

**Theoretical Investigations of the Small Molecule Acceptor
Materials Based on Oligothiophene - Naphthalene Diimide in
Organic Solar Cells**

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Table S1 The test HOMO energy levels of **NDI-2T₃Me** at the **6-31G**** and **6-311G**** basis set levels. All energies are in eV.

Opt test	B3LYP	M06	PBE0	PBE0	Exp
	6-31G**			6-311G**	
HOMO	-5.18	-5.47	-5.43	-5.64	-5.6

Table S2 The test electronic absorption spectra of **NDI-2T₃Me** at the **6-311G**** basis set levels based on the optimization structure.

Absorption test	B3LYP	PBE0	BMK	CAM-B3LYP	Exp
λ/nm	778	718	599	521	607

Table S3 The calculated spectroscopic properties of **NDI-T₁Me—NDI-T₄Me**. All absorption wavelengths are in nm. (Assignment: H=HOMO, L=LUMO, L+2=LUMO+2.)

Oligomers	Transition	λ_{max}	f	Main configuration	λ_{EXP}
NDI-2T₁Me	S ₀ -S ₁	497	0.4836	H→L (96%)	525
	S ₀ -S ₉	292	0.4365	H→L+2 (49%)	
NDI-2T₂ Me	S ₀ -S ₁	576	0.8875	H→L (95%)	604
	S ₀ -S ₄	340	1.4851	H→L+2 (64%)	
NDI-2T₃ Me	S ₀ -S ₁	614	1.1813	H→L (92%)	632
	S ₀ -S ₄	376	2.1228	H→L+2 (63%)	
NDI-2T₄ Me	S ₀ -S ₁	631	1.4124	H→L (86%)	638
	S ₀ -S ₄	407	2.7418	H→L+2 (54%)	

Table S4 Electron density difference plots of electronic transition $S_0 \rightarrow S_1$ for **NDI-T₁Me** — **NDI-T₄Me**. ΔD is the electron transfer distance, Δq is the fraction of electron exchange, Ω is the overlap (normalization to the exchanged charge) between the regions of density depletion and increment.

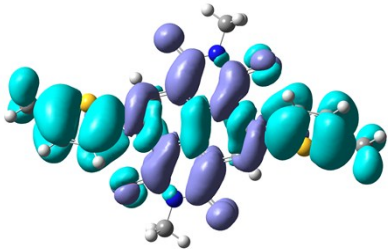
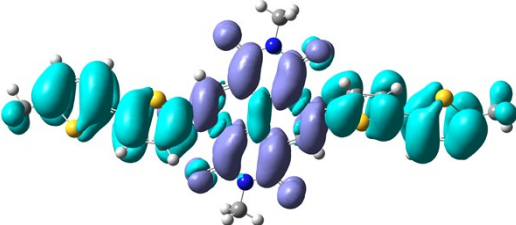
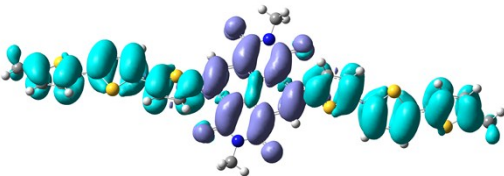
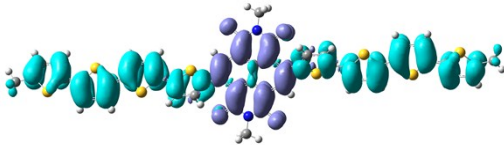
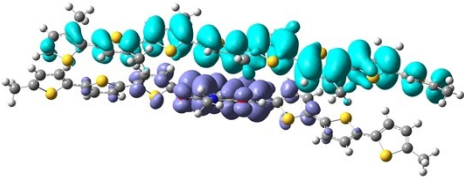
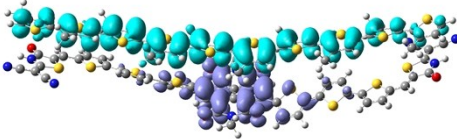
Oligomers	Electronic differential density plots	Data analysed
NDI-2T ₁ Me		$\Delta D=0.202\text{\AA}$ $\Delta q=1.0148$ $\Omega=0.6016$ $S_0 \rightarrow S_1$
NDI-2T ₂ Me		$\Delta D=0.269\text{\AA}$ $\Delta q=1.1073$ $\Omega=0.5797$ $S_0 \rightarrow S_1$
NDI-2T ₃ Me		$\Delta D=0.497\text{\AA}$ $\Delta q=1.1352$ $\Omega=0.5246$ $S_0 \rightarrow S_1$
NDI-2T ₄ Me		$\Delta D=0.817\text{\AA}$ $\Delta q=1.1017$ $\Omega=0.4201$ $S_0 \rightarrow S_1$

Table S5 The hole mobility (μ_h) and its relative parameters (r is the intermolecular stacking distance, t_+ is the charge transfer integral, k_{CT-h} is the charge transfer rate) for **NDI-T₁DCRD**, **NDI-T₄DCRD** and **NDI-T₃Me**.

Oligomers	λ_h/eV	r/eV	V_+/eV	K_{CT-h}/S^{-1}	$\mu_h/cm^2 \cdot (V \cdot S)^{-1}$
NDI-2T₃Me	0.257	4.950	0.0044	5.32×10^{10}	2.54×10^{-3}
NDI-2T₃DCRD	0.181	4.825	0.0129	1.14×10^{12}	5.15×10^{-2}
NDI-2T₄DCRD	0.182	4.825	0.0021	2.98×10^{10}	1.35×10^{-3}

Table S6 Electron density difference plots of electronic transition $S_0 \rightarrow S_1$ for active blend layers **P3HT/Acceptor**. ΔD is the electron transfer distance, Δq is the fraction of electron exchange, Ω is the overlap (normalization to the exchanged charge) between the regions of density depletion and increment.

Oligomers	Electronic differential density plots	Data analysed
P3HT/NDI-2T ₃ Me		S1: $\Delta D=5.342\text{\AA}$ $\Delta q=1.4357$ $\Omega=0.0987$
P3HT/NDI-2T ₃ DCRD		S1: $\Delta D=5.924\text{\AA}$ $\Delta q=1.3537$ $\Omega=0.0168$

Oligomers

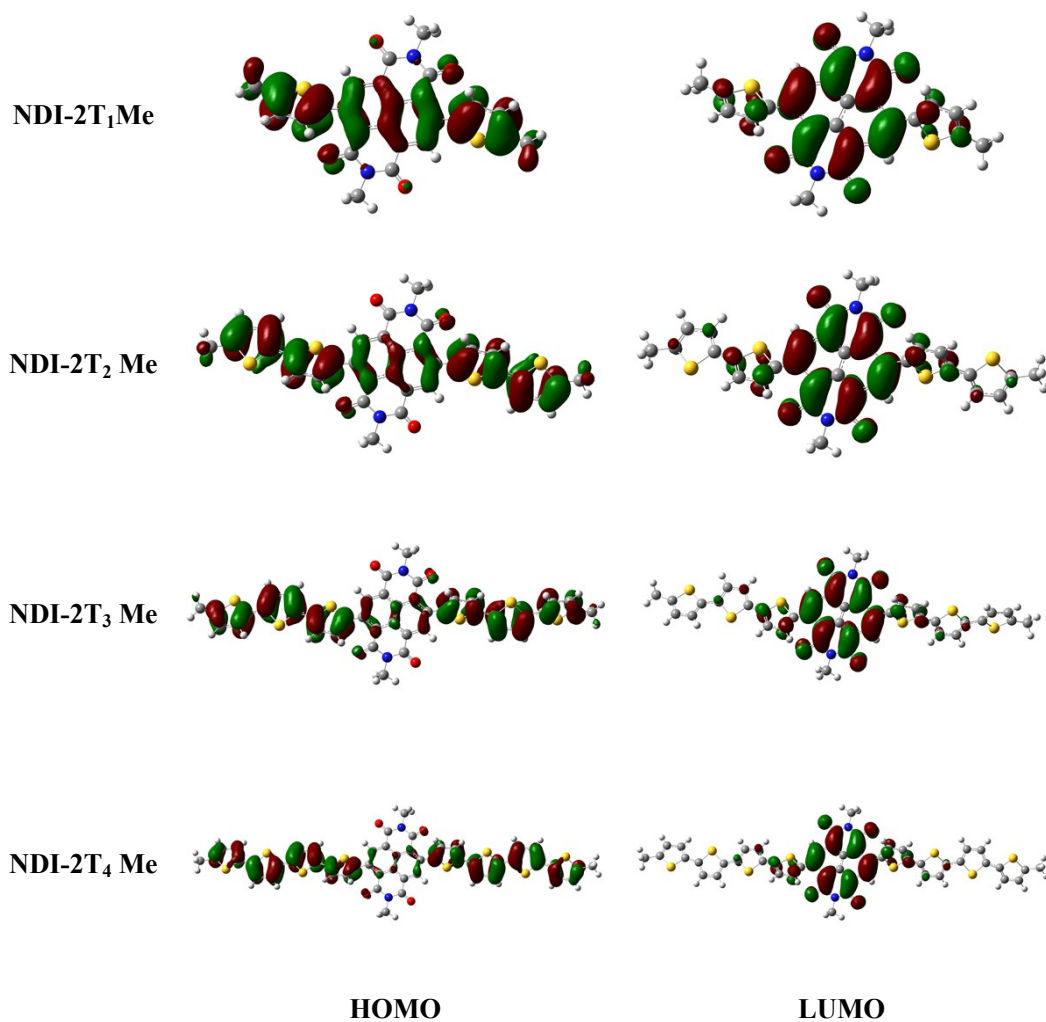
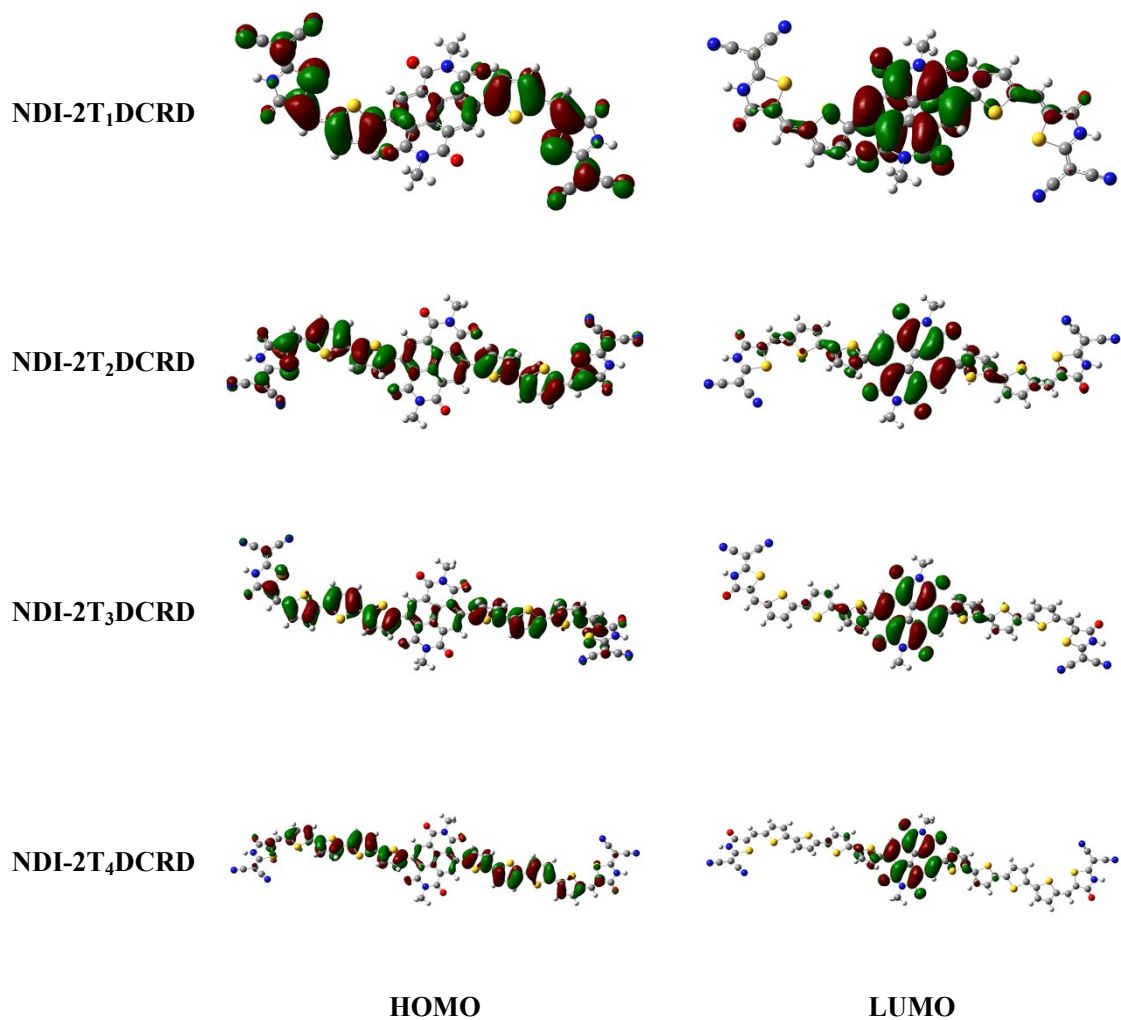


Fig. S1 The distributions of Frontier Molecular Orbitals of NDI-T₁Me—NDI-T₄Me.

Oligomers



HOMO

LUMO

Fig. S2 The distributions of Frontier Molecular Orbitals.

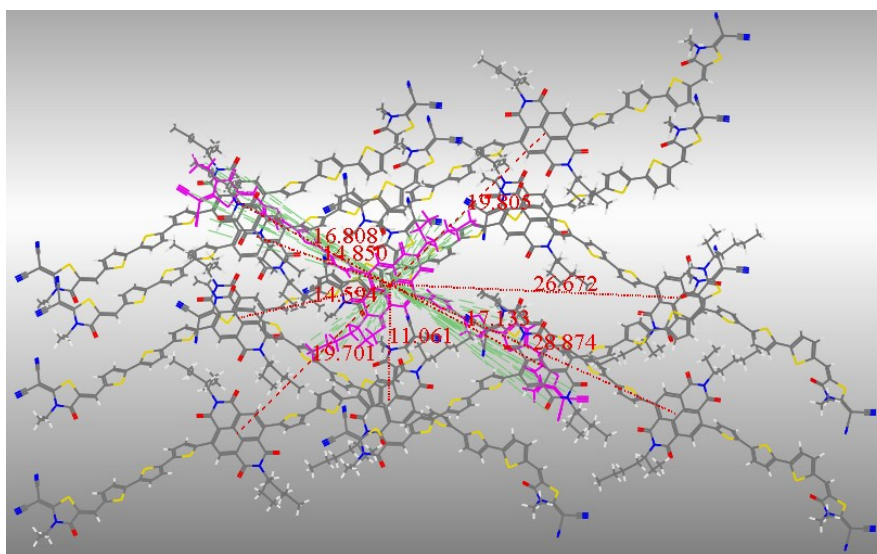


Fig. S3 The possible transfer paths of **NDI-2T₃DCVRD** in a supercell.

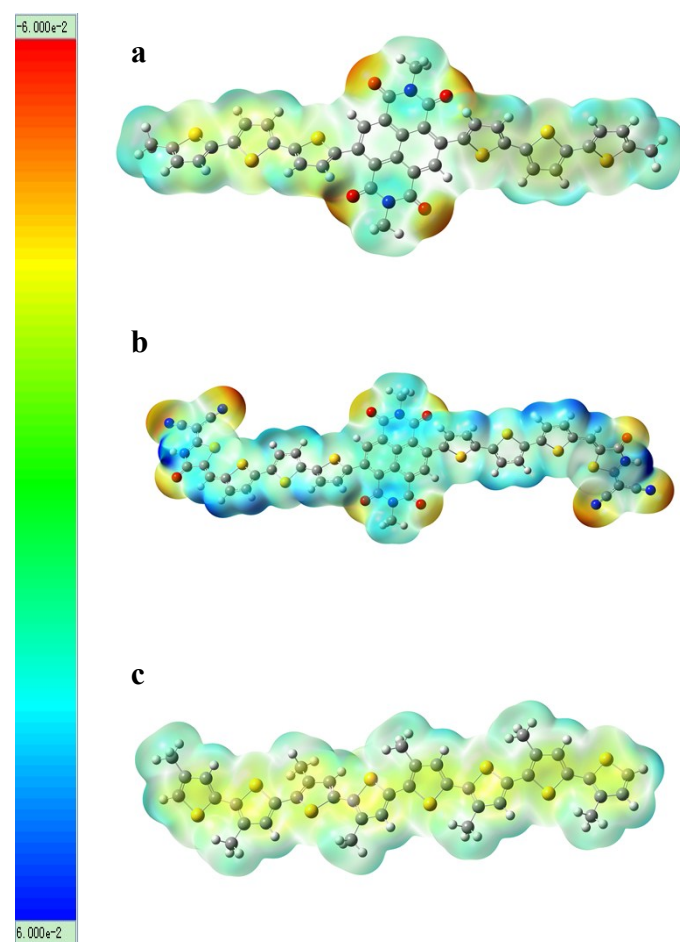


Fig. S4 The scatter diagrams of electrostatic potential of **NDI-2T₃Me** (a), **NDI-2T₃DCRD** (b) and **P3HT** (c).

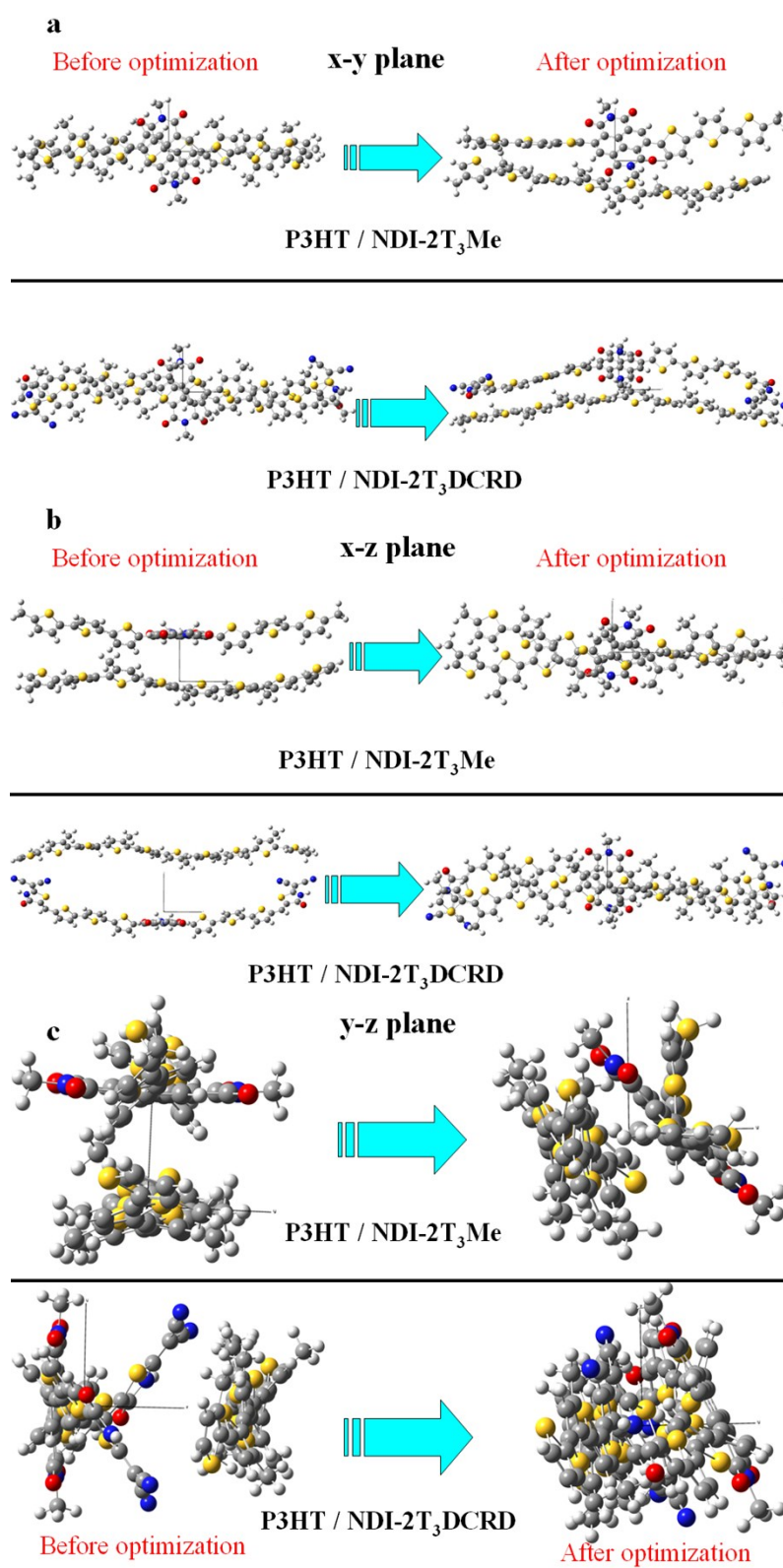
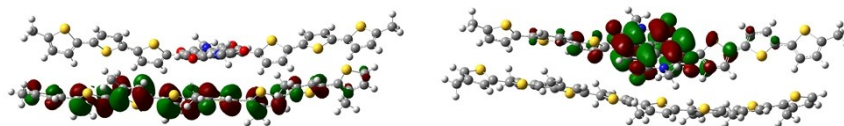
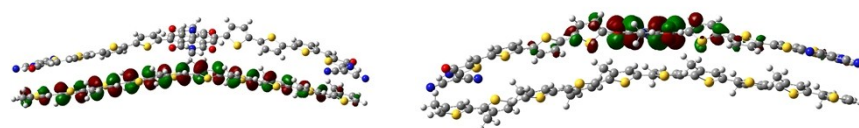


Fig. S5 The different-plane spatial structure of **P3HT/Acceptor** based on the optimized geometry.

P3HT/NDI-2T₃
Me



P3HT/NDI-
2T₃DCRD



HOMO

LUMO

Fig. S6 The distributions of Frontier Molecular Orbitals of blend **P3HT/Acceptor**.