

Optical and surface band bending mediated fluorescence sensing properties of MoS₂ quantum dots

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Supporting Information

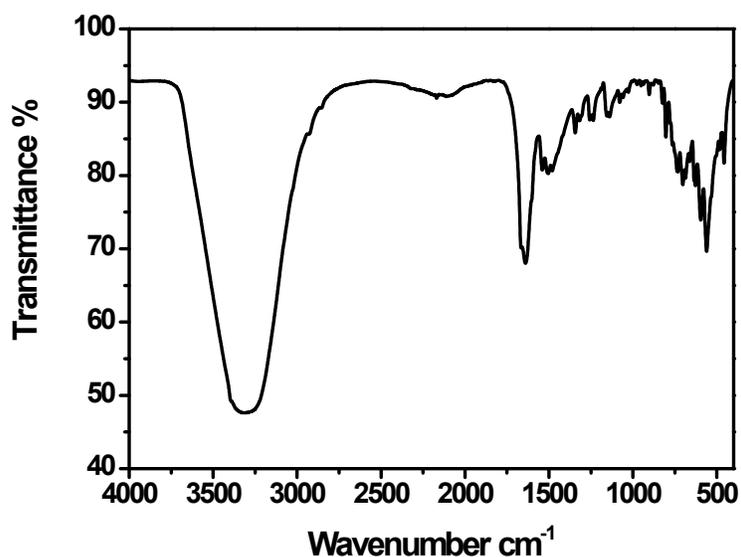


Figure S1: FTIR spectrum of prepared MoS₂ quantum dots.

Figure S1 depicts the FTIR spectrum of the prepared MoS₂ quantum dots. A characteristic peak found at 480 cm⁻¹ is attributed to the Mo-S vibrations. The bands observed around 581, 1112 and 1397 cm⁻¹ are assigned to the sulfate vibrations. The peaks around 1572

and 1637 cm^{-1} are assigned to the N-H in-plane stretching vibration and N-H bending vibrations respectively. The broad band around 3315 cm^{-1} is attributed to the N-H stretching vibrations.

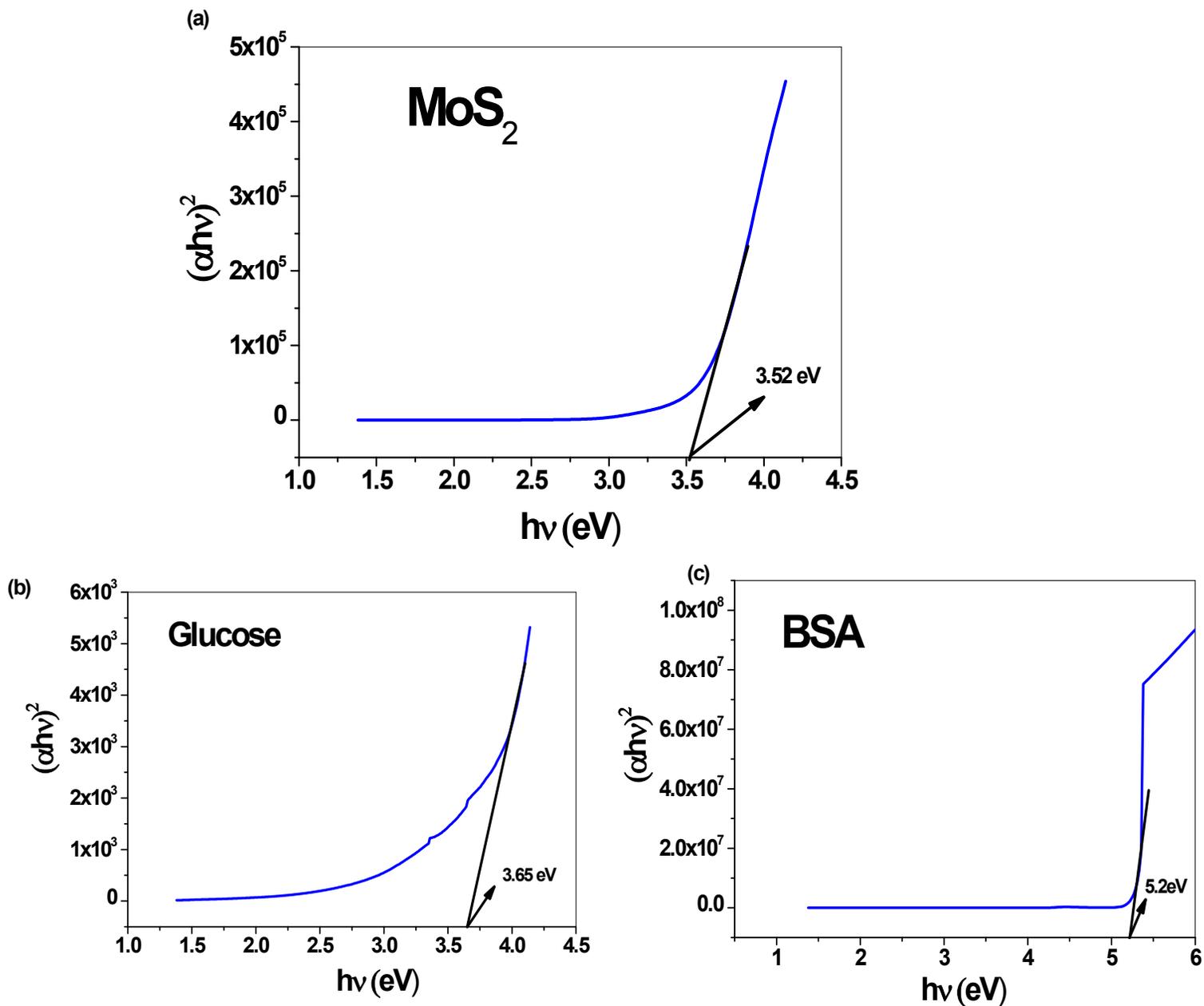


Figure S2: $(\alpha h\nu)^2$ vs $h\nu$ plots of (a) MoS_2 (b) glucose and (c) BSA

From the $(\alpha h\nu)^2$ vs $h\nu$ plots, the direct band gap of MoS₂ is found to be 3.52eV which is lesser than the HOMO-LUMO gap of glucose (3.65 eV) and BSA (5.2 eV). It supports that the electron transfer is possible only from glucose and BSA to MoS₂.