

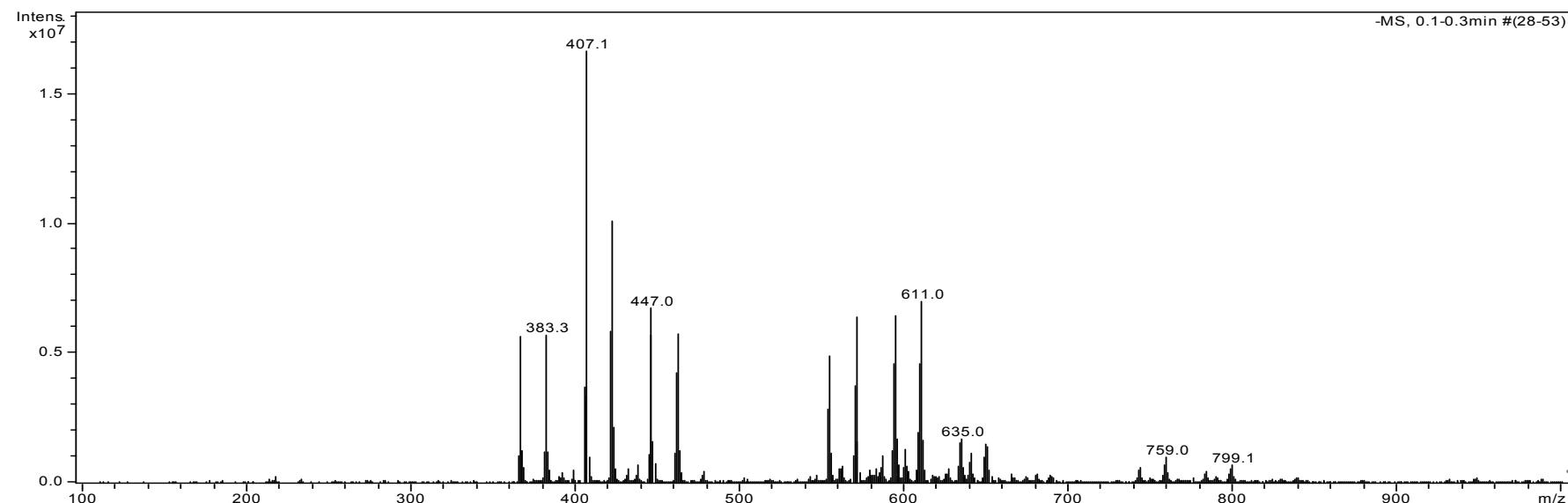
Electronic supplementary information

Chiral separation by nonaqueous capillary electrophoresis using L-sorbose–boric acid complexes as chiral ion-pair selectors

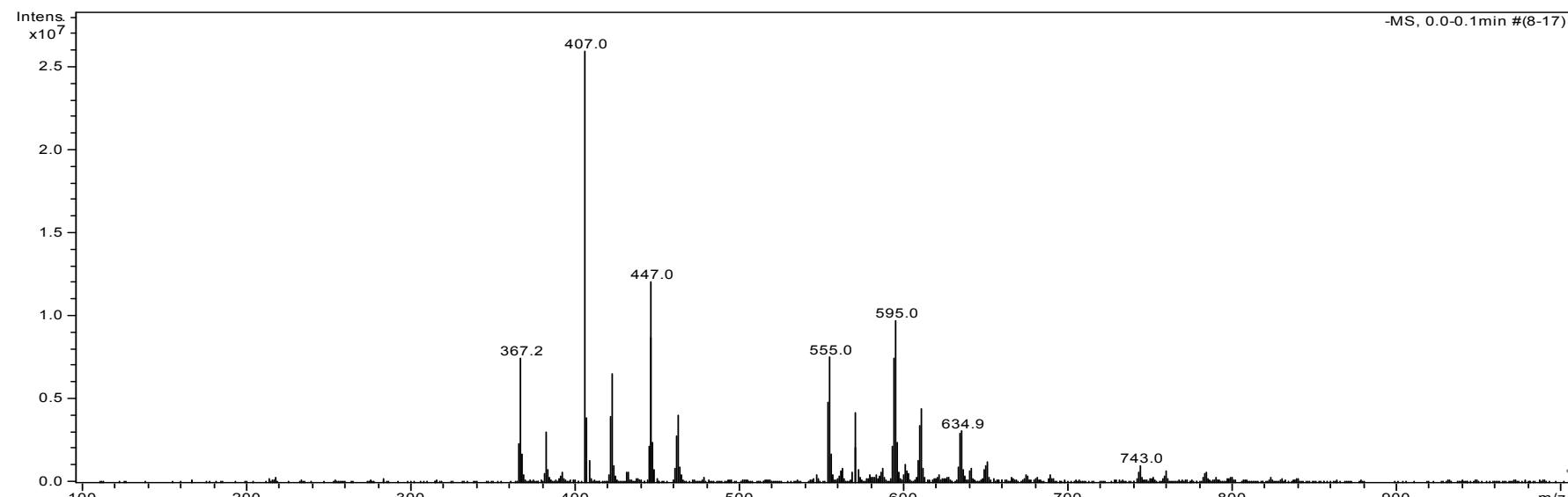
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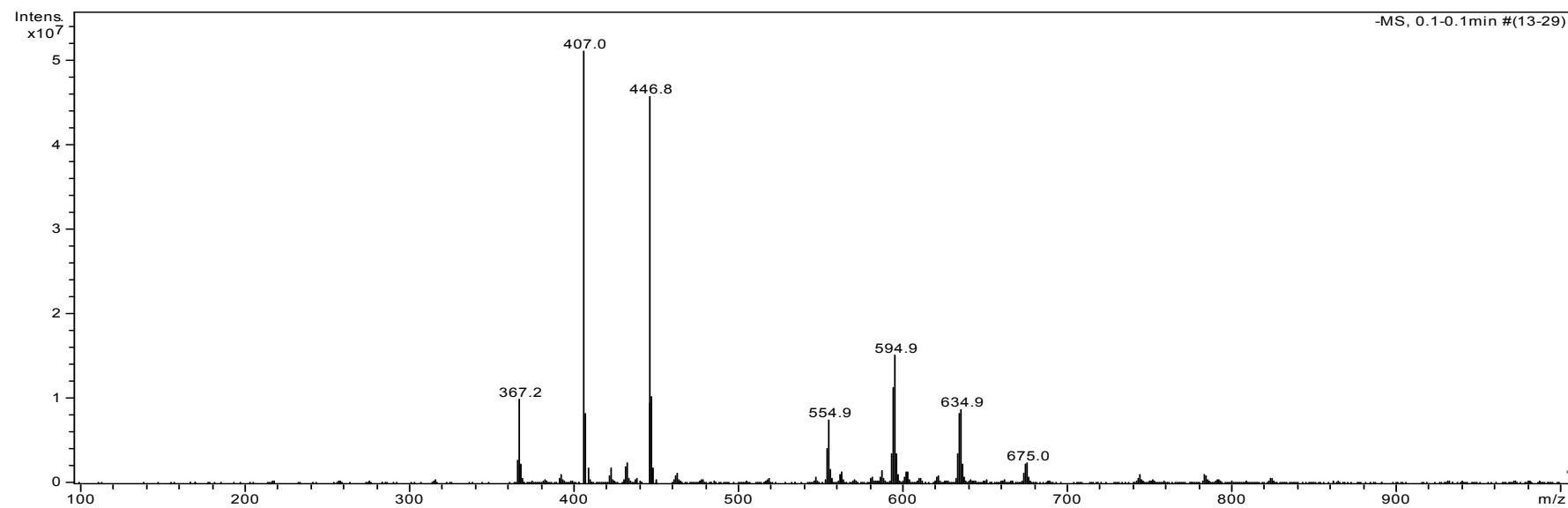
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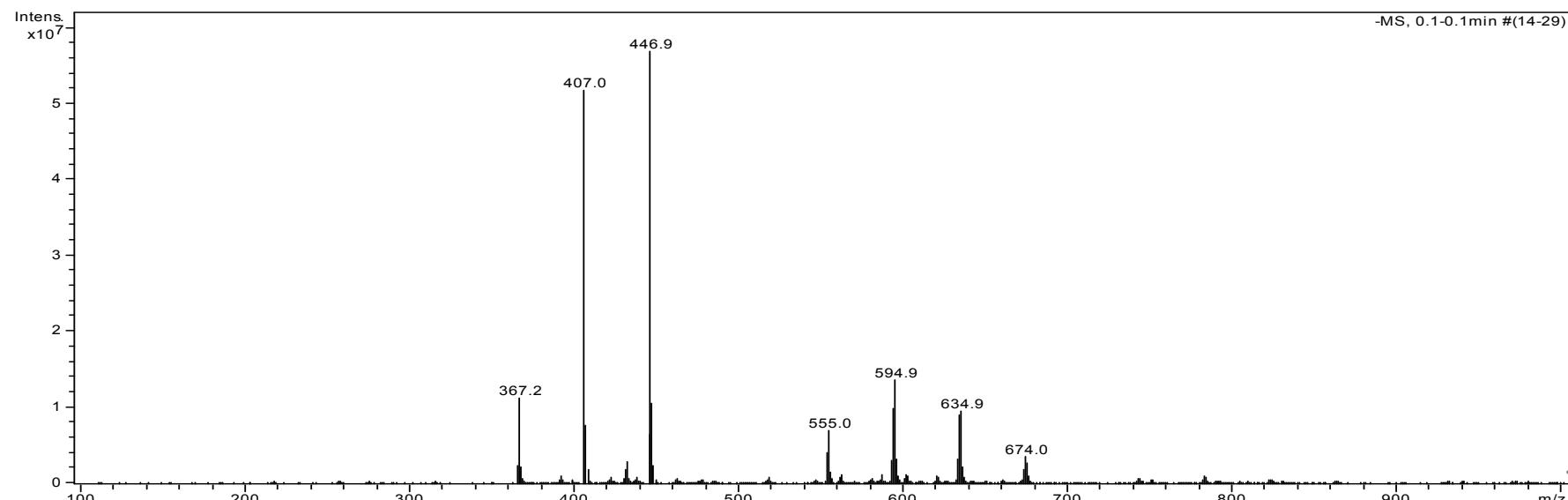
(A) 0 mM triethylamine



(B) 7.2 mM triethylamine



(C) 14.4 mM triethylamine



(D) 21.6 mM triethylamine

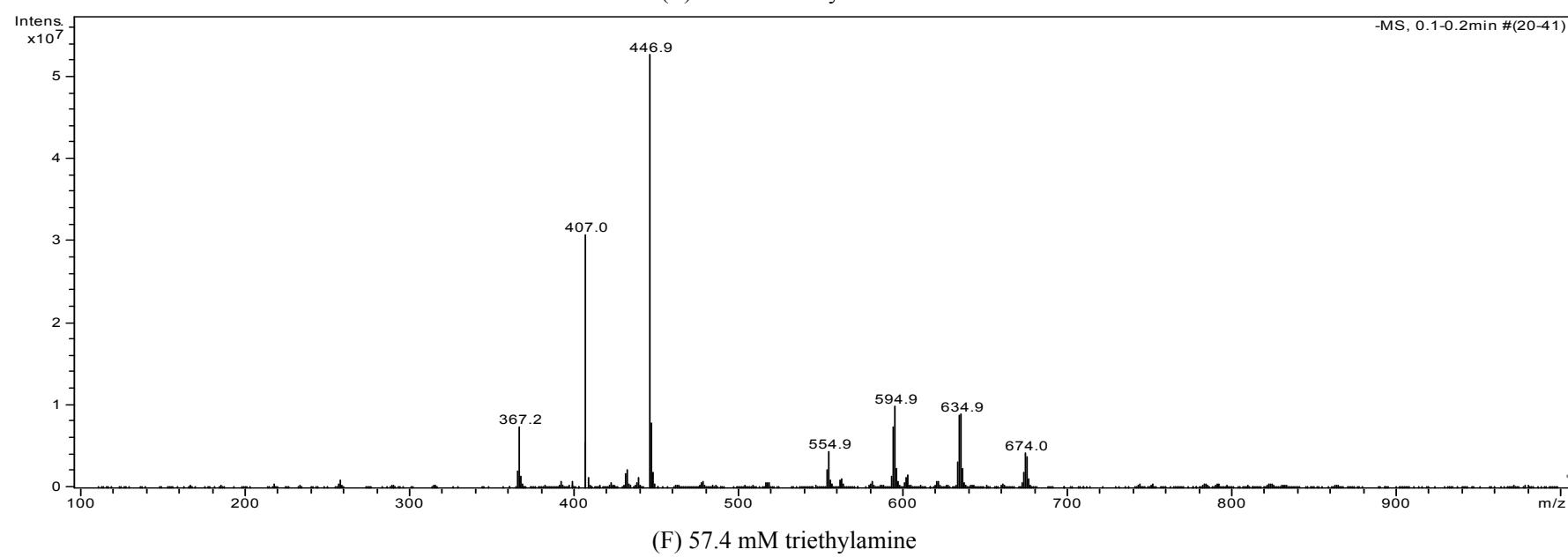
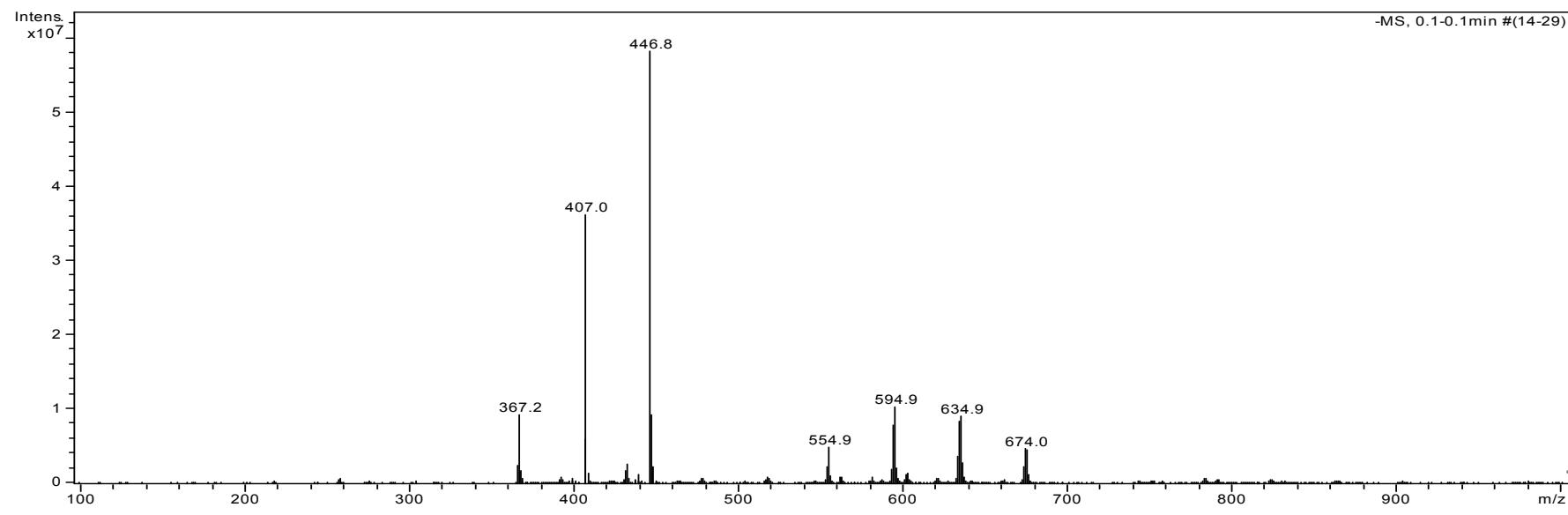
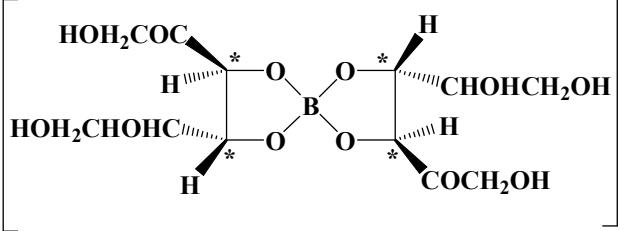
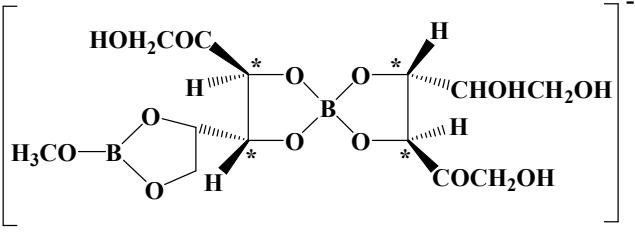
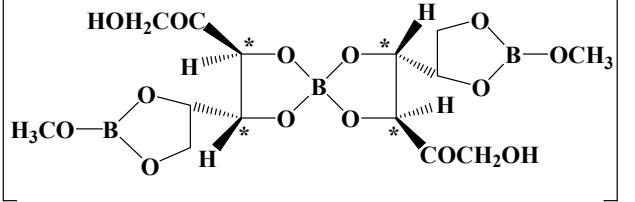


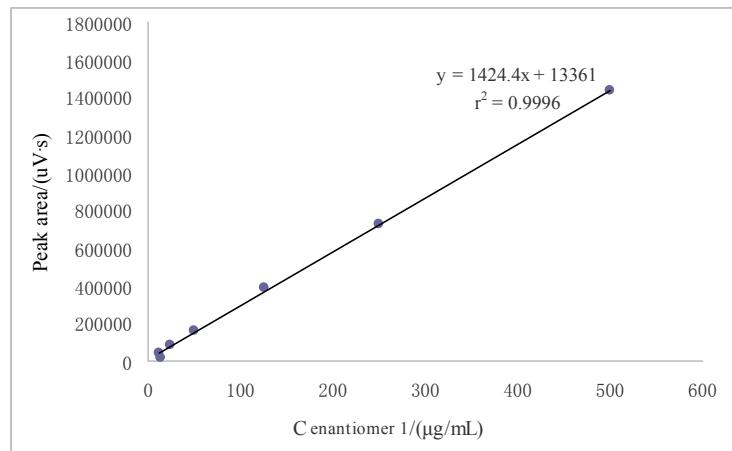
Fig. S1 MS spectra of NACE buffers with different concentrations of triethylamine.

Buffer composition in addition to triethylamine is 40 mM L-sorbose and 100 mM boric acid in methanol.

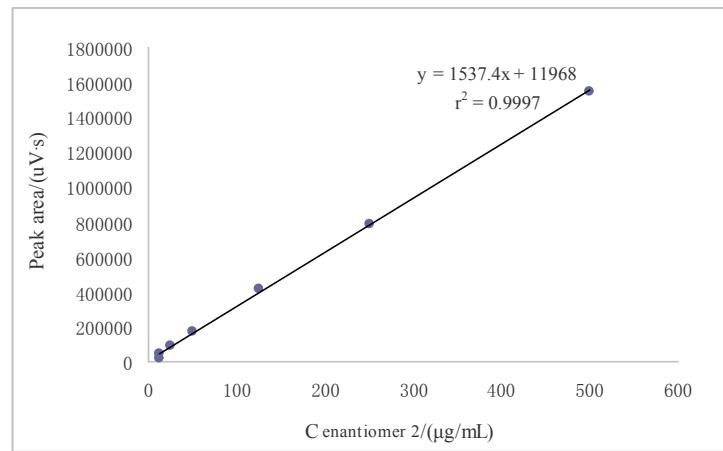
MS conditions: negative ESI-mode, spray voltage 3000V, nebulizer gas (N_2) flow set at 15 psi, dry gas (N_2) flow 5 L/min with a temperature of 325 °C.

Table S1 The structural formulae of pseudo-molecular ions in the MS experiments.

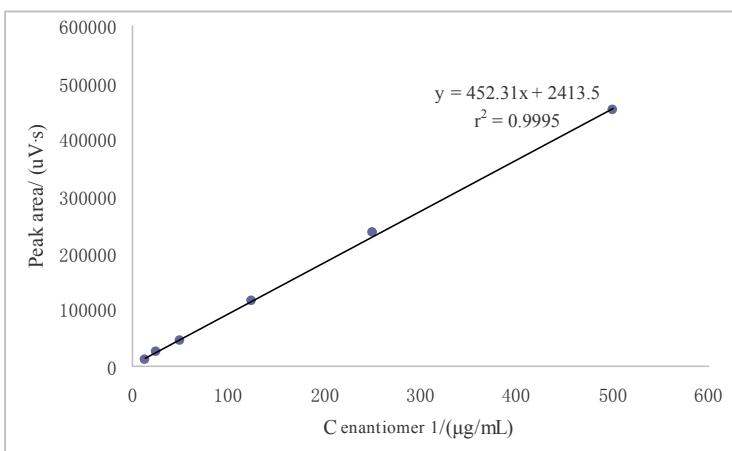
[M-H ⁺] ⁻	m/z	Structural formula
$[\text{C}_{12}\text{H}_{20}\text{BO}_{12}-\text{H}^+]$ ⁻	367	
$[\text{C}_{13}\text{H}_{21}\text{B}_2\text{O}_{13}-\text{H}^+]$ ⁻	407	
$[\text{C}_{14}\text{H}_{22}\text{B}_3\text{O}_{14}-\text{H}^+]$ ⁻	447	



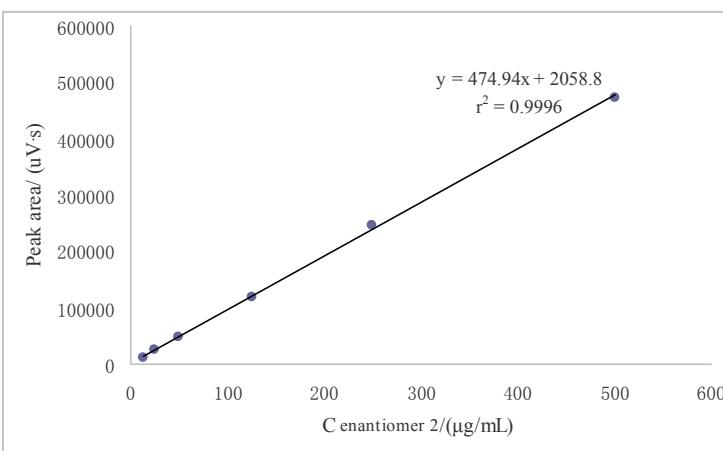
(A) The enantiomer 1 of clenbuterol



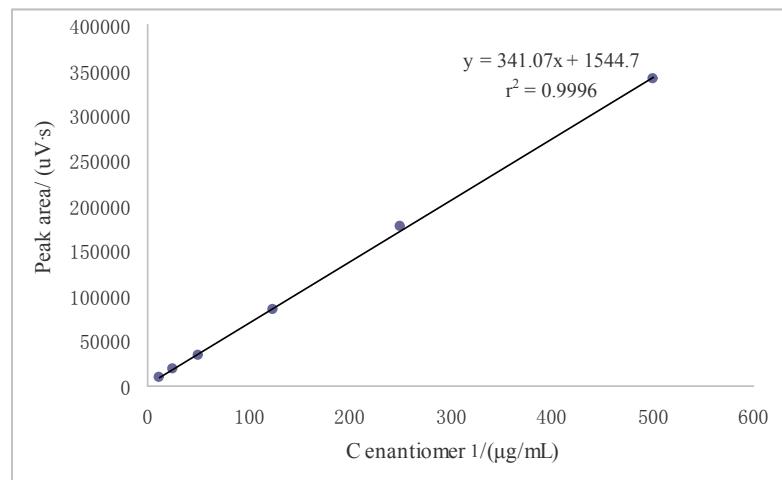
(B) The enantiomer 2 of clenbuterol



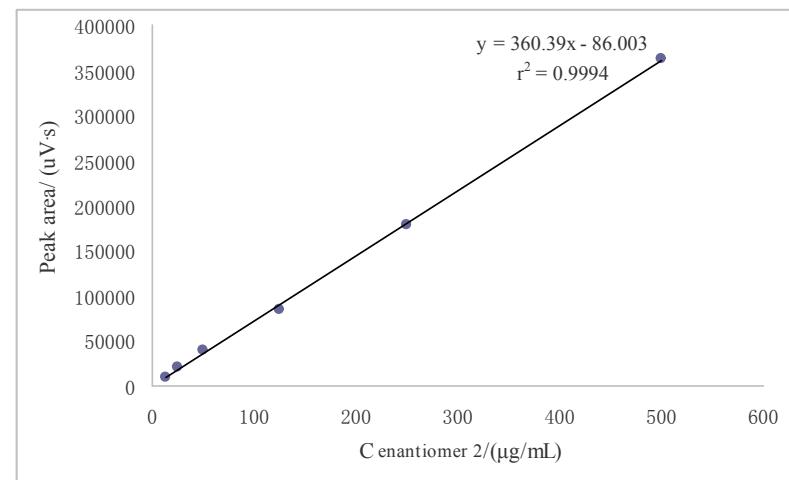
(C) The enantiomer 1 of esmolol



(D) The enantiomer 2 of esmolol



(E) The enantiomer 1 of metoprolol



(F) The enantiomer 2 of metoprolol

Fig. S2 The calibration curves and linear equation of the NACE method using L-sorbose–boric acid complexes as the chiral selectors in NACE. NACE conditions are the same as in Fig. 4. The concentration of each enantiomer was calculated as a half of its racemate.