

Highly efficient continuous-flow oxidative coupling of amines using promising nanoscale CeO₂-M/SiO₂ (M = MoO₃ and WO₃) solid acid catalysts

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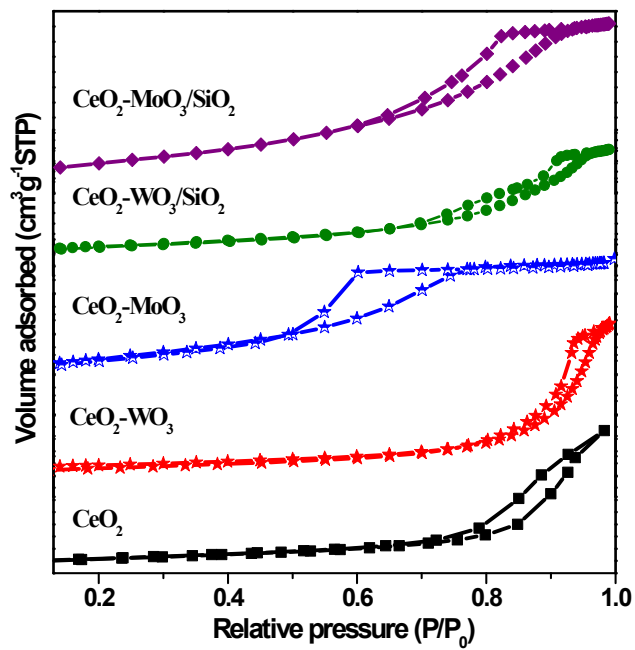


Fig.S1 N₂ adsorption–desorption isotherms of CeO₂, CeO₂-WO₃, CeO₂-MoO₃, CeO₂-WO₃/SiO₂ and CeO₂-MoO₃/SiO₂ catalysts.

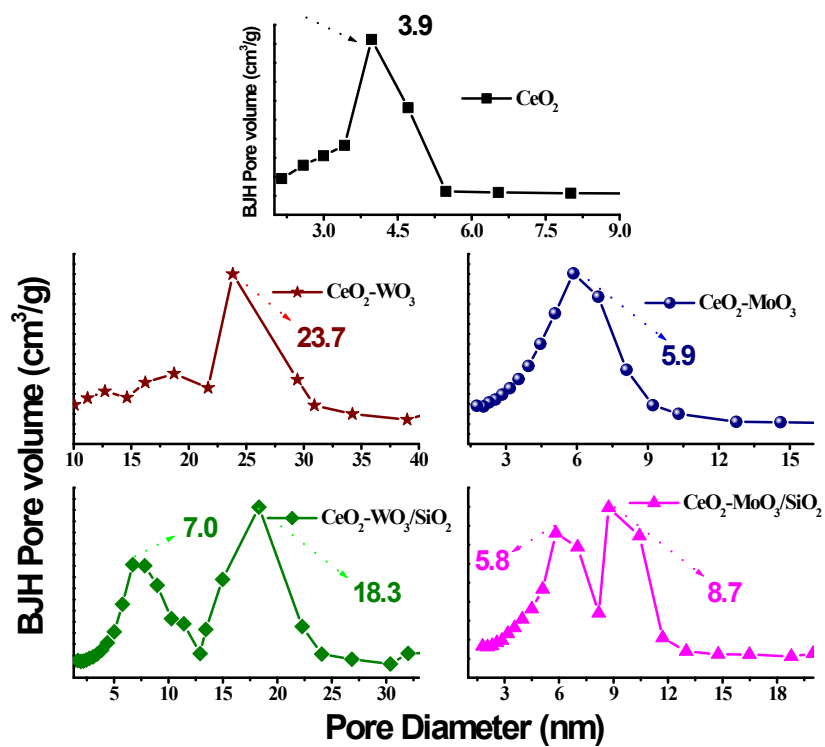


Fig.S2 Pore size distribution profiles of CeO₂, CeO₂-WO₃, CeO₂-MoO₃, CeO₂-WO₃/SiO₂ and CeO₂-MoO₃/SiO₂ samples.

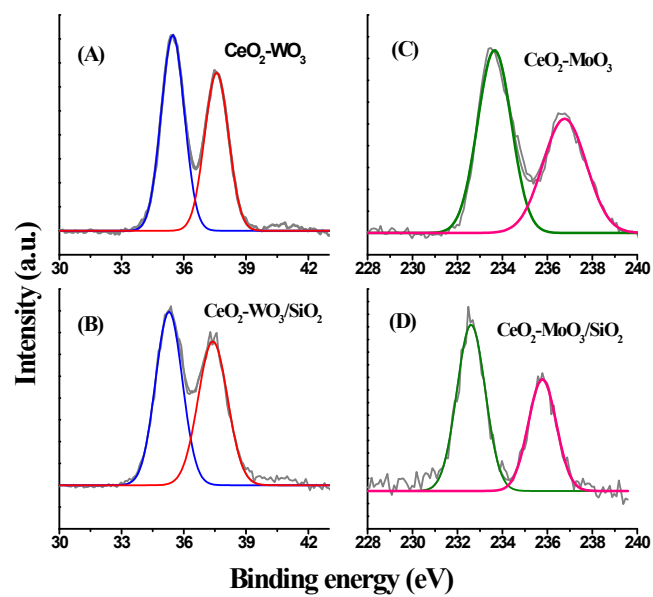


Fig. S3 (A-B) W 4f core level XP spectra of $\text{CeO}_2\text{-WO}_3$ and $\text{CeO}_2\text{-WO}_3/\text{SiO}_2$ catalysts and (C-D) Mo 3d core level XP spectra of $\text{CeO}_2\text{-MoO}_3$ and $\text{CeO}_2\text{-MoO}_3/\text{SiO}_2$ catalysts.

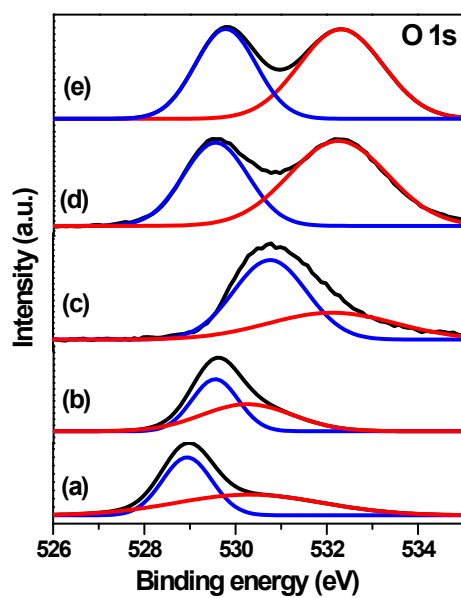


Fig. S4 The O 1s core level XP spectra of (a) CeO₂, (b) CeO₂-WO₃, (c) CeO₂-MoO₃, (d) CeO₂-WO₃/SiO₂ and (e) CeO₂-MoO₃/SiO₂ catalysts.

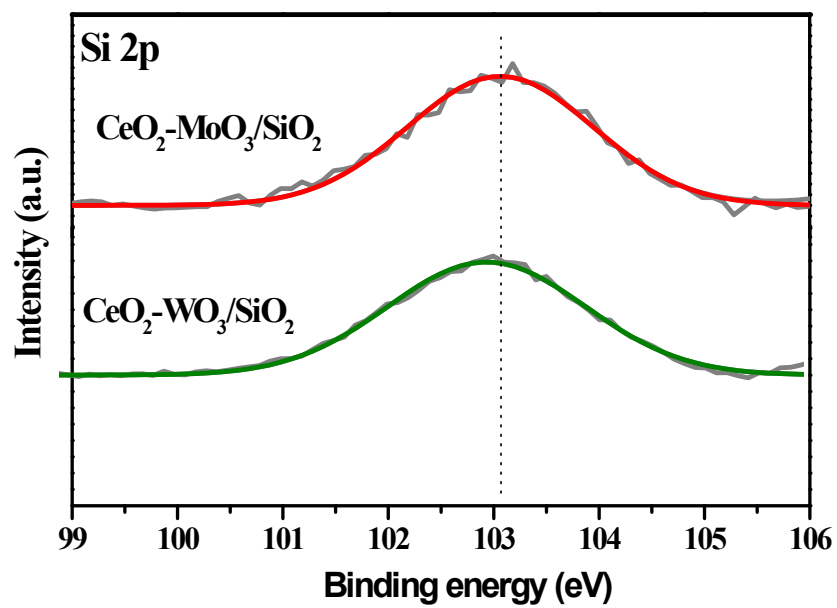


Fig. S5 The Si 2p XP spectra of $\text{CeO}_2\text{-WO}_3/\text{SiO}_2$ and $\text{CeO}_2\text{-MoO}_3/\text{SiO}_2$ samples.

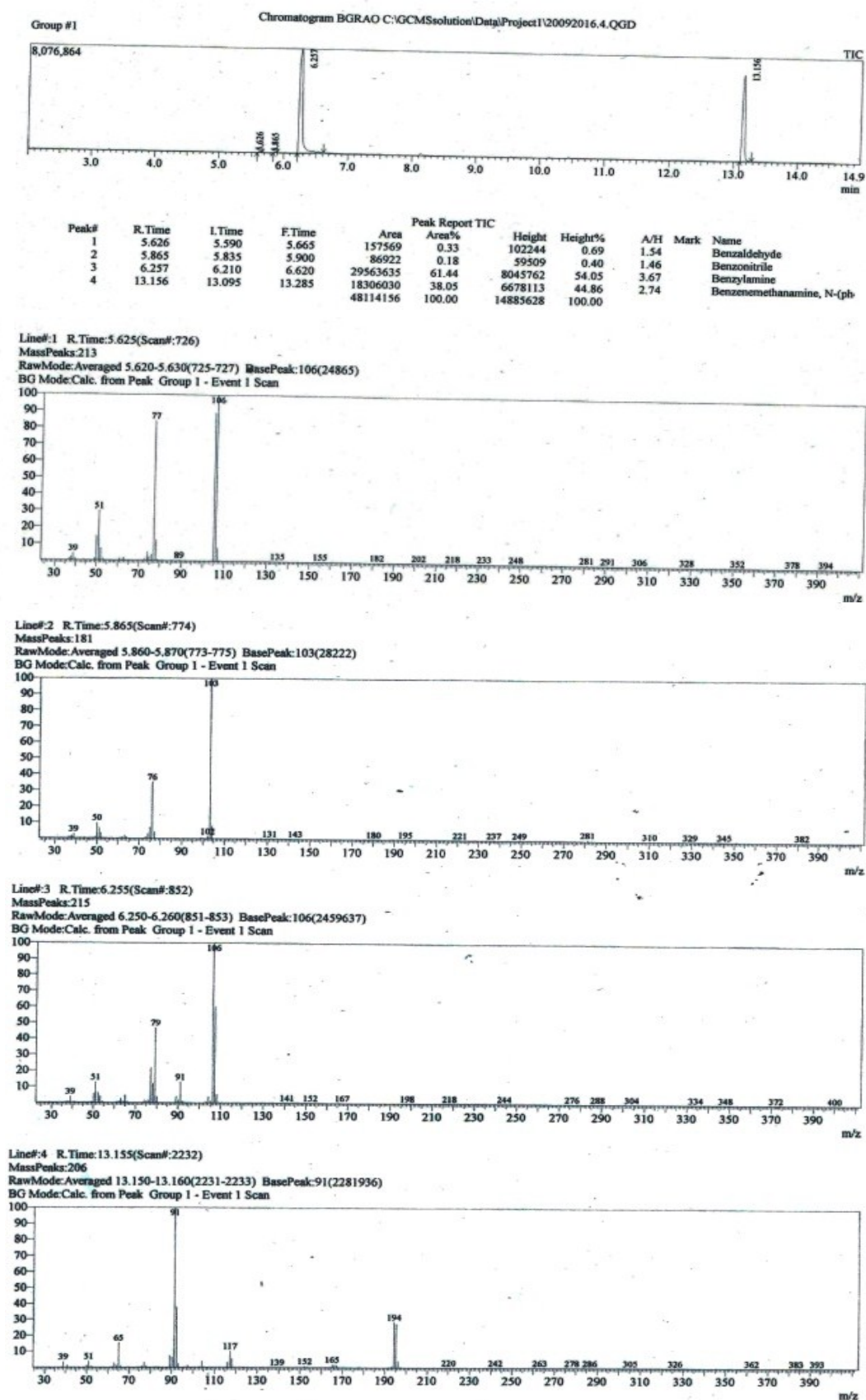


Fig. S6 GC-MS data of the reaction products.

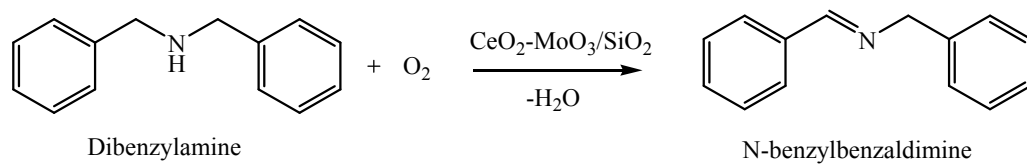


Fig. S7 Continuous-flow oxidation of dibenzylamine using CeO_2-MoO_3/SiO_2 solid acid catalyst.