

Supporting Information

Cu ion-exchanged and Cu nanoparticles decorated mesoporous ZSM-5 catalysts for the activation and utilization of phenylacetylene in the sustainable chemical synthesis

Bhaskar Sarmah^a, Biswarup Satpati^b, and Rajendra Srivastava^{*a}

^aDepartment of Chemistry, Indian Institute of Technology Ropar, Rupnagar-140001, India

^bSurface Physics and Material Science Division, Saha Institute of Nuclear Physics, 1/AF, Bidhannagar, Kolkata 700 064, India

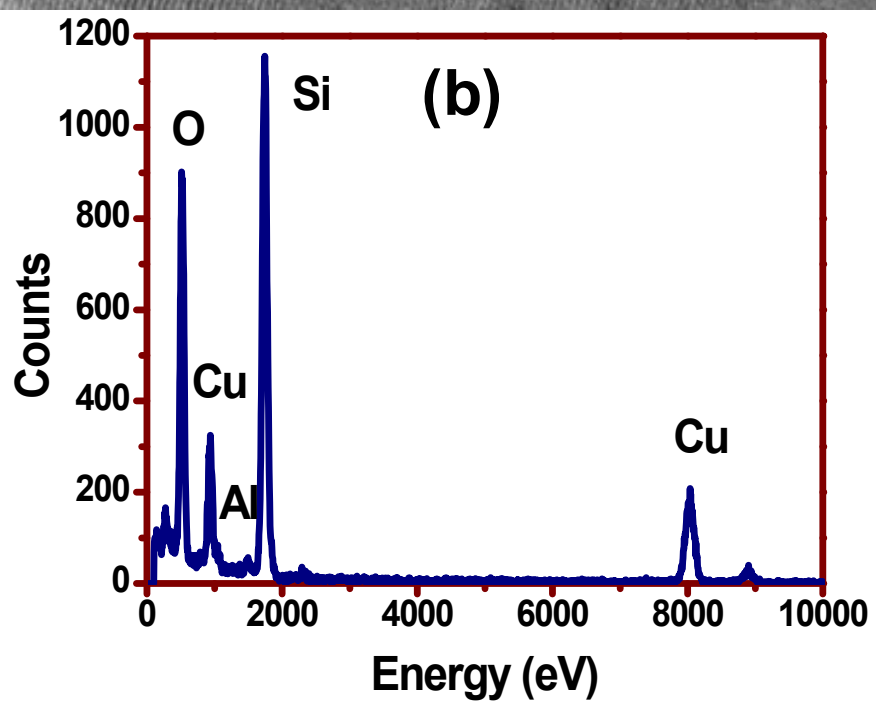
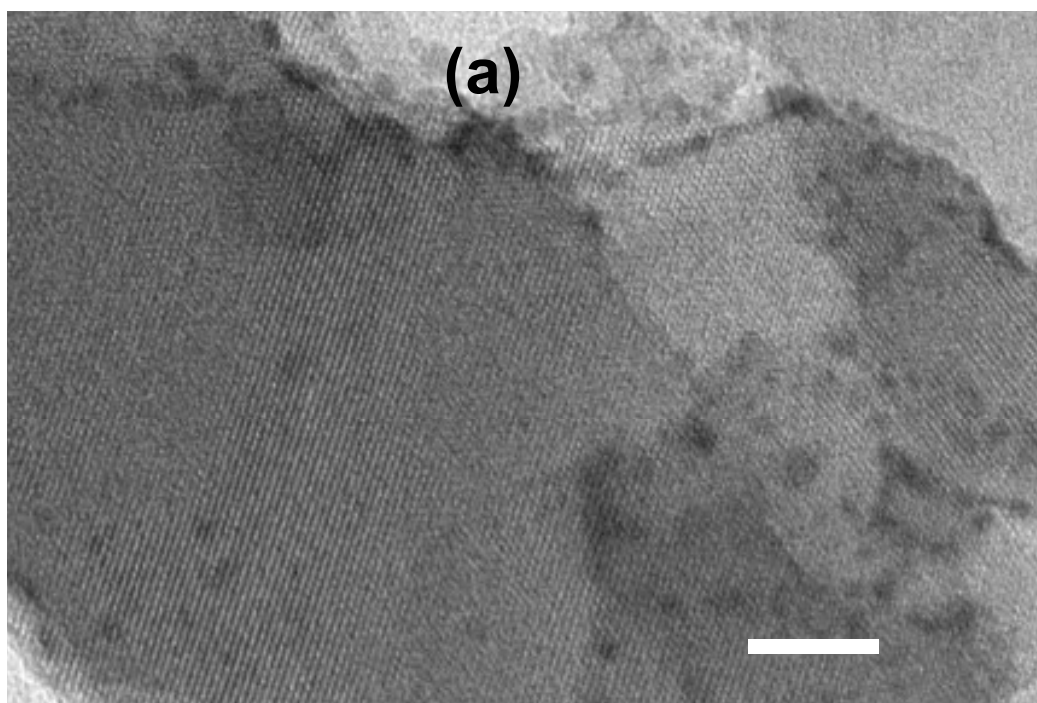


Figure S1. (a) HRTEM image and (b) EDX spectrum of Cu(10%)/Meso-ZSM-5

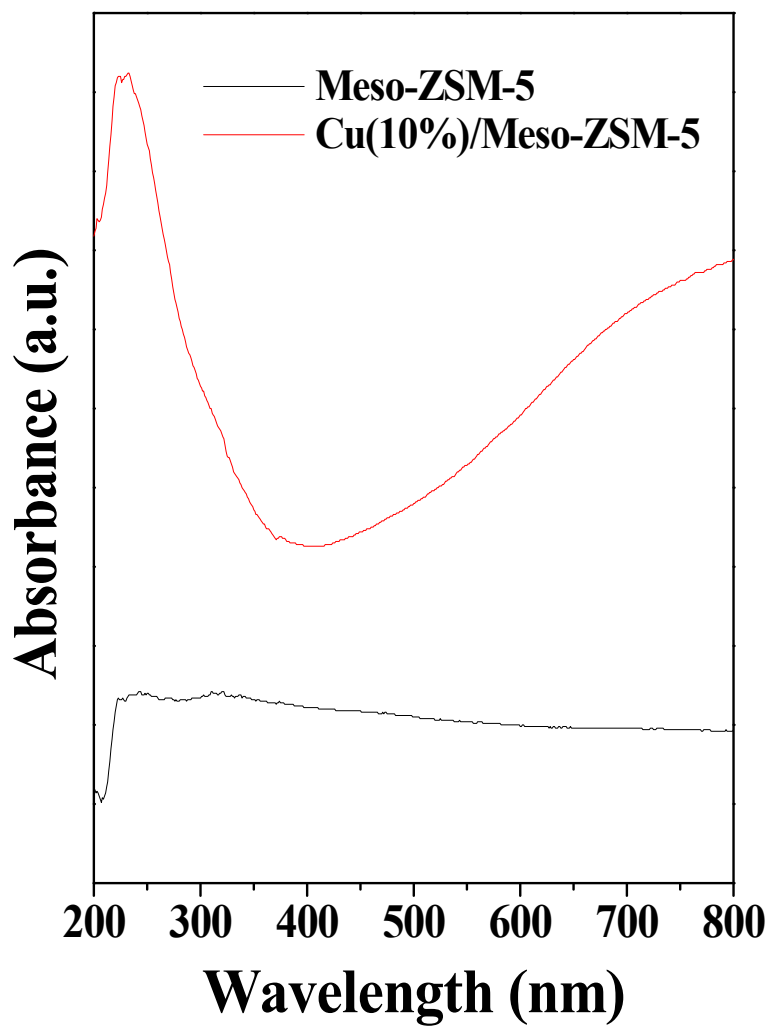


Figure S2. DR UV-Visible spectra of Meso-ZSM-5 and Cu(10%)/Meso-ZSM-5.

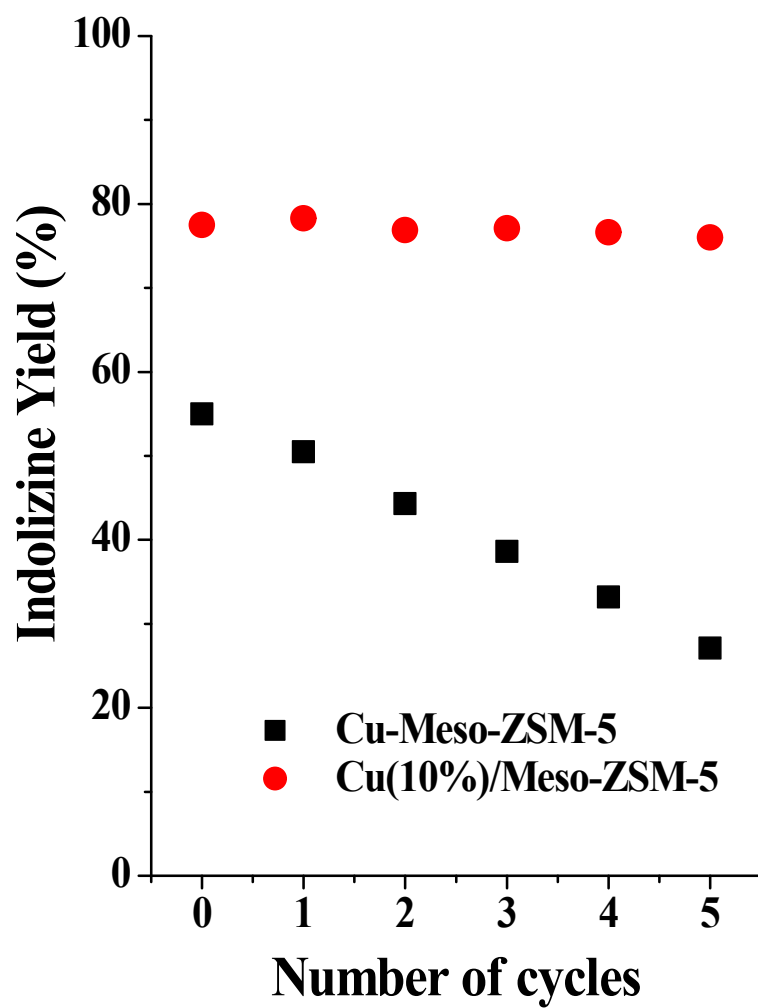


Figure S3. Catalytic performance of Cu-Meso-ZSM-5 and Cu(10%)/Meso-ZSM-5 during the recycling experiments up to five cycles.

Synthesis of SBA-15

Mesoporous SBA-15 material was prepared using molar gel composition 1TEOS: 0.016 P123: 127 H₂O: 0.46 HCl (35%) by the following reported procedure.^{1,2} In a typical synthesis 6.11 g amphiphilic triblock copolymer P123 was dispersed in 100 g distilled water and stirred for 4 h at 300 K. Afterwards, 3.028 g conc. HCl diluted with 42.47 g distilled water was mixed to the above dispersed solution and stirred for 2 h to get a clear solution. After complete dissolution of P123, 14 g of TEOS was added drop wise to the resulting solution and allowed to stir for 20 h at 313 K. Finally, the resulting mixture was hydrothermally treated in a Teflon-lined autoclave at 373 K for 2 days. The solid product was filtered through Buchner funnel and washed several times with distilled water and dried at oven at 353 K and calcined at 823 K for 6 h.

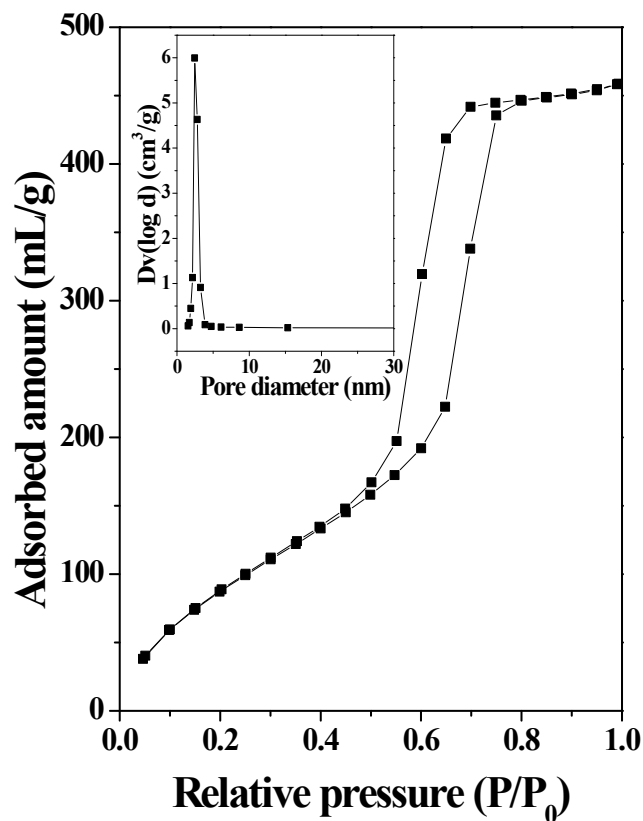


Fig. S4. N₂-adsorption-desorption isotherm for SBA-15. Inset shows pore size distribution.

Table S1. Textural properties of SBA-15.

Material	S_{BET} (m ² /g)	Ex. SA (m ² /g)	V_{Total} (mL/g)	Mesopore volume (mL/g)
SBA-15	624	553	0.84	0.79

Reference:

1. R. Srivastava, D. Srinivas and P. Ratnasamy, *J. Catal.*, 2005, **233**, 1-15.

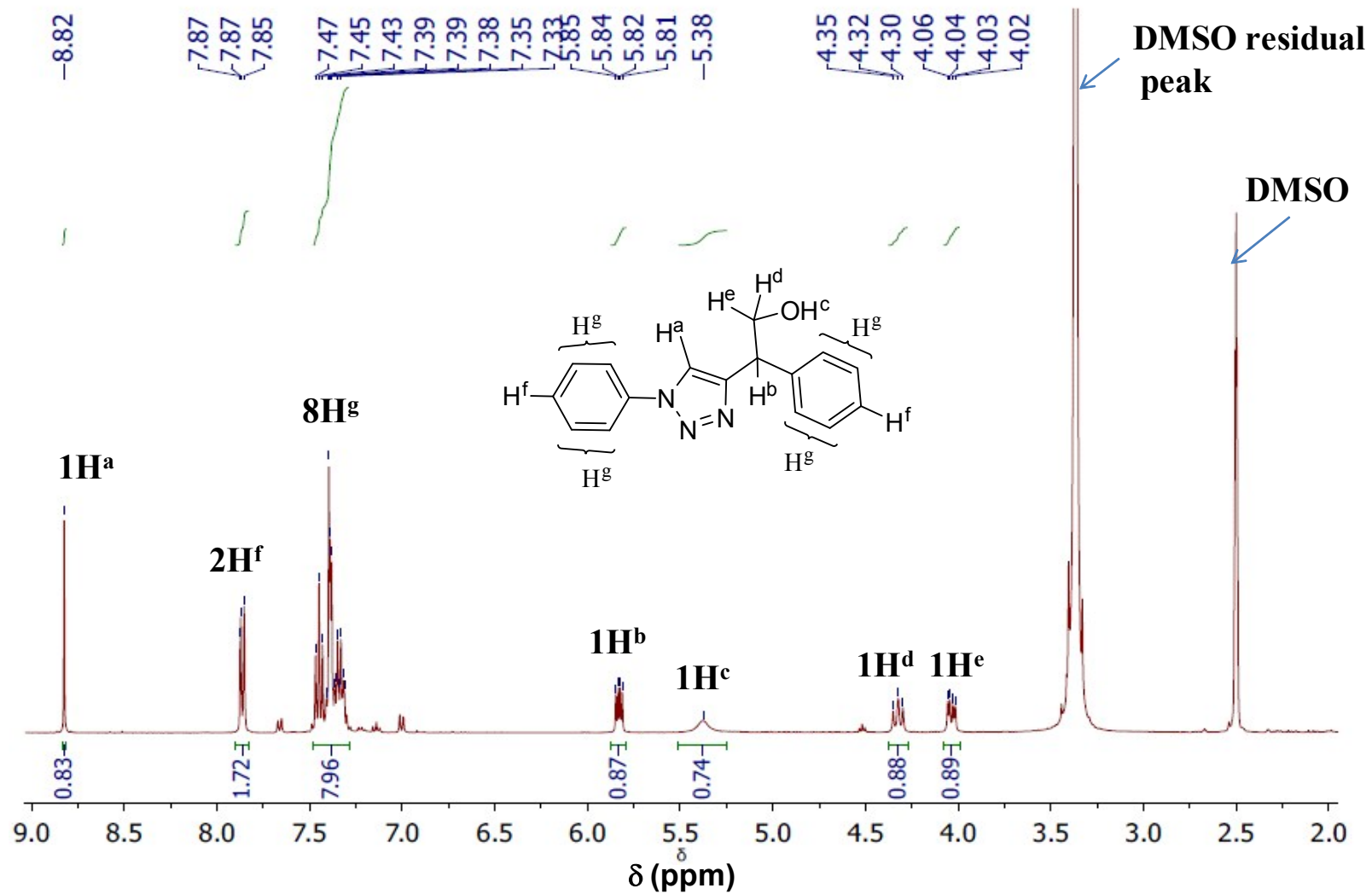


Figure S5. 1H NMR spectrum of 2-Phenyl-2-(4-phenyl-1H-1,2,3-triazol-1-yl)ethanol.

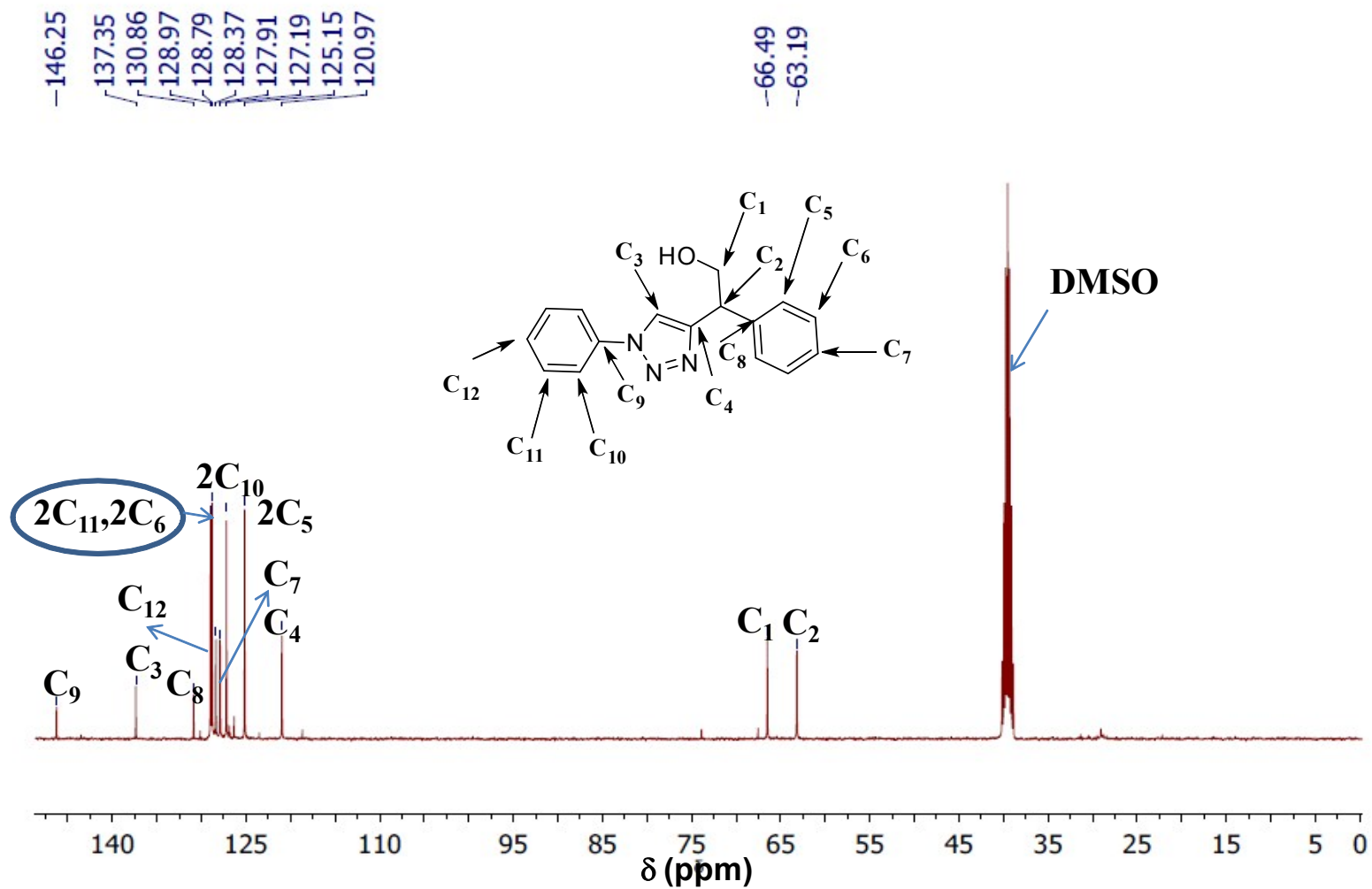


Figure S6. ¹³C NMR spectrum of 2-Phenyl-2-(4-phenyl-1H-1,2,3-triazol-1-yl)ethanol.

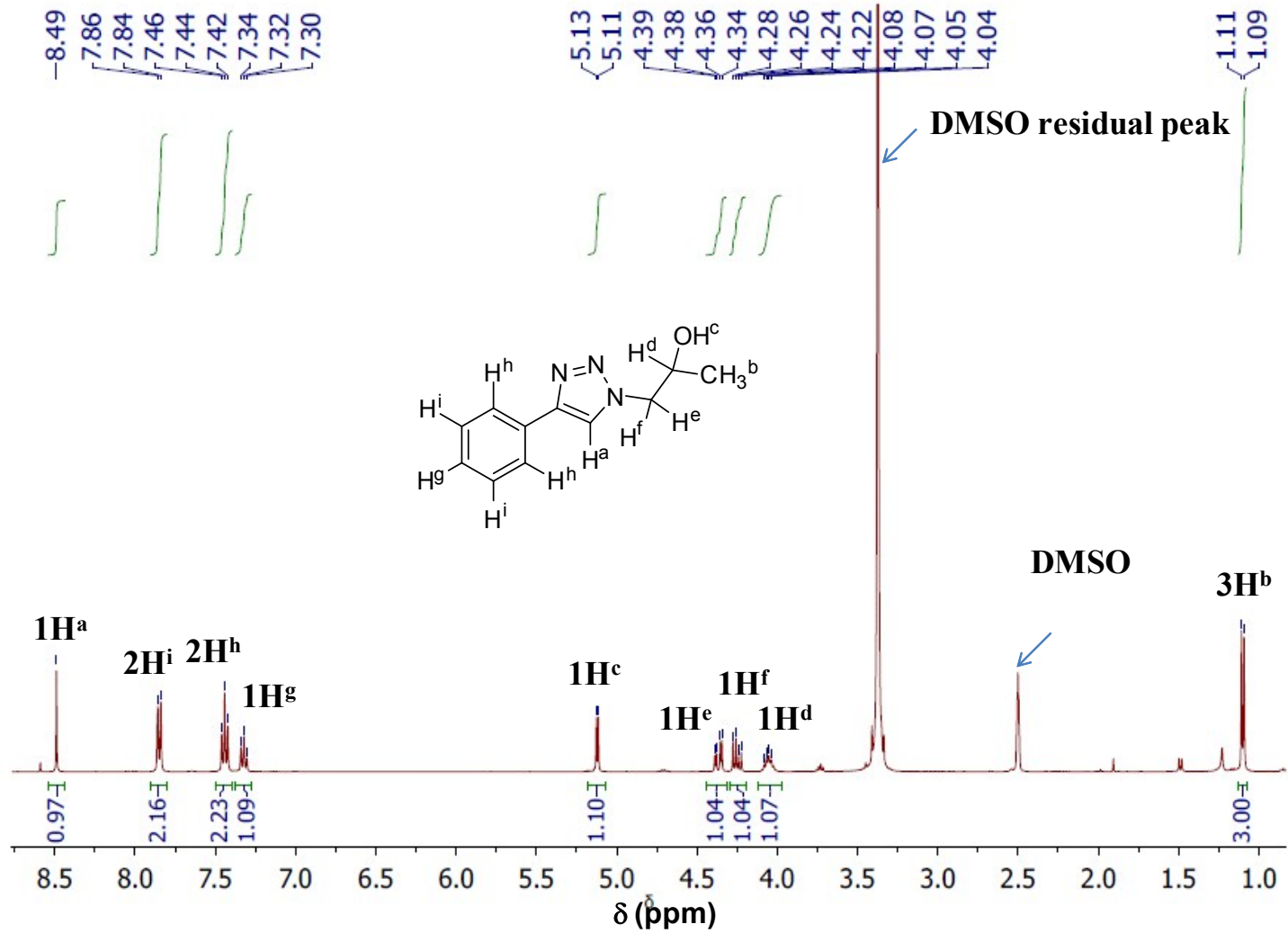


Figure S7. 1H NMR spectrum of 1-(4-Phenyl-1H-(4-phenyl-1,2,3-triazol-1-yl) propan-2-ol).

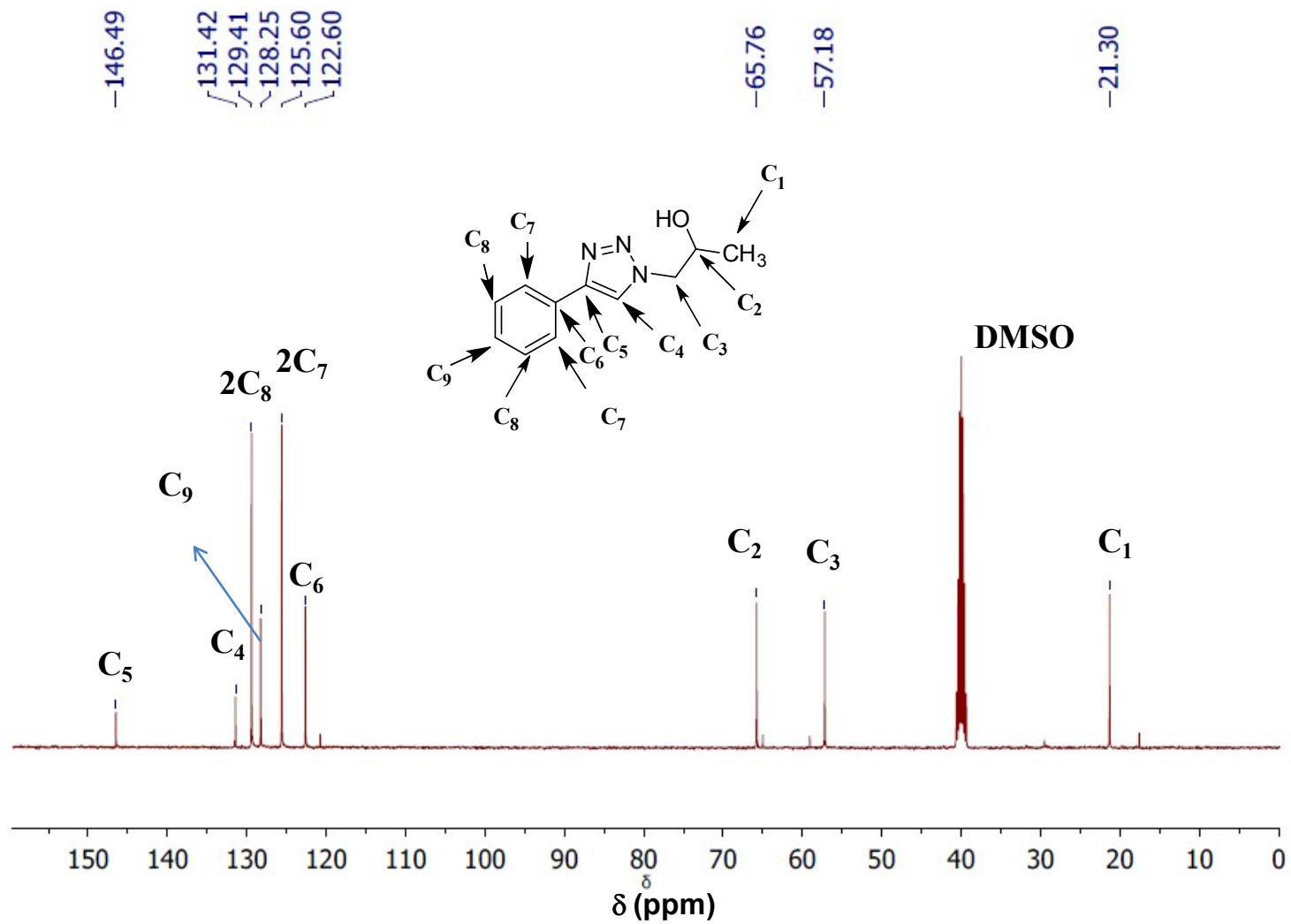


Figure S8. ¹³C NMR spectrum of 1-(4-Phenyl-1H-(4-phenyl-1,2,3-triazol-1-yl)propan-2-ol).

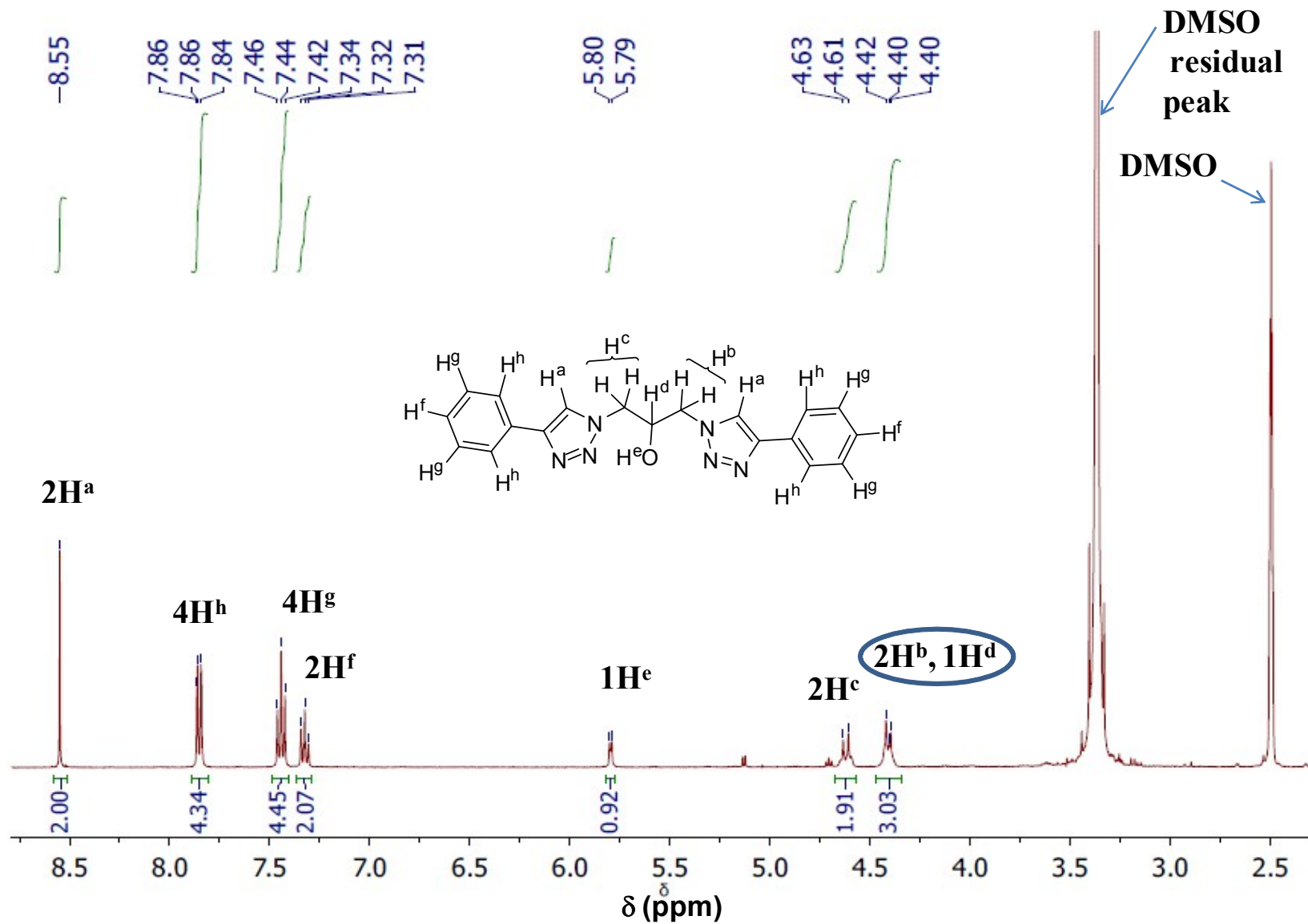


Figure S9. 1H NMR spectrum of 1,3-bis(5-phenyl-1H-1,2,3-triazol-1-yl)propan-2-ol.

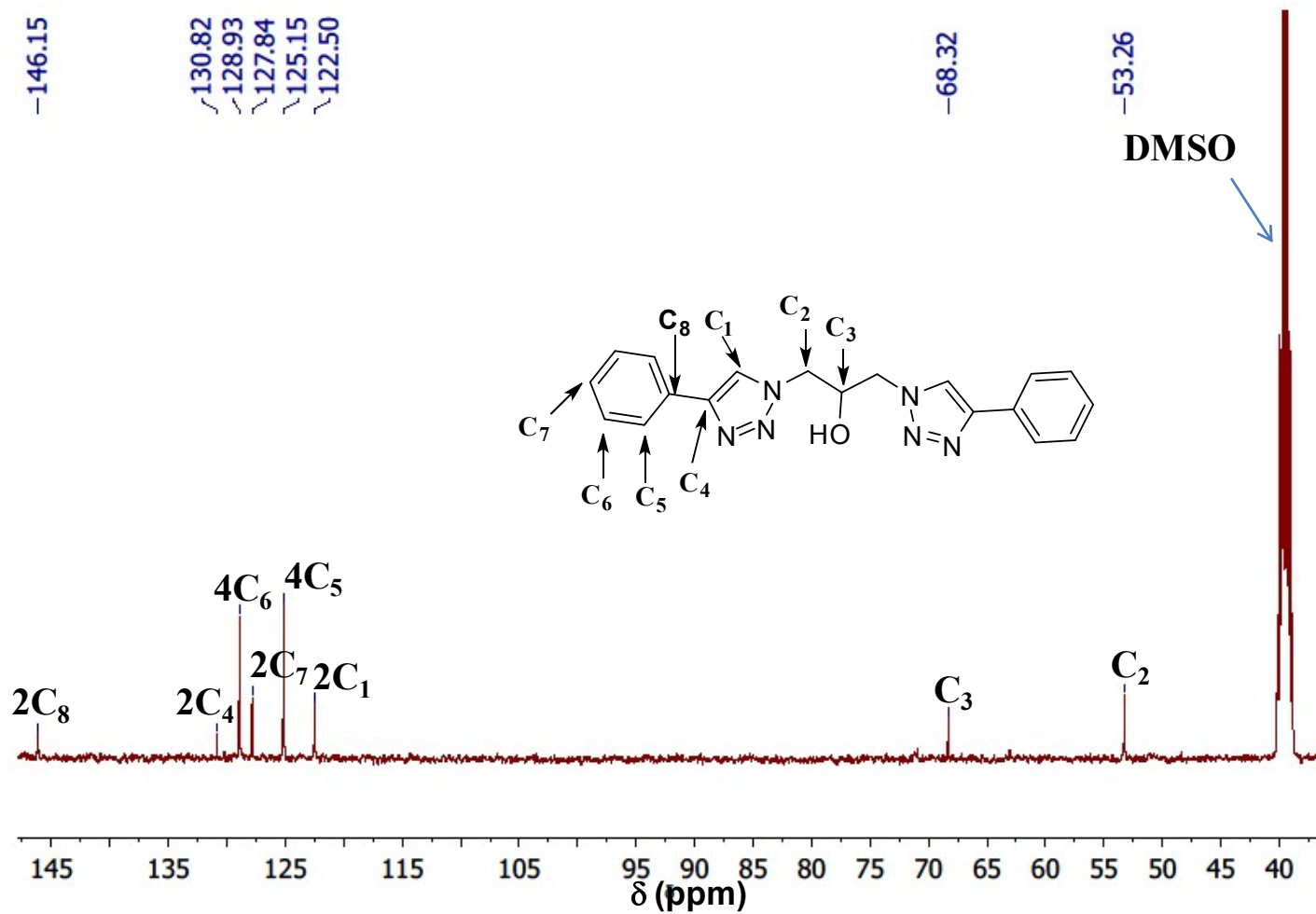


Figure S10. ¹³C NMR spectrum of 1,3-bis(5-phenyl-1H-1,2,3-triazol-1-yl)propan-2-ol.

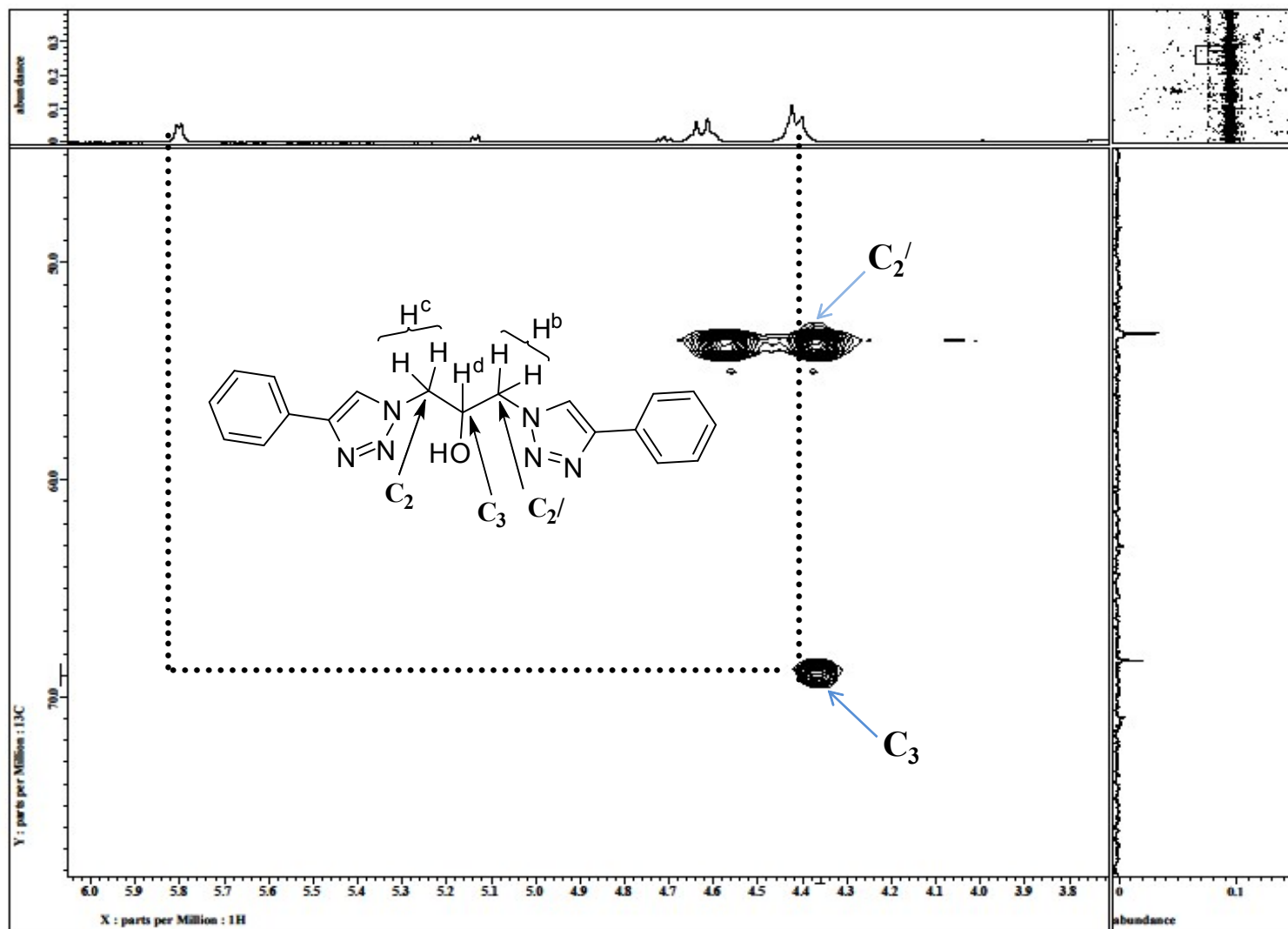


Figure S11. ^1H - ^{13}C HETCOR NMR spectrum of 1,3-bis(5-phenyl-1H-1,2,3-triazol-1-yl)propan-2-ol.

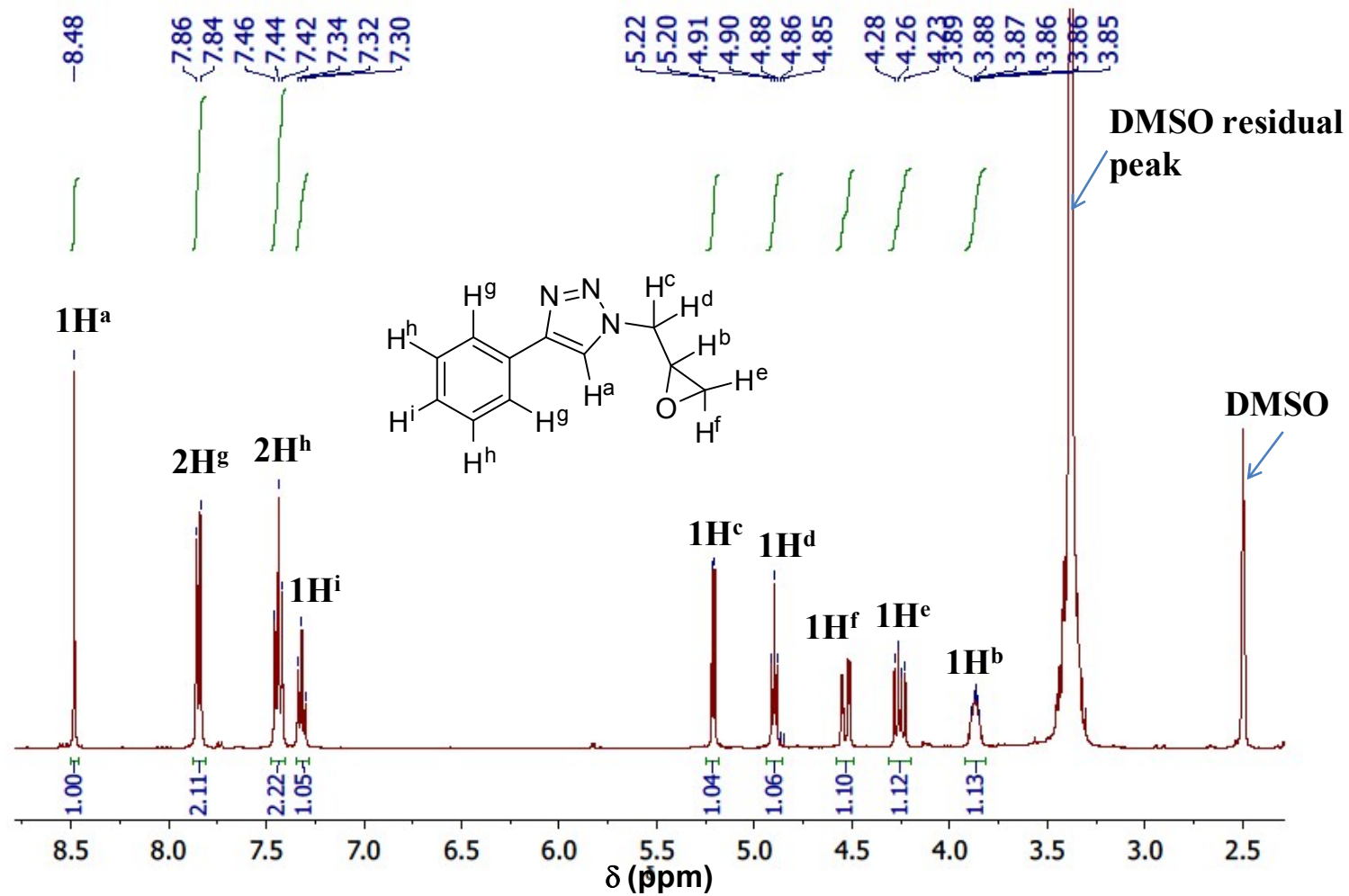


Figure S12. ¹H NMR spectrum of 1-(oxiran-2-ylmethyl)-4-phenyl-1H-1,2,3-triazole.

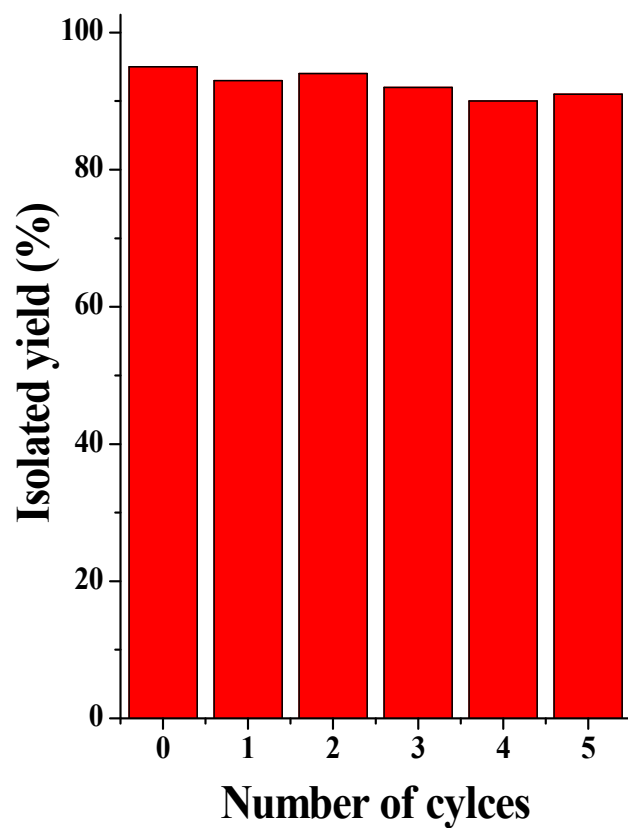


Figure S13. Catalytic performance of Cu(10%)/Meso-ZSM-5 during the multi-component reactions of phenyl acetylene, styrene oxide and sodium azide.