



RSC Advances

ARTICLE

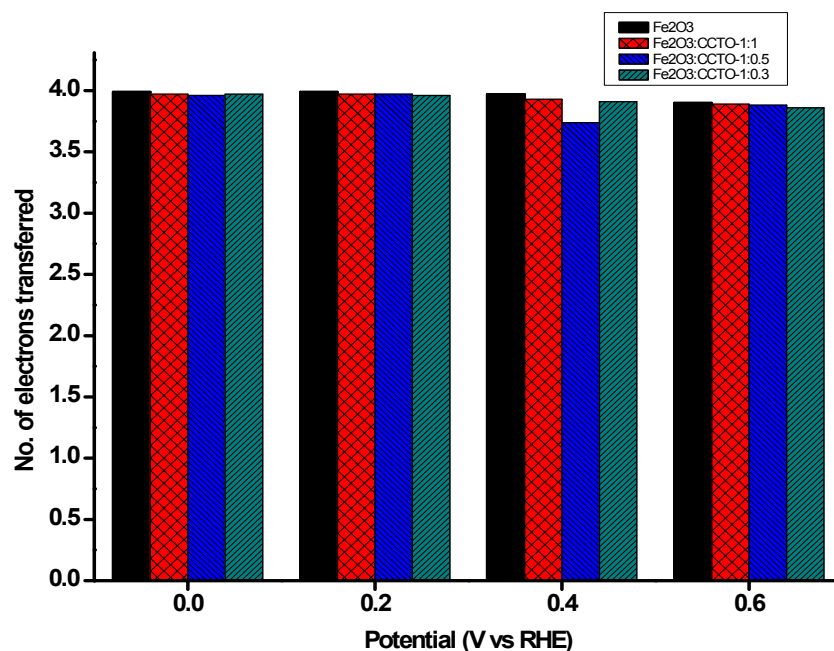
Supporting Information

S1 Table 1. Performance of different perovskite electrocatalysts for ORR

S.No.	Perovskite material	Onset potential (V vs RHE)	%H ₂ O ₂	References
1	LaNiO ₃	0.67V	<2%	1
2	LaCoO ₃	0.72V	<2%	1
3	LaFeO ₃	0.67V	<2%	1
4	LaMnO ₃	0.68V	<2%	1
5	LaCrO ₃	0.67V	<2%	1
6	La _{0.6} Ca _{0.4} CoO ₃ -C	1.30V (vs Hg/HgO)	3-14%	6
7	La _{0.6} Ca _{0.4} CoO ₃	-0.70V (vs Hg/HgO)	3-14%	6
8	BaTiO _{3-x}	0.85V	15%	10
9	La _x (Ba _{0.5} Sr _{0.5}) _{1-x} Co _{0.8} Fe _{0.2} O _{3-δ} (BSCF)	0.70V	15%	12
10	BaMnO ₃	0.77V	--	15
11	BaMnO ₃ @5%C	0.84V	--	15
12	La _{0.8} Sr _{0.2} MnO ₃	---	0%	18
13	Sr ₂ FeMoO ₆	-0.15V (vs Hg/HgO)	---	20
14	Sr ₂ CoMoO ₆	-0.15V (vs Hg/HgO)	---	20
15	CaCu ₃ Ti ₄ O ₁₂ - Fe ₂ O ₃	0.77 V	2.33%	in this article

S2. Table 2 Performance of synthesized catalyst for ORR

S.No.	Name of Material	Onset Potential (V vs RHE)	% H ₂ O ₂ (V vs RHE)			
			0V	0.2V	0.4V	0.6V
1	Fe ₂ O ₃	0.73	0.263	0.468	1.28	4.80
2	Fe ₂ O ₃ - CCTO- 1:0.3	0.75	1.52	1.509	3.58	5.72
3	Fe ₂ O ₃ - CCTO- 1:0.5	0.76	1.84	1.38	4.58	5.56
4	Fe ₂ O ₃ - CCTO- 1:1	0.77	1.37	1.76	4.31	7.08

S3. Plot showing number of electrons transferred by Fe₂O₃, Fe₂O₃ – CCTO 1:1, Fe₂O₃ – CCTO 1:0.5 and Fe₂O₃ – CCTO 1:0.3 at potentials 0V, 0.2V, 0.4V and 0.6V (vs RHE).

S4. Elemental analysis confirming presence of components from both CCTO and Fe_2O_3 in the composites

