

Electronic Supplementary Information for:

Fluorine-Free Salts for Aqueous Lithium-ion and Sodium-ion Battery Electrolytes

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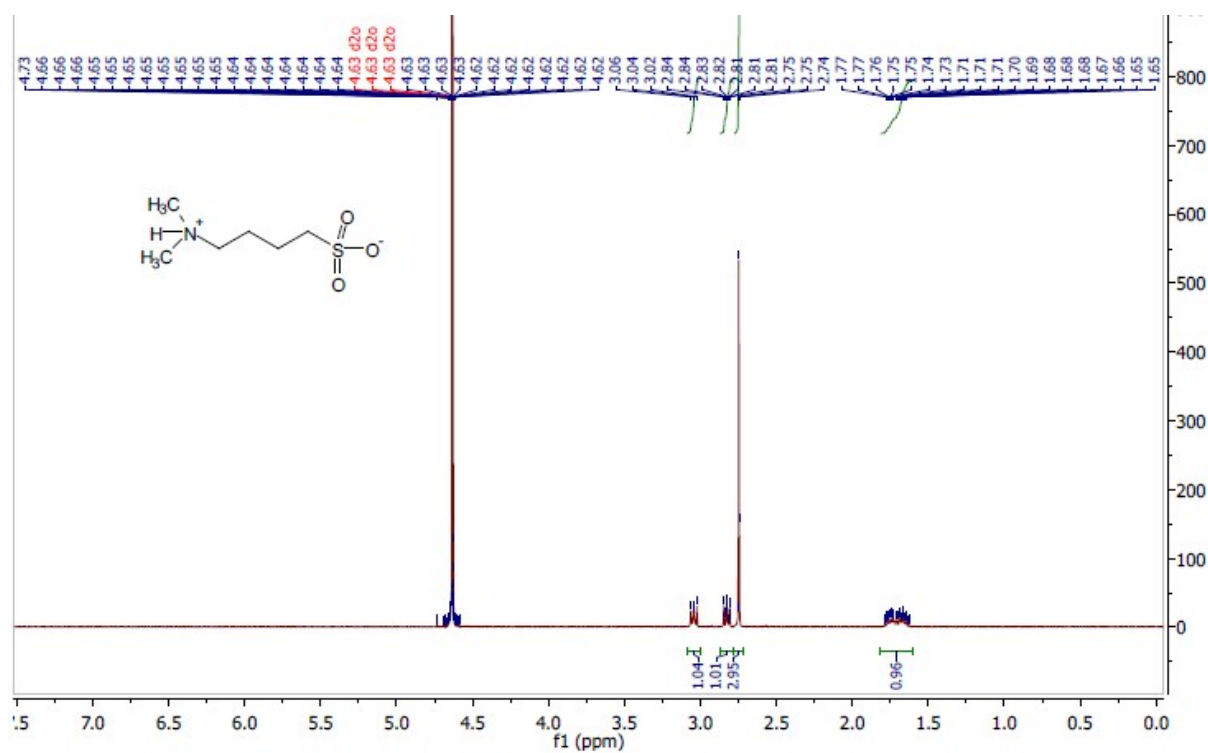


Figure S1: ¹H-NMR of ZDMBS

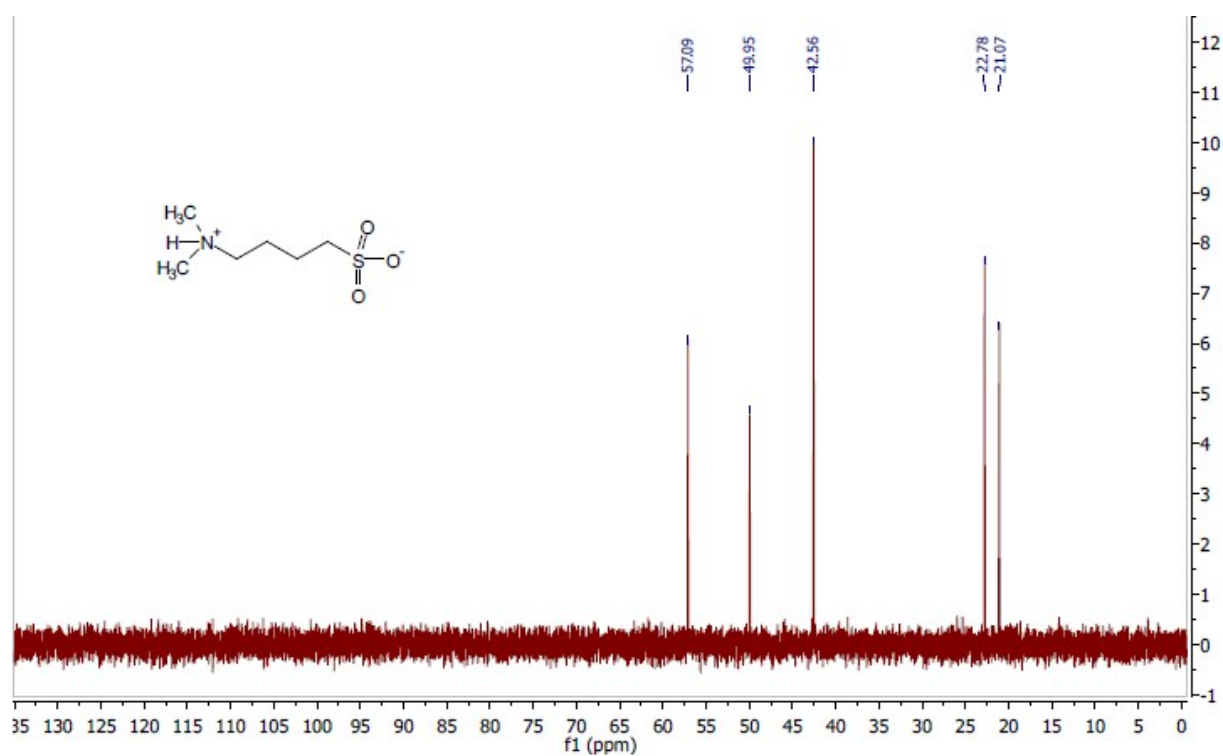


Figure S2: ¹³C-NMR of ZDMBS

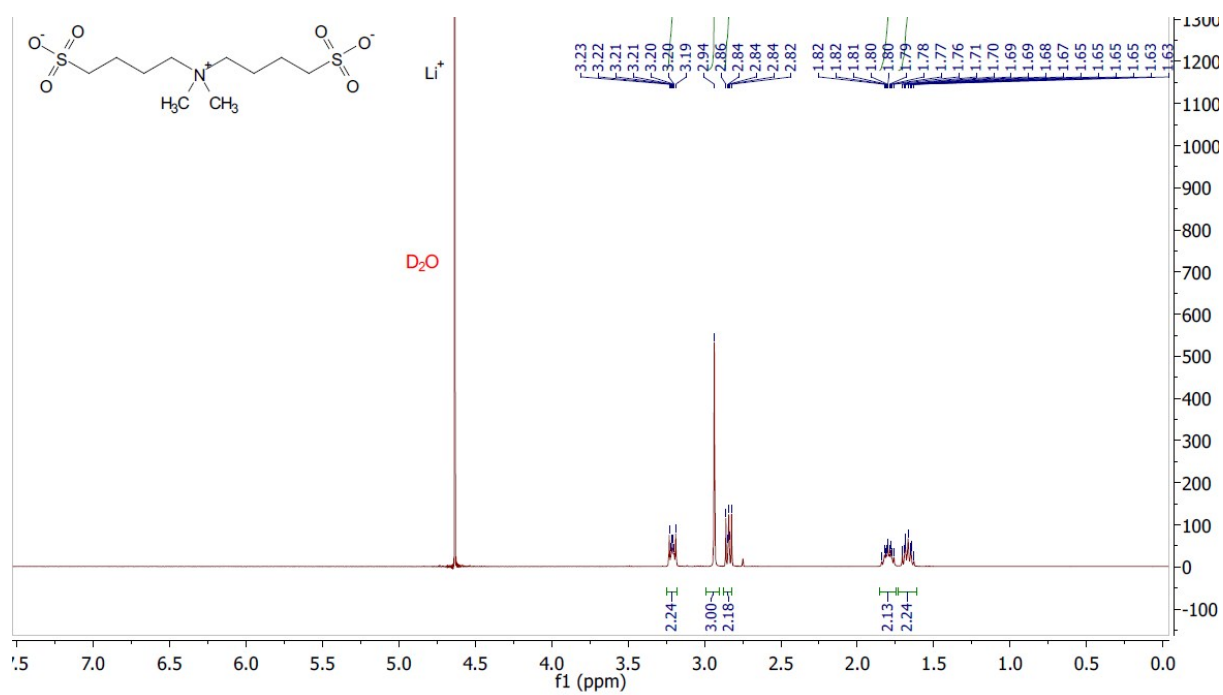


Figure S3: ¹H-NMR of LiMM4411

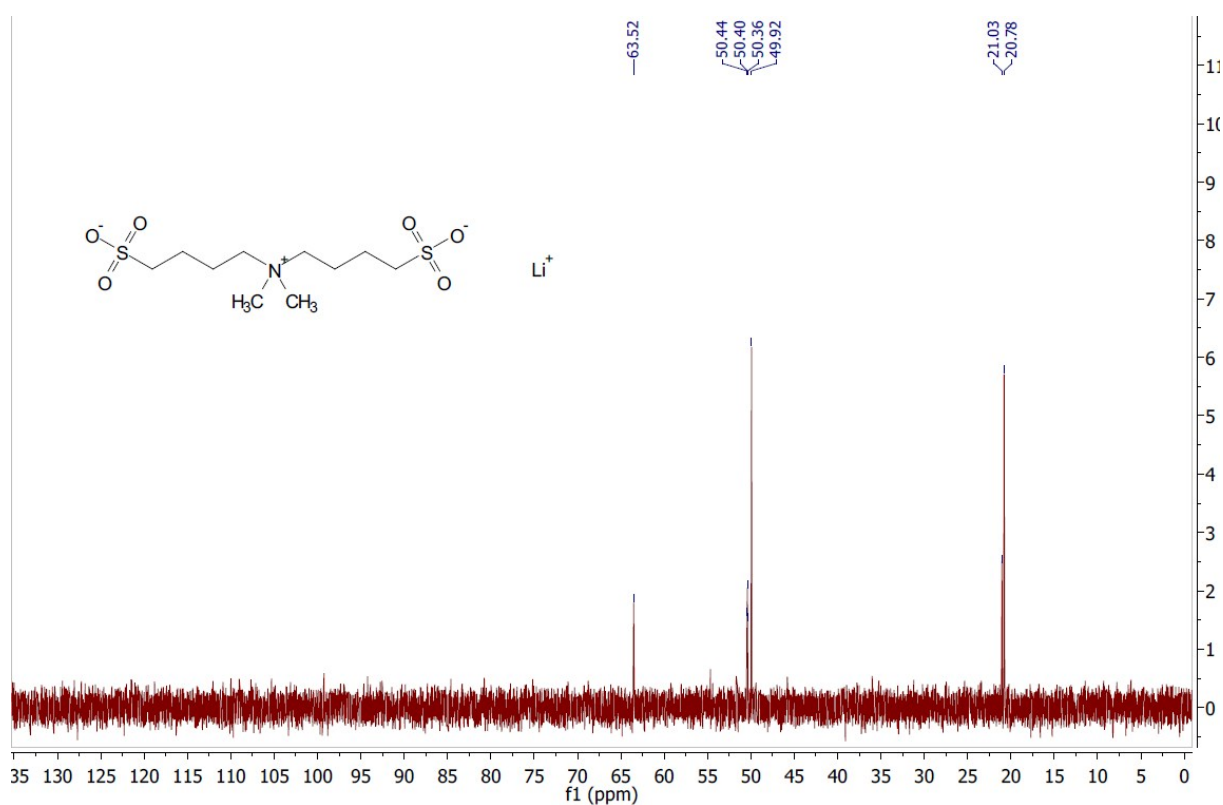


Figure S4: ¹³C-NMR of LiMM4411

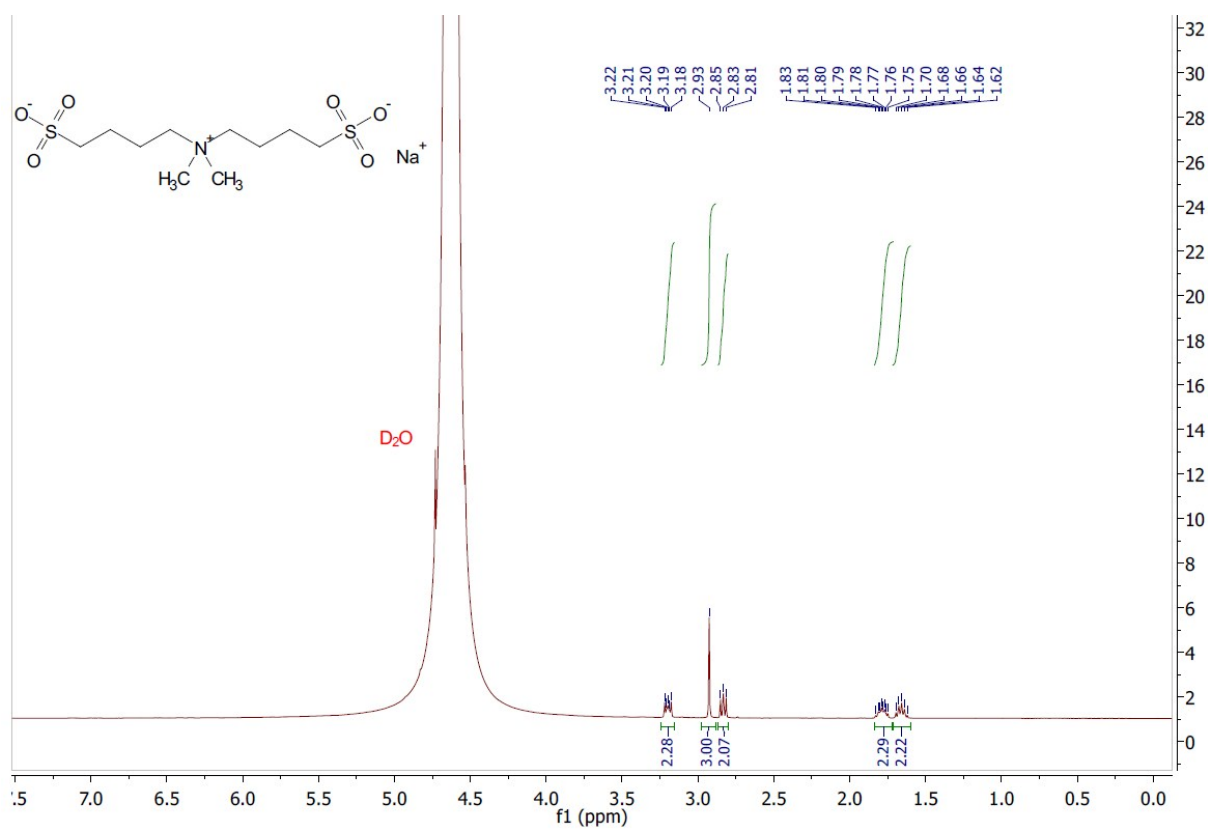


Figure S5: ¹H-NMR of NaMM4411

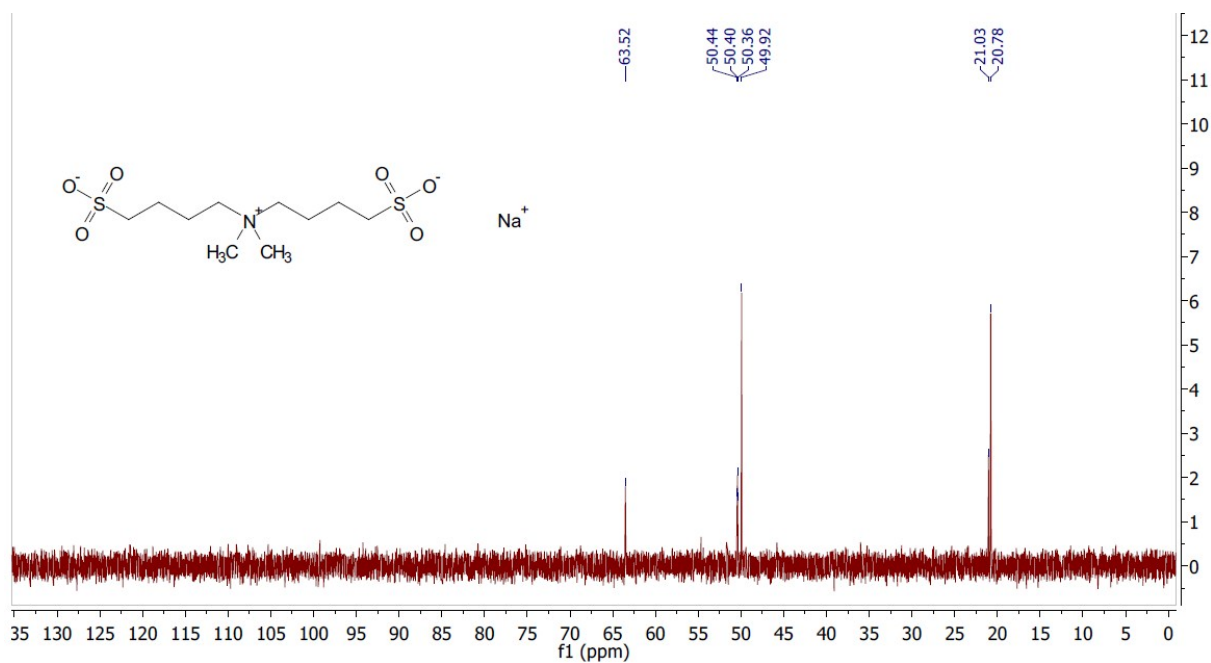


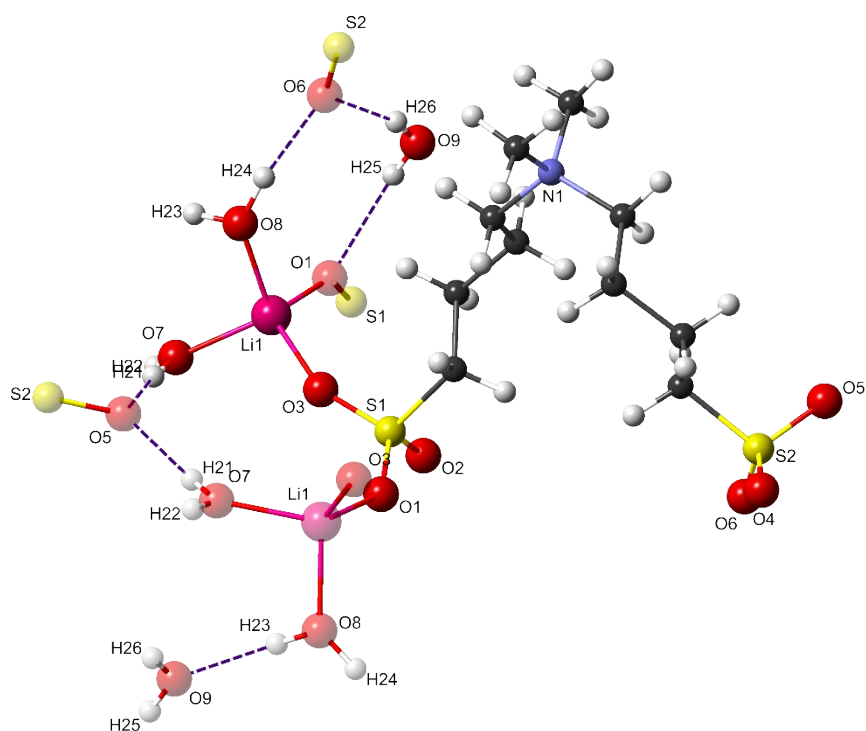
Figure S6: ¹³C-NMR of NaMM4411

Table S1. Selected dihedral angles in the crystals of LiMM4411 and NaMM4411. Atomic numbering is the same as in Figure 2.

	Dihedral angle / °	
	LiMM4411	NaMM4411
N1–C3–C4–C5	175.80	179.29
C3–C4–C5–C6	–70.29	–175.69
C4–C5–C6–S1	–175.86	177.18
N1–C7–C8–C9	165.08	172.01
C7–C8–C9–C10	–80.06	–73.31
C8–C9–C10–S2	–177.75	–170.68

Table S2. Hydrogen bond (HB) distances in the LiMM4411 crystal. The atomic numbering corresponds to the model illustrated below, where dotted lines indicate HBs.

LiMM4411	HB distance / Å
O7–H21...O5	2.047
O7–H22...O5	1.997
O8–H23...O9	1.959
O8–H24...O6	1.852
O9–H25...O1	2.222
O9–H26...O6	2.176



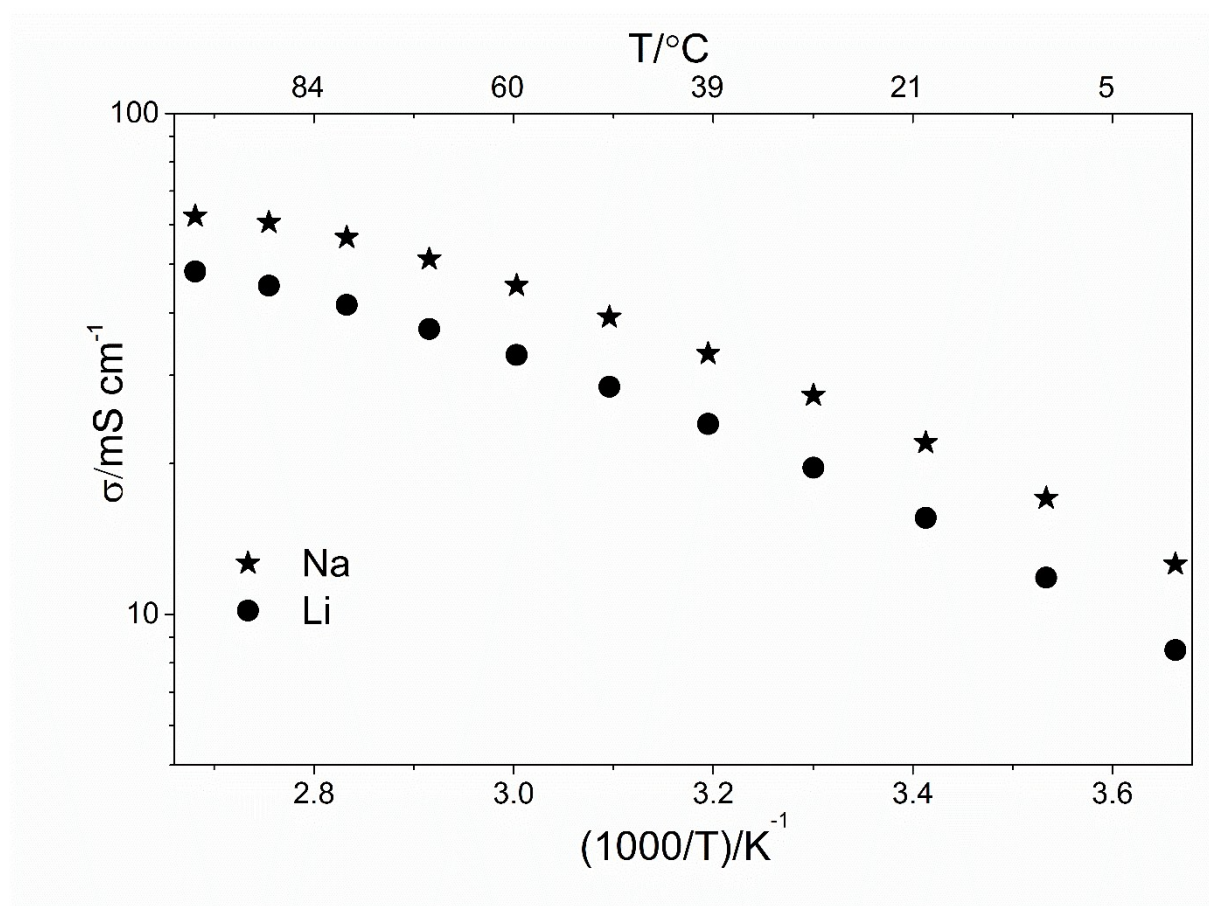


Figure S7: Arrhenius plots of ionic conductivities for heating scans of NaMM4411 and LiMM4411 1.5 m aqueous electrolytes.

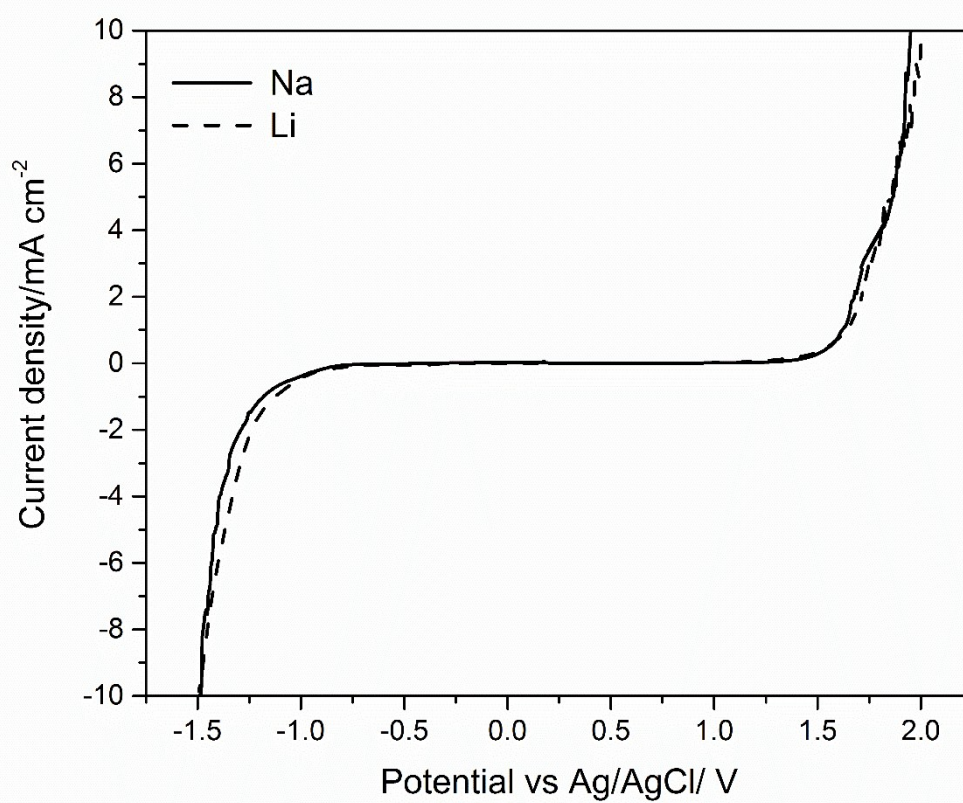


Figure S8: Electrochemical stability window of NaMM4411 and LiMM4411 1.5 m aqueous electrolytes.