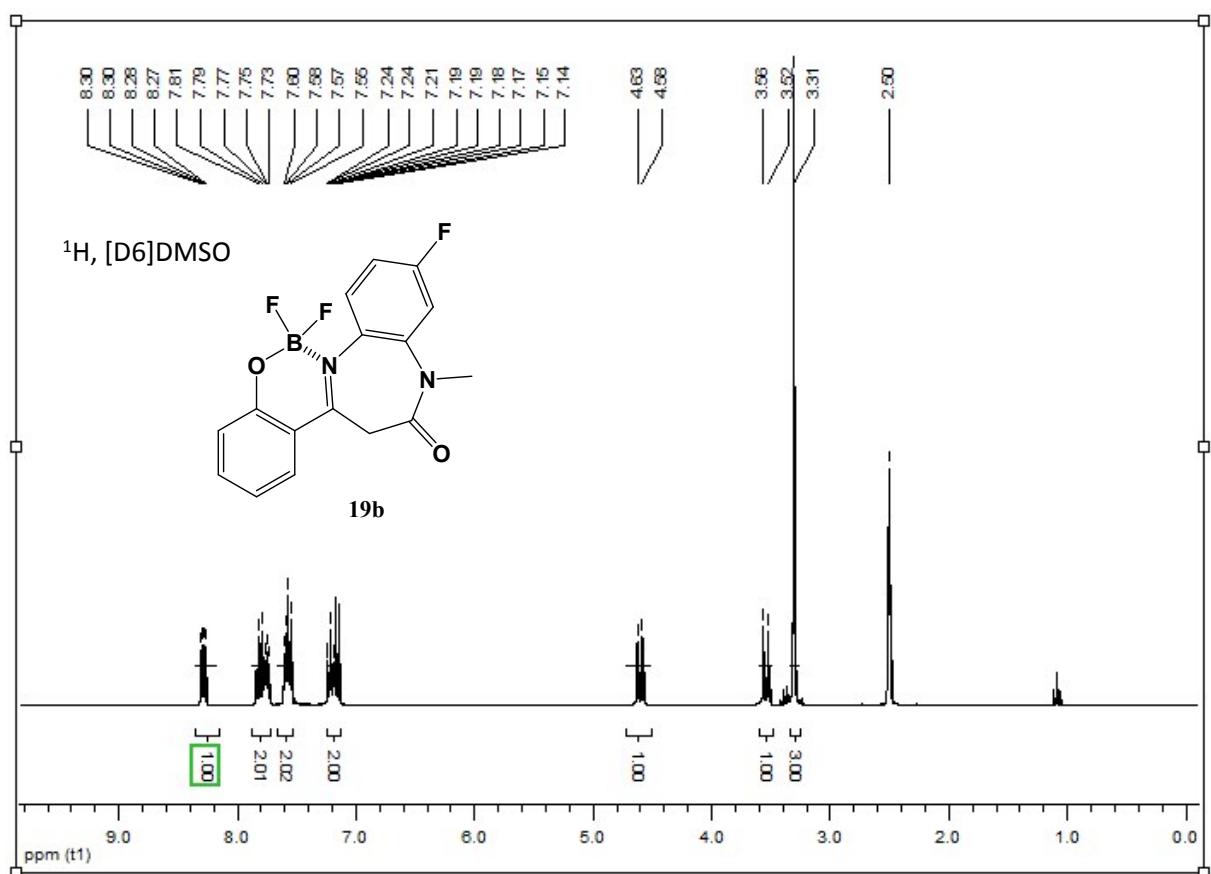


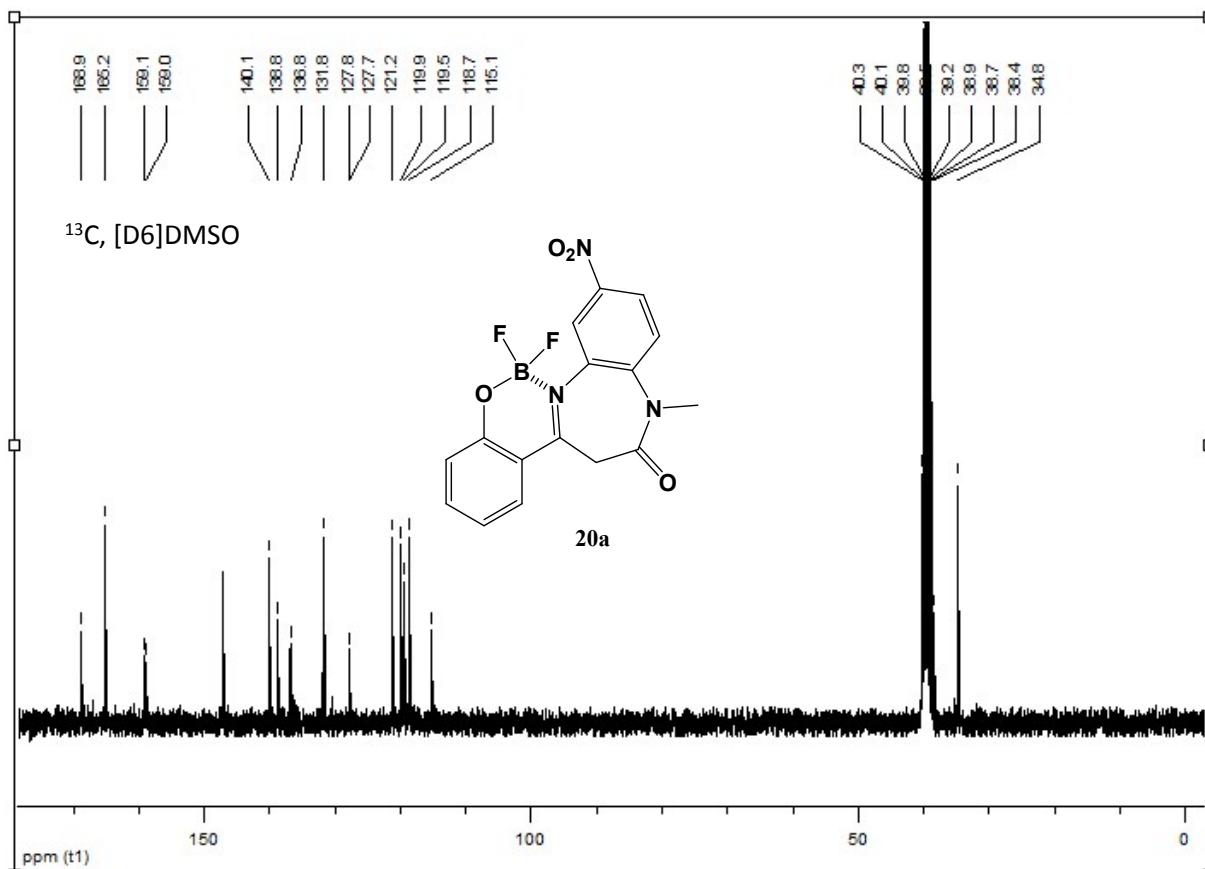
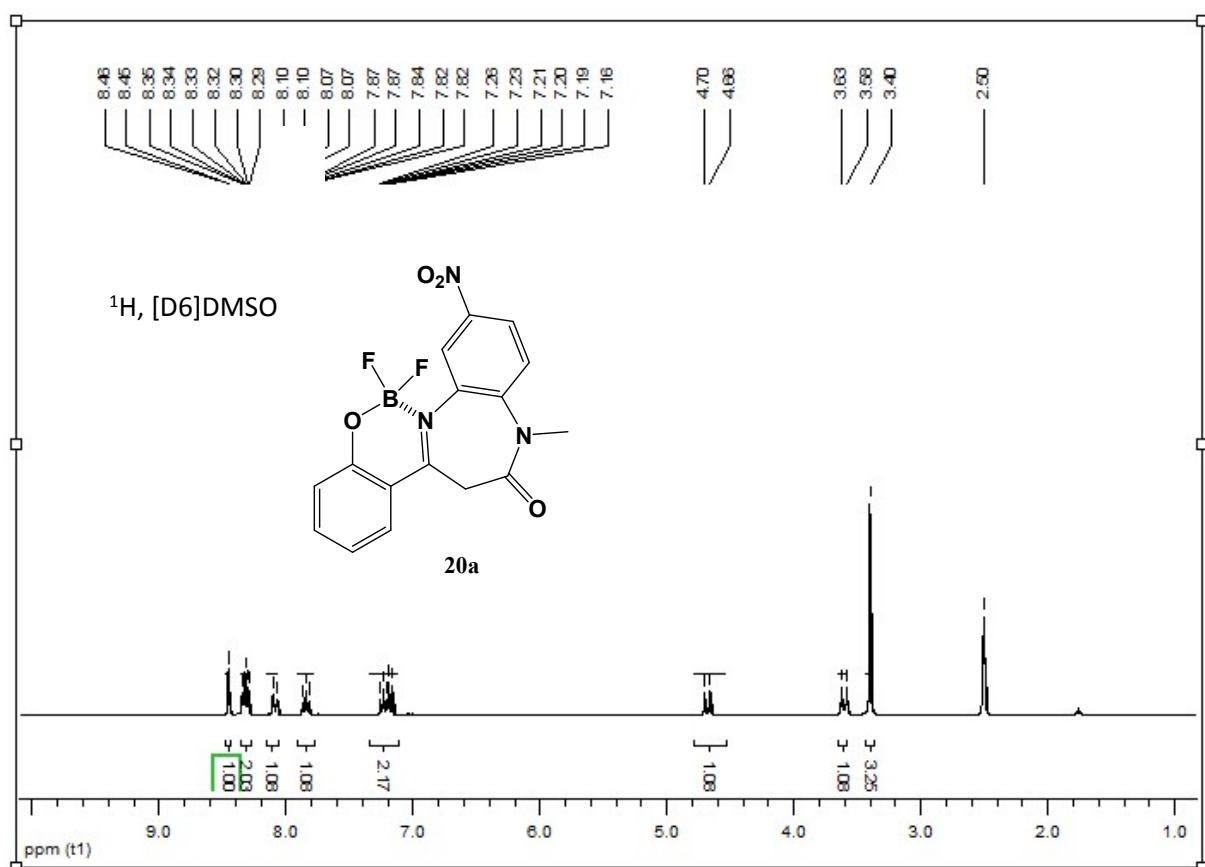


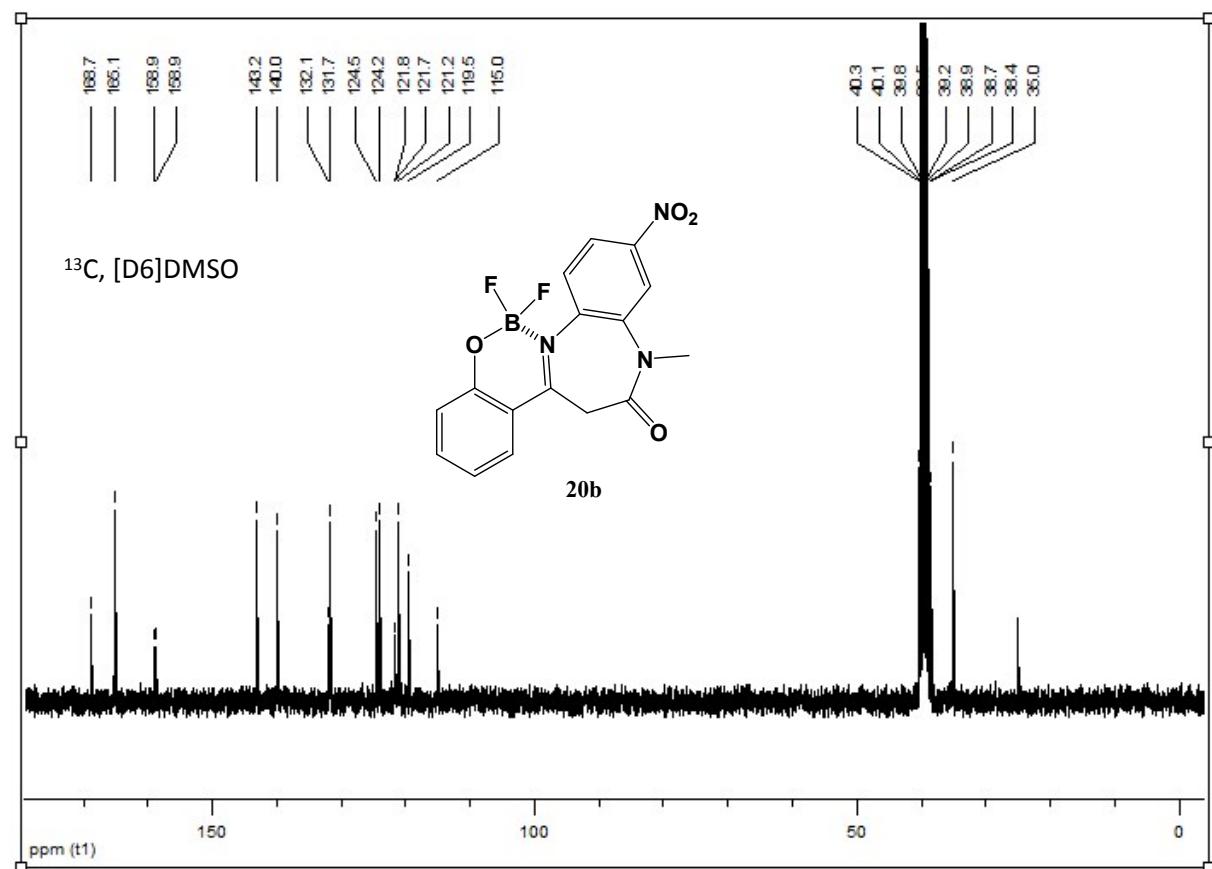
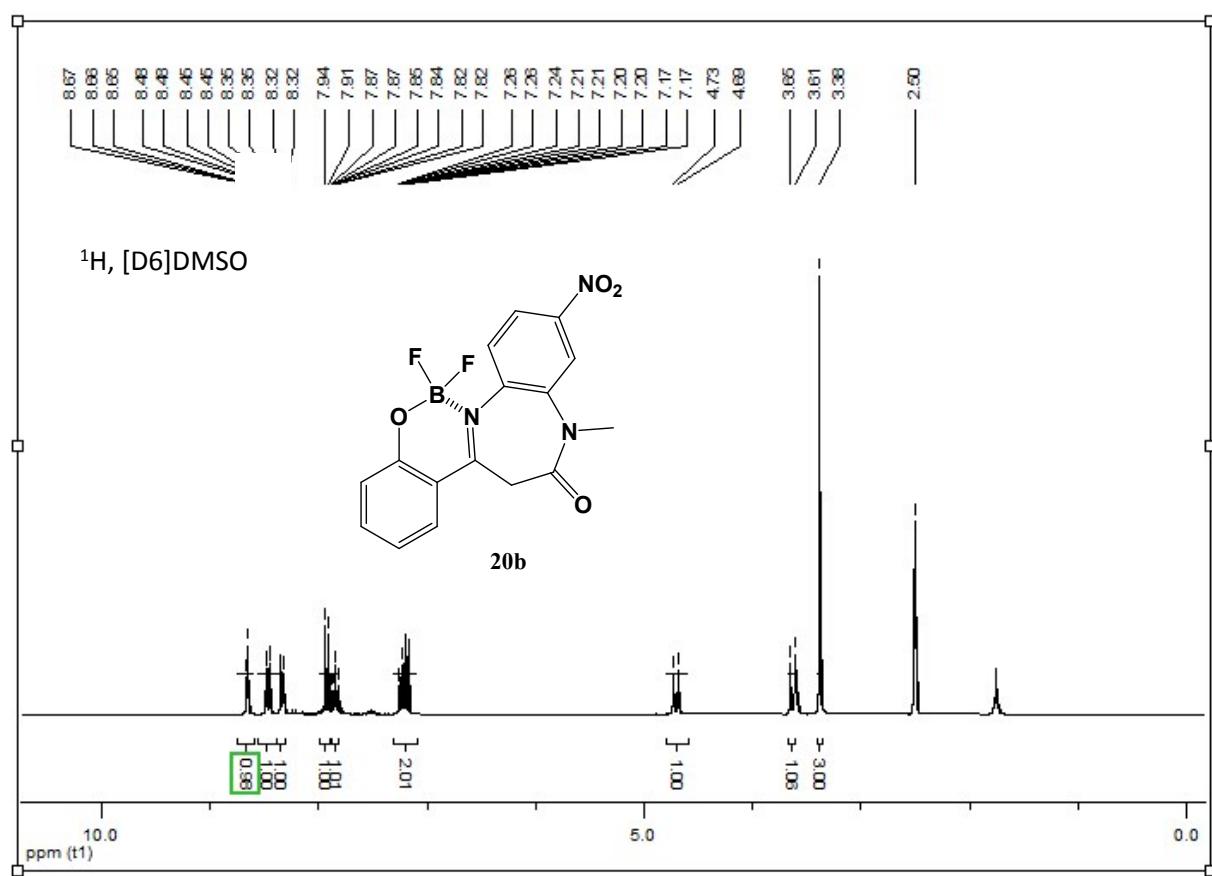


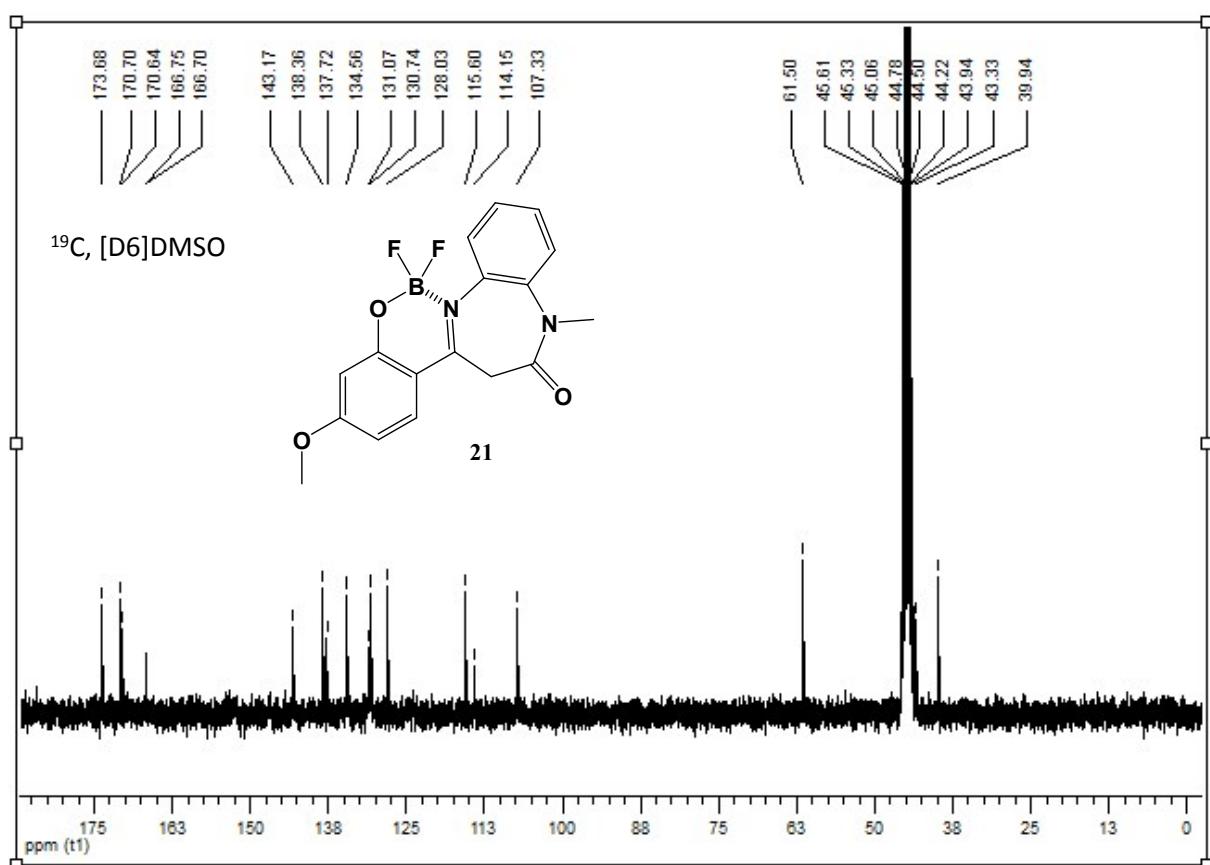
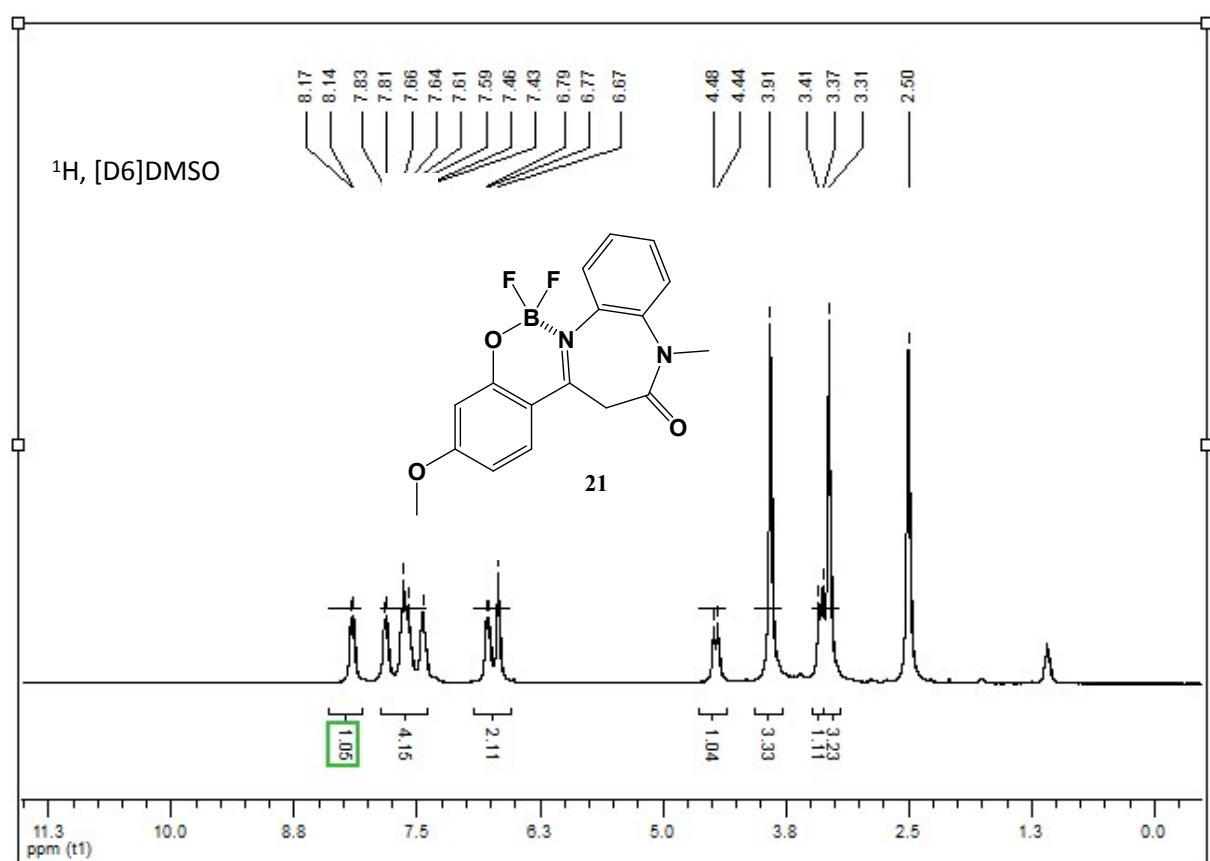


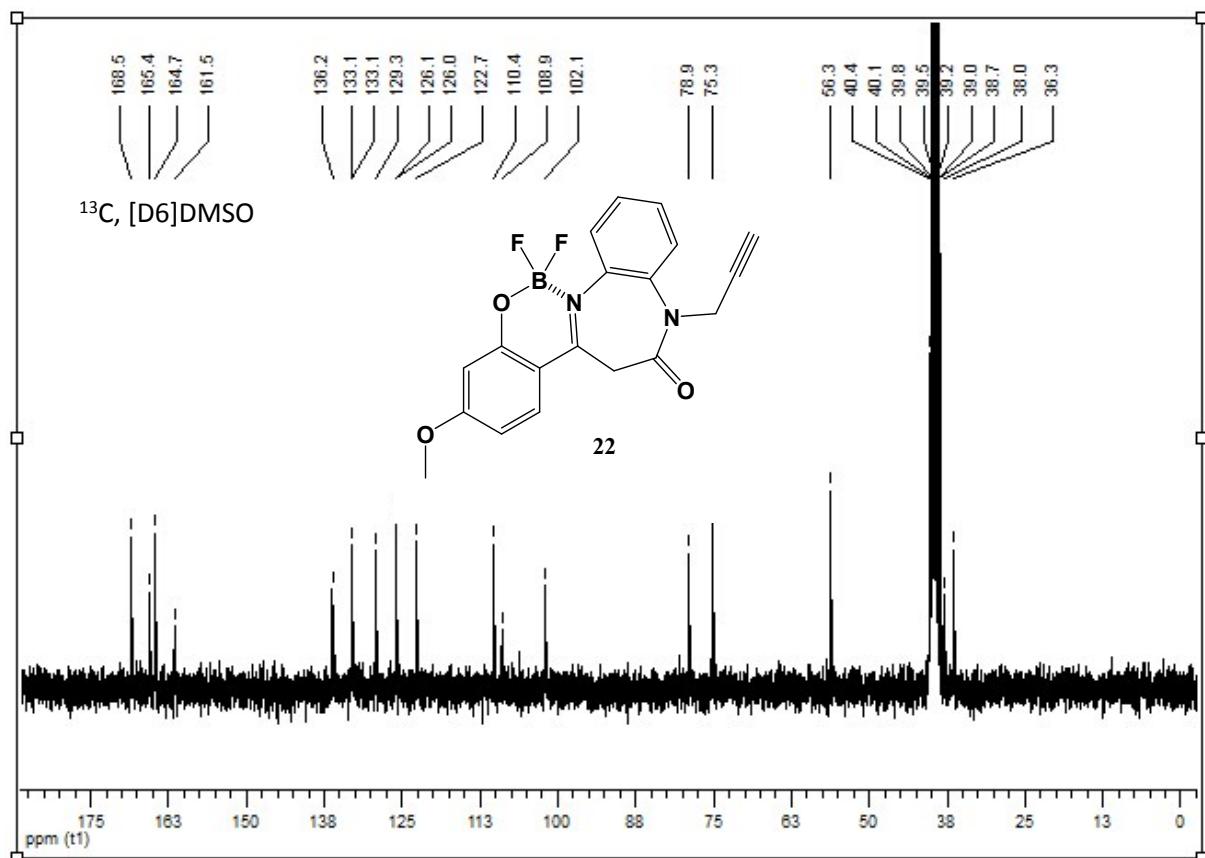
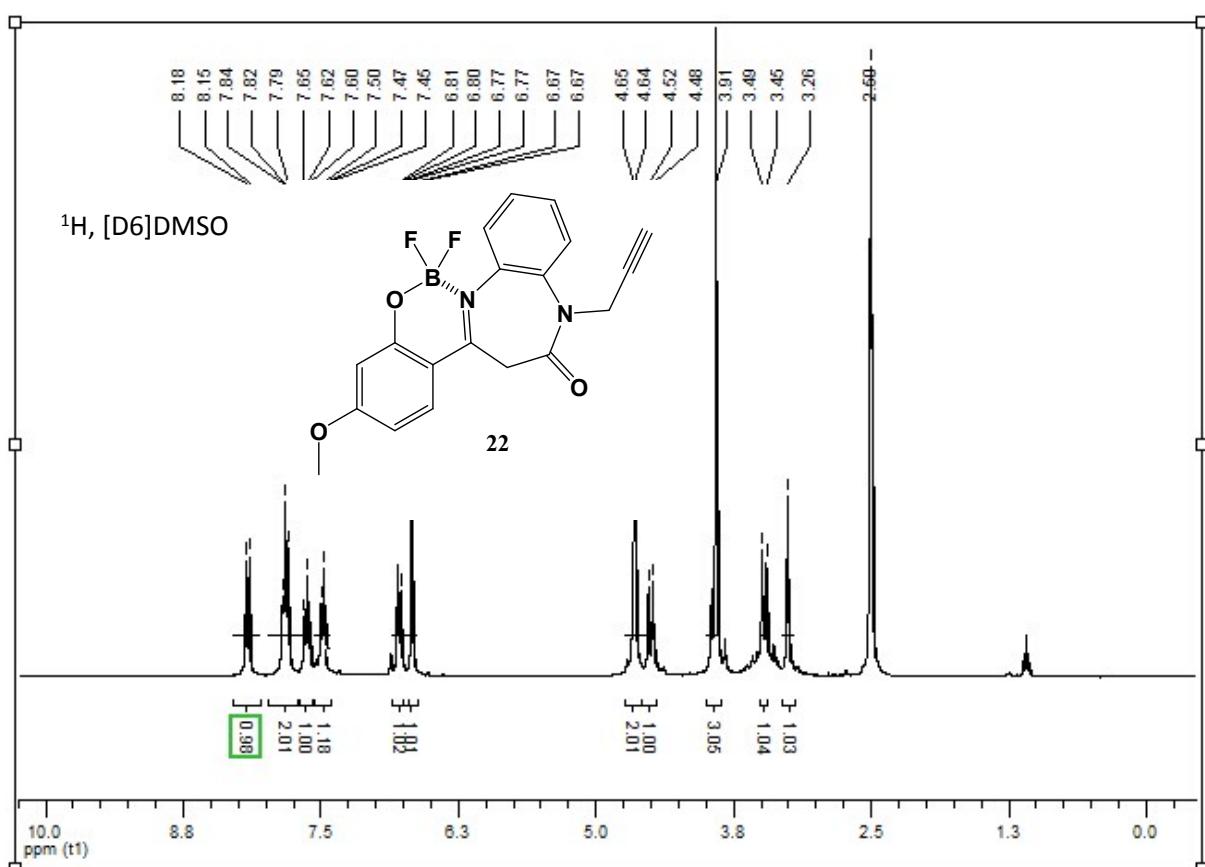


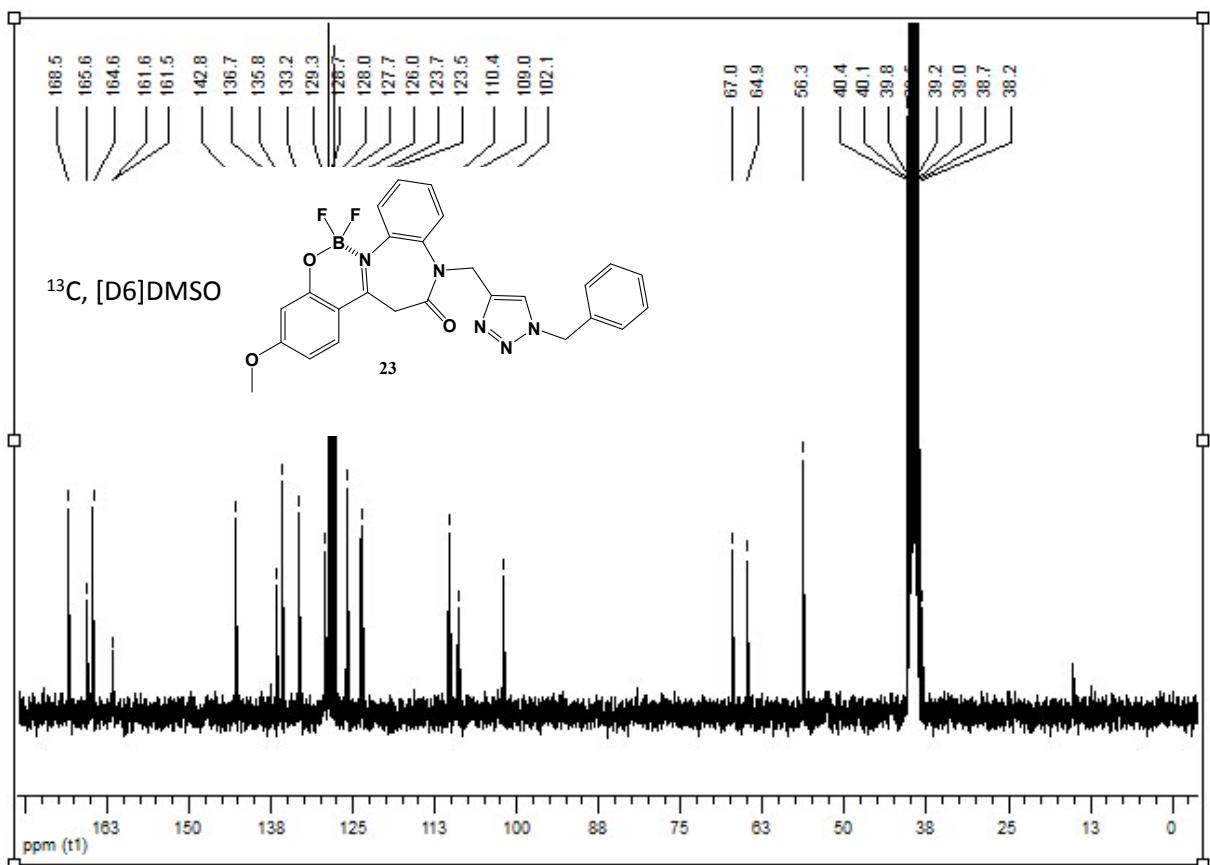
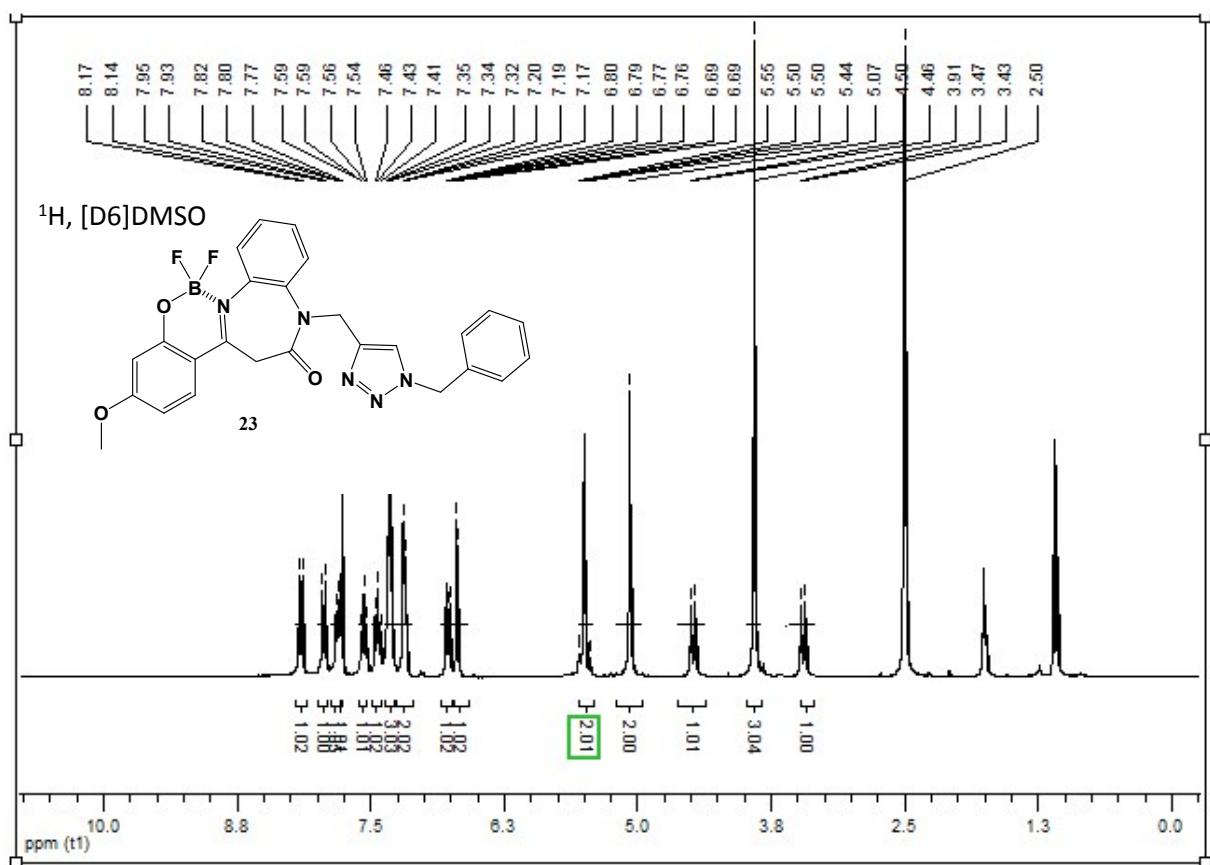












## II. Crystallographic data

### DATA COLLECTION

The crystal structure of compound **13** [ $C_{16}H_{13}BF_2N_2O_2$ ] has been determined from single crystal X-Ray diffraction. The chosen crystal was stuck on a glass fibre and mounted on the full three-circle goniometer of a Bruker SMART APEX diffractometer with a CCD area detector. Three sets of exposures (a total of 1800 frames) were recorded, corresponding to three  $\omega$  scans (steps of  $0.3^\circ$ ), for three different values of  $\phi$ . The details of data collection are given in annexe 1.

The cell parameters and the orientation matrix of the crystal were preliminary determined by using SMART Software<sup>1</sup>. Data integration and global cell refinement were performed with SAINT Software<sup>2</sup>. Intensities were corrected for Lorentz, polarisation, decay and absorption effects (SAINT and SADABS Softwares) and reduced to  $F_O^2$ . The program package WinGX<sup>3</sup> was used for space group determination, structure solution and refinement.

### DATA REFINEMENT

The standard space group  $P2_1/c$  (n°14) was determined from systematic extinctions and relative  $F_O^2$  of equivalent reflections. The structure was solved by direct methods<sup>4</sup>. Anisotropic displacement parameters were refined for all non-hydrogen atoms. Every Hydrogen atoms were located from subsequent difference Fourier syntheses and placed with geometrical constraints (SHELXL<sup>5</sup>). The final cycle of full-matrix least-square refinement on  $F^2$  was based on 2887 observed reflections and 228 variable parameters and converged with unweighted and weighted agreement factors of:

$R_1 = 0.0357$ ,  $wR_2 = 0.0977$  for 2458 reflections with  $I > 2\sigma I$  and  $R_1 = 0.0420$ ,  $wR_2 = 0.1022$  for all data.

The refinement data are given in annexe 1 table 2

## CRYSTALLOGRAPHIC DATA AND STRUCTURAL DESCRIPTION

### *Crystallographic data*

The crystal data are collected in Table 1. The full crystallographic parameters (atomic coordinates, bond length, angles and anisotropic displacements) are reported in annexe 2.

Chemical Formula	C <sub>16</sub> H <sub>13</sub> BF <sub>2</sub> N <sub>2</sub> O <sub>2</sub>
Molecular Weight / g.mol <sup>-1</sup>	314.09
Crystal System	Monoclinic
Space Group	P21/c
Z , Z' (asymmetric units per unit cell)	4,1
a / Å	10.173(1)
b / Å	7.374(1)
c / Å	19.011(2)
α / °	90
β / °	94.747(2)
γ / °	90
V / Å <sup>3</sup>	1421.4(2)
d <sub>calc</sub> / g.cm <sup>-3</sup>	1.468
F(000) / e <sup>-</sup>	648
Absorption coefficient μ (MoK $\alpha_1$ ) / mm <sup>-1</sup>	0.114

### **Structural description**

The asymmetric unit is composed of one molecule of  $C_{16}H_{13}N_2O_2B$  (figure1). The Fluorine atoms are presenting a disordered distribution on two crystallographic sites with 73/27 % of statistical occupancy.

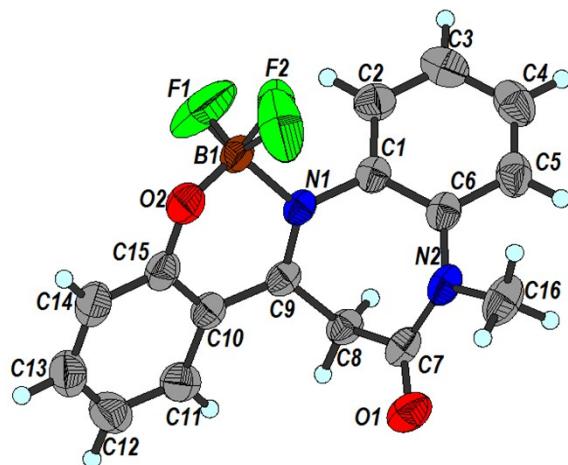


Figure 1: Asymmetric unit in ORTEP representation , with atom labels.

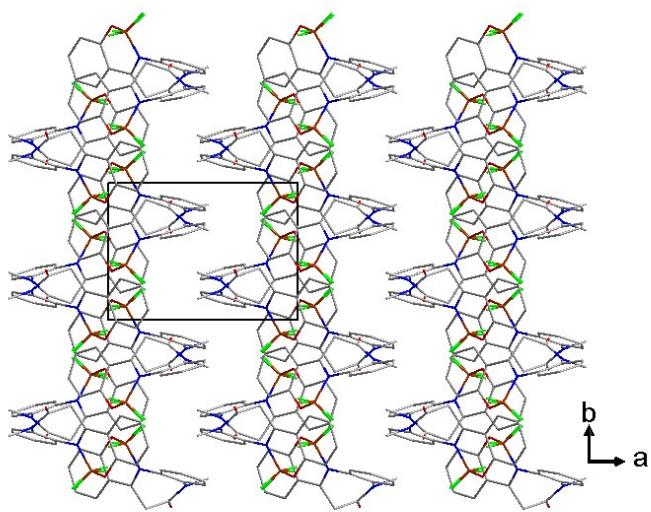


Figure 2: Projection along c.

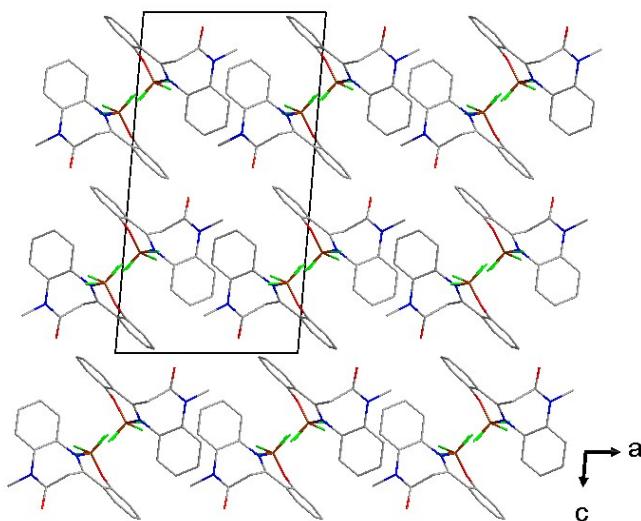


Figure 3: Projection along b

#### Sofwares :

(1)- SMART for WNT/2000 V5.622 (2001), Smart software reference manual, Bruker Advanced X Ray Solutions, Inc., Madison, Wisconsin, USA.

(2)- SAINT+ V6.02 (1999), Saint software reference manual, Bruker Advanced X Ray Solutions, Inc., Madison, Wisconsin, USA.

(3)-WinGX: Version 1.70.01: An integrated system of Windows Programs for the solution, refinement and analysis of Single Crystal X-Ray Diffraction Data, By LouisJ. Farrugia, Dept. of chemistry, University of Glasgow.

L. J. Farrugia (1999) J. Appl. Cryst. 32, 837-838.

(4)-include in WinGX suite : SIR 92: A. Altomare, G. Cascarano, & A. Gualandi (1993) J. Appl. Cryst. 26, 343-350; SHELXS-97: Sheldrick, G. M., (1990) Acta cryst, A46, 467.

(5)-include in WinGX suite: SHELXL-97 – a program for crystal structure refinement, G. M. .Sheldrick, University of Goettingen, Germany, 1997, release 97-2.

(6)-PowderCell for Windows (version 2.4) by Kraus W. & Nolze G., Federal institute for materials Research and testing, Rudower Chausse 5, 12489 Berlin Germany.

**ANNEXE 1 :**

- **Table 1 : DATA COLLECTION FOR COMPOUND 13**

Date	<b>10/06/16</b>
Temperature / K	RT
Radiation	Mo-K $\alpha_1$ ( $\lambda = 0.71073\text{\AA}$ )
Monochromator	Graphite
Collimator / mm	0.5
Generator set	50 kV 40mA
Crystal-detector distance / mm	60
Detector 2 $\theta$ angle / °	-28
$\omega$ oscillations / °	-0.3
$\omega$ scan 1	$\chi = 54.7^\circ, \phi = 0^\circ, -28^\circ \leq \omega \leq -208^\circ$
$\omega$ scan 2	$\chi = 54.7^\circ, \phi = 120^\circ, -28^\circ \leq \omega \leq -208^\circ$
$\omega$ scan 3	$\chi = 54.7^\circ, \phi = 240^\circ, -28^\circ \leq \omega \leq -208^\circ$
Time exposure / s	20
Total number of reflections	10887
Unique reflections [ $F_o > 4.0 \sigma(F_o)$ ]	2887 / 2458
$\theta$ range / °	2.01 to 26.34
hkl range	-12 ≤ h ≤ 12, -9 ≤ k ≤ 9, -23 ≤ l ≤ 23
$R_{int} = \Sigma [  F_o ^2 - F_o^2(\text{mean}) ] / \Sigma [ F_o^2 ]$	0.018
Completeness to $\theta = 26.40$ / %	99.7

- **Table 2 : REFINEMENT DATA FOR COMPOUND 13**

Number of reflections (n) (with $F_o > 4.0 \sigma(F_o)$ )	2458
Number of refined parameters (p) / restraints	228
Reflection / parameter ratio	
Final R indices [ I >2sigma(I)]	R1=0.0357 wR2=0.0977
R indices (all data)	R1=0.0420, wR2=0.1022
Goodness of Fit indicator (Restrained GooF)	1.061
Maximum peak in Final Difference Map / e $^{-\text{\AA}^3}$	0.178
Maximum hole in Final Difference Map / e $^{-\text{\AA}^3}$	-0.173

$$R_1 = \Sigma (|F_o| - |F_c|) / \Sigma |F_o|$$

$$wR_2 = [\Sigma [w (F_o^2 - F_c^2)^2] / \Sigma [w (F_o^2)^2]]^{1/2}$$

$$\text{GooF} = [\Sigma [w (F_o^2 - F_c^2)^2] / (n - p)]^{1/2}$$

## ANNEXE 2 : CRYSTALLOGRAPHIC DATA

Table 1a: Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ).

$U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor

	x	y	z	$U(\text{eq})$
C(1)	2820(1)	6230(2)	2426(1)	37(1)
C(2)	2780(1)	5949(2)	3147(1)	48(1)
C(3)	3888(2)	6221(2)	3601(1)	57(1)
C(4)	5053(2)	6749(2)	3335(1)	58(1)
C(5)	5106(1)	7040(2)	2626(1)	51(1)
C(6)	3991(1)	6802(2)	2159(1)	40(1)
C(7)	3186(1)	8205(2)	1050(1)	41(1)
C(8)	1992(1)	8767(2)	1431(1)	40(1)
C(9)	1207(1)	7079(2)	1513(1)	35(1)
C(10)	36(1)	6674(2)	1055(1)	39(1)
C(11)	-640(1)	8003(2)	636(1)	49(1)
C(12)	-1705(2)	7547(2)	184(1)	60(1)
C(13)	-2099(1)	5742(3)	125(1)	60(1)
C(14)	-1470(1)	4420(2)	527(1)	55(1)
C(15)	-416(1)	4878(2)	1010(1)	43(1)
C(16)	5256(1)	6561(2)	1098(1)	60(1)
B(1)	900(1)	3988(2)	2057(1)	42(1)
F(1)	1807(5)	2676(7)	2256(2)	60(1)
F(2)	56(2)	4253(5)	2594(2)	53(1)
F(1B)	230(30)	3960(40)	2549(9)	118(6)
F(2B)	1900(20)	2760(30)	2030(30)	109(8)
N(1)	1653(1)	5876(1)	1974(1)	35(1)
N(2)	4067(1)	7127(2)	1430(1)	43(1)
O(1)	3312(1)	8634(2)	442(1)	57(1)
O(2)	180(1)	3559(1)	1400(1)	51(1)

Table 1b: Hydrogen coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ).  
 U(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor

	x	y	z	U(eq)
H(2)	1999	5575	3324	57
H(3)	3854	6051	4084	68
H(4)	5807	6908	3640	69
H(5)	5895	7401	2454	61
H(8A)	2268	9282	1889	47
H(8B)	1472	9662	1157	47
H(11)	-360	9203	666	59
H(12)	-2163	8437	-82	72
H(13)	-2805	5431	-195	72
H(14)	-1744	3220	479	65
H(16A)	5027	6288	609	89
H(16B)	5624	5503	1333	89
H(16C)	5893	7524	1134	89

Table 2: Bond lengths ( $\text{\AA}$ )

C(1)-C(2)	1.3901(18)	C(10)-C(11)	1.4060(19)
C(1)-C(6)	1.3979(17)	C(11)-C(12)	1.369(2)
C(1)-N(1)	1.4308(15)	C(11)-H(11)	0.93
C(2)-C(3)	1.376(2)	C(12)-C(13)	1.392(2)
C(2)-H(2)	0.93	C(12)-H(12)	0.93
C(3)-C(4)	1.382(2)	C(13)-C(14)	1.366(2)
C(3)-H(3)	0.93	C(13)-H(13)	0.93
C(4)-C(5)	1.371(2)	C(14)-C(15)	1.3938(19)
C(4)-H(4)	0.93	C(14)-H(14)	0.93
C(5)-C(6)	1.3919(19)	C(15)-O(2)	1.3380(17)
C(5)-H(5)	0.93	C(16)-N(2)	1.4703(16)
C(6)-N(2)	1.4147(18)	C(16)-H(16A)	0.96
C(7)-O(1)	1.2165(16)	C(16)-H(16B)	0.96
C(7)-N(2)	1.3605(17)	C(16)-H(16C)	0.96
C(7)-C(8)	1.5221(17)	B(1)-F(1B)	1.20(3)
C(8)-C(9)	1.4940(16)	B(1)-F(1)	1.369(5)
C(8)-H(8A)	0.97	B(1)-F(2B)	1.37(2)
C(8)-H(8B)	0.97	B(1)-F(2)	1.400(4)
C(9)-N(1)	1.3029(16)	B(1)-O(2)	1.4302(19)
C(9)-C(10)	1.4463(17)	B(1)-N(1)	1.6031(16)
C(10)-C(15)	1.4030(19)		

Table 3: Angles ( $^{\circ}$ )

C(2)-C(1)-C(6)	119.89(12)	C(13)-C(12)-H(12)	120.2
C(2)-C(1)-N(1)	118.32(11)	C(14)-C(13)-C(12)	121.27(14)
C(6)-C(1)-N(1)	121.78(11)	C(14)-C(13)-H(13)	119.4
C(3)-C(2)-C(1)	120.46(13)	C(12)-C(13)-H(13)	119.4
C(3)-C(2)-H(2)	119.8	C(13)-C(14)-C(15)	119.69(15)
C(1)-C(2)-H(2)	119.8	C(13)-C(14)-H(14)	120.2
C(2)-C(3)-C(4)	119.61(14)	C(15)-C(14)-H(14)	120.2
C(2)-C(3)-H(3)	120.2	O(2)-C(15)-C(14)	118.55(13)
C(4)-C(3)-H(3)	120.2	O(2)-C(15)-C(10)	121.39(11)
C(5)-C(4)-C(3)	120.59(13)	C(14)-C(15)-C(10)	120.00(13)
C(5)-C(4)-H(4)	119.7	N(2)-C(16)-H(16A)	109.5
C(3)-C(4)-H(4)	119.7	N(2)-C(16)-H(16B)	109.5
C(4)-C(5)-C(6)	120.69(13)	H(16A)-C(16)-H(16B)	109.5
C(4)-C(5)-H(5)	119.7	N(2)-C(16)-H(16C)	109.5
C(6)-C(5)-H(5)	119.7	H(16A)-C(16)-H(16C)	109.5
C(5)-C(6)-C(1)	118.74(13)	H(16B)-C(16)-H(16C)	109.5
C(5)-C(6)-N(2)	119.62(11)	F(1B)-B(1)-F(1)	100.5(16)
C(1)-C(6)-N(2)	121.63(11)	F(1B)-B(1)-F(2B)	119.5(13)
O(1)-C(7)-N(2)	122.70(12)	F(1)-B(1)-F(2B)	19(2)
O(1)-C(7)-C(8)	122.18(12)	F(1B)-B(1)-F(2)	9.9(17)
N(2)-C(7)-C(8)	115.06(11)	F(1)-B(1)-F(2)	109.7(2)
C(9)-C(8)-C(7)	106.05(10)	F(2B)-B(1)-F(2)	129(2)
C(9)-C(8)-H(8A)	110.5	F(1B)-B(1)-O(2)	113.2(9)
C(7)-C(8)-H(8A)	110.5	F(1)-B(1)-O(2)	111.7(2)
C(9)-C(8)-H(8B)	110.5	F(2B)-B(1)-O(2)	98.3(18)
C(7)-C(8)-H(8B)	110.5	F(2)-B(1)-O(2)	111.52(16)
H(8A)-C(8)-H(8B)	108.7	F(1B)-B(1)-N(1)	113.5(12)
N(1)-C(9)-C(10)	119.22(11)	F(1)-B(1)-N(1)	108.9(2)
N(1)-C(9)-C(8)	118.55(11)	F(2B)-B(1)-N(1)	101.9(12)
C(10)-C(9)-C(8)	122.01(11)	F(2)-B(1)-N(1)	106.11(18)
C(15)-C(10)-C(11)	118.73(12)	O(2)-B(1)-N(1)	108.71(11)
C(15)-C(10)-C(9)	118.78(11)	C(9)-N(1)-C(1)	120.40(10)
C(11)-C(10)-C(9)	122.45(12)	C(9)-N(1)-B(1)	120.98(10)
C(12)-C(11)-C(10)	120.63(15)	C(1)-N(1)-B(1)	118.61(10)
C(12)-C(11)-H(11)	119.7	C(7)-N(2)-C(6)	122.14(10)
C(10)-C(11)-H(11)	119.7	C(7)-N(2)-C(16)	117.83(12)
C(11)-C(12)-C(13)	119.57(15)	C(6)-N(2)-C(16)	119.14(12)
C(11)-C(12)-H(12)	120.2	C(15)-O(2)-B(1)	119.95(11)

Table 4: Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) The anisotropic displacement factor exponent takes the form:  $-2 \pi^2 [ h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12} ]$

	U11	U22	U33	U23	U13	U12
C(1)	37(1)	30(1)	45(1)	-2(1)	6(1)	1(1)
C(2)	53(1)	46(1)	46(1)	2(1)	9(1)	3(1)
C(3)	72(1)	49(1)	48(1)	-2(1)	-5(1)	10(1)
C(4)	57(1)	43(1)	70(1)	-6(1)	-17(1)	4(1)
C(5)	40(1)	39(1)	74(1)	-1(1)	-1(1)	-2(1)
C(6)	37(1)	29(1)	53(1)	-2(1)	7(1)	-1(1)
C(7)	43(1)	36(1)	46(1)	-3(1)	13(1)	-11(1)
C(8)	41(1)	32(1)	46(1)	2(1)	9(1)	0(1)
C(9)	35(1)	33(1)	39(1)	-1(1)	14(1)	2(1)
C(10)	35(1)	43(1)	39(1)	-2(1)	12(1)	1(1)
C(11)	47(1)	53(1)	49(1)	3(1)	10(1)	7(1)
C(12)	48(1)	81(1)	50(1)	7(1)	5(1)	12(1)
C(13)	43(1)	90(1)	47(1)	-6(1)	4(1)	-6(1)
C(14)	47(1)	68(1)	50(1)	-7(1)	9(1)	-14(1)
C(15)	38(1)	50(1)	44(1)	-2(1)	13(1)	-4(1)
C(16)	48(1)	54(1)	81(1)	-7(1)	31(1)	-2(1)
B(1)	35(1)	34(1)	57(1)	7(1)	8(1)	-3(1)
F(1)	44(1)	34(1)	100(2)	10(1)	-6(1)	2(1)
F(2)	47(2)	59(1)	56(1)	11(1)	23(1)	-6(1)
F(1B)	177(15)	117(10)	66(6)	-17(6)	53(8)	-89(9)
F(2B)	71(7)	33(5)	220(20)	-15(9)	-37(10)	6(4)
N(1)	32(1)	32(1)	41(1)	-1(1)	11(1)	-1(1)
N(2)	36(1)	40(1)	55(1)	-2(1)	17(1)	-3(1)
O(1)	64(1)	62(1)	47(1)	3(1)	19(1)	-10(1)
O(2)	53(1)	40(1)	60(1)	1(1)	4(1)	-11(1)

Table 5: Torsion angles ( $^{\circ}$ )

C(6)-C(1)-C(2)-C(3)	-0.4(2)	C(10)-C(9)-N(1)-B(1)	-1.82(16)
N(1)-C(1)-C(2)-C(3)	178.16(12)	C(8)-C(9)-N(1)-B(1)	-176.56(10)
C(1)-C(2)-C(3)-C(4)	-1.0(2)	C(2)-C(1)-N(1)-C(9)	130.07(12)
C(2)-C(3)-C(4)-C(5)	1.4(2)	C(6)-C(1)-N(1)-C(9)	-51.43(16)
C(3)-C(4)-C(5)-C(6)	-0.3(2)	C(2)-C(1)-N(1)-B(1)	-49.07(15)
C(4)-C(5)-C(6)-C(1)	-1.1(2)	C(6)-C(1)-N(1)-B(1)	129.43(12)
C(4)-C(5)-C(6)-N(2)	179.41(12)	F(1B)-B(1)-N(1)-C(9)	-100.6(15)
C(2)-C(1)-C(6)-C(5)	1.40(18)	F(1)-B(1)-N(1)-C(9)	148.3(2)
N(1)-C(1)-C(6)-C(5)	-177.08(11)	F(2B)-B(1)-N(1)-C(9)	130(2)
C(2)-C(1)-C(6)-N(2)	-179.08(11)	F(2)-B(1)-N(1)-C(9)	-93.69(17)
N(1)-C(1)-C(6)-N(2)	2.44(18)	O(2)-B(1)-N(1)-C(9)	26.37(15)
			78.5(15)
O(1)-C(7)-C(8)-C(9)	109.22(13)	F(1B)-B(1)-N(1)-C(1)	
N(2)-C(7)-C(8)-C(9)	-68.13(14)	F(1)-B(1)-N(1)-C(1)	-32.6(2)
C(7)-C(8)-C(9)-N(1)	73.19(13)	F(2B)-B(1)-N(1)-C(1)	-51(2)
C(7)-C(8)-C(9)-C(10)	-101.40(12)	F(2)-B(1)-N(1)-C(1)	85.44(16)
N(1)-C(9)-C(10)-C(15)	-14.00(16)	O(2)-B(1)-N(1)-C(1)	-154.50(10)
C(8)-C(9)-C(10)-C(15)	160.56(11)	O(1)-C(7)-N(2)-C(6)	172.79(12)
N(1)-C(9)-C(10)-C(11)	168.25(11)	C(8)-C(7)-N(2)-C(6)	-9.88(16)
C(8)-C(9)-C(10)-C(11)	-17.20(17)	O(1)-C(7)-N(2)-C(16)	3.74(19)
C(15)-C(10)-C(11)-C(12)	-1.05(19)	C(8)-C(7)-N(2)-C(16)	-178.93(11)
C(9)-C(10)-C(11)-C(12)	176.71(12)	C(5)-C(6)-N(2)-C(7)	-129.51(13)
C(10)-C(11)-C(12)-C(13)	-1.8(2)	C(1)-C(6)-N(2)-C(7)	50.98(17)
C(11)-C(12)-C(13)-C(14)	2.2(2)	C(5)-C(6)-N(2)-C(16)	39.39(17)
C(12)-C(13)-C(14)-C(15)	0.3(2)	C(1)-C(6)-N(2)-C(16)	-140.11(13)
C(13)-C(14)-C(15)-O(2)	179.70(12)	C(14)-C(15)-O(2)-B(1)	-156.93(12)
C(13)-C(14)-C(15)-C(10)	-3.20(19)	C(10)-C(15)-O(2)-B(1)	26.01(17)
C(11)-C(10)-C(15)-O(2)	-179.43(11)	F(1B)-B(1)-O(2)-C(15)	89.2(18)
C(9)-C(10)-C(15)-O(2)	2.73(17)	F(1)-B(1)-O(2)-C(15)	-158.1(2)
C(11)-C(10)-C(15)-C(14)	3.55(18)	F(2B)-B(1)-O(2)-C(15)	-143.6(18)
C(9)-C(10)-C(15)-C(14)	-174.29(11)	F(2)-B(1)-O(2)-C(15)	78.7(2)
C(10)-C(9)-N(1)-C(1)	179.06(10)	N(1)-B(1)-O(2)-C(15)	-37.92(15)
C(8)-C(9)-N(1)-C(1)	4.32(15)		

Table 6: Calculated reflections from PowderCell\*

$h$	$k$	$l$	$2\theta/^\circ$	$d/\text{\AA}$	$I/\text{rel.}$	$ F(hkl) $
1	0	0	8.72	10.14	51.01	59.64
0	0	2	9.33	9.47	100.00	89.42
-1	0	2	12.24	7.23	40.24	74.62
0	1	1	12.87	6.87	16.86	35.95
1	0	2	13.30	6.65	10.34	41.16
1	1	0	14.84	5.96	36.99	61.55
0	1	2	15.21	5.82	50.92	74.05
-1	1	1	15.35	5.77	93.52	101.24
-1	1	2	17.17	5.16	3.18	20.95
2	0	0	17.48	5.07	7.46	46.22
1	1	2	17.94	4.94	5.45	28.67
-2	0	2	19.15	4.63	3.55	34.99
-1	1	3	19.96	4.45	59.56	105.81
-1	0	4	20.01	4.43	6.25	48.59
2	0	2	20.53	4.32	4.79	43.70
1	1	3	20.96	4.23	8.67	42.48
2	1	0	21.25	4.18	41.19	93.92
1	0	4	21.34	4.16	17.15	86.05
-2	1	1	21.45	4.14	5.88	35.82
2	1	1	22.08	4.02	3.94	30.23
0	1	4	22.29	3.99	20.84	70.20
-1	1	4	23.39	3.80	15.63	63.94
0	2	0	24.12	3.69	12.59	83.77
1	1	4	24.54	3.62	11.95	58.79
0	2	1	24.58	3.62	3.99	34.03
-2	0	4	24.62	3.61	60.15	187.14
-2	1	3	24.72	3.60	14.81	65.93
1	2	0	25.69	3.47	9.58	55.23
-1	2	1	25.99	3.43	10.19	57.67
1	2	1	26.25	3.39	60.12	141.56
2	1	3	26.35	3.38	2.07	26.35
-1	2	2	27.13	3.28	17.05	78.05
-2	1	4	27.47	3.24	6.77	49.81
0	2	3	28.00	3.18	3.26	35.28
3	1	1	29.79	3.00	3.56	39.41

Source: Cu-K $\alpha_1$  ( $\lambda = 1.540598 \text{ \AA}$ )

Condition on reflections:  $I \geq 2$

Range ( $2\theta$ ): From  $3^\circ$  to  $30^\circ$

\*PowderCell for Windows (version 2.4) by Kraus W. & Nolze G., Federal institute for materials Research and testing, Rudower Chausse 5, 12489 Berlin Germany.